Supplementary information

Insight into the chemical adsorption properties of CO molecules on supported Au or Cu and hybridized Au-CuO nanoparticles

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Supplementary information



Figure S1. Photos displaying the distinctive colors of samples before and after calcination process.



Figure S2. TEM images of CuO/AO (a), Au/AO (b), and size distribution of gold nanoparticles in Au/AO (c).



Figure S3. XRD patterns of typical samples (left) and zoomed spectra in the range of 33-40° (right).



Figure S4. CO-TPD profiles of pure Al_2O_3 powder. The deposition compounds are monitored with m/z intensity of 28 and 44 for CO and CO₂, respectively, at a heating rate of 5 °C min⁻¹ in helium.



Figure S5. H₂-TPR profiles of the Au/AO, CuO/AO and Au-CuO/AO catalysts. The reduction compounds are monitored by mass spectroscopy with m/z intensity of 18 for H₂O, at a heating rate of 10 °C \cdot min⁻¹ in 5% H₂ balanced with helium.



Figure S6. STEM image and size distribution of Au/AO-3.2 with smaller AuNPs.



Figure S7. CO conversion over three catalysts in CO oxidation as a function of reaction temperature. Conditions: Gas flow rate of 20 ml \cdot min⁻¹, 50 mg catalyst, 1 vol % CO in air. The error bars were marked based on three continuous reaction cycles from room temperature to 300 °C.



Figure S8. FT-IR spectra of different samples as CO adsorption for 30 min in the range of 2250-1980 cm⁻¹.



Figure S9. Operando DRIFT spectra of CuO/AO (a-b) and Au/AO (c-d) samples as CO adsorption for 30 min in the range of 2250-1980 cm⁻¹.



Figure S10. Operando DRIFT spectra of Au/AO- $_{3,2}$ at 25 °C under: CO adsorption (left) and He purging (right) until the adsorption peaks become stable. Conditions: Gas flow rate of 20 ml·min⁻¹, 2 vol % CO in helium, 30 mg catalyst.



Figure S11. FT-IR spectra of Au-CuO/AO under different temperature at steady state in CO adsorption in the range of 2250-1980 cm⁻¹.



Figure S12. CO conversion over Au-CuO/AO as a function of reaction temperature, 2^{nd} reaction cycle of calcined Au-CuO/AO (\blacksquare), 1^{st} (\triangleleft) and 2^{nd} (\bigstar) reaction cycles after reduction at 300 °C in H₂ for 1 h.

Catalyst	Au/Cu		Au 4f			Cu 2p	
	XPS	ICP	Au ⁰	Au^{δ^+}	BE of Au ⁰ (eV)	Cu ⁰ or Cu ⁺	Cu^{2+}
Au/AO	-	-	80.3%	19.7%	83.7	-	-
Cu/AO	-	-	-	-	-	88.1%	11.9%
Au-Cu/AO	1:5.2	1:1.6	90.5%	9.5%	83.4	89.0%	11.0%

Table S1 Information of surface metal compositions revealed by XPS analysis.