

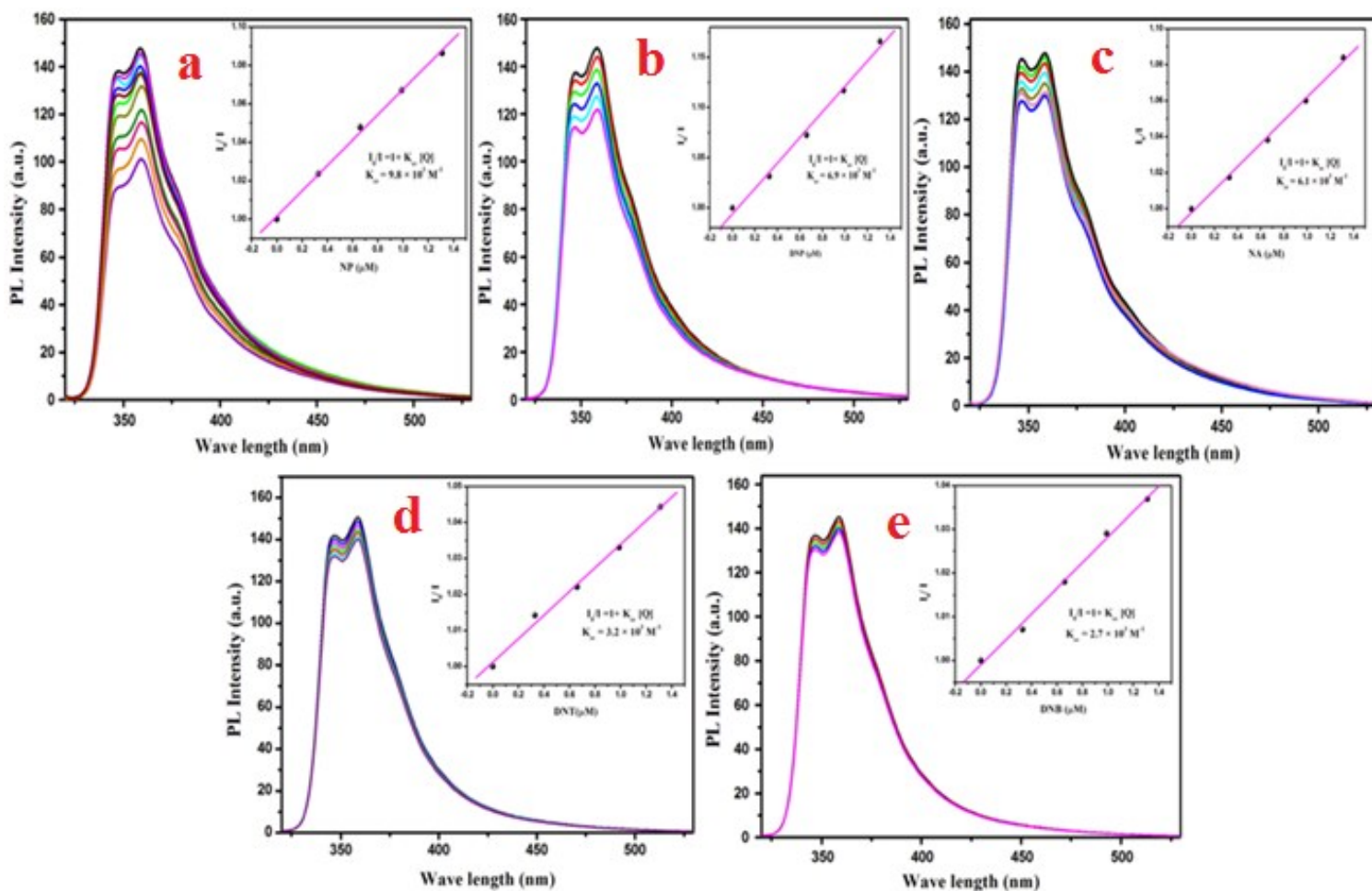
## Supporting Information

### Proton triggered emission and selective sensing of 2,4,6-trinitrophenol by fluorescent hydrosol of 2-phenylquinoline

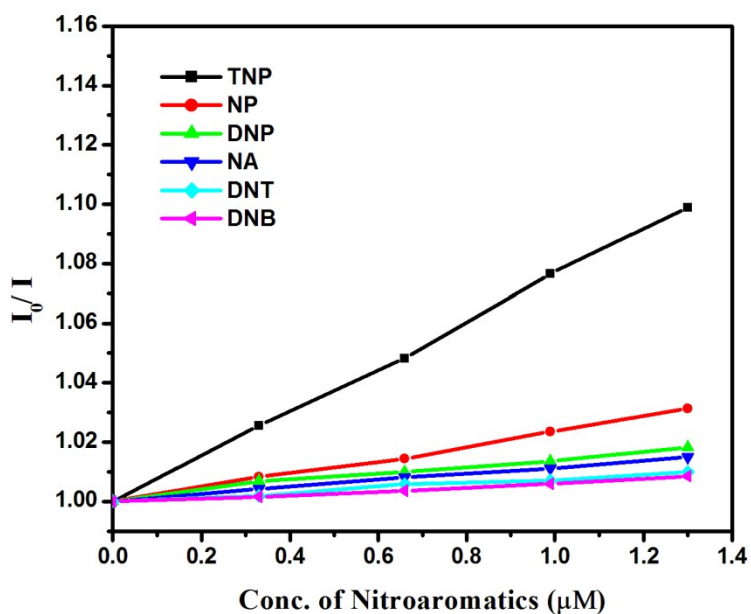
Samir Maity, Milan Shyamal, Debasish Das, Ashim Maity, Sudipto Dey and Ajay Misra

Department of Chemistry and Chemical Technology, Vidyasagar University,

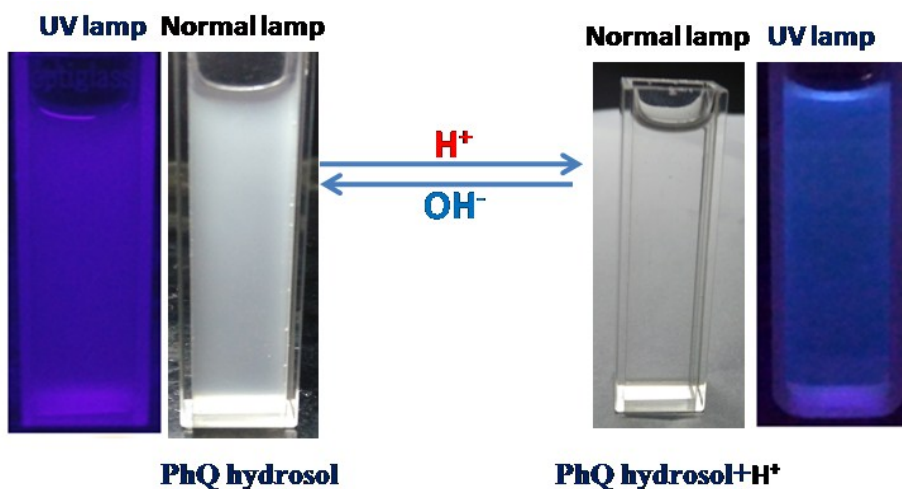
Midnapore 721 102, W.B, India.



**Fig.S1** Fluorescence Quenching of PhQ hydrosol in presence of (a) 4-Nitro-phenol (NP), (b) 2,4-dinitro-phenol (DNP), (c) 4-Nitroaniline (NA), (d) Dinitro-toluene (DNT), (e) Dinitro-benzene (DNB).



**Fig.S2** Comparison of Stern-Volmer plot of different nitroaromatics.



**Fig.S3:** Photographic picture of PhQ and protonated PhQ hydrosol under normal light and UV light (365 nm) irradiation.

**Table S1:** Stern-Volmer quenching constant ( $K_{SV} \text{ M}^{-1}$ ) for different nitroaromatics

<b>Nitroaromatics</b>	<b>R</b>	<b><math>K_{SV} (\text{M}^{-1})</math></b>
<b>TNP</b>	<b>0.9930</b>	<b><math>7.8 \times 10^4</math></b>
<b>NP</b>	<b>0.9981</b>	<b><math>9.8 \times 10^3</math></b>
<b>DNP</b>	<b>0.9995</b>	<b><math>6.9 \times 10^3</math></b>
<b>NA</b>	<b>0.9967</b>	<b><math>6.1 \times 10^3</math></b>
<b>DNT</b>	<b>0.9975</b>	<b><math>3.2 \times 10^3</math></b>
<b>DNB</b>	<b>0.9983</b>	<b><math>2.7 \times 10^3</math></b>

**Table S2:** A comparative study of our work with the other reported works in the literature with reference to  $K_{sv}$ , Detection limit and Anti interference study.

<b>References</b>	<b><math>K_{sv} (\text{M}^{-1})</math></b>	<b>Detection limit</b>	<b>Anti interference</b>
<b>Anal. Methods, 2013,5,6228-6233</b>	<b>0.33</b>	<b>1 <math>\mu\text{M}</math></b>	<b>-</b>
<b>Dalton Trans. 2013,42,12403-12409</b>	<b>-</b>	<b><math>4.5 \times 10^{-7} \text{ M}</math></b>	<b>-</b>
<b>ACS Appl. Mater. Interfaces 2013, 5, 672–679</b>	<b><math>1.55 \times 10^4</math></b>	<b><math>3.5 \times 10^{-7} \text{ M}</math></b>	<b>-</b>
<b>RSC Adv., 2016, 6, 37929-37932</b>	<b>-</b>	<b><math>2.4 \times 10^{-7} \text{ M}</math></b>	<b>-</b>
<b>J. Mol. Liq. 224, 2016, 255–264</b>	<b><math>6.02 \times 10^3</math></b>	<b><math>2.98 \times 10^{-6} \text{ M}</math></b>	<b>-</b>
<b>J. Photo Chem. &amp; Photo Biology A. 2011,217,356-362</b>	<b><math>5.1 \times 10^4</math></b>	<b><math>1 \times 10^{-8} \text{ M}</math></b>	<b>-</b>
<b>Org. Lett. 2012,14,3112-3115</b>	<b><math>6.9 \times 10^4</math></b>	<b>500 ppb</b>	<b>-</b>

<b>J.Mater.Chem.A,2013, 1,8745-8752</b>	<b>3.4 x 10<sup>4</sup></b>	<b>8 x 10<sup>-8</sup> M</b>	<b>-</b>
<b>Chem.commun.,2013, 49,4764</b>	<b>-</b>	<b>70 nM</b>	<b>-</b>
<b>J.Org. Chem., 2013,78,1306- 1310</b>	<b>3.3 x 10<sup>4</sup></b>	<b>354 ppb</b>	<b>-</b>
<b>J. Mater. Chem., 2012,22,11574</b>	<b>3.04 x 10<sup>4</sup></b>	<b>2.38 x 10<sup>-8</sup> M</b>	<b>-</b>
<b>Our present study</b>	<b>7.8 x 10<sup>4</sup></b>	<b>4.3 x 10<sup>-7</sup> M</b>	<b>Yes</b>