

Supporting Information

Magneto-structural correlation of cyano-substituted
3-*tert*-butyl-1-phenyl-1,2,4-benzotriazin-4-yl: spin
transition behaviour observed in a 6-cyano derivative

Yusuke Takahashi, Naoya Tsuchiya, Youhei Miura and Naoki Yoshioka*

*Department of Applied Chemistry, Faculty of Science and Technology,
Keio University, Yokohama 223-8522, Japan.
E-mail: yoshioka@applc.keio.ac.jp*

Table of Contents

Magnetic Properties	2
Crystallographic Data	3
Computational Data	6
EPR Data	8
UV-Vis Spectra	9
Cyclic Voltammograms	10
IR Spectra	11

Magnetic Properties

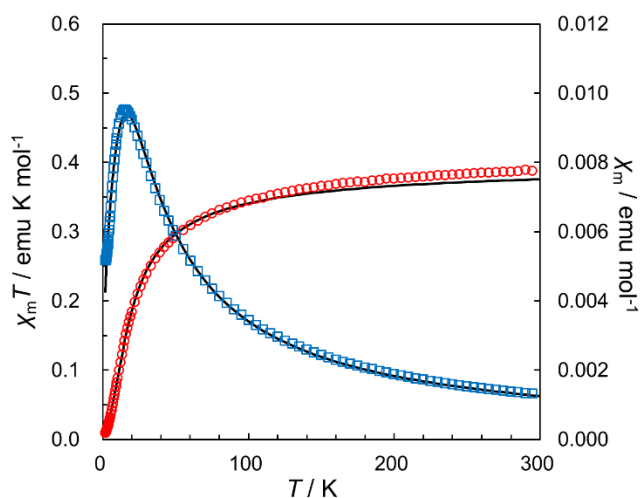


Fig. S1 Thermal dependence of $\chi_m T$ and χ_m for **7CN**. The solid lines represent the best-fit result of the 1D antiferromagnetic chain model (correlation factor $R = \Sigma[\chi_{m \text{ obs}} - \chi_{m \text{ calcd}}]^2 / \Sigma[\chi_{m \text{ obs}}]^2 = 5.25 \times 10^{-4}$).

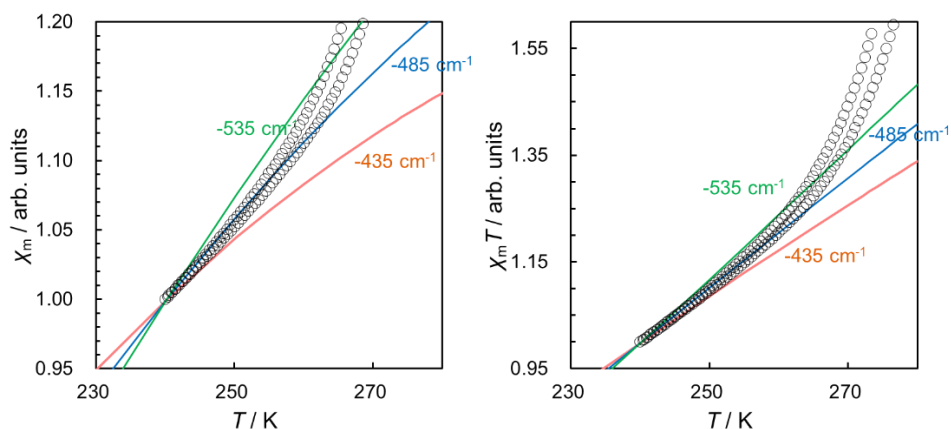


Fig. S2 Temperature dependence of normalized χ_m and $\chi_m T$ of LT phase of Al foil wrapped sample of **6CN** for the rough estimation of the magnitude of the intermolecular magnetic interaction using the Bleaney-Bowers model. The experimental and theoretical values of χ_m and $\chi_m T$ were normalized by the corresponding value of 240 K.

Crystallographic Data

Table S1 Crystallographic data of **7CN** and **6CN**.

	7CN	6CN (HT)	6CN (LT)^a
Chemical formula	C ₁₈ H ₁₇ N ₄	C ₁₈ H ₁₇ N ₄	C ₁₈ H ₁₇ N ₄
Formula weight	289.36	289.36	289.36
Temperature / K	296	300	263
Crystal size / mm ³	0.74 × 0.17 × 0.08	1.24 × 0.54 × 0.06	1.24 × 0.54 × 0.06
Space group	<i>I</i> 2/ <i>a</i> (No. 15)	<i>I</i> 2/ <i>a</i> (No. 15)	<i>P</i> -1 (No. 2)
<i>a</i> / Å	18.709(3)	26.317(4)	6.8527(6)
<i>b</i> / Å	7.8316(12)	6.7524(7)	9.1813(9)
<i>c</i> / Å	21.501(6)	18.003(2)	13.1091(12)
α / °	90	90	86.834(3)
β / °	90.945(4)	90.578(3)	80.616(3)
γ / °	90	90	74.983(3)
<i>V</i> / Å ³	3149.9(11)	3199.0(7)	785.88(13)
<i>Z</i>	8	8	2
<i>d</i> _{calc} / g cm ⁻³	1.220	1.202	1.223
μ / mm ⁻¹	0.075	0.074	0.075
F(000)	1224	1224	306
θ (min, max) / °	2.18, 23.02	2.73, 24.44	2.77, 25.10
	-19 ≤ <i>h</i> ≤ 20	-30 ≤ <i>h</i> ≤ 30	-7 ≤ <i>h</i> ≤ 8
Index ranges	-8 ≤ <i>k</i> ≤ 8	-7 ≤ <i>k</i> ≤ 7	-10 ≤ <i>k</i> ≤ 10
	-23 ≤ <i>l</i> ≤ 22	-18 ≤ <i>l</i> ≤ 20	0 ≤ <i>l</i> ≤ 15
Measured reflection	10772	13670	2512
Independent reflection (<i>R</i> _{int})	2196 (0.0486)	2605 (0.0641)	2512 (-)
Observed reflection (<i>I</i> ≥ 2σ(<i>I</i>))	1422	1929	1707
Goodness of fit on <i>F</i> ²	1.380	1.396	1.103
<i>R</i> , <i>R</i> _w (<i>I</i> ≥ 2σ(<i>I</i>))	0.0702, 0.1891	0.0712, 0.2160	0.0647, 0.1763
<i>R</i> , <i>R</i> _w (all data)	0.1153, 0.2190	0.0901, 0.2370	0.0975, 0.2128
Resd density (min, max) / e Å ⁻³	-0.300, 0.438	-0.190, 0.289	-0.276, 0.220

^a The data of LT phase for **6CN** was obtained as twinning crystal and it was analyzed using *TWINABS* program.

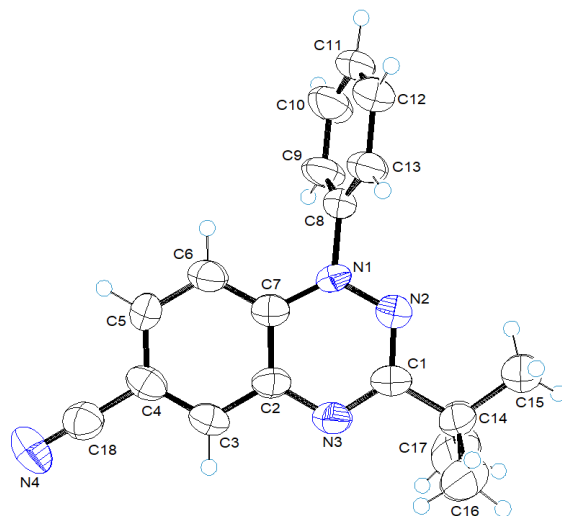


Fig. S3 ORTEP drawing of LT phase of **6CN** (50% thermal ellipsoids).

Table S2 Selected bond lengths and dihedral angle of benzotriazinyl radicals.

Compound	$d_{N1-N2} / \text{\AA}$	$d_{N2-C1} / \text{\AA}$	$d_{C1-N3} / \text{\AA}$	dihedral angle / $^{\circ}$
7CN	1.375(3)	1.324(4)	1.335(4)	45.236(80)
6CN	HT	1.360(3)	1.332(3)	68.408(97)
	LT	1.364(3)	1.325(4)	69.858(130)

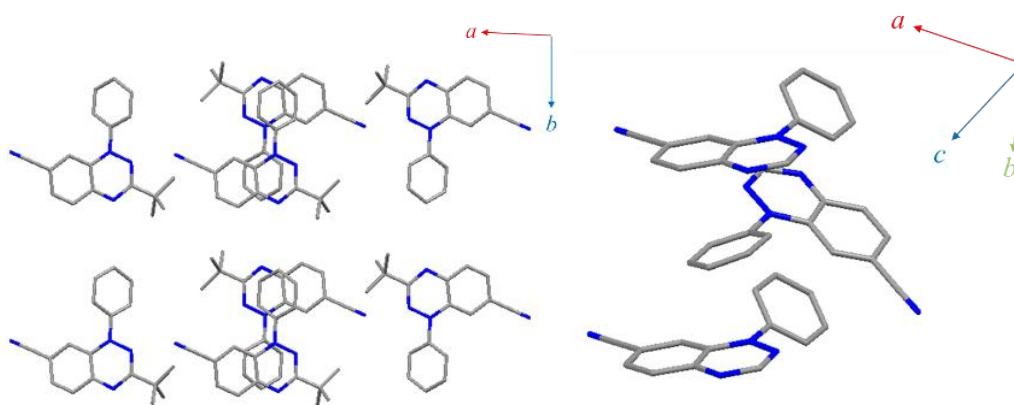


Fig. S4 Molecular arrangement of 7CN. The selected *tert*-butyl groups were omitted for clarity.

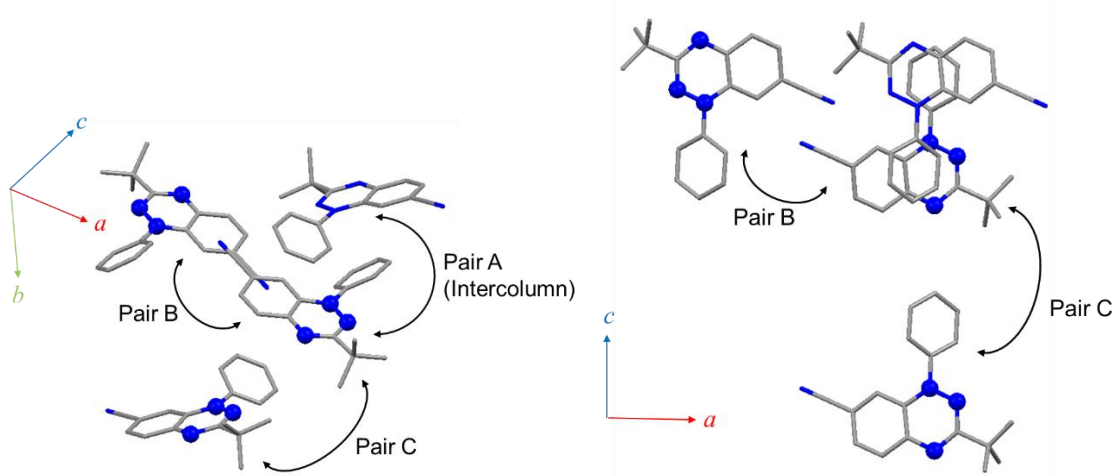
Table S3 The longitudinal and latitudinal slippages of 6CN.

Pair	$T / \text{K}(\text{phase})$	$\phi_1 / ^\circ$	$\phi_2 / ^\circ$
I-II	300 (HT)	50.8	66.0
	263 (LT)	57.2	65.2
II-I'	300 (HT)	47.2	64.5
	263 (LT)	43.2	67.1

Computational Data

Table S4 The calculation result of the intermolecular interaction for **7CN** at UB3LYP and UBLYP level, and 6-31G(d) basis set. BS and T represent broken symmetry singlet and triplet state.

Pair	Function	Spin state	Energy / a.u.	$\langle S^2 \rangle$	J / cm^{-1}
Pair A	UB3LYP	BS	-1829.6052263	1.027120	-3.4
		T	-1829.6052110	2.027955	
	UBLYP	BS	-1828.7856173	1.003589	-8.7
		T	-1828.7855779	2.008456	
Pair B	UB3LYP	BS	-1829.6155189	1.027472	+0.2
		T	-1829.6155196	2.027491	
	UBLYP	BS	-1828.7979382	1.008204	-0.04
		T	-1828.7979380	2.008332	
Pair C	UB3LYP	BS	-1829.6079463	1.027894	-0.04
		T	-1829.6079461	2.027898	
	UBLYP	BS	-1828.7908988	1.008365	-0.11
		T	-1828.7908983	2.008430	



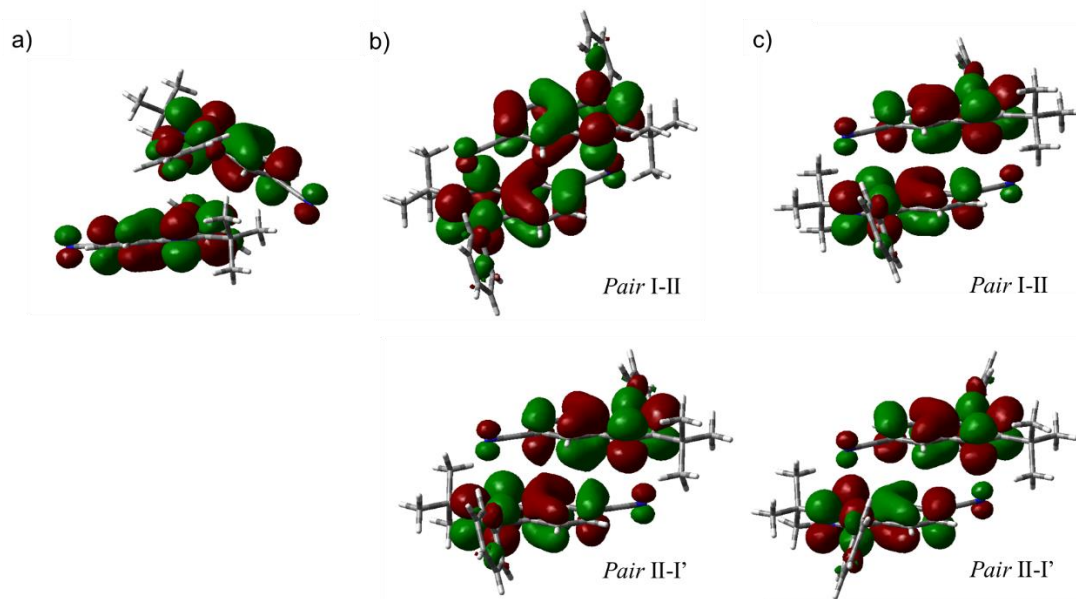


Fig. S5 Overlap of SOMOs for (a) **7CN**, (b) **6CN(HT)** and (c) **6CN(LT)**.

Table S5 The calculation results of **6CN** at UB3LYP/6-31G(d) level. BS and T represent broken symmetry singlet and triplet state.

Pair (Phase)	Spin state	Energy / a.u.	$\langle S^2 \rangle$	$2J / \text{cm}^{-1}$
I-II(HT)	BS	-1829.5998682	1.025442	+4.5
	T	-1829.5998784	2.027256	
I-II(LT)	BS	-1829.6097942	0.980403	-587.5
	T	-1829.6085162	2.027684	
II-I'(HT)	BS	-1829.5974589	1.026153	+10.9
	T	-1829.5974837	2.027356	
II-I'(LT)	BS	-1829.6107765	1.025442	-147.5
	T	-1829.6104445	2.027256	

EPR Data

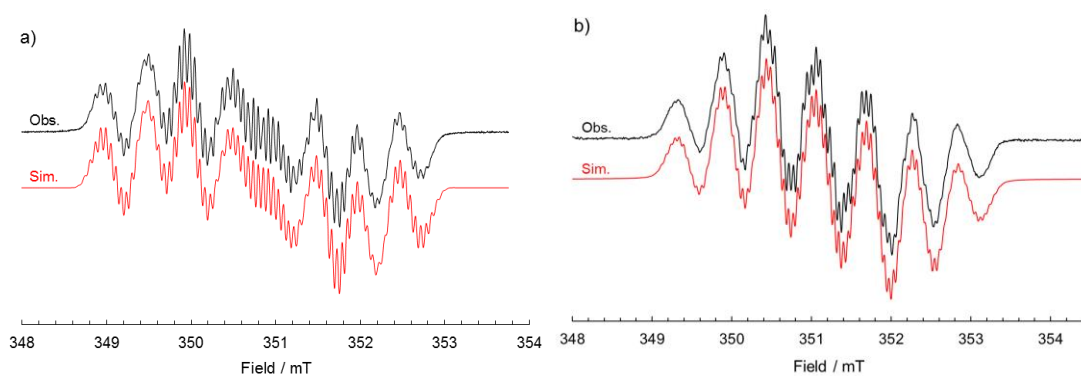
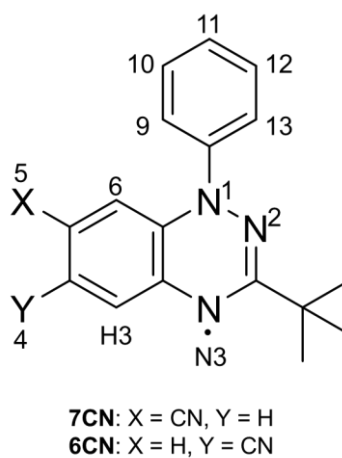


Fig. S6 EPR spectra of (a) **7CN** and (b) **6CN** in toluene at room temperature; black and red lines denote the observed and simulated spectra, respectively.

Table S6 The hfccs and *g*-factor of **7CN** and **6CN**.

nuclei	hfcc / mT	
	7CN	6CN
N1	0.752	0.687
N2	0.462	0.516
N3	0.495	0.517
CN	0.113	0.800
H3	0.125	0.120
H4	0.117	-
H5	-	0.169
H6	0.113	0.101
H9,13	0.061	0.066
H10, 12	0.060	0.053
H11	0.061	0.048
<i>g</i> -factor	2.0039	2.0040



UV-Vis Spectra

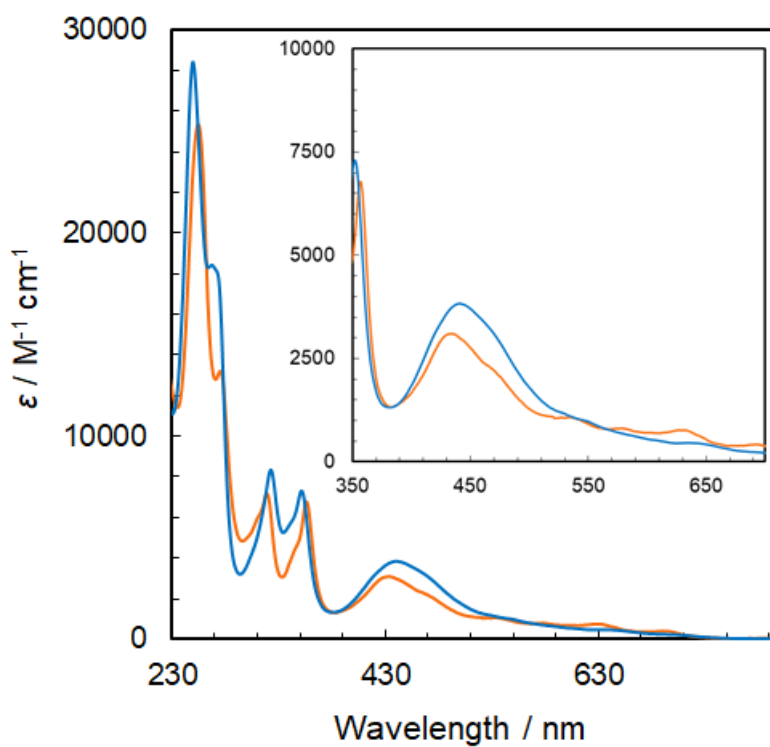


Fig. S7 UV-Vis spectra of **7CN**(blue) and **6CN**(orange) in CH₂Cl₂.

Table S7 The wavelength of maximum absorption (λ_{\max}) of **7CN** and **6CN**.

Compound	λ_{\max} (molar absorptance coefficient, $\epsilon/10^3 \text{ M}^{-1} \text{ cm}^{-1}$)
7CN	250(28.4), 263(18.4), 323(8.33), 352(7.30), 441(3.82)
6CN	255(25.4), 276(13.2), 319(7.17), 356(6.78), 435(3.10), 632(0.764)

Cyclic Voltammograms

The cyclic voltammograms of **7CN** and **6CN** are shown in Figure S8. Two reversible waves corresponding to the reduction and oxidation processes were observed.

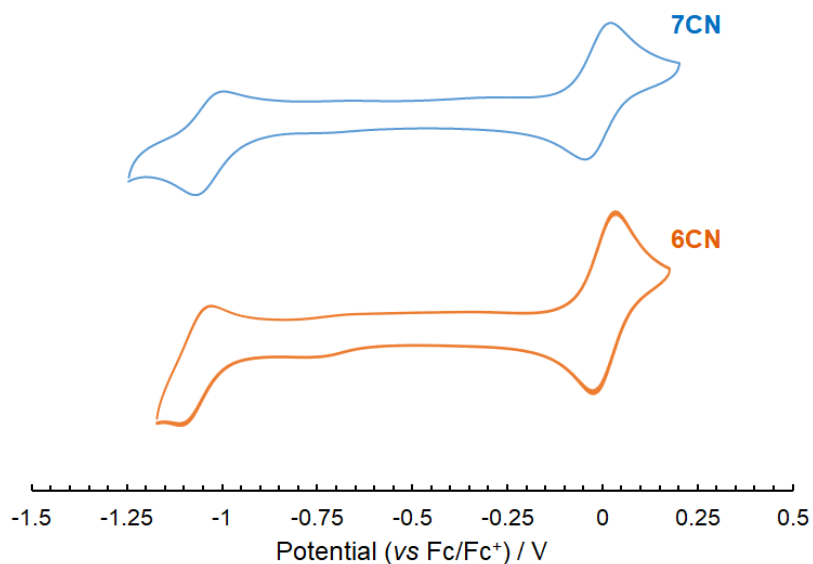


Figure S8. Cyclic voltammograms of **7CN** and **6CN** in CH_3CN containing 0.1 M tetrabutylammonium hexafluorophosphate as the supporting electrolyte.

Table S8 The redox potential of **7CN** and **6CN**.

Compound	Potential (vs. Fc/Fc^+)	
	E_1	E_2
7CN	-1.03	-0.01
6CN	-1.07	0.174

IR Spectra

