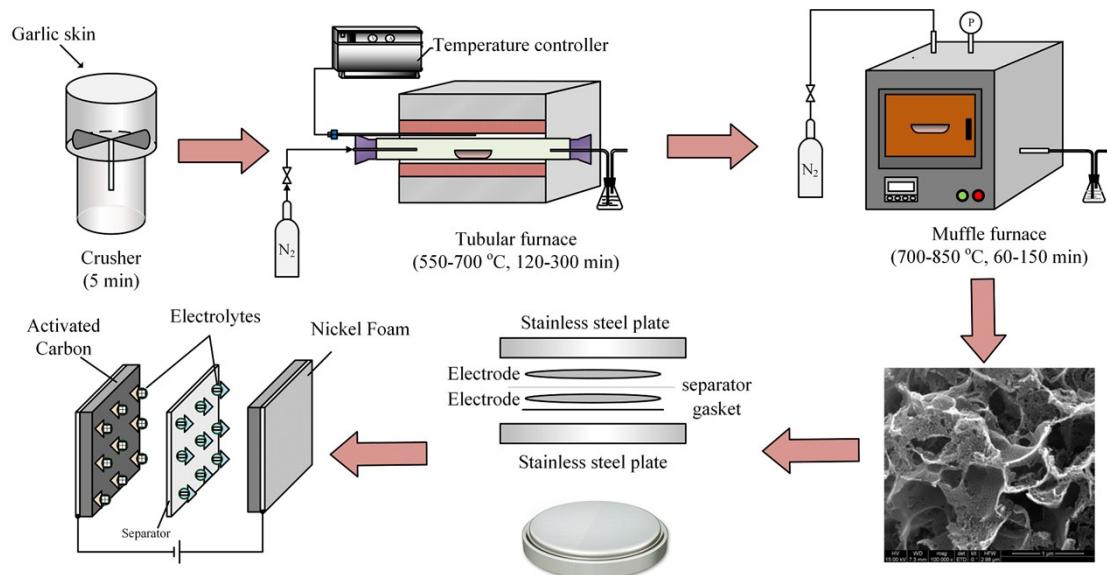


Electronic Supporting Information for

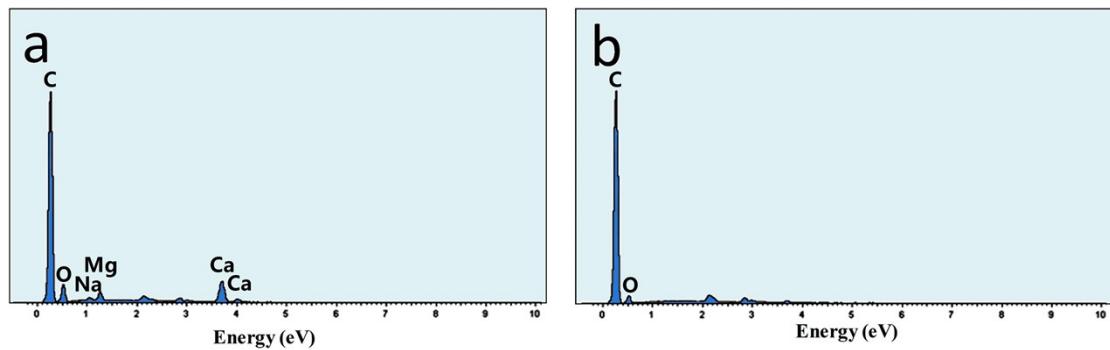
**Synthesis of Garlic Skin-Derived 3D Hierarchical  
Porous Carbon for High-Performance  
Supercapacitors**

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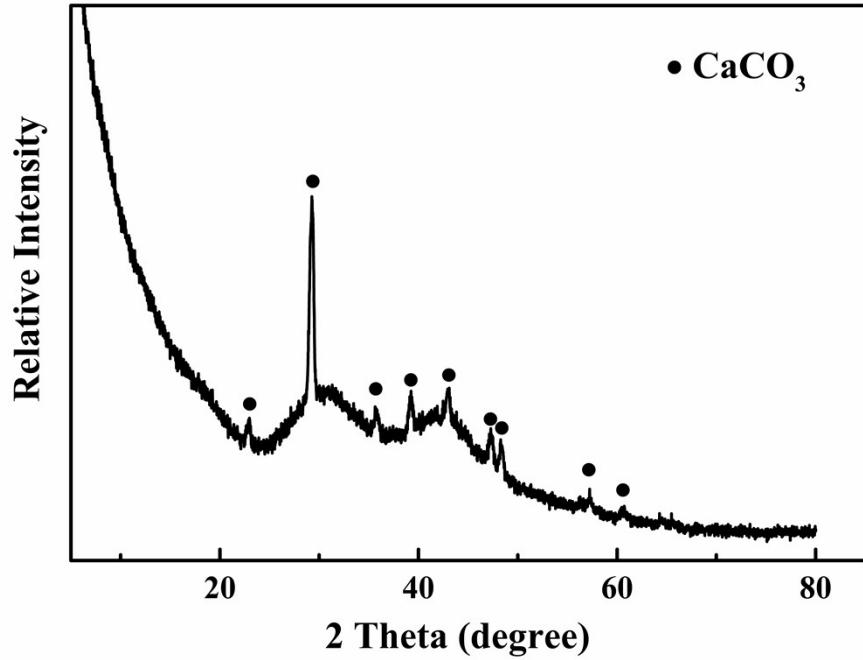
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China. E-mail: hankh@163.com



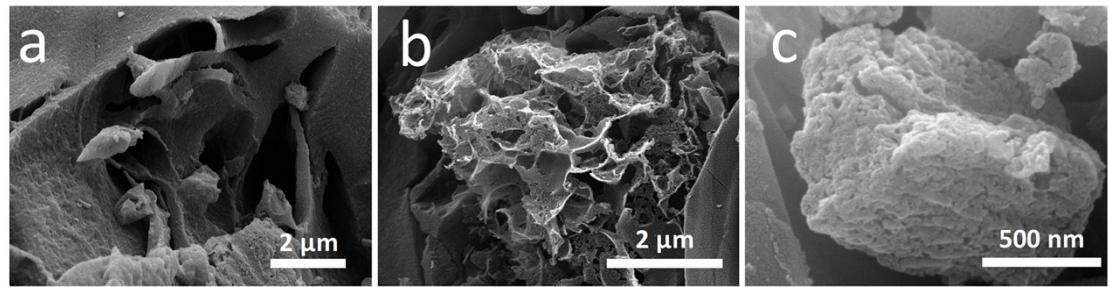
**Figure S1** Schematic illustration for the preparation of garlic-skin-derived hierarchical porous carbon.



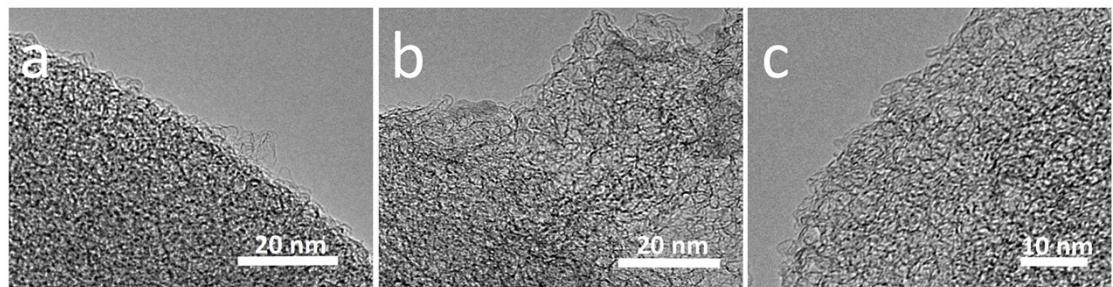
**Figure S2** (a) SEM-EDS image of GHCs before acid washing. (b) SEM-EDS image of GHCs after acid washing.



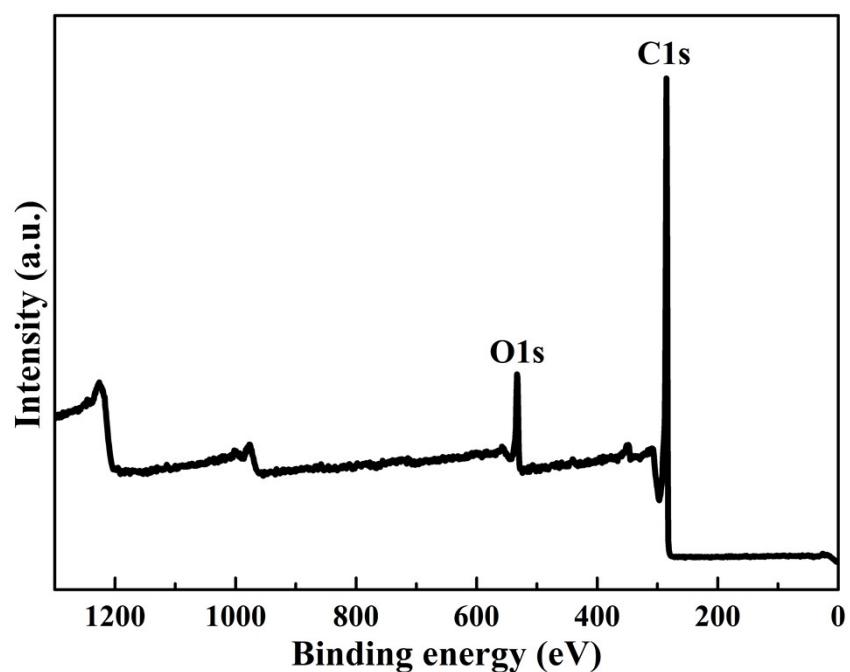
**Figure S3** XRD pattern of GHCs before acid washing.



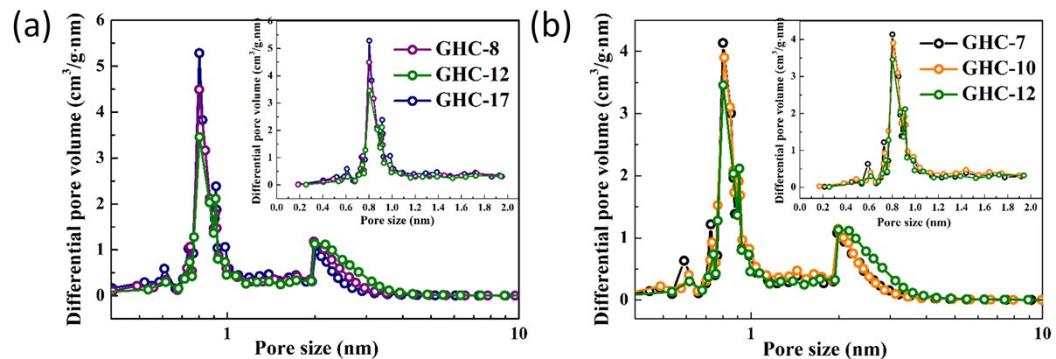
**Figure S4** SEM images of GHCs (a and b) before and (c) after acid washing.



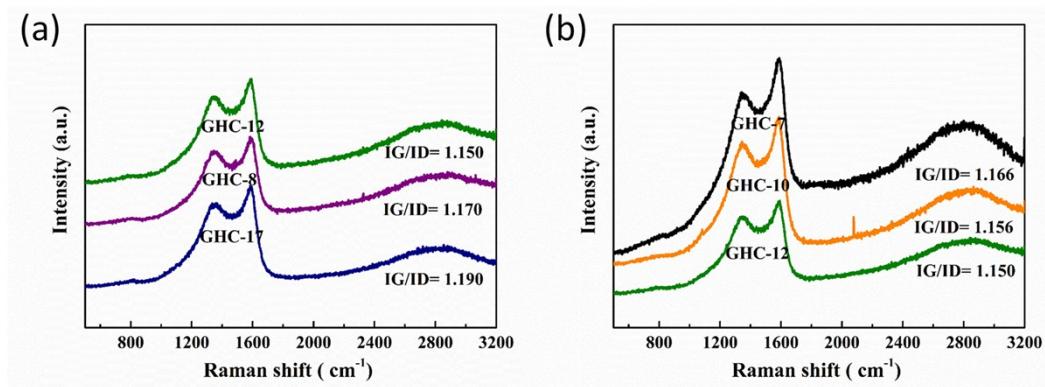
**Figure S5** TEM images of GHCs.



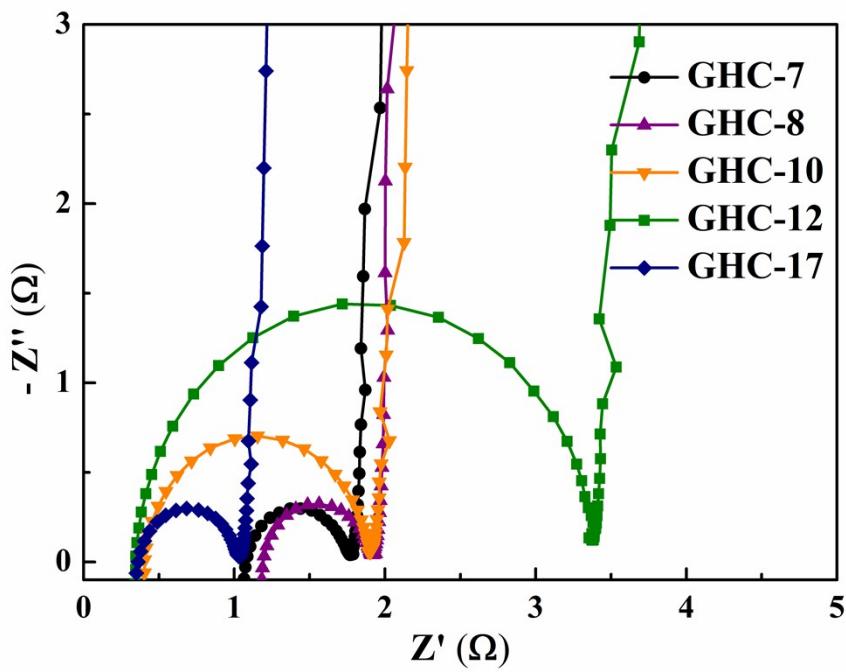
**Figure S6** X-ray photoelectron spectroscopy (XPS) analysis of GHC-17.



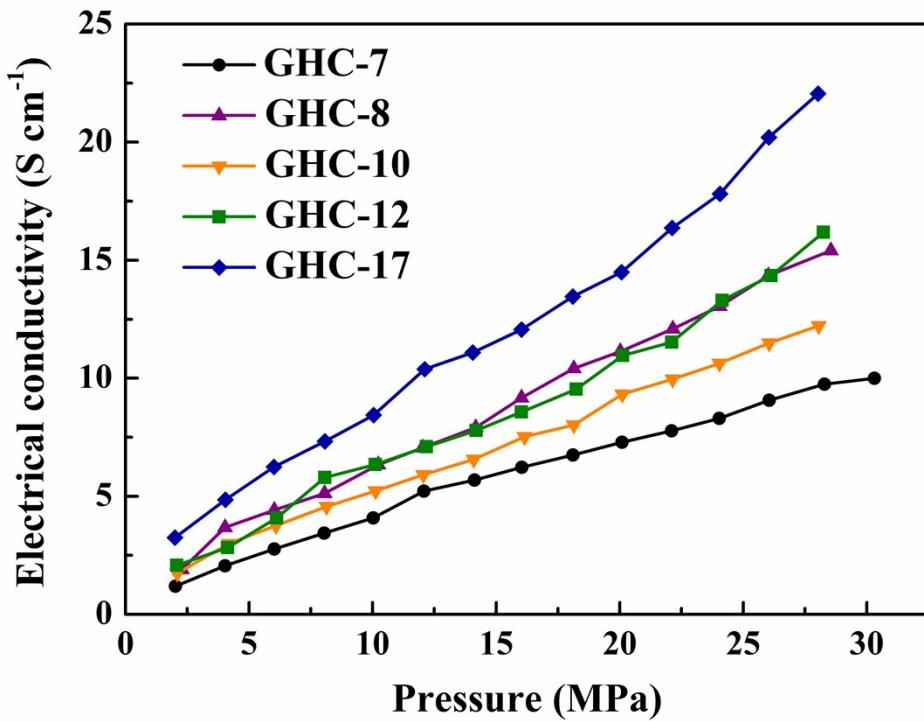
**Figure S7** Pore size distribution curve of (a) GHC-8, GHC-12, and GHC-17; (b) GHC-7, GHC-10, and GHC-12.



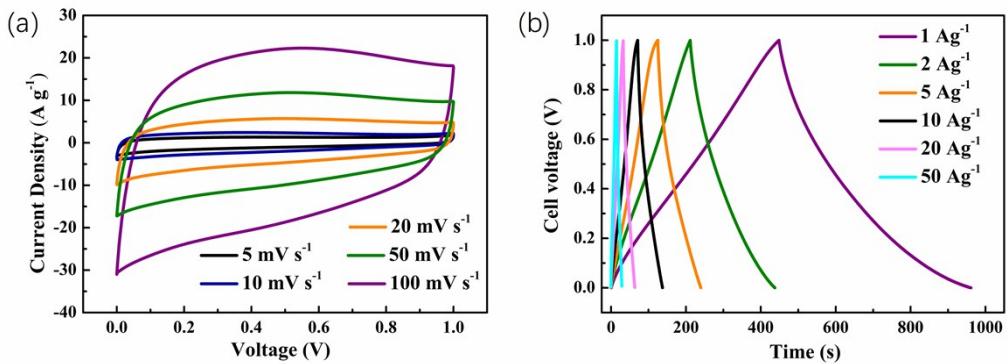
**Figure S8** (a) Raman spectrum of (a) GHC-8, GHC-12, and GHC-17; (b) GHC-7, GHC-10, and GHC-12.



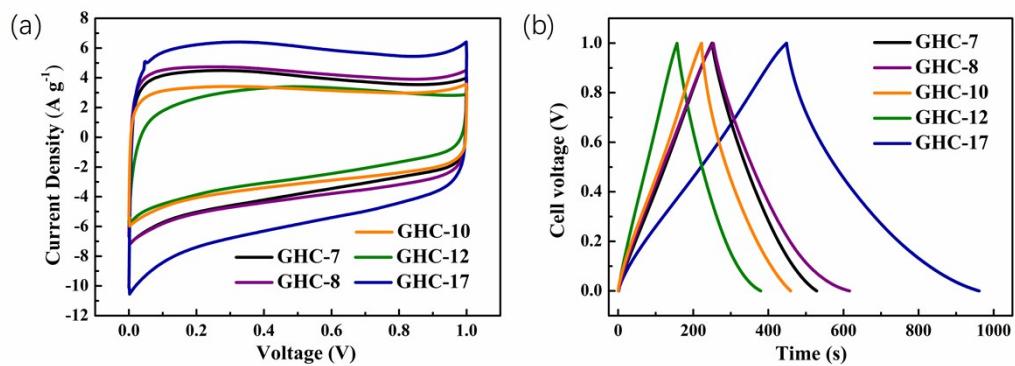
**Figure S9** Nyquist plots of GHC-7, GHC-8, GHC-10, GHC-12, and GHC-17.



**Figure S10** The electrical conductivity of the GHC based electrodes measured by four-point method (25 °C)



**Figure S11** Electrochemical performance of the GHC-17 sample measured in a three-electrode system using  $6 \text{ mol} \cdot \text{L}^{-1}$  KOH electrolyte. (a) CV curves of the GHC-17 at different scan rates from 5 to  $100 \text{ mV} \cdot \text{s}^{-1}$ . (b) Charge-discharge curves of the GHC-17 at different current densities from 1 to  $50 \text{ A} \cdot \text{g}^{-1}$



**Figure S12** Electrochemical performance of the GHC-7, GHC-8, GHC-10, GHC-12, GHC-17 samples measured in a three-electrode system using  $6 \text{ mol L}^{-1}$  KOH as the electrolyte. (a) CV curves at  $20 \text{ mV} \cdot \text{s}^{-1}$ ; (b) The galvanostatic charge-discharge profiles at  $1 \text{ A} \cdot \text{g}^{-1}$

**Table S1** Factors and levels of the orthogonal design

Levels	Factors				
	Impregnatio n ratio	Activation temperature	Activation time	Carbonization temperature	Carbonization time
	A	B (°C)	C (min)	D (°C)	E (min)
1	3.0:1	700	60	550	120
2	3.5:1	750	90	600	180
3	4.0:1	800	120	650	240
4	4.5:1	850	150	700	300

**Table S2** Orthogonal experimental arrangements

Experiment No.	Range and levels				
	Impregnation ratio	Activation temperature B	Activation time	Carbonization temperature	Carbonization time
	A	(°C)	C (min)	D (°C)	E (min)
1	3.0:1	700	60	550	120
2	3.0:1	750	90	600	180
3	3.0:1	800	120	650	240
4	3.0:1	850	150	700	300
5	3.5:1	700	90	650	300
6	3.5:1	750	60	700	240
7	3.5:1	800	150	550	180
8	3.5:1	850	120	600	120
9	4.0:1	700	150	700	180
10	4.0:1	750	60	650	120
11	4.0:1	800	90	600	300
12	4.0:1	850	90	550	240
13	4.5:1	700	150	600	240
14	4.5:1	750	120	550	300
15	4.5:1	800	90	700	120
16	4.5:1	850	60	600	180

**Table S3** Specific surface area and pore volume at different fabrication conditions

Experiment No.	Variables					Responses	
	A	B	C	D	E	Specific surface area (m <sup>2</sup> /g)	Pore volume (cm <sup>3</sup> /g)
1	1	1	1	1	1	1468.33	0.759
2	1	2	2	2	2	1701.82	0.896
3	1	3	3	3	3	1714.46	0.922
4	1	4	4	4	4	1568.98	0.852
5	2	1	2	3	4	1725.09	0.938
6	2	2	1	4	3	1442.16	0.795
7	2	3	4	1	2	2268.66	1.162
8	2	4	3	2	1	2542.77	1.378
9	3	1	3	4	2	1561.91	0.838
10	3	2	4	3	1	2259.80	1.218
11	3	3	1	2	4	2077.21	1.165
12	3	4	2	1	3	2254.06	1.336
13	4	1	4	2	3	1886.07	1.063
14	4	2	3	1	4	1507.58	0.865
15	4	3	2	4	1	1161.90	0.664
16	4	4	1	3	2	1470.07	0.824
17	3	3	4	2	1	2818.22	1.327

**Table S4** Range analysis results

Indicators	Levels	Factors				
		A	B	C	D	E
<b>Specific surface area <math>S_{BET}</math></b>	$K_{1j}$	1613.40	1660.35	1614.44	1874.66	1858.20
	$K_{2j}$	1994.67	1727.84	1710.72	2051.97	1750.62
	$K_{3j}$	2038.25	2194.99	1831.68	1792.36	1824.19
	$K_{4j}$	1506.41	1958.97	1995.88	1433.74	1719.72
	$R_j$	531.84	534.64	381.44	618.23	138.49
<b>The optimal solution</b>		A <sub>3</sub>	B <sub>3</sub>	C <sub>4</sub>	D <sub>2</sub>	E <sub>1</sub>
<b>Pore volume</b>	Levels	Factors				
		A	B	C	D	E
	$K_{1j}$	0.86	0.90	0.89	1.03	1.03
	$K_{2j}$	1.07	0.94	0.96	1.13	0.93
	$K_{3j}$	1.14	1.20	1.00	0.98	1.00
	$K_{4j}$	0.85	1.10	1.07	0.79	0.96
	$R_j$	0.29	0.30	0.19	0.34	0.10
<b>The optimal solution</b>		A <sub>3</sub>	B <sub>3</sub>	C <sub>4</sub>	D <sub>2</sub>	E <sub>1</sub>

**Table S5** Characteristics of garlic skin-derived activated carbons without KOH activation  
 (Samples are prepared with carbonization time for 2h)

Experiment No.	Carbonization temperature D (°C)	Specific surface area (m <sup>2</sup> /g)	Pore volume (cm <sup>3</sup> /g)	Average pore diameter (nm)
1-1	550	81.44	0.079	3.812
1-2	600	178.16	0.143	3.315
1-3	650	266.48	0.216	2.947
1-4	700	420.61	0.251	2.759

**Table S6** Peak assignment of C 1s and O 1s of the GHC-17

Peak	Binding energy (eV)	Assignment	Fraction of species	Atom (%)
<b>C 1s</b>	284.7	Sp <sup>3</sup> C=C	32.7	87.08
	285.6	Sp <sup>2</sup> C-C	23.2	
	286.5	C-O	21.4	
	288.3	C-OOH	22.7	
<b>O 1s</b>	531.5	C=O Carbonyl O in COOR	25.0	12.92
	532.4	C-OH, C-O-C	25.3	
	533.8	Non-carbonyl O in COOR	25.2	
	535.5	H <sub>2</sub> O <sub>(ads)</sub> /O <sub>2(ads)</sub>	24.6	

**Table S7** Pore parameters of GHC-7, GHC-8, GHC-10, GHC-12, GHC-17 ( $S_{BET}$ : total surface area;  $S_{micro}$ : micropore specific surface area;  $S_{meso}$ : mesoporous specific surface area;  $V_{tot}$ : total pore volume;  $V_{micro}$ : micropore volume;  $V_{meso}$ : mesoporous volume;  $d_a$ : average pore diameter)

Sample	$S_{BET}$ ( $m^2$ $g^{-1}$ )	$S_{micro}$ ( $m^2$ $g^{-1}$ )	$S_{meso}$ ( $m^2$ $g^{-1}$ )	$V_{tot}$ ( $cm^3$ $g^{-1}$ )	$V_{micro}$ ( $cm^3$ $g^{-1}$ )	$V_{meso}$ ( $cm^3$ $g^{-1}$ )	$d_a$ (nm)	Capacitance (F g <sup>-1</sup> )
7	2268	1606	662	1.162	0.687	0.475	2.049	245
10	2260	1311	949	1.218	0.535	0.683	2.008	207
12	2254	851	1403	1.336	0.366	0.970	2.355	173
8	2543	1721	822	1.378	0.808	0.570	2.055	268
17	2818	2422	396	1.327	1.022	0.305	1.883	427

**Table S8** The electrical conductivity of the GHC based electrodes measured by four-point method (25 °C)

	Electrical conductivity (S cm <sup>-1</sup> )	
	2 MPa	28 MPa
<b>GHC-7</b>	1.19	9.74
<b>GHC-8</b>	1.9	15.17
<b>GHC-10</b>	1.74	12.22
<b>GHC-12</b>	2.08	16.18
<b>GHC-17</b>	3.25	22.05

**Table S9** Summary of capacitive performance of carbon-based supercapacitor electrodes

Carbon material	$S_{BET}$ ( $m^2g^{-1}$ )	Gravimetric capacitance (F g <sup>-1</sup> )	Volumetric capacitance (F cm <sup>-3</sup> )	Electrolyte & Test set-up	Ref.

<b>GHC</b>	2818	427 (0.5 A g <sup>-1</sup> )	162 (0.5 A g <sup>-1</sup> )	6 M KOH Two electrode	This work
<b>Reed straw derived carbon</b>	2387	355 (1 A g <sup>-1</sup> )	--	6 M KOH Two electrode	1
<b>N-Doped carbon nanocages</b>	1794	313 (1 A g <sup>-1</sup> )	--	1 M KOH Two electrode	2
<b>Activated carbon fiber from sawdust</b>	2395	242 (0.5 A g <sup>-1</sup> )	--	6 M KOH Two electrode	3
<b>Seaweed derived carbon</b>	3270	425 (0.1 A g <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	4
<b>Hierarchical porous graphene</b>	1533	280 (1 A g <sup>-1</sup> )	224 (1 A g <sup>-1</sup> )	6 M KOH Two electrode	5
<b>Polypyrrole-Derived activated carbon</b>	3432	300 (1 mV s <sup>-1</sup> )	--	EMMBF <sub>4</sub> Two electrode	6
<b>Holey graphene frameworks</b>	810	310 (1 A g <sup>-1</sup> )	220 (1 A g <sup>-1</sup> )	6 M KOH Two electrode	7
<b>Graphene</b>	--	200 (1 A g <sup>-1</sup> )	255 (0.1 A g <sup>-1</sup> )	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	8
<b>Dead leaves derived carbons</b>	1230	400 (0.5 A g <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	9
<b>Reduced graphene oxide</b>	2400	310 (20 A g <sup>-1</sup> )	--	1 M KOH Two electrode	10
<b>Graphene-based porous frameworks</b>	350	226 (1 mV s <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	11
<b>Eggshell membranes derived carbon</b>	221	297 (0.2 A g <sup>-1</sup> )	--	1 M KOH Two electrode	12
<b>3D microporous graphene frameworks</b>	194	202 (1 A g <sup>-1</sup> )	--	1 M Na <sub>2</sub> SO <sub>4</sub> Two electrode	13
<b>Natural organic chemicals derived carbons</b>	2967	230 (0.1 A g <sup>-1</sup> )	100 (0.1 A g <sup>-1</sup> )	1 M TEAFB <sub>4</sub> /AN Two electrode	14
<b>Activated graphene</b>	3100	165 (1.4 A g <sup>-1</sup> )	60 (1.4 A g <sup>-1</sup> )	1 M	15

<b>3D-printed graphene aerogel lattice</b>	418	64 (10 A g <sup>-1</sup> )	2 (10 A g <sup>-1</sup> )	BMIMBF <sub>4</sub> /AN Two electrode	
<b>Carbon cryogel</b>	1025	161 (10 A g <sup>-1</sup> )	80 (10 A g <sup>-1</sup> )	3 M KOH Two electrode	16
<b>Nanoporous carbon</b>	1472	220 (0.5 A g <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	17
<b>KOH activated C<sub>70</sub> microstructure</b>	1249	192 (5 A g <sup>-1</sup> )	--	6 M H <sub>2</sub> SO <sub>4</sub> Two electrode	18
<b>Superactivated carbon</b>	1600	246 (0.2 A g <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Two electrode	19
<b>Graphene/carbon</b>	492	250 (5 A g <sup>-1</sup> )	--	1 M Na <sub>2</sub> SO <sub>4</sub> Two electrode	20
<b>Hierarchically porous carbons</b>	2988	238 (0.2 A g <sup>-1</sup> )	--	6 M KOH <sub>4</sub> Two electrode	21
<b>Fungi derived porous carbon</b>	80	196 (5 mV s <sup>-1</sup> ))	180 (5 mV s <sup>-1</sup> ))	6 M KOH <sub>4</sub> Two electrode	22
<b>Ordered mesoporous carbide-derived carbon</b>	2250	170 (0.1 A g <sup>-1</sup> )	178 (0.1 A g <sup>-1</sup> )	1 M TEAFB <sub>4</sub> /AN Two electrode	23
<b>Dry elm samara</b>	1947	470 (1 A g <sup>-1</sup> )	--	6 M KOH Three electrode	24
<b>Protein</b>	805.7	390.4 (0.2 A g <sup>-1</sup> )	--	1 M H <sub>2</sub> SO <sub>4</sub> Three electrode	25
<b>Human hair</b>	1306	340 (1 A g <sup>-1</sup> )	--	6 M KOH Three electrode	26
<b>Fungus</b>	1103	373 (0.5 A g <sup>-1</sup> )	--	6 M KOH Three electrode	27
<b>Shiitake mushroom</b>	2988	306 (1 A g <sup>-1</sup> )	--	6 M KOH Three electrode	28

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