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Plasmonic Core-Shell Ionic Microgels for Photo-Tuning Catalytic Applications

Yitong Wang, Ling Wang, Jingcheng Hao, Shuli Dong*

Key Laboratory of Colloid and Interface Chemistry & Key Laboratory of Special Aggregated Materials, Shandong University, Ministry of Education, Jinan 250100, P. R. China.

* Corresponding author. Tel: +86-531-88363768. E-mail: shuli@sdu.edu.cn

Supplementary Figures



Figure S1. UV-Vis spectra of Au nanorods.



Figure S2. TEM image of Au nanoparticles.



Figure S3. UV-Vis spectra of Au nanoparticles.



Figure S4. Zeta potential of Au NRs@PNV (200 μ L) with the addition of Au nanoparticles.



AuNRs@SiO2@PNV microgels and AuNRs@SiO2@PNV@Au NPs.

Figure



Figure S6. Variation of hydrodynamic diameter of Au NRs@PNV microgels with different concentrations as a function of NIR laser.



Figure S7. TEM image (a) and hydrodynamic diameter (b) of the nanocomposites after being irradiated.



Figure S8. UV-Vis spectra of 4-NP before and after adding NaBH₄ aqueous solution.



Figure S9. Time-dependent UV-Vis absorption spectra for the reduction of 4-NP with NaBH₄ in the presence of laser irradiation (a) and Au NRs (b).



Figure S10. TEM image (a) and hydrodynamic diameter (b) of the nanocomposites

after catalytic reaction.



Figure S11. Catalytic activity of Au NRs @PNV microgel@Au NPs catalyst after two cycles.

No	Catalyst	Time (s)	Rate constant, κ , (s ⁻¹)	Reference
1	Au@SiO ₂	1800	14 × 10 ⁻³	Adv. Mater. 2008, 20, 1523.
2	Au@CPF-1	720	5.05 × 10 ⁻³	Chem. Eur. J. 2016, 22,
				17029-17036.
3	Au ₍₀₎ @TpPa-1	780	5.35 × 10 ⁻³	Chem. Commun. 2014, 50,
				3169-3172.
4	Au@MgO	420	7.60 × 10 ⁻³	Green Chem. 2012, 14,
				3164-3174.
5	Au core/porous shell	286	3.65×10^{-3}	Nanoscale, 2016, 8, 11707-
	nanoparticles			11717.
6	Au NRs@PNV@Au NPs	240	18×10^{-3}	This work

Table S1. Summary of rate constants of other similar 4-nitrophenol reductionreactions catalyzed by previously reported Au-based catalysts.