# The Application of Perfluoroheteroaromatic Reagents in the Preparation of Modified Peptide Systems

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# Materials and general methods

All chemicals and solvents were analytical grade and used without further purification. Liquid chromatography (LC)-mass spectrometry (ESI+) analyses were performed on a Acquity UPLC BEH C18 column (1.7 µm, 2.1 mm x 50 mm) using a Waters Acquity UPLC system equipped with a photodiode array detector, providing absorbance data from 210 nm to 400 nm. A gradient with eluent I (0.1% HCOOH in water) and eluent II (0.1% HCOOH in acetonitrile) rising linearly from 5 to 95% of II during t=0.2-4.0 min was applied at a flow rate of 0.6 ml/min after 0.2 min of 95% solvent I equilibration. High-resolution QToF-LC/MS and QToF-MS/MS analyses were performed on a Acquity UPLC BEH C18 column also (1.7 µm, 2.1 mm x 50 mm) using a Waters Acquity UPLC system coupled in this case to a Micromass QToF Premier mass spectrometer, also equipped with a photodiode array detector providing absorbance data from 210 nm to 400 nm. A gradient with eluent I (0.1% HCOOH in water) and eluent II (0.1% HCOOH in acetonitrile) rising linearly from 0 to 99% of II during t=0.0-5.0 min was applied at a flow rate of 0.6 ml/min. <sup>19</sup>F NMR spectra studies were recorded at 376 MHz in a Bruker Advance spectrometer at 298 K, using 8 scans with a relaxation delay of 1s. All data has been processed using Mestrenova® software.

### Solid phase synthesis of model peptides pep1-3:

### General experimental procedure:

Each of the described peptide sequences was prepared using automated Fmoc-SPPS methods on a Liberty 1 peptide synthesiser (CEM) with microwave-assisted couplings (single coupling per amino acid; 10 min, 75 °C (50 °C for Fmoc-Cys(Trt)-OH coupling). Solid phase synthesis was conducted using Rink amide resin (0.7 mol/g loading) on a 0.1 mmol scale, employing HBTU and DIPEA as the activator and base respectively. Standard acid-sensitive amino acid side chain protecting groups e.g. Boc, tBu, trityl were employed. Following the on-resin growing of the appropriate sequence, N-terminal acetylation was achieved by using a solution of Ac<sub>2</sub>O in DMF (20% v/v, 5 mL) while shaking at room temperature for 15 min. Once completed, the beads were washed with DMF three times and the acetylation step repeated. The resin was then carefully washed with DMF, MeOH and CH<sub>2</sub>Cl<sub>2</sub> (3 x 10 mL). Finally, peptides were cleaved from the resin as the corresponding C-terminal amides by treatment with a mixture of TFA/TIPS/H<sub>2</sub>O 95:5:5% v/v (2 mL) at room temperature for 4 h. After removal of volatiles in vacuum, the product was triturated and carefully washed with Et<sub>2</sub>O (3 x 10 mL). The final precipitated solid residue was then re-dissolved in H<sub>2</sub>O/TFA 99:1 (15 ml) and lyophilised affording the corresponding peptides as a white/yellow powder. The crude peptides were used without further chromatographic purification. LC/MS data is given next.

LC/MS characterization of crude peptides pep1-3:

Pep1: Ac-YCGGGCAL-NH<sub>2</sub>



LC/MS: Calculated m/z: 783.30, observed m/z: 784.12 [M+H]<sup>+</sup>. Retention time: 2.237 min. Elemental composition: C<sub>32</sub> H<sub>49</sub> N<sub>9</sub> O<sub>10</sub> S<sub>2</sub>.



**Figure SI01**. LC/MS trace at  $\lambda$ = 280 nm of crude peptide **pep1**.



LC/MS: Calculated m/z: 751.35, observed m/z: 752.03 [M+H]<sup>+</sup>. Retention time: 2.045 min. Elemental composition: C<sub>32</sub> H<sub>49</sub> N<sub>9</sub> O<sub>12</sub>.



**Figure SI02**. LC/MS trace at  $\lambda$ = 280 nm of crude peptide **pep2**.



LC/MS: Calculated m/z: 833.47, observed m/z: 834.33 [M+H]<sup>+</sup>. Retention time: 1.732 min. Elemental composition: C<sub>38</sub> H<sub>63</sub> N<sub>11</sub> O<sub>10</sub>.



**Figure SI03**. LC/MS trace at  $\lambda$ = 280 nm of crude peptide **pep3**.

# Model peptide tagging and stapling with perfluoroaromatics:

# General procedure for solution phase peptide tagging and stapling

Solid crude peptides **pep1-3** (2 mg, approx. 2.5  $\mu$ mol) were dissolved in the DMF (0.5 mL) in a 1.5 ml plastic Eppendorf tube, to which a Cs<sub>2</sub>CO<sub>3</sub> or DIPEA stock solution (50 mM in appropriate solvent, 0.5 mL) was then added. 25 equivalents of the corresponding perfluoroaromatic were slowly added and the tube was shaken vigorously at room temperature for 4.5 h (*method A*, unless otherwise stated). After removal of volatiles under vacuum, all products were dissolved in an 8:1:1 mixture of DMF/H<sub>2</sub>O/MeCN (1 mL) and characterised by LC/MS (ESI+). When formation of novel compounds was observed, 10-fold scaled reactions were employed in all cases for product isolation in order to afford a complete characterization. Scaled reactions were run under exactly the same exposed conditions but in argon flushed syringes, to avoid air bubbles where volatile aromatic compounds could concentrate. LC/MS data for crude reactions is provided next.

LC/MS analysis of small scale reactions

Pyridine based systems vs. hexafluorobenze. Table 1, Entries 1-5



**Entry 1:** Ac-YCGGGCAL-NH<sub>2</sub> + *hexafluorobenzene* (1:25) in DMF/DIPEA:

**Figure SI04**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with hexafluorobenzene when using DIPEA as a base in DMF.





**Figure SI05**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with pentafluoropyridine when using DIPEA as a base in DMF.

Entry 3: Ac-YCGGGCAL- NH<sub>2</sub> + 4-aminotetrafluoropyridine (1:25) in DMF/DIPEA:



**Figure SI06**. LC/MS traces  $\lambda$ =220 nm of crude reaction of peptide **pep3** with hexafluorobenzene when using DIPEA as a base in DMF.





**Figure SI07.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with pentafluoropyridine when using DIPEA as a base in DMF.



Entry 5: Ac-YSGGGSAL- NH<sub>2</sub> + 4-Amino-3,5-dichloro-2,6-difluoropyridine (1:25) in DMF/DIPEA:

1.0 1.1 1.2 1.3 1.4 1.5 1.6 1.7 1.8 1.9 2.0 2.1 2.2 2.3 2.4 2.5 2.6 2.7 2.8 2.9 3.0 3.1 3.2 3.3 3.4 3.5 3.6 3.7 3.8 3.9 Retention time (min)



**Figure SI08**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with pentafluoropyridine when using DIPEA as a base in DMF.

In order to explore the impact of changing different reaction parameters alternative methods have been employed based in the above mentioned general procedure A, in all cases employing **pep1** as the model nucleophile. Main results and modifications are summarized in following **Table SI\_001**. No other changes aside of those exposed were made to the reactions, work-up or analysis methods.

Procedure	Conditions	Comments/ Products formed
A (Entry 2)	PFP <u>25</u> equivalents DIPEA 25 mM Vf= 1 mL RT	$ \begin{array}{c} F \\ F \\$
В	PFP <u>5</u> equivalents DIPEA 25 mM Vf= 1 mL RT	F = F + F + F + F + F + F + F + F + F +
С	PFP 25 equivalents DIPEA 25 mM Vf= <u>5 mL</u> RT	HO = F + F + F + F + F + F + F + F + F + F



**Table SI01:** Procedures and overall observed effect of changing some reaction conditions over the SNAr of pentafluoropyridine by the model di-cysteine nucleophilic peptide **pep-1**.

### **Reactions in Table SI-001**

REACTION Ac-YSGGGSAL- NH<sub>2</sub> (2 mg/mL) + pentafluoropyridine (1:25) at rt :



**Figure SI09**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with pentafluoropyridine in a ratio 1:25, when using DIPEA as a base in DMF.





Peak	Retention time	m/z	Identity		
1	<b>1</b> 1.946 820		Starting peptide MeCN adduct		
2	<b>2</b> 2.411 1076, 113		Unknown		
3	2.714	1082	Double ArF addition		
4	3.074	1138	[2(M+2ArF)+TFA]2+		
5	3.175	1231	Triple ArF addition		

**Figure SI10**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with pentafluoropyridine in a ratio 1:5, when using DIPEA as a base in DMF.





**Figure SI11**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with pentafluoropyridine in 5x more diluted conditions, when using DIPEA as a base in DMF.

REACTION Ac-YSGGGSAL- NH<sub>2</sub> (2 mg/mL) + pentafluoropyridine (1:25) at 50°C:



**Figure SI12**. LC/MS traces at  $\lambda$ =280 nm of crude reaction of peptide **pep1** with pentafluoropyridine in a ratio 1:25 at 50°C, when using DIPEA as a base in DMF.



Entry 6: Ac-YCGGGCAL- NH<sub>2</sub> + *hexafluorobenzene* (1:25) in DMF/DIPEA:

**Figure SI13**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with hexafluorobenzene when using DIPEA as a base in DMF.

Entry 7: Ac-YKGGGKAL- NH<sub>2</sub> + *hexafluorobenzene* (1:25) in DMF/DIPEA:



**Figure SI14**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep3** with hexafluorobenzene when using DIPEA as a base in DMF.

Entry 8: Ac-YSGGGSAL- NH<sub>2</sub> + *pentafluoropyridine* (1:25) in DMF/DIPEA:



**Figure SI15**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with pentafluoropyridine when using DIPEA as a base in DMF.

Entry 9: Ac-YKGGGKAL- NH<sub>2</sub> + *pentafluoropyridine* (1:25) in DMF/DIPEA:



**Figure SI16.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep3** with pentafluoropyridine when using DIPEA as a base in DMF.



Entry 10: Ac-YCGGGCAL- NH<sub>2</sub> + hexafluorobenzene (1:25) in DMF/Cs<sub>2</sub>CO<sub>3</sub>:

**Figure SI17.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with pentafluoropyridine when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF.





**Figure SI18.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with pentafluoropyridine when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF

Entry 12: Ac-YKGGGKAL- NH<sub>2</sub> + *hexafluorobenzene* (1:25) in DMF/Cs<sub>2</sub>CO<sub>3</sub>:



**Figure SI19**. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep3** with hexafluorobenzene when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF.



Entry 13: Ac-YCGGGCAL- NH<sub>2</sub> + pentafluoropyridine (1:25) in DMF/Cs<sub>2</sub>CO<sub>3</sub>:

Figure SI20. LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep1** with hexafluorobenzene when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF. Inset showing the scheme corresponding to adduct formation in the basis of the observed masses.

Entry 14: Ac-YSGGGSAL- NH<sub>2</sub> + pentafluoropyridine (1:25) in DMF/Cs<sub>2</sub>CO<sub>3</sub>:



**Figure SI21.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep2** with hexafluorobenzene when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF.

Entry 15: Ac-YKGGGKAL- NH<sub>2</sub> + pentafluoropyridine (1:25) in DMF/Cs<sub>2</sub>CO<sub>3</sub>:



**Figure SI22.** LC/MS traces at  $\lambda$ =220 nm of crude reaction of peptide **pep3** with pentafluoropyridine when using Cs<sub>2</sub>CO<sub>3</sub> as a base in DMF.

# **Isolation and characterization of compounds 11-23**

## General methods:

Products from large-scale reactions were purified and isolated by semi-preparative reverse phase HPLC performed on a Discovery Bio wide pore C<sub>18</sub>-5 column from Supelco (5  $\mu$ m, 25 cm  $\times$  10 mm), using a Pelking-Elmer 200 lc pump coupled to a Waters 486 tunable absorbance detector settled up at  $\lambda$ =220 nm. A gradient with eluent A (95:5:0.1% H<sub>2</sub>O:MeCN:TFA) and eluent B (5:95:0.1% H<sub>2</sub>O:MeCN:TFA) was applied, where solvent B was firstly rose linearly from 0 to 100% during t=60 min and finally maintained isocratically for 5 min at a flow rate of 2 mL/min. Purified pooled fractions were then freeze-dried and the identity and purity of the different compounds verified by LC/MS. The desired pure compounds were then further characterised by  $^{19}$ F NMR (2) mg/mL in H<sub>2</sub>O/CD<sub>3</sub>CN 1:1, unless otherwise stated), high resolution LC/MS-QToF and ion directed tandem mass spectrometry (MS/MS), allowing to obtain the characteristic rupture profile for each product. In MS/MS fragmentation analysis we have made use of the accepted nomenclature for fragment ions firstly proposed by Roepstorff and Fohlman (P. Roepstorff and J. Fohlman, Biol. Mass Spectrom. 1984, 11, 601-601.), and subsequently modified by Johnson et al. (R. S. Johnson, S. A. Martin, K. Biemann J. T. Stults and J. T. Watson, Anal. Chem., 1987, 59, 2621-2625). Note that, in peptides and proteins, ions arising from fragmentation series  $\gamma$  or b are expected to be predominant.

## Product 11.



Isolated mass: 6.2 mg (26%).

QToF LC/MS: Calculated *m/z*: 929.96, observed *m/z*: 931.49  $[M+H]^+$  Retention time: 2.400 min. Elemental composition: C<sub>38</sub> H<sub>47</sub> F<sub>4</sub> N<sub>9</sub> O<sub>10</sub> S<sub>2</sub>.



**Figure SI23**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =280 nm and composition of isolated compound **11**.

F NMR (D <sub>2</sub> O/MeCN) (1:1)				
	MU			
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121 -122 -123 -124 -125 -126 -127 -	128 -129 -130 -131 f1 (ppm)	-132 -133 -134	-135 -136	-137 -138 -139
<sup>19</sup> F NMR Expansion				
	$\Lambda \Lambda$		ΛΛΛ.	
		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	~/VVV	manum
30.05 -130.10 -130.15 -130.20 -130.25 -130.30 -130.35 -130.40 -130	.45 -130.50 -130.55 -130.60 -130.6	55 -130.70 -130.75 -130.80 -	130.85 -130.90 -130.95 -:	31.00 -131.05 -131.10

**Figure SI024**. <sup>19</sup>F NMR spectrum of compound **11** as recorded in  $H_2O/ACN-d_3$  1:1 at room temperature.

Product 12.



Isolated mass: 6.4 mg (20%).

QToF LC/MS: Calculated m/z: 1230.27, observed m/z: 1231.28 [M+H]<sup>+</sup> Retention time: 3.442 min. Elemental composition: C<sub>47</sub> H<sub>46</sub> F<sub>12</sub> N<sub>12</sub> O<sub>10</sub> S<sub>2</sub>.

QToF-MS/MS: Calculated m/z: 1103.18  $[b7+H]^+$ , 1032.13  $[b6+H]^+$ , 779.14  $[b5+H]^+$ , 722.11  $[b4+H]^+$ , 665.09  $[b3+H]^+$ , 608.07  $[b2+H]^+$ , 356.07  $[b1+H]^+$ , 495.15  $[z4+H]^+$  Da.

Observed m/z: 1103.21  $[b7+H]^+$ , 1032.18  $[b6+H]^+$ , 780.17  $[b5+H]^+$ , 722.14  $[b4+H]^+$ , 665.12  $[b3+H]^+$ , 608.13  $[b2+H]^+$ , 355.08  $[b1+H]^+$ , 495.14  $[z4+H]^+$  Da.

<sup>19</sup>F NMR (376 MHz, DMSO-*d*<sub>6</sub>) δ -91.20 (m, 2F), -93.42 (m, 4F), -137.49 (m, 4F), -155.63 (m, 2F).



**Figure SI25**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **12**.



Figure SI26. MS/MS analysis of compound 12 showing its characteristic rupture pattern and the assignation of the main ions observed.



Figure SI27. <sup>19</sup>F -NMR spectrum of compound 12 as recorded in DMSO-d<sub>6</sub> at room temperature.

Product 13.



Isolated mass: 2.7 mg (11%).

QToF LC/MS: Calculated m/z: 961.25, observed m/z: 1018.33 [2M+TFA]<sup>2+</sup> Retention time: 2.425 min. Elemental composition: C<sub>37</sub> H<sub>50</sub> Cl<sub>2</sub> F N<sub>11</sub> O<sub>10</sub> S<sub>2</sub>.

QToF-MS/MS: Calculated m/z: 889.65  $[b7/b7^{+} TFA+H]^{+}$ , 818.65  $[b6/b6^{+} TFA+H]^{+}$ , 730.32  $[b5+TFA+H]^{+}$ , 658.58  $[b4+TFA+H]^{+}$ , 601. 08  $[b3+TFA+H]^{+}$ , 537.66  $[b5^{+} TFA+H]^{+}$ , 480.16  $[b4^{+} TFA+H]^{+}$ , 423.13  $[b3^{+} TFA+H]^{+}$ , 366.11  $[b2^{+} TFA+H]^{+}$ , 331  $[x3+H]^{+}$  Da.

Observed *m/z*: 890.19 [b7/b7'+ TFA+H]<sup>+</sup>, 819.20 [b6/b6'+TFA+H]<sup>+</sup>, 729.13 [b5+ TFA+H]<sup>+</sup>, 660.12 [b4+ TFA+H]<sup>+</sup>, 603.08 [b3+ TFA+H]<sup>+</sup>, 536.23 [b5'+ TFA+H]<sup>+</sup>, 479.20 [b4'+TFA+H]<sup>+</sup>, 422.17 [b3'+TFA+H]<sup>+</sup>, 365.14 [b2'+TFA+H]<sup>+</sup>, 332.12 [x3+H]<sup>+</sup> Da.

<sup>19</sup>F NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1) δ -73.68 (m, 1F), -73.80 (m, 1F).



**Figure SI28**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **13**.



Figure SI29. MS/MS analysis of compound 13 showing its characteristic rupture pattern and the main ions observed.



**Figure SI30**. <sup>19</sup>F NMR spectrum of compound **13** as recorded in  $H_2O/MeCN-d_3$  1:1 at room temperature.

Product 14.



Isolated mass: 6.1 mg (25%).

QToF LC/MS: Calculated m/z: 900.34, observed m/z: 901.34 [M+H]<sup>+</sup> Retention time: 2.375 min. Elemental composition: C<sub>37</sub> H<sub>48</sub> F<sub>4</sub> N<sub>10</sub> O<sub>12</sub>.

QToF-MS/MS: Calculated m/z: 771.24  $[b7]^+$ , 700.20  $[b6]^+$ , 613.17  $[b5]^+$ , 499.12  $[b3]^+$ , 442.35  $[b2]^+$ , 355.07  $[b1]^+$  Da.

Observed m/z: 771.27  $[b7]^+$ , 700.24  $[b6]^+$ , 613.19  $[b5]^+$ , 499.10  $[b3]^+$ , 442.08  $[b2]^+$ , 355.11  $[b1]^+$  Da.

 $^{19}F$  NMR (376 MHz, H\_2O/MeOD 1:1)  $\delta$  -91.32 (m, 2F), -155.98 (m, 2F).



**Figure SI31**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound 14.



**Figure SI32**. MS/MS analysis of compound **14** showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI33**. <sup>19</sup>F NMR spectrum of compound **14** as recorded in  $H_2O/MeOD$  1:1 at room temperature.

Product 15.



Isolated mass: 7.2 mg (23%).

QToF LC/MS: Calculated m/z: 1280.44, observed m/z: 1281.45 [M+H]<sup>+</sup> Retention time: 3.492 min. Elemental composition: C<sub>53</sub> H<sub>60</sub> F<sub>12</sub> N<sub>14</sub> O<sub>10</sub>.

QToF-MS/MS:

Calculated m/z: 1152.34 [b7+H]<sup>+</sup>, 1081.30 [b6+H]<sup>+</sup>, 804.22 [b5+H]<sup>+</sup>, 691.2337 [b3+H]<sup>+</sup>, 747.20 [b4+H]<sup>+</sup>, 911.36 [z7+H]<sup>+</sup>, 634.28 [z6+H]<sup>+</sup>, 520.23 [z4+H]<sup>+</sup> Da.

Observed m/z: 1152.40 [b7+H]<sup>+</sup>, 1082.36 [b6+H]<sup>+</sup>, 804.26 [b5+H]<sup>+</sup>, 690.23 [b3+H]<sup>+</sup>, 747.24 [b4+H]<sup>+</sup>, 911.36 [z7+H]<sup>+</sup>, 634.29 [z6+H]<sup>+</sup>, 520.23 [z4+H]<sup>+</sup> Da.

 $^{19}{\rm F}$  NMR (376 MHz, H2O/MeOD 1:1)  $\delta$  -91.66 (m, 2F), -98.17 (m, 4F), -156.29 (m, 4F), -165.54 (m, 2F).



**Figure SI34**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **15**.



Figure SI35. MS/MS analysis of compound 15 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI36**. <sup>19</sup>F NMR spectrum of compound **15** as recorded in  $H_2O/MeOD$  1:1 at room temperature.

Product 16.



Isolated mass: 6.0 mg (24%).

QToF LC/MS: Calculated m/z: 917.33, observed m/z: 918.34  $[M+H]^+$ . Retention time: 2.842 min. Elemental composition: C<sub>38</sub> H<sub>48</sub> F<sub>5</sub> N<sub>9</sub> O<sub>12</sub>.

QToF MS/MS:

Calculated m/z: 789.23 [b7+H]<sup>+</sup>, 771.23 [b7-H<sub>2</sub>O+H]<sup>+</sup>, 718.19 [b6+H]<sup>+</sup>, 700.19 [b6-H<sub>2</sub>O+H]<sup>+</sup>, 631.16 [b5+H]<sup>+</sup>, 574.14 [b4+H]<sup>+</sup>, 531.26 [z7+H]<sup>+</sup> Da.

Observed m/z: 790.20  $[b7+H]^+$ , 772.19  $[b7-H_2O+H]^+$ , 718.16  $[b6+H]^+$ , 700.15  $[b6-H_2O+H]^+$ , 631.13  $[b5+H]^+$ , 574.12  $[b4+H]^+$ , 531.23  $[z7+H]^+$  Da.

 $^{19}\text{F}$  NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1)  $\delta$  -156.46 (m, 2F), -162.76 (m, 1F), -164.40 (m, 2F).



**Figure SI37**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **16**.



**Figure SI38**. MS/MS analysis of compound **16** showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI39**. <sup>19</sup>F NMR spectrum of compound **16** as recorded in  $H_2O/MeCN-d_3$  1:1 at room temperature.

Product 17.



Isolated mass: 3.4 mg (14%).

QToF LC/MS: Calculated m/z: 879.32, observed m/z: 880.29  $[M+H^+]^+$ . Retention time: 2.633 min. Elemental composition: C<sub>38</sub> H<sub>45</sub> F<sub>4</sub> N<sub>9</sub> O<sub>11</sub>.

QToF MS/MS:

Calculated m/z: 752.22 [b7-(OH)+H]<sup>+</sup>, 611.16 [b5+H]<sup>+</sup>, 554.13 [b4+H]<sup>+</sup>, 497.11 [b3+H]<sup>+</sup>, 440.09 [b2+H]<sup>+</sup> Da.

Observed m/z: 752.19 [b7-(OH)+H]<sup>+</sup>, 611.14 [b5+H]<sup>+</sup>, 554.12 [b4+H]<sup>+</sup>, 496.17 [b3+H]<sup>+</sup>, 440.07 [b2+H]<sup>+</sup> Da.

<sup>19</sup>F NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1) δ -159.46 (m, 2F), -163.63 (m, 2F).



**Figure SI40**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **17**.



Figure SI41. MS/MS analysis of compound 17 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI42**. <sup>19</sup>F NMR spectrum of compound **17** as recorded in H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1 at room temperature.

Product 18.



Isolated mass: 2.2 mg (9%).

QToF LC/MS: Calculated m/z: 881.31, observed m/z: 882.32  $[M+H]^+$ . Retention time: 2.633 min. Elemental composition: C<sub>38</sub> H<sub>44</sub> F<sub>5</sub> N<sub>9</sub> O<sub>10</sub>.

QToF MS/MS:

Calculated m/z: 753.21 [b7+H]<sup>+</sup>, 682.17 [b6+H]<sup>+</sup>, 613.15 [b5+H]<sup>+</sup>, 556.13 [b4+H]<sup>+</sup>, 499.11 [b3+H]<sup>+</sup>, 495.53 [z7+H]<sup>+</sup>, 384.20 [y5+H]<sup>+</sup>, 312.36 [z4+H]<sup>+</sup> Da.

Observed m/z: 754.19 [b7+H]<sup>+</sup>, 683.15 [b6+H]<sup>+</sup>, 613.13 [b5+H]<sup>+</sup>, 556.11 [b4+H]<sup>+</sup>, 499.10 [b3+H]<sup>+</sup>, 495.22 [z7+H]<sup>+</sup>, 382.14 [y5+H]<sup>+</sup>, 310.11 [z4+H]<sup>+</sup> Da.

 $^{19}{\rm F}$  NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1)  $\delta$  -156.24 (d, J= 23.56 Hz, 2F), -161.81 (t, J= 23.40 Hz, 1F), -163.74 (m, 2F).



**Figure SI43**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **18**.



**Figure SI44**. MS/MS analysis of compound **18** showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI45**. <sup>19</sup>F NMR spectrum of compound **18** as recorded in  $H_2O/MeCN-d_3$  1:1 at room temperature.

Product 19 (Mixture of regioisomers).



Isolated mass: 3.2 mg (11%).

QToF LC/MS: Calculated m/z: 1081.28, observed m/z: 1138.35  $[2M+TFA+H]^{2+}$ . Retention time: 3.300 min. Elemental composition: C<sub>42</sub> H<sub>47</sub> F<sub>8</sub> N<sub>11</sub> O<sub>10</sub> S<sub>2</sub>.

QToF MS/MS:

Calculated *m/z*: 938.64  $[2(b6//b^{\circ}6)+TFA+H]^{2+}$ , 919.18  $[(b7/b^{\circ}7)+TFA-4fPyr+H]^{+}$ , 848.15  $[b6/b^{\circ}6+TFA-4fPyr+H]^{+}$ , 778.61  $[2b^{\circ}4+TFA+H]^{2+}$ , 722.59  $[2b^{\circ}3+TFA+H]^{2+}$ , 686.60  $[2b5 + TFA + H]^{2+}$ , 664.57  $[2b^{\circ}3+TFA + H]^{2+}$ , 629.62  $[2b4 + TFA + H]^{2+}$ , 480.15  $[2b5 + 4fPyr+H]^{2+}$ , 423.13  $[2b4 + -4fPyr+H]^{2+}$  Da.

Observed *m/z*: 939.24 [2(b6/b'6)+TFA+H<sup>+</sup>]<sup>2+</sup>, 920.56 [b7 /b'7)+TFA-4*f*Pyr+H<sup>+</sup>]<sup>+</sup>, 849.17 [b6/b'6)+TFA-4*f*Pyr+H<sup>+</sup>]<sup>+</sup>, 779.14 [2b'4+TFA+H<sup>+</sup>]<sup>2+</sup>, 723.14 [2b'3+TFA+H<sup>+</sup>]<sup>2+</sup>, 687.23 [2b5 +TFA+H<sup>+</sup>]<sup>2+</sup>, 664.12 [2b'3+TFA+H<sup>+</sup>]<sup>2+</sup>, 630.18 [2b4 +TFA+H<sup>+</sup>]<sup>2+</sup>, 481.13 [2b5+-4*f*Pyr+H<sup>+</sup>]<sup>2+</sup>, 424.08 [2b4 +-4*f*Pyr+H<sup>+</sup>]<sup>2+</sup> Da.

<sup>19</sup>F NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1) δ -91.57 (m, 2F), -93.40 (m, 2F), -137.69 (m, 2F), -156.12 (m, 2F).



**Figure SI46**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **19**.



**Figure SI47**. MS/MS analysis of compound **19** showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI48**. <sup>19</sup>F NMR spectrum of compound **19** as recorded in H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1 at room temperature.

Product 20 (Mixture of regioisomers).



Isolated mass: 3.1 mg (13%).

QToF LC/MS: Calculated m/z: 932.29, observed m/z: 1048.30 [M+TFA+H]<sup>+</sup> Retention time: 3.008 min. Elemental composition: C<sub>37</sub> H<sub>48</sub> F<sub>4</sub> N<sub>10</sub> O<sub>10</sub> S<sub>2</sub>.

#### QToF MS/MS:

Calculated m/z: 918.19 [b7/b'7/b''7+TFA+H]<sup>+</sup>, 847.15 [b6/b'6/b''6+ TFA +H]<sup>+</sup>, 595.16 [b5+ TFA +H]<sup>+</sup>, 538.46 [b4+ TFA +H]<sup>+</sup>, 481.41 [b3+ TFA +H]<sup>+</sup>, 423.46 [b4+H]<sup>+</sup> Da.

Observed m/z: 920.21 [b7/b'7/b'7+TFA+H]<sup>+</sup>, 849.17 [b6/b'6/b''6+TFA+H]<sup>+</sup>, 596.17 [b5+TFA+H]<sup>+</sup>, 538.15 [b4+TFA+H]<sup>+</sup>, 478.07 [b3+TFA+H]<sup>+</sup>, 424.08 [b4+H]<sup>+</sup> Da.

<sup>19</sup>F NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1) δ -91.65 (m, 2F), -93.49 (m, 2F), -137.74 (m, 2F), -156.21 (m, 2F).



**Figure SI49**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **20**.



Figure SI50. MS/MS analysis of compound 20 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI51**. <sup>19</sup>F NMR spectrum of compound **20** as recorded in  $H_2O/MeCN-d_3$  1:1 at room temperature.

Product 21 (Mixture of regioisomers).



Isolated mass: 6.5 mg (27%).

QToF LC/MS: Calculated m/z: 900.34, observed m/z: 901.34  $[M+H]^+$ . Retention time: 2.667 min. Elemental composition: C<sub>37</sub> H<sub>48</sub> F<sub>4</sub> N<sub>10</sub> O<sub>12</sub>.

QToF MS/MS:

Calculated m/z: 772.24  $[b7/b'7/b''7+H]^+$ , 754.24  $[b7/b'7/b''7-H_2O+H]^+$ , 701.20  $[b6/b'6/b''6+H]^+$ , 683.20  $[b6/b'6/b''6-H_2O+H]^+$ , 614.17  $[b5/b5''+H]^+$ , 557.45  $[b4/b4''+H]^+$ , 421.15  $[z'3+H]^+$ , 403.15  $[z'3-H_2O+H]^+$  Da.

Observed m/z: 773.21 [b7/b'7/b''7+H]<sup>+</sup>, 755.20 [b7/b'7/b''7-H<sub>2</sub>O+H]<sup>+</sup>, 702.18 [b6/b'6/b''6+H]<sup>+</sup>, 683.16 [b6/b'6/b''6-H<sub>2</sub>O+H]<sup>+</sup>, 614.15 [b5/b5''+H]<sup>+</sup>, 557.13 [b4/b4'' +H]<sup>+</sup>, 419.16 [z'3+H]<sup>+</sup>, 401.15 [z'3-H<sub>2</sub>O+H]<sup>+</sup> Da.

 $^{19}{\rm F}$  NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1)  $\delta$  -91.64 (m, 2F), -97.10 (m, 2F), -156.16 (m, 2F), -165.96 (m, 2F).





**Figure SI52**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **21**.

Figure SI53. MS/MS analysis of compound 21 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI54**. <sup>19</sup>F NMR spectrum of compound **21** as recorded in  $H_2O/MeCN-d_3$  1:1 at room temperature.

Product 22 (Mixture of regioisomers).



Isolated mass: 2.7 mg (9%).

QToF LC/MS: Calculated m/z: 1049.33, observed m/z: 1050.34 [M+H]<sup>+</sup> Retention time: 2.667 min. Elemental composition: C<sub>42</sub> H<sub>47</sub> F<sub>8</sub> N<sub>11</sub> O<sub>12</sub>.

QToF MS/MS:

Calculated m/z: 921.23  $[b7/b'7+H]^+$ , 903.22  $[b7/b'7-H_2O+H]^+$ , 903.22  $[b7/b'7-4fPyr+H]^+$ , 850.19  $[b6/b'6+H]^+$ , 833.19  $[b6/b'6-H_2O+H]^+$ , 683.57  $[b7/b'7-4fPyr+H]^+$ , 613.18  $[b'5+H]^+$  Da.

Observed m/z: 922.20  $[b7/b'7+H]^+$ , 904.19  $[b7/b'7-H_2O+H]^+$ , 755.21  $[b7/b'7-4fPyr+H]^+$ , 850.17  $[b6/b'6+H]^+$ , 833.16  $[b6/b'6-H_2O+H]^+$ , 680.23  $[b7/b'7-4fPyr+H]^+$ , 614.15  $[b'5+H]^+$  Da.

 $^{19}{\rm F}$  NMR (376 MHz, H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1)  $\delta$  -91.63 (m,2F), -93.39 (m,2F), -156.17 (m,2F), -159.31 (m,2F).



**Figure SI55**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **22**.



Figure SI56. MS/MS analysis of compound 22 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI57**. <sup>19</sup>F NMR spectrum of compound **22** as recorded in H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1 at room temperature.

#### **Product 23**



Isolated mass: 3.5 mg (11%).

QToF LC/MS: Calculated m/z: 1198.32, observed m/z: 1199.33 [M+H]<sup>+</sup> Retention time: 3.558 min. Elemental composition: C<sub>47</sub> H<sub>46</sub> F<sub>12</sub> N<sub>12</sub> O<sub>12</sub>.

### QToF MS/MS:

Calculated m/z: 1070.21 [b7+H]<sup>+</sup>, 999.18 [b6+H]<sup>+</sup>, 904.22 [b7-4*f*PyrOH+H]<sup>+</sup>, 829.24 [z7+H]<sup>+</sup>, 763.16 [b5+H]<sup>+</sup>, 706.13 [b4+H]<sup>+</sup>, 649.11 [b3+H]<sup>+</sup> Da.

Observed m/z: 1071.18 [b7+H]<sup>+</sup>, 1000.14 [b6+H]<sup>+</sup>, 904.19 [b7-4*f*PyrOH+H]<sup>+</sup>, 832.15 [z7+H]<sup>+</sup>, 763.13 [b5+H]<sup>+</sup>, 706.11 [b4+H]<sup>+</sup>, 649.09 [b3+H]<sup>+</sup> Da.

 $^{19}{\rm F}$  NMR (376 MHz, H2O/MeCN-d3 1:1)  $\delta$  -91.63 (m, 2F), -93.49 (m, 4F), -156.17 (m, 2F), -159.24 (m, 4F).



**Figure SI58**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **23**.



Figure SI59. MS/MS analysis of compound 23 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI60**. <sup>19</sup>F NMR spectrum of compound **23** as recorded in H<sub>2</sub>O/MeCN-d<sub>3</sub> 1:1 at room temperature.

# Effect of perfluoro-tagging over Vassopresin and Oxytocin bio-stability:

### General procedure for solution phase peptide tagging

Commercially available C-teminal amide oxytocin (24) and vasopressin (25) were purchased from Cayman Chemical Company and Apollo Scientific Ltd. respectively, and used as starting materials without further purification. In both cases 10 mg of the starting products were dissolved in 5 mL of DMF to which another 5 mL of a freshly made stock solution of DIPEA 50 mM in DMF was then added. Solutions where then placed in argon flushed syringes, and 25 equivalents of either pentafluoropyridine or tetrafluoropyridazine were slowly added. The systems were then allowed to react while vigorously shaking at room temperature for 4.5 h. After removal of DMF and volatiles under vacuum, the remaining oily residue was dissolved in H<sub>2</sub>O/MeCN (10 mL) and freeze-dried. Crude materials were characterised by LC/MS (ESI+) to confirm the presence of the desired products and further purified by semi-preparative HPLC (as described previously). Final pure compounds were further characterised by <sup>19</sup>F-NMR (2 mg/mL in CD<sub>3</sub>CN), high resolution LC/MS-QToF and ion directed tandem mass spectrometry (MS/MS) in order to confirm the tagging positions at the tyrosine and N-terminus positions.

### LC/MS analysis:

Oxytocin (24) arylation using pentafluoropyridine (1) and tetrafluoropyridazine (3) :



**Figure SI61**. LC/MS trace at  $\lambda$ =220 nm of crude reaction of Oxytocin with pentafluoropyridine (1) in the presence of DPEA as a base.



**Figure SI62**. LC/MS trace at  $\lambda$ =220 nm of crude reaction of Oxytocin with tetrafluoropyridazine (3) in the presence of DIPEA as a base.

Vasopressin (25) arylation using pentafluoropyridine (1) and tetrafluoropyridazine (3) :



**Figure SI63**. LC/MS trace at  $\lambda$ =220 nm of crude reaction of Vasopressin with pentafluoropyridine (1) in the presence of DIPEA as a base.



**Figure SI64.** LC/MS trace at  $\lambda$ =220 nm of crude reaction of Vasopressin with tetrafluoropyridazine (3) in the presence of DIPEA as a base.

# **Compound 26**



Isolated mass: 4.6 mg (36%).

QToF LC/MS: Calculated m/z: 1270.43, observed m/z: 1271.42 [M+H]<sup>+</sup> Retention time: 2.858 min. Elemental composition: C<sub>51</sub> H<sub>64</sub> F<sub>6</sub> N<sub>16</sub> O<sub>12</sub> S<sub>2</sub>.

QToF MS/MS:

Calculated m/z: 1198.38 [b8+H]<sup>+</sup>, 1085.30 [b7+H]<sup>+</sup>, 988.25 [b6+H]<sup>+</sup>, 960.25 [a6+H]<sup>+</sup>, 284.18 [y3+H]<sup>+</sup> Da.

Observed m/z: 1199.38 [b8+H]<sup>+</sup>, 1086.29 [b7+H]<sup>+</sup>, 989.25 [b6+H]<sup>+</sup>, 961.25 [a6+H]<sup>+</sup>, 285.19 [y3+H]<sup>+</sup> Da.

<sup>19</sup>F NMR (376 MHz, MeCN- $d_3$ ): δ -88.15 (1F), -89.18 (1F), -95.84 (1F), -102.08 (1F), -140.29 (1F), -151.91 (1F).



**Figure SI65**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **26**.



Figure SI66. MS/MS analysis of compound 26 showing its characteristic rupture pattern and the assignation of the main ions observed.



**Figure SI67**. <sup>19</sup>F NMR spectrum of compound **26** as recorded in MeCN-d<sub>3</sub> at room temperature. **Compound 27** 



Isolated mass: 5.4 mg (43%).

QToF LC/MS: Calculated m/z: 1347.42, observed m/z: 1348.43 [M+H]<sup>+</sup>. Retention time: 2.300 min. Elemental composition: C<sub>54</sub> H<sub>63</sub> F<sub>6</sub> N<sub>19</sub> O<sub>12</sub> S<sub>2</sub>.

QToF MS/MS:

Calculated m/z: 1022.23 [b6+H]<sup>+</sup>, 920.23 [b5+H]<sup>+</sup>, 806.19 [b4+H<sup>+</sup>]<sup>+</sup>, 786.34 [z7-NH<sub>3</sub>+H]<sup>+</sup>, 678.13 [b3+H]<sup>+</sup>, 531.06 [b2+H]<sup>+</sup>, 503.06 [a3+H]<sup>+</sup>, 327.19 [y3+H]<sup>+</sup>, 310.18 [y3+NH<sub>3</sub>+H]<sup>+</sup>, 213.12 [y2-NH<sub>3</sub>+H]<sup>+</sup> Da.

Observed m/z: 1022.13 [b6+H]<sup>+</sup>, 922.21 [b5+H]<sup>+</sup>, 807.20 [b4+H<sup>+</sup>]<sup>+</sup>, 786.11 [z7-NH<sub>3</sub>+H]<sup>+</sup>, 679.13 [b3+H]<sup>+</sup>, 531.06 [b2+H]<sup>+</sup>, 503.07 [a3+H]<sup>+</sup>, 328.19 [y3+H]<sup>+</sup>, 311.18 [y3+NH<sub>3</sub>+H]<sup>+</sup>, 214.13 [y2-NH<sub>3</sub>+H]<sup>+</sup> Da.

<sup>19</sup>F NMR (376 MHz, MeCN- $d_3$ ): δ -88.05 (m, 1F), -89.17 (m, 1F), -95.69 (m, 1F), -101.88 (1F), -140.14 (m, 1F), -151.79 (m, 1F).



**Figure SI68**. Structure, high resolution QToF-LC/MS trace at  $\lambda$ =220 nm and composition of isolated compound **27**.



Figure SI69. MS/MS analysis of compound 27 showing its characteristic rupture pattern and the assignation of the main ions observed.



Figure SI70. <sup>19</sup>FNMR spectrum of compound 27 as recorded in MeCN-d<sub>3</sub> at room temperature.

# General procedure for in vitro digestion using Chymotrypsin

For bio-stability assays  $\alpha$ -chymotrypsin, from bovine pancreas, was purchased from Sigma-Aldrich. For digestion, peptides were dissolved in a stock solution of Tris buffer (25mM, pH: 7.6) up to a concentration of 1 mg/mL concentration. Simultaneously a freshly made stock solution of enzyme in the same chilled buffer was prepared. The samples were immediately prepared by mixing in an eppendorf an aliquot of 100 µL of peptide solution with the corresponding amount of enzyme stock solution to lead a final peptide to enzyme ratio of 25:1. The sample mixtures were then incubated in a thermally controlled water bath at 20°C for a minimum of 24 h. During the incubation time, 15 µL aliquots of sample mixture were withdrawn at different times, quenched at once with 15 µL of 30% MeCN/TFA solution and analyzed by analytical reverse phase HPLC, employing a X-Bridge C<sub>18</sub> column (5.3 µm, 4.6 x 100 mm, 40°C) in a Pelkin-Elmer 200 series lc system supplied with auto-sampler, UV/Vis detector and a Peltier column oven. In all samples a linear gradient rising from eluent A (95:5:0.05% H<sub>2</sub>O:MeCN:TFA) to 100% of eluent B (5:95:0.03% H<sub>2</sub>O:MeCN:TFA) during t=30 min plus 5 min of isocratic B elution was applied at a constant flow rate of 1 mL/min. Detection was performed at  $\lambda$ =220 nm. In order to properly analyze the evolution of digested peptide, and due to the fact that in some cases close proximity did not permit a direct integration of well resolved peaks, the signals corresponding to both peptide and enzyme in the raw chromatograms were baseline corrected and then fitted into separate Gaussian equations using Origin8 software. The area enclosed by each function was taken as the peak neat integrated area. The ratio of non-digested peptide is then estimated as:

 $Peptide Ratio = \frac{[Peptide Area/Enzyme Area]_t}{[Peptide Area/Enzyme Area]_{t=0}}.$ 

For each peptide a minimum of two independent assays were performed, and the final Peptide Ratio vs Incubation time analyzed in terms of an exponential decay behavior where:

Peptide Ratio =  $y_0 + Ae^{-kt}$ 

allowing the comparison of the resulting fitted models. Note that due to enzyme aggregation, the proposed method may be only valid for the short initial times window.

In order to discard possible peptide degradation during experimental time, control samples were run by subjecting each peptide to the same procedure and conditions using the corresponding volume of Tris buffer instead of enzyme solution. In any case degradation of the starting materials was observed.





**Figure SI71.** A) HPLC traces showing the progression of parent oxytocin (24) digestion in the presence of chymotrypsin ( $\lambda$ = 220 nm). B) Experimental points (n=2) and exponential fit showing the ratio of non-digested peptide left in the sample with the incubation time. C) Maldi-ToF analysis of the final digested products after 24h of incubation.

In vitro bio-stability of perfluoro-tagged Oxytocin 26



**Figure SI72.** A) HPLC traces showing the progression of perfluoro-tagged oxytocin (26) digestion in the presence of chymotrypsin ( $\lambda$ = 220 nm). B) Experimental points (n=2) and exponential fit showing the ratio of non-digested peptide left in the sample with the incubation time. C) Maldi-ToF analysis of the final digested products after 24h of incubation.





**Figure SI73**. A) HPLC traces showing the progression of parent vasopressin (25) digestion in the presence of chymotrypsin ( $\lambda$ = 220 nm). B) Experimental points (n=2) and exponential fit showing the ratio of non-digested peptide left in the sample with the incubation time. C) Maldi-ToF analysis of the final digested products after 24h of incubation.

In vitro bio-stability of perfluoro-tagged Vasopressin 27



**Figure SI74.** A) HPLC traces showing the progression of perfluoro-tagged vasopressin (27) digestion in the presence of chymotrypsin ( $\lambda$ = 220 nm). B) Experimental points (n=2) and exponential fit showing the ratio of non-digested peptide left in the sample with the incubation time. C) Maldi-ToF analysis of the final digested products after 24h of incubation.







**Figure SI75**. Control reactions showing the HPLC trace progression and the final Maldi-ToF analysis after 24 h of incubation at 20°C in the absence of chymotrypsin. A) parent oxytocin 24, B) "tagged" analogue 26, C) parent vasopressin 25 and D) "tagged" vasopressin 27.