

# Controllable supramolecular self-assemblies (rods-wires-spheres) and ICT/PET based perylene probe for palladium detection in solution and solid state

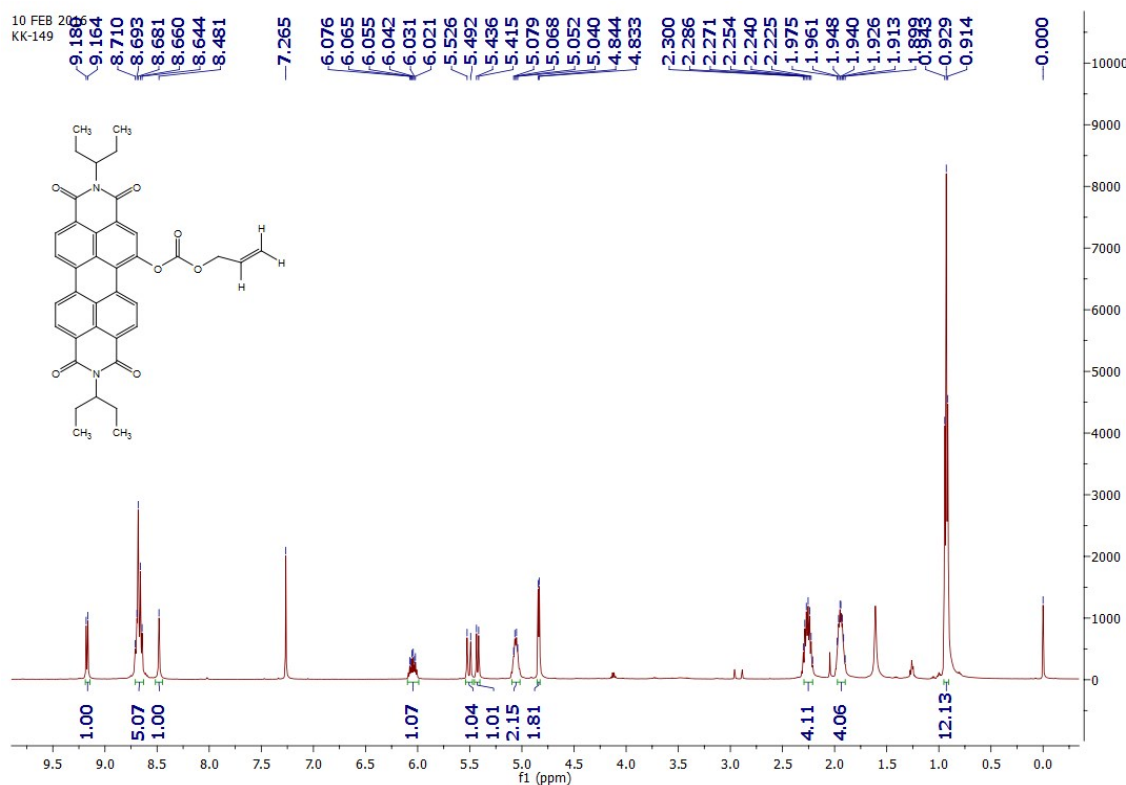
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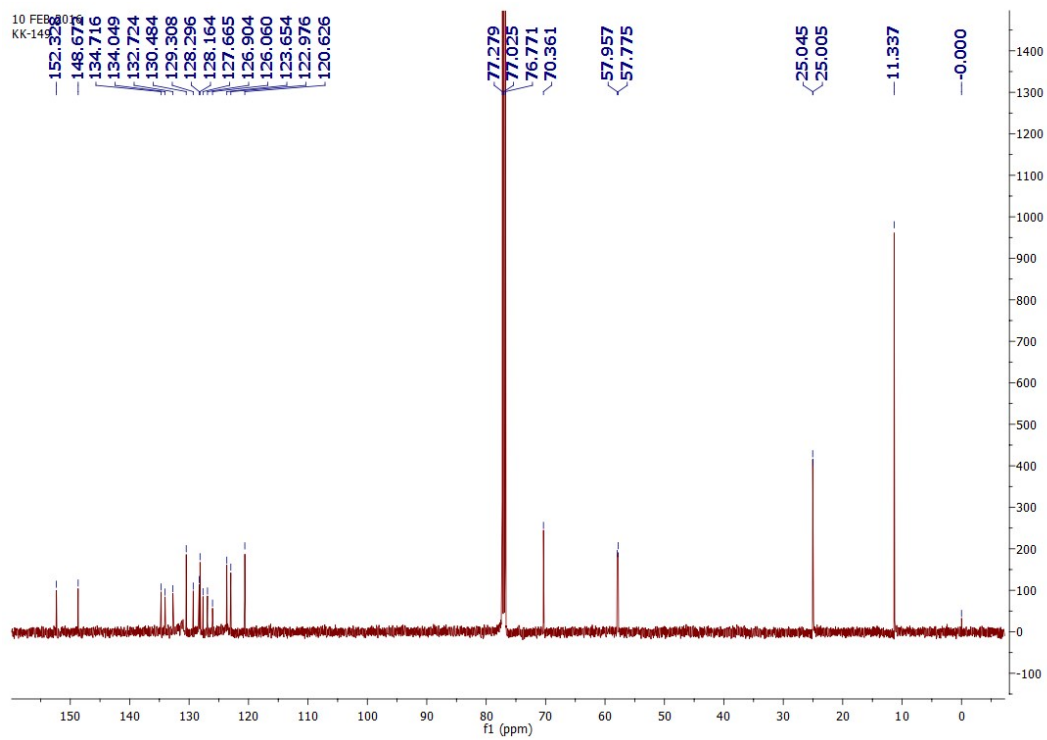
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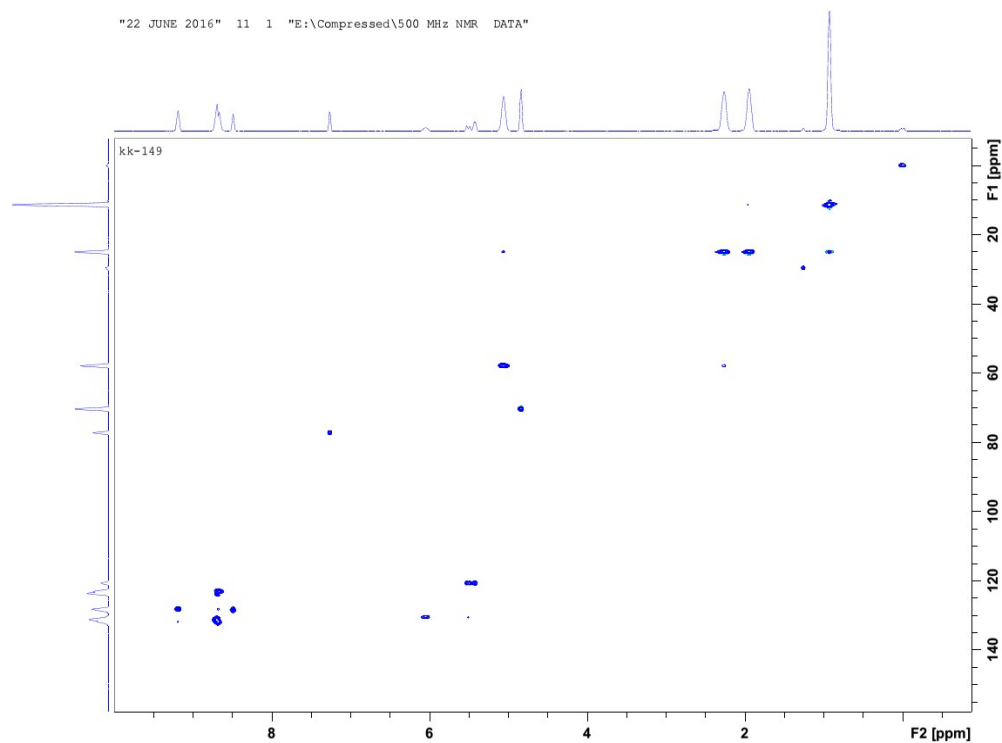
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## EXPERIMENTAL

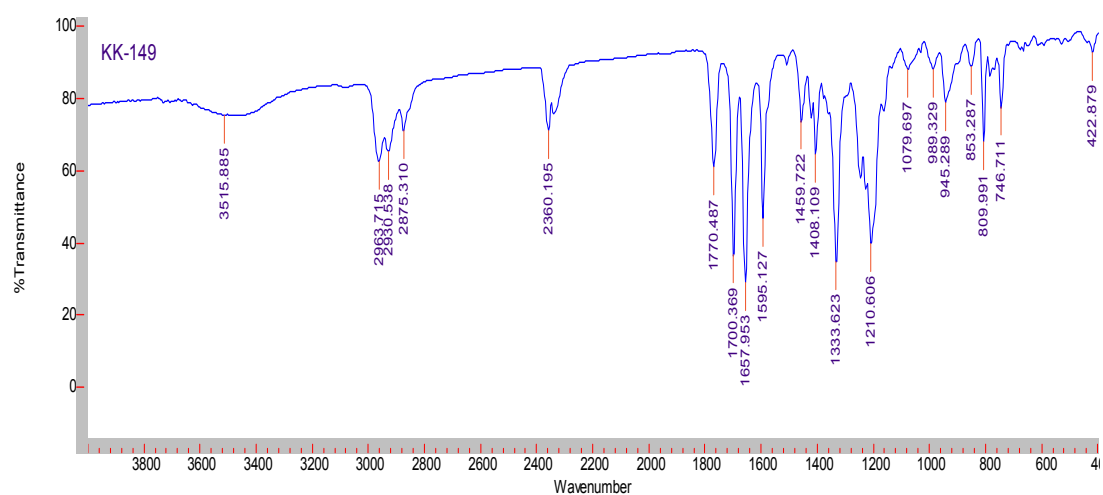




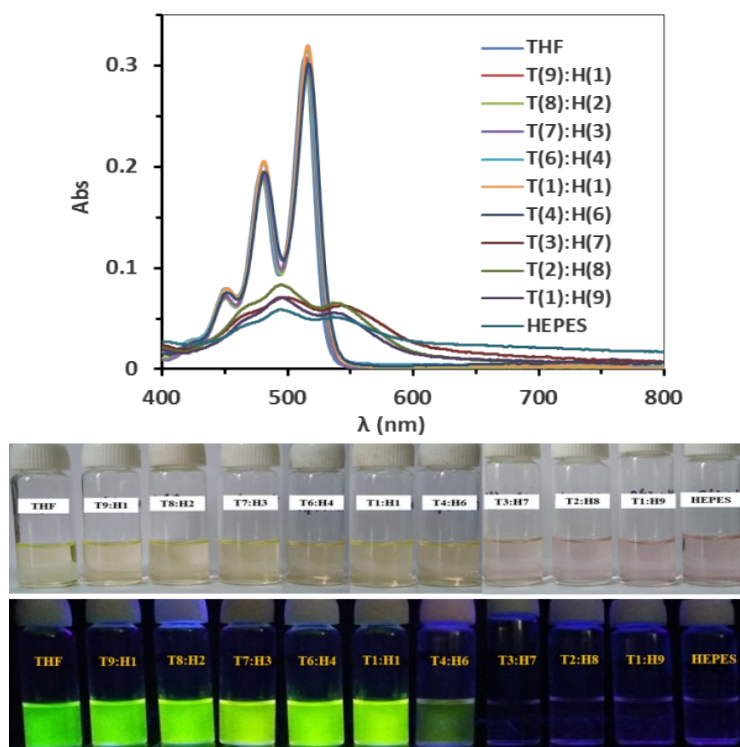
**Figure S1a:**  $^1\text{H}$  NMR and  $^{13}\text{C}$  Spectrum of AC-PDI.



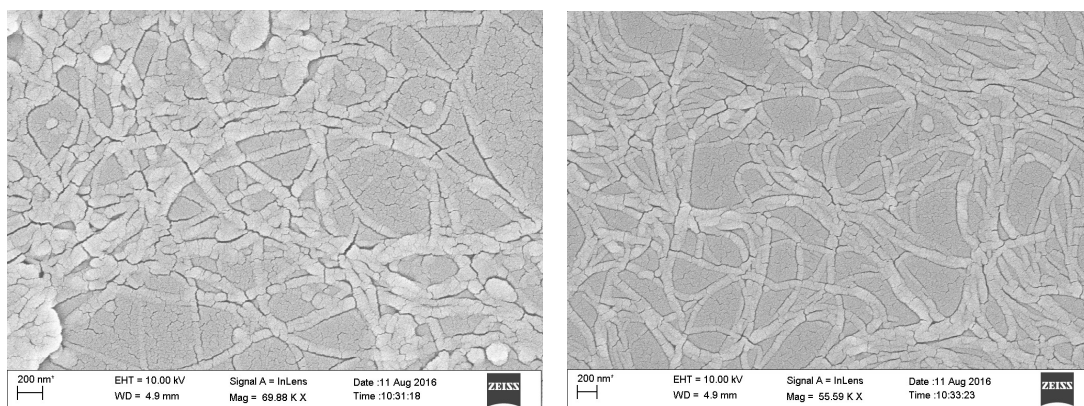
**Figure S1b:** HSQC spectrum of AC-PDI.



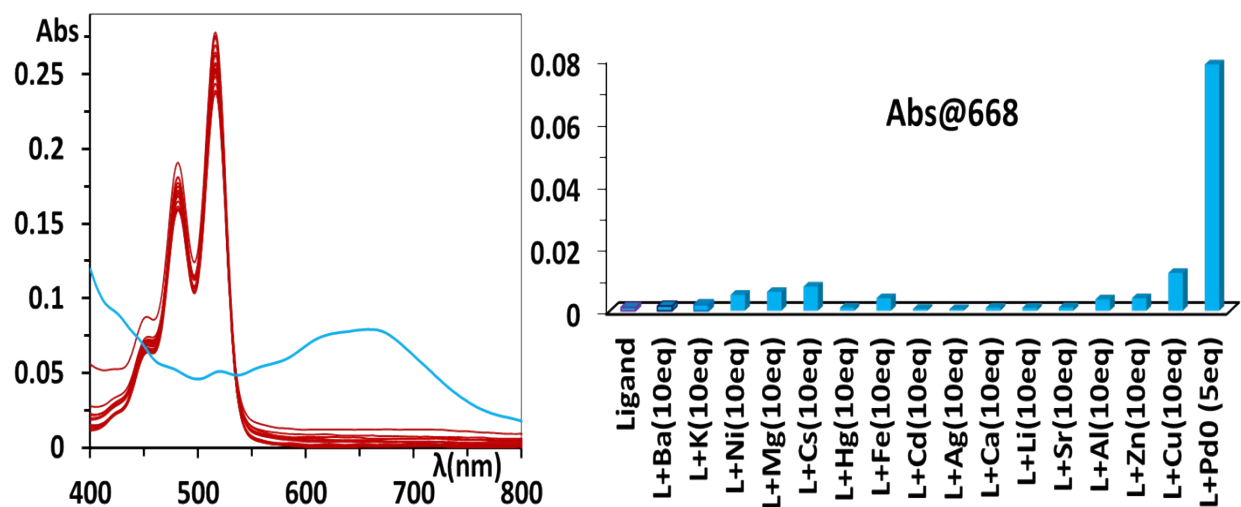
**Figure S1c:** IR spectrum of AC-PDI.



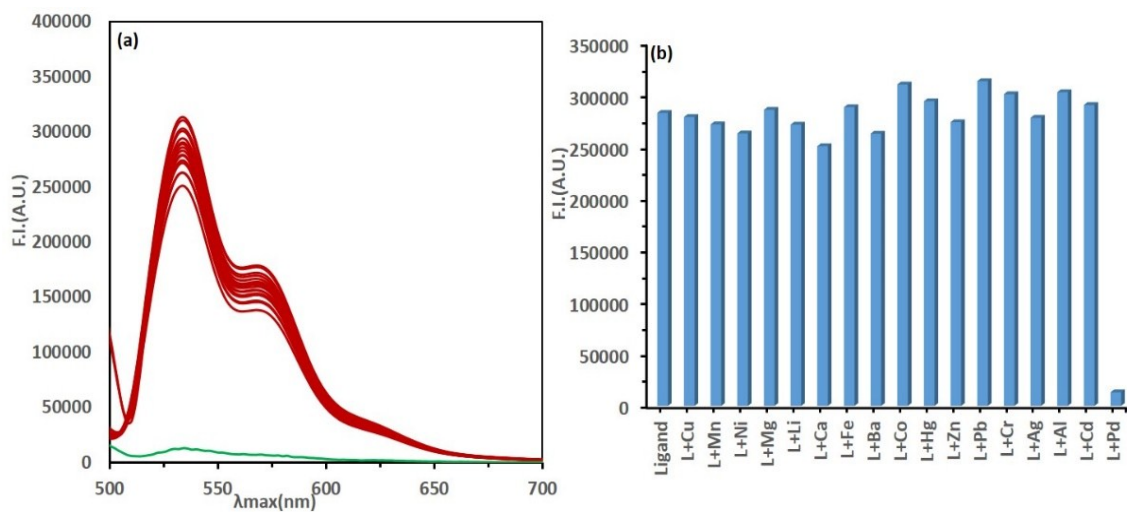
**Figure S2** UV-Vis absorption spectra of AC-PDI (10 μM) on gradual addition of H<sub>2</sub>O in THF (top); room light and after 365 nm UV irradiation photographs showing the visible colour and emission changes of AC-PDI in aqueous THF mixture (bottom); T = THF; D = DMF; H = H<sub>2</sub>O in v/v ratio.



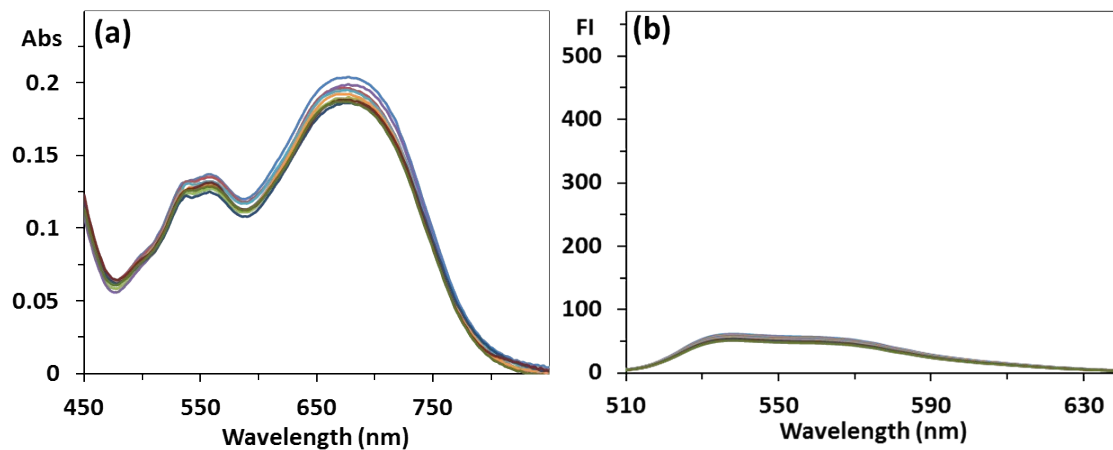
**Figure S3** SEM micrographs of self-assembled regular nanorods/tube of AC-PDI (10 μM) in methylcyclohexane.



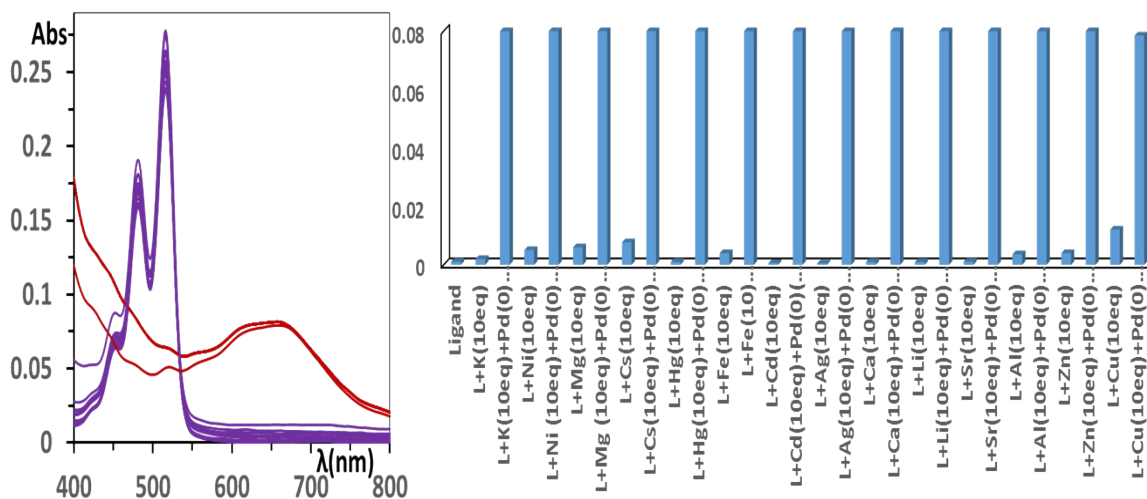
**Figure S4a** UV-Vis absorption spectrum and (b) bar diagram showing the effect of individual metal ions on the absorption spectra of ACF-PDI



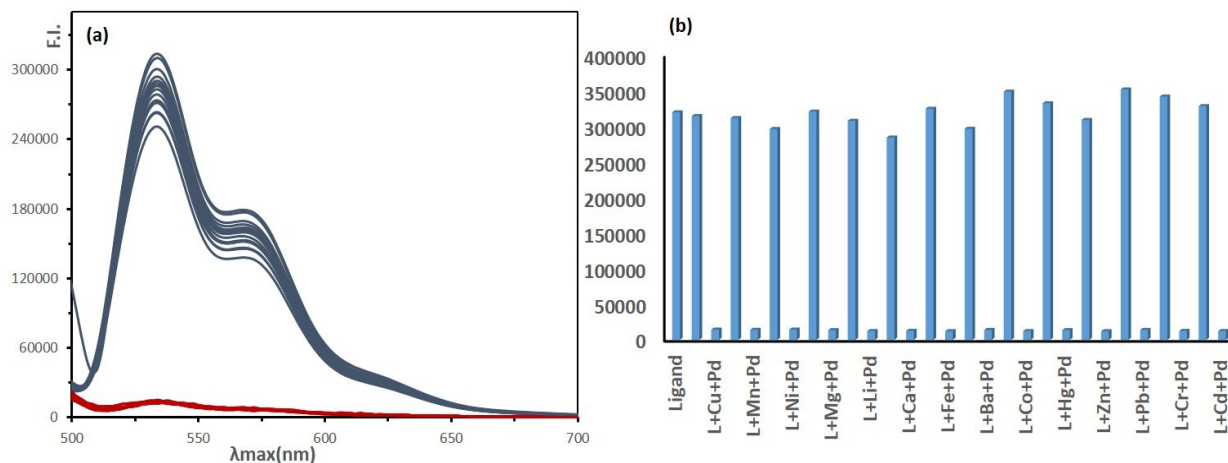
**Figure S4b** Emission spectra of AC-PDI and bar diagram indicating the effect of Pd<sup>0</sup> and other metal ions ( $\lambda_{\text{ex}} = 490$  nm; Ex = 1:1 nm, Em = 0.5:0.5 nm).



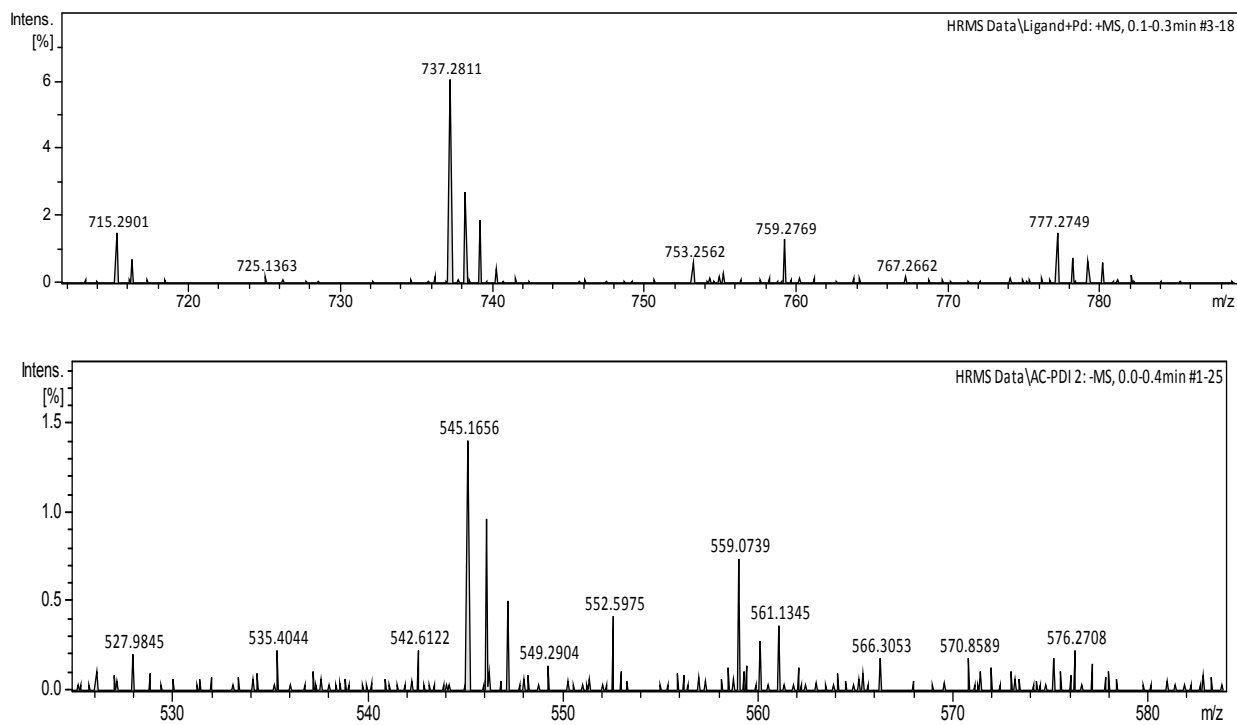
**Figure S5** (a) UV-Vis and (b) Fluorescence spectra of **PDI 1** (10 μM) on incremental addition of Pd<sup>0</sup> in HEPES buffer-CH<sub>3</sub>CN (1:1, v/v, pH 7.2).



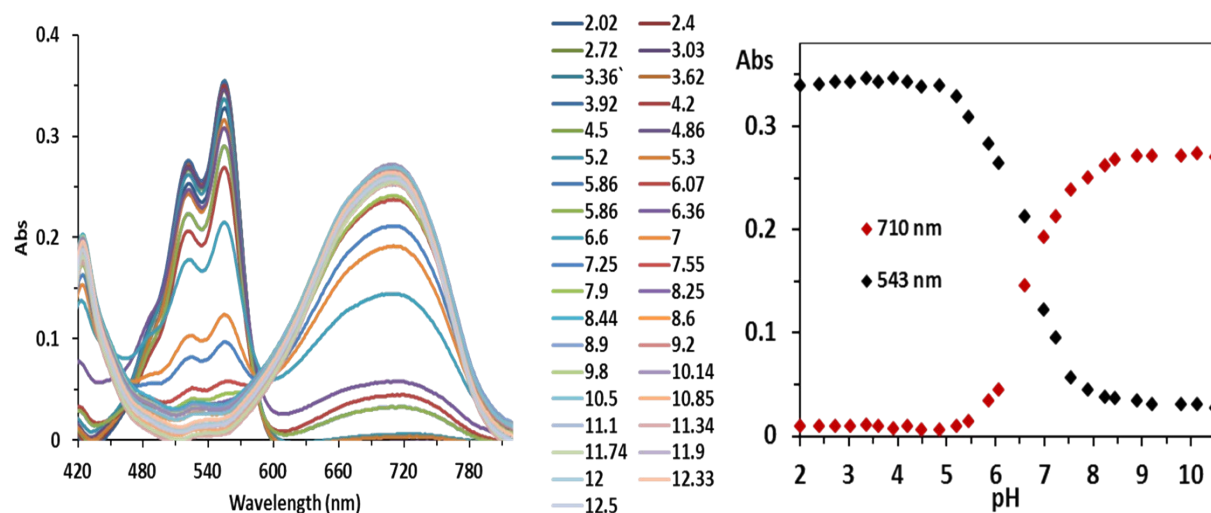
**Figure S6a** (a) UV-Vis absorption spectra and (b) bar diagram showing the effect of individual metal ions (100 μM) on the absorption spectra of **AC-PDI**.



**Figure S6b** (a) Emission spectra and (b) bar graph of AC-PDI representing the response towards  $\text{Pd}^0$  in the presence of other metal ions ( $\lambda_{\text{ex}} = 490 \text{ nm}$ ; Slit width;  $\text{Ex} = 1:1 \text{ nm}$ ,  $\text{Em} = 0.5:0.5 \text{ nm}$ )



**Figure S7** HRMS spectrum of AC-PDI +  $\text{Pd}^0$  in positive and negative mode.



**Figure S8.** (Left) Effect of pH changes on the UV-Vis absorption spectrum of **PDI-OH** (10  $\mu\text{M}$ ) in water (50% THF); (Right) Plot of absorbance at 543 nm and 710 nm vs. change in pH of the solution.

**Table S1:** Time dependent and concentration dependent absorbance data of **AC-PDI** (10  $\mu\text{M}$ ) on addition of  $\text{Pd}^0$  in (a) HEPES buffer-THF (1:1, v/v, pH 7.2); (b) HEPES buffer-DMSO (3:7, v/v, pH 7.2).

Conc. of $\text{Pd}^0$	Time taken	50% $\text{H}_2\text{O}:\text{CH}_3\text{CN}$		Conc. of $\text{Pd}^0$	Time taken	30% $\text{H}_2\text{O}:\text{DMSO}$	
		Rate constant $K_1$	Rate constant $K_2$			Rate constant $K_1$	Rate constant $K_2$
45 $\mu\text{M}$	26 min	0.25 $\text{min}^{-1}$ (up to 0-4 min)	0.07 $\text{min}^{-1}$ (up to 4-18 min)	35 $\mu\text{M}$	20 min	0.23 $\text{min}^{-1}$ (up to 0-8 min)	0.08 $\text{min}^{-1}$ (up to 8-20 min)
90 $\mu\text{M}$	18 min	0.30 $\text{min}^{-1}$ (up to 0-4 min)	0.07 $\text{min}^{-1}$ (up to 4-16 min)	70 $\mu\text{M}$	12 min	0.36 $\text{min}^{-1}$ (up to 0-4 min)	0.10 $\text{min}^{-1}$ (up to 4-12 min)
135 $\mu\text{M}$	16 min	0.22 $\text{min}^{-1}$ (up to 0-6 min)	0.08 $\text{min}^{-1}$ (up to 6-14 min)	105 $\mu\text{M}$	12 min	0.48 $\text{min}^{-1}$ (up to 0-4 min)	0.12 $\text{min}^{-1}$ (up to 4-12 min)