Electronic Supplementary Material (ESI) for New Journal of Chemistry.

Supplementary Information:

Transformation of CO_2 to polyureas with 3-Amino-1,2,4-triazole potassium as the solid base catalyst

Peixue Wang,[a,b] Yuqing Fei,[a] Youquan Deng,*[a,b]

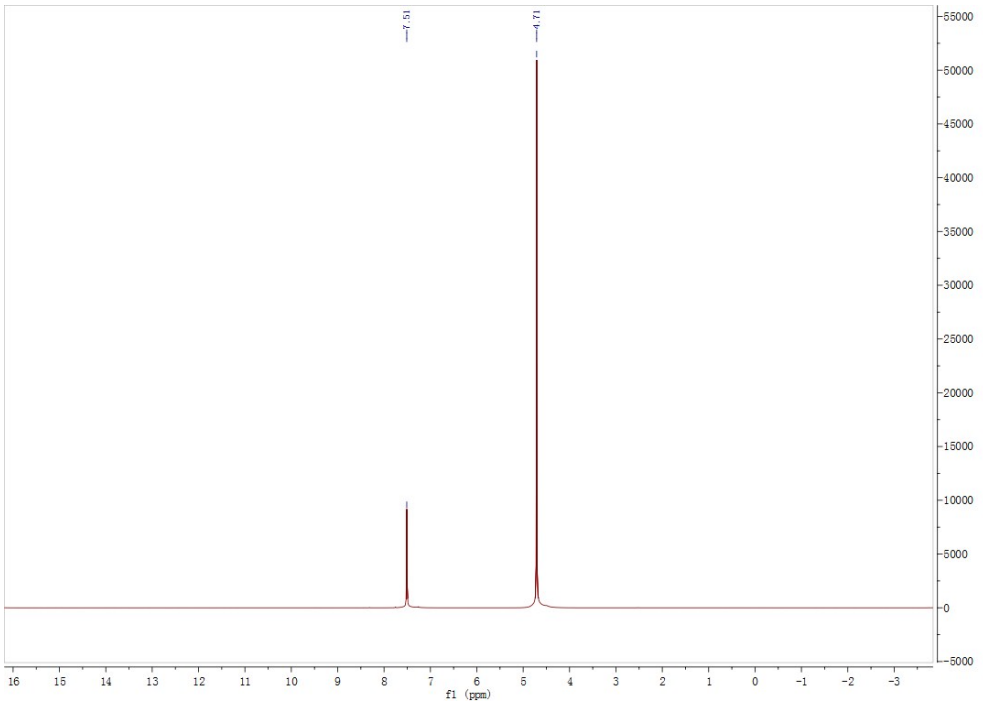


Fig. S1 ^{1}H NMR spectra of 3- amino-1,2,4 Triazole lithium salt.

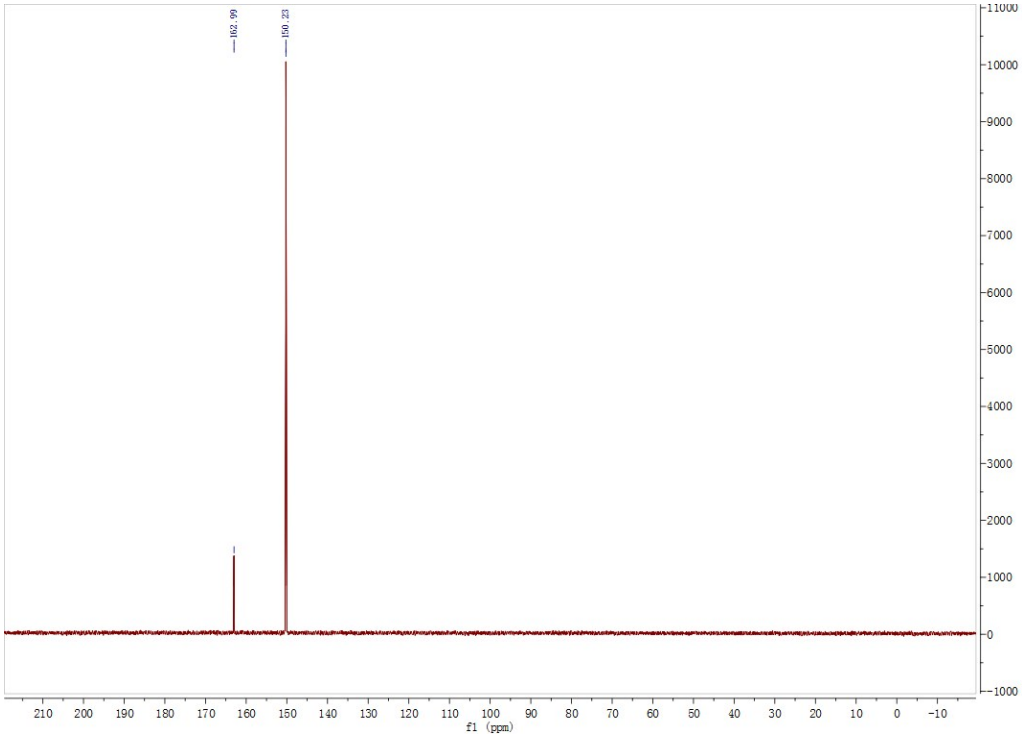


Fig. S2 ^{13}C NMR spectra of 3- amino-1,2,4 Triazole lithium salt.

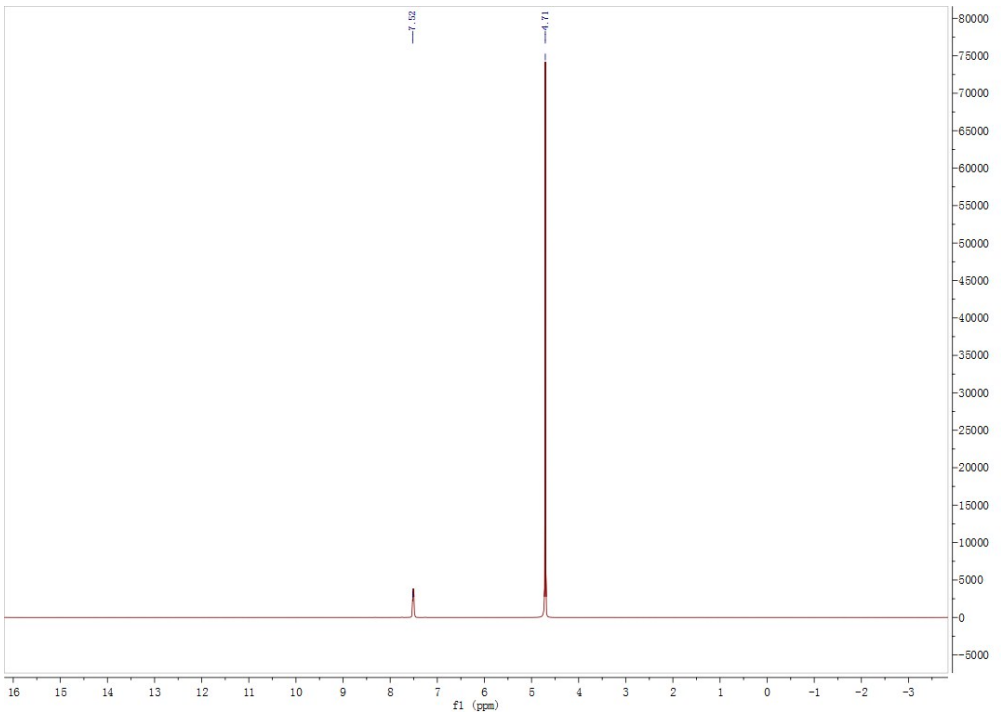


Fig. S3 ^{1}H NMR spectra of 3- amino-1,2,4 Triazole sodium salt.

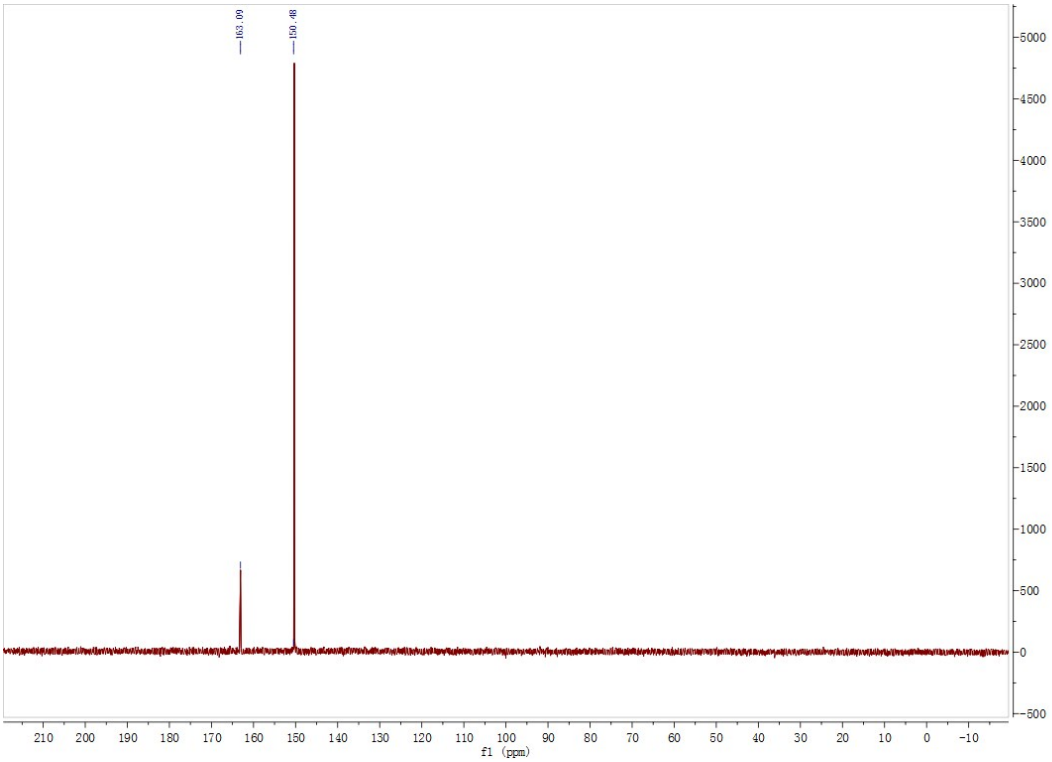


Fig. S4 ^{13}C NMR spectra of 3- amino-1,2,4 Triazole sodium salt.

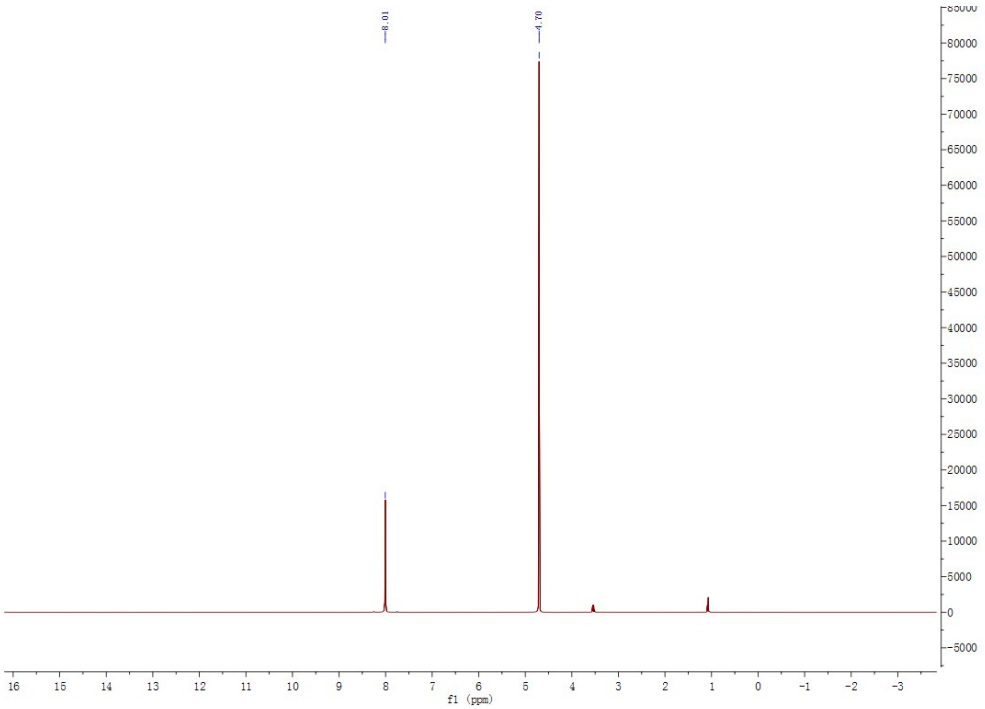


Fig. S5 ^{1}H NMR spectra of 3- amino-1,2,4-Triazole potassium salt.

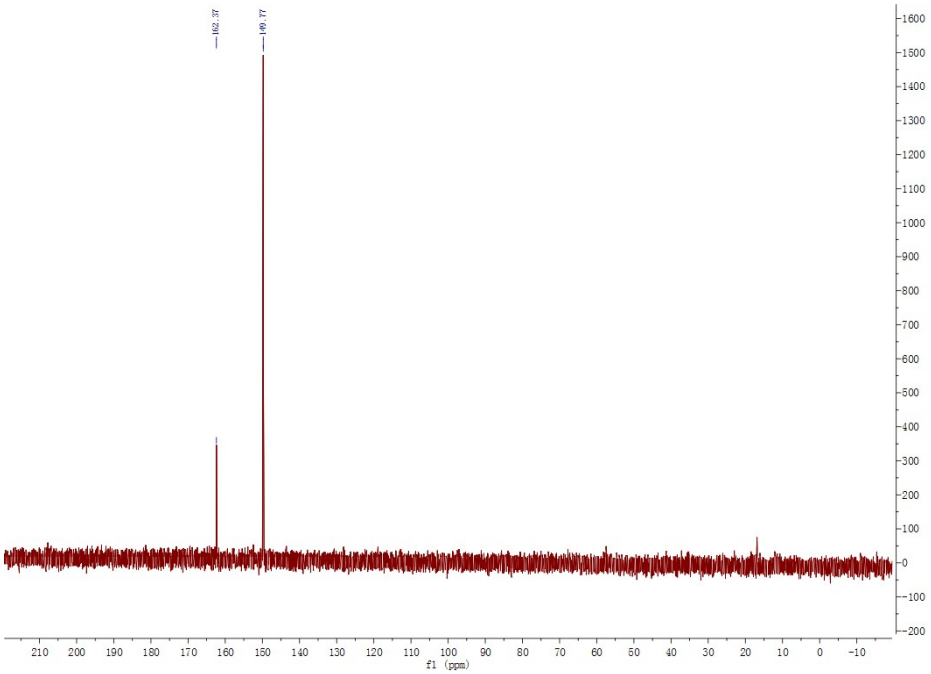


Fig. S6 ^{13}C NMR spectra of 3- amino-1,2,4 Triazole potassium salt.

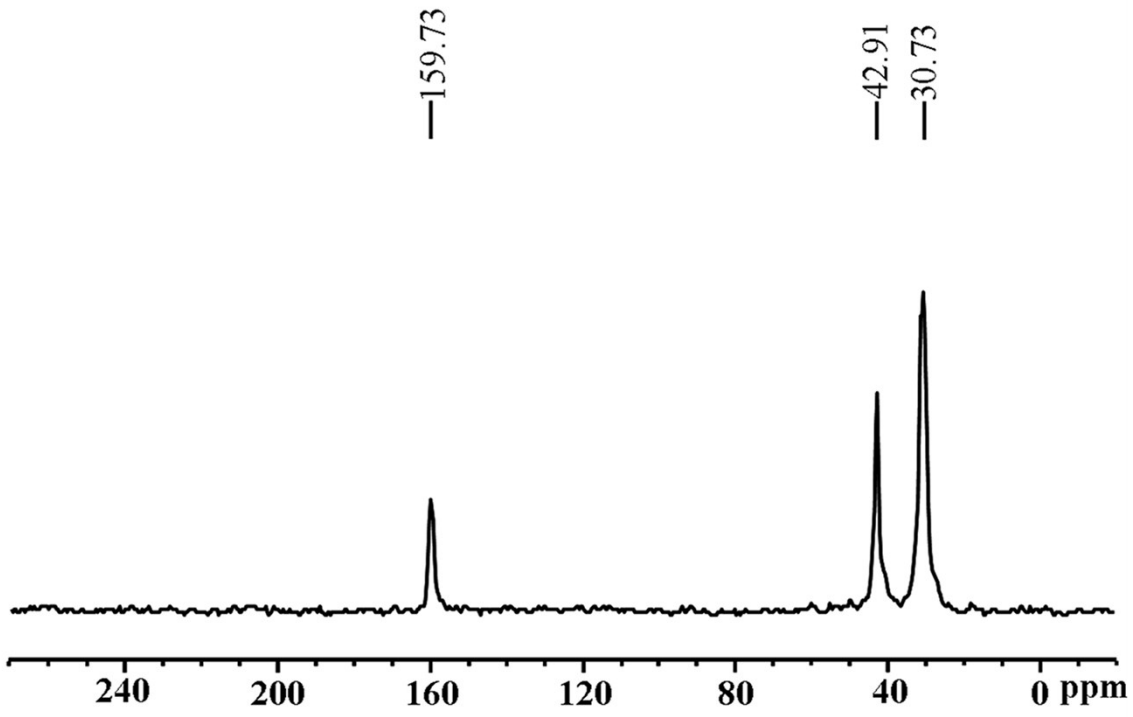
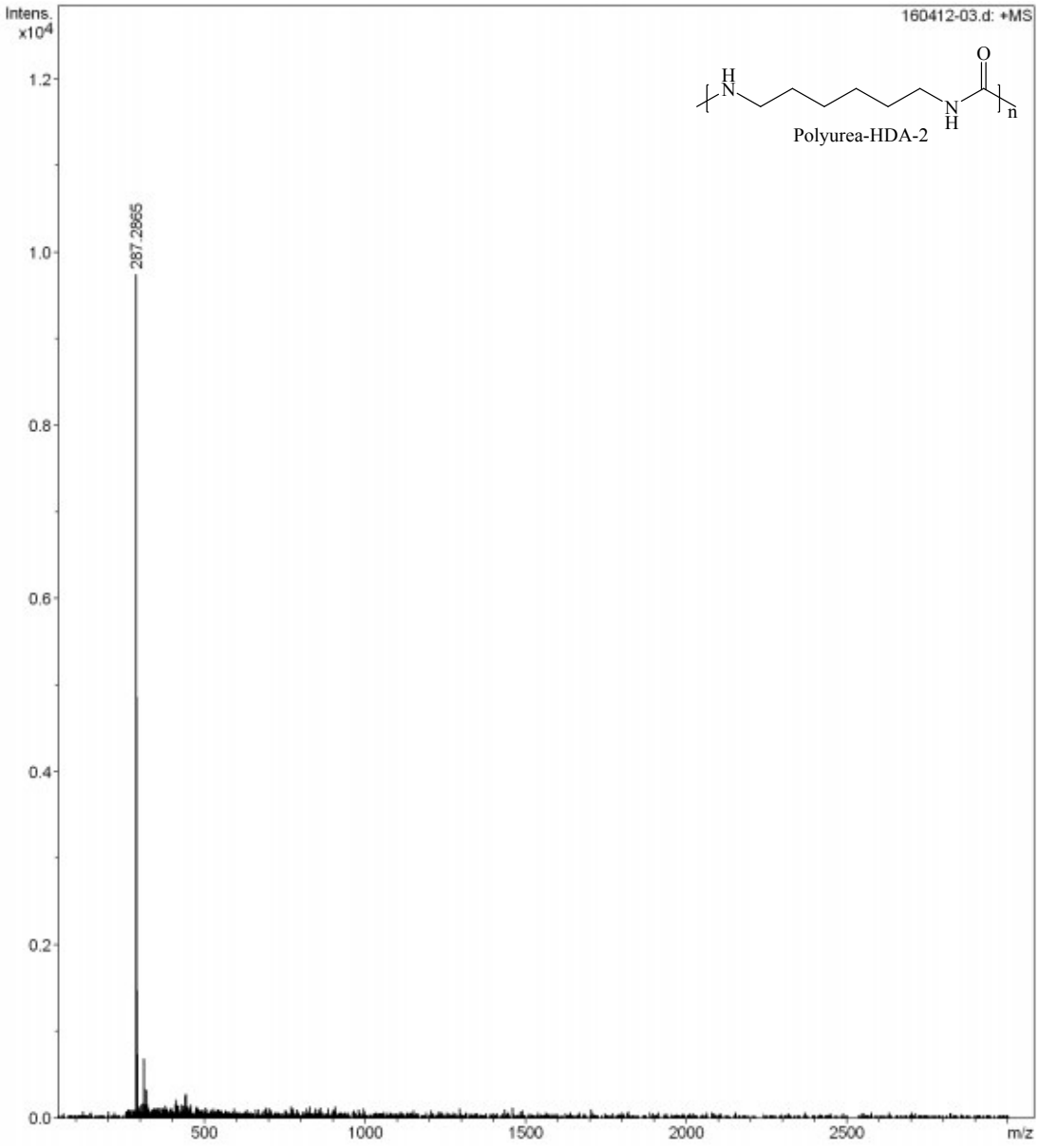
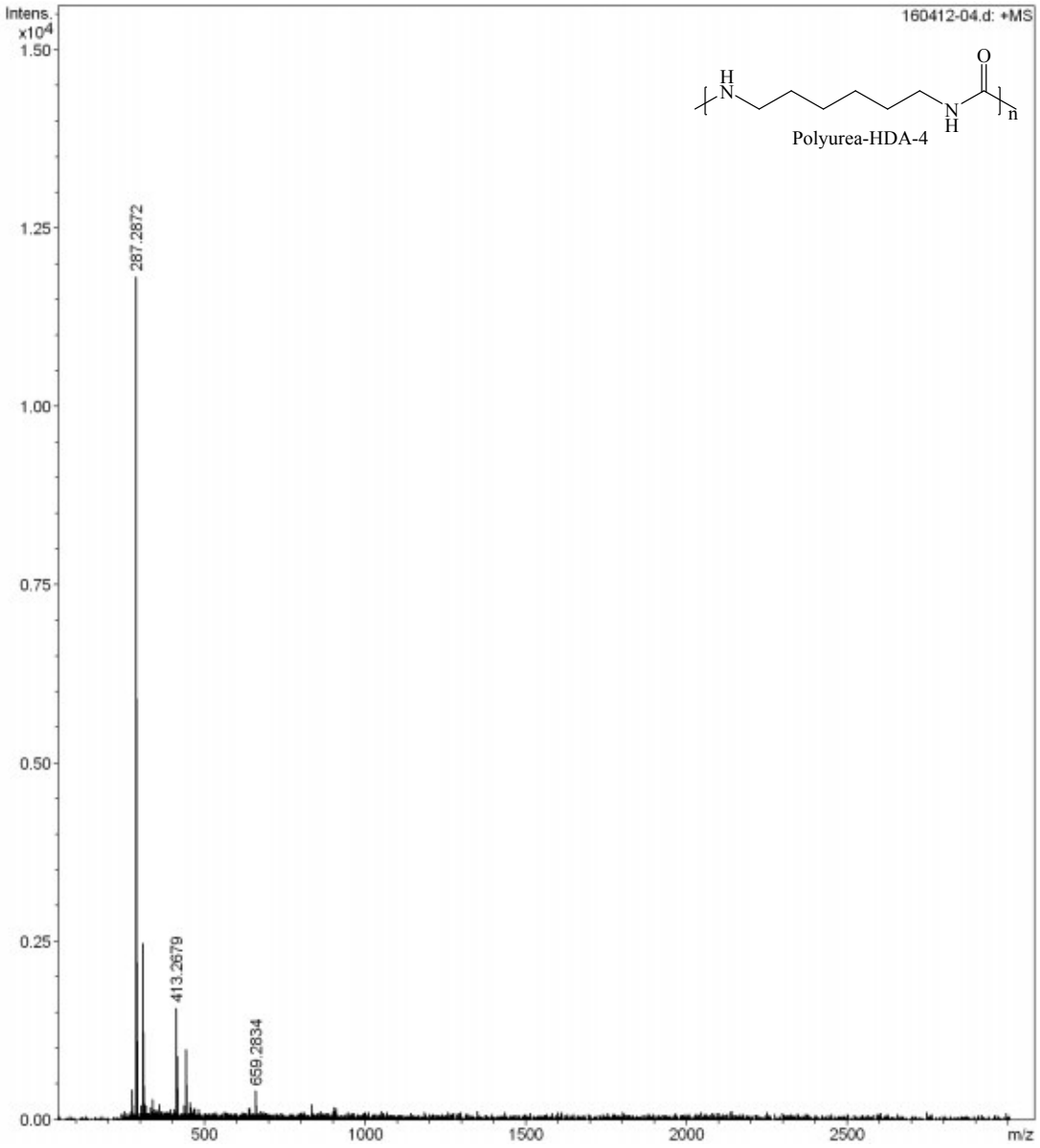


Fig. S7 CP/MAS ^{13}C NMR spectrum of the solid product of the reaction of HDA with CO_2



Polyurea-HDA-2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1.25
1.00
0.75
0.50
0.25
0.00
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659.2834
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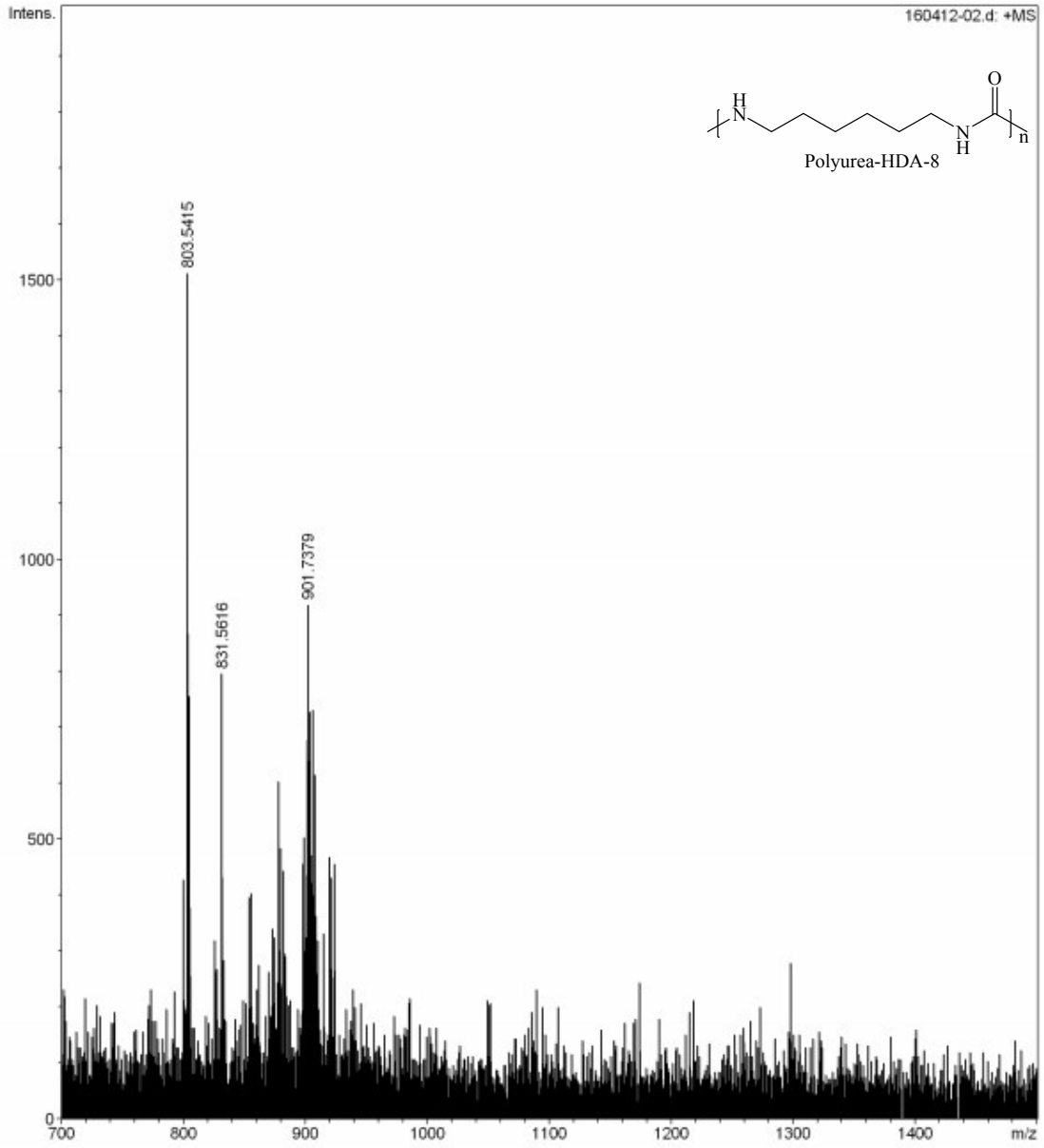
1500
2000
2500
m/z
160412-04.d: +MS
Polyurea-HDA-4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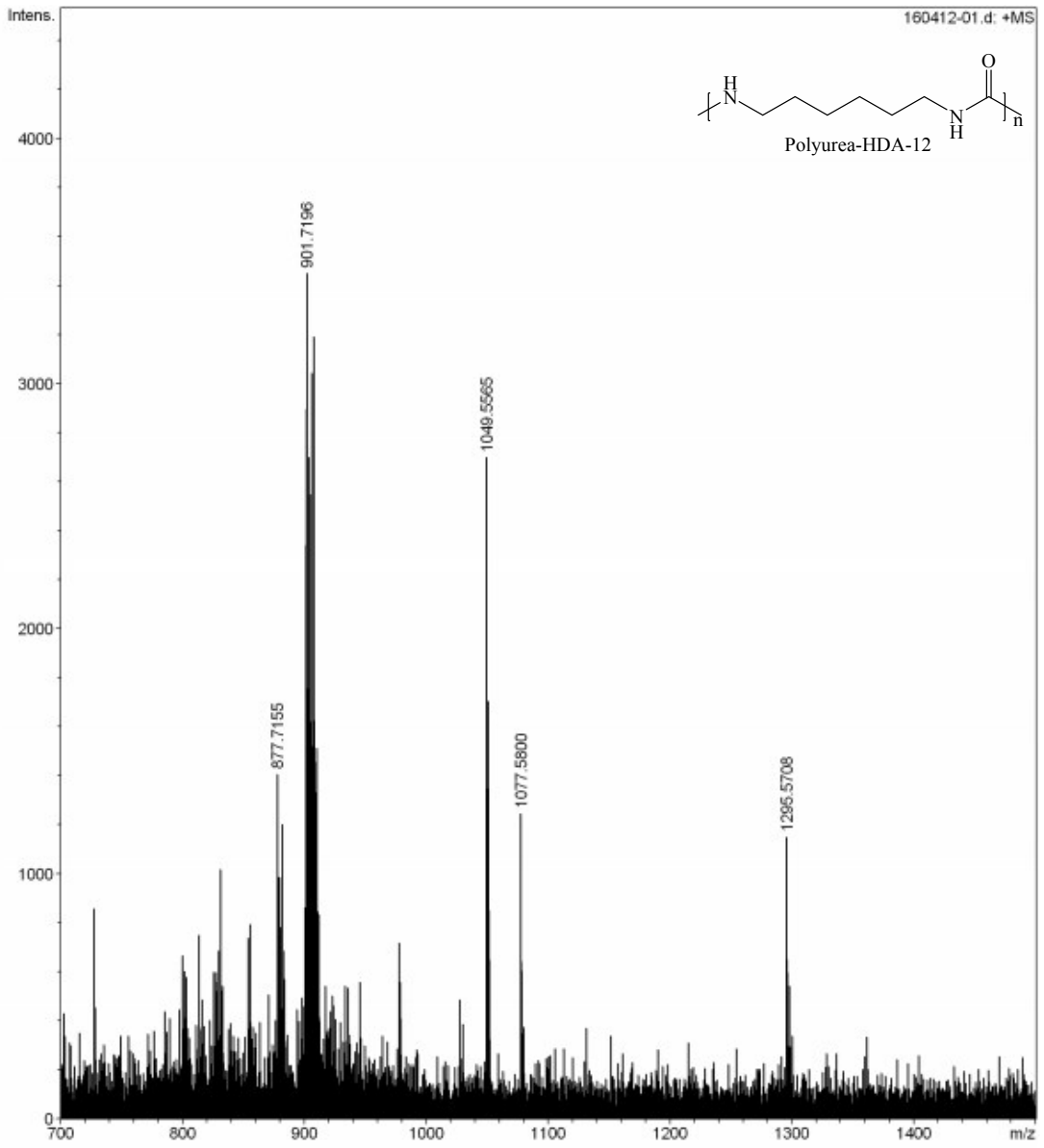
1300
1400
m/z
Polyurea-HDA-8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m/z
Polyurea-HDA-12

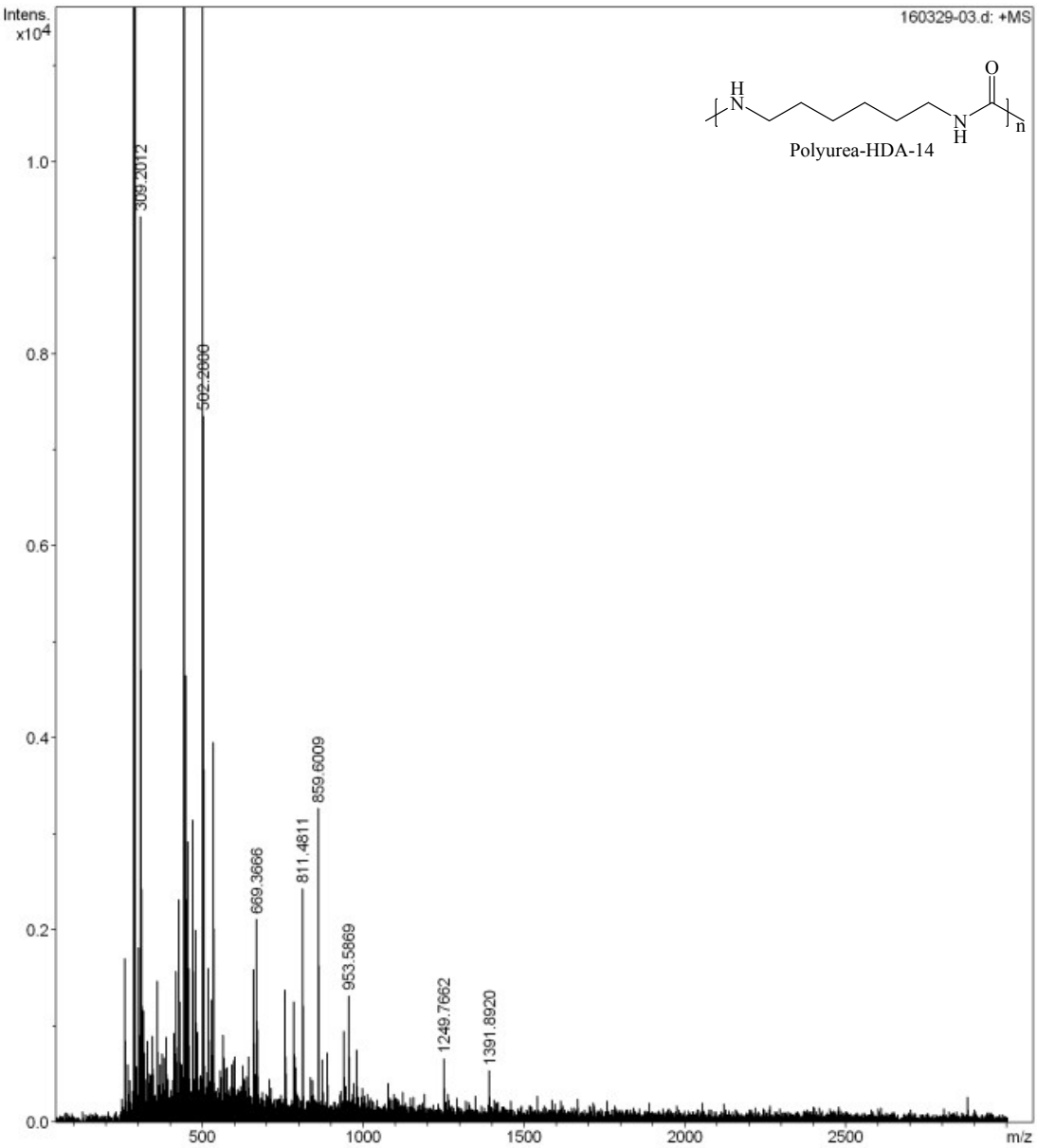

Figure S8. The HRMS spectra of polyurea-HDA-X (X = 2,4,8,12,16, which represents the reaction time of polyurea-HDA preparation) collected by a Bruker micrOTOF Q II

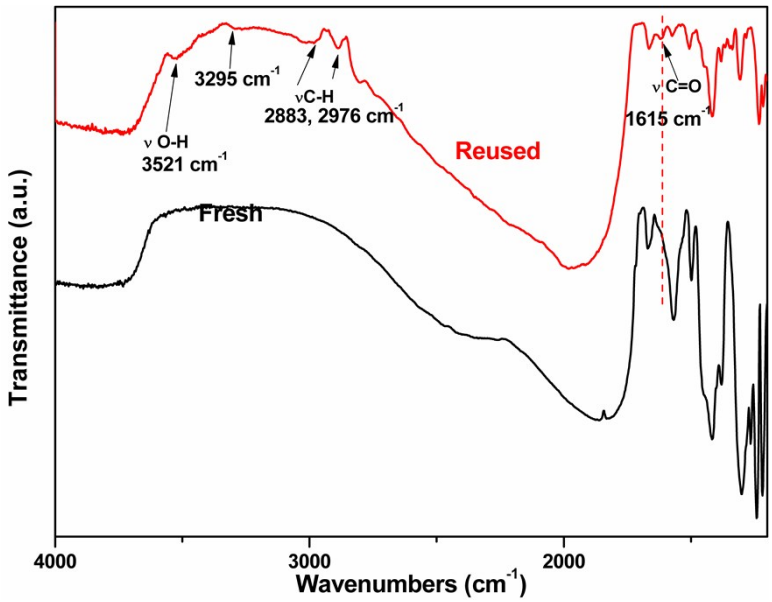

Fig. S9 IR of the KATriz salts before and after use.