Supporting Information

One-Pot Mass Preparation of MoS_2/C Aerogel for High-Performance Supercapacitors and Lithium-Ion Batteries

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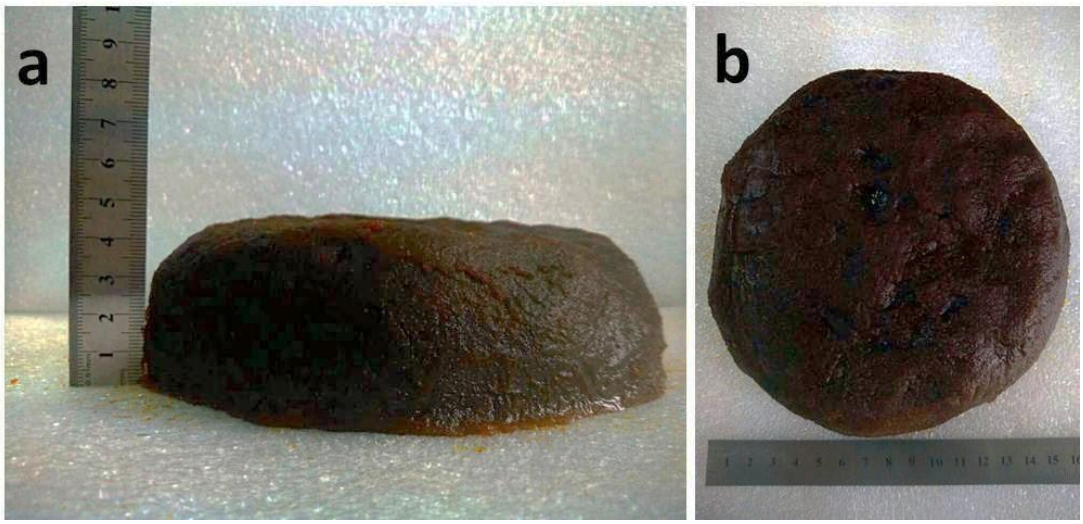

Fig. S1 Optical photographs of the obtained precursor gel in a larger scale.

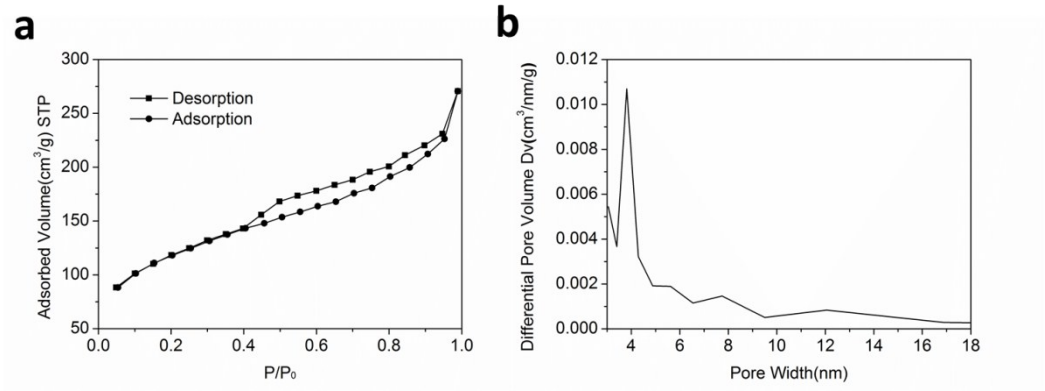

Fig. S2 (a) Nitrogen adsorption-desorption isotherms and (b) pore-size distribution curves of the MoS_2 composite aerogel.

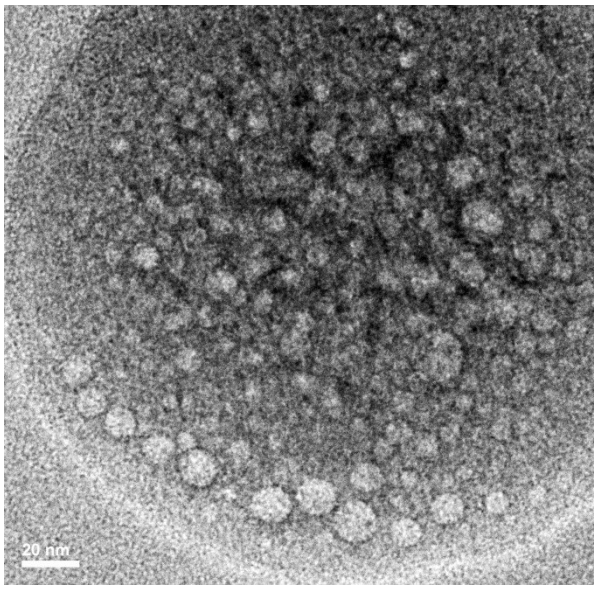

Fig. S3 TEM image of the MoS_2/C composite aerogel.

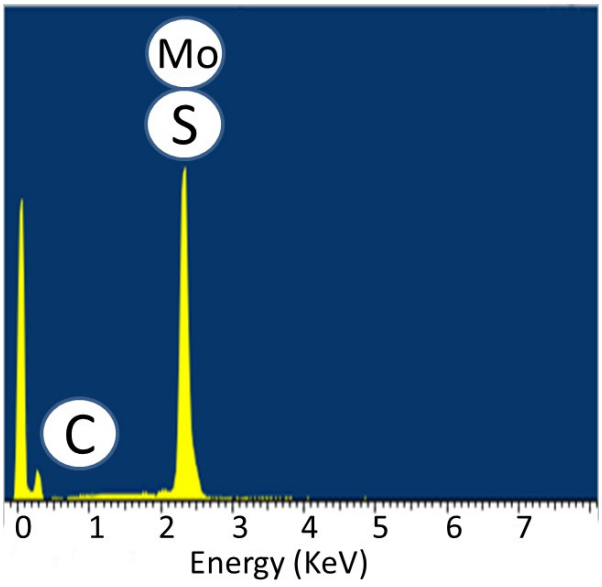

Fig. S4 Energy dispersive X-ray spectrometry (EDS) spectrum of the MoS_2/C composite aerogel.

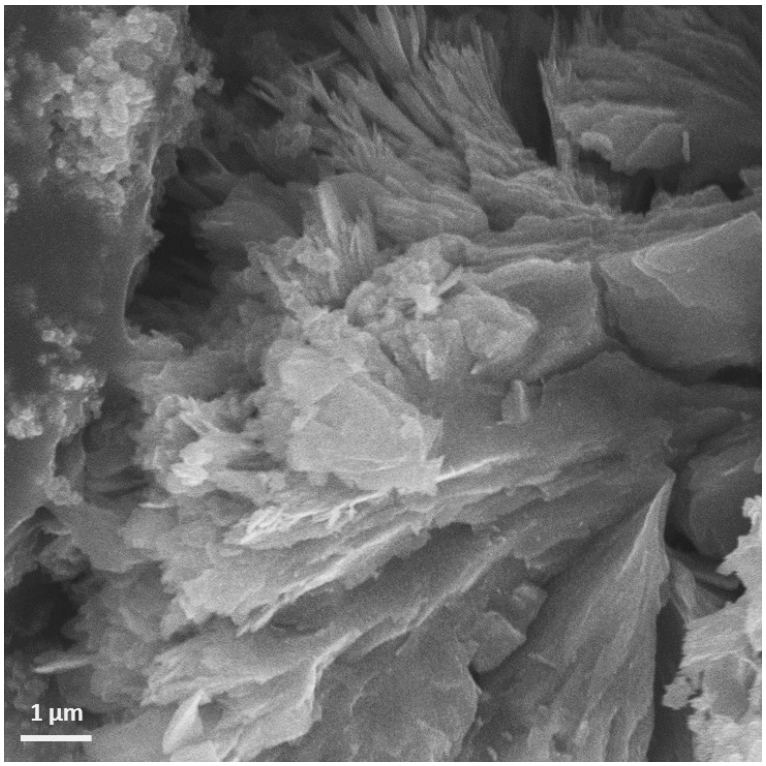

Fig. S5 SEM image of MC-3 overloaded with MoS_2.

Tab. S1 Summary of electrochemical performance comparisons of various MoS_2/C composites used as electrode materials in supercapacitors.

	Material	Capacity at (X) Current Density	Coulombic Efficiency after (Y) Cycles	At Current Density
a	MoS_2/RGO@PANI	1224 F g^{-1} (1 A g^{-1})	82.5% (3 000)	10 A g^{-1}
b	three-dimensional graphene/MoS_2	410 F g^{-1} (1 A g^{-1})	80.3% (10 000)	2 A g^{-1}
c	MoS_2/N-doped graphene	245 F g^{-1} (0.25 A g^{-1})	91.3% (1 000)	2 A g^{-1}
d	MoS_2/microporous carbons	189 F g^{-1} (1 A g^{-1})	98% (3 000)	10 A g^{-1}
e	MoS_2/PANI	552 F g^{-1} (0.5 A g^{-1})	79% (6 000)	1 A g^{-1}
	This work	712.6 F g^{-1} (1 A g^{-1})	97.3% (13 000)	6 A g^{-1}

Note: a, b, c, d and e correspond to Ref.s 29, 30, 31, 32 and 33, respectively, in the main text.

Tab. S2 The mass percentage of MoS_2 and C in the three samples, calculated based on ICP results.

Sample	Mass percentage of MoS_2 (%)	Mass percentage of C (%)
MC-1	22.9	77.1
MC-2	41.3	58.7
MC-3	59.2	40.8