

Supporting Information

The NICS Values Scan in Three-dimensional Space of the hoop-shaped π -conjugated molecules:

[6]₈cyclacene and [16]trannulene

Qing Li,^{a,b}Hong-Liang Xu^{a*} and Zhong-Min Su^{a,b,*}

^a*Institute of Functional Material Chemistry, Faculty of Chemistry, Northeast Normal University, Changchun, Jilin 130024, China.*

^b*Department of Chemistry, Faculty of Science, Yanbian University, Yanji, Jilin, 133002, China.*

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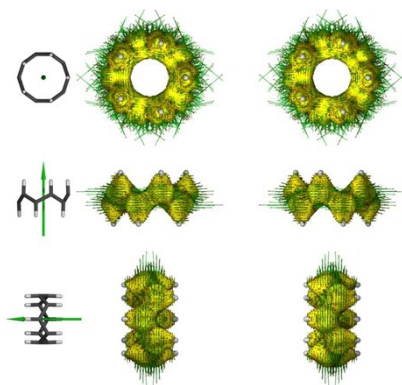


Figure S1. The ACID of [10]trannulene.

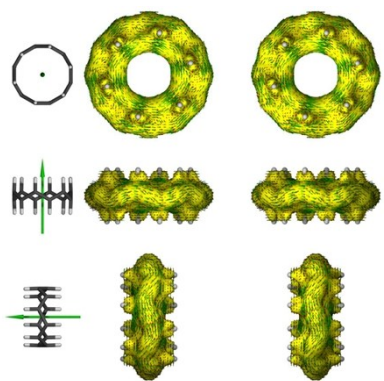


Figure S2. The ACID of [12]trannulene.

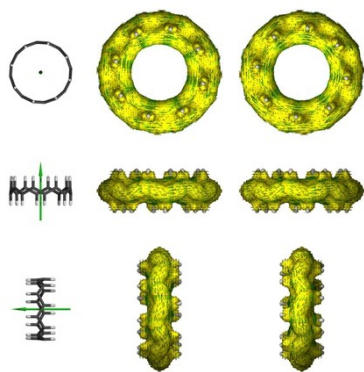


Figure S3. The ACID of [14]trannulene.

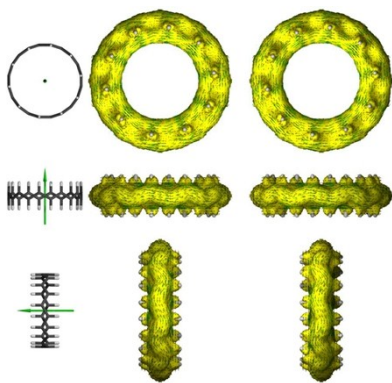


Figure S4. The ACID of [20]trannulene.

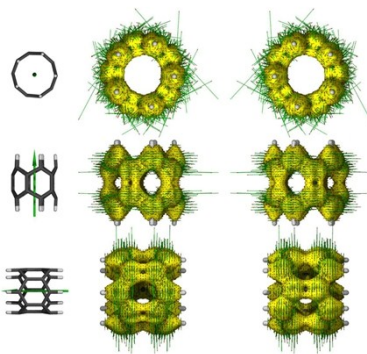


Figure S5. The ACID of [6]₅cyclacene.

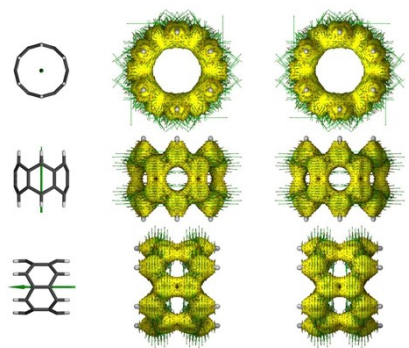


Figure S6. The ACID of [6]₆cyclacene.

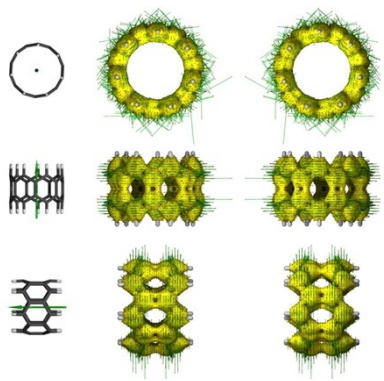


Figure S7. The ACID of [6]cyclacene.

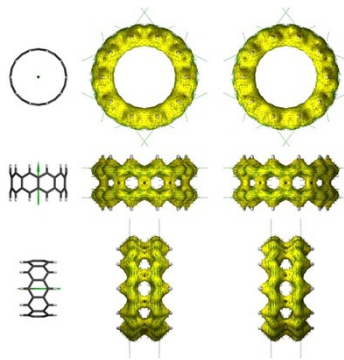


Figure S8. The ACID of [6]₁₀cyclacene.

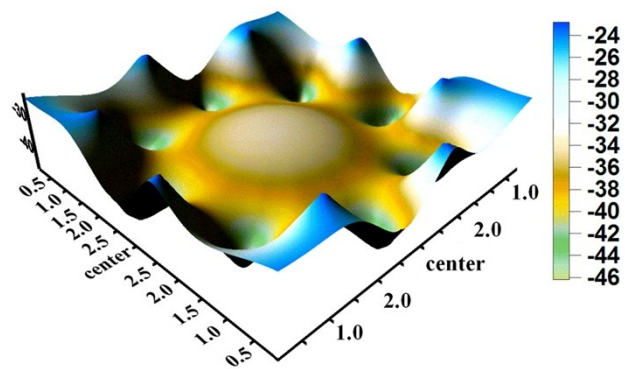


Figure S9. The planar NICS values scan of $P(L=0.2)$ for $[6]_8$ cycloacene.

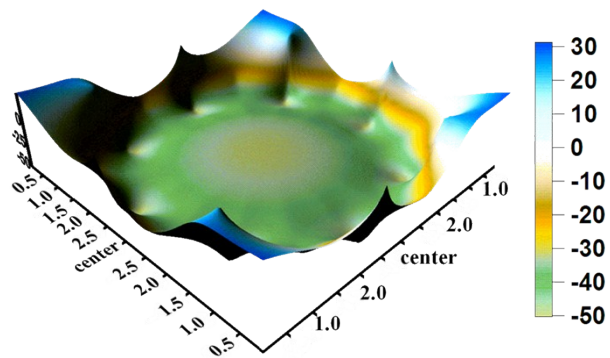


Figure S10. The planar NICS values scan of $P(L=0.6)$ for $[6]_8$ cycloacene.

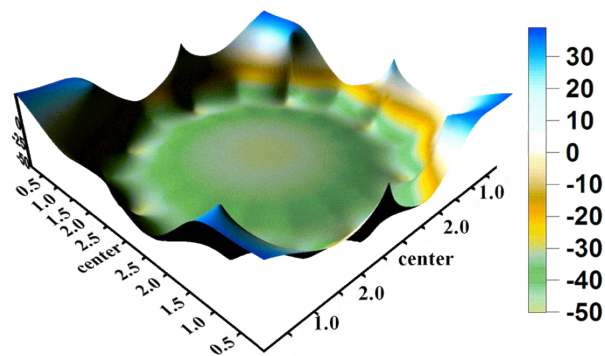


Figure S11. The planar NICS values scan of $P(L=0.8)$ for $[6]_8$ cycloacene.

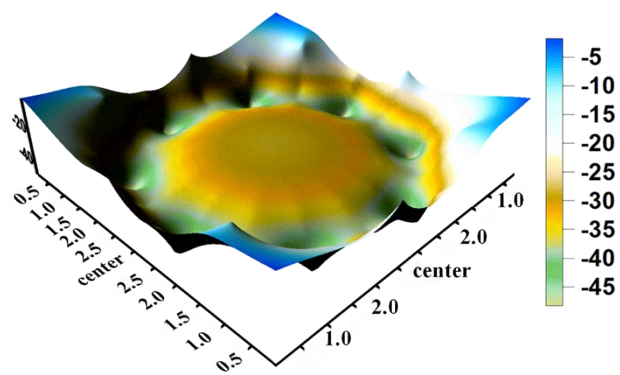


Figure S12. The planar NICS values scan of $P(L=1.2)$ for $[6]_8$ cyclacene.

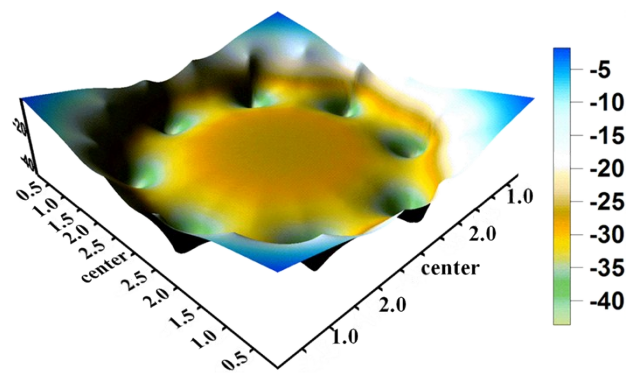


Figure S13. The planar NICS values scan of $P(L=1.4)$ for $[6]_8$ cycloacene.

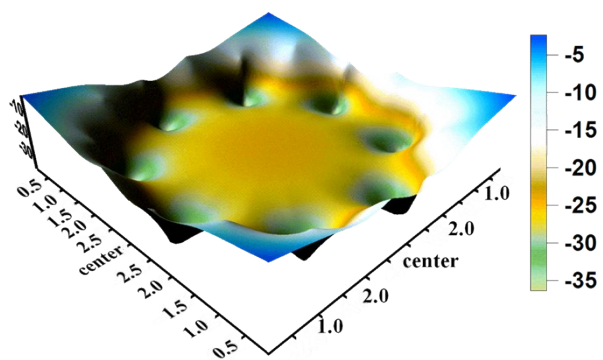


Figure S14. The planar NICS values scan of $P(L=1.6)$ for $[6]_8$ cycloacene.

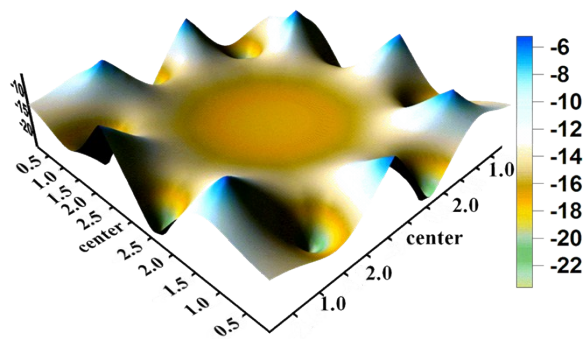


Figure S15. The planar NICS values scan of $P(L=2.4)$ for $[6]_3$ cycloclacene.

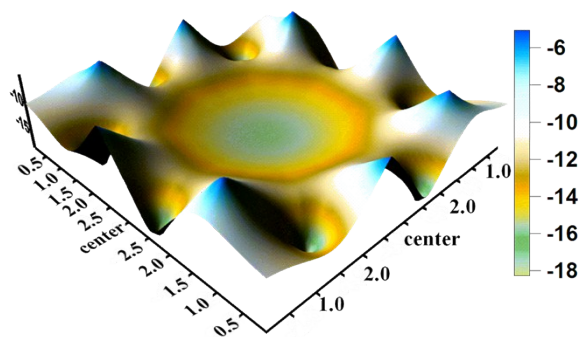


Figure S16. The planar NICS values scan of $P(L=2.6)$ for $[6]_3$ cycloacene.

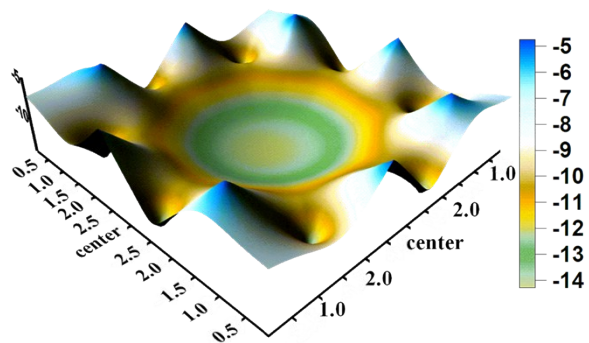


Figure S17. The planar NICS values scan of $P(L=2.8)$ for $[6]_8$ cyclacene.

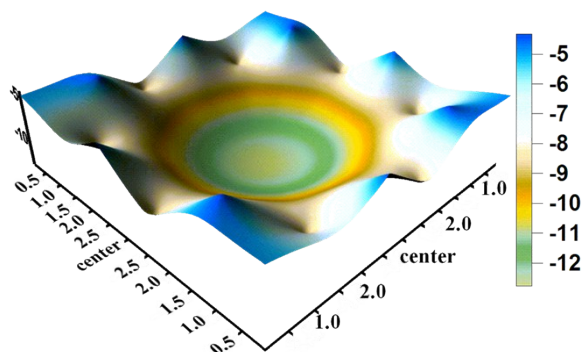


Figure S18. The planar NICS values scan of $P(L=3.0)$ for $[6]_8$ cyclacene.

The situation between the internal magnetic field during the course of two $[16]$ trannulenes stacked into $[6]_8$ cyclacene.

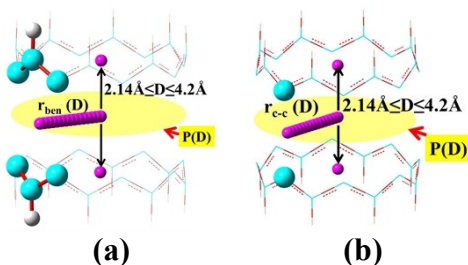


Figure S19. The diagram for the NICS values scan about the center cross-section $P(D=2.14\sim 4.20\text{Å})$ in the stack process.

In order to explicitly describe the situation between the internal magnetic field during the course of two $[16]$ trannulenes stacked into $[6]_8$ cyclacene. We simulated the change of the center cross-section internal magnetic field in the stack process by the NICS values scan method. Firstly, the center cross-section is defined as $P(D=2.14\sim 4.20\text{Å})$, in which D ($2.14\sim 4.20\text{Å}$) is defined as the distance of two $[16]$ trannulenes in the stack process (Figure S19). On the other hand, the two kinds radius ($r_{\text{ben}}(D=2.14\sim 4.20\text{Å})$ and $r_{\text{c-c}}(D=2.14\sim 4.20\text{Å})$) in center cross-section are listed in Figure S19. The distance from the center of the six-member ring to the center of the cross-section is represented by the $r_{\text{ben}}(D=2.14\sim 4.20\text{Å})$, and the distance from the center of the C-C bond to the center of the cross-section is represented by the $r_{\text{c-c}}(D=2.14\sim 4.20\text{Å})$.

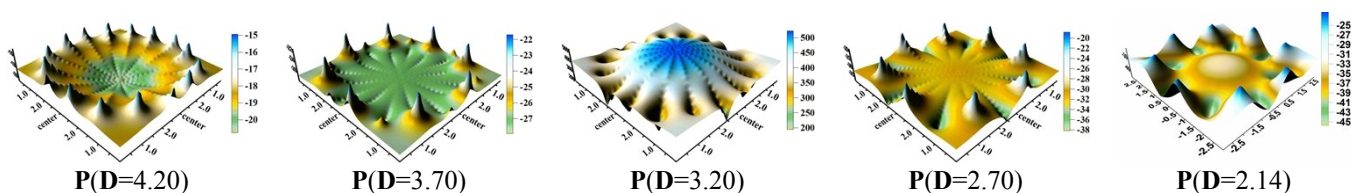


Figure S20. The NICS scan in the cross-section $P(D=2.14\sim 4.20\text{Å})$.

Figure S20 lists the typical schematic diagram of the NICS values scan in the center cross-section $\mathbf{P}(\mathbf{D}=2.14\sim 4.20)$ based on the results of the NICS values scan in the $\mathbf{r}_{\text{ben}}(\mathbf{D}=2.14\sim 4.20)$ and $\mathbf{r}_{\text{c-c}}(\mathbf{D}=2.14\sim 4.20)$. There is a significant difference of the internal magnetic field with one [16]trannulenes closed to another [16]trannulenes. When $\mathbf{D}>3.2\text{\AA}$, there is a induced aromaticity due to the influence of two [16]trannulenes, and the aromaticity is enhancing with the \mathbf{D} decreasing. However, when $\mathbf{D}\approx 3.2\text{\AA}$, the drastic anti-aromaticity has existed between two [16]trannulenes. With the formation of [6]₈cyclacene, the obvious aromaticity has been appeared gradually.