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## Para aminosalicylic acid in the treatment of manganese toxicity. Complexation of Mn<sup>2+</sup> with 4amino-2-hydroxybenzoic acid and its N-acetylated metabolite.

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## **Electronic Supplementary Material**



**Figure S1.** UV spectra of SAL (A-C), PAS (D-F), Ac-PAS (G-I) and 4-aminobenzoic acid (J). The concentration of the ligands was 0.5 mM, at 0.1 M NaCl ionic strength, 25 °C and 0.2 cm optical path length.



**Figure S2**. Minimum energy geometries of the MnL and  $[MnL_2]^{2-}$  species with no extra explicit water molecules added. Optimizations performed at the B3LYP/6-31G(d,p)/SDD level of theory in the CPCM solvation model (solvent: water). A)  $[Mn(H_2O)_4(SAL)]$ ; B)  $[Mn(H_2O)_4(PAS)]$ ; C)  $[Mn(H_2O)_4(AcPAS)]$ ; D)  $[Mn(H_2O)_2(SAL)_2]^{2-}$ ; E)  $[Mn(H_2O)_2(PAS)_2]^{2-}$ ; F)  $[Mn(H_2O)_2(AcPAS)_2]^{2-}$ .

	НОМО	LUMO
SALH <sub>2</sub>		
SALH		
SAL		
PASH <sub>3</sub>		
PASH <sub>2</sub> (b)		
PASH		



**Figure S3.** Plots of the HOMO LUMO orbitals of the various protonated forms of the salicylic acid derivatives.



**Figure S4.** Speciation plots of manganese complexes with SAL (A), PAS (B) and Ac-PAS (C) calculated using the stability constants in Table 2. Ligand concentration was 0.5mM and a 1:5 metal to ligand ratio was used.



**Figure S5.** Speciation plots of copper complexes with SAL (A), PAS (B) and Ac-PAS (C) calculated using the stability constants in Table 6. Ligand concentration was 0.5 mM. A 1:5 metal to ligand ratio and 0.1 M NaCl ionic strength were used. The pH of complex precipitation is marked with the blue line.



**Figure S6.** Speciation plots relative of iron and SAL (A), PAS (B) and Ac-PAS (C) complexes calculated using the stability constants in Table 6. Ligand concentration were 0.5 mM and a 1:5 metal to ligand ratio, 0.1 M NaCl ionic strength were used. The pH of complex precipitation is marked with the blue line.