

Electronic Supplementary Material (ESI)

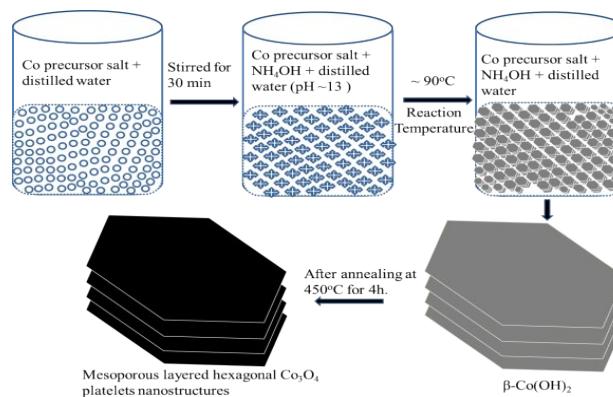
Mesoporous Layered Hexagonal Platelets of Co_3O_4 nanoparticle with (111) facets for battery: High Performance and Ultra-high Rate Capability

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Scheme 1 Growth mechanism of mesoporous layered hexagonal platelets of Co_3O_4 structures.

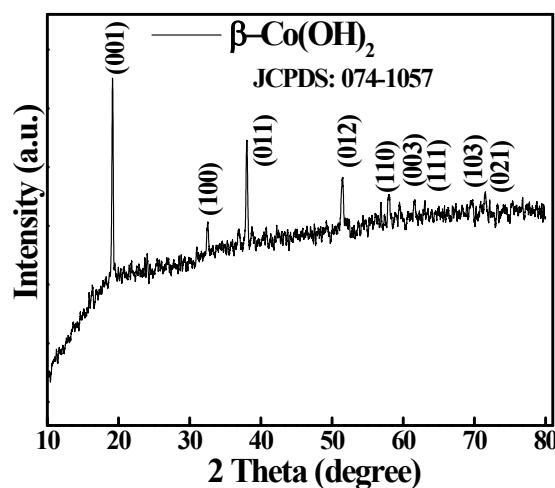


Fig. S1 shows the XRD pattern of as-prepared ($\beta\text{-Co(OH)}_2$) sample.

The XRD pattern shown in figure S1 matches well with the JCPDS file 074-1057 belongs to hexagonal crystal system, space group $P\bar{3}m1$ (164) confirms the formation of $\beta\text{-Co(OH)}_2$. The XRD pattern shows the presence of diffraction peaks at 2θ of 19.14° , 32.49° , 38.02° , 51.46° , 57.92° and 59.54° , etc. which are assigned to (001), (100), (011), (012), (110) and (003) diffraction peaks of hexagonal $\beta\text{-Co(OH)}_2$ phase and matches well with the above JCPDS file.

Table S1: Specific capacity as a function of current densities.

Current density (mAg^{-1})	Capacity ($\text{mA}\text{h}\text{g}^{-1}$)
434.78	137.58
869.57	135.59
1304.35	132.32
2608.7	124.00
3478.26	123.18
4347.83	119.66
5217.39	119.53
6956.52	117.10
8695.65	114.92
10434.78	111.62
12173.91	108.60
14782.61	106.79
16521.74	105.34
19130.43	103.82
20869.57	101.72
26086.96	98.06
30434.78	95.83
34782.61	91.71
39130.43	89.25
43478.26	84.59

Table S2. The comparison of specific capacity obtained from various synthesis methodologies and their morphology of Co_3O_4 electrodes.

Morphology	Synthesis method	Specific capacity ($\text{mA}\text{h}\text{g}^{-1}$)	Current density (mAg^{-1})	Electrolyte used	Capacity retention, Cycles@ mAg^{-1}	Ref
Mesoporous microdisks like Co_3O_4	Solvo-thermal	1032	100	1 M LiPF_6 solution in (EC/DMC/DEC) (1:1:1)	74.12%, 30 Cycles@100	1
Mesoporous Co_3O_4	Two Solvent method	1489	50	1 M LiPF_6 solution in (EC/DMC) (1:1)	76%, 25 cycles@ 50	2
Co_3O_4 /porous electrospun carbon nanofibers	Electro-spinning	952	100	1 M LiPF_6 solution in (EC/DMC) (1:1)	96.95%, 100 cycles@100	3
Self-assembled hairy ball like Co_3O_4 nanostructure	Hydrothermal	1768	100	1 M LiPF_6 solution in (EC/DMC) (1:1)	48%, 50 cycles@100	4
Mesoporous Co_3O_4 nanoflakes	Microwave assisted Hydrothermal	1192	89	1 M LiPF_6 solution in (EC/DMC) (1:1)	74%, 300 cycles@445	5
Co_3O_4 hollow-structured nanoparticles	Impregnation reduction	1107	50	1 M LiPF_6 solution in (EC/DMC) (1:1)	79%, 50 cycles@50	6
Co_3O_4 nanocages	Hydrothermal	1116	50	1 M LiClO_4 solution in (EC/DMC/DEC) (1:1:1)	77%, 50 cycles@178	7
Nanobowl array and Nanotubes of Co_3O_4	Thermal Decomposition and	1293 1250	35	1 M LiPF_6 solution in (EC/DMC/DEC) (1:1:1)	57% and 69% after 10 cycles@35	8
Co_3O_4 nanoparticles with opened-book morphology	Solvothermal	1408	100	1 M LiPF_6 solution in (EC/DMC/DEC) (1:1:1)	67%, 25 cycles@100	9
Bare/Intrinsic Mesoporous Co_3O_4 layered hexagonal platelets	Chemical bath	137	434.78	2 M KOH electrolyte	81.25%@ ~12170, 2020 cycles	This work

Table S3. Energy density and power density comparison with previous reports.

Sample name	Energy density	Power density	Ref.
	(Wh kg ⁻¹)	(kW kg ⁻¹)	
Co ₃ O ₄ hexagonal platelets (powder)	42.56	1.56	10
Co ₃ O ₄ nanowire arrays (Ni-foam)	25.5	11	11
3D nanoporous Co ₃ O ₄ /Carbon (powder)	20.44	16	12
1D Co-Ni/ Co ₃ O ₄ -NiO core/shell	23	5.5	13
Mesoporous layered hexagonal platelets of Co₃O₄	32.03	9.33	(This work)

References

1. Y. Jin, L. Wang, Y. Shang, J. Gao, J. Li and X. He, *Electrochim. Acta*, 2015, **151**, 109–117.
2. S. Sun, X. Zhao, M. Yang, L. Wu, Z. Wen and X. Shen, *Sci. Rep.*, 2016, **6**, 19564.
3. S. Abouali, M. Akbari Garakani, B. Zhang, H. Luo, Z. Xu, J.-Q. Huang, J. Huang and J.-K. Kim, *J. Mater. Chem. A*, 2014, **2**, 16939–16944.
4. D. Fang, L. Li, W. Xu, G. Li, Z. Luo, C. Liang, Y. Ji, J. Xu and C. Xiong, *J. Mater. Chem. A*, 2013, **1**, 13203–13208.
5. S. Chen, Y. Zhao, B. Sun, Z. Ao, X. Xie, Y. Wei and G. Wang, *ACS Appl. Mater. Interfaces*, 2015, **7**, 3306–3313.
6. D. Wang, Y. Yu, H. He, J. Wang, W. Zhou and H. D. Abruña, *ACS Nano*, 2015, **9**, 1775–1781.
7. D. Liu, X. Wang, X. Wang, W. Tian, Y. Bando and D. Golberg, *Sci. Rep.*, 2013, **3**, 2543.
8. G. Tong, Y. Liu and J. Guan, *J. Alloys Compd.*, 2014, **601**, 167–174.
9. B. Yan, L. Chen, Y. Liu, G. Zhu, C. Wang, H. Zhang, G. Yang, H. Ye and A. Yuan, *CrystEngComm*, 2014, **16**, 10227–10234.
10. K. Deori, S. K. Ujjain, R. K. Sharma and S. Deka, *ACS Appl. Mater. Interfaces*, 2013, **5**, 10665.
11. X. Xia, J. Tu, Y. Zhang, Y. Mai, X. Wang, C. Gu and X. Zhao, *RSC Adv.*, 2012, **2**, 1835–1841.
12. N. Wang, Q. Liu, D. Kang, J. Gu, W. Zhang and D. Zhang, *ACS Appl. Mater. Interfaces*, 2016, **8**, 16035–16044.
13. A. K. Singh, D. Sarkar, G. Gopal Khan and K. Mandal, *Appl. Phys. Lett.*, 2014, **104**, 133904.