## **Supporting Information**

## An Efficient Ternary CoP<sub>2x</sub>Se<sub>2(1-x)</sub> Nanowire Array for overall Water Splitting

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**Fig. S1**. SEM (a) and TEM (b) images of  $Co(OH)_2$ , XRD pattern of  $Co(OH)_2$  (c).



Fig. S2. Low-magnification SEM (a) and high-magnification SEM (b) of  $CoP_{2x}Se_{2(1-x)}NWs$ .



**Fig. S3**Polarization curves of CoSe<sub>2</sub>NWs for HER in both 0.5 M H<sub>2</sub>SO<sub>4</sub> and 1.0 M KOH solution.



**Fig. S4.** Polarization curves of  $CoP_{0.45}Se_{1.55}$ ,  $CoP_{0.97}Se_{1.03}$  and  $CoP_{1.46}Se_{0.54}$  NWs for HER in both 0.5 M H<sub>2</sub>SO<sub>4</sub> and 1.0 M KOH solution (a and c). The corresponding Tafel plots for these ternary samples (b and d).



**Fig. S5.** Cyclic voltammograms (CV) curves of  $CoP_{2x}Se_{2(1-x)}$  NWs (a) and  $CoSe_2$  NWs (b) at scan rates of 20mv/s-200 mv/s in 0.5 M H<sub>2</sub>SO<sub>4</sub>. (c) Linear fitting of C<sub>dl</sub> plots at 0.05 V vs. RHE of the catalysts vs. scan rate in 0.5 M H<sub>2</sub>SO<sub>4</sub>. Cyclic voltammograms (CV) curves of  $CoP_{2x}Se_{2(1-x)}$  NWs (d) and  $CoSe_2$  NWs(e) at scan rates of 10mv/s-100 mv/s in 1.0 M KOH. (f) Linear fitting of the C<sub>dl</sub> plots at 0.10 V vs. RHE of the catalysts vs. scan rate in 1.0 KOH.



**Fig. S6.** SEM image of  $CoP_{1.37}Se_{0.63}$  NWs after HER test in 0.5 M H<sub>2</sub>SO<sub>4</sub>. SEM (b), TEM (c) and HRTEM (d) of  $CoP_{1.37}Se_{0.63}$  NWs after HER test in1.0 M KOH.



Fig. S7. XRD patterns of  $CoP_{1.37}Se_{0.63}$  NWs after HER test in both 0.5 M  $H_2SO_4$  and 1.0 M KOH solution.

**Table S1.** Comparison of HER performance of  $CoP_{1.37}Se_{0.63}$  NWswith other reported catalysts in both acidic and alkalinesolution.<sup>1-10</sup>

Catalyst	Electrolyte	η @ 10 mA cm <sup>-2</sup> (mV vs. RHE)	Tafel slope (mV dec⁻¹)	Reference
CoSe <sub>2</sub> nanoparticles	0.5 M H <sub>2</sub> SO <sub>4</sub>	137	42.1	Ref. 1
NiP <sub>1.93</sub> Se <sub>0.07</sub> nanoflakes	0.5 M H <sub>2</sub> SO <sub>4</sub>	84	41	Ref. 2
porous NiSe <sub>2</sub> nanosheets	0.5 M H <sub>2</sub> SO <sub>4</sub>	135	37.2	Ref. 3
Ni <sub>5</sub> P <sub>4</sub> films	0.5 M H <sub>2</sub> SO <sub>4</sub>	140	40	Ref. 4
CoS <sub>2x</sub> Se <sub>2(1-x)</sub> NWs	0.5 M H <sub>2</sub> SO <sub>4</sub>	129.5	44	Ref. 5
CP@Ni-P	0.5 M H <sub>2</sub> SO <sub>4</sub>	98	58.8	Ref. 6
MoSSe	0.5 M H <sub>2</sub> SO <sub>4</sub>	164±2	48±2	Ref. 7
MoP S	0.5 M H <sub>2</sub> SO <sub>4</sub>	64	50	Ref. 8
CoP <sub>1.37</sub> Se <sub>0.63</sub> NWs	0.5 M H <sub>2</sub> SO <sub>4</sub>	70	54	This work
Ni <sub>5</sub> P <sub>4</sub> films	1 М КОН	150	53	Ref. 4
porous NiSe <sub>2</sub> nanosheets	1 M KOH	184	76.6	Ref. 3
CP@Ni-P	1 М КОН	117	85.4	Ref. 6
porous CoP nanowires	1 M KOH	209	129	Ref. 9
CP/CTs/Co-S	1 М КОН	190	131	Ref. 10
CoP <sub>1.37</sub> Se <sub>0.63</sub> NWs	1 M KOH	98	74	This work

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