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Ruthenium-arene complexes with NSAIDs: Synthesis, characterization and bioactivity

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# **Supplementary material**

# S1. NMR spectra of synthesized complexes



Figure S1. Parallel <sup>1</sup>H NMR spectra of ligand1 and complex 1



Figure S2. Parallel <sup>13</sup>C NMR spectra of ligand 1 and complex 1



Figure S4. Parallel <sup>13</sup>C NMR spectra of ligand 2 and complex 2



Figure S5. Parallel <sup>1</sup>H NMR spectra of ligand1 and complex 3



Figure S6. Parallel <sup>13</sup>C NMR spectra of ligand 1 and complex 3



Figure S7. Parallel <sup>1</sup>H NMR spectra of ligand 2 and complex 4



Figure S8. Parallel <sup>13</sup>C NMR spectra of ligand 2 and complex 4

# S2. MTT assay

Compound	K562	A549	MDA-MB-231	MRC-5	*SI <sub>K562</sub>	*SI A549	*SI <sub>MDA-MB-231</sub>
1	11.9±4.4	45.5±2.7	22±3.6	39.6±3.7	3.33	0.87	1.80
2	96.4±2	145.1±6.4	153±1.2	222.6±23.9	2.31	1.53	1.45
3	13.2±6.2	31.7±1.15	26±1.7	42±1.3	3.18	1.32	1.62
4	133±7	142.4±9.3	$121.4{\pm}1.8$	275.7±14.5	2.07	1.94	2.27
Hindo	155.9±11.4	161.5±13.9	244.7±17.8	230.5±17.8	1.48	1.43	0.94
Hmef	143.9±4.1	217.3±46.7	237.9±18.8	>300	>2.08	>1.38	>1.26
CDDP	10.3±1.2	13.6±1.8	15.9±2.1	9.3±0.9	0.90	0.68	0.58

Table 1. IC<sub>50</sub> [µM] values obtained after 72 h of continuous drug action.

\* IC<sub>50</sub> [ $\mu$ M] values are presented as the mean  $\pm$  SEM of three independent experiments. > 300 denotes that IC<sub>50</sub> was not obtained in the range of concentrations tested up to 300  $\mu$ M.

\*SI-selectivity index for tested complexes, ligands and cisplatin, in tumor cell lines (K562, A549 and MDA-MB-231), related to non-tumor MRC-5 cell line:  $SI_{K562}$  (IC<sub>50</sub> MRC-5/IC<sub>50</sub> K562),  $SI_{A549}$  (IC<sub>50</sub> MRC-5/IC<sub>50</sub> A549),  $SI_{MDA-MB-231}$ (IC<sub>50</sub> MRC-5/IC<sub>50</sub> MDA-MB-231). SI values for tested complexes and ligands were obviously higher than for cisplatin, particularly in MDA-MB-231 and K562.

### **S3.** Interaction with biomolecules

#### S3-1. Interaction with serum albumins

The extent of the inner-filter effect can be roughly estimated with the following formula:

$$I_{corr} = I_{meas} \times 10^{\frac{\epsilon(\lambda_{exc})cd}{2}} \times 10^{\frac{\epsilon(\lambda_{em})cd}{2}}$$
(eq. S1)

where  $I_{corr}$  = corrected intensity,  $I_{meas}$  = the measured intensity, c = the concentration of the quencher, d = the cuvette (1 cm),  $\epsilon(\lambda_{exc})$  and  $\epsilon(\lambda_{em})$  = the  $\epsilon$  of the quencher at the excitation and the emission wavelength, respectively, as calculated from the UV-vis spectra of the complexes.<sup>1</sup>

The Stern-Volmer and Scatchard graphs are used in order to study the interaction of a quencher with serum albumins. According to Stern-Volmer quenching equation: <sup>2</sup>

$$\frac{Io}{I} = 1 + k_{q} \tau_{0}[Q] = 1 + K_{SV}[Q]$$
(eq. S2)

where Io = the initial tryptophan fluorescence intensity of SA, I = the tryptophan fluorescence intensity of SA after the addition of the quencher (i.e. complexes **1-4**),  $k_q$  = the quenching constant,  $K_{SV}$  = the Stern-Volmer constant,  $\tau_o$  = the average lifetime of SA without the quencher, [Q] = the concentration of the quencher)  $K_{SV}$  (in M<sup>-1</sup>) can be obtained by the slope of the diagram Io/I versus [Q], and subsequently the quenching constant ( $k_q$ , in M<sup>-1</sup>s<sup>-1</sup>) is calculated from eq. S3, with  $\tau_o$ = 10<sup>-8</sup> s as fluorescence lifetime of tryptophan in SA,

$$K_{SV} = k_q \tau_o$$
 (eq. S3)

From the Scatchard equation:<sup>3</sup>

$$\frac{\Delta I}{[Q]} = nK - K\frac{\Delta I}{Io}$$
 (eq. S4)

where n is the number of binding sites per albumin and K is the SA-binding constant, K (in M<sup>-1</sup>) is calculated from the slope in plots ( $\Delta I/Io$ )/[Q] versus  $\Delta I/Io$  and n is given by the ratio of y intercept to the slope.<sup>3</sup>

### S3-2. Interaction with CT DNA

The DNA-binding constant (K<sub>b</sub>, in M<sup>-1</sup>) can be obtained by monitoring the changes in the absorbance at the corresponding  $\lambda_{max}$  with increasing concentrations of CT DNA and it is given by the ratio of slope to the y intercept in plots [DNA]/( $\epsilon_A$ - $\epsilon_f$ ) versus [DNA], according to the Wolfe-Shimer equation:<sup>4</sup>

$$\frac{[\text{DNA}]}{(\varepsilon_{\text{A}} - \varepsilon_{\text{f}})} = \frac{[\text{DNA}]}{(\varepsilon_{\text{b}} - \varepsilon_{\text{f}})} + \frac{1}{K_{\text{b}}(\varepsilon_{\text{b}} - \varepsilon_{\text{f}})}$$
(eq. S5)

where [DNA] is the concentration of DNA in base pairs,  $\varepsilon_A = A_{obsd}/[compound]$ ,  $\varepsilon_f$  = the extinction coefficient for the free compound and  $\varepsilon_b$  = the extinction coefficient for the compound in the fully bound form.

### S3-3. Competitive studies with EB

The Stern-Volmer constant (K<sub>SV</sub>, in M<sup>-1</sup>) is used to evaluate the quenching efficiency for each compound according to the Stern-Volmer equation (eq. S2),<sup>2</sup> where Io and I are the emission intensities of the EB-DNA solution in the absence and the presence of the quencher, respectively, [Q] is the concentration of the quencher (i.e. complexes **1-4**),  $\tau_0$  = the average lifetime of the emitting system without the quencher and k<sub>q</sub> = the quenching constant. K<sub>SV</sub> may be obtained from the Stern-Volmer plots by the slope of the diagram Io/I versus [Q]. Taking  $\tau_0$  = 23 ns as the fluorescence lifetime of the EB-DNA system,<sup>5</sup> the quenching constants (k<sub>q</sub>, in M<sup>-1</sup>s<sup>-1</sup>) of the compounds can be determined according to eq. (S3).

#### References

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Table S2. 7	The BSA a	and HSA	binding	constants and	parameters	(K <sub>sv</sub> , 1	Ka, K	, n) fo	or complexes	1-4.
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Compound	Ksv (M <sup>-1</sup> )	$k_{q} (M^{-1}s^{-1})$	K (M <sup>-1</sup> )	n
BSA				
$K[Ru(\eta^6-p-cymene)(indo)Cl_2], 1$	$1.25(\pm 0.05) \times 10^5$	$1.25(\pm 0.05) \times 10^{13}$	$4.49(\pm 0.30) \times 10^5$	0.68
$(NH_4)[Ru(\eta^6-p-cymene)(mef)Cl_2], 2$	$1.70(\pm 0.08) \times 10^{5}$	$1.70(\pm 0.08) \times 10^{13}$	$3.63(\pm 0.15) \times 10^5$	0.86
K[Ru( $\eta^6$ - <i>p</i> -toluene)(indo)Cl <sub>2</sub> ], <b>3</b>	$4.85(\pm 0.11) \times 10^4$	$4.85(\pm 0.11) \times 10^{12}$	$5.30(\pm 0.18) \times 10^4$	0.96
$(NH_4)[Ru(\eta^6-p-toluene)(mef)Cl_2], 4$	$1.30(\pm 0.04) \times 10^5$	1.30(±0.04)×10 <sup>13</sup>	$2.63(\pm 0.10) \times 10^5$	0.84
HSA				
K[Ru( $\eta^6$ - <i>p</i> -cymene)(indo)Cl <sub>2</sub> ], <b>1</b>	$6.10(\pm 0.29) \times 10^4$	$6.10(\pm 0.29) \times 10^{12}$	$2.15(\pm 0.08) \times 10^5$	0.57
$(NH_4)[Ru(\eta^6-p-cymene)(mef)Cl_2], 2$	$5.46(\pm 0.19) \times 10^4$	$5.46(\pm 0.19) \times 10^{12}$	9.79(±0.34)×10 <sup>4</sup>	0.78
K[Ru( $\eta^6$ - <i>p</i> -toluene)(indo)Cl <sub>2</sub> ], <b>3</b>	$2.04(\pm 0.13) \times 10^4$	$2.04(\pm 0.13) \times 10^{12}$	$9.44(\pm 0.40) \times 10^4$	0.31
$(NH_4)[Ru(\eta^6-p-toluene)(mef)Cl_2], 4$	4.32(±0.29)×104	4.32(±0.29)×1012	$4.24(\pm 0.12) \times 10^{5}$	0.37



Figure S9. Stern-Volmer quenching plot of HSA for complexes (A)-(D) 1-4, respectively.



Figure S10. Stern-Volmer quenching plot of BSA for complexes (A)-(D) 1-4, respectively.



Figure S11. Scatchard plot of HSA for complexes (A)-(D) 1-4, respectively.



Figure S12. Scatchard plot of BSA for complexes (A)-(D) 1-4, respectively.



Figure S13. Plot of  $[DNA]/(\epsilon_A - \epsilon_f)$  vs [DNA] for complexes (A)-(D) 1-4, respectively.



**Figure S14**. Stern-Volmer quenching plot of EB-DNA fluorescence for complexes (A)-(D) **1-4**, respectively.