

Supporting Information

Preparation of BiVO₄ nanoporous photoanode based on peroxovanadate reduction and conversion for efficient photoelectrochemical performance

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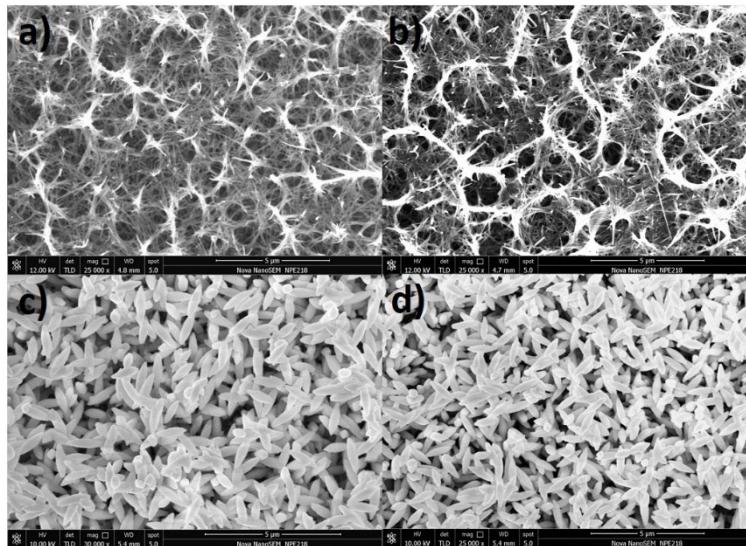


Fig. S1 SEM images of V₂O₅•xH₂O films deposited for 2h (a) and 4h (b), and the corresponding converted BiVO₄ films prepared from V₂O₅•xH₂O film after annealing treatment: (c) 2h-BiVO₄, (d) 4h-BiVO₄.

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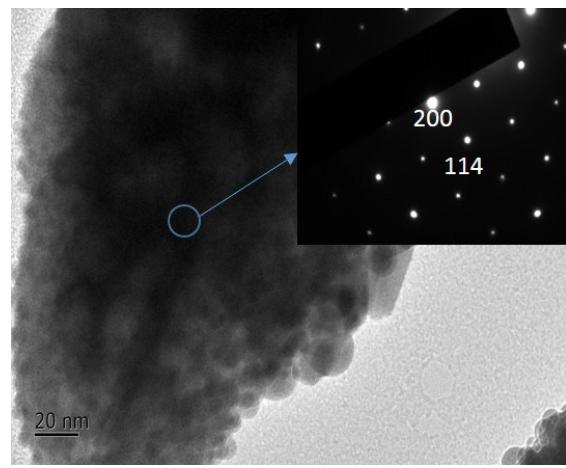


Fig. S2 TEM image of the shuttle like BiVO₄ rod and the SAED pattern.

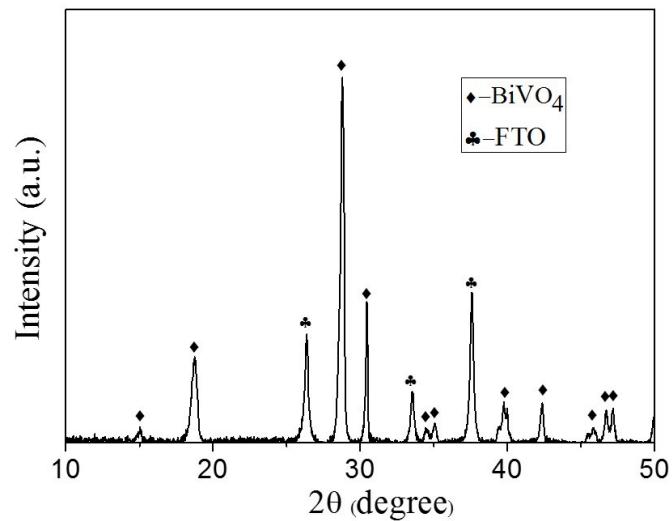


Fig.S3 XRD pattern of BiVO₄ film prepared from V₂O₅ film deposited for 2 h.

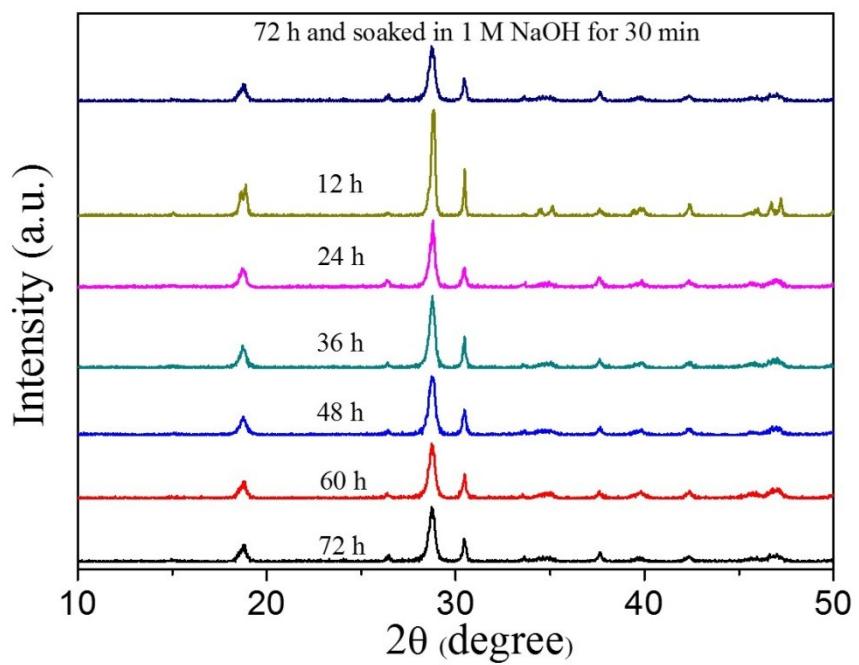


Fig. S4 XRD patterns of samples formed by the annealed $\text{V}_2\text{O}_5 \cdot x\text{H}_2\text{O}$ film with varied impregnation time by Bi^{3+} .

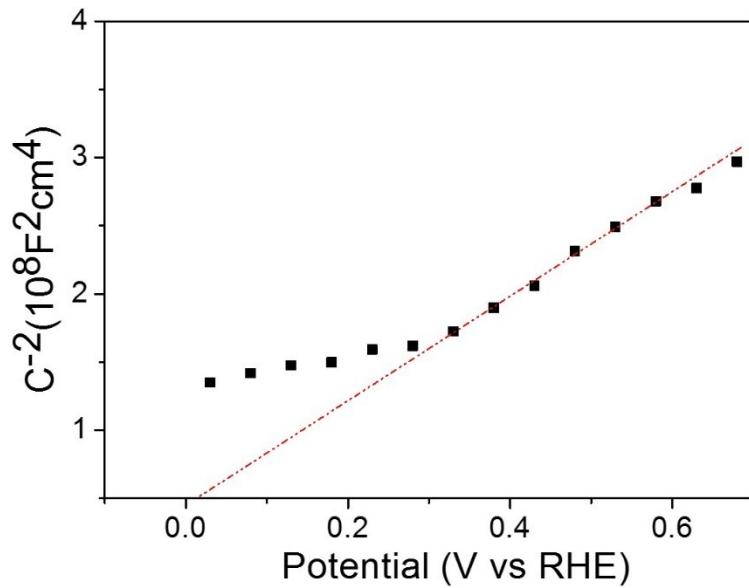


Fig. S5 Mott–Schottky plot for BiVO_4 photoanode prepared in this work.

Table S1 Comparison of PEC properties with BiVO₄ photoanodes reported in the literatures

Bare BiVO ₄	Electrolyte	Photocurrent densiy(mAcm ⁻²)@1.23V vs RHE	IPCE@400nm and 1.23V vs RHE	Reference
Na ₂ SO ₄ (pH 6.6)		~1.5	~20%	[1]
natural seawater		<0.5	~10%	[2]
NaHCO ₃ (pH 8.5)		0.5-1.0	<20%	[3]
Na ₂ SO ₄ (pH 7)		~0.4	<20%(Mo doped)	[4]
Na ₂ SO ₄ (pH 7)		<1(@1.41 V)	-	[5]
Na ₂ SO ₄ (pH 7)		<1	-	[6]
KH ₂ PO ₄ (pH 7)		<0.5	<10%	[7]
Na ₂ SO ₄ and PBS (pH 6.6)		0.5-1.0	20-25%(@1.59 V)	[8]
Na ₂ SO ₄ (pH 6.8)		0.2(@1.0 V)	<10%	[9]
PBS (pH=7)		<0.5	<20%	[10]
PBS (pH=7)		0.5-1.0	<20%	[11]
PBS (pH=7)		<1	-	[12]
PBS (pH=7)		~1.5	-	[13]
PBS (pH=7)		1.10	22.4%	This paper

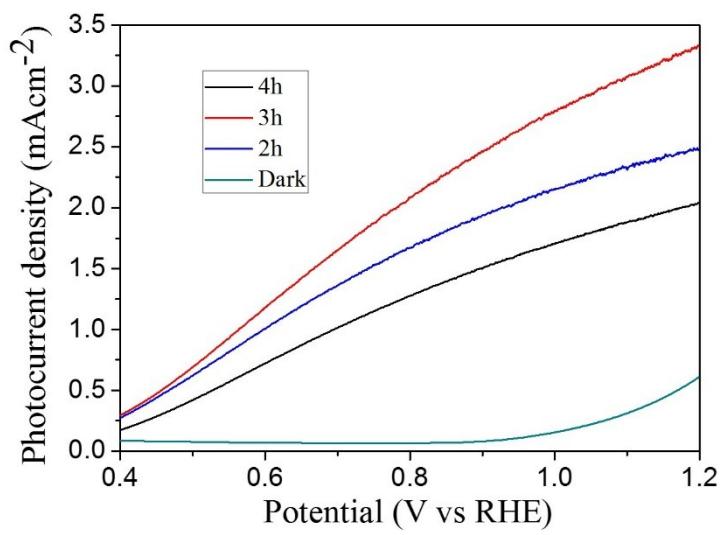


Fig. S6 I-V curves of BiVO_4 electrodes measured in a 0.1 M phosphate buffer (pH 7) containing 0.1 M Na_2SO_3 as hole scavenger under AM 1.5G, 100 mW/cm^2 illumination (scan rate, 10 mV/s).

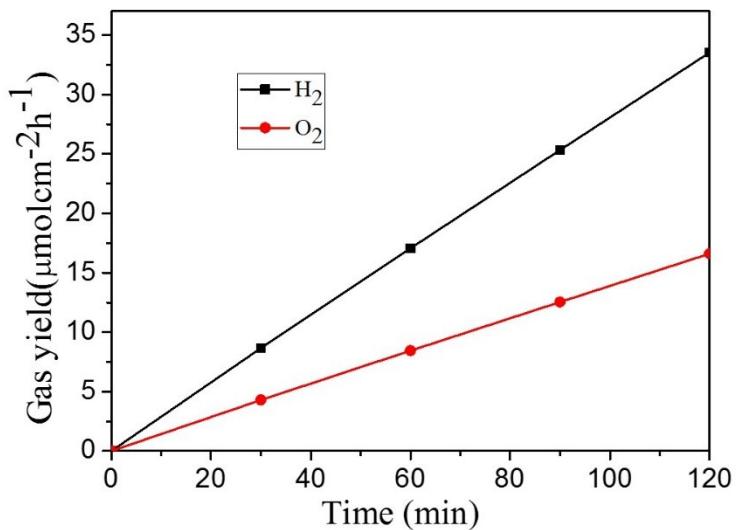


Fig. S7 Gas yield during water oxidation under AM 1.5G illumination at 1.23 V vs RHE.

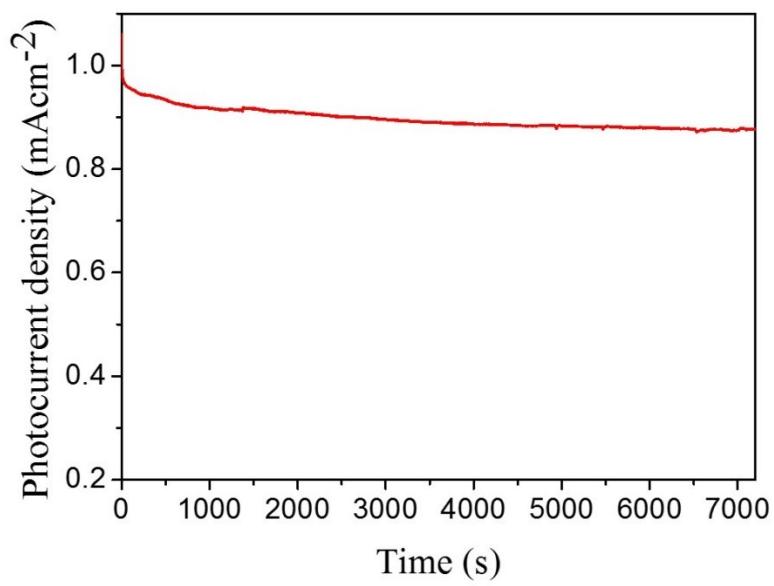


Fig. S8 I-t curves of BiVO_4 photoanode at 1.23 V vs RHE under AM 1.5G illumination.

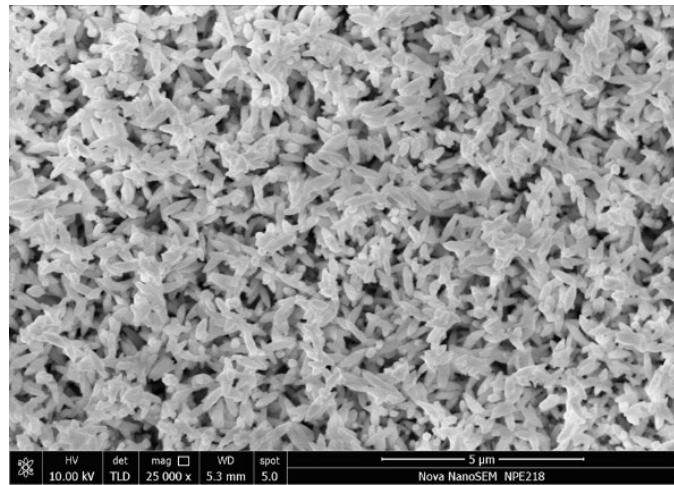


Fig. S9 SEM image of BiVO_4 film after 2 h stability test.

Supplementary references

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