

## Nanocrystal Heterostructures of LiCoO<sub>2</sub> with Conformal Passivating Shells

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## Supplementary information

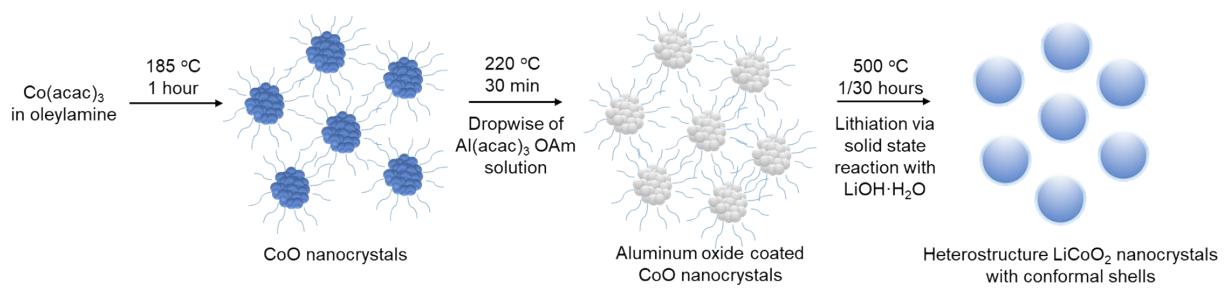


Figure S1. A scheme for the preparation of heterostructure  $\text{LiCoO}_2$  nanocrystals with conformal shells.

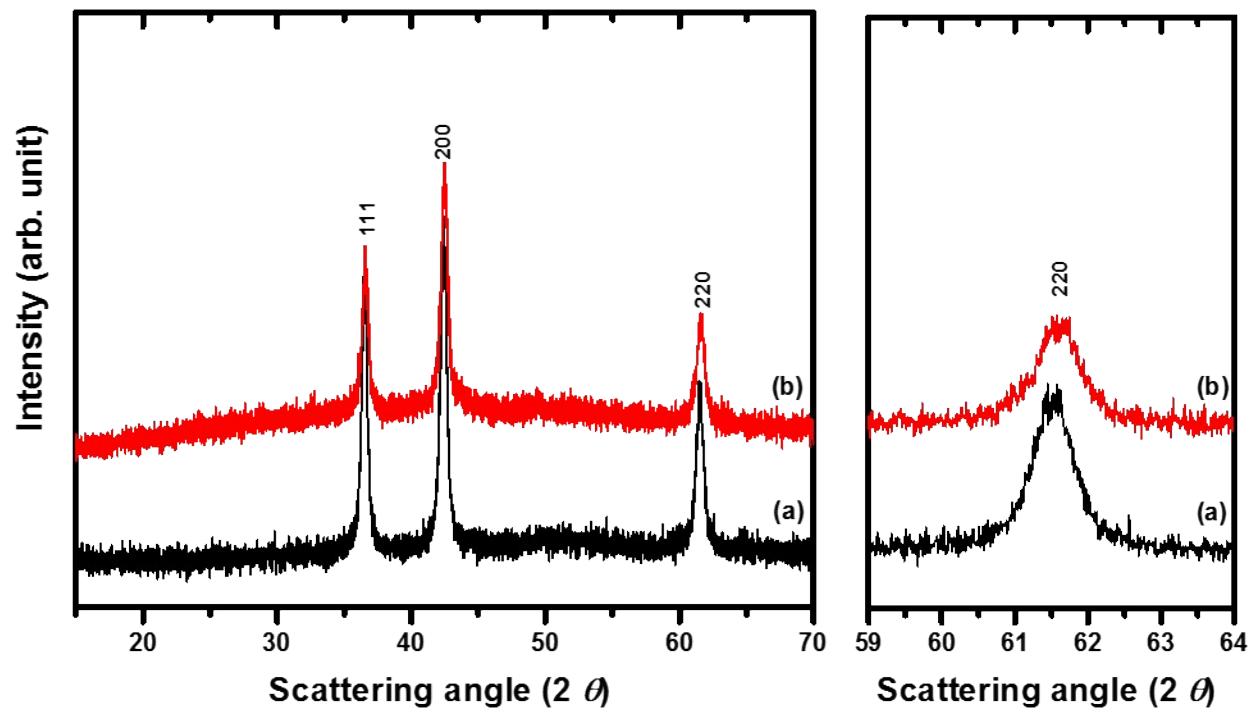


Figure S2. X-ray diffraction patterns of (a) CoO and (b) core-shell CoO nanocrystals.

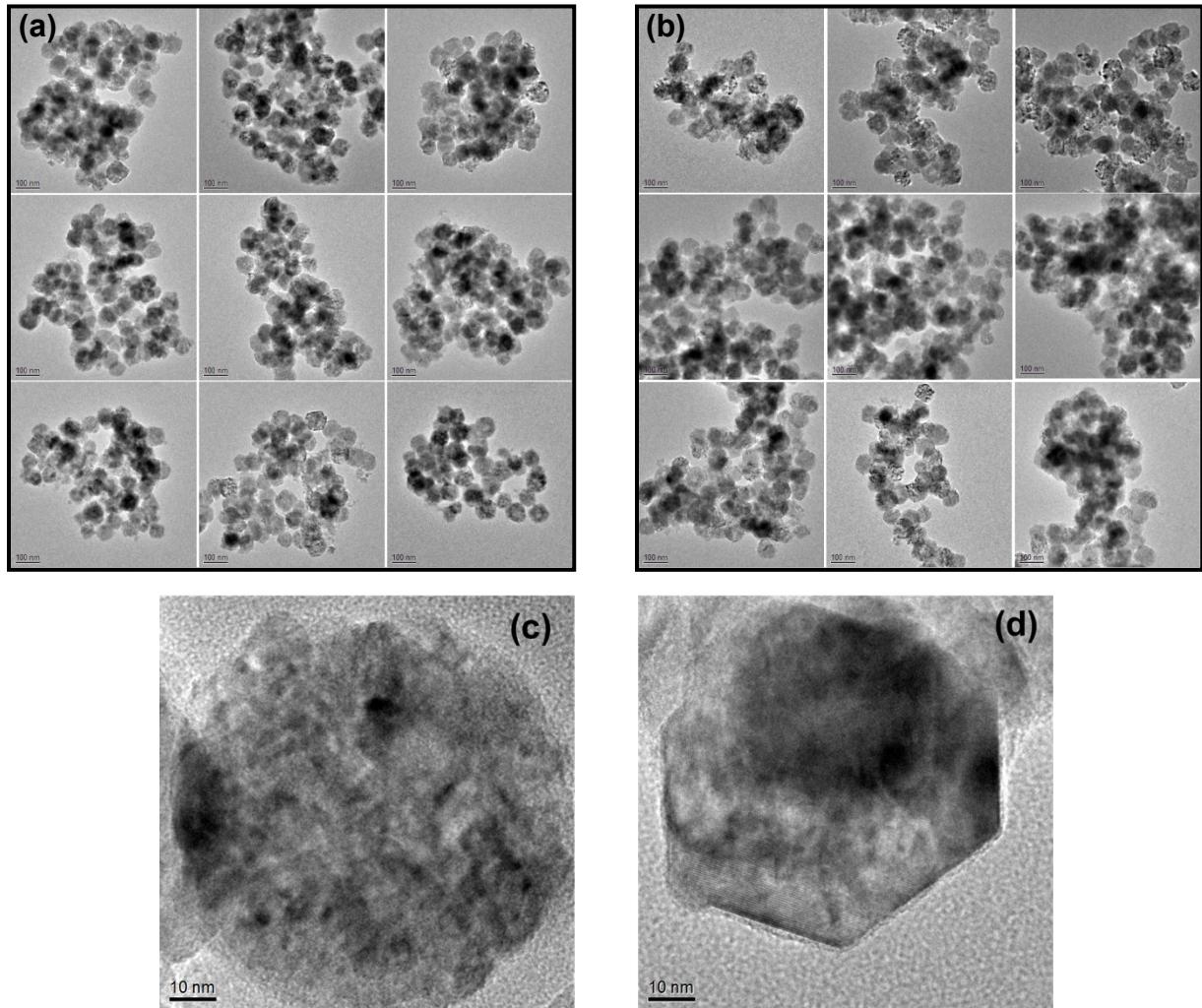


Figure S3. TEM images of (a) LiCoO<sub>2</sub>-1h and (b) LiCoO<sub>2</sub>-30h nanocrystals used to measure the populations with spherical or platelet morphology. Representative TEM images of spherical (c) and faceted (plate-like) (d) morphologies in the LiCoO<sub>2</sub> nanocrystals.

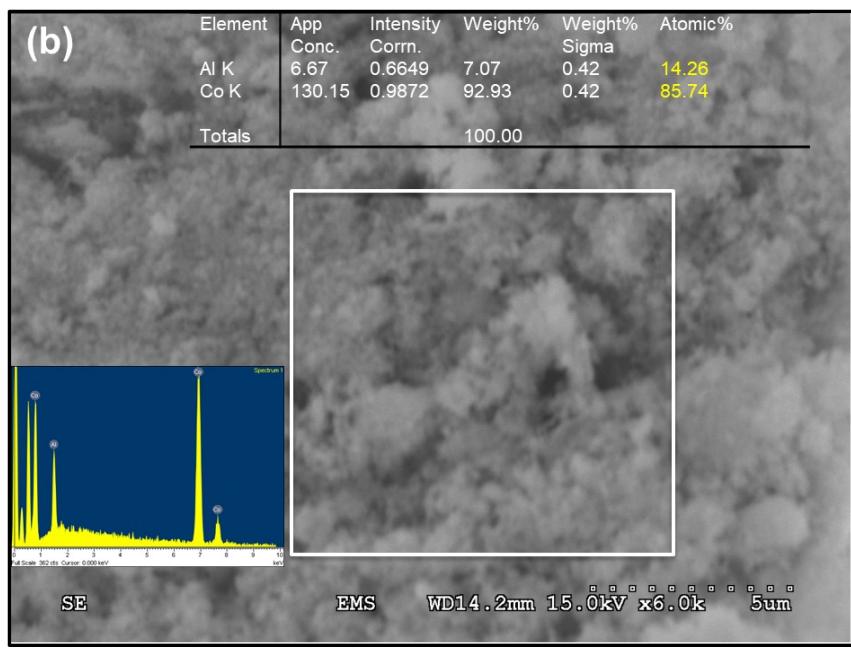
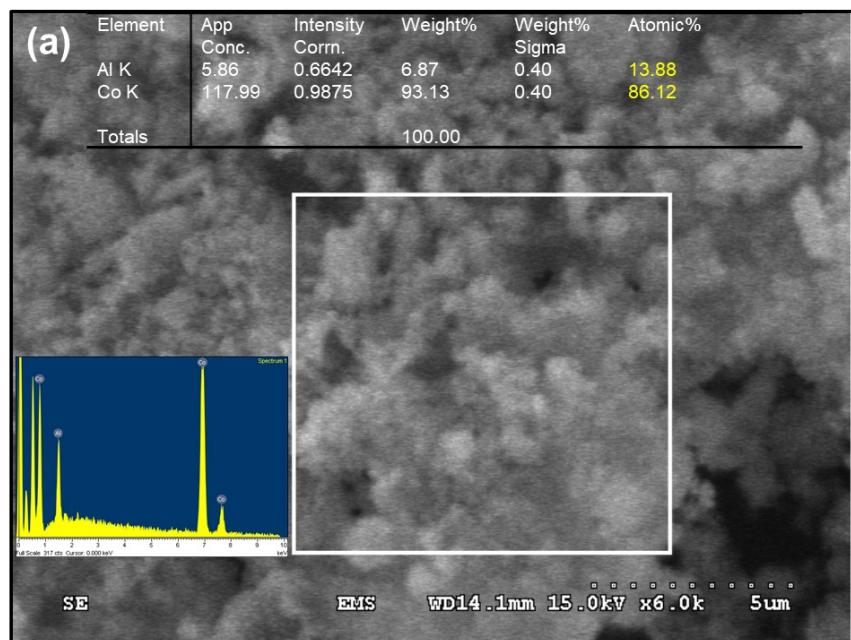


Figure S4. SEM image and corresponding EDX spectrum of (a) CS-LiCoO<sub>2</sub>-1h and (b) CS-LiCoO<sub>2</sub>-30h nanocrystals.

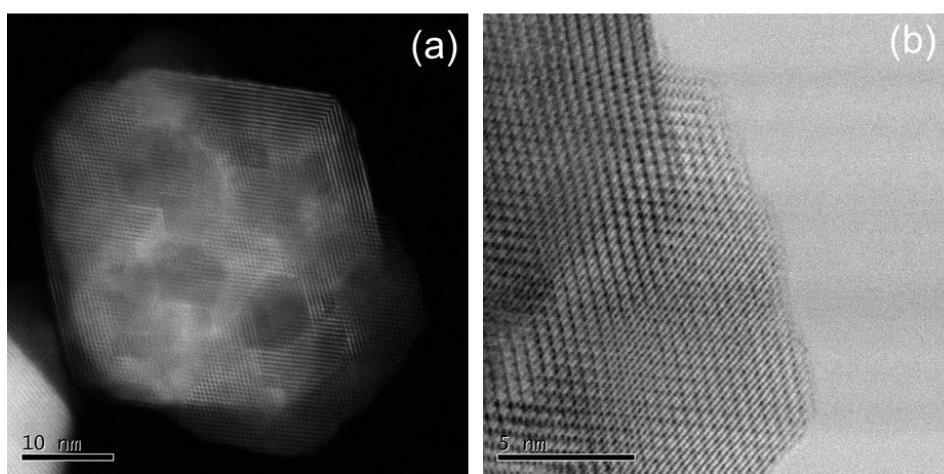


Figure S5. (a) LAADF and (b) ABF images of CS-LiCoO<sub>2</sub>-30h nanocrystals

	Charge capacity above 3.9 V (mAh/g)	Charge capacity below 3.9 V (mAh/g)	% capacity of LT
LCO 1h	<b><math>84.1 \pm 0.3</math></b>	<b><math>59.6 \pm 2.1</math></b>	<b><math>41.5 \pm 0.8</math></b>
CS-LCO 1h	<b><math>70.6 \pm 1.4</math></b>	<b><math>41.4 \pm 2.6</math></b>	<b><math>36.9 \pm 1.7</math></b>
LCO 30h	<b><math>93.6 \pm 1.8</math></b>	<b><math>58.6 \pm 3.7</math></b>	<b><math>38.5 \pm 1.6</math></b>
CS-LCO 30h	<b><math>73.4 \pm 0.4</math></b>	<b><math>49.3 \pm 0.8</math></b>	<b><math>40.2 \pm 0.3</math></b>

Table S1. Specific capacities below and above 3.9 V during the first charge. These values are considered qualitatively representative of the ratio of HT- and LT-phase in the nanocrystals. Percent capacities of LT-phase were calculated as  $(LT * 100 / (HT + LT))$ . All values were taken by averaging three cells.

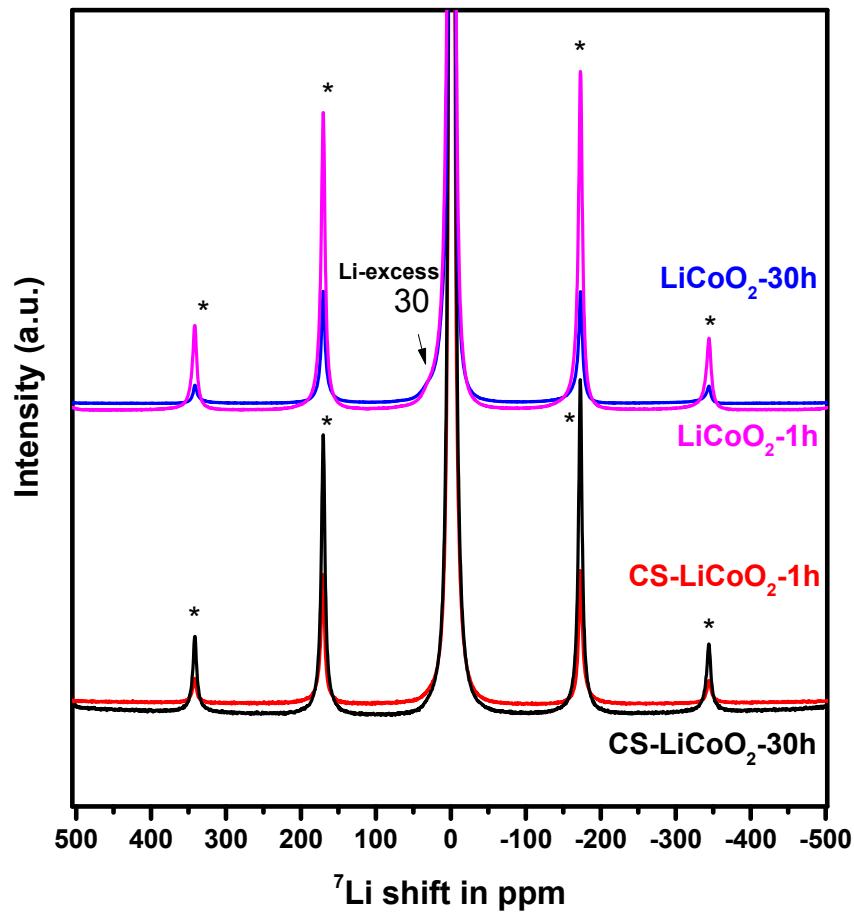


Figure S6. Alternate view of the  ${}^7\text{Li}$  MAS NMR spectra of bare and core shell  $\text{LiCoO}_2$ , presented in the manuscript.

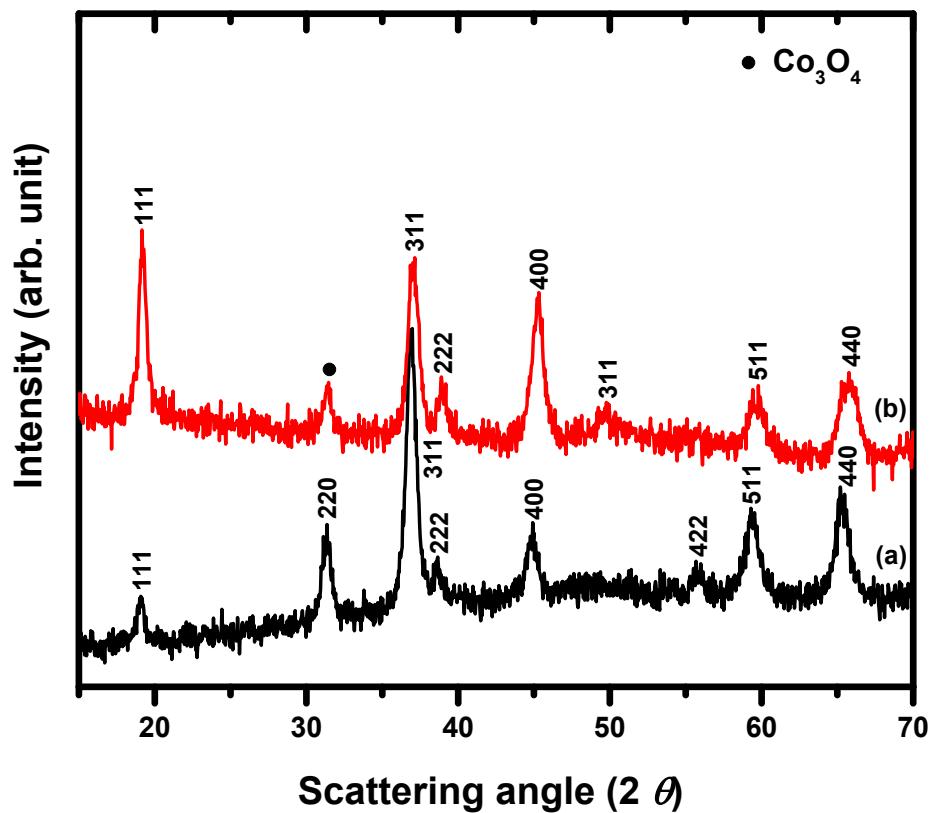


Figure S7. XRD patterns of (a)  $\text{Co}_3\text{O}_4$  prepared by calcination of  $\text{CoO}$  nanocrystals at 300 °C for 4 hours and (b) spinel  $\text{LiCoO}_2$  nanocrystals.

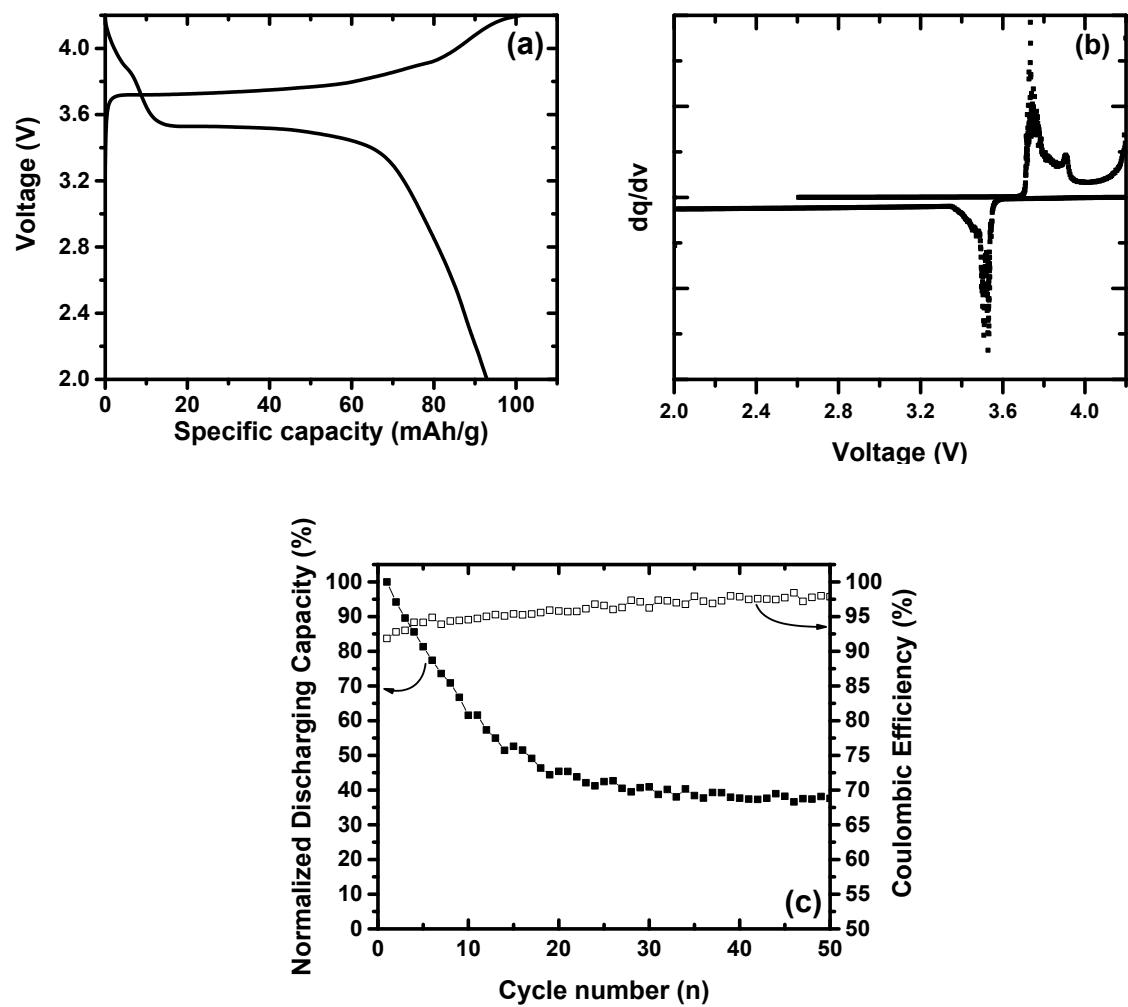


Figure S8. (a) Voltage versus specific capacity profiles, (b) corresponding  $dQ/dV$  plots of spinel  $\text{LiCoO}_2$  nanocrystals (Figure S5) and (c) evolution of normalized specific capacity (solid symbol) and Coulombic efficiency (open symbol) when cycling with a discharge cutoff voltage of 2.0 V and a charge cutoff voltage of 4.2 V vs  $\text{Li}^+/\text{Li}^0$  at C/20 rate.

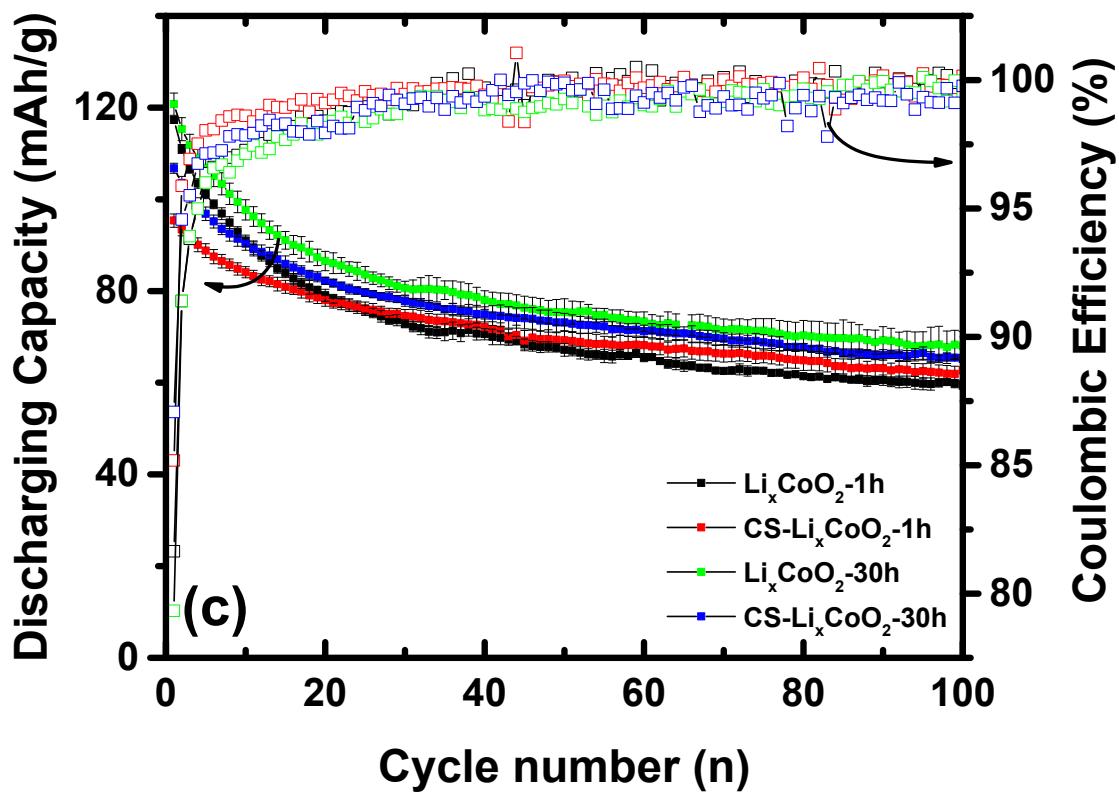


Figure S9. Evolution of discharging specific capacity (solid symbol) and Coulombic efficiency (open symbol) when cycling at C/20.

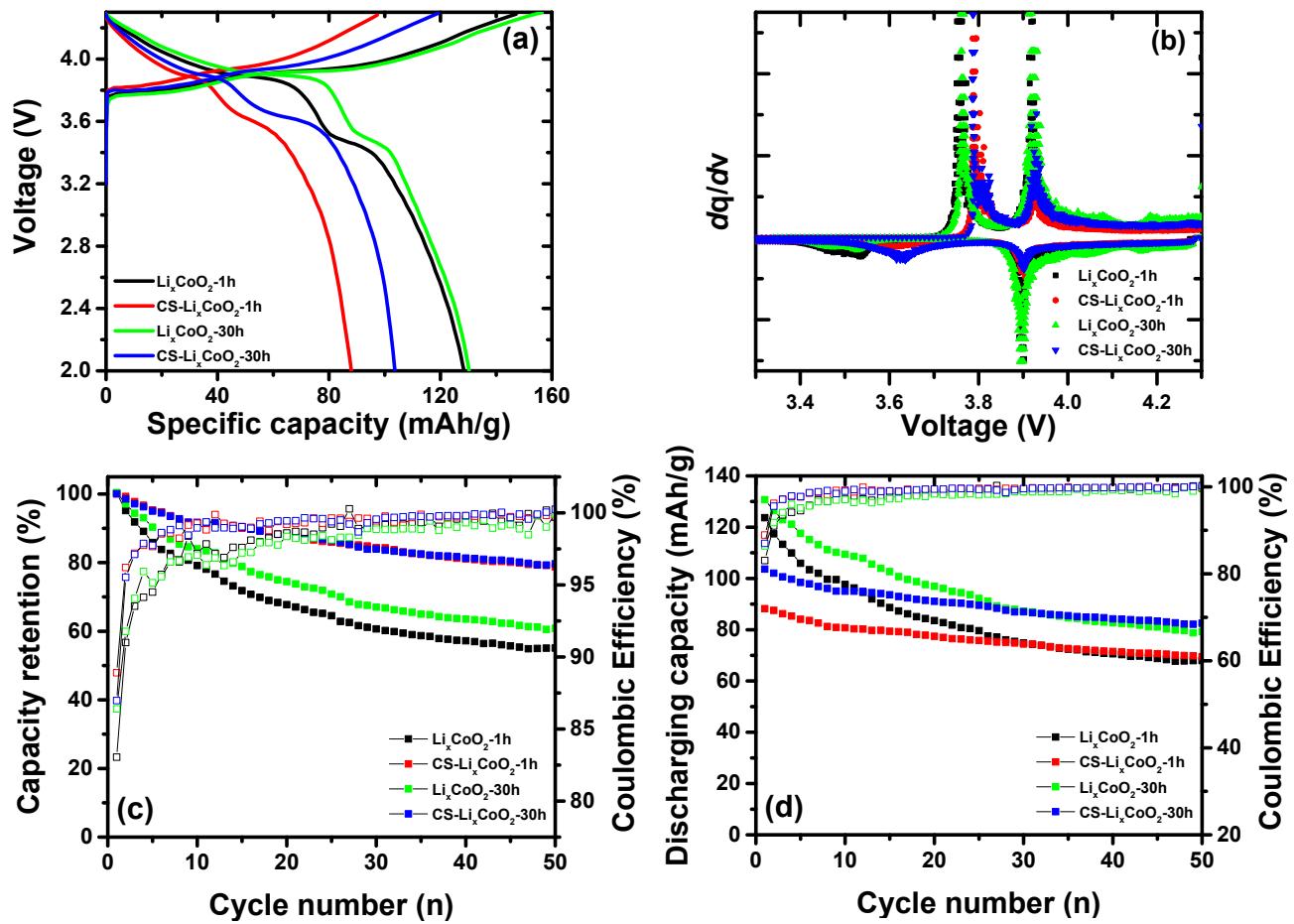


Figure S10. (a) Voltage versus specific profiles, (b) corresponding  $dQ/dV$  plots, (c) normalized specific capacity (solid symbol) and coulombic efficiency (open symbol), and (d) absolute discharge capacity (solid symbol) and coulombic efficiency of bare and core-shell  $\text{LiCoO}_2$  nanocrystal electrodes when cycled at C/20 in an electrolyte consisting of 1.2 M  $\text{LiPF}_6$  in a 3:7, w/w, mixture of ethylene carbonate (EC) and ethyl methyl carbonate (EMC).

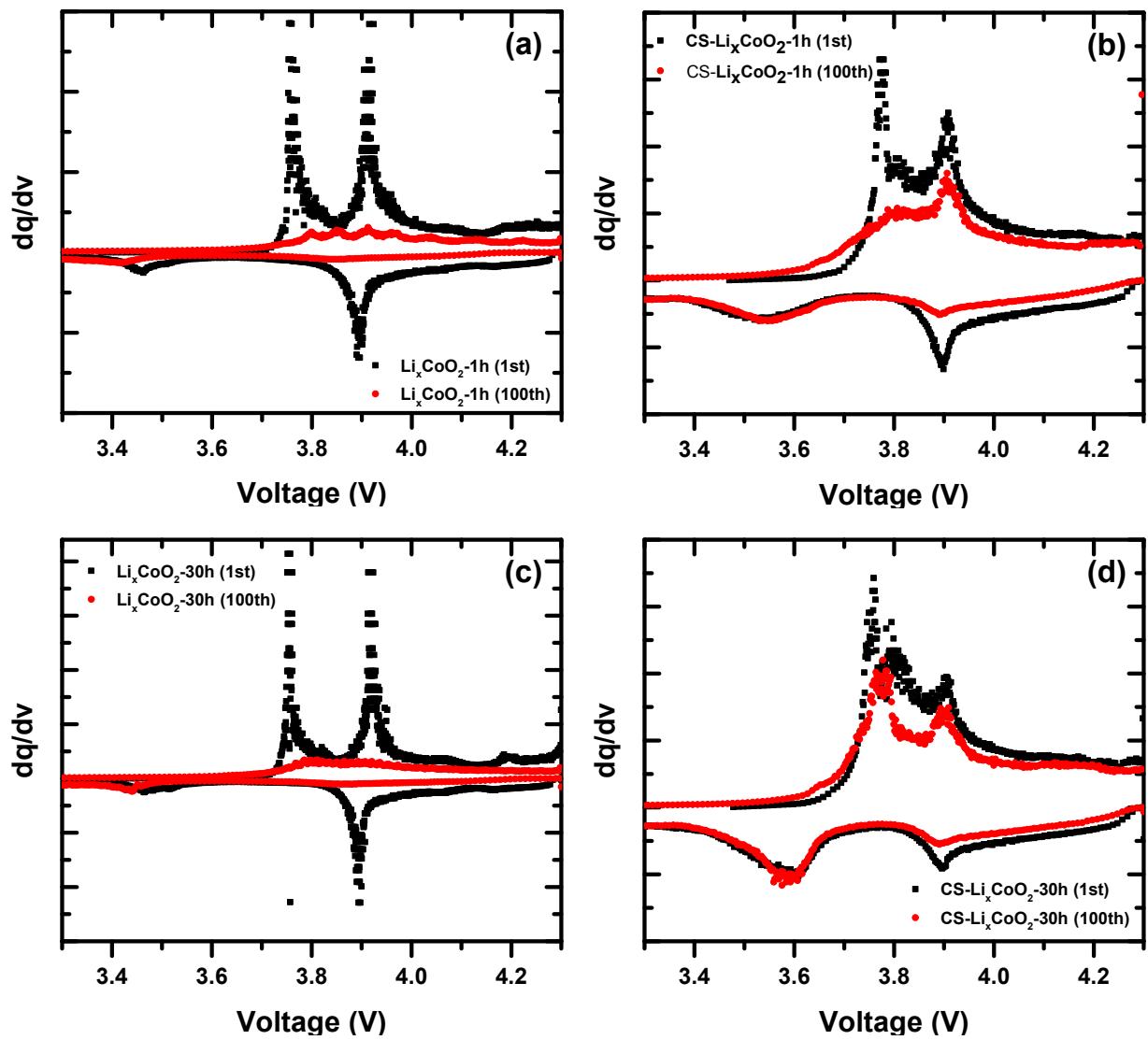


Figure S11.  $dQ/dV$  plots of (a)  $\text{LiCoO}_2$ -1h, (b)  $\text{CS-LiCoO}_2$ -1h, (c)  $\text{CS-LiCoO}_2$ -30h, and (d)  $\text{CS-LiCoO}_2$ -30h in the 1<sup>st</sup> (black) and 100<sup>th</sup> cycle (red), corresponding to the cells in Figure 6.