

HPLC-ICP-AES technique for the screening of $[XW_{11}NbO_{40}]^{n-}$ aqueous solutions

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Supporting information

Fig. S1. The HPLC-UV chromatograms of reaction product for different initial W:Nb ratios.

Fig. S2. The HPLC-ICP-AES chromatograms of reaction product at different initial W:Nb ratios in coordinates “retention time–line intensity”.

Table S1. Atomic ratios W/Nb for $[PW_{11}NbO_{40}]^{4-}$ peak calculated from HPLC-ICP-AES-data.

Fig. S3. ESI-MS spectrum of $[PW_{11}NbO_{40}]^{4-}$ after the reaction of $(TBA)_4H_3[PW_{11}O_{39}]$ and Nb-Ox for 12 hours at 60°C in DMF.

Fig. S4. ^{31}P NMR spectrum of $(TBA)_4[PW_{11}NbO_{40}]$ in CD_3CN .

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Fig. S9. Crystal packing of **2**.

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Fig. S11. Crystal packing of **3**.

Fig. S12. ESI-MS(-) spectrum of **5** in acetonitrile.

Table S2. Peak assignment for ESI-MS(-) spectrum of **5** in acetonitrile.

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Fig. S15. Crystal packing of **4**.

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Fig. S17. The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of $[GeW_{11}NbO_{39}]^{5-}$ solution.

Fig. S18. ESI-MS(-) spectrum of **6** in acetonitrile.

Table S3. Peak assignment for ESI-MS(-) spectrum of **6** in acetonitrile.

Table S4. Experimental details.

Table S5. Selected geometric parameters (Å).

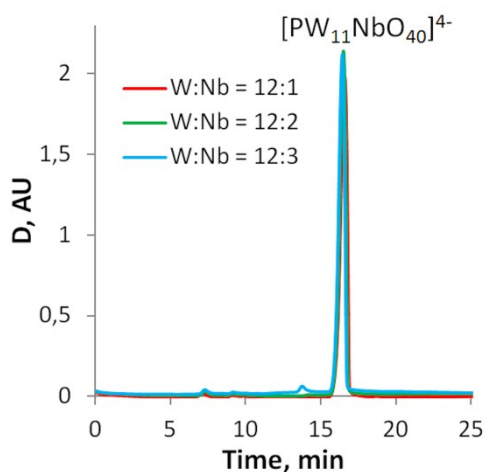


Fig. S1. The HPLC-UV chromatograms of reaction product for different initial W:Nb ratios. The separation conditions: ProntoSIL 120-5-C18AQ, eluents: A – 0.04% TBAH, 1% AcOH (pH 2.7), B – acetonitrile; gradient mode: 0–4.6 min, 0–30% B; 4.6–14.2 min, 30–55% B; 14.2–23.1 min, 55% B; 23.1–29.2 min, 100% B; flow rate – 0.13 mL min⁻¹.

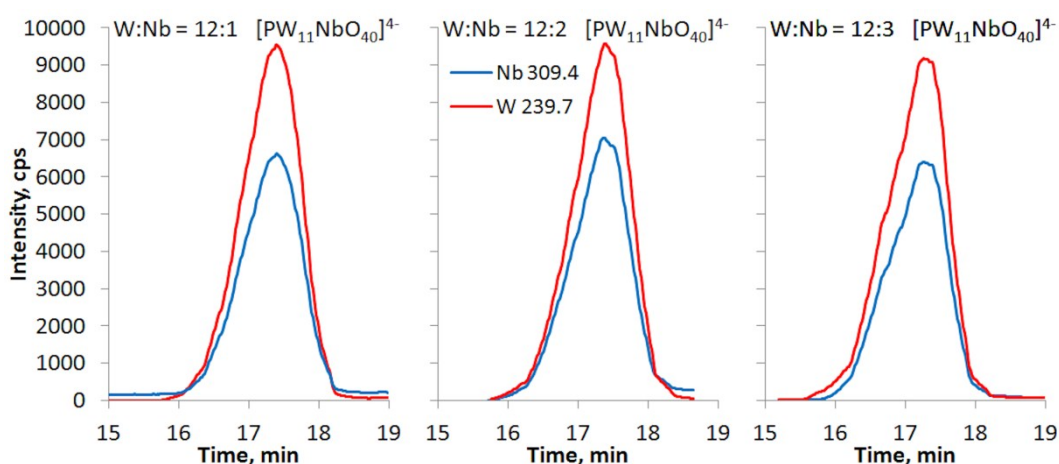


Fig. S2. The HPLC-ICP-AES chromatograms of reaction product at different initial W:Nb ratios in coordinates “retention time–line intensity”. Separation conditions: the same as in Fig. S1.

Table S1. Atomic ratios W/Nb for $[PW_{11}NbO_{40}]^{4-}$ peak calculated from HPLC-ICP-AES-data.

W / Nb	Nb 202.9 nm	Nb 309.4 nm	Nb 316.3	Nb 319.4
W 202.6 nm	11.6±0.7	11.2±0.6	10.6±0.6	10.9±0.7
W 209.8 nm	11.4±0.6	11.1±0.5	10.7±0.5	10.8±0.5
W 229.4 nm	11.5±0.6	11.4±0.6	10.9±0.5	10.7±0.6
W 239.7 nm	11.1±0.6	10.9±0.5	10.8±0.5	10.7±0.5

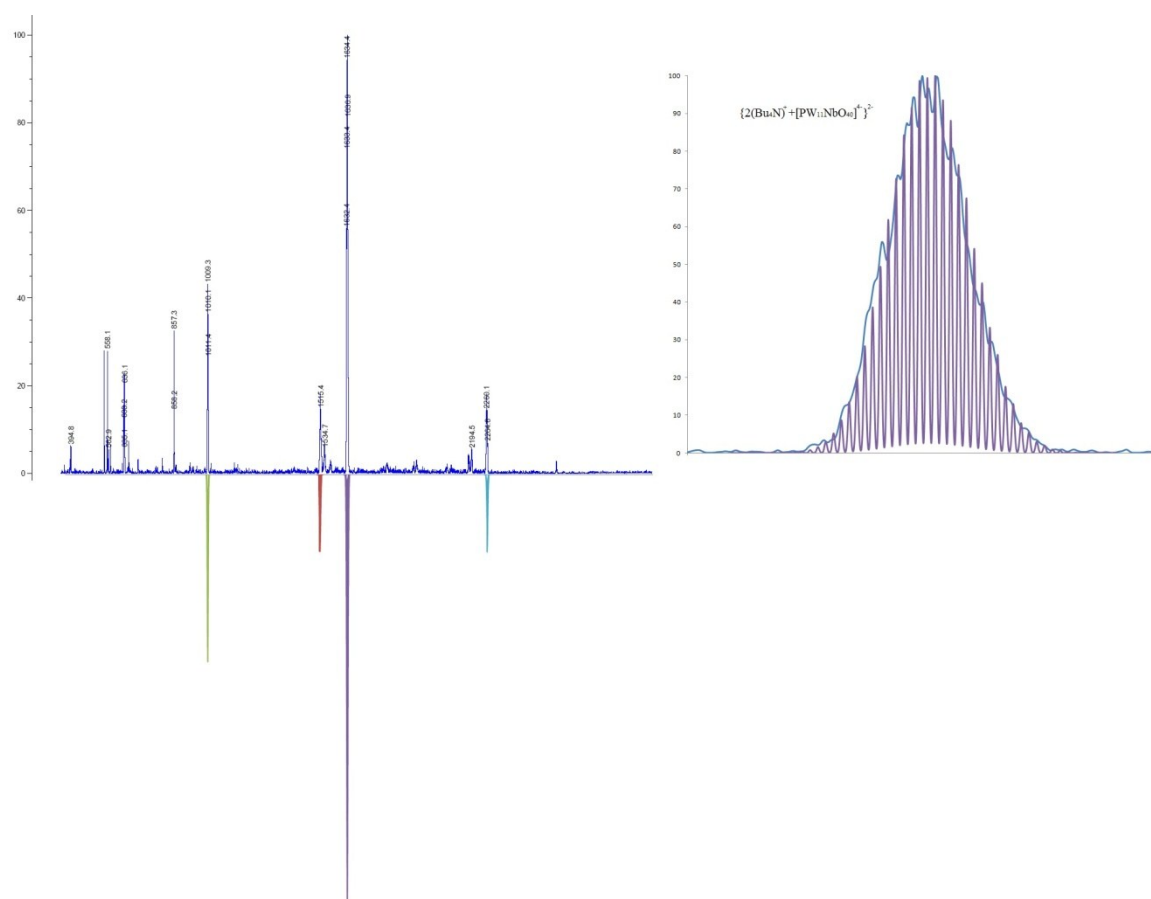


Fig. S3. ESI-MS spectrum of $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ after the reaction of $(\text{TBA})_4\text{H}_3[\text{PW}_{11}\text{O}_{39}]$ and Nb-Ox for 12 hours at 60°C in DMF.

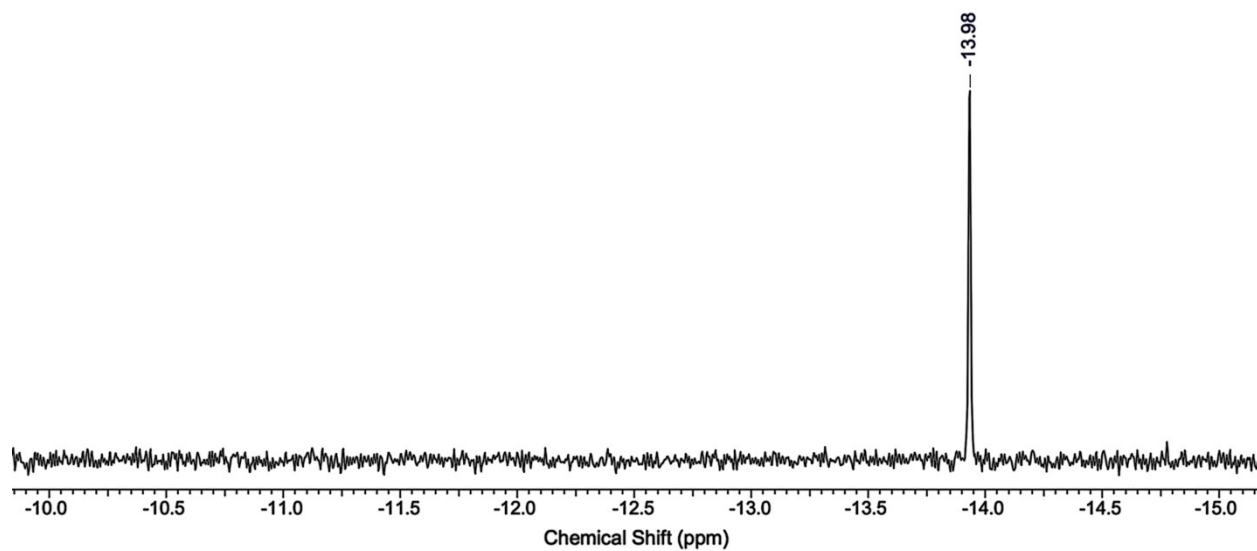


Fig. S4. ^{31}P NMR spectrum of $(\text{TBA})_4[\text{PW}_{11}\text{NbO}_{40}]$ in CD_3CN .

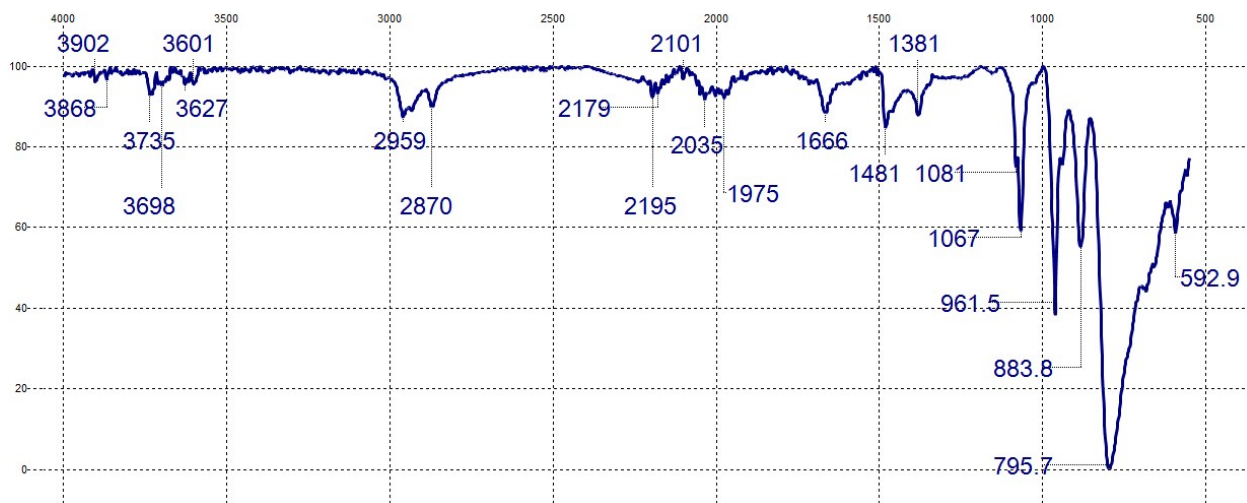


Fig. S5. FT-IR of (TBA)₄[PW₁₁NbO₄₀].

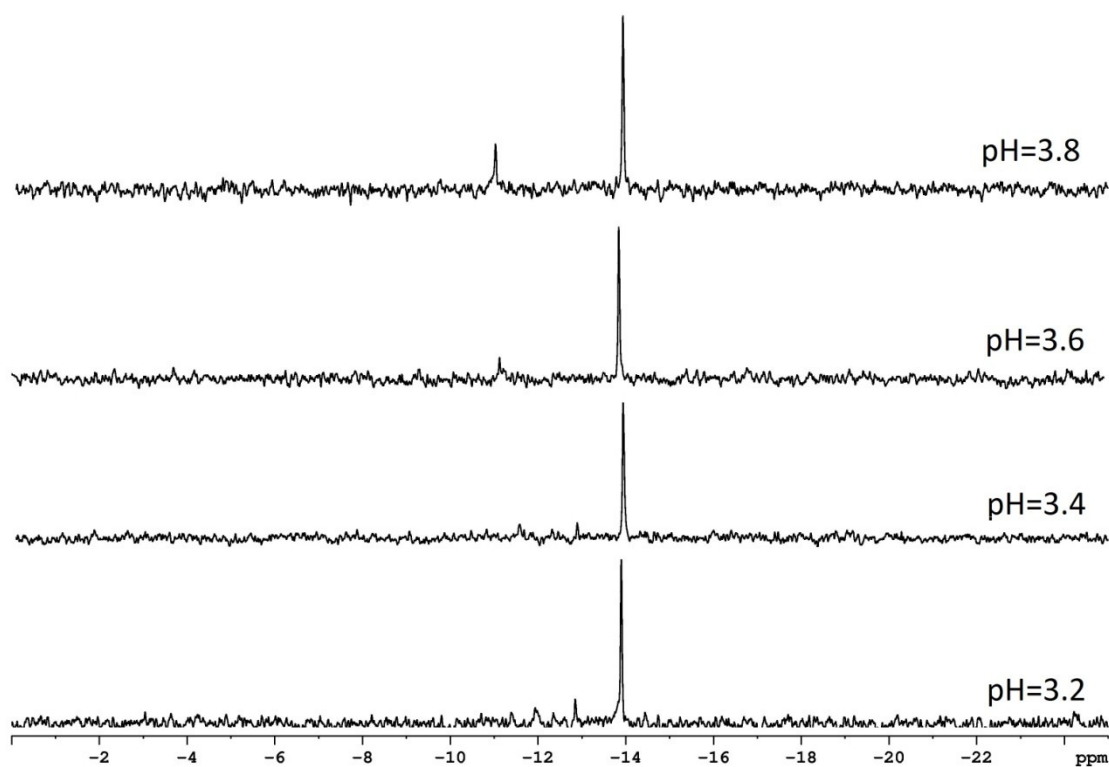


Fig. S6. pH dependency of ³¹P NMR chemical shifts for [PW₁₁NbO₄₀]⁴⁻ between pH 3.2 and 3.8.

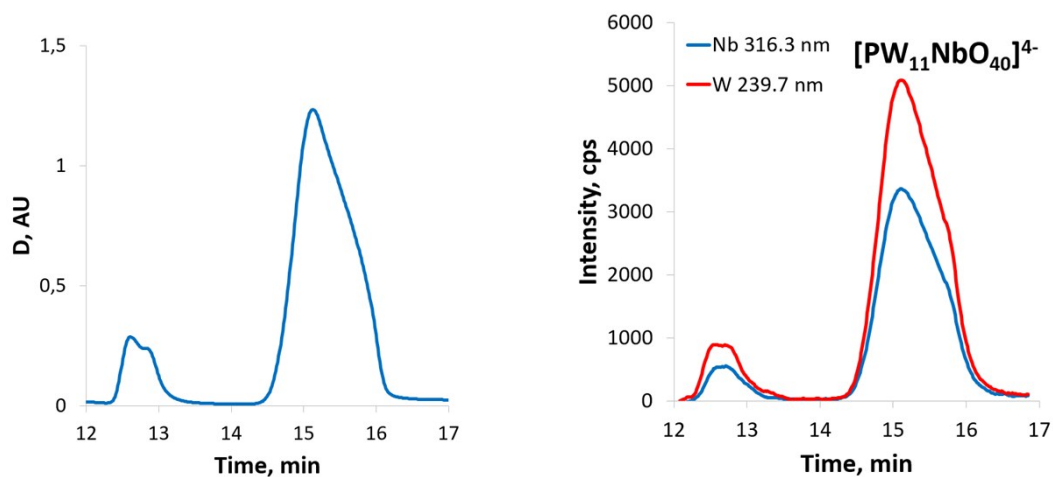


Fig. S7. a) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 3.15.

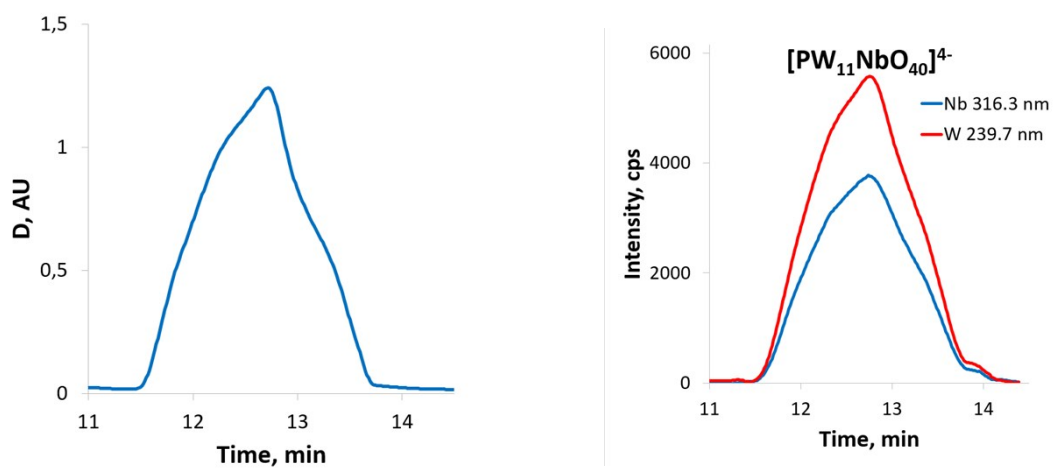


Fig. S7. b) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 4.0.

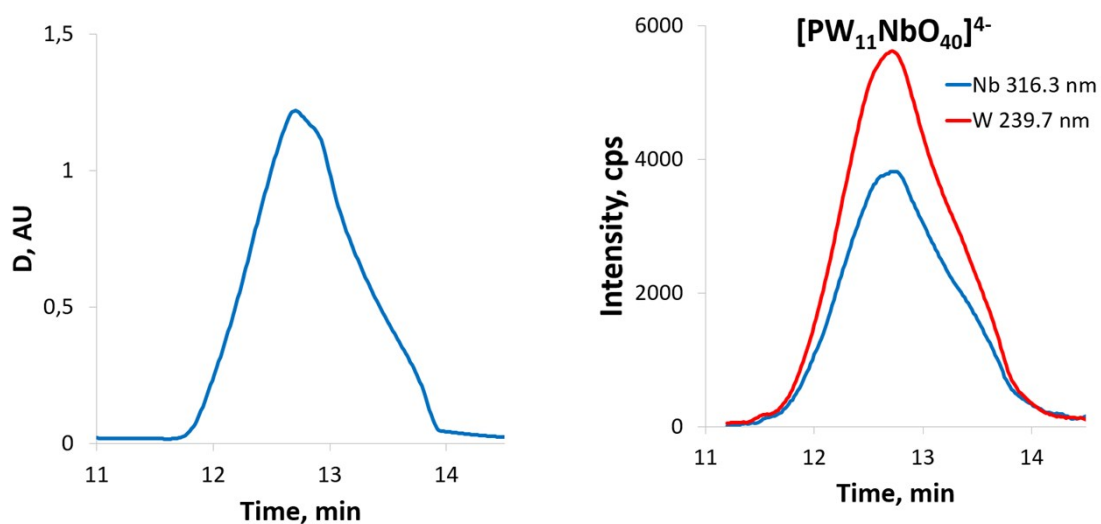


Fig. S7. c) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 4.6.

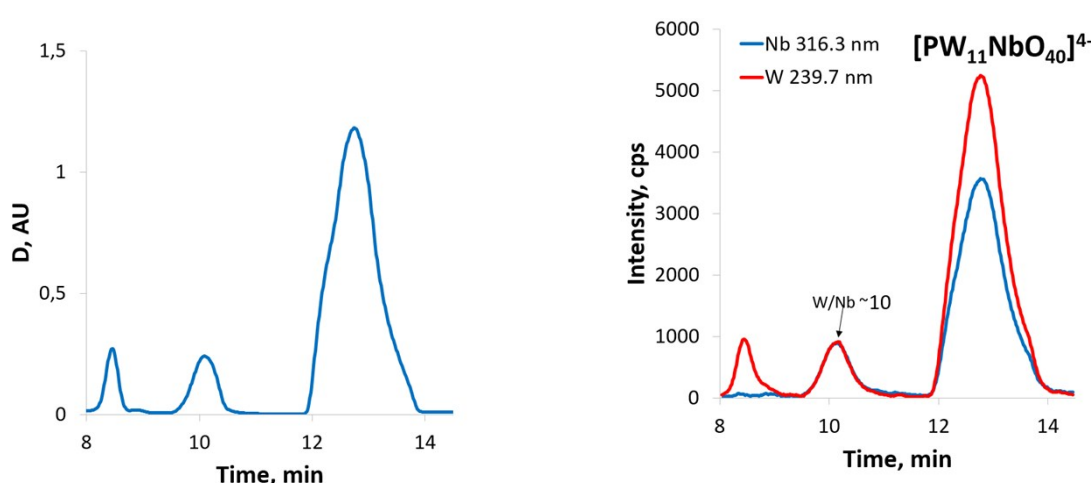


Fig. S7. d) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 5.5.

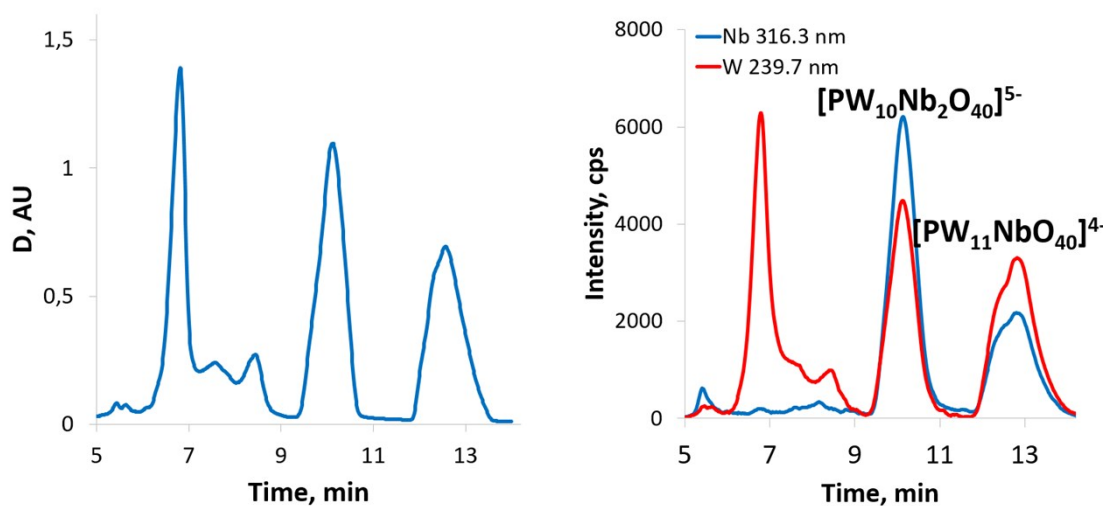


Fig. S7. e) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 6.0.

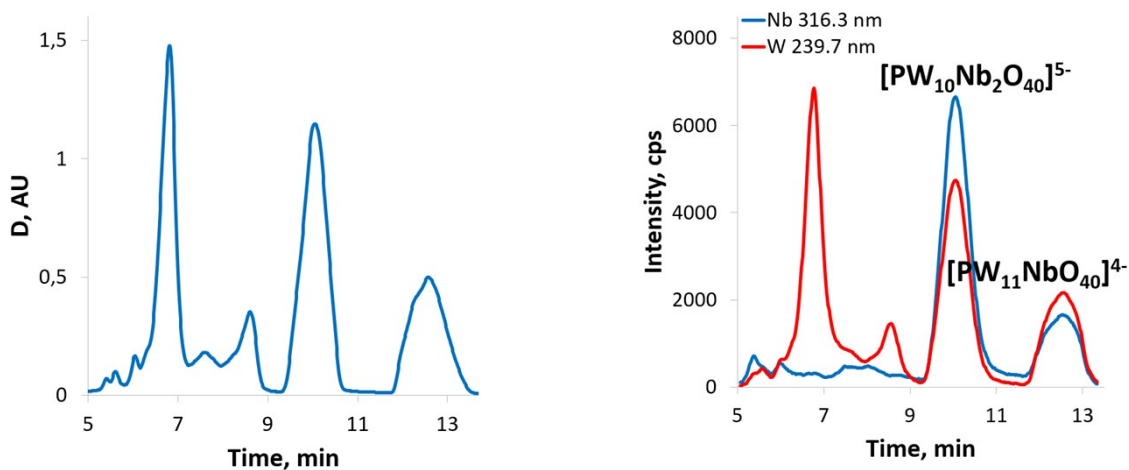


Fig. S7. f) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at pH = 6.4.

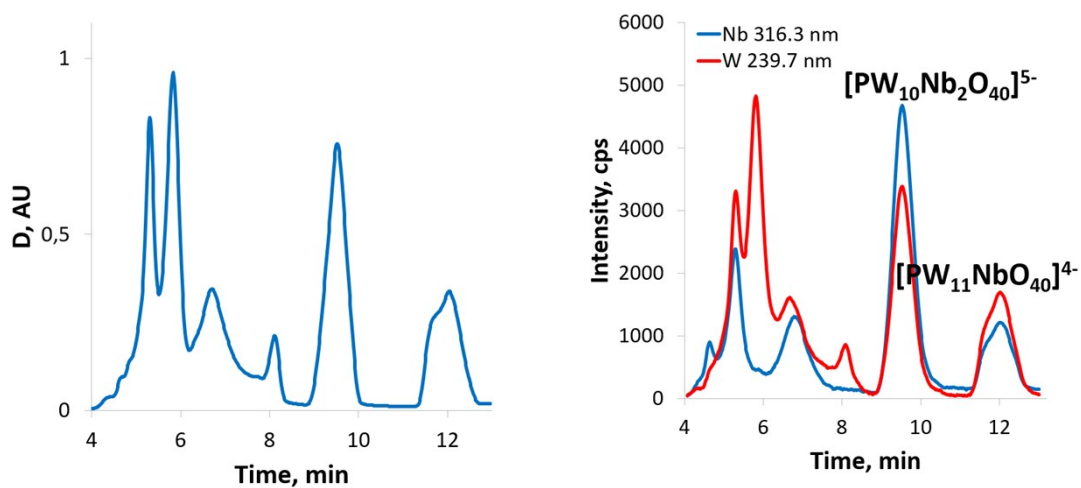


Fig. S7. g) The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of aged $[\text{PW}_{11}\text{NbO}_{40}]^{4-}$ solution at $\text{pH} = 7.08$.

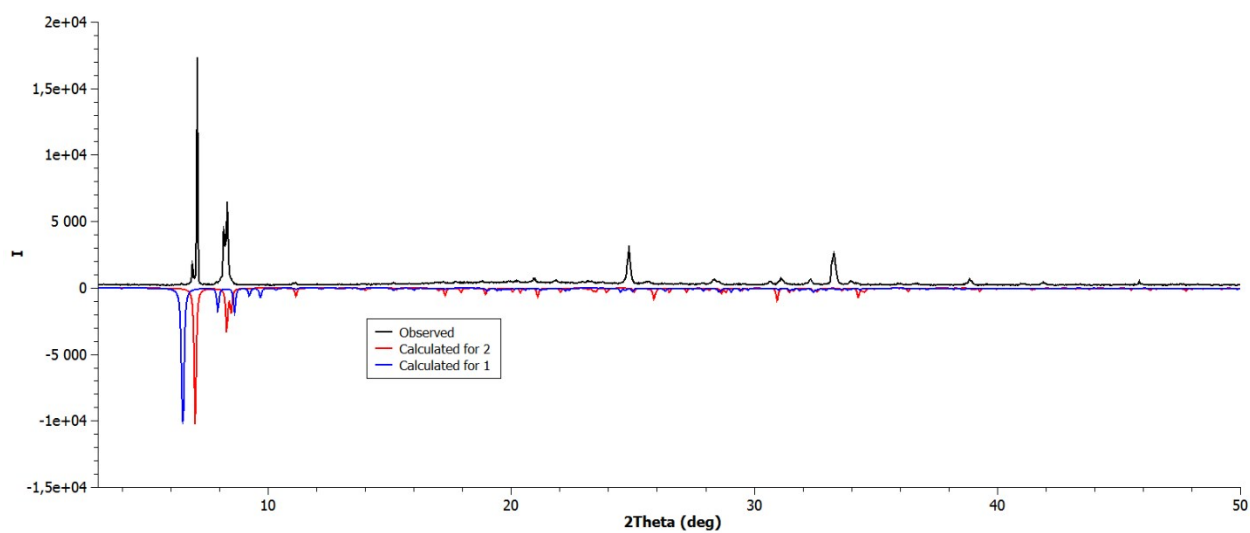


Fig. S8. XRPD patterns for **2** in vaseline oil.

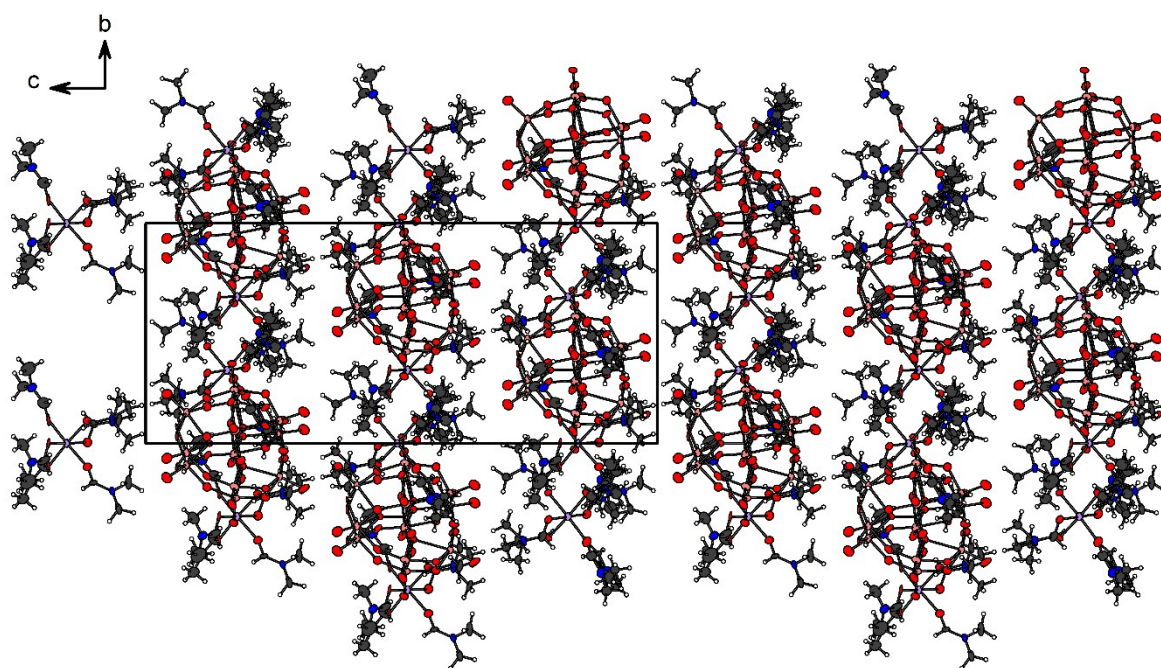


Fig. S9. Crystal packing of **2**.

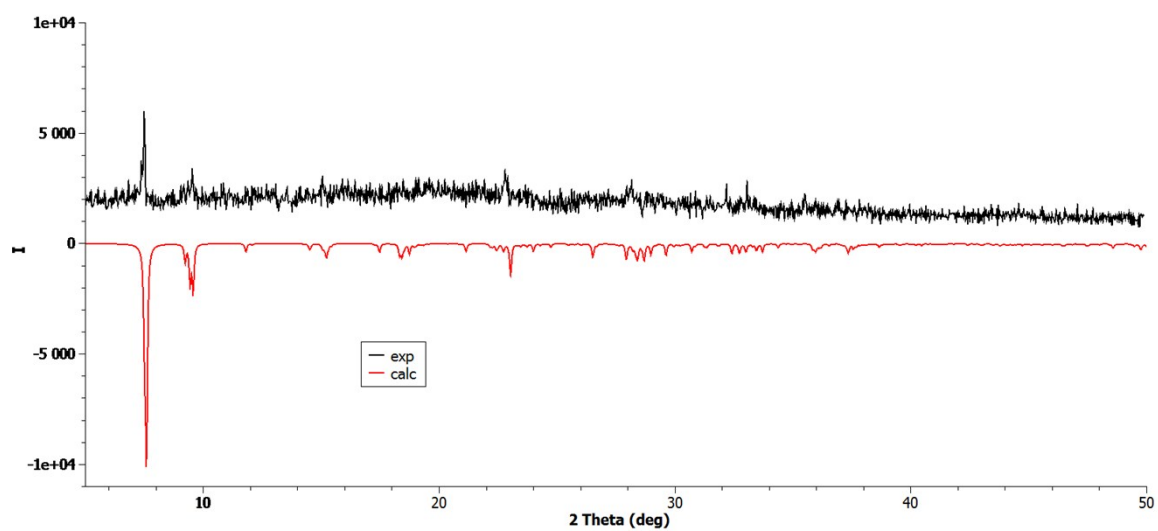


Fig. S10. XRPD pattern for **3** in vaseline oil.

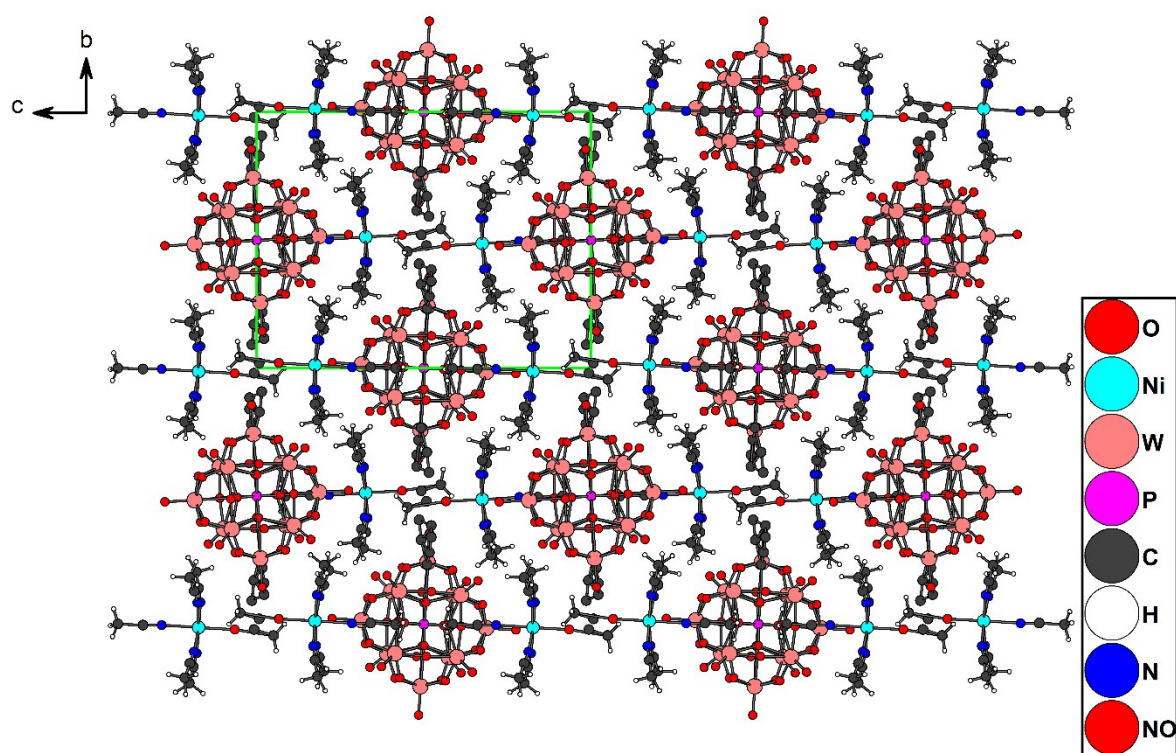


Fig. S11. Crystal packing of **3**.

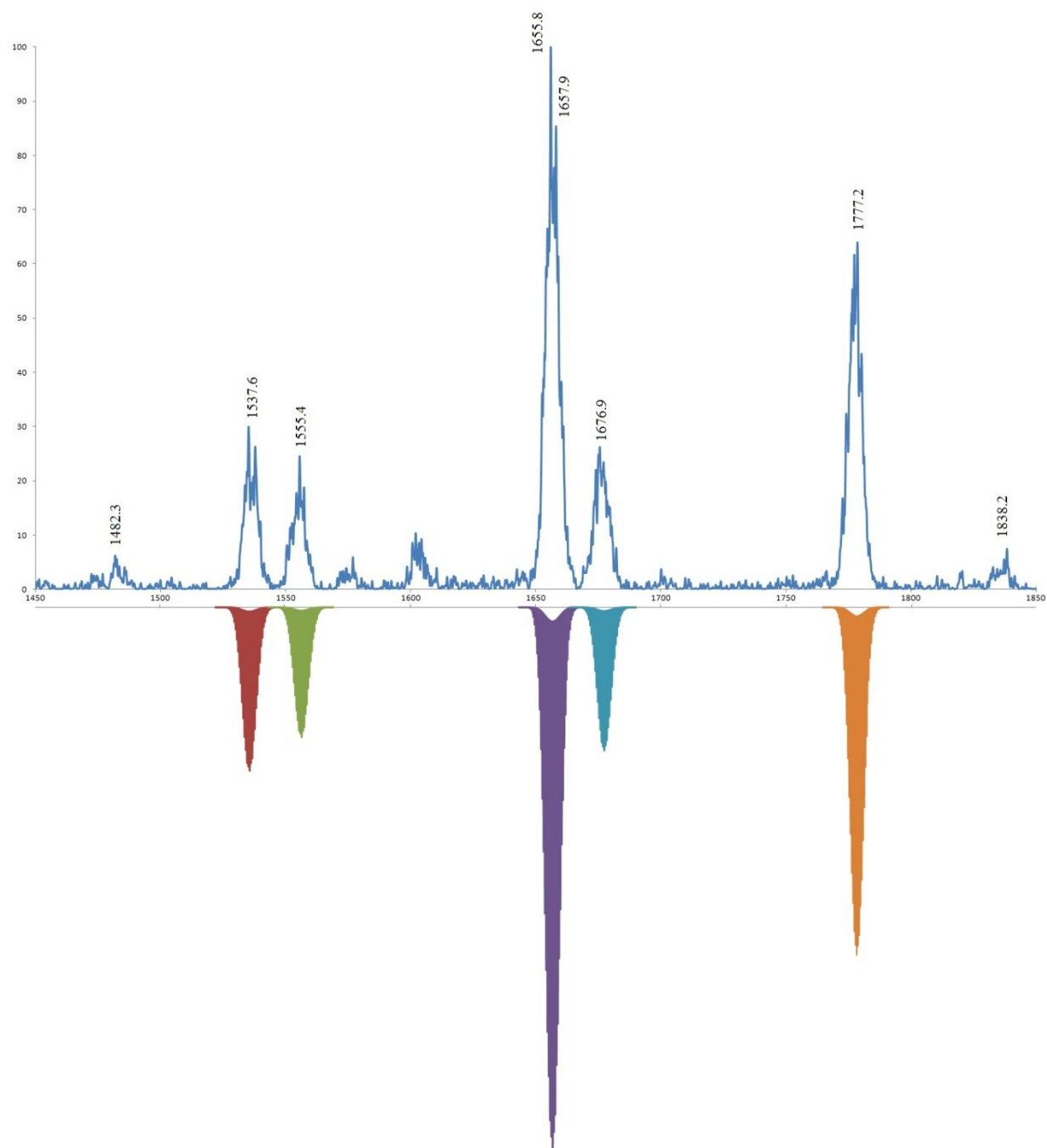


Fig. S12. ESI-MS(-) spectrum of **5** in acetonitrile, up - experimental, down - calculated.

Table S2. Peak assignment for ESI-MS(-) spectrum of **5** in acetonitrile.

Mass/charge	Assignment
1537,6	$\{(\text{Bu}_4\text{N})^+ + 2\text{H}^+ + [\text{GeW}_{11}\text{NbO}_{40}]^{5-}\}^{2-}$
1555,4	$\{(\text{Bu}_4\text{N})^+ + 2\text{H}^+ + [\text{GeW}_{11}\text{NbO}_{40}]^{5-} + \text{MeCN}\}^{2-}$
1655,8	$\{2(\text{Bu}_4\text{N})^+ + \text{H}^+ + [\text{GeW}_{11}\text{NbO}_{40}]^{5-}\}^{2-}$
1676,9	$\{2(\text{Bu}_4\text{N})^+ + \text{H}^+ + [\text{GeW}_{11}\text{NbO}_{40}]^{5-} + \text{MeCN}\}^{2-}$
1777,2	$\{3(\text{Bu}_4\text{N})^+ + [\text{GeW}_{11}\text{NbO}_{40}]^{5-}\}^{2-}$

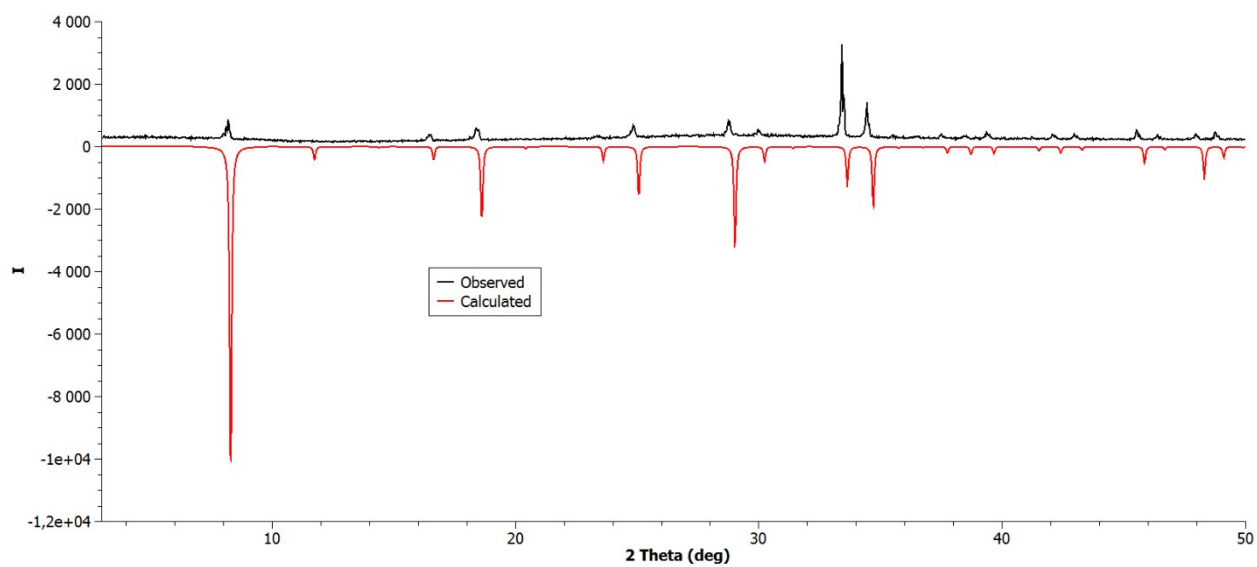


Fig. S13. XRPD pattern for **4**.

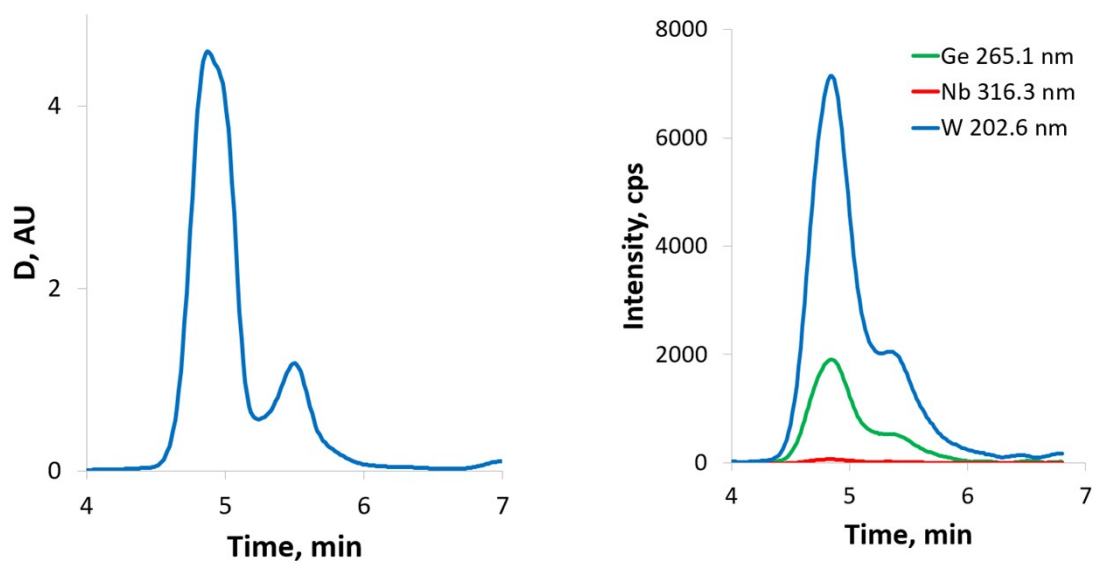


Fig. S14. The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of $[\text{GeW}_{11}\text{O}_{39}]^{8-}$ solution.

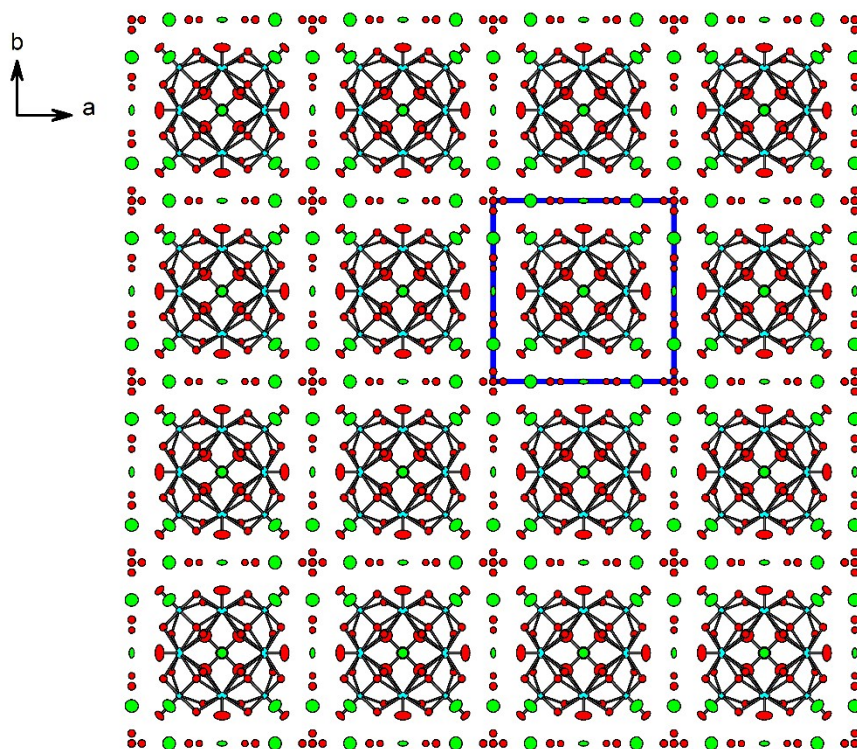


Fig. S15. Crystal packing of **4**.

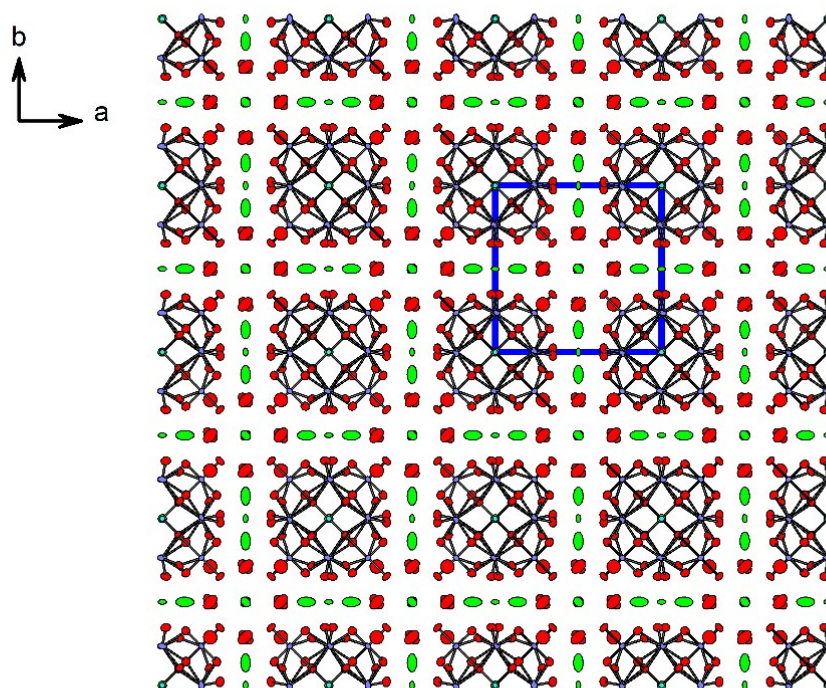


Fig. S16. Crystal packing of $\text{K}_8[\text{GeW}_{11}\text{O}_{39}] \cdot 13\text{H}_2\text{O}$.

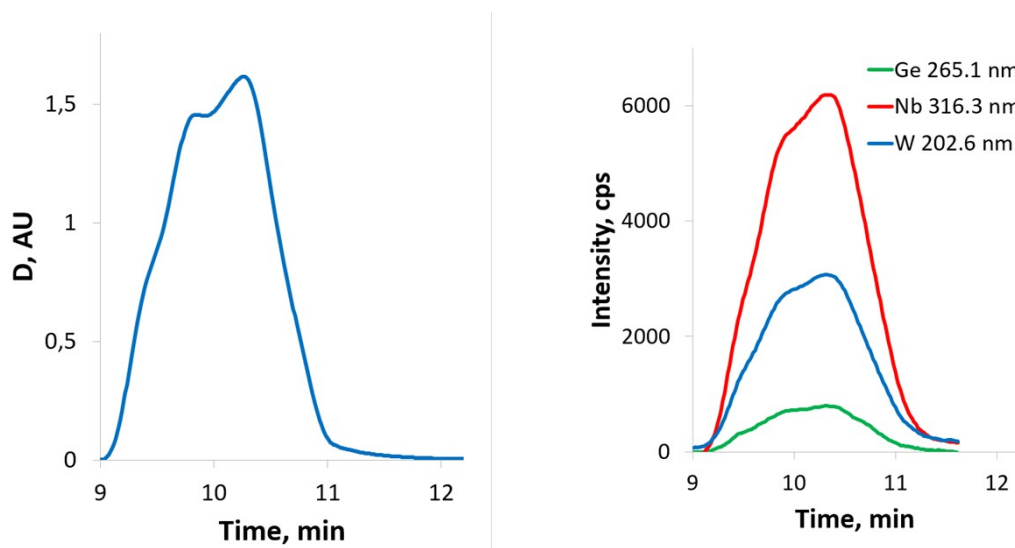


Fig. S17. The HPLC-UV chromatogram (left) and HPLC-ICP-AES chromatogram (right) of $[\text{GeW}_{11}\text{NbO}_{39}]^{5-}$ solution.

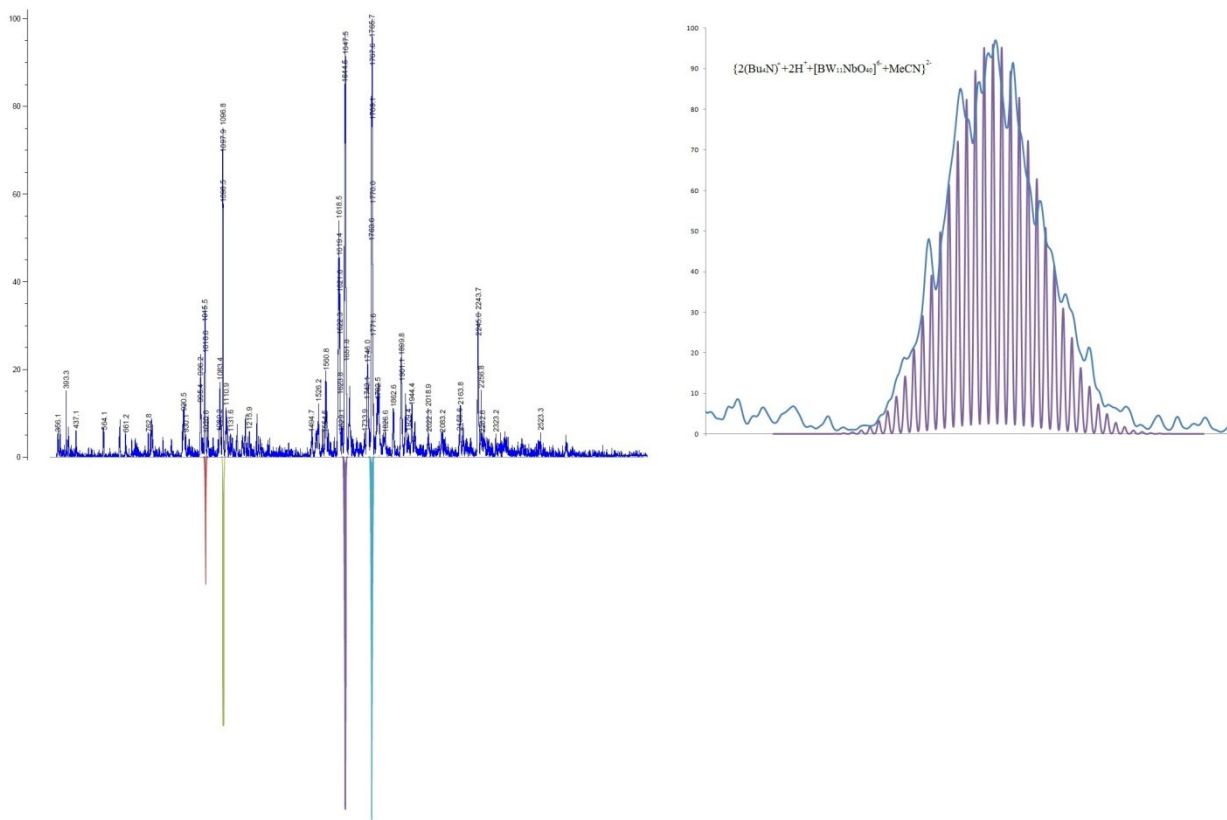


Fig. S18. ESI-MS(-) spectrum of **6** in acetonitrile.

Table S3. Peak assignment for ESI-MS(-) spectrum of **6** in acetonitrile.

Mass/charge	Assignment
1015,5	$\{(\text{Bu}_4\text{N})^+ + 2\text{H}^+ + [\text{BW}_{11}\text{NbO}_{40}]^{6-} + \text{MeCN}\}^{3-}$
1096,8	$\{2(\text{Bu}_4\text{N})^+ + \text{H}^+ + [\text{BW}_{11}\text{NbO}_{40}]^{6-} + \text{MeCN}\}^{3-}$
1647,5	$\{(\text{Bu}_4\text{N})^+ + 3\text{H}^+ + [\text{BW}_{11}\text{NbO}_{40}]^{6-} + \text{MeCN}\}^{2-}$
1765,7	$\{2(\text{Bu}_4\text{N})^+ + 2\text{H}^+ + [\text{BW}_{11}\text{NbO}_{40}]^{6-} + \text{MeCN}\}^{2-}$

Table S4. Experimental details.

Parameter	2	3	4
Chemical formula	C ₃₆ H ₈₄ Co ₂ N ₁₂ O ₅₂ PW ₁₂	C ₂₇ H ₃₃ N ₁₁ Ni ₂ O ₄₁ PW ₁₂	GeK ₅ O _{50.29} W ₁₂
M_r	3872.18	3522.23	3278.95
Crystal system, space group	Trigonal, $R\bar{3}$	Monoclinic, $P2_1/n$	Cubic, $Pm\bar{3}m$
Temperature (K)	130	130	296
a, b, c (Å)	15.8751 (6), 15.8751 (6), 31.9698 (14)	12.6148 (8), 14.6506 (7), 19.8436 (12)	10.6496 (1), 10.6496 (1), 10.6496 (1)
α, β, γ (°)	90, 90, 120	90, 105.190 (6), 90	90, 90, 90
V (Å ³)	6977.6 (6)	3539.3 (4)	1207.81 (3)
Z	3	2	1
μ (mm ⁻¹)	15.22	20.04	29.60
Crystal size (mm)	0.22 × 0.22 × 0.06	0.15 × 0.10 × 0.08	0.10 × 0.10 × 0.10
Diffractometer	New Xcalibur, AtlasS2	New Xcalibur, AtlasS2	Bruker Apex Duo
Absorption correction	Multi-scan <i>CrysAlis PRO</i> 1.171.38.41 (Rigaku Oxford Diffraction, 2015) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.	Multi-scan <i>CrysAlis PRO</i> 1.171.38.41 (Rigaku Oxford Diffraction, 2015) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.	Multi-scan <i>SADABS</i> (Bruker-AXS, 2004)
$T_{\min}, T_{\max}$	0.144, 1.000	0.321, 1.000	0.551, 0.747
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	6745, 4974, 4549	27962, 6230, 3763	6866, 507, 443
R_{int}	0.042	0.114	0.047
θ values (°)	$\theta_{\max} = 29.6, \theta_{\min} = 3.5$	$\theta_{\max} = 25.0, \theta_{\min} = 3.3$	$\theta_{\max} = 33.0, \theta_{\min} = 1.9$
$(\sin \theta/\lambda)_{\max}$ (Å ⁻¹)	0.694	0.595	0.767
Range of h, k, l	$-17 \leq h \leq 16, -21 \leq k \leq 9, -43 \leq l \leq 28$	$-15 \leq h \leq 15, -16 \leq k \leq 17, -23 \leq l \leq 22$	$-16 \leq h \leq 8, -7 \leq k \leq 16, -15 \leq l \leq 9$
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.051, 0.125, 1.01	0.074, 0.182, 1.06	0.035, 0.110, 1.16
No. of reflections, parameters, restraints	4974, 343, 52	6230, 414, 78	507, 29, 0
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained	H-atoms were not refined
Weighting scheme	$w = 1/[\sigma^2(F_o^2) + (0.0706P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$	$w = 1/[\sigma^2(F_o^2) + (0.0581P)^2 + 97.5865P]$ where $P = (F_o^2 + 2F_c^2)/3$	$w = 1/[\sigma^2(F_o^2) + (0.0523P)^2 + 20.1908P]$ where $P = (F_o^2 + 2F_c^2)/3$
$\Delta\rho_{\max}, \Delta\rho_{\min}$ (e Å ⁻³)	3.51, -4.37	2.56, -2.35	2.42, -2.38

Experiments were carried out with Mo $K\alpha$ radiation.

Computer programs: *CrysAlis PRO* 1.171.38.41 (Rigaku OD, 2015), *APEX2* (Bruker-AXS, 2004), *SAINT* (Bruker-AXS, 2004), *SHELXS2014* (Sheldrick, 2014), *SHELXL2014* (Sheldrick, 2014), *ShelXle* (Hübschle, 2011), *CIFTAB-2014* (Sheldrick, 2014).

Table S5. Selected geometric parameters (Å).

2			
O10—Co2	2.049 (18)	O6—W3	1.918 (14)
O11—Co2	2.111 (16)	O7—W2	1.931 (14)
O12—Co1	2.095 (17)	O7—W3	1.915 (15)
O19—Co1	2.054 (15)	O8—W1	1.981 (16)
O1—P1	1.52 (3)	O8—W2	1.889 (14)
O17—P1	1.556 (13)	O9—W1	1.672 (16)
O1—W4 ⁱ	2.464 (14)	O13—W2 ⁱⁱ	1.884 (14)
O1—W4 ⁱⁱ	2.464 (14)	O13—W2	1.926 (14)
O1—W4	2.464 (14)	O14—W2	1.714 (18)
O2—W4 ⁱⁱ	1.878 (14)	O15—W3	1.882 (16)
O2—W4	1.930 (15)	O15—W4	1.924 (16)
O3—W4	1.705 (19)	O16—W1	1.853 (17)
O4—W1 ⁱⁱ	1.898 (16)	O16—W4 ⁱ	1.945 (17)
O4—W3	1.916 (16)	O17—W1 ⁱ	2.426 (13)
O5—W3	1.730 (16)	O17—W2 ⁱ	2.417 (14)
O6—W1	1.914 (15)	O17—W3 ⁱ	2.445 (13)
3			
Ni1—N1	2.07 (2)	O11—W5	1.845 (19)
Ni1—N2	2.04 (2)	O12—W1	1.87 (2)
Ni1—N3	2.12 (2)	O12—W4	1.905 (18)
Ni1—N4	2.05 (2)	O13—W3	1.90 (2)
Ni1—N5	2.07 (2)	O13—W5	1.91 (2)
Ni1—N6	2.08 (3)	O14—W4	1.88 (2)
O18—P1	1.57 (3)	O14—W6 ⁱⁱⁱ	1.93 (2)
O19—P1	1.59 (3)	O15—W2	1.86 (2)
O22—P1	1.49 (4)	O15—W6	1.94 (2)
O23—P1	1.43 (3)	O16—W1	1.90 (2)
O1—W5	1.699 (19)	O16—W6	1.90 (3)
O2—W4	1.86 (2)	O17—W1	1.94 (3)
O2—W5	1.89 (2)	O17—W2	1.85 (3)
O3—W4	1.70 (2)	O18—W1	2.43 (3)
O4—W3 ⁱⁱⁱ	1.918 (18)	O18—W2	2.41 (3)
O4—W4	1.872 (19)	O18—W6	2.53 (3)
O5—W2	1.94 (2)	O19—W4	2.45 (3)
O5—W3	1.863 (19)	O19—W5	2.37 (3)
O6—W6	1.649 (17)	O19—W6 ⁱⁱⁱ	2.51 (3)
O7—W1 ⁱⁱⁱ	1.90 (2)	O20—W5 ⁱⁱⁱ	1.93 (2)
O7—W3	1.90 (2)	O20—W6	1.85 (2)

O8—W1	1.689 (19)	O21—W3	1.71 (3)
O10—W2	1.665 (18)	O22—W3	2.38 (4)
O11—W2	1.935 (19)	O23—W3	2.50 (3)
4			
O3—Ge1	1.76 (2)	O2—W1	1.659 (11)
O1B—W1 ^{iv}	1.850 (4)	O3—W1 ^{iv}	2.342 (10)
O1B—W1	1.850 (4)	O3—W1 ^v	2.342 (10)
O1A—W1 ^{iv}	2.046 (7)	O3—W1	2.342 (10)
O1A—W1	2.046 (7)		

Symmetry code(s): (i) $-y+1, x-y+1, z$; (ii) $-x+y, -x+1, z$; (iii) $-x+1, -y+2, -z+1$; (iv) $y, -z+1, x$; (v) $z, -x+1, -y+1$.