

SUPPORTING INFORMATION

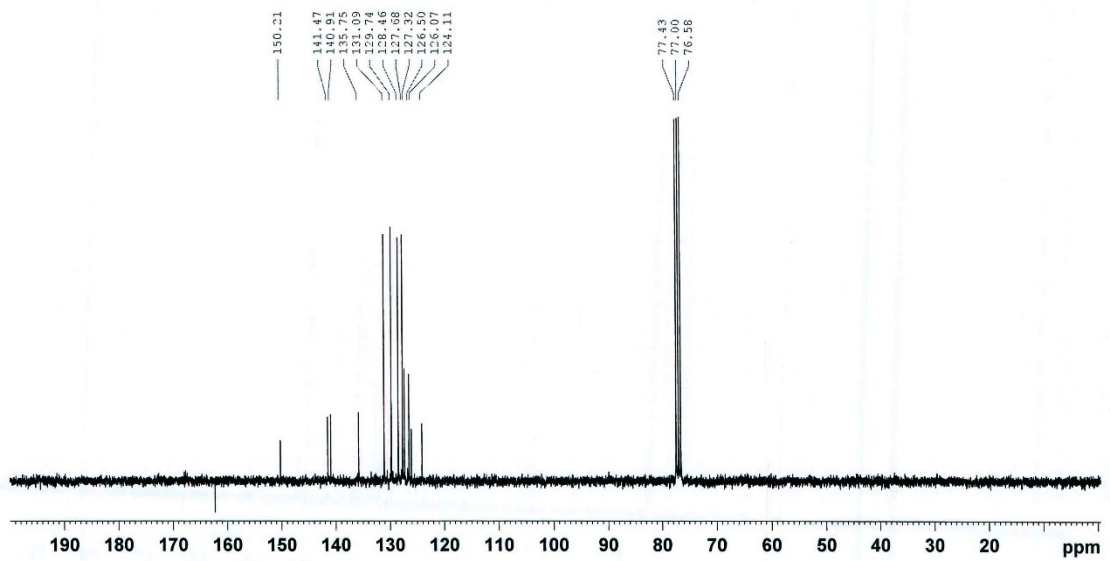
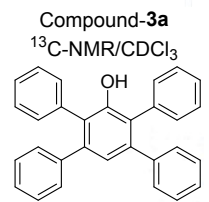
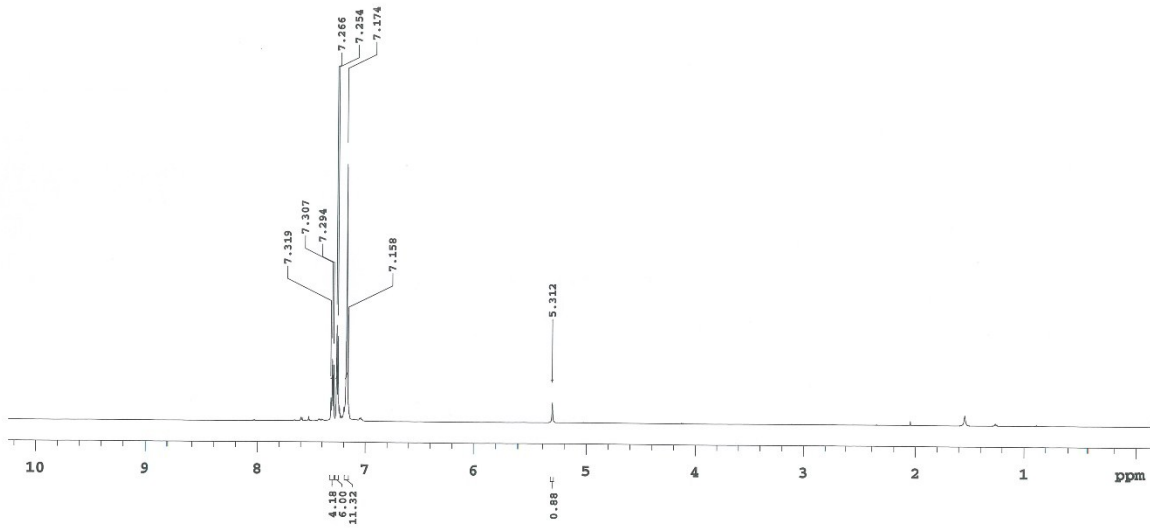
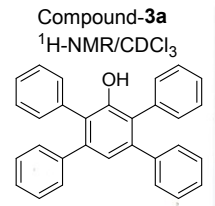
One-pot construction of sterically challenging polyarylphenols *via* transition-metal-free benzannulation and their *in-vitro* antioxidant activity

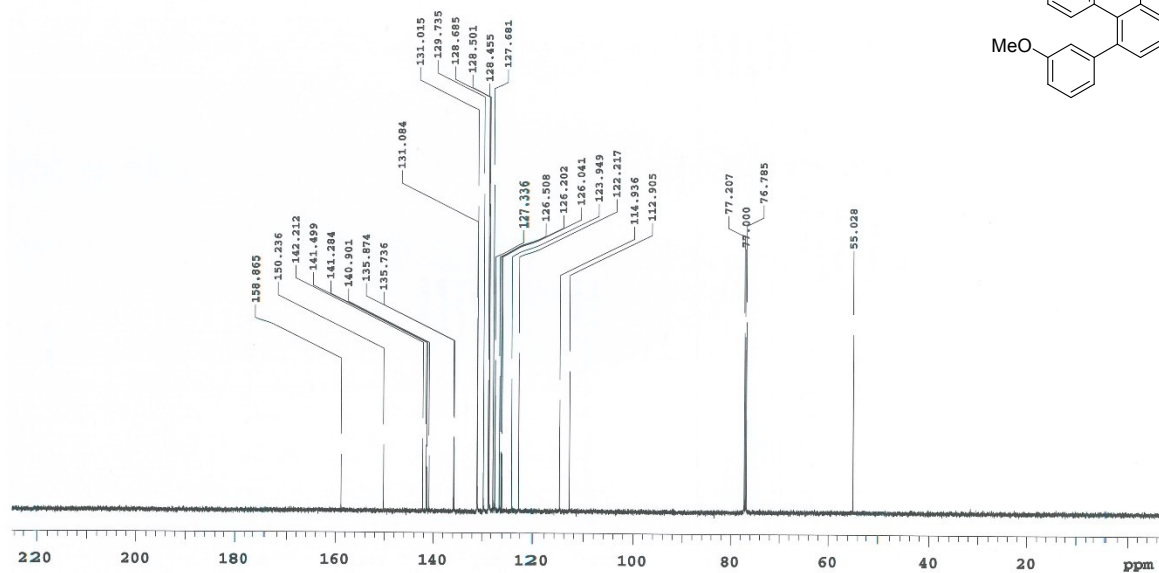
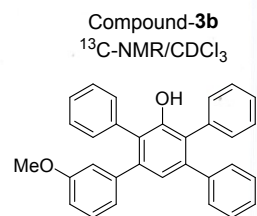
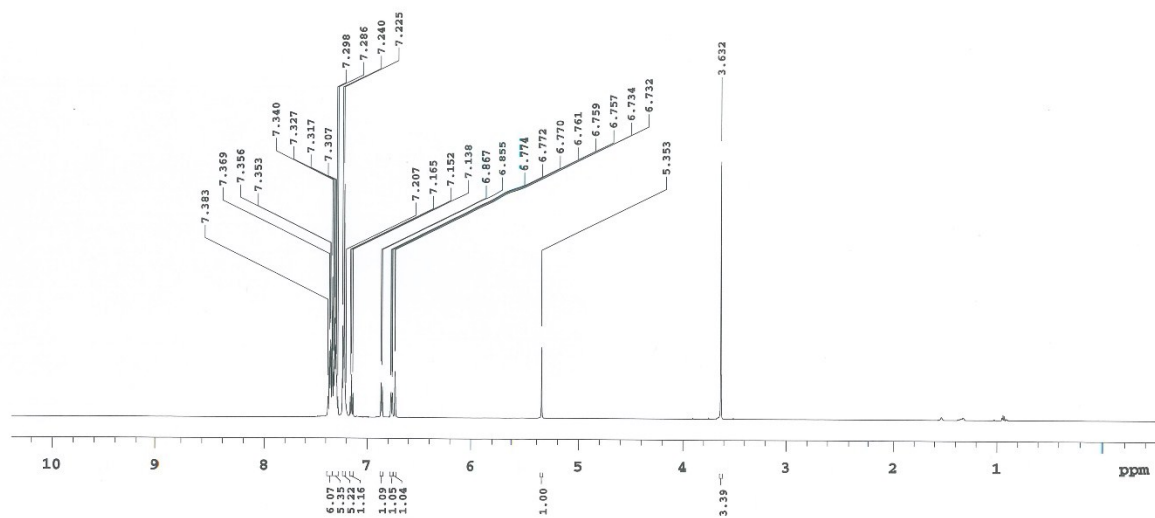
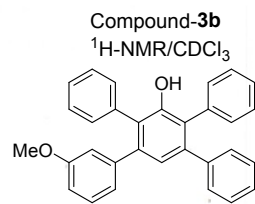
Shizuka Mei Bautista Maezono,^a Tej Narayan Poudel,^a and Yong Rok Lee^{a*}

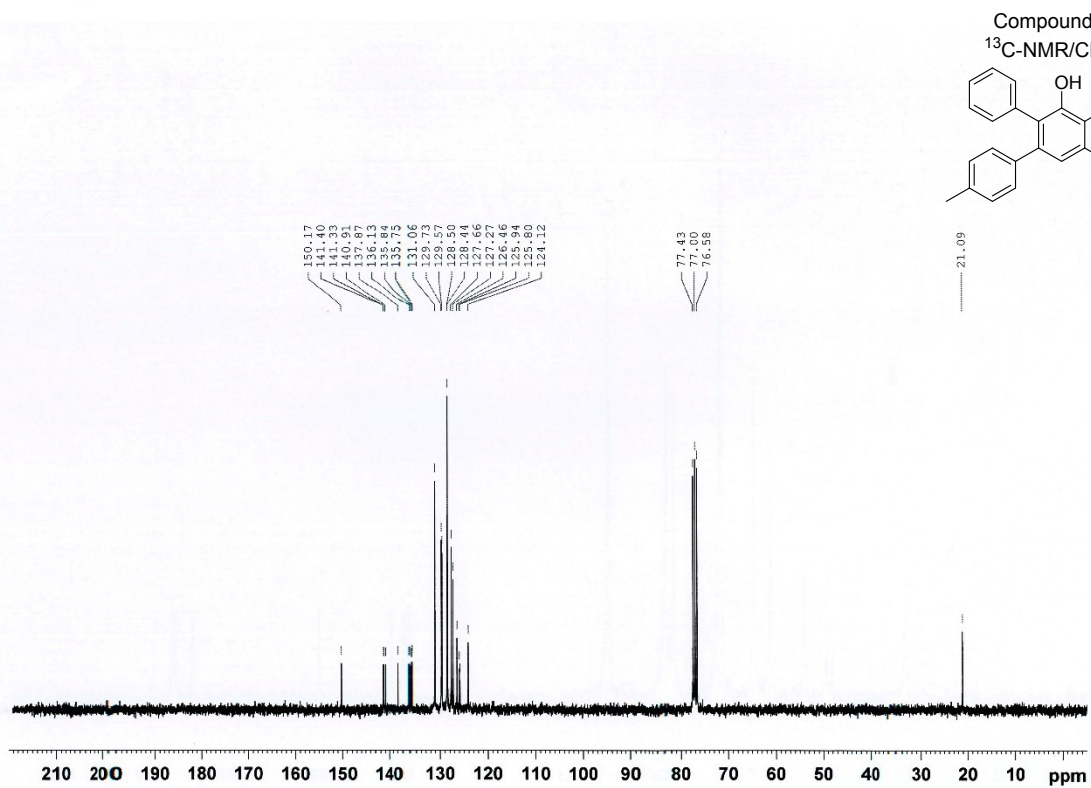
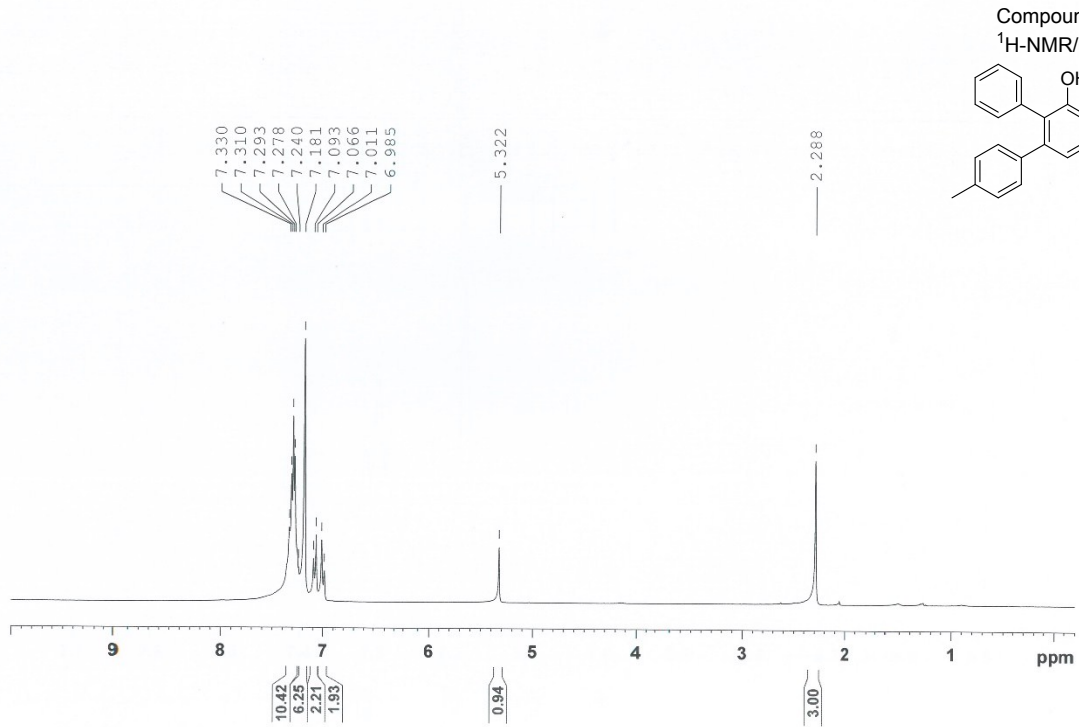
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Email: yrlee@yu.ac.kr, Phone: +82-53-810-2529; Fax: +82-53-810-4631

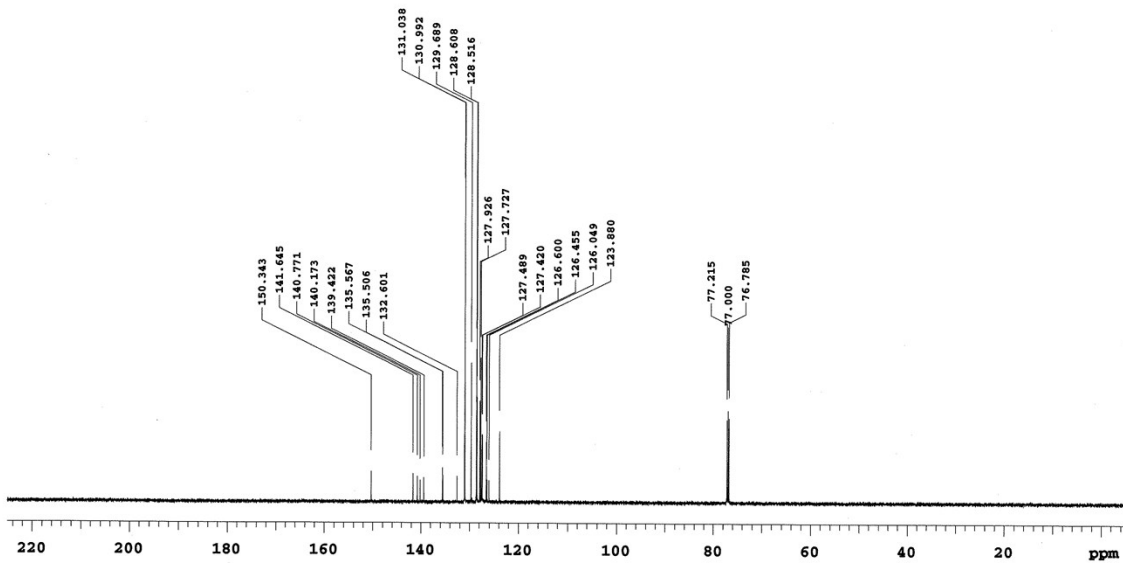
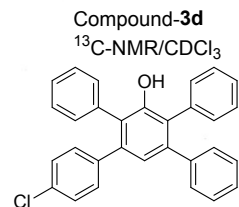
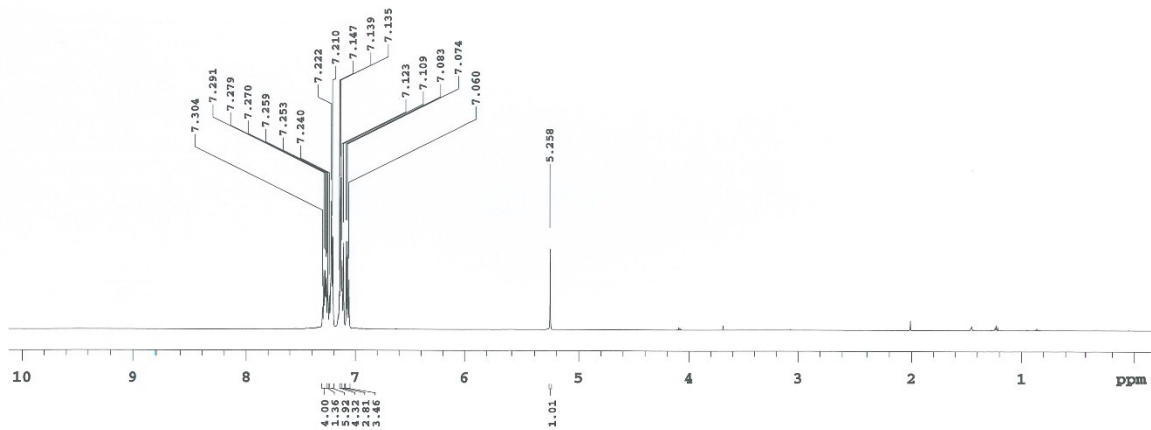
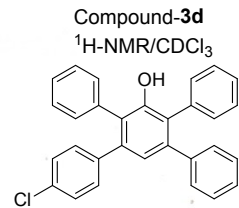
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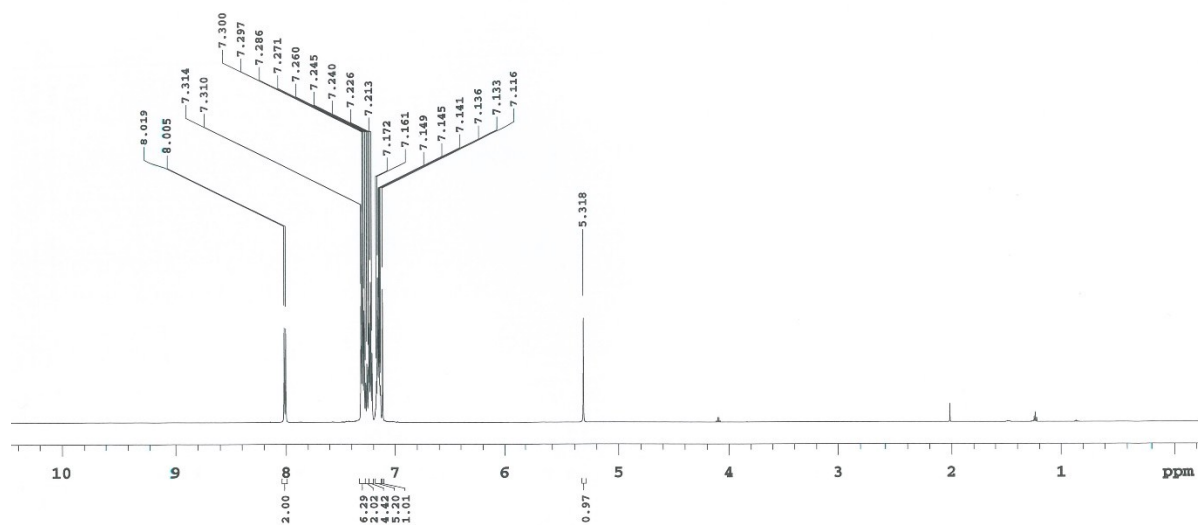
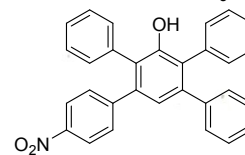






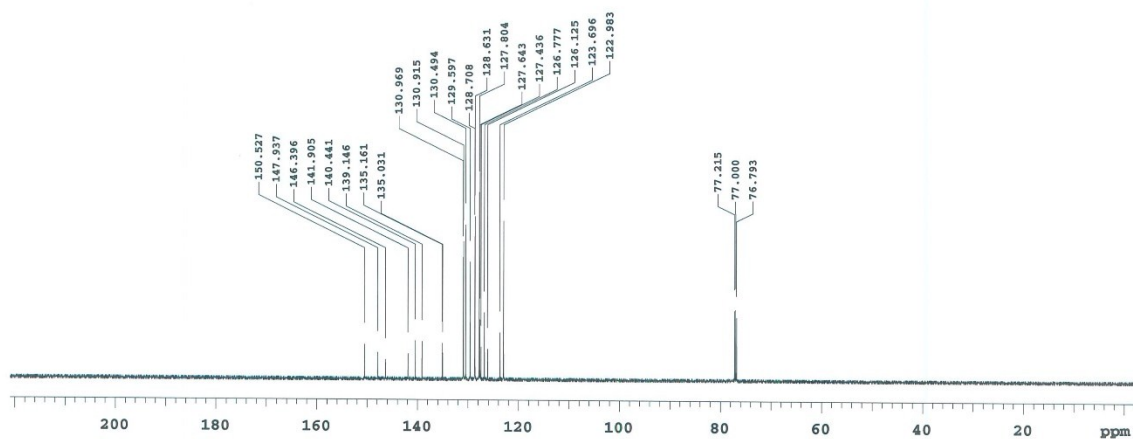
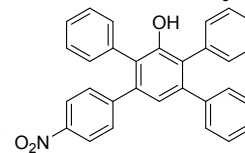
Compound-3e

$^1\text{H-NMR}/\text{CDCl}_3$

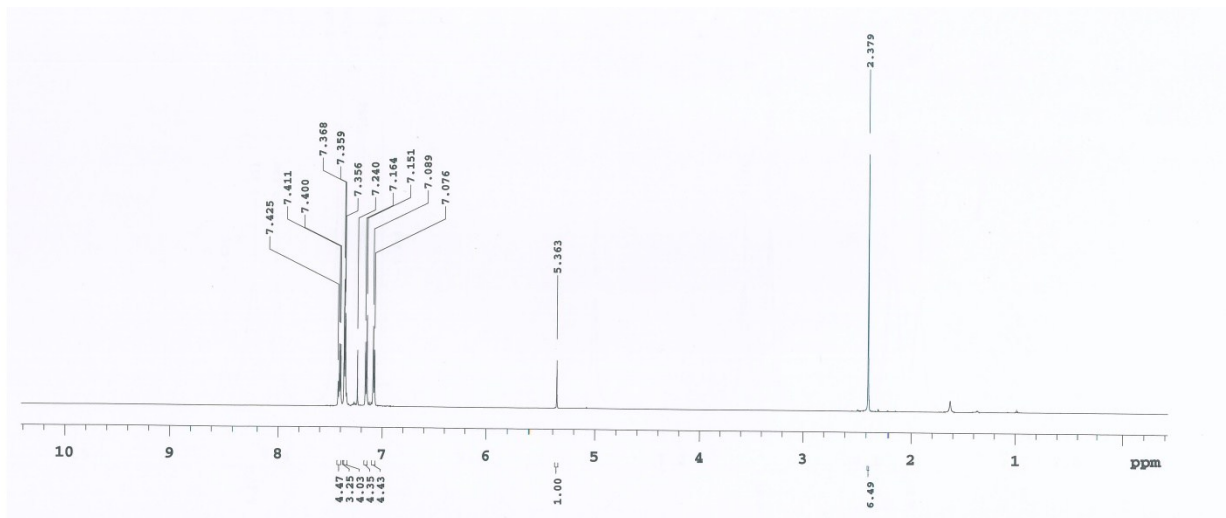
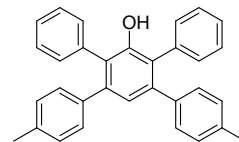


Compound-3e

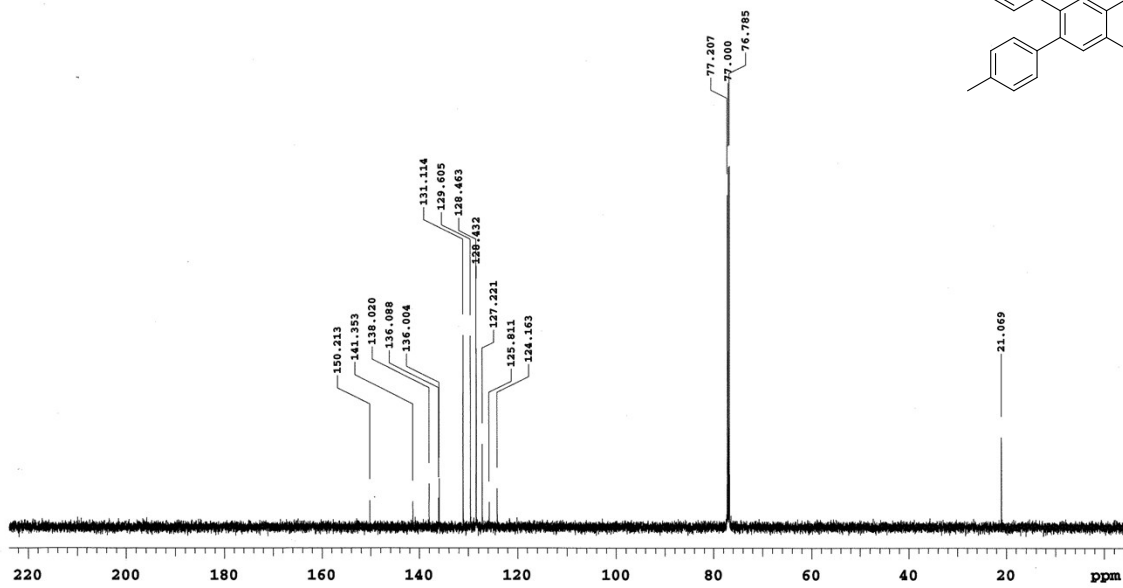
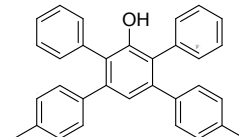
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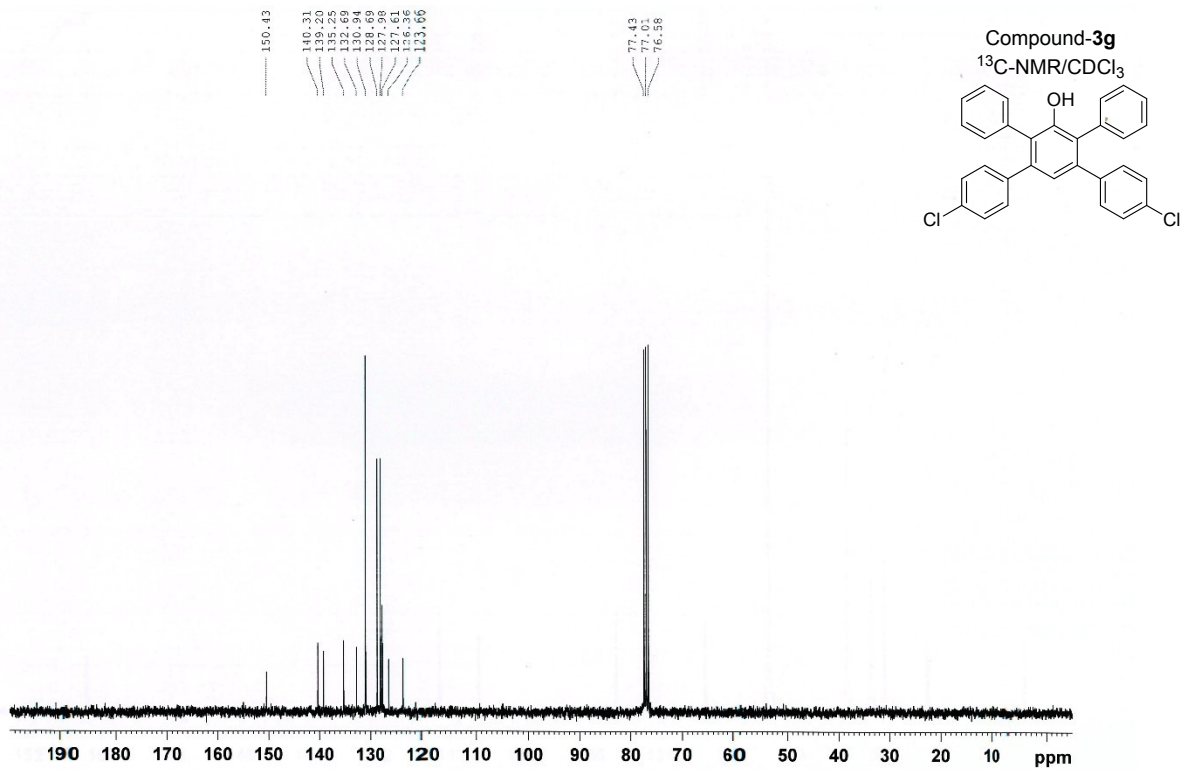
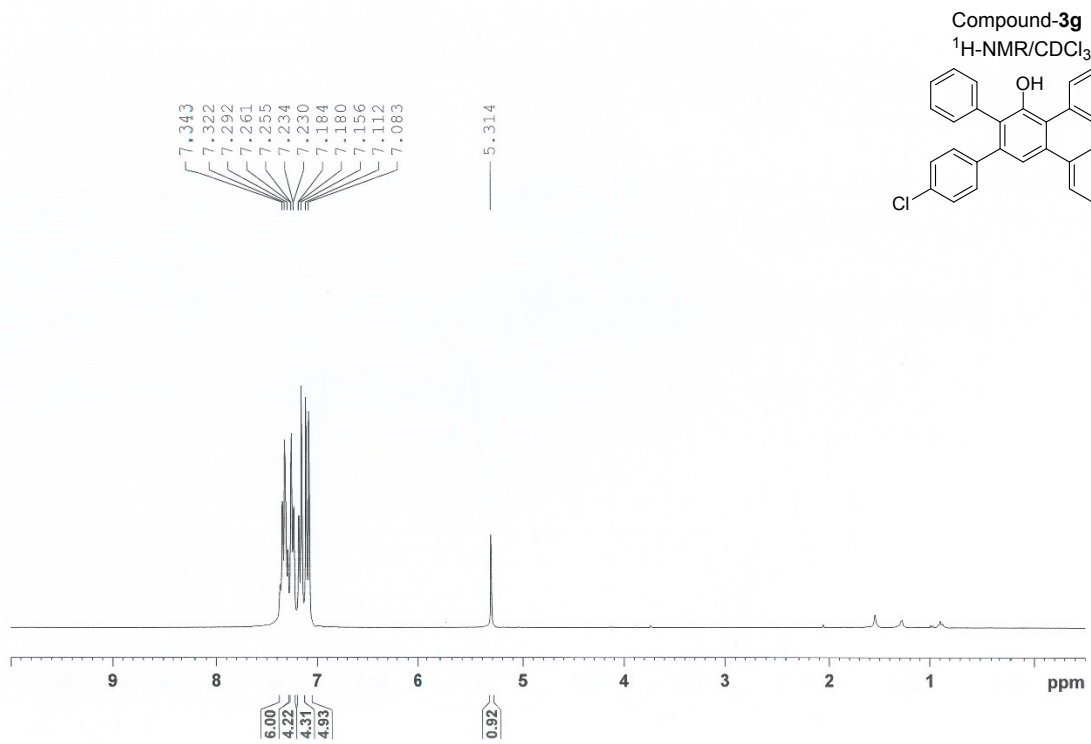


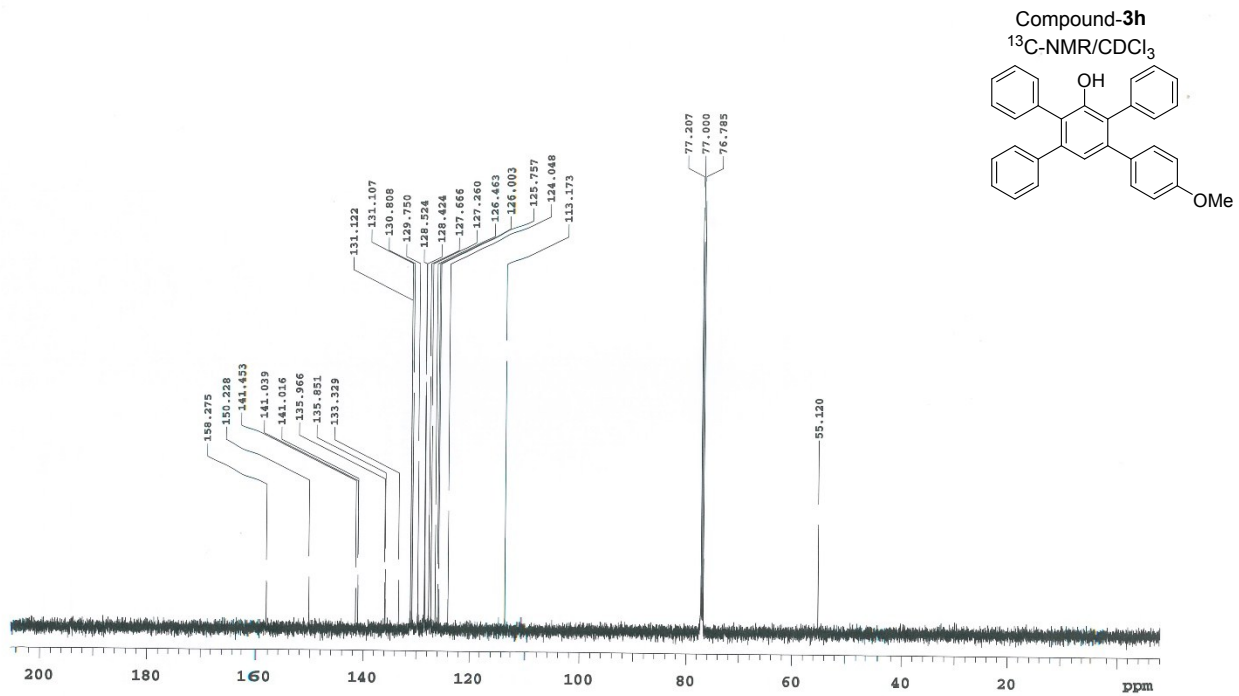
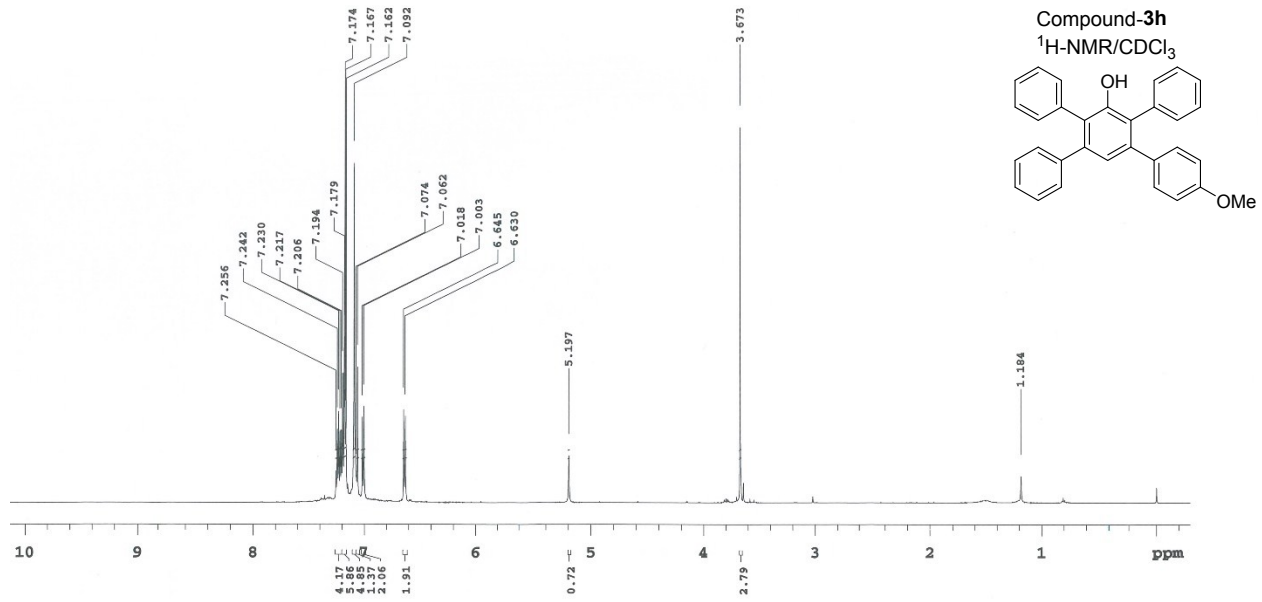
Compound-3f
¹H-NMR/CDCl₃

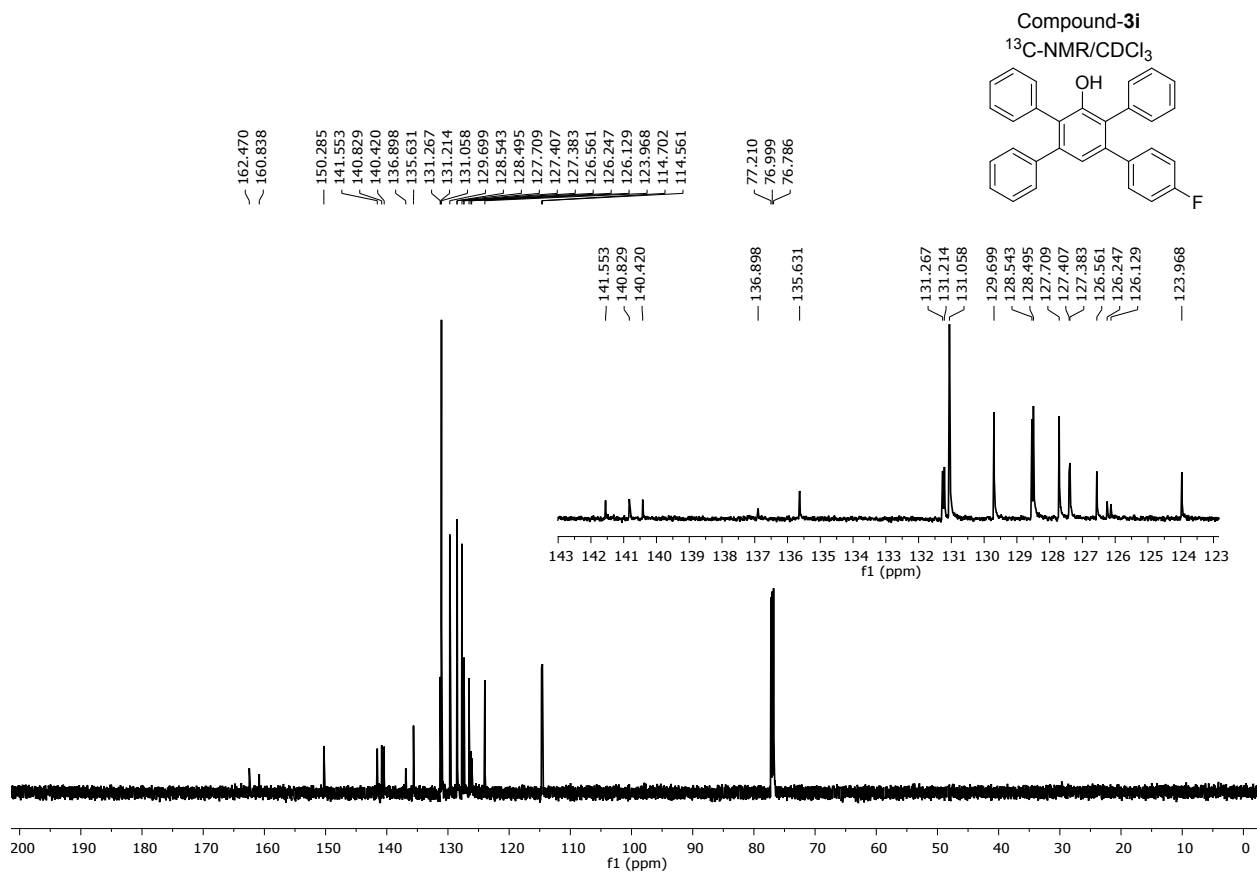
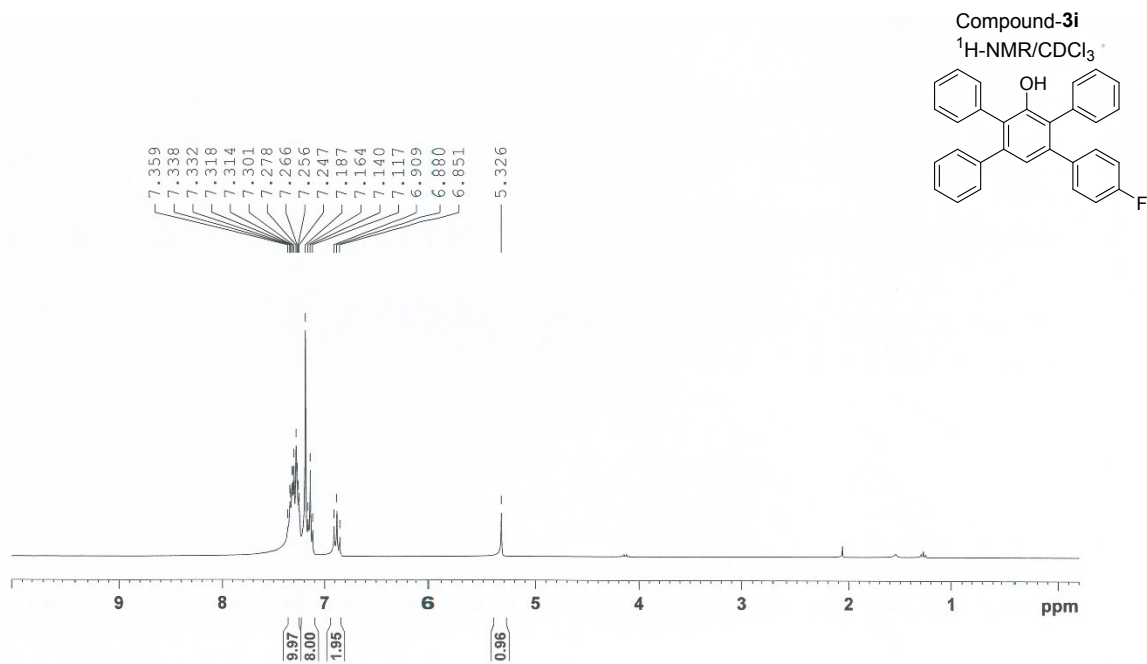


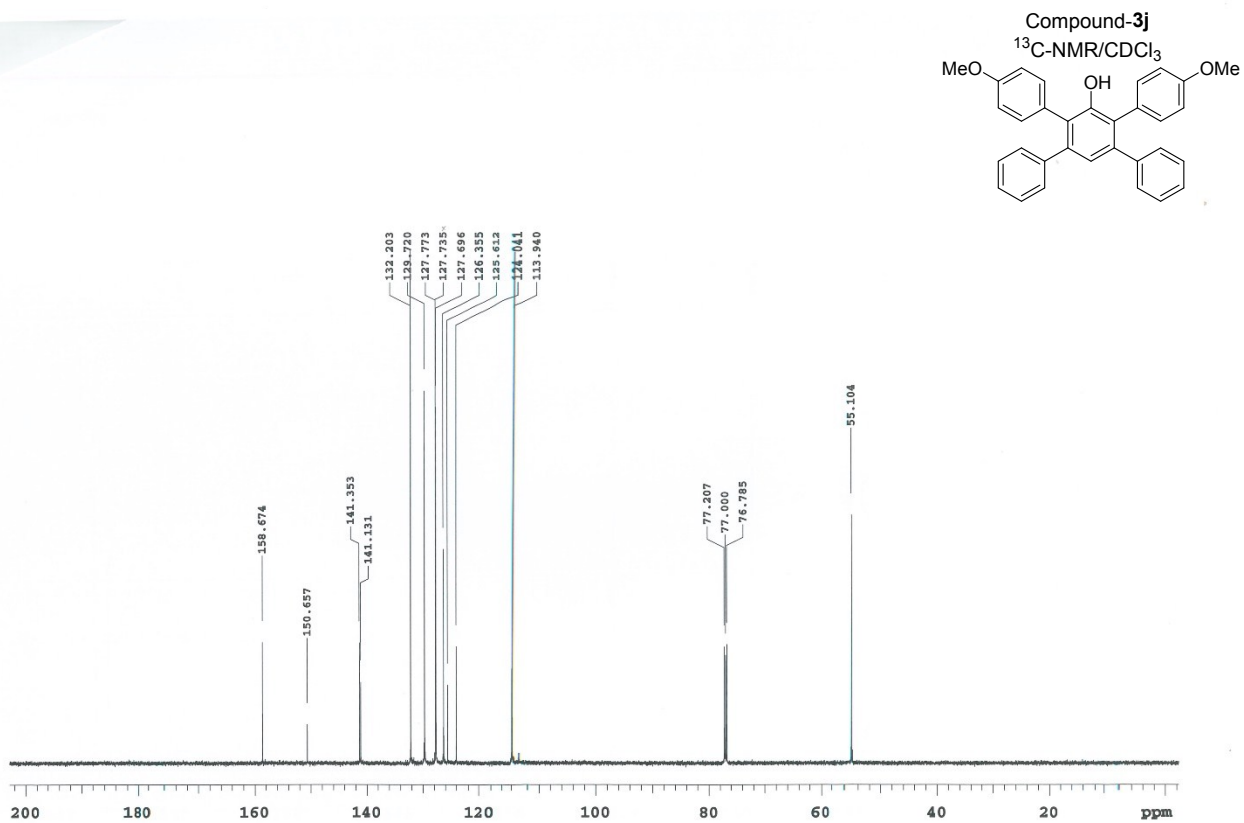
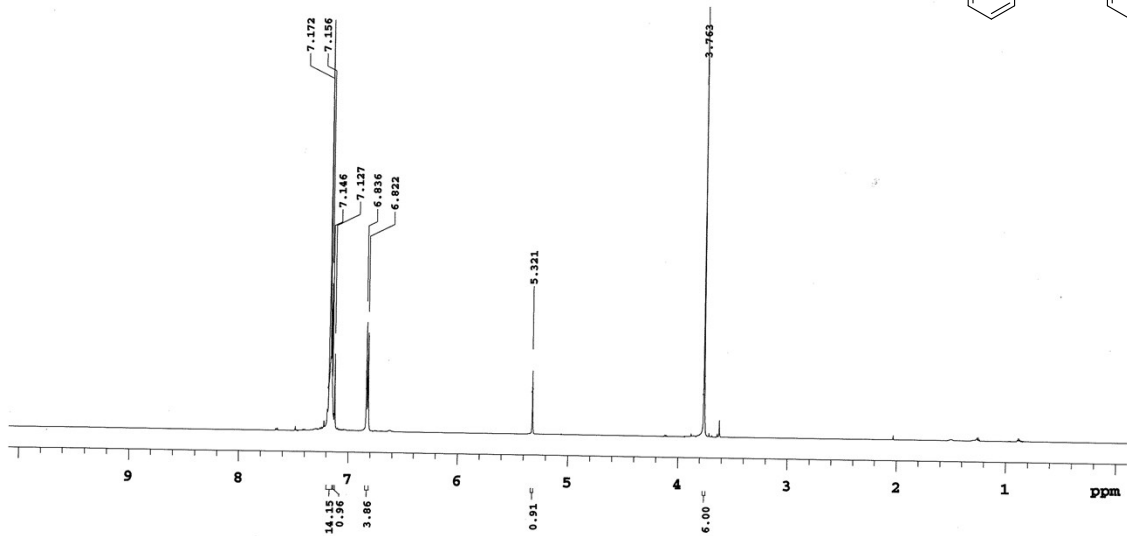
Compound-3f
¹³C-NMR/CDCl₃

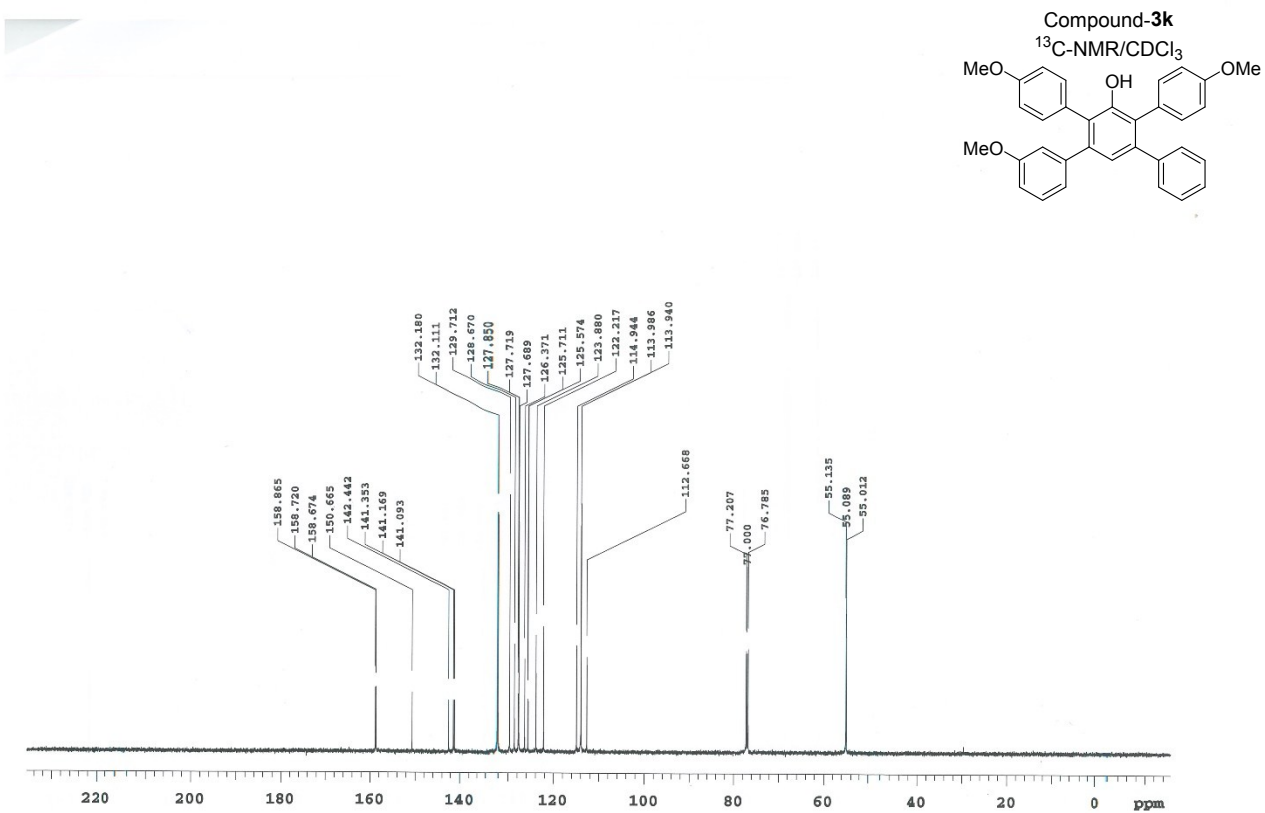
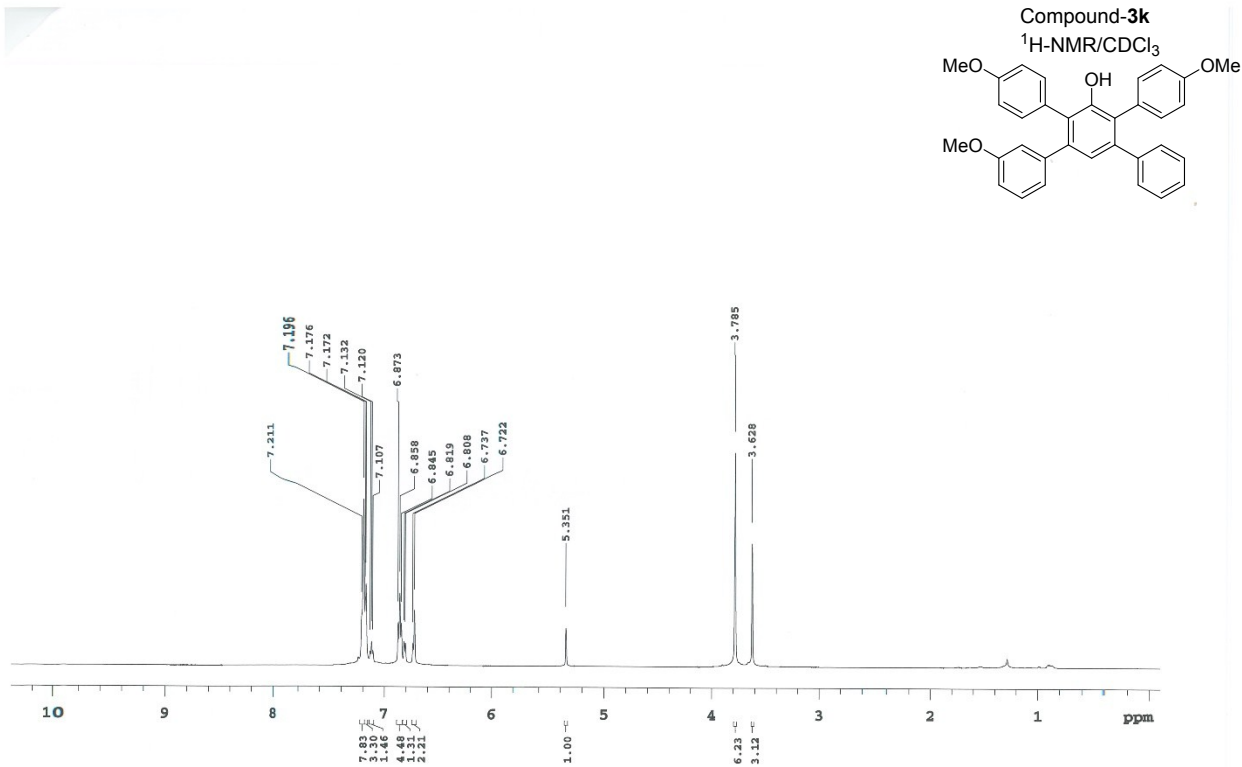


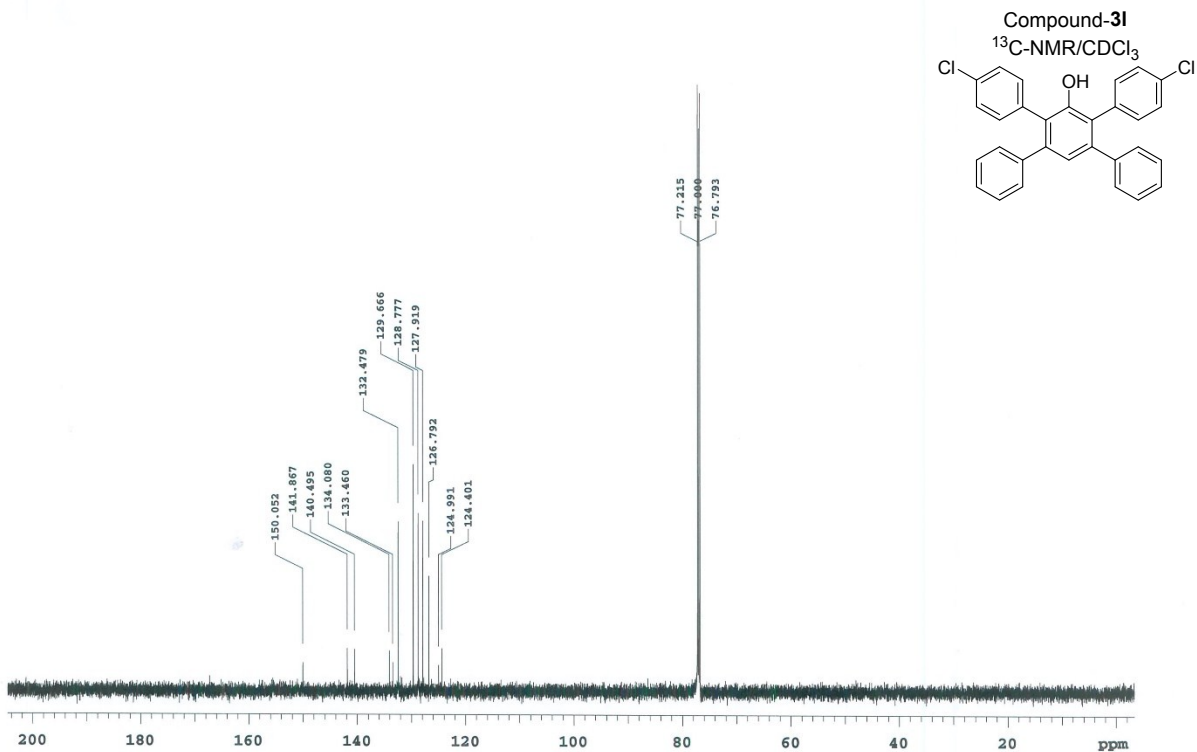
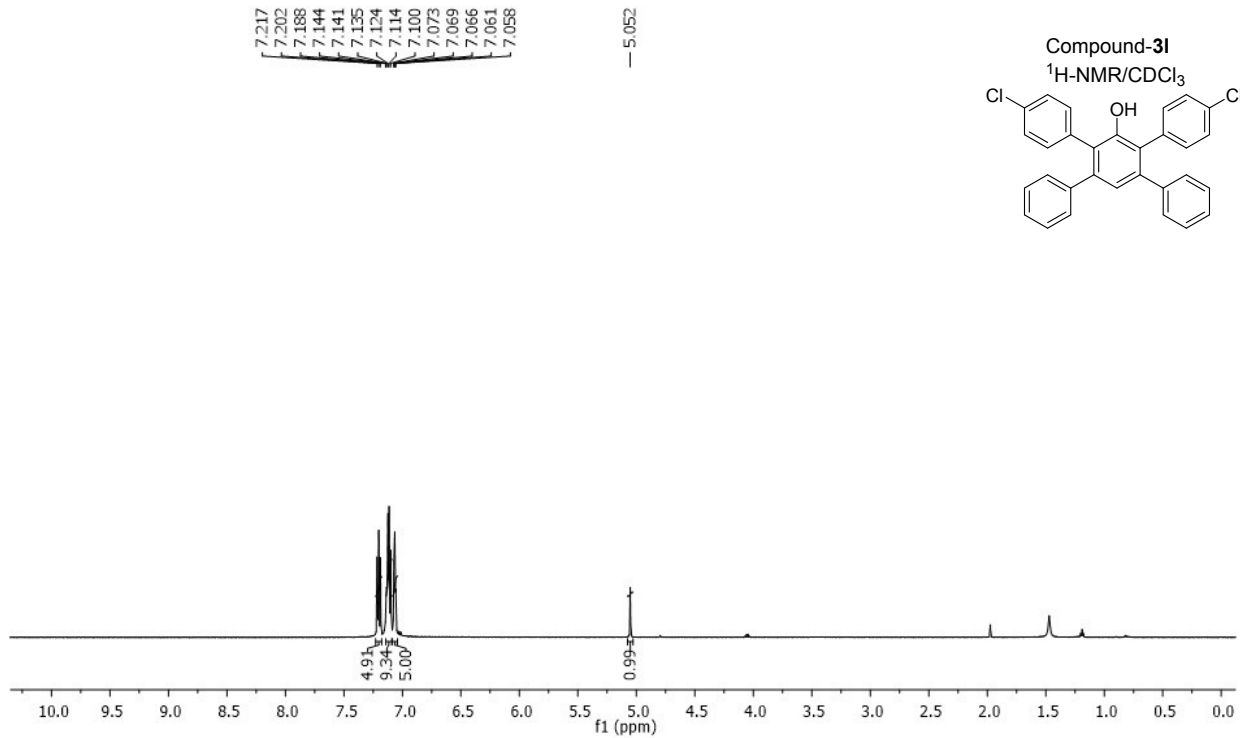


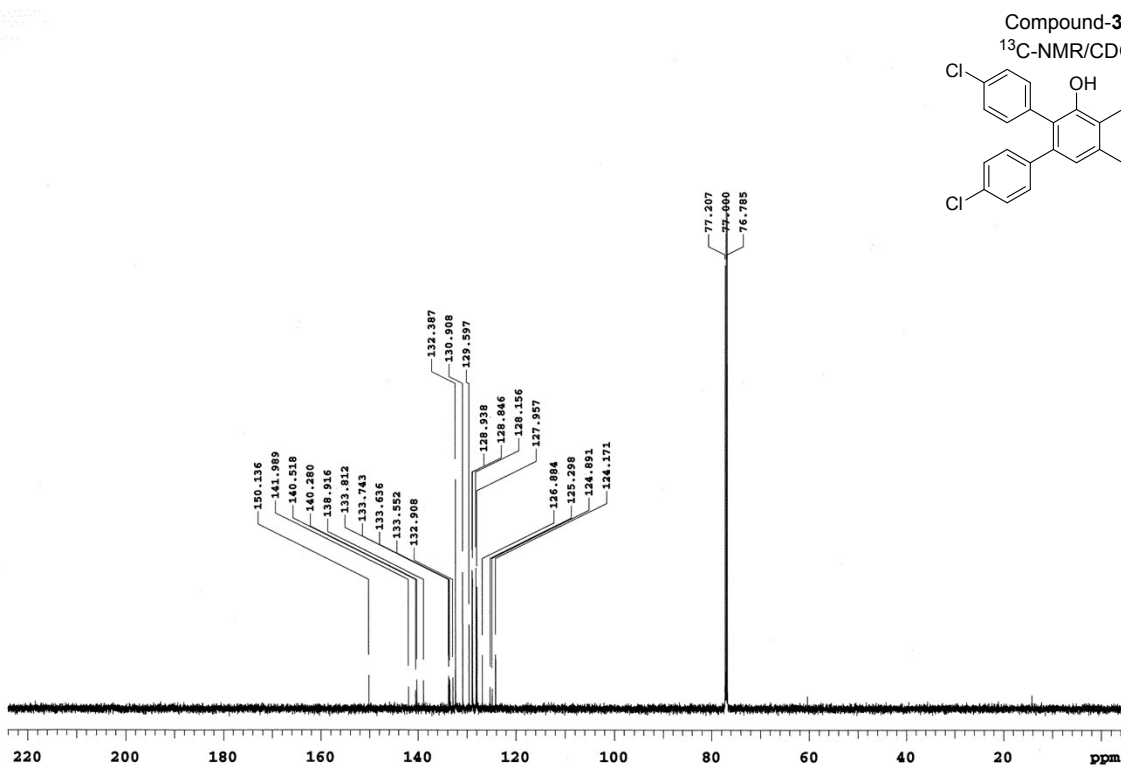
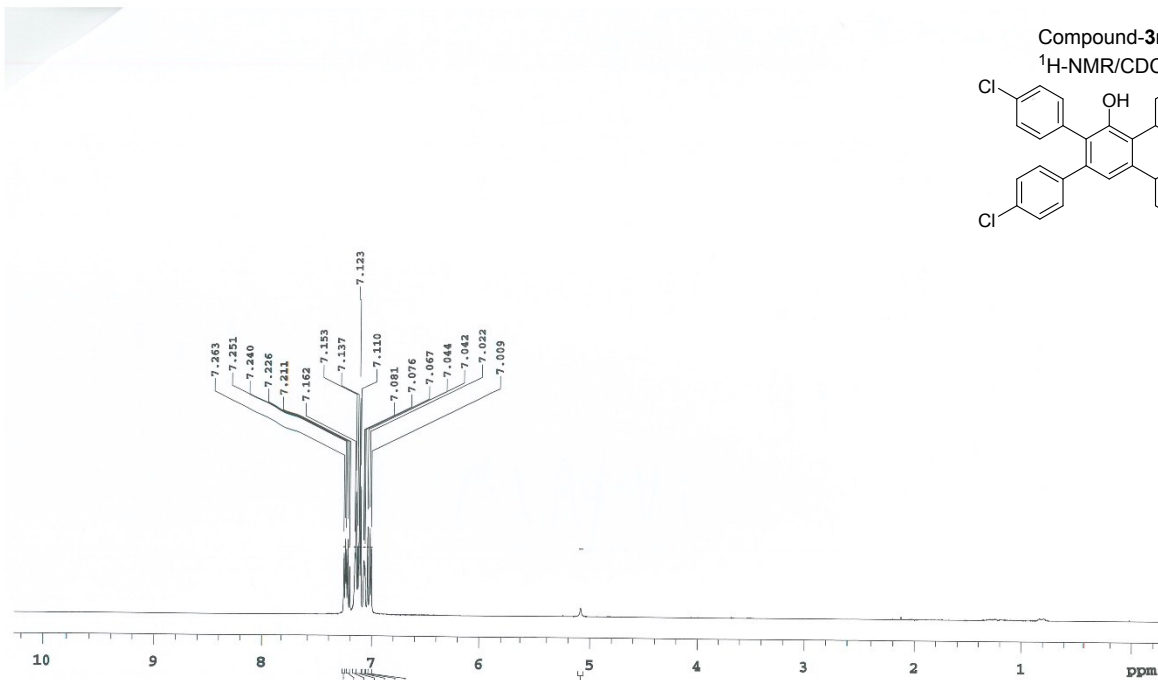


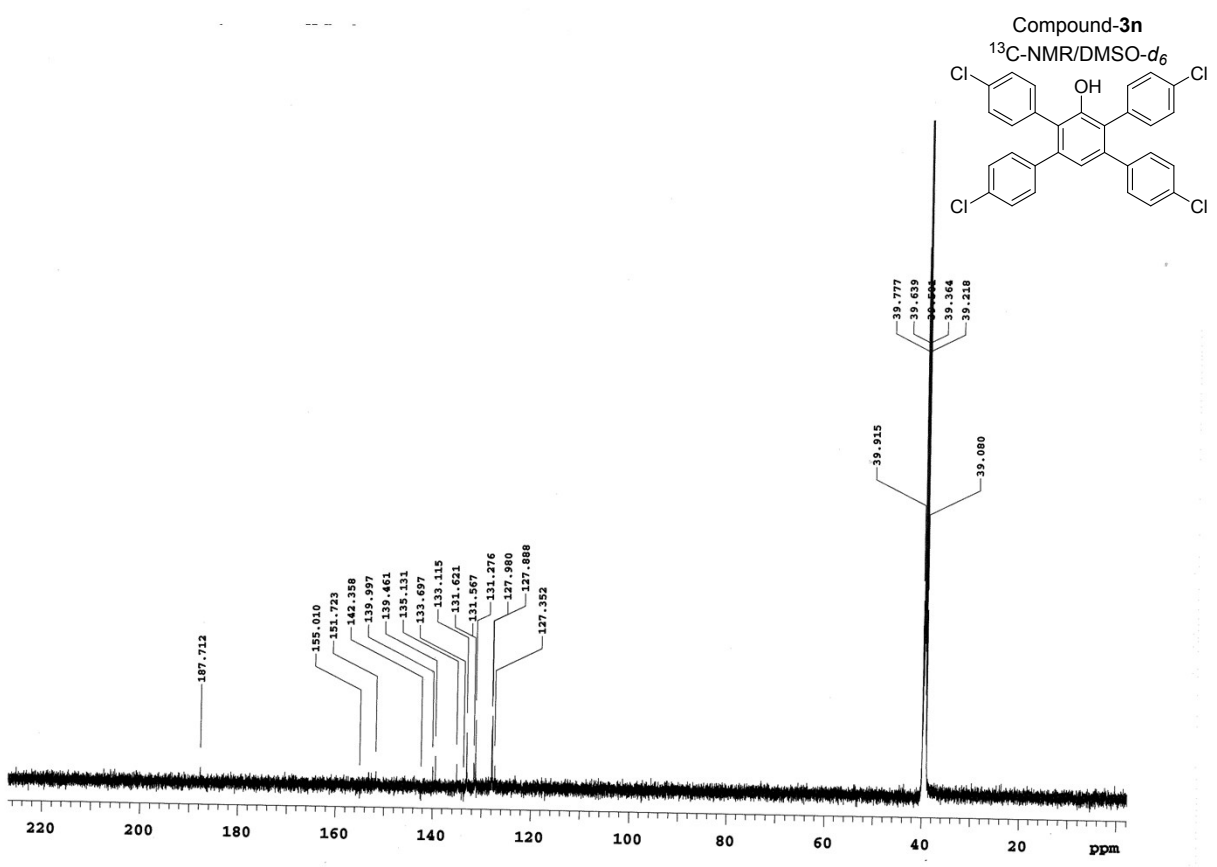
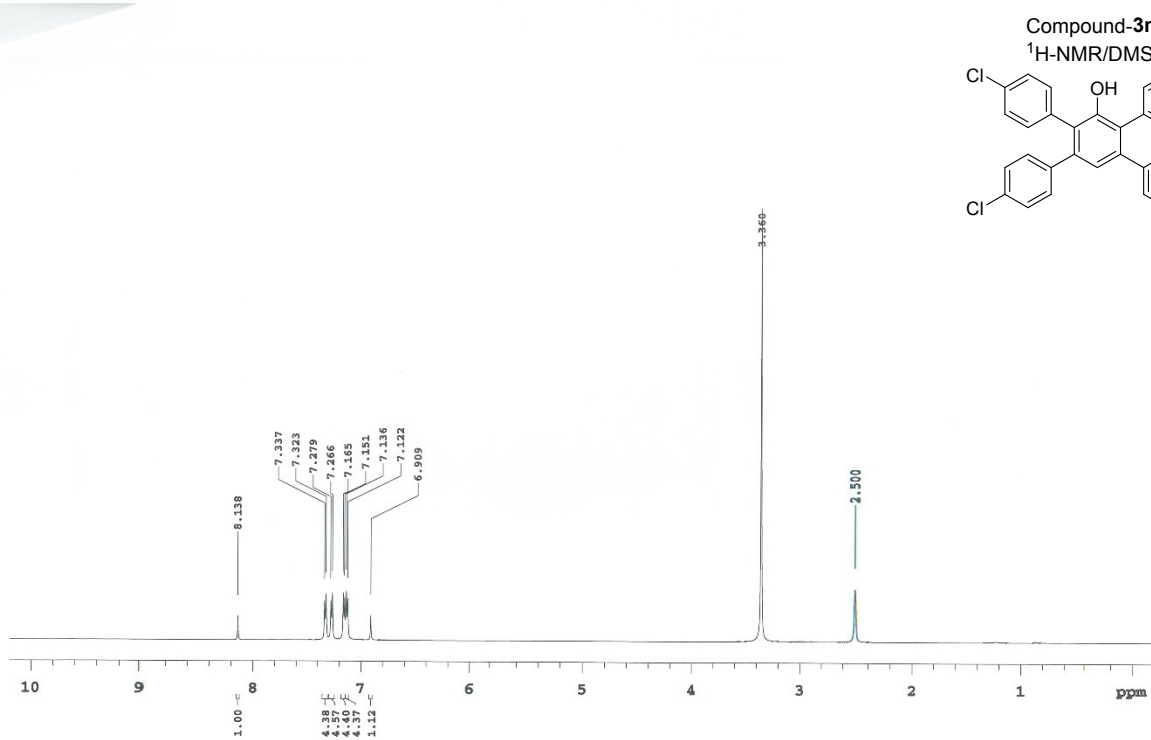


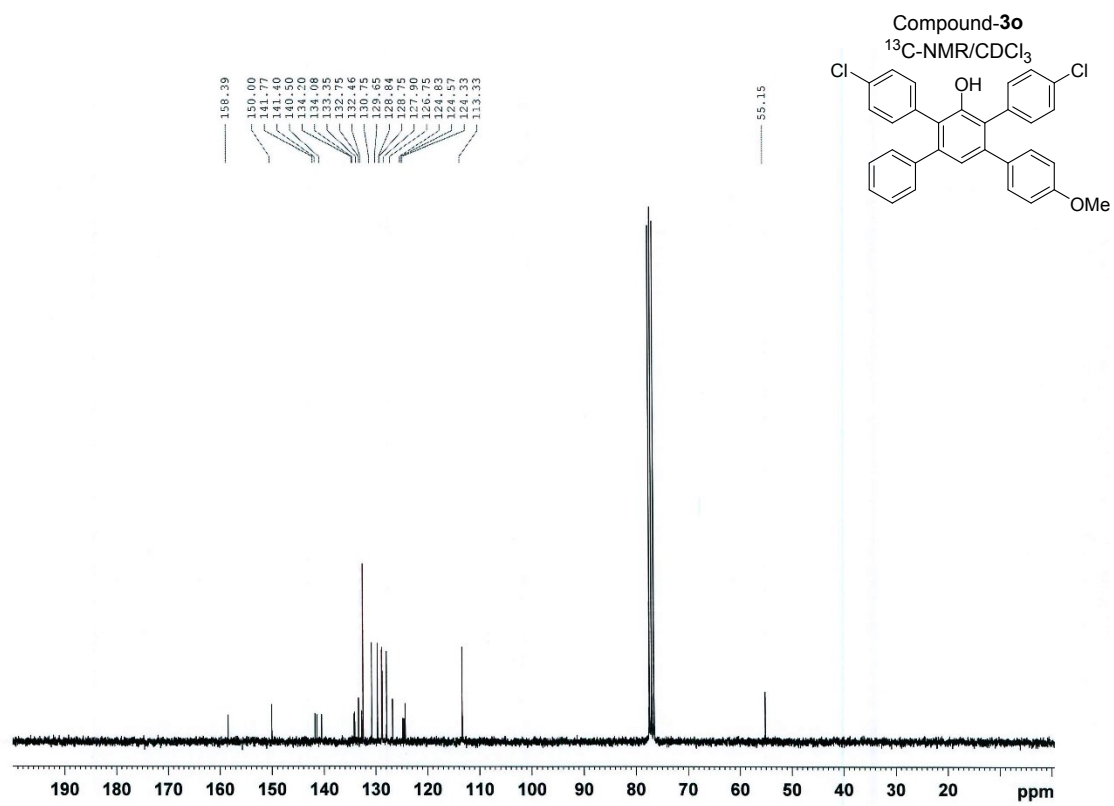
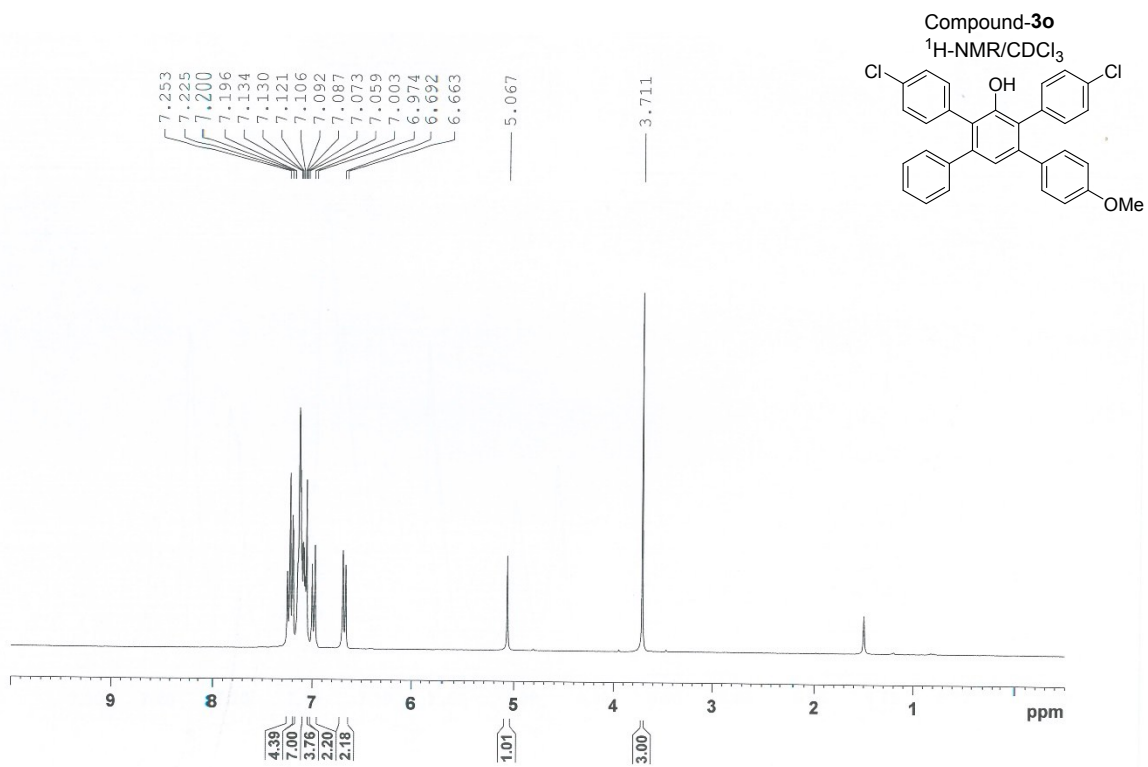


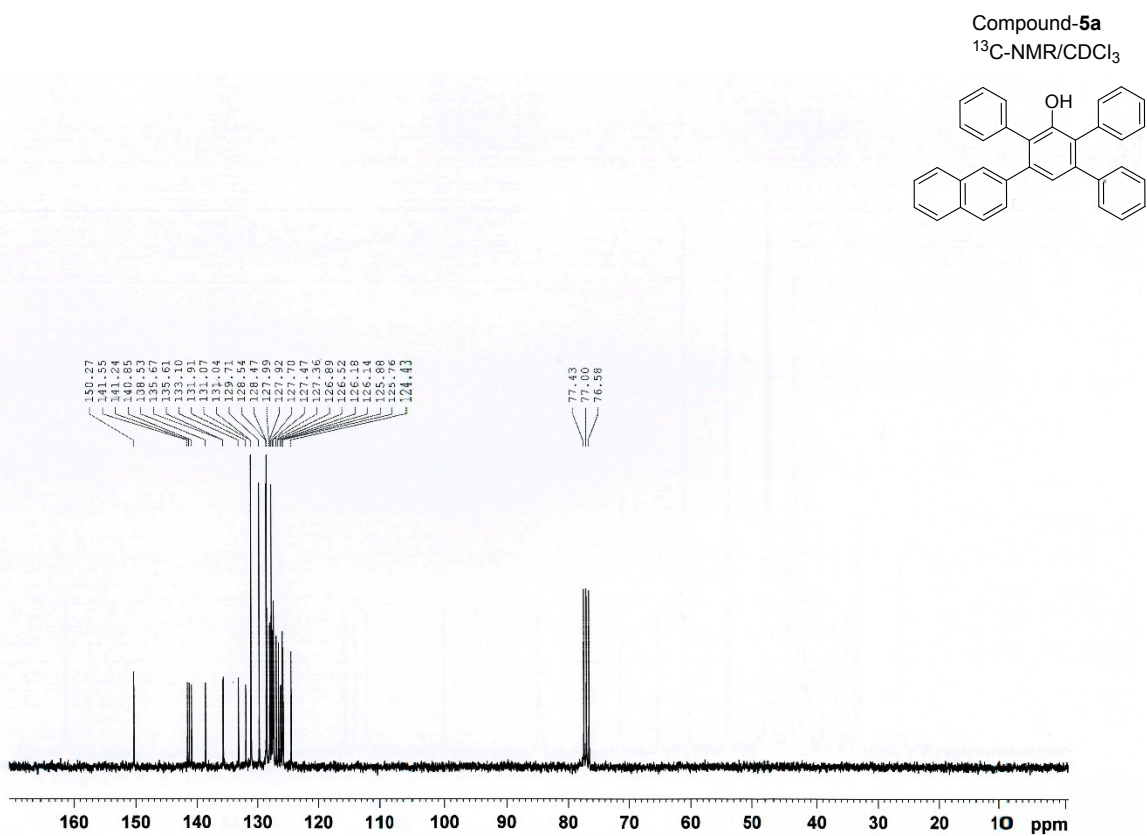
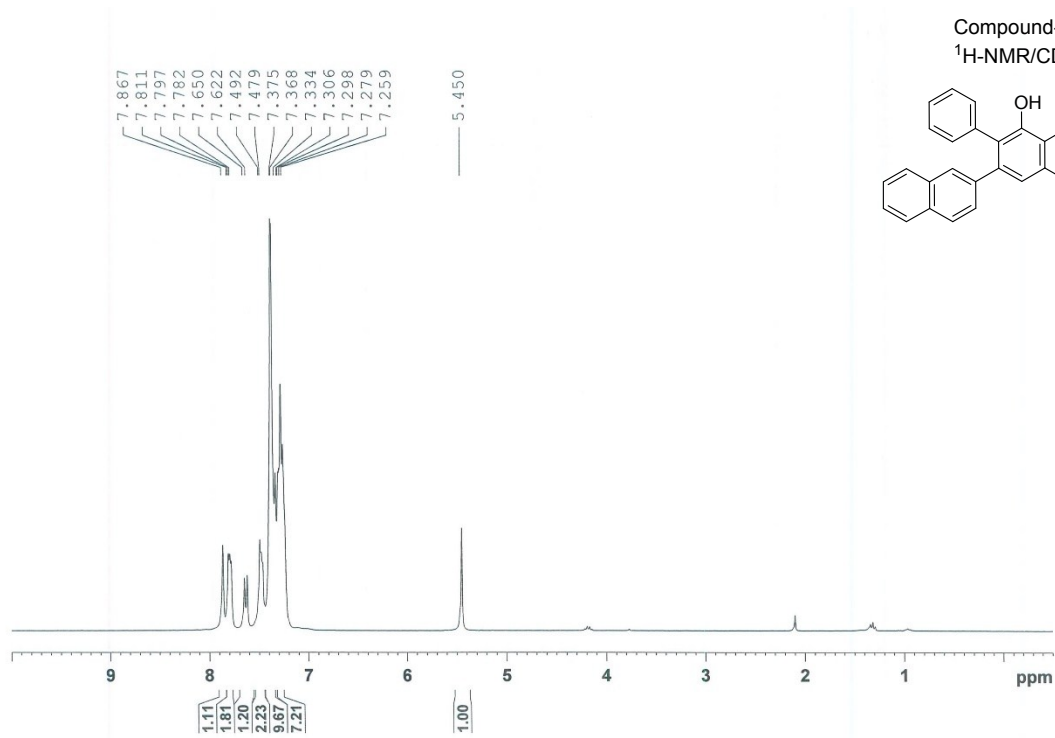


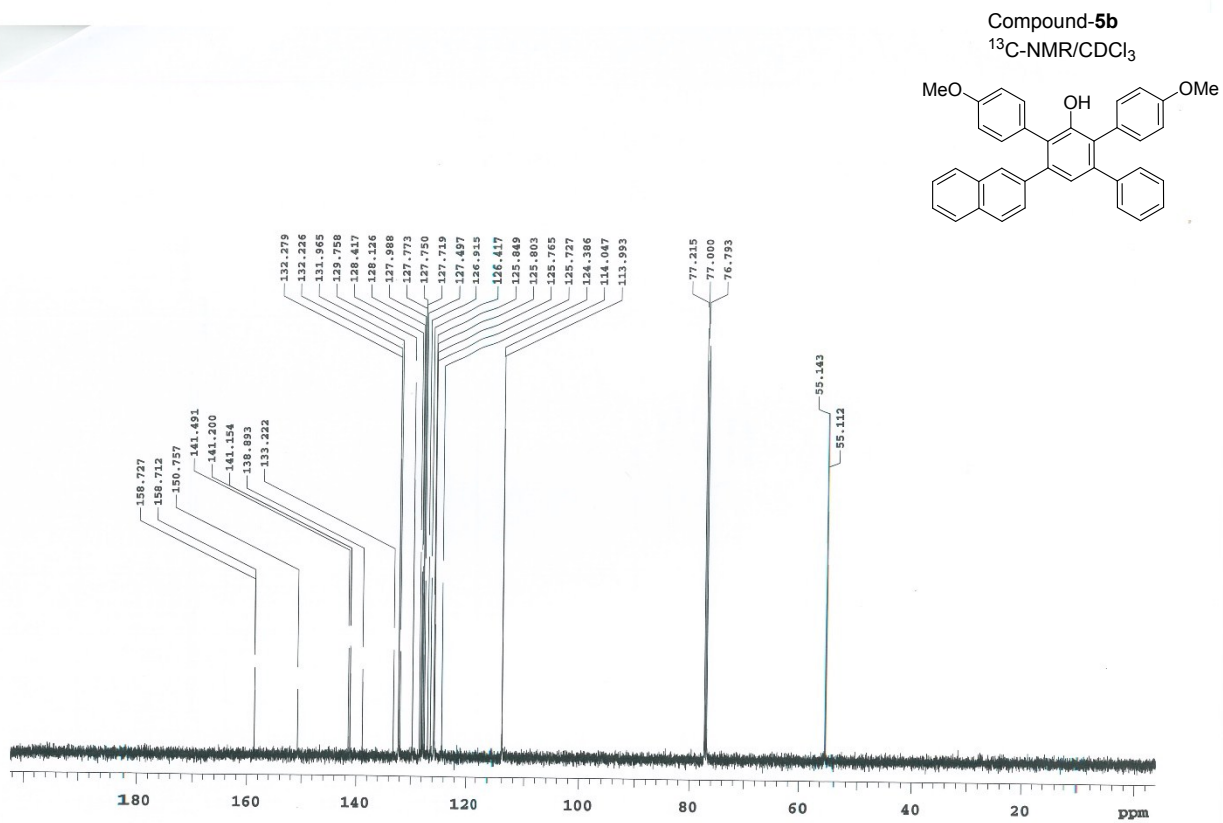
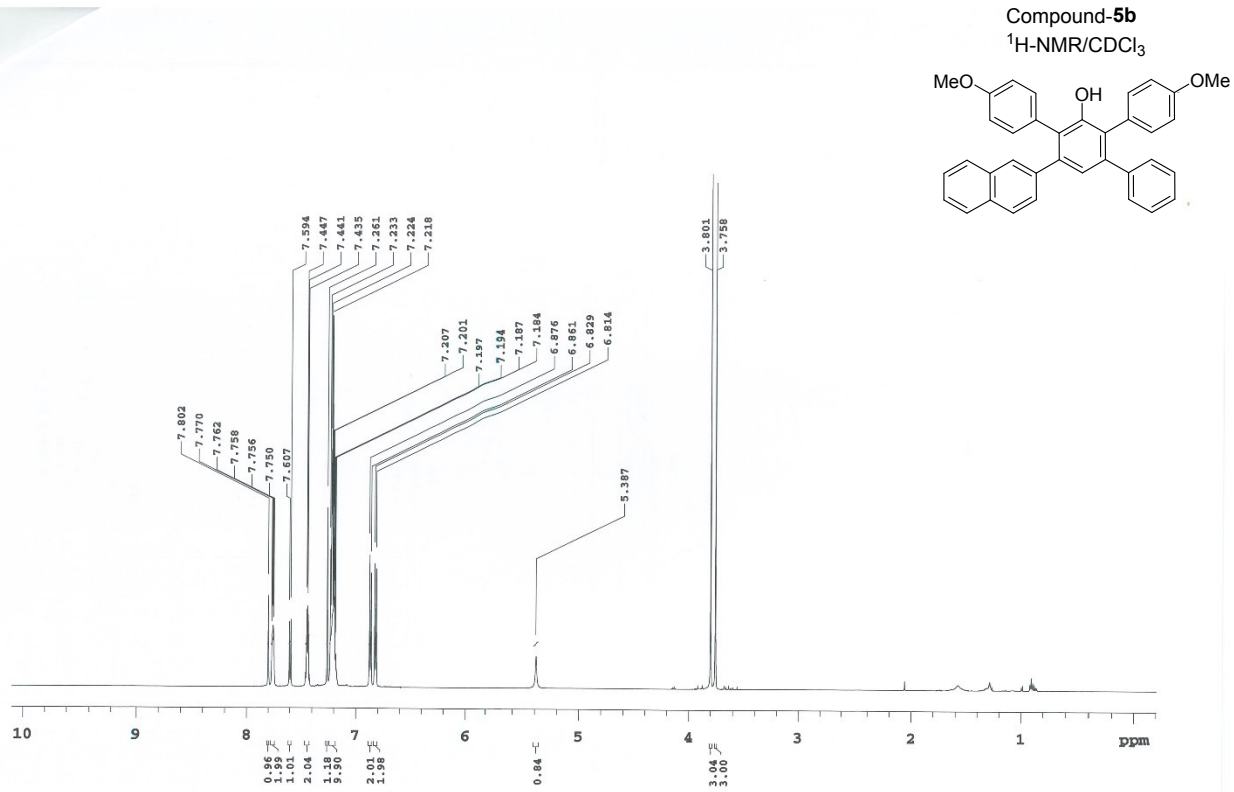




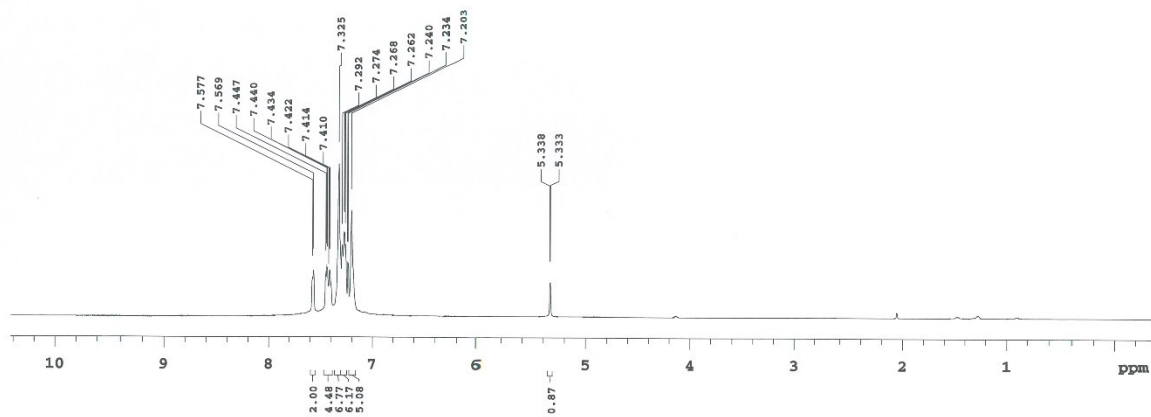
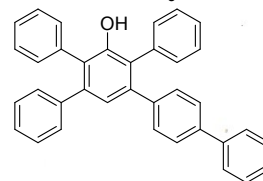




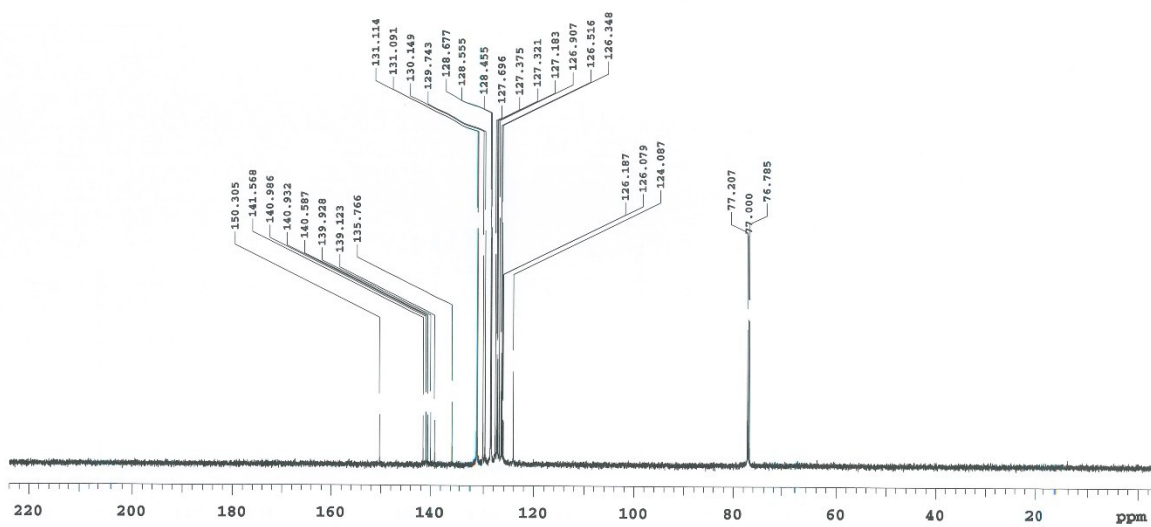
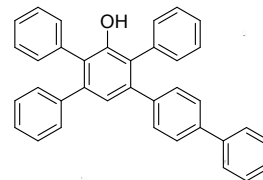


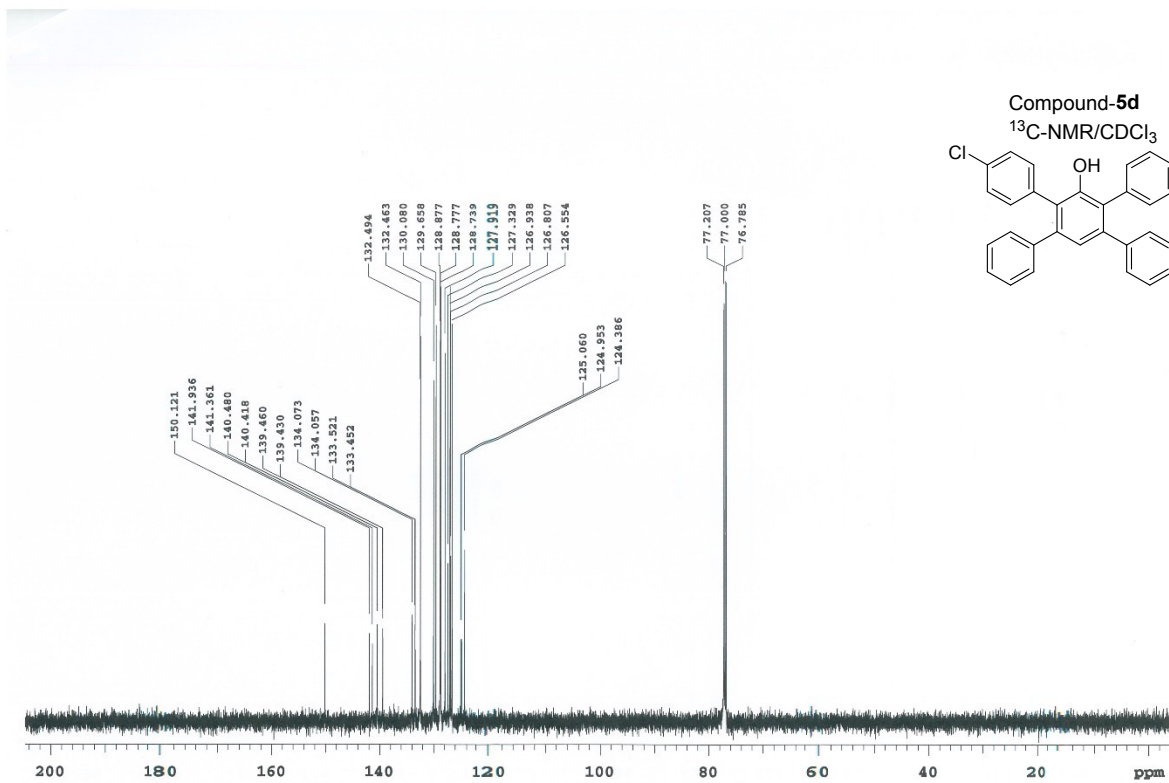
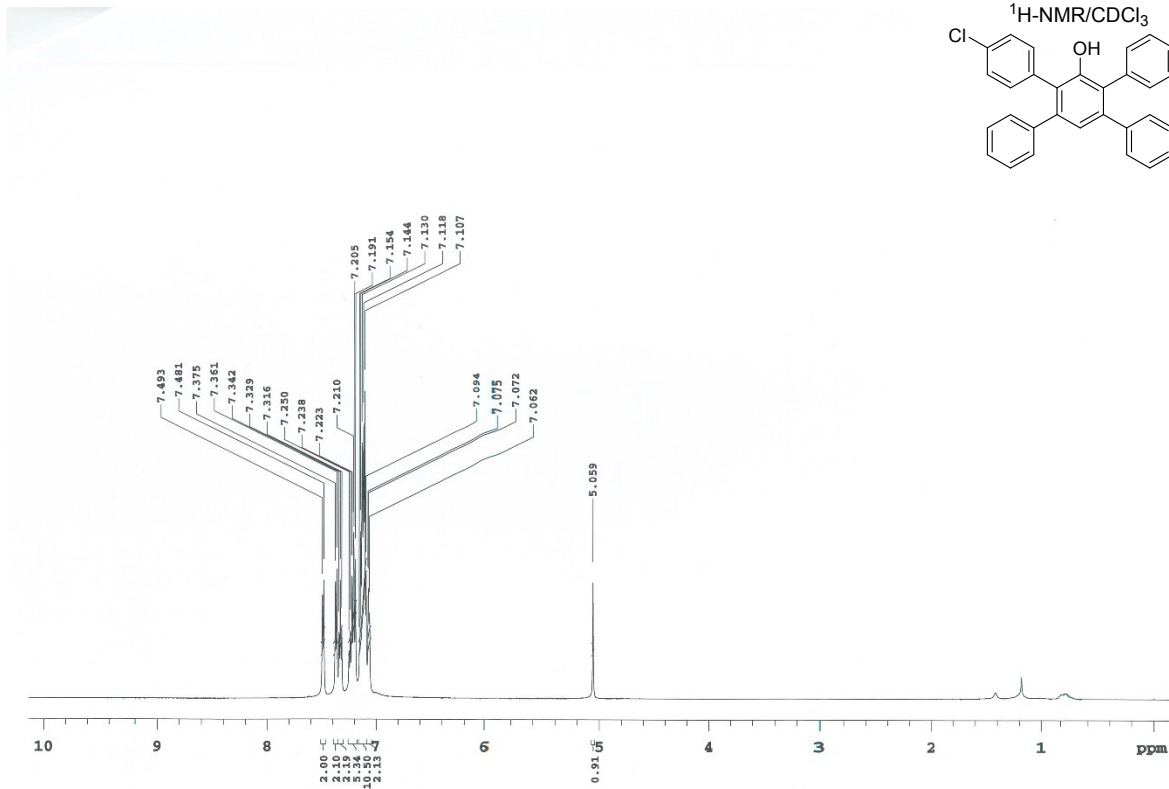


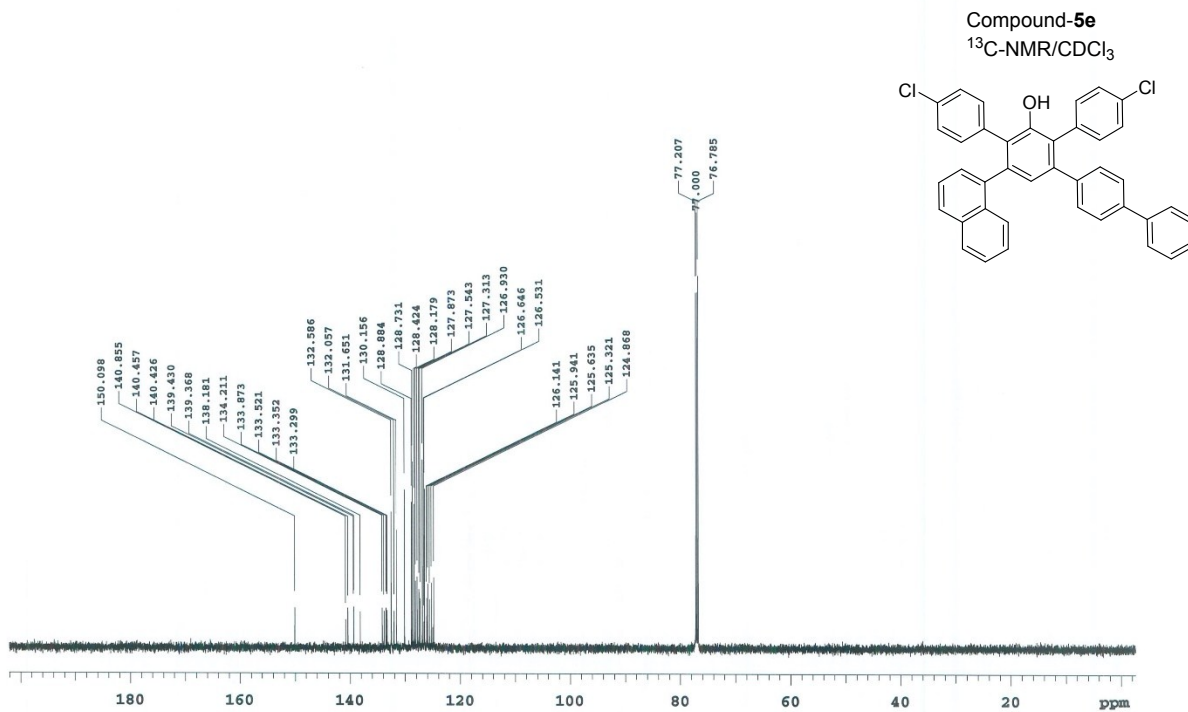
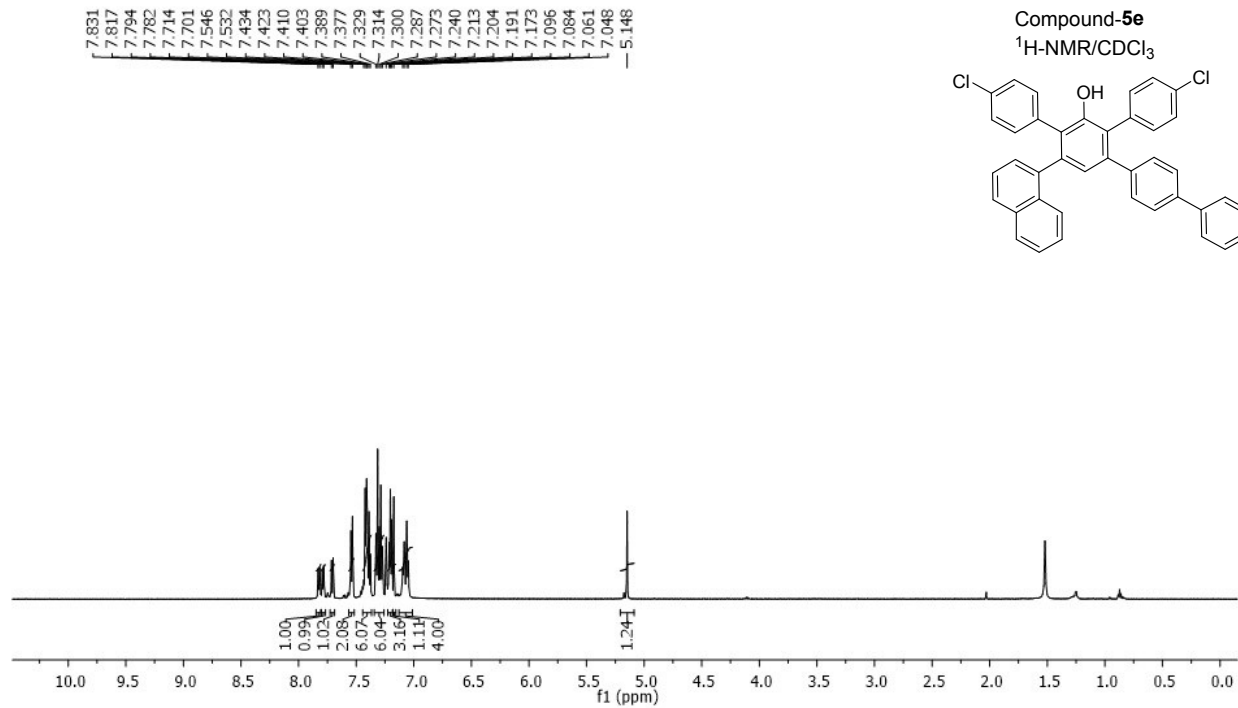
Compound-5c
¹H-NMR/CDCl₃

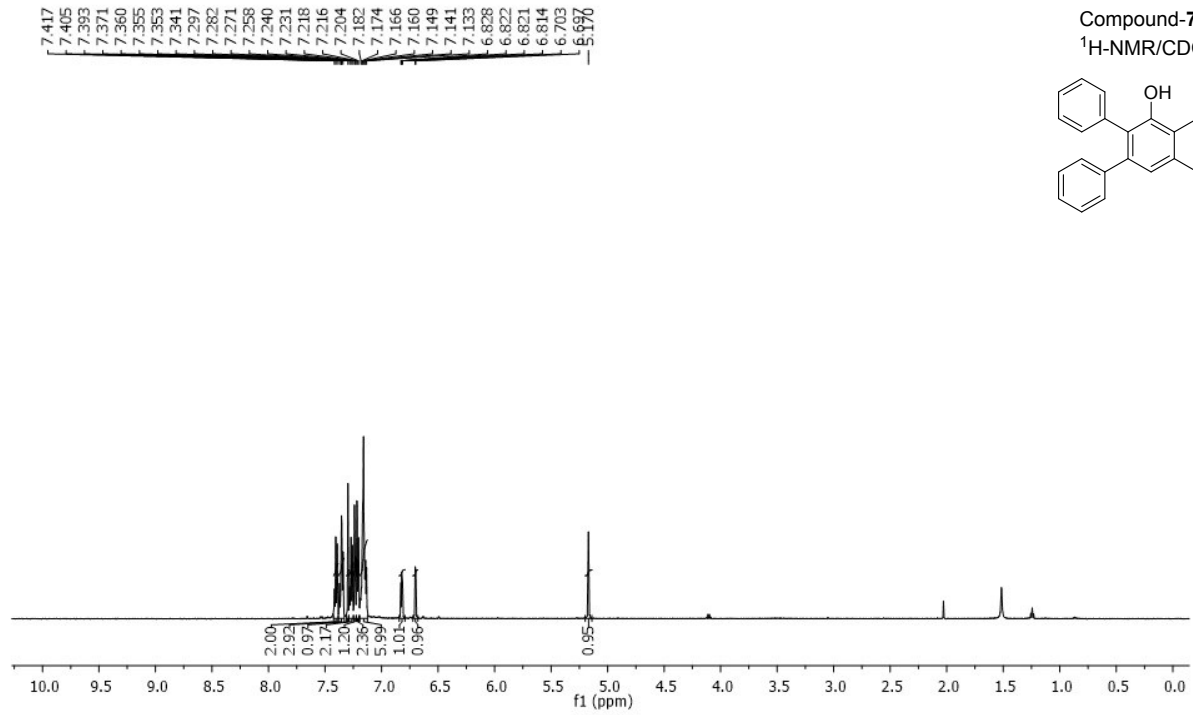


Compound-5c
¹³C-NMR/CDCl₃

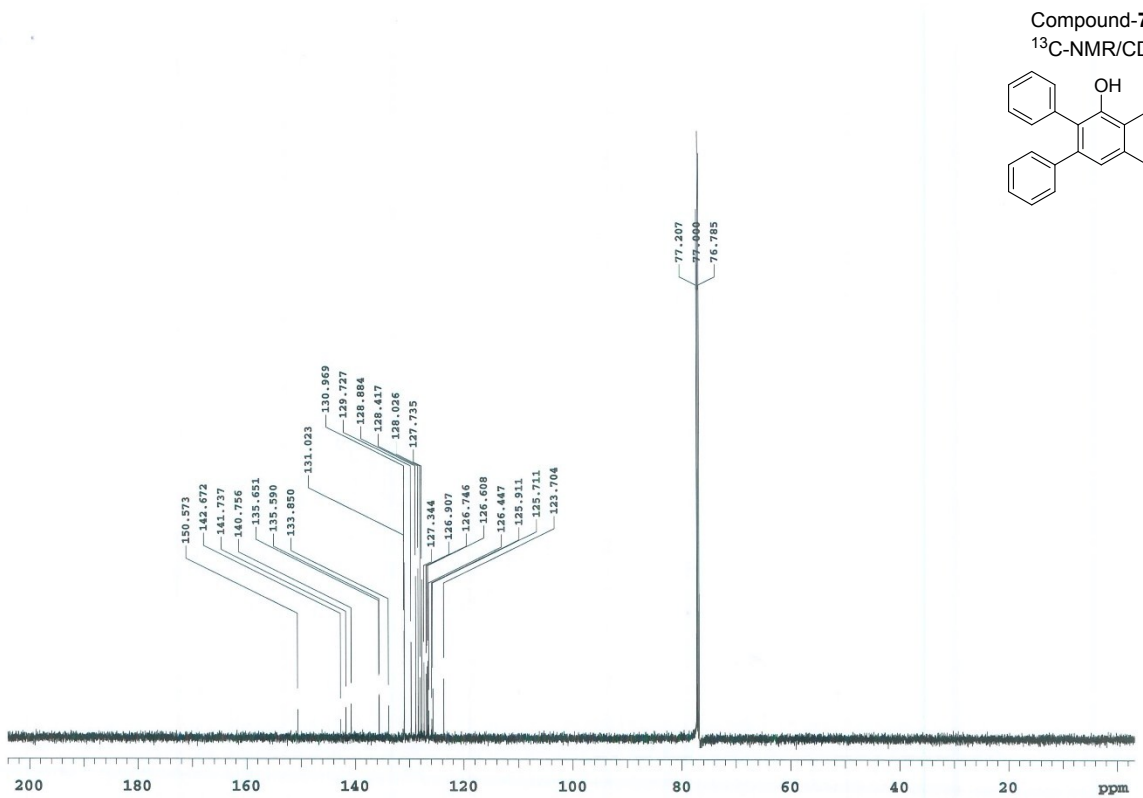
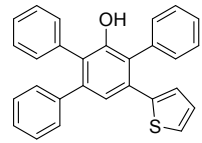




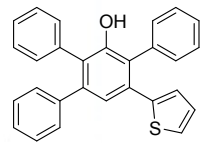


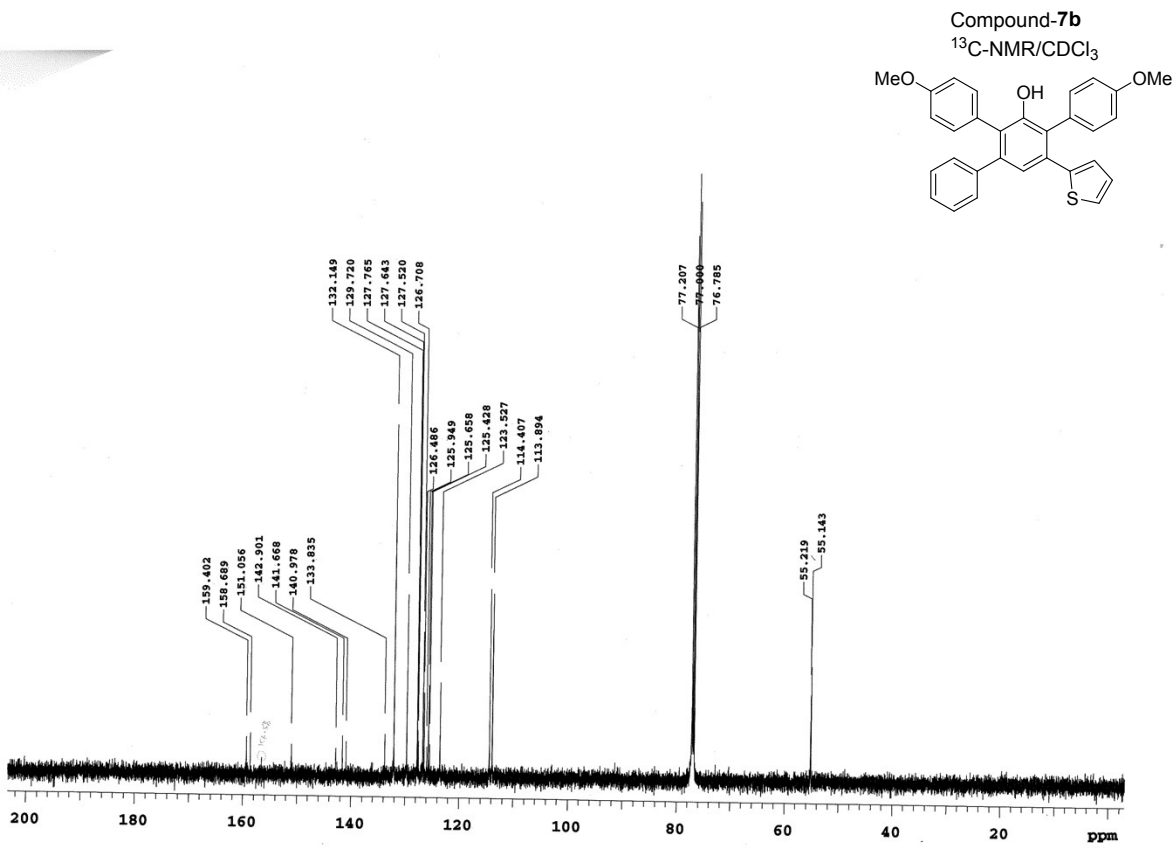
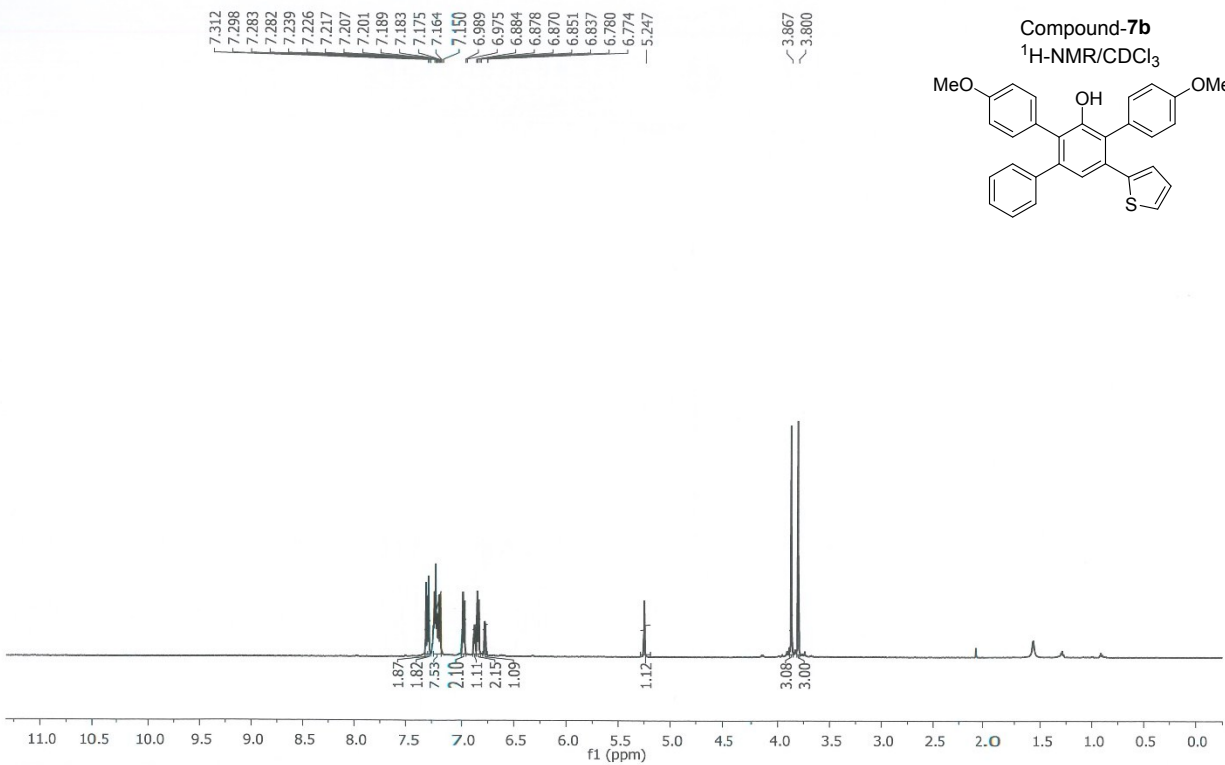


Compound-7a
¹H-NMR/CDCl₃

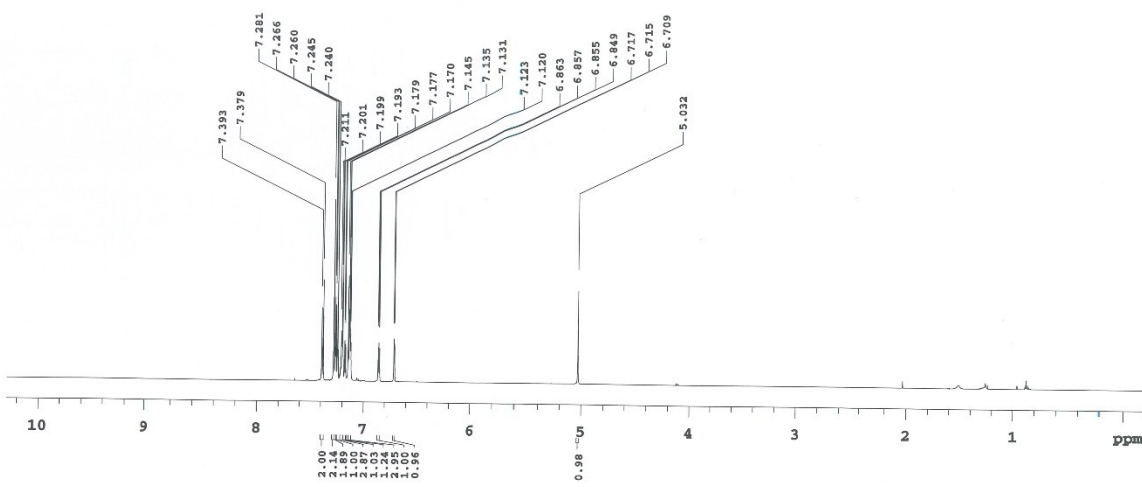
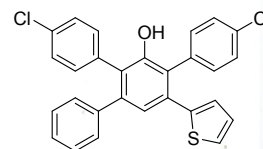


Compound-7a
¹³C-NMR/CDCl₃

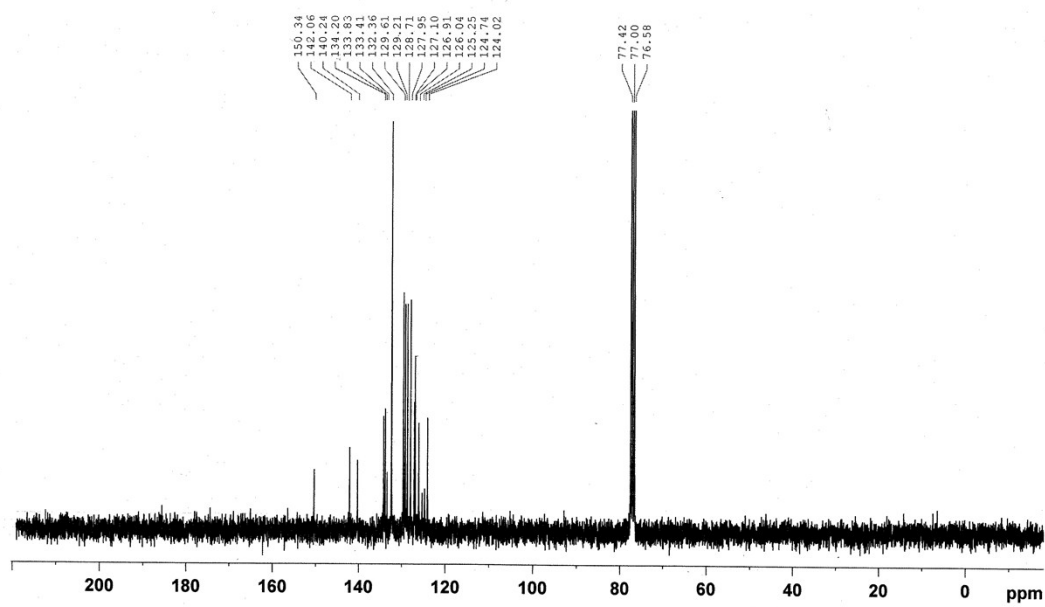
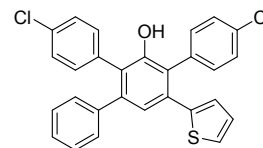


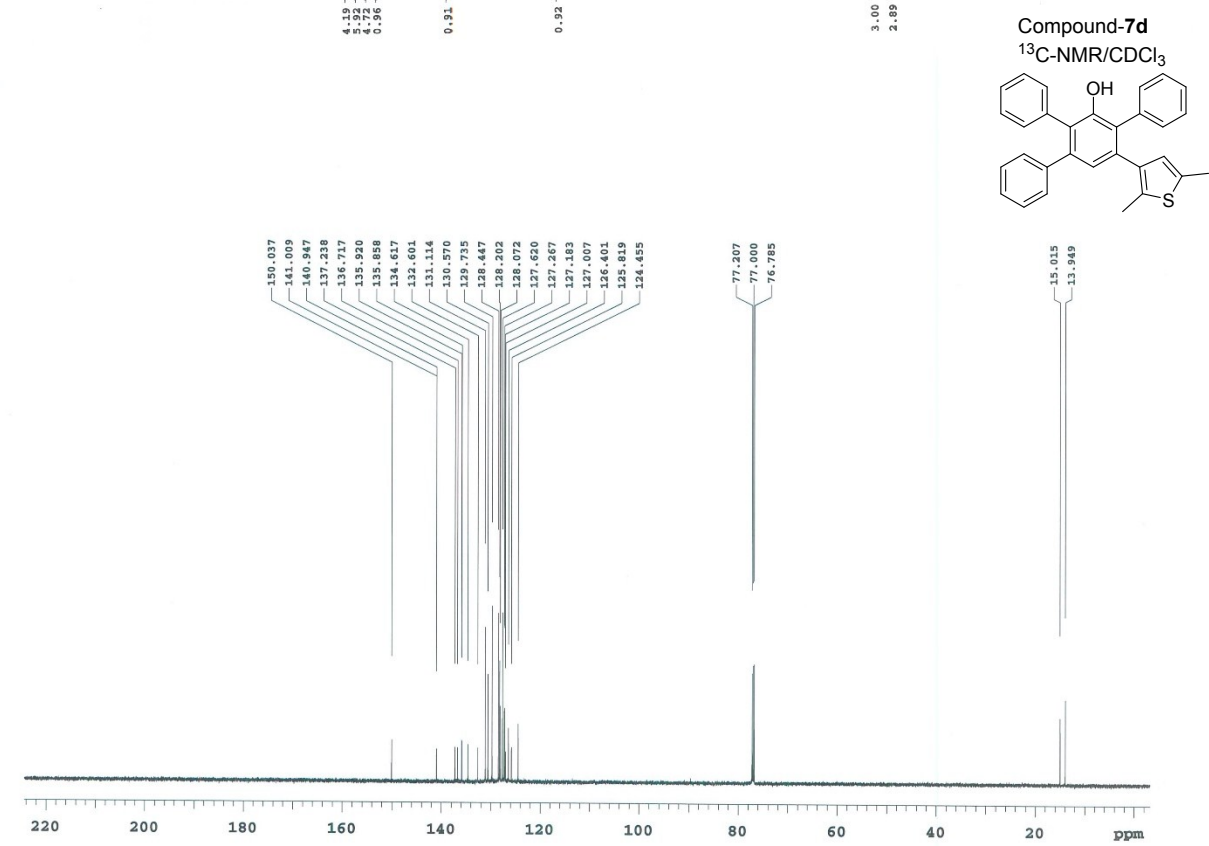
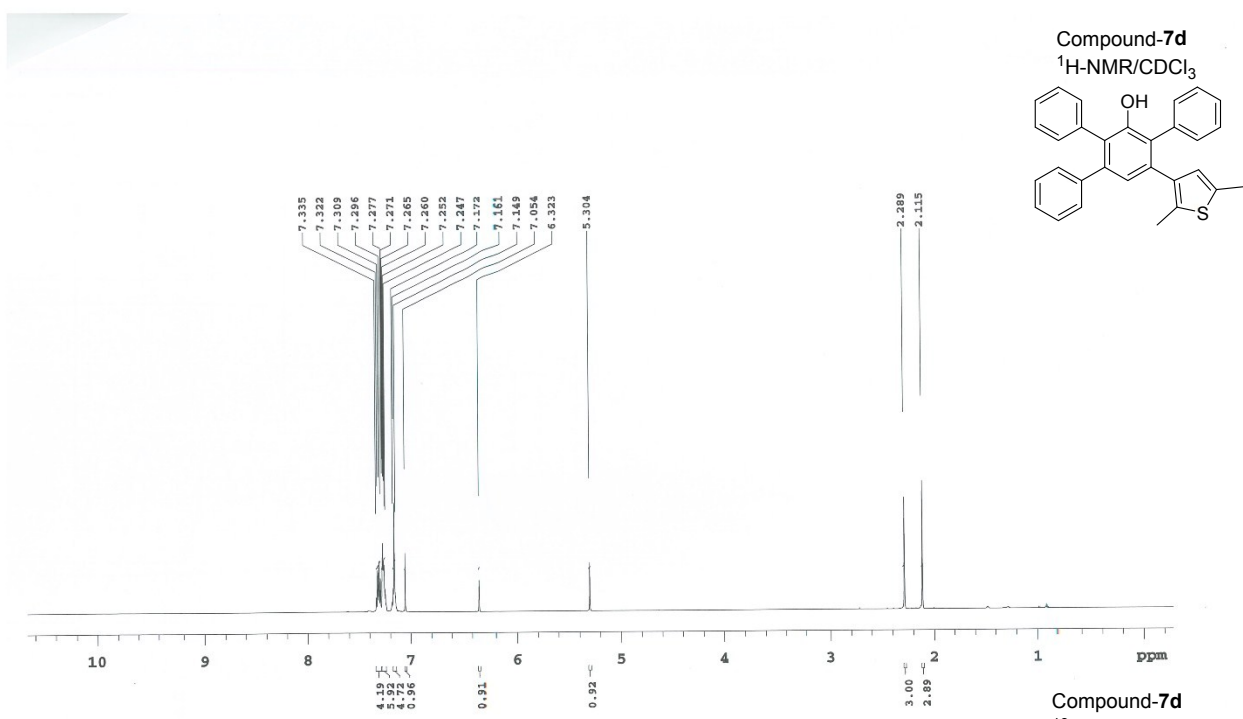


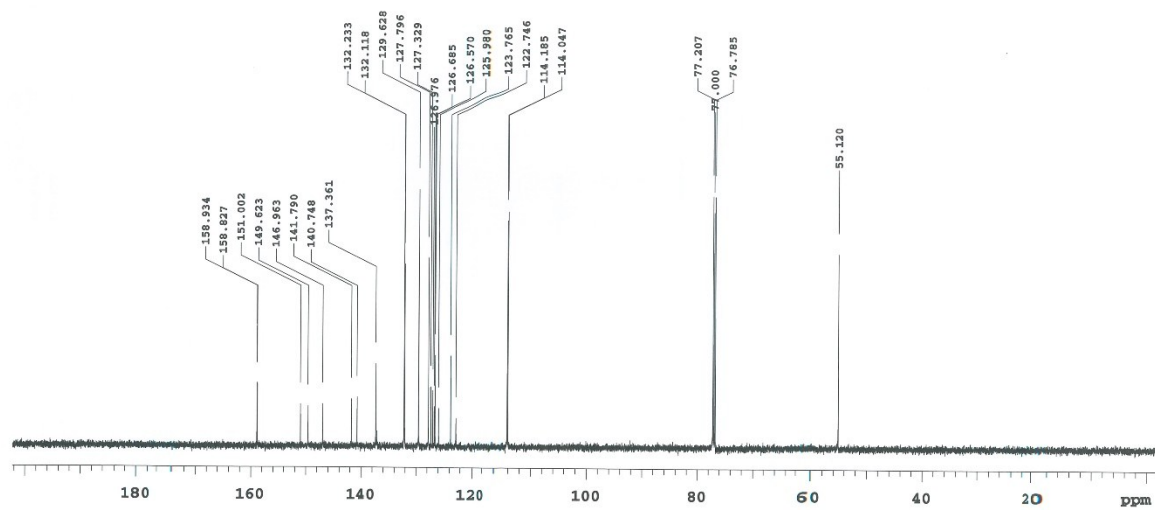
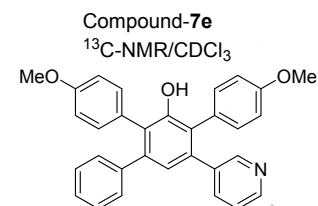
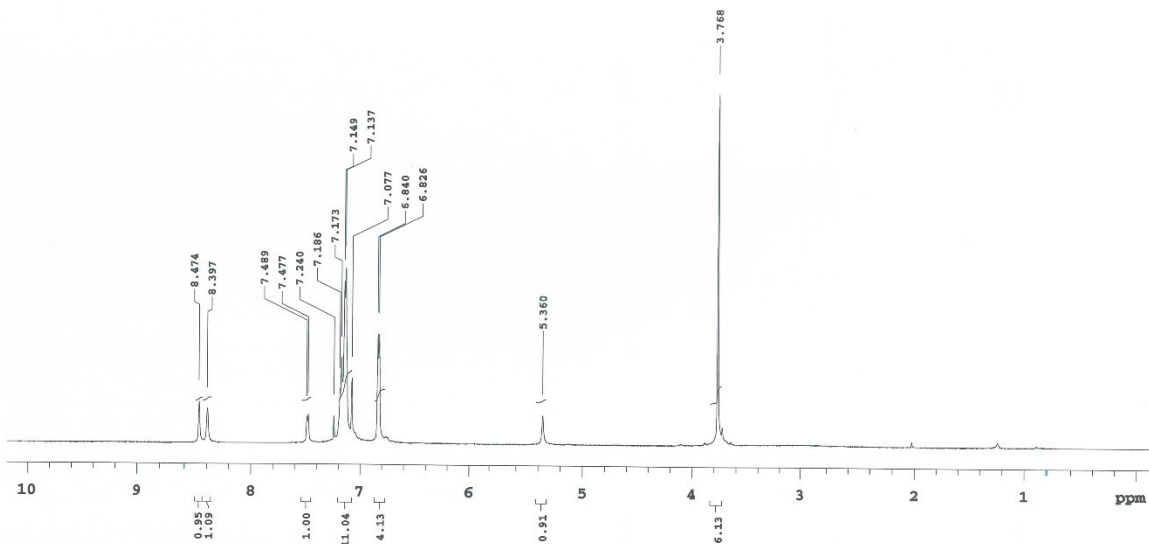
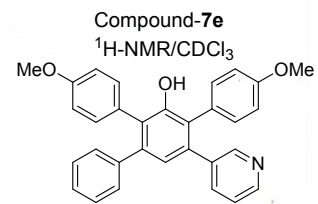
Compound-7c
¹³C-NMR/CDCl₃

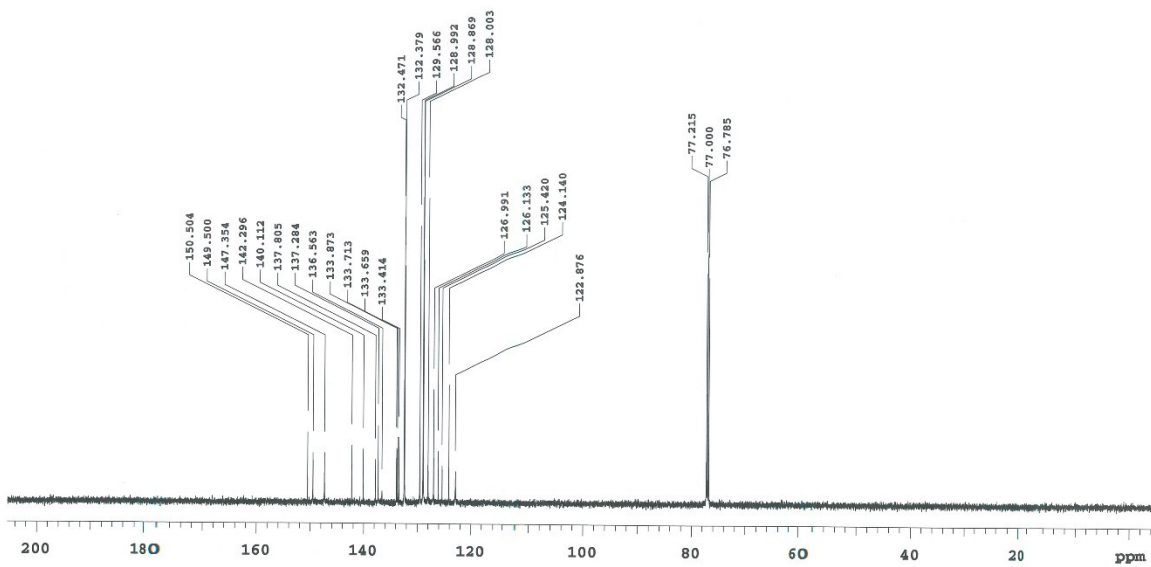
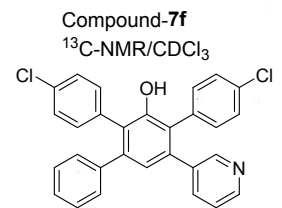
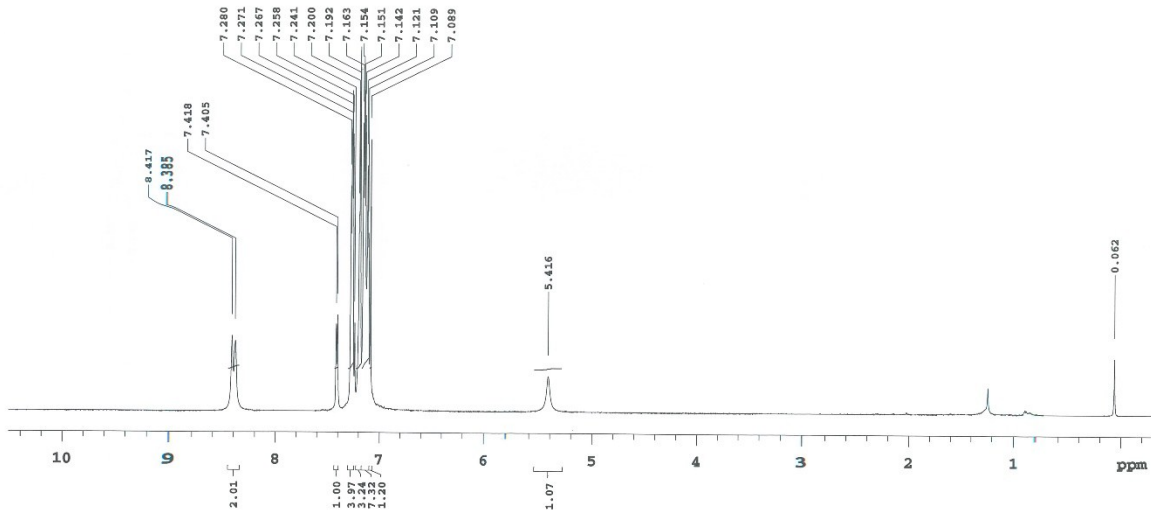
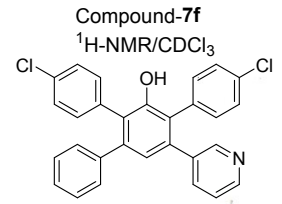


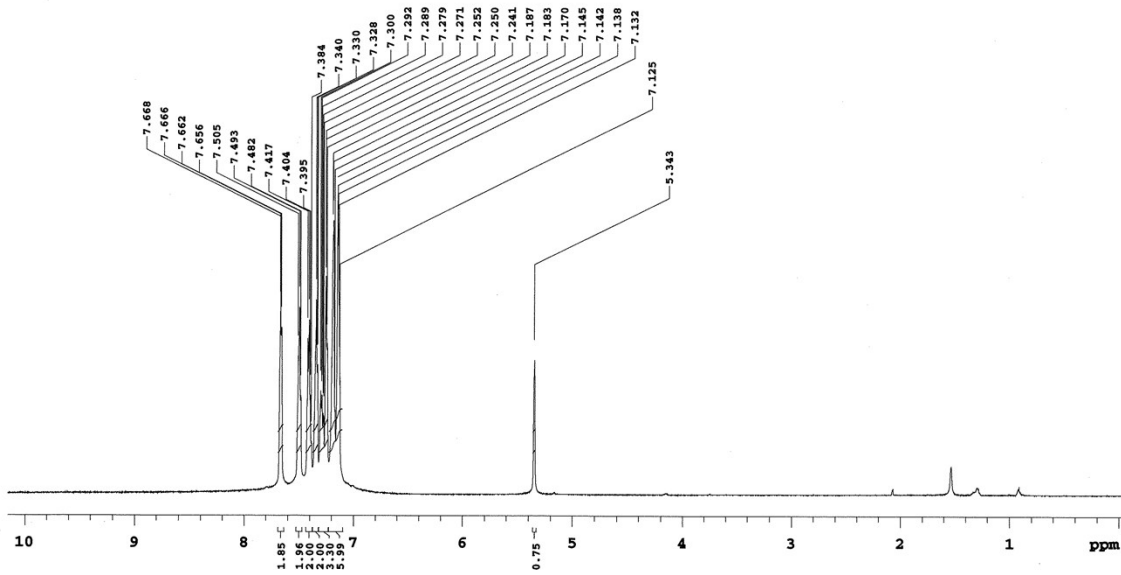
Compound-7c
¹H-NMR/CDCl₃



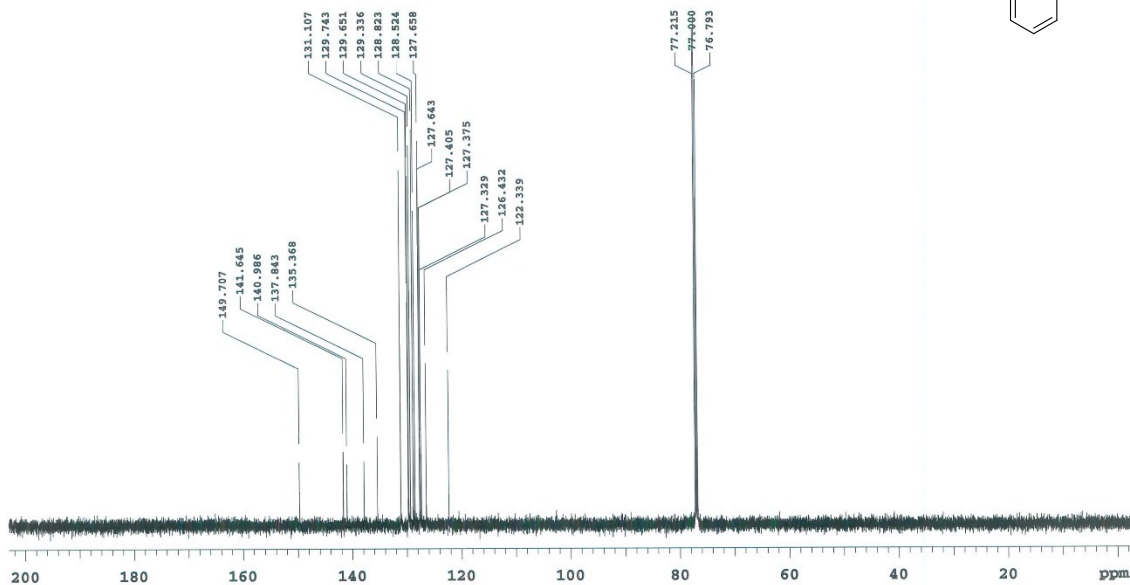
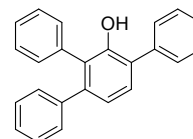


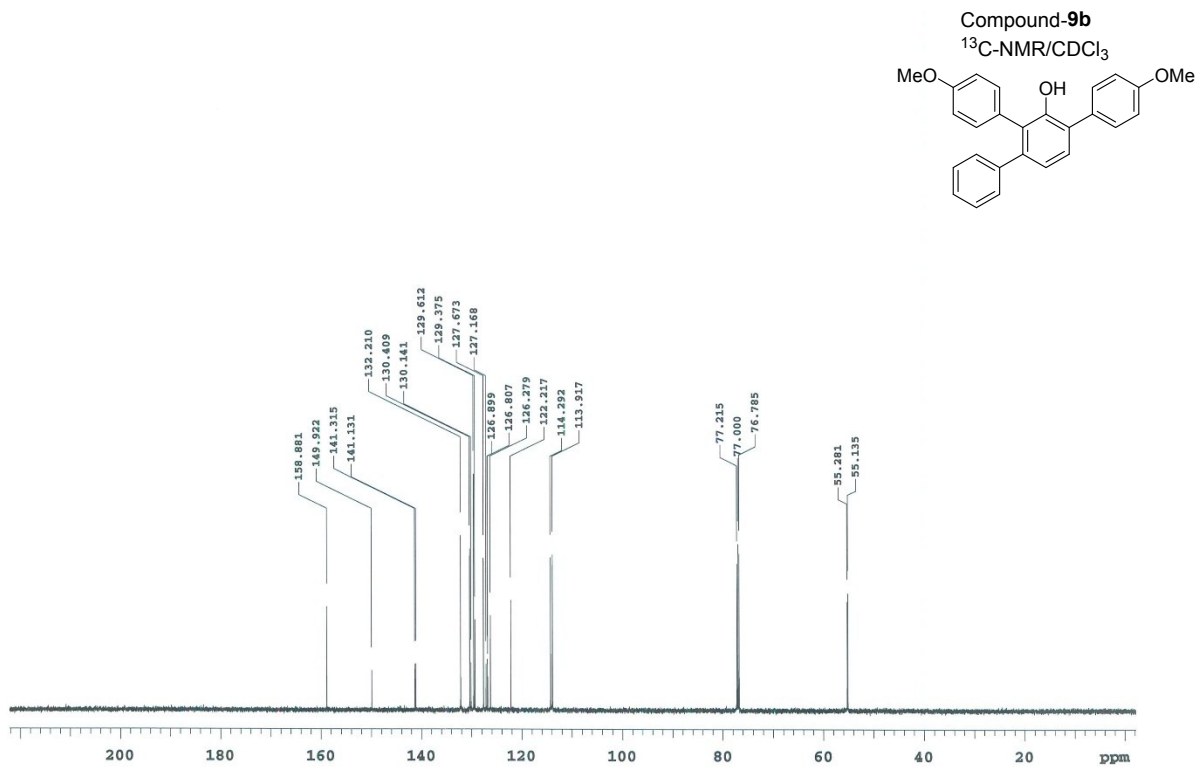
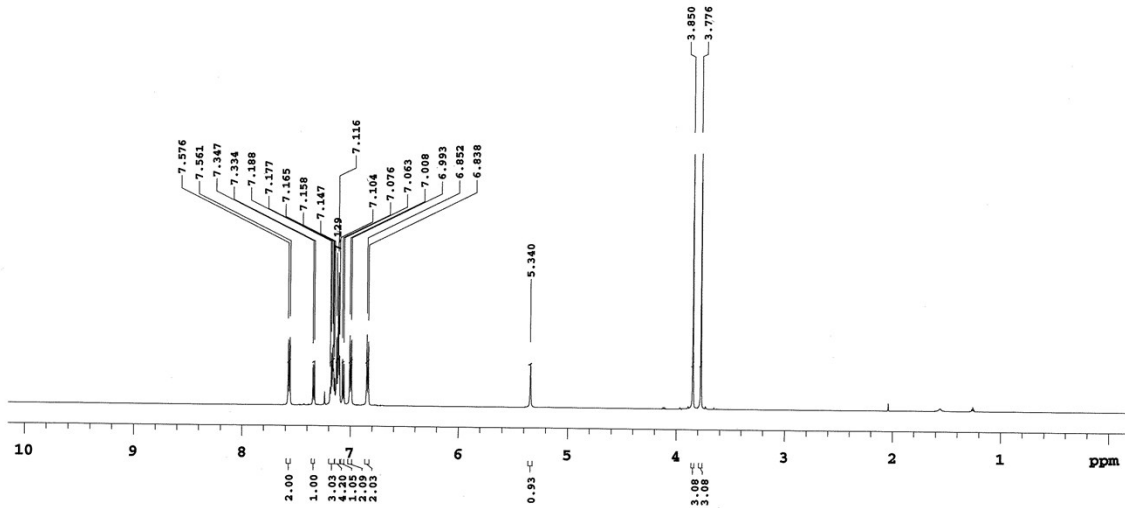


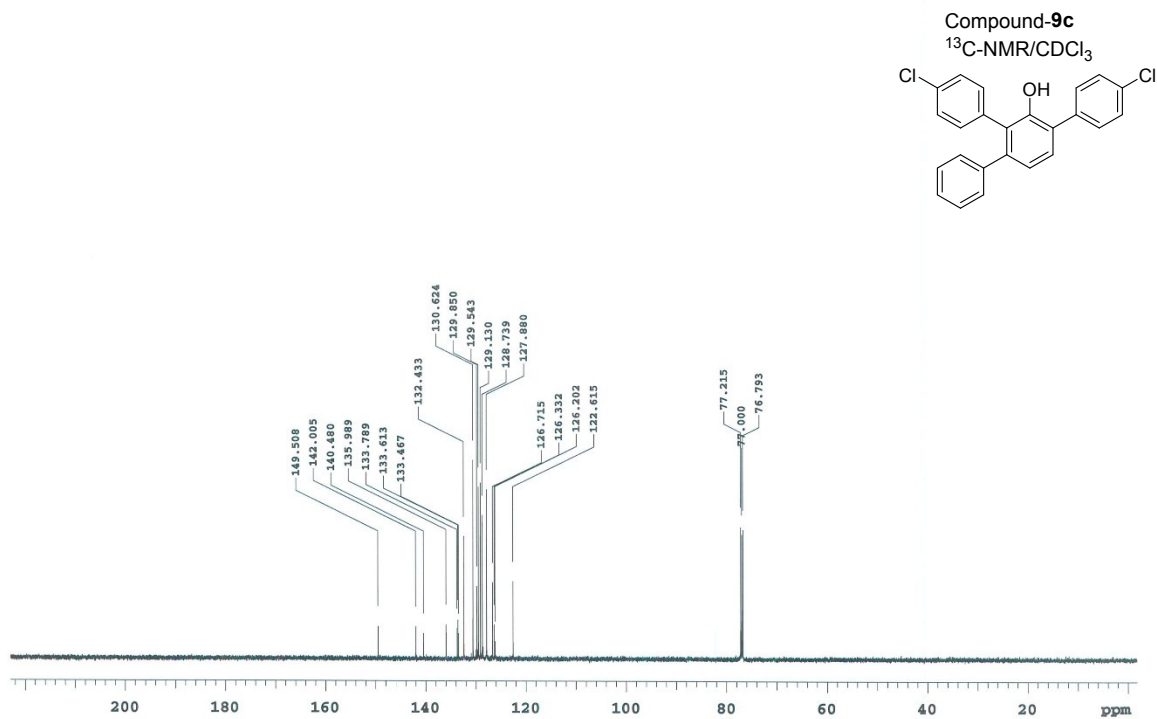
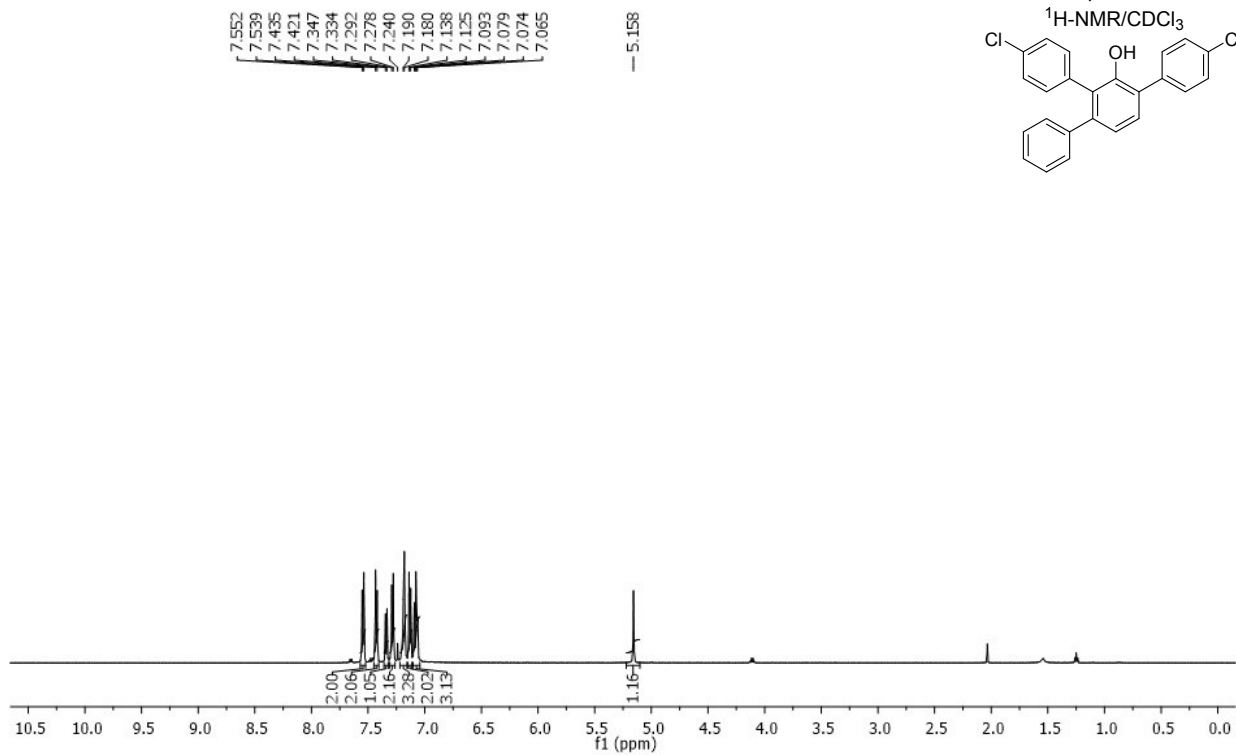




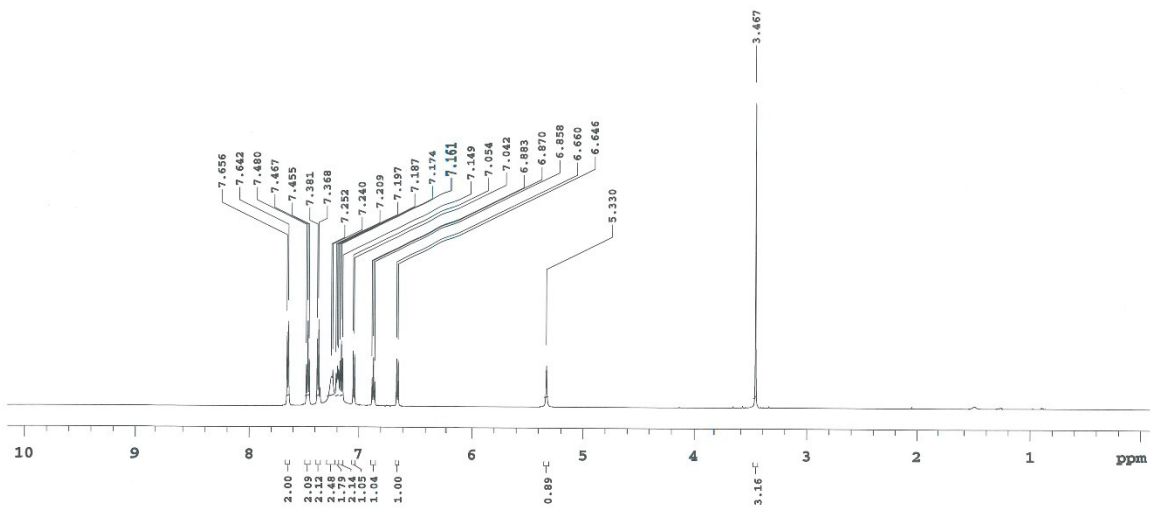
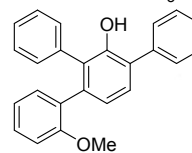
Compound-9a
¹³C-NMR/CDCl₃



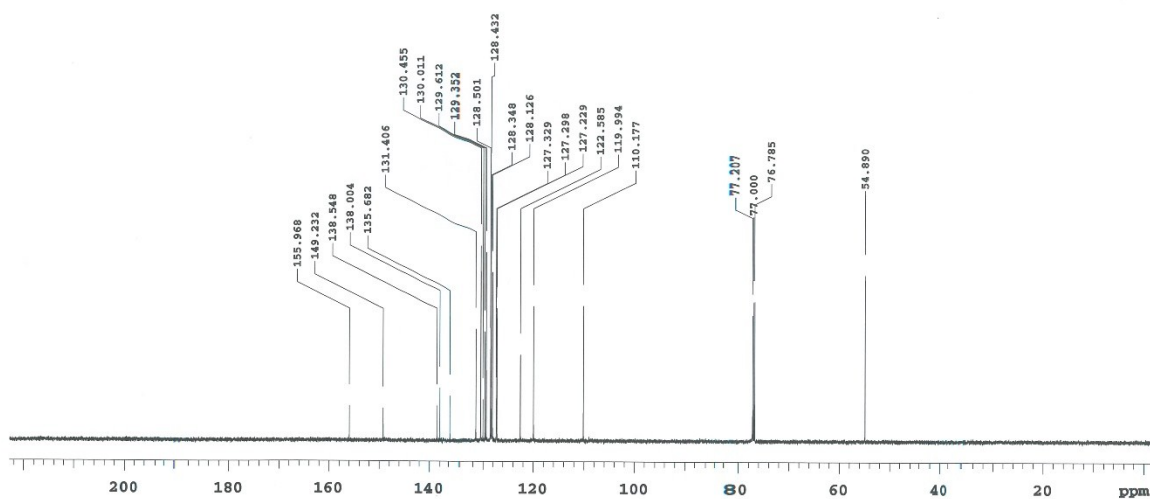
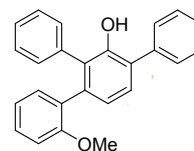


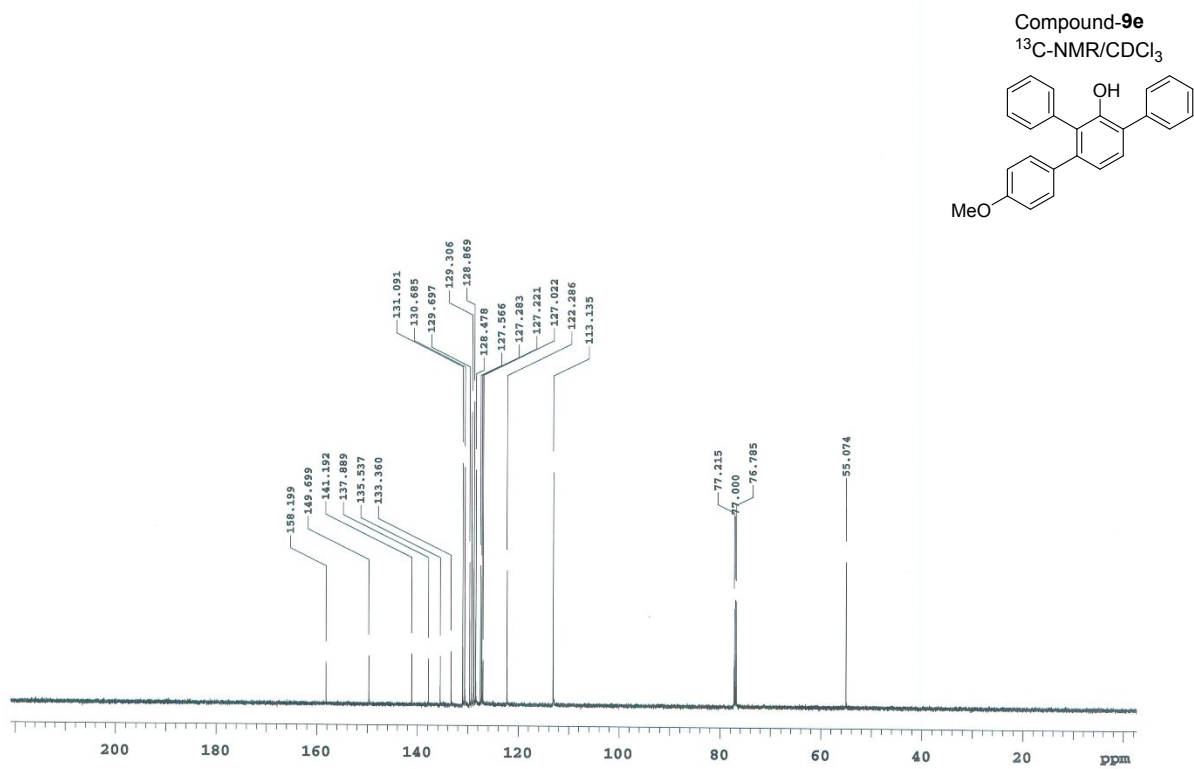
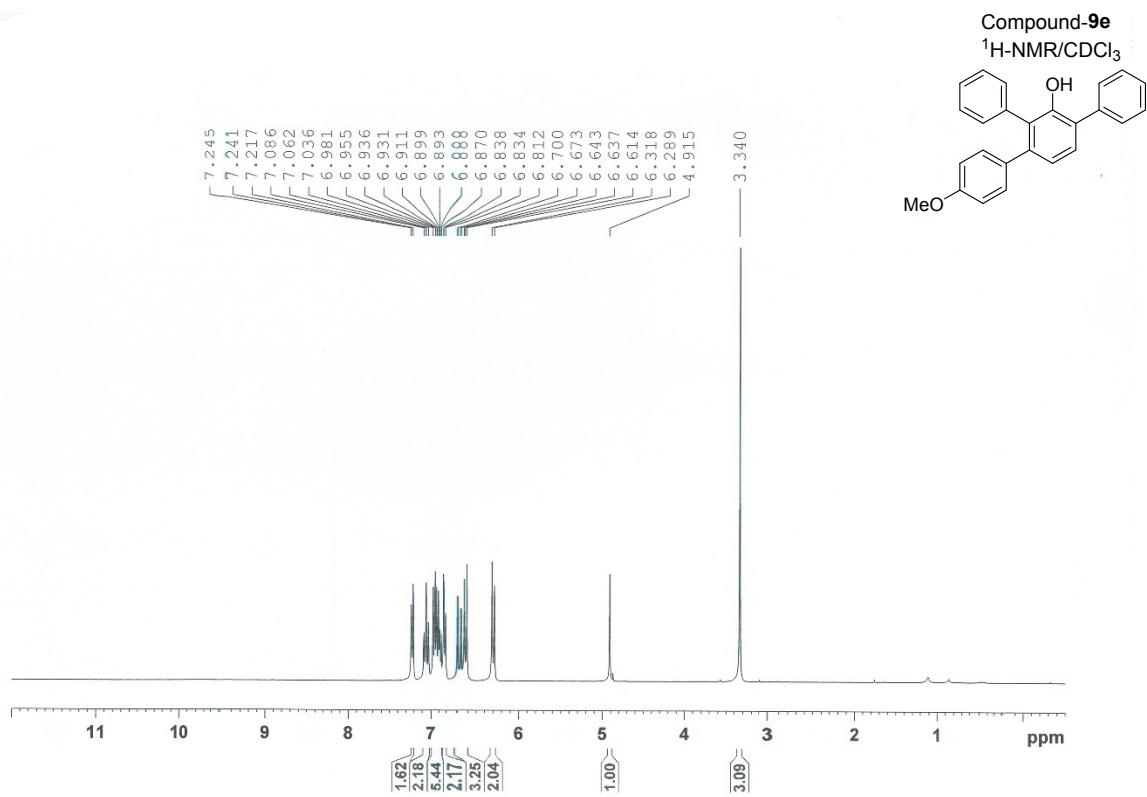


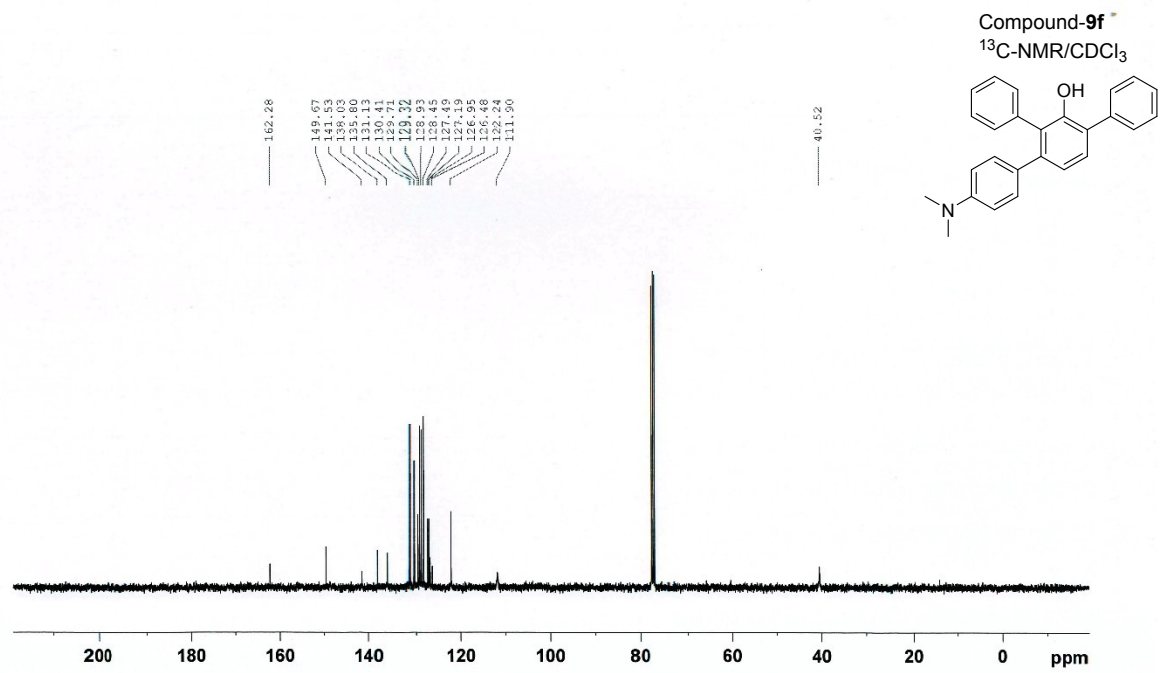
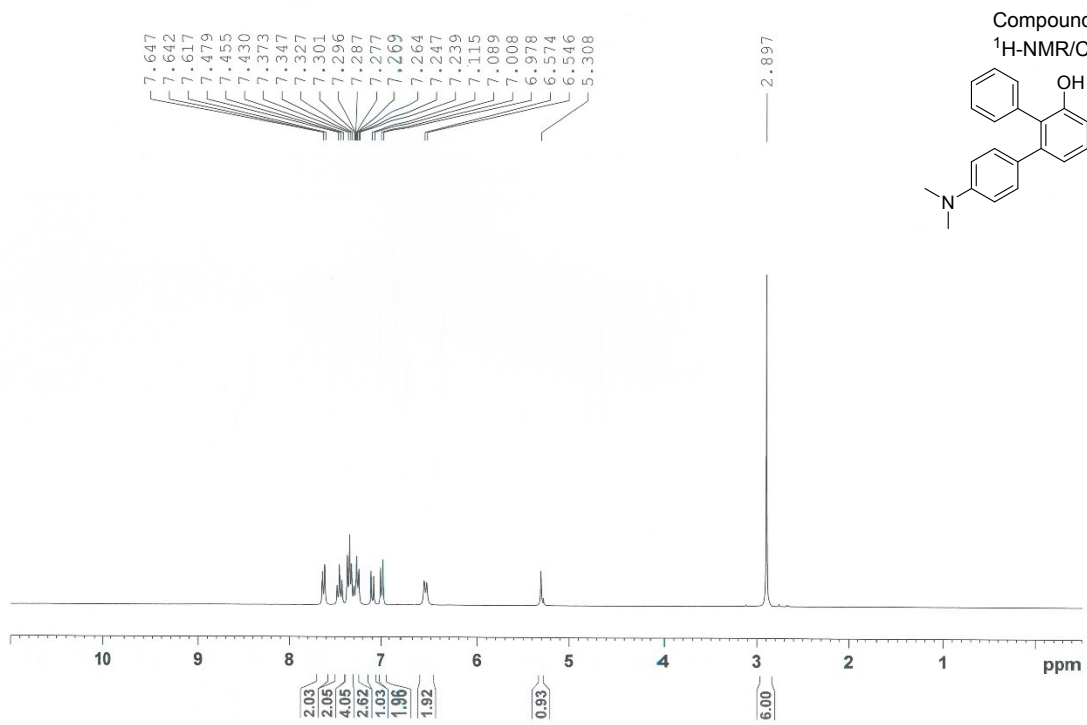
Compound-9d
¹H-NMR/CDCl₃

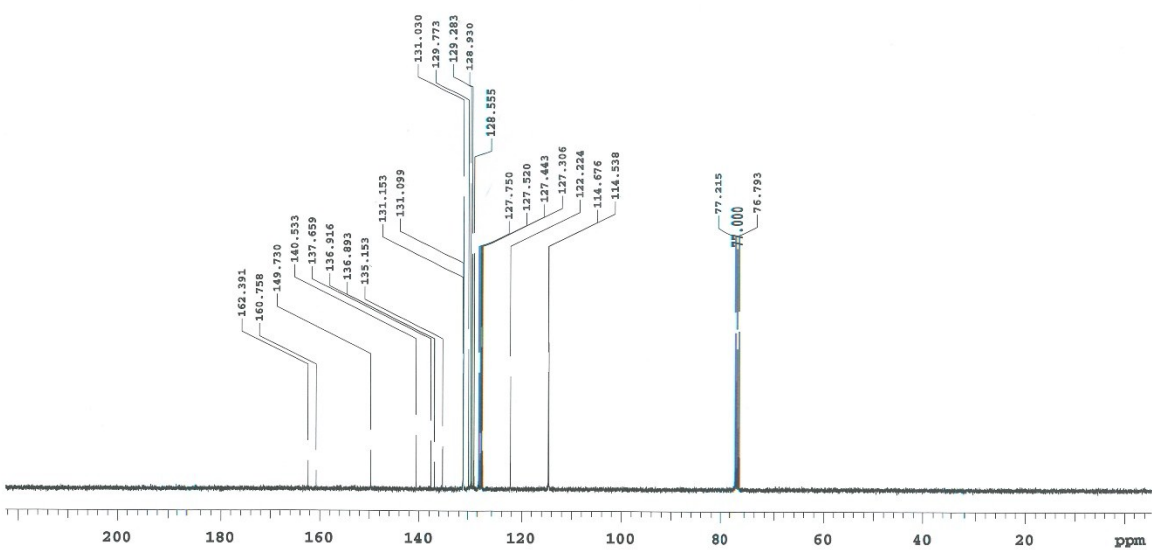
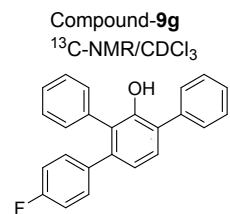
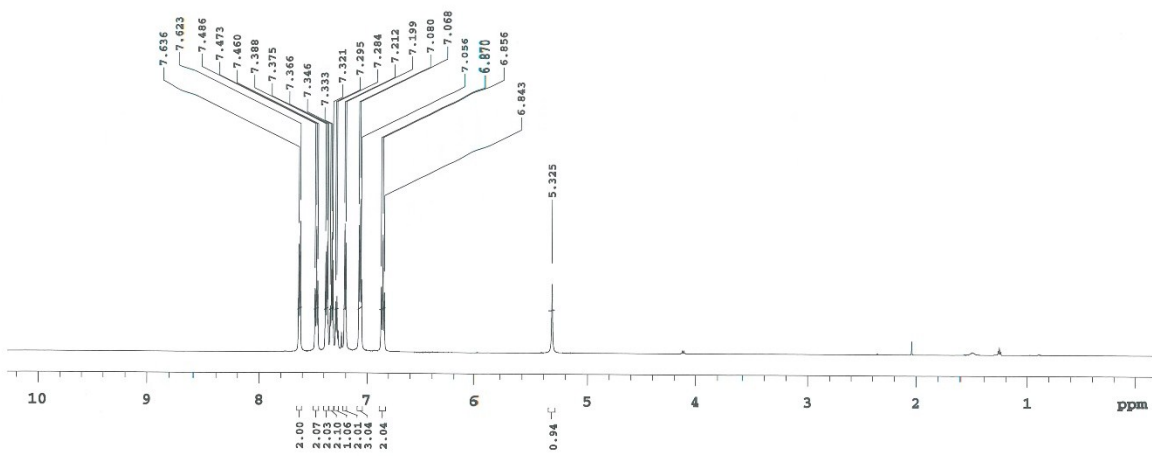
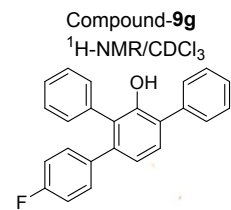


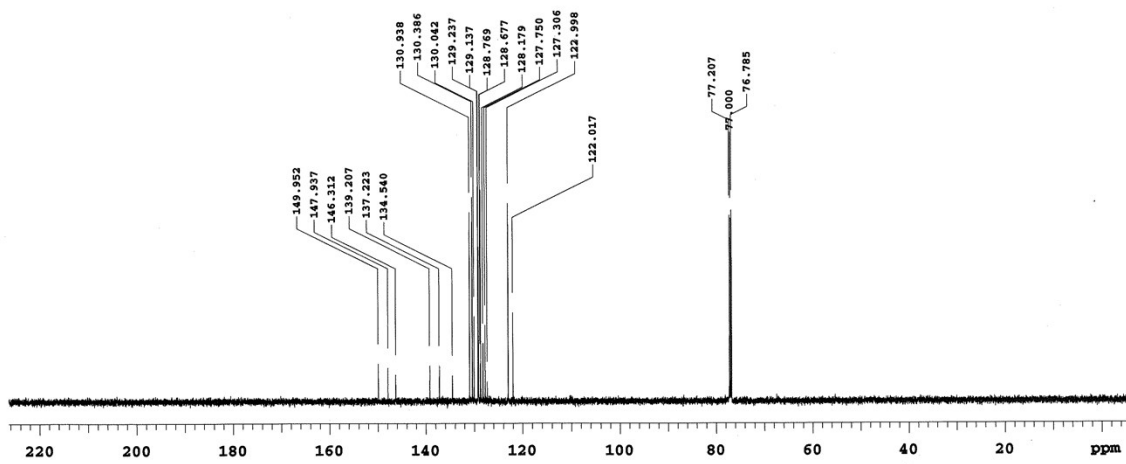
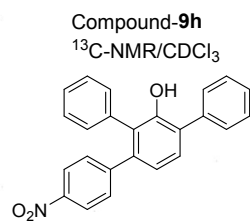
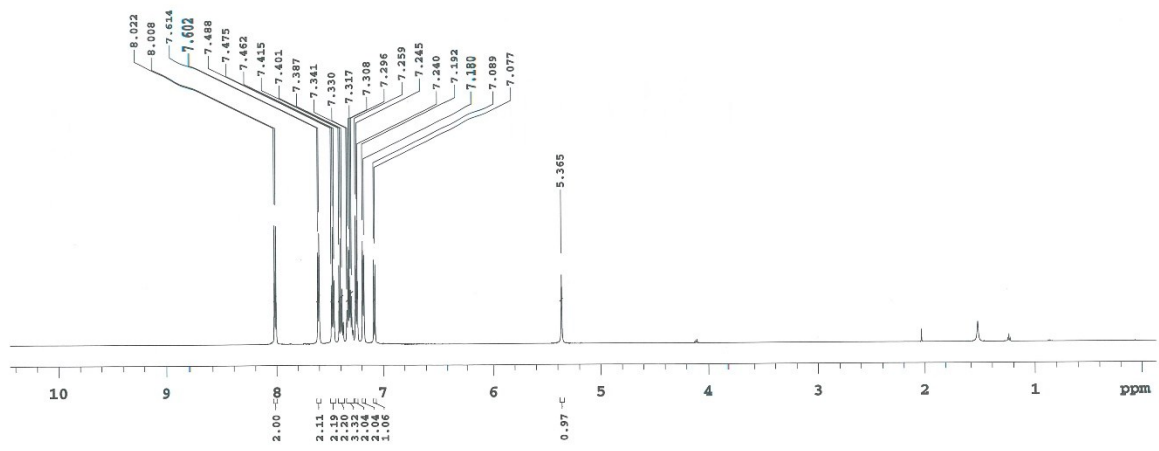
Compound-9d
¹³C-NMR/CDCl₃

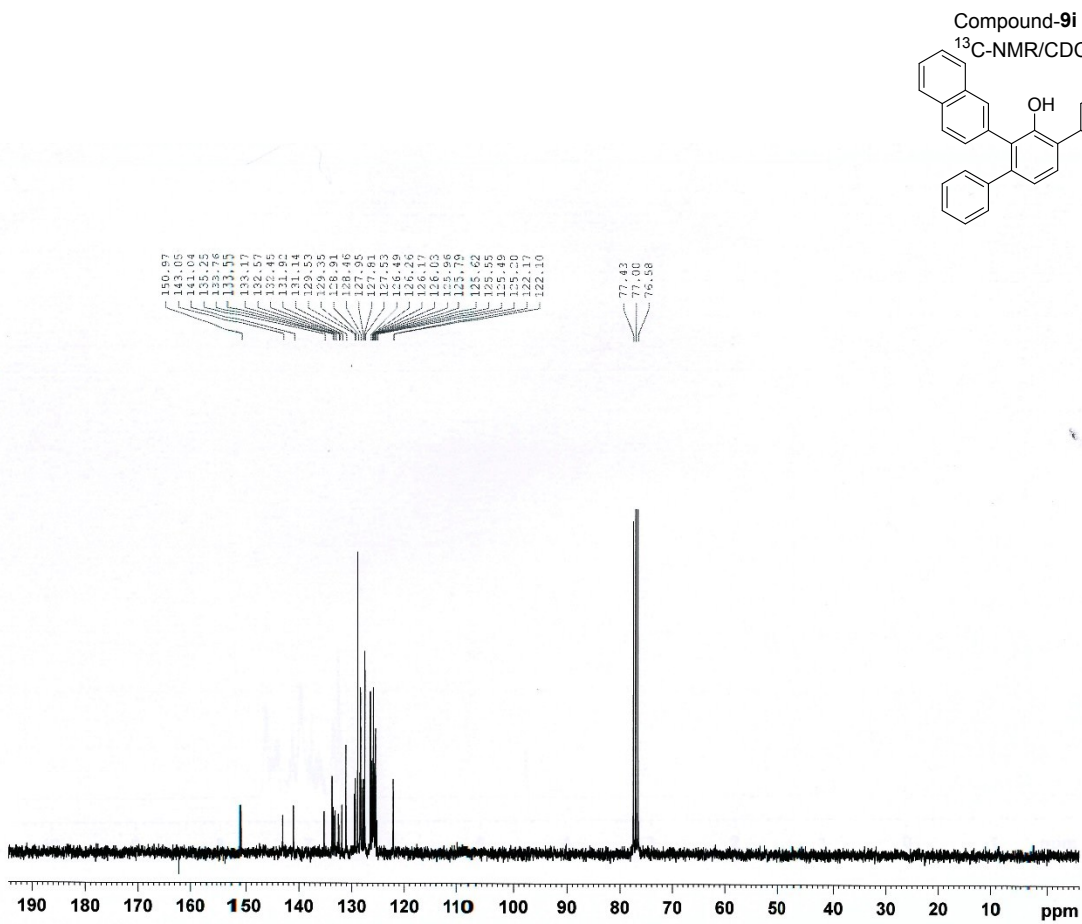
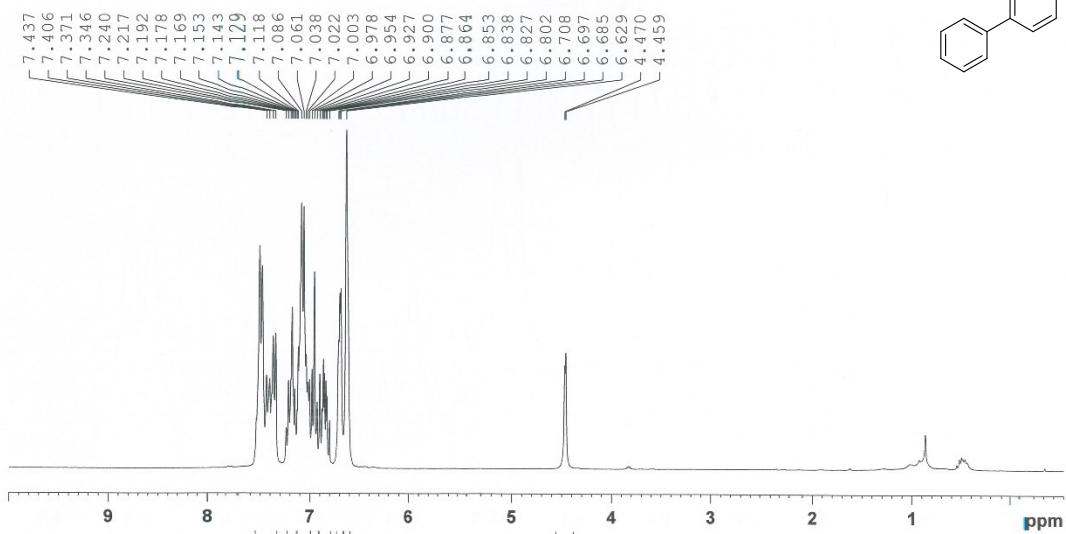




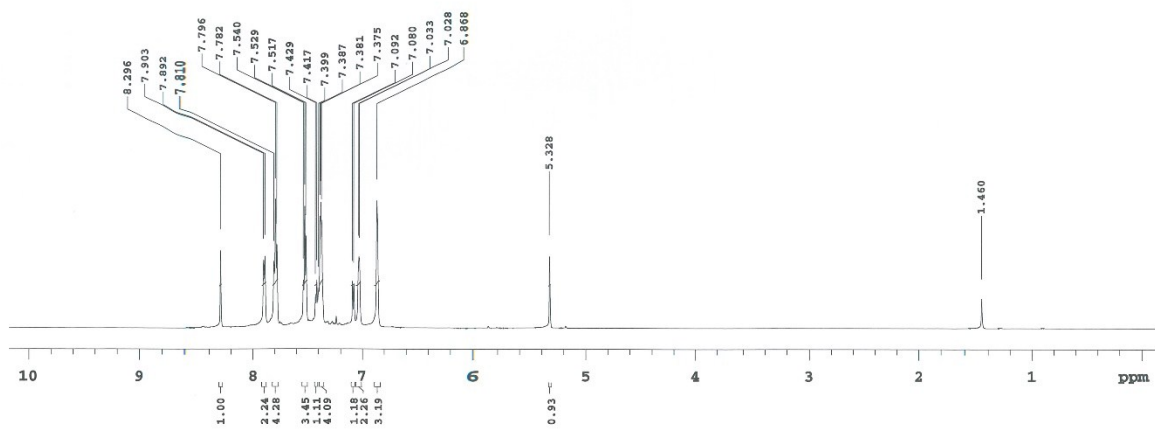
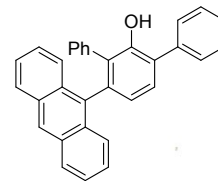




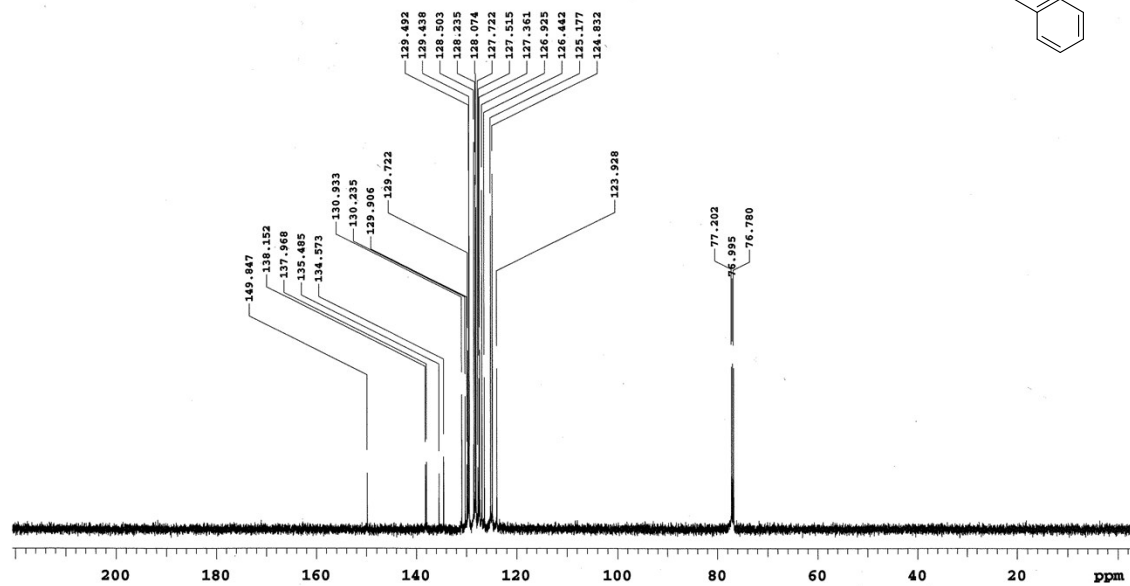
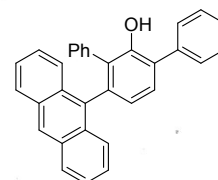


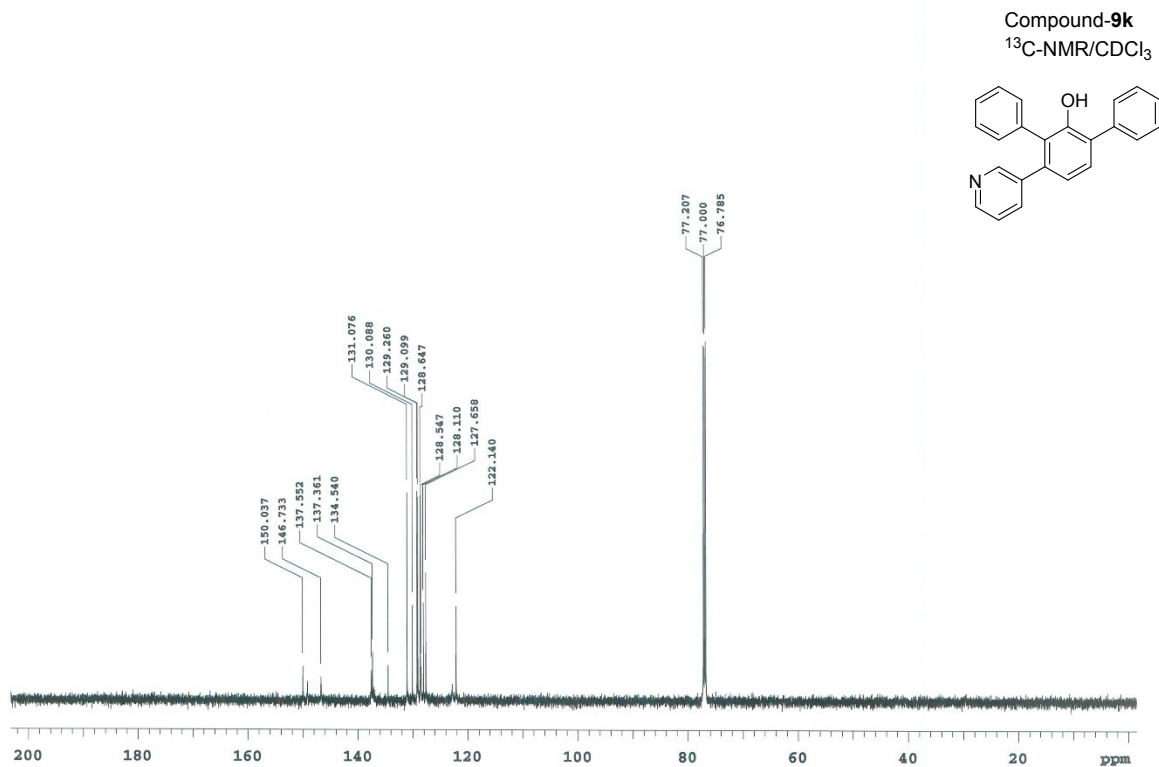
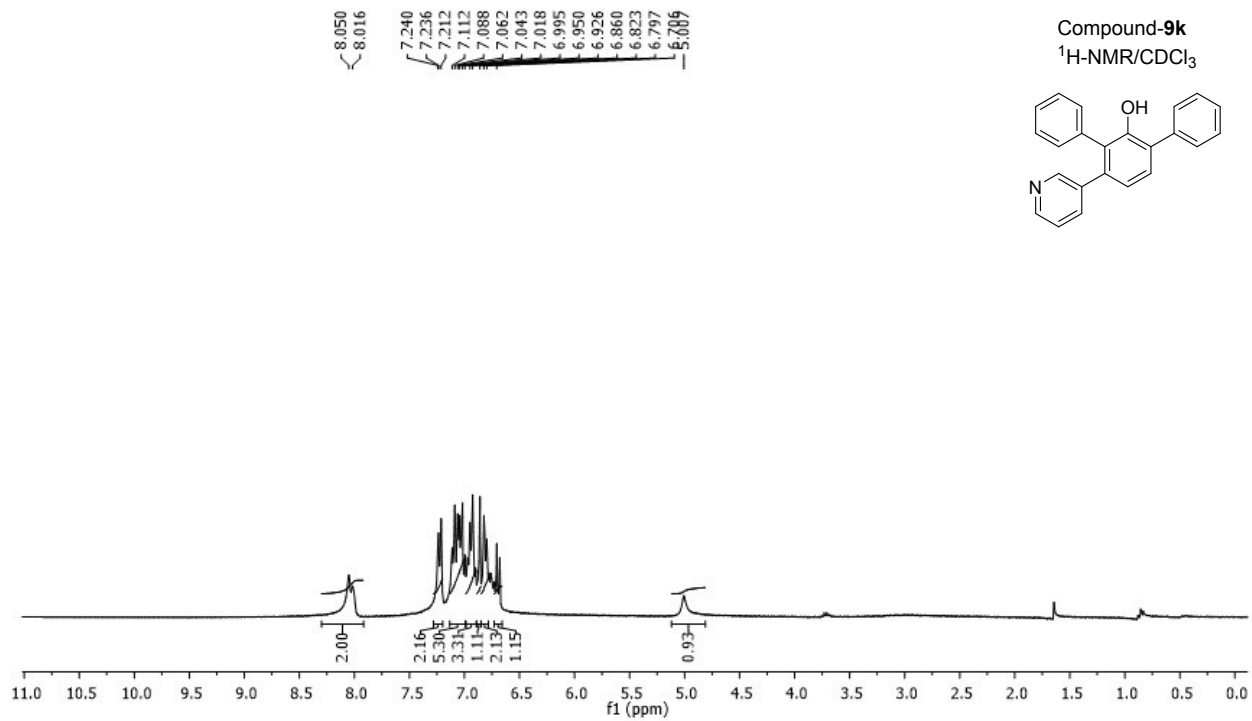


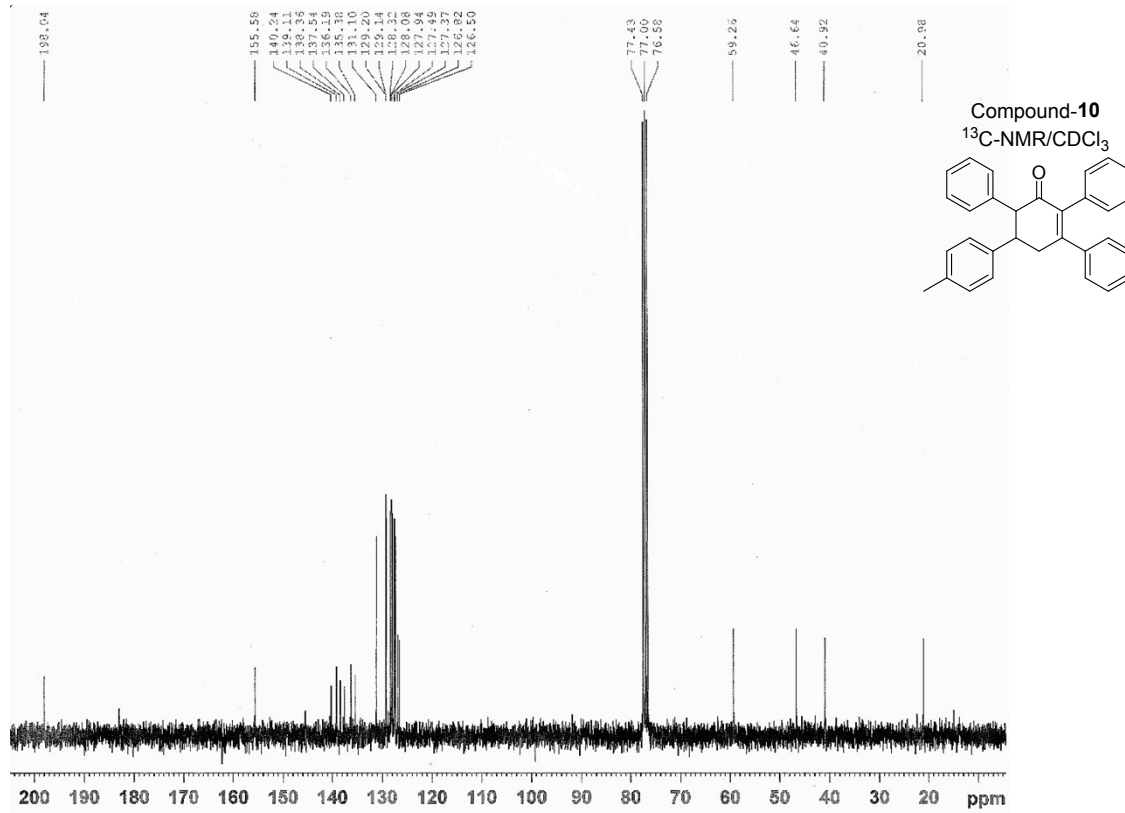
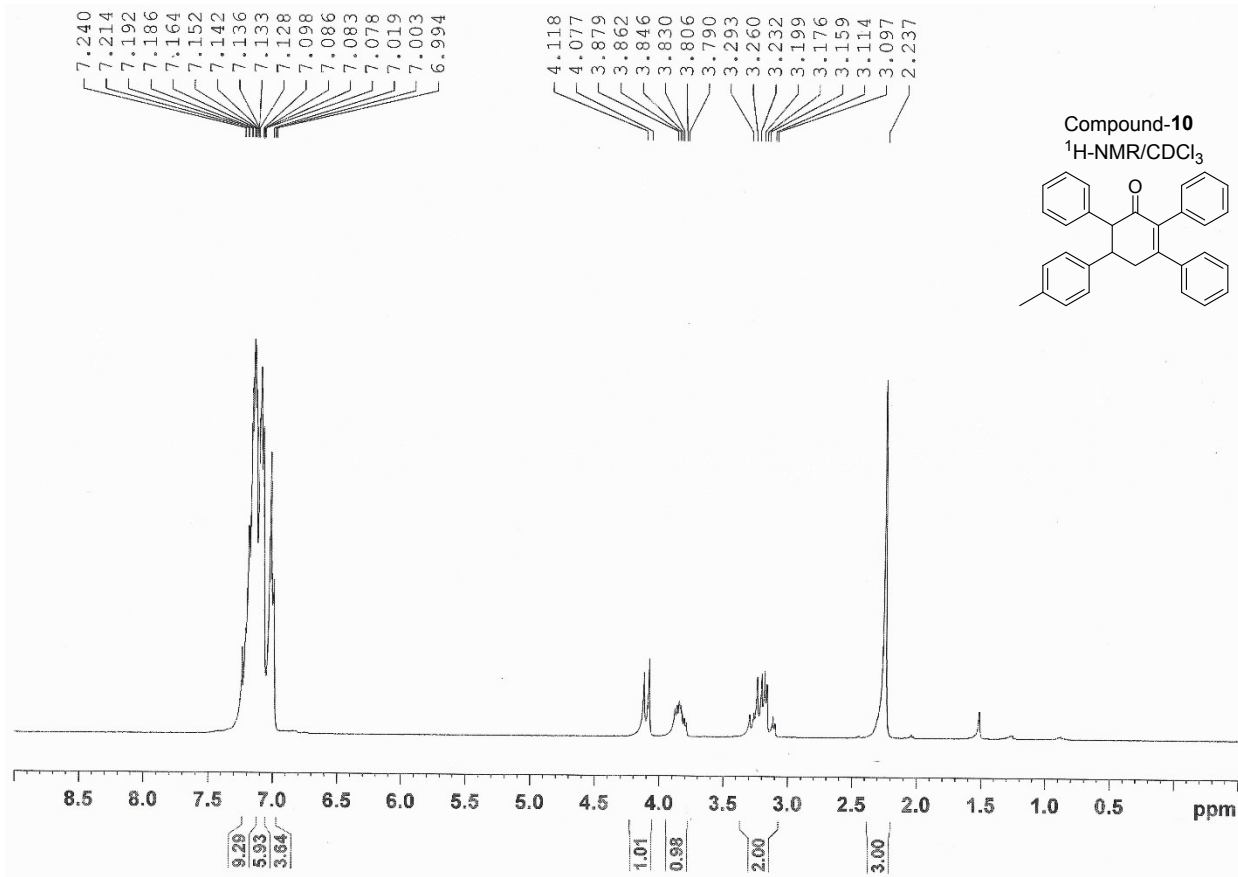
Compound-9j
¹H-NMR/CDCl₃



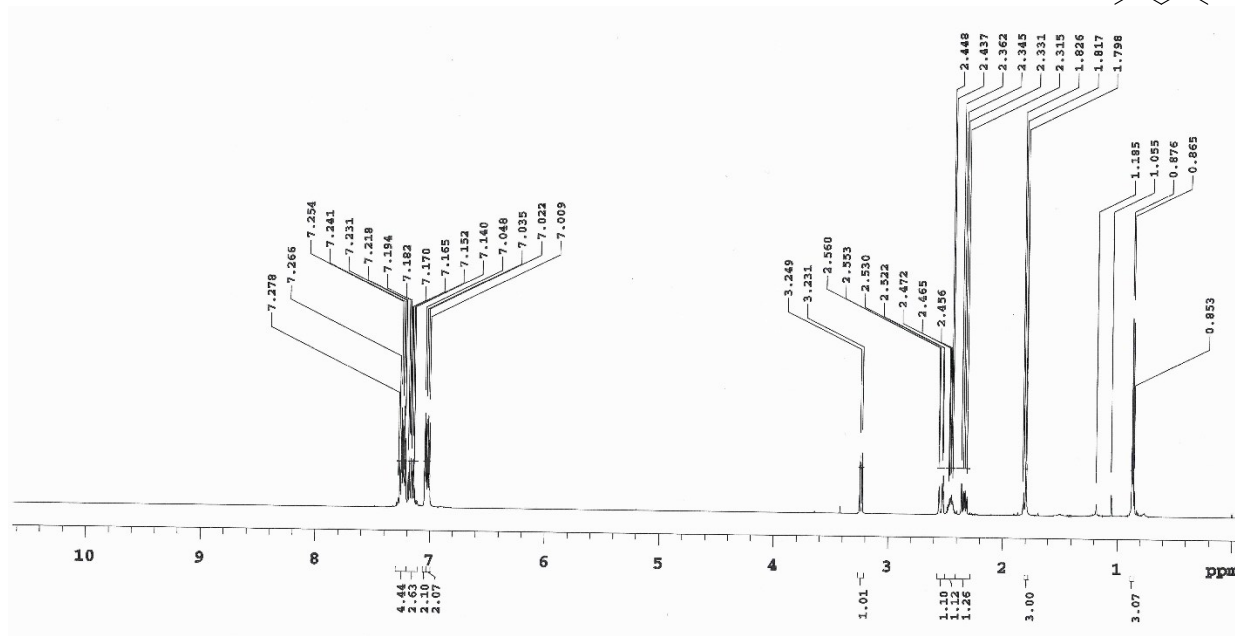
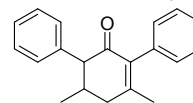
Compound-9j
¹³C-NMR/CDCl₃



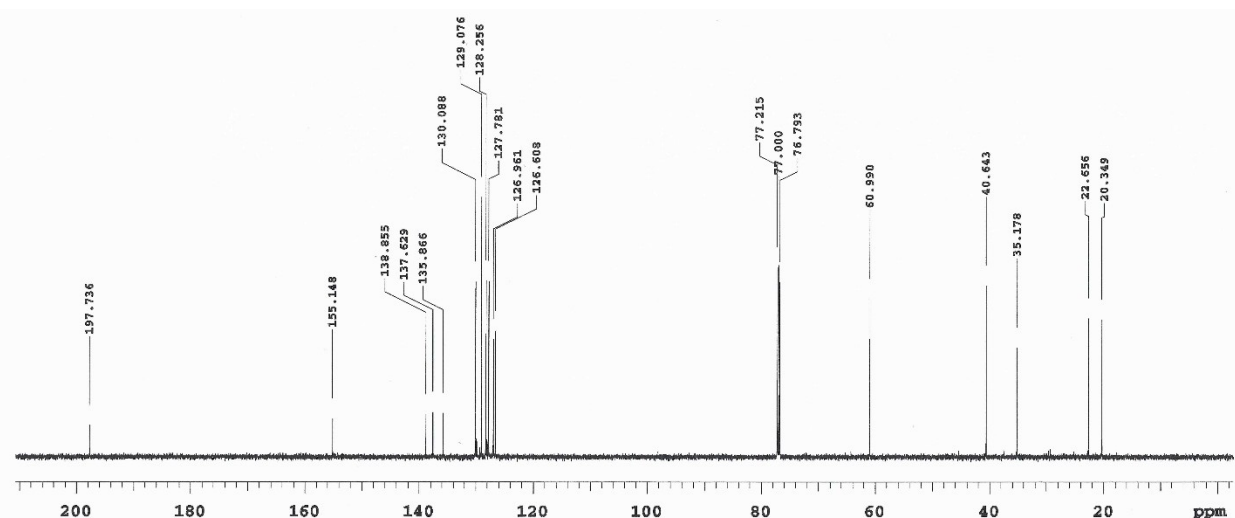
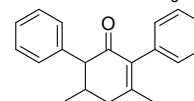




Compound-15
¹H-NMR/CDCl₃



Compound-15
¹³C-NMR/CDCl₃



X-Ray crystallographic structure and data of compound 9d: Empirical Formula- $C_{25}H_{20}O_2$, $M = 352.41$, Triclinic, Space group P_{bca} , $a = 9.5607(9) \text{ \AA}$, $b = 10.2840(10) \text{ \AA}$, $c = 10.9413(10) \text{ \AA}$, $V = 916.80(15) \text{ \AA}^3$, $Z = 2$, $T = 223(2) \text{ K}$, $\rho_{\text{calcd}} = 1.277 \text{ Mg/m}^3$, $2\Theta_{\text{max.}} = 28.413^\circ$, Refinement of 246 parameters on 4586 independent reflections out of 54615 collected reflections ($R_{\text{int}} = 0.0425$) led to $R_1 = 0.0429 [I > 2\sigma(I)]$, $wR_2 = 0.1258$ (all data) and $S = 1.069$ with the largest difference peak and hole of 0.239 and $-0.190 \text{ e. \AA}^{-3}$ respectively. The crystal structure has been deposited at the Cambridge Crystallographic Data Centre (CCDC 1526470). The data can be obtained free of charge via the Internet at www.ccdc.cam.ac.uk/data_request/cif.

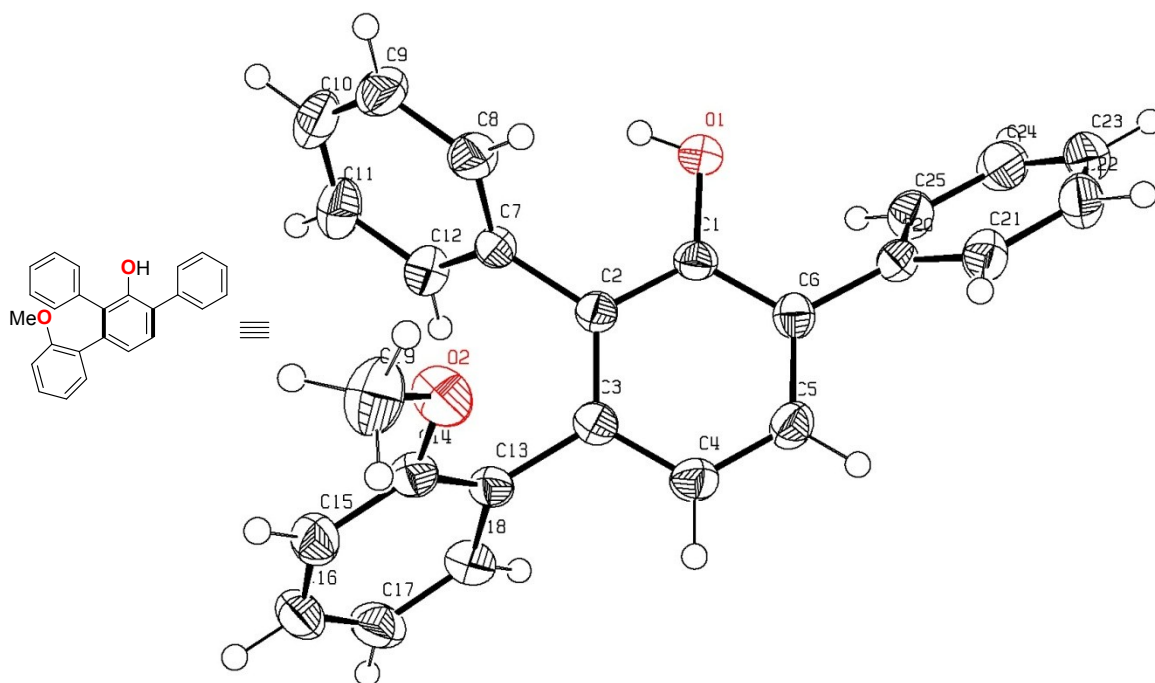


Fig. S1 X-Ray Structure of Compound 9d.

Table 1. Crystal data and structure refinement for **9d**

Identification code	9d	
Empirical formula	C ₂₅ H ₂₀ O ₂	
Formula weight	352.41	
Temperature	223(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	a = 9.5607(9) Å	α = 100.114(4)°.
	b = 10.2840(10) Å	β = 114.210(4)°.
	c = 10.9413(10) Å	γ = 102.160(4)°.
Volume	916.80(15) Å ³	
Z	2	
Density (calculated)	1.277 Mg/m ³	
Absorption coefficient	0.080 mm ⁻¹	
F(000)	372	
Crystal size	0.180 x 0.150 x 0.100 mm ³	
Theta range for data collection	2.119 to 28.413°.	
Index ranges	-12 ≤ h ≤ 12, -13 ≤ k ≤ 13, -14 ≤ l ≤ 14	
Reflections collected	54615	
Independent reflections	4586 [R(int) = 0.0425]	
Completeness to theta = 25.242°	99.8 %	
Absorption correction	Semi-empirical from equivalents	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	4586 / 0 / 246	
Goodness-of-fit on F ²	1.069	
Final R indices [I > 2σ(I)]	R1 = 0.0429, wR2 = 0.1164	
R indices (all data)	R1 = 0.0621, wR2 = 0.1258	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.239 and -0.190 e.Å ⁻³	

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$)

for **9d**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
C(1)	7188(1)	2846(1)	438(1)	29(1)
C(2)	7528(1)	1863(1)	1165(1)	28(1)
C(3)	8095(1)	842(1)	677(1)	29(1)
C(4)	8344(2)	846(1)	-491(1)	35(1)
C(5)	8045(2)	1847(1)	-1164(1)	35(1)
C(6)	7440(1)	2852(1)	-736(1)	30(1)
O(1)	6592(1)	3847(1)	845(1)	40(1)
C(7)	7232(1)	1929(1)	2412(1)	30(1)
C(8)	8074(2)	3099(1)	3580(1)	39(1)
C(9)	7750(2)	3208(2)	4709(2)	46(1)
C(10)	6593(2)	2151(2)	4698(2)	49(1)
C(11)	5764(2)	971(2)	3561(2)	48(1)
C(12)	6072(2)	858(2)	2416(1)	38(1)
C(13)	8398(2)	-287(1)	1344(1)	31(1)
C(14)	9658(2)	-20(1)	2691(1)	33(1)
C(15)	9916(2)	-1106(2)	3260(2)	41(1)
C(16)	8941(2)	-2460(2)	2475(2)	45(1)
C(17)	7727(2)	-2747(1)	1141(2)	45(1)
C(18)	7462(2)	-1657(1)	586(2)	39(1)
O(2)	10594(1)	1343(1)	3359(1)	47(1)
C(19)	12081(2)	1622(2)	4568(2)	55(1)
C(20)	7071(2)	3899(1)	-1502(1)	31(1)
C(21)	8269(2)	4791(1)	-1650(1)	37(1)
C(22)	7935(2)	5789(2)	-2328(2)	46(1)
C(23)	6391(2)	5880(2)	-2906(2)	50(1)
C(24)	5185(2)	4981(2)	-2801(2)	47(1)
C(25)	5520(2)	4006(2)	-2093(1)	39(1)

Table 3. Bond lengths [\AA] and angles [$^\circ$] for **9d**.

C(1)-O(1)	1.3705(14)
C(1)-C(6)	1.4019(16)
C(1)-C(2)	1.4027(16)
C(2)-C(3)	1.4005(16)
C(2)-C(7)	1.4955(16)
C(3)-C(4)	1.3936(17)
C(3)-C(13)	1.4935(16)
C(4)-C(5)	1.3790(18)
C(4)-H(4)	0.9400
C(5)-C(6)	1.3875(17)
C(5)-H(5)	0.9400
C(6)-C(20)	1.4898(16)
O(1)-H(1)	0.8300
C(7)-C(12)	1.3911(18)
C(7)-C(8)	1.3921(18)
C(8)-C(9)	1.3835(19)
C(8)-H(8)	0.9400
C(9)-C(10)	1.373(2)
C(9)-H(9)	0.9400
C(10)-C(11)	1.380(2)
C(10)-H(10)	0.9400
C(11)-C(12)	1.3907(19)
C(11)-H(11)	0.9400
C(12)-H(12)	0.9400
C(13)-C(18)	1.3873(18)
C(13)-C(14)	1.4000(18)
C(14)-O(2)	1.3699(16)
C(14)-C(15)	1.3891(18)
C(15)-C(16)	1.387(2)
C(15)-H(15)	0.9400
C(16)-C(17)	1.371(2)
C(16)-H(16)	0.9400
C(17)-C(18)	1.3866(19)
C(17)-H(17)	0.9400

C(18)-H(18)	0.9400
O(2)-C(19)	1.4161(18)
C(19)-H(19A)	0.9700
C(19)-H(19B)	0.9700
C(19)-H(19C)	0.9700
C(20)-C(25)	1.3921(18)
C(20)-C(21)	1.3932(17)
C(21)-C(22)	1.3847(19)
C(21)-H(21)	0.9400
C(22)-C(23)	1.379(2)
C(22)-H(22)	0.9400
C(23)-C(24)	1.377(2)
C(23)-H(23)	0.9400
C(24)-C(25)	1.3825(19)
C(24)-H(24)	0.9400
C(25)-H(25)	0.9400

O(1)-C(1)-C(6)	116.81(10)
O(1)-C(1)-C(2)	121.40(10)
C(6)-C(1)-C(2)	121.79(11)
C(3)-C(2)-C(1)	118.79(10)
C(3)-C(2)-C(7)	122.62(10)
C(1)-C(2)-C(7)	118.57(10)
C(4)-C(3)-C(2)	119.49(11)
C(4)-C(3)-C(13)	118.41(11)
C(2)-C(3)-C(13)	122.07(10)
C(5)-C(4)-C(3)	120.60(11)
C(5)-C(4)-H(4)	119.7
C(3)-C(4)-H(4)	119.7
C(4)-C(5)-C(6)	121.62(11)
C(4)-C(5)-H(5)	119.2
C(6)-C(5)-H(5)	119.2
C(5)-C(6)-C(1)	117.66(11)
C(5)-C(6)-C(20)	121.01(11)
C(1)-C(6)-C(20)	121.32(11)
C(1)-O(1)-H(1)	109.5

C(12)-C(7)-C(8)	118.27(12)
C(12)-C(7)-C(2)	121.33(11)
C(8)-C(7)-C(2)	120.34(11)
C(9)-C(8)-C(7)	121.02(13)
C(9)-C(8)-H(8)	119.5
C(7)-C(8)-H(8)	119.5
C(10)-C(9)-C(8)	120.26(14)
C(10)-C(9)-H(9)	119.9
C(8)-C(9)-H(9)	119.9
C(9)-C(10)-C(11)	119.65(13)
C(9)-C(10)-H(10)	120.2
C(11)-C(10)-H(10)	120.2
C(10)-C(11)-C(12)	120.46(14)
C(10)-C(11)-H(11)	119.8
C(12)-C(11)-H(11)	119.8
C(11)-C(12)-C(7)	120.32(13)
C(11)-C(12)-H(12)	119.8
C(7)-C(12)-H(12)	119.8
C(18)-C(13)-C(14)	118.35(11)
C(18)-C(13)-C(3)	119.36(12)
C(14)-C(13)-C(3)	122.20(11)
O(2)-C(14)-C(15)	123.98(12)
O(2)-C(14)-C(13)	115.72(11)
C(15)-C(14)-C(13)	120.29(12)
C(16)-C(15)-C(14)	119.59(13)
C(16)-C(15)-H(15)	120.2
C(14)-C(15)-H(15)	120.2
C(17)-C(16)-C(15)	121.03(13)
C(17)-C(16)-H(16)	119.5
C(15)-C(16)-H(16)	119.5
C(16)-C(17)-C(18)	119.07(13)
C(16)-C(17)-H(17)	120.5
C(18)-C(17)-H(17)	120.5
C(17)-C(18)-C(13)	121.63(13)
C(17)-C(18)-H(18)	119.2
C(13)-C(18)-H(18)	119.2

C(14)-O(2)-C(19)	117.43(12)
O(2)-C(19)-H(19A)	109.5
O(2)-C(19)-H(19B)	109.5
H(19A)-C(19)-H(19B)	109.5
O(2)-C(19)-H(19C)	109.5
H(19A)-C(19)-H(19C)	109.5
H(19B)-C(19)-H(19C)	109.5
C(25)-C(20)-C(21)	118.35(12)
C(25)-C(20)-C(6)	120.83(11)
C(21)-C(20)-C(6)	120.81(11)
C(22)-C(21)-C(20)	120.72(13)
C(22)-C(21)-H(21)	119.6
C(20)-C(21)-H(21)	119.6
C(23)-C(22)-C(21)	120.06(13)
C(23)-C(22)-H(22)	120.0
C(21)-C(22)-H(22)	120.0
C(24)-C(23)-C(22)	119.83(13)
C(24)-C(23)-H(23)	120.1
C(22)-C(23)-H(23)	120.1
C(23)-C(24)-C(25)	120.38(14)
C(23)-C(24)-H(24)	119.8
C(25)-C(24)-H(24)	119.8
C(24)-C(25)-C(20)	120.60(13)
C(24)-C(25)-H(25)	119.7
C(20)-C(25)-H(25)	119.7

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **9d**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
C(1)	32(1)	25(1)	31(1)	6(1)	16(1)	9(1)
C(2)	29(1)	26(1)	28(1)	7(1)	13(1)	8(1)
C(3)	28(1)	29(1)	31(1)	8(1)	12(1)	9(1)
C(4)	36(1)	38(1)	37(1)	12(1)	20(1)	18(1)
C(5)	35(1)	43(1)	34(1)	14(1)	20(1)	16(1)
C(6)	29(1)	30(1)	30(1)	10(1)	14(1)	7(1)
O(1)	60(1)	34(1)	41(1)	16(1)	31(1)	25(1)
C(7)	33(1)	32(1)	29(1)	11(1)	15(1)	15(1)
C(8)	44(1)	37(1)	34(1)	8(1)	16(1)	14(1)
C(9)	56(1)	53(1)	29(1)	9(1)	16(1)	26(1)
C(10)	53(1)	78(1)	33(1)	24(1)	25(1)	35(1)
C(11)	44(1)	66(1)	45(1)	26(1)	26(1)	18(1)
C(12)	37(1)	42(1)	34(1)	12(1)	16(1)	11(1)
C(13)	33(1)	30(1)	35(1)	11(1)	19(1)	15(1)
C(14)	34(1)	34(1)	36(1)	10(1)	19(1)	14(1)
C(15)	44(1)	50(1)	44(1)	23(1)	24(1)	25(1)
C(16)	57(1)	41(1)	62(1)	30(1)	39(1)	28(1)
C(17)	52(1)	30(1)	60(1)	14(1)	31(1)	15(1)
C(18)	40(1)	33(1)	42(1)	9(1)	19(1)	14(1)
O(2)	41(1)	39(1)	42(1)	9(1)	6(1)	8(1)
C(19)	40(1)	69(1)	38(1)	14(1)	9(1)	3(1)
C(20)	36(1)	30(1)	25(1)	7(1)	14(1)	10(1)
C(21)	39(1)	39(1)	32(1)	11(1)	16(1)	8(1)
C(22)	61(1)	38(1)	42(1)	16(1)	28(1)	10(1)
C(23)	79(1)	44(1)	43(1)	23(1)	32(1)	31(1)
C(24)	53(1)	57(1)	39(1)	20(1)	21(1)	32(1)
C(25)	39(1)	44(1)	36(1)	14(1)	18(1)	15(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^{-3}$) for **9d**.

	x	y	z	U(eq)
H(4)	8719	159	-824	42
H(5)	8256	1849	-1930	41
H(1)	6362	3689	1470	60
H(8)	8875	3826	3602	47
H(9)	8324	4009	5486	55
H(10)	6367	2230	5462	59
H(11)	4987	238	3560	58
H(12)	5493	55	1642	45
H(15)	10747	-923	4172	50
H(16)	9115	-3193	2863	54
H(17)	7083	-3670	610	54
H(18)	6627	-1851	-327	47
H(19A)	11875	1319	5287	83
H(19B)	12651	2613	4915	83
H(19C)	12734	1124	4331	83
H(21)	9316	4714	-1286	45
H(22)	8762	6405	-2394	55
H(23)	6163	6555	-3371	60
H(24)	4128	5031	-3212	56
H(25)	4692	3410	-2011	46

Table 6. Torsion angles [°] for **9d**.

O(1)-C(1)-C(2)-C(3)	178.34(11)
C(6)-C(1)-C(2)-C(3)	-1.70(17)
O(1)-C(1)-C(2)-C(7)	-0.20(17)
C(6)-C(1)-C(2)-C(7)	179.77(11)
C(1)-C(2)-C(3)-C(4)	1.57(17)
C(7)-C(2)-C(3)-C(4)	-179.97(11)
C(1)-C(2)-C(3)-C(13)	-176.56(11)
C(7)-C(2)-C(3)-C(13)	1.91(18)
C(2)-C(3)-C(4)-C(5)	0.31(18)
C(13)-C(3)-C(4)-C(5)	178.50(12)
C(3)-C(4)-C(5)-C(6)	-2.2(2)
C(4)-C(5)-C(6)-C(1)	2.00(19)
C(4)-C(5)-C(6)-C(20)	-177.96(12)
O(1)-C(1)-C(6)-C(5)	179.91(11)
C(2)-C(1)-C(6)-C(5)	-0.06(17)
O(1)-C(1)-C(6)-C(20)	-0.13(17)
C(2)-C(1)-C(6)-C(20)	179.90(11)
C(3)-C(2)-C(7)-C(12)	-63.18(16)
C(1)-C(2)-C(7)-C(12)	115.29(13)
C(3)-C(2)-C(7)-C(8)	119.50(13)
C(1)-C(2)-C(7)-C(8)	-62.03(16)
C(12)-C(7)-C(8)-C(9)	-1.02(19)
C(2)-C(7)-C(8)-C(9)	176.37(12)
C(7)-C(8)-C(9)-C(10)	0.5(2)
C(8)-C(9)-C(10)-C(11)	0.6(2)
C(9)-C(10)-C(11)-C(12)	-1.2(2)
C(10)-C(11)-C(12)-C(7)	0.7(2)
C(8)-C(7)-C(12)-C(11)	0.42(19)
C(2)-C(7)-C(12)-C(11)	-176.95(12)
C(4)-C(3)-C(13)-C(18)	-63.52(16)
C(2)-C(3)-C(13)-C(18)	114.62(14)
C(4)-C(3)-C(13)-C(14)	113.04(13)
C(2)-C(3)-C(13)-C(14)	-68.81(16)
C(18)-C(13)-C(14)-O(2)	176.63(11)

C(3)-C(13)-C(14)-O(2)	0.03(17)
C(18)-C(13)-C(14)-C(15)	-2.12(18)
C(3)-C(13)-C(14)-C(15)	-178.72(11)
O(2)-C(14)-C(15)-C(16)	-177.26(12)
C(13)-C(14)-C(15)-C(16)	1.38(19)
C(14)-C(15)-C(16)-C(17)	0.2(2)
C(15)-C(16)-C(17)-C(18)	-1.1(2)
C(16)-C(17)-C(18)-C(13)	0.3(2)
C(14)-C(13)-C(18)-C(17)	1.30(19)
C(3)-C(13)-C(18)-C(17)	178.00(12)
C(15)-C(14)-O(2)-C(19)	12.71(19)
C(13)-C(14)-O(2)-C(19)	-165.98(12)
C(5)-C(6)-C(20)-C(25)	121.14(14)
C(1)-C(6)-C(20)-C(25)	-58.82(16)
C(5)-C(6)-C(20)-C(21)	-58.53(17)
C(1)-C(6)-C(20)-C(21)	121.51(13)
C(25)-C(20)-C(21)-C(22)	2.15(19)
C(6)-C(20)-C(21)-C(22)	-178.17(12)
C(20)-C(21)-C(22)-C(23)	-2.1(2)
C(21)-C(22)-C(23)-C(24)	0.4(2)
C(22)-C(23)-C(24)-C(25)	1.4(2)
C(23)-C(24)-C(25)-C(20)	-1.3(2)
C(21)-C(20)-C(25)-C(24)	-0.44(19)
C(6)-C(20)-C(25)-C(24)	179.88(12)

Symmetry transformations used to generate equivalent atoms: