Supporting Information

Colloidal synthesis of urchin-like Fe doped NiSe₂ for efficient

oxygen evolution

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	Table S1 Composition of NiFeSe catalysts							
Catalysts	1	2	3	4				
Precursor ratio Ni:Fe:Se	NiFe _{0.23} Se ₂	NiFe _{0.53} Se ₂	NiFe _{0.75} Se ₂	NiFeSe ₂				
ICP result	$Ni_{1.16}Fe_{0.27}Se_2$	$Ni_{1.12}Fe_{0.49}Se_2$	$Ni_{1.06}Fe_{0.58}Se_2$	$Ni_{1.02}Fe_{0.88}Se_2$				

Catalyst	electrolyte	support	loading	J	Overpotential	ref
			(mg/cm^2)	(mA/cm ²)	(mV)	
Ni ₃ Se ₂	0.3 M KOH	GC	0.217	10	290	1
$Co_{0.13}Ni_{0.87}Se_2$	1 M KOH	Ti	1.67	100	320	2
NiSe ₂	1 M KOH	Ti	2.5	20	295	3
NiSe ₂	1 M KOH	GC	1	10	250	4
FeNiSe	1 M KOH	FeNi foam	4.8	100	264	5
NiSe	1 M KOH	Ni foam	2.8	20	270	6
NiSe	1 M KOH	Ni foam		10	320	7
Ni ₃ Se ₂	1 M KOH	CF	3	50	340	8
Ni ₃ Se ₂	1 M KOH	Ni foam	8.87	20	242	9
(Ni, Co) _{0.85} Se	1 M KOH	carbon	5	10	255	10
		cloth				
(Ni,Co) _{0.85} Se @ NiCo	1 M KOH	carbon	6	10	216	10
LDH		cloth				
NiSe ₂	1 M KOH	Ni foam		50	350	11
$Ni_{0.5}Fe_{0.5}Se_2$	1 M KOH	CFC		50	350	12
CoSe	1 M KOH	GC	0.28	10	295	13
Co ₇ Se ₈	1 M KOH	GC		10	290	14
NiCo-selenide	0.1 M KOH	CFP	0.2	10	393	15
CoSe	1 M KOH	Ti	3.8	10	292	16
CoSe	0.1 M KOH	NG	0.2	10	366	17
Co _{0.85} Se/NiFe-LDH	1 M KOH	EG		150	270	18
CoSe ₂	0.1 M KOH	GC	0.142	10	320	19

Table S2 Com	narison (of OFR	catalytic	nerformances	with re	norted meta	l selenides
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CoOx–CoSe	1 M KOH	Ni foam	1.7	100	300	20
$CeO_2/CoSe_2$	0.1 M KOH	GC	0.2	10	288	21
$Ni_{0.75}Fe_{0.25}Se_2$	1 M KOH	CFC	1.5	100	277	22
NixFe _{1-x} Se ₂ -DO	1 M KOH	Ni foam		10	195	23
Ni _{1.12} Fe _{0.49} Se ₂	1 M KOH	XC-72	0.45	10	227	this
						work



Figure S1. XRD patterns of $Ni_{1.12}Fe_{0.49}Se_2$ before (a) and after (b) OER stability test.



Figure S2. TEM image of $Ni_{1.12}Fe_{0.49}Se_2\!/XC\text{-}72$ after OER stability test.



Figure S3. XPS spectra of Ni 2p (a), Fe 2p (b), Se 3d (c) in Ni_{1.12}Fe_{0.49}e₂ catalyst before (i) and after (ii) OER stability test.

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