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Surfactant free one pot synthesis of CeO₂, TiO₂ and Ti@Ce oxide nanoparticles for ultra fast removal of Cr(VI) from aqueous media

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Caption for Supplementary Data

Fig. S1. BET-BJH pore size distribution of (a) CeO_2 (b) $Ti_{0.1}$ - $Ce_{0.9}$ oxide nanoparticles (c) $Ti_{0.3}$ - $Ce_{0.7}$ oxide nanoparticles (d) TiO_2 .

Fig. S2. Zeta potential at variable pH.

Fig. S3. EDAX mapping image of CeO₂ adsorbed Cr(VI).

Fig. S4. EDAX mapping image of Ti_{0.3}@ Ce0.7 oxide nanoparticles adsorbed Cr(VI).

Fig. S5. EDAX mapping image of TiO₂ adsorbed Cr(VI).

Fig. S6. The reusability study for removal of Cr(VI) using Ti_{0.3}@Ce_{0.7} oxide nanoparticles. (b) Surface area of Ti_{0.3}@Ce_{0.7} after adsorption of Cr(VI).

Fig. S7. (a) Langmuir Isotherm plot (b) Freundlich Isotherm plot (c) Redlich-Peterson Isotherm plot (d) D-R Isotherm plot.

Fig. S8. Pseudo Second Order Reaction.

Fig. S9. FTIR spectra of $Ti_{0.3}$ ($Ce_{0.7}$ oxide nanoparticles before and after adsorption of Cr(VI).

Table 1. Standard Error for Langmuir Isotherm.

Table 2. Standard Error for Freundlich Isotherm.

 Table 3. Standard Error for R-P Isotherm.

Table 4. Standard Error for D-R Isotherm.

Supplementary Data



Fig. S1 BET-BJH pore size distribution of (a) CeO₂; (b) Ti_{0.1}@Ce_{0.9} oxide nanoparticles; (c) Ti_{0.3}@Ce_{0.7} oxide nanoparticles and (d) TiO₂.



Fig. S2 Zeta potential at variable pH.



Fig. S3 EDAX mapping image of CeO₂ adsorbed Cr(VI).



Fig. S4 EDAX mapping image of Ti_{0.3}@Ce_{0.7} oxide nanoparticles adsorbed Cr(VI).



Fig. S5 EDAX mapping image of TiO₂ adsorbed Cr(VI).



Fig. S6 (a) The reusability study for removal of Cr(VI) using Ti_{0.3}@Ce_{0.7} oxide nanoparticles.



Fig. S6 (b) Surface area of Ti_{0.3}@Ce_{0.7} oxide nanoparticles after adsorption of Cr(VI).



Fig. S7 (a) Langmuir Isotherm plot.



Fig. S7 (b) Freundlich Isotherm plot.



Fig. S7 (c) Redlich-Peterson Isotherm plot.



Fig. S7 (d) D-R Isotherm plot.



Fig. S8 Pseudo Second Order Reaction.



Fig. S9 FTIR spectra of Ti@Ce oxide Before and after adsorption of Cr(VI).

Parameters	Values (CeO ₂)	Values (TiO ₂)	Values Ti _{0.1} @Ce _{0.9} oxide	Values Ti _{0.3} @Ce _{0.7} oxide
Intercept error	0.0216	0.0216	0.00587	0.00675
Slope error	6.34*10-4	6.34*10-4	2.93*10-4	3.01*10-4
Adjacent R ²	0.982	0.982	0.991	0.991
Pearson R ²	0.9928	0.9928	0.996	0.996
Residual sum	0.00394	0.01674	3.98*10-4	5.1*10-4
of squares				
Sum of squares	0.02752	0.02752	0.057	0.0744
F values	275.248	275.248	572.34	578.22
Prob>F	7.7 * 10 ⁻⁵	7.7 * 10 ⁻⁵	1.8*10-5	1.7*10-4

Table 1. Standard Error for Langmuir Isotherm

Table 2. Standard Error for Freundlich Isotherm

Parameters	Values (CeO ₂)	Values (TiO ₂)	Values Ti _{0.1} @Ce _{0.9} oxide	Values Ti _{0.3} @Ce _{0.7} oxide
Intercept error	0.166	0.112	0.150	0.207
Slope error	0.050	0.0318	0.060	0.087
Adjacent R ²	0.69	0.983	0.865	0.740
Pearson R ²	0.89	0.959	0.944	0.890
Residual sum	0.027	0.021	0.205	0.398
Sum of squares	0.091	0.648	1.713	1.52
F values	10.16	119.68	33.29	15.22
Prob>F	0.0498	3.9*10-4	0.00448	0.0174

Parameters	Values (CeO ₂)	Values (TiO ₂)	Values Ti _{0.1} @Ce _{0.9} oxide	Values Ti _{0.3} @Ce _{0.7} oxide
Intercept error	0.0188	0.0331	0.00445	0.00469
Slope error	0.00107	0.001	4.4*10-4	3.9*10-4
Adjacent R ²	0.997	0.995	0.996	0.997
Pearson R ²	0.994	0.998	0.998	0.994
Residual sum	0.0027	0.0059	1.8*10-4	2.3*10-4
of squares				
Sum of squares	0.272	1.688	0.0747	0.0572
F values	394.43	1140.18	1583.3	985.5
Prob>F	3.7*10-5	4.5*10 ⁻⁶	2.3*10-6	6.1*10-6

Table 3. Standard Error for R-P Isotherm

Table 4. Standard Error for D-R Isotherm

Parameters	Values	Values	Values	Values
	(CeO ₂)	(TiO ₂)	Ti _{0.1} @Ce _{0.9} oxide	Ti _{0.3} @Ce _{0.7}
				oxide
Intercept error	0.175	0.101	0.411	0.413
Slope error	0.00019	0.00143	0.0040	0.00403
Adjacent R ²	0.918	0.975	0.924	0.782
Pearson R ²	0.967	0.990	0.969	0.908
Residual sum	0.0787	0.0129	0.383	0.334
of squares				
Sum of squares	1.134	0.657	1.00	1.588
F values	57.62	202.3	62.008	19.01
Prob>F	0.0016	1.4*10-4	0.00141	0.0120