Indocyanine green-loaded gold nanostars for sensitive SERS imaging and subcellular monitoring of photothermal therapy

Jing Chen,¹ Zonghai Sheng,² Penghui Li,³ Manxiang Wu,⁴ Nisi Zhang,² Xue-Feng Yu,³ Yuanwen Wang,¹ Dehong Hu,² Hairong Zheng,^{*2} Guo Ping Wang^{*1}

¹College of Electronic Science and Technology, Guangdong Provincial Key Laboratory of Micro/Nano Optomechatronics Engineering and Key Laboratory of Optoelectronic Devices and Systems of Ministry of Education and Guangdong Province, Shenzhen University, Shenzhen 518060, China. E-mail: gpwang@szu.edu.cn

²Paul C. Lauterbur Research Center for Biomedical Imaging, Institute of Biomedical and Health Engineering, Shenzhen Institutes of Advanced Technology, Chinese Academy of Sciences, Shenzhen 518055, China. E-mail: hr.zheng@siat.ac.cn

³Institute of Biomedicine and Biotechnology, Shenzhen Institutes of Advanced Technology, Chinese Academy of Sciences, Shenzhen 518055, China.

⁴Department of Medicine Ultrasonics, Nanfang Hospital, Southern Medical University, Guangzhou, China.

Enhancement Factor (EF) Calculation

To calculate the EF of GNS-ICG-BSA SERS nanotags, a 10^{-2} M of ICG solution is used for normal Raman detection, and 10 μ L of GNS-ICG-BSA solution with the adsorbed ICG concentration of 10^{-4} M is used for SERS detection.

The most widely used definition for EF is^[1]

$$EF = \frac{I_{surf}}{I_{bulk}} \times \frac{N_{bulk}}{N_{surf}}$$

where I_{surf} and I_{bulk} are the integrated intensities of ICG molecules adsorbed on GNS-ICG-BSA SERS nanoprobes and from 10⁻² M of ICG bulk solution, respectively. N_{surf} and N_{bulk} are the corresponding numbers of ICG molecules adsorbed on GNS-ICG-BSA SERS nanoprobes and in the bulk solution effectively illuminated by the laser beam, respectively.

$$N_{bulk} = Ahc_{bulk}N_A$$

where *A* is the area of the laser focal spot, *h* is the confocal depth of the laser, and *h* is 13 μ m according to our previous work,^[2] c_{bulk} is the concentration of ICG bulk

solution, here $c_{bulk}=10^{-2}$ M, N_A is the Avogadro constant.

$$N_{surf} = \frac{c_{surf} v N_A A}{\pi r^2}$$

where c_{surf} is the concentration of ICG molecules adsorbed on GNS-ICG-BSA SERS nanoprobes, here $c_{surf}=10^{-4}$ M. v is the volume of ICG solution contained in 10 µL of GNS-ICG-BSA solution. According to the procedure 2.2 in the manuscript, the ratio of ICG reporter solution to GNS-ICG-BSA solution is 0.03, so here v is 0.3 µL. r is the radius of 10 µL of GNS-ICG-BSA solution formed on the silicon substrate, r=3.5 mm.

Figure S1a, S1b are the normal Raman spectrum of 10^{-2} M ICG solution and SERS spectrum of 10^{-4} M GNS-ICG-BSA solution, respectively. The integrated intensities of the bands for I_{bulk} (1431 cm⁻¹) and I_{surf} (1444 cm⁻¹) are 3089 and 373862 cps, respectively. Considering the different number of molecules in each unit volume for normal Raman spectrum and SERS spectrum acquisition,^[2] I_{surf} / I_{bulk} =373862×10² / 3089.

Finally, the EF of GNS-ICG-BSA SERS nanoprobes can be calculated as 2.02×10⁶.



Figure S1 (a) Normal Raman spectrum of 10⁻² M ICG solution. Laser power: 15.7 mW. (b) SERS spectrum of 10⁻⁴ M GNS-ICG-BSA solution. Laser power: 15.7 mW.



Figure S2 (a) TEM image and (b) size distribution of GNS-ICG-BSA nanoprobe after laser irradiation (808 nm, 3.0 W/cm²) for 7 min.



Figure S3 GNS-ICG-BSA nanotags were heated to 45 °C by 808 nm laser (1.0 W/cm^2), and the SERS spectra were acquired immediately and after the storage at room temperature for 24 h, respectively.



Figure S4 The SERS spectra of (a) GNS-R6G-BSA and (b) GNS-CV-BSA nanotags without and with 808 nm laser irradiation.



Scheme S1 Schematic diagram of ICG decomposition mechanism upon laser irradiation. An alkene (double bond in a polymethine chain) with a general structure is shown, and ¹R to ⁴R represent different functional groups.

References

[1] E. C. Le Ru, E. Blackie, M. Meyer and P. G. Etchegoin, J. Phys. Chem. C 2007, 111, 13794.

[2] J. Chen, B. Shen, G. W. Qin, X. W. Hu, L. H. Qian, Z. W. Wang, S. Li, Y. P. Ren and L. Zuo, J. Phys. Chem. C, 2012, 116, 3320.