

## Electronic Supplementary Information

### Rapidly accessible “click” rotaxanes utilizing a single amide hydrogen bond motif

Beth E. Fletcher, Michael J. G. Peach and Nicholas H. Evans\*

#### Contents

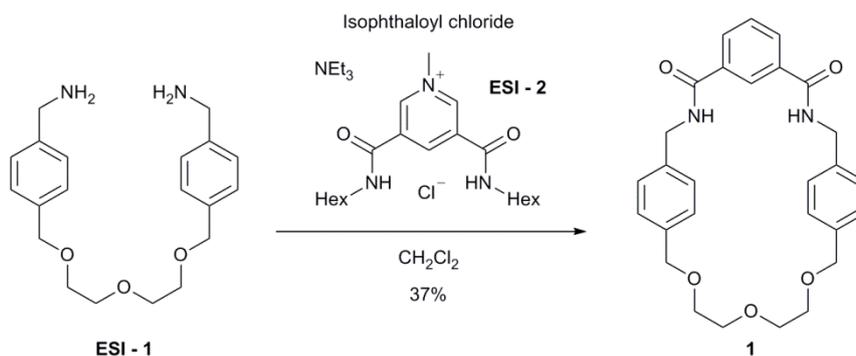
<b>Part I: Synthesis</b> .....	<b>S2</b>
Additional Notes on Experimental Procedures.....	S2
<b>Part II: Spectral Characterisation</b> .....	<b>S3</b>
Macrocycle <b>1</b> .....	S3
Alkyne <b>3</b> .....	S4
Bromo-amide <b>4</b> .....	S6
Azide <b>5</b> .....	S8
Rotaxane <b>6</b> .....	S10
Rotaxane <b>7</b> .....	S16
Axle <b>8</b> .....	S22
Axle <b>9</b> .....	S27
<sup>1</sup> H NMR Spectral Comparison of Macrocycle <b>1</b> , Rotaxane <b>7</b> and Axle <b>9</b> .....	S32
<b>Part III: Computational Modelling</b> .....	<b>S33</b>
Minimum Energy Structures of Rotaxane <b>6</b> .....	S33
Energies of Minimised Structures of Rotaxanes <b>6</b> and <b>7</b> .....	S33
Hydrogen Bond Distances in Minimised Structures of Rotaxanes <b>6</b> and <b>7</b> .....	S33
<b>Part IV: References and Notes</b> .....	<b>S34</b>

## Part I: Synthesis

### Additional Notes on Experimental Procedures

#### Preparation of macrocycle **1**

The synthesis of macrocycle **1** using semi-high dilution macrocyclisation conditions in 12% yield has been reported previously.<sup>1</sup> We have since adapted the template synthesis method of Hancock and Beer.<sup>2</sup>



*Scheme ESI-1: Templated synthesis of macrocycle **1***

Dimethanamine **ESI-1**<sup>3</sup> (169 mg, 0.49 mmol) and methyl pyridinium chloride template **ESI-2**<sup>4,5</sup> (188 mg, 0.49 mmol) were dissolved in dry CH<sub>2</sub>Cl<sub>2</sub> (15 mL) under an Ar (g) atmosphere. Then NEt<sub>3</sub> (0.14 mL, 99 mg, 0.98 mmol) was added, immediately followed by the dropwise addition of a solution of isophthaloyl chloride (100 mg, 0.49 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (10 mL). The reaction was stirred for 30 minutes under an Ar (g) atmosphere. Then, the reaction mixture was washed with 10% HCl (aq) (2 × 25 mL) and H<sub>2</sub>O (2 × 25 mL). The organic layer was dried (MgSO<sub>4</sub>), filtered and solvent removed *in vacuo*. The crude material was submitted to silica gel column chromatography (9:1 EtOAc/CH<sub>2</sub>Cl<sub>2</sub>) to yield the title compound as a white solid (86 mg, 37%).

Mp 192-196 °C (Lit: 198-200 °C).<sup>1</sup>

$R_f = 0.54$ , 9:1 EtOAc/CH<sub>2</sub>Cl<sub>2</sub> (Lit:  $R_f = 0.55$ , 9:1 EtOAc/CH<sub>2</sub>Cl<sub>2</sub>).<sup>1</sup>

$\delta$ H(400 MHz; CDCl<sub>3</sub>) 7.97 (2H, dd, <sup>3</sup> $J = 7.7$  Hz <sup>4</sup> $J = 0.9$  Hz), 7.79 (1H, s), 7.50 (1H, t, <sup>3</sup> $J = 7.7$  Hz, C<sup>1</sup>H), 7.25-7.30 (8H, m), 6.76 (2H, br s), 4.49-4.53 (8H, m), 3.60-3.69 (8H, m).

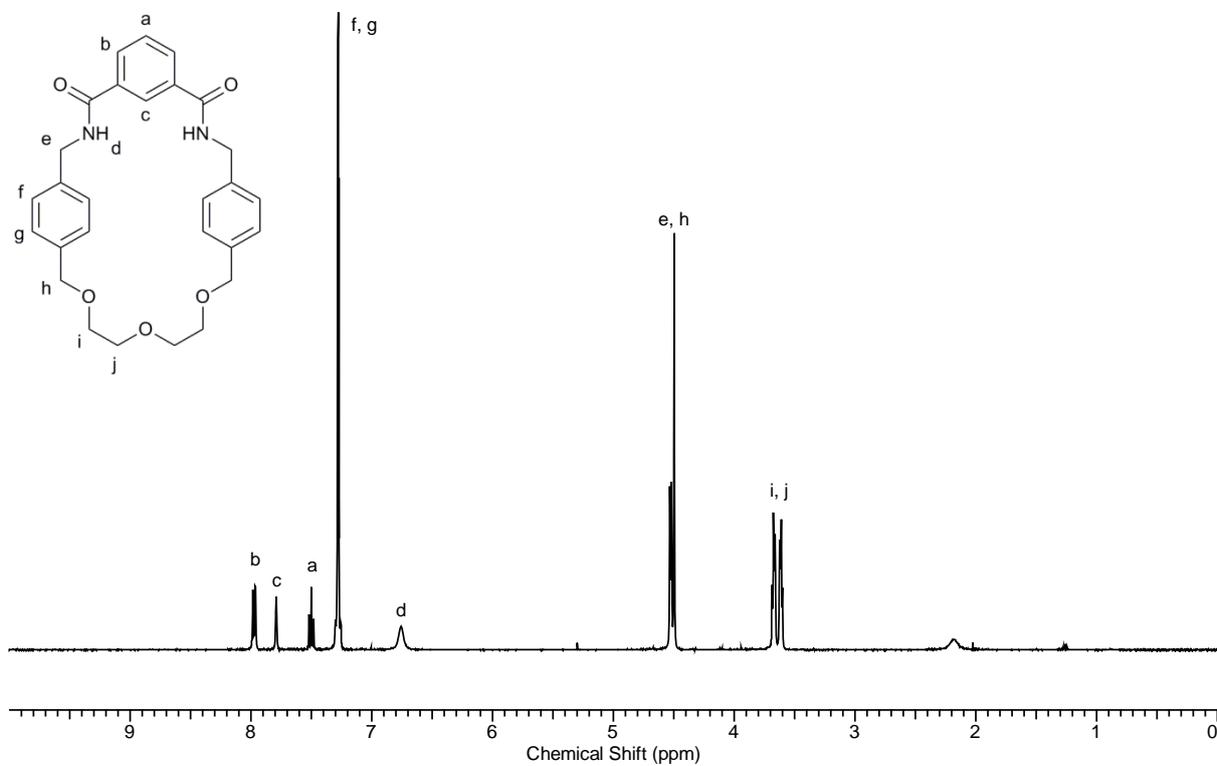
$\delta$ C(100 MHz; CDCl<sub>3</sub>) 166.9, 137.5, 137.2, 134.6, 131.0, 129.5, 128.6, 128.3, 123.7, 72.9, 70.5, 69.5, 44.1.

<sup>1</sup>H and <sup>13</sup>C NMR data are consistent with literature values.<sup>1</sup>

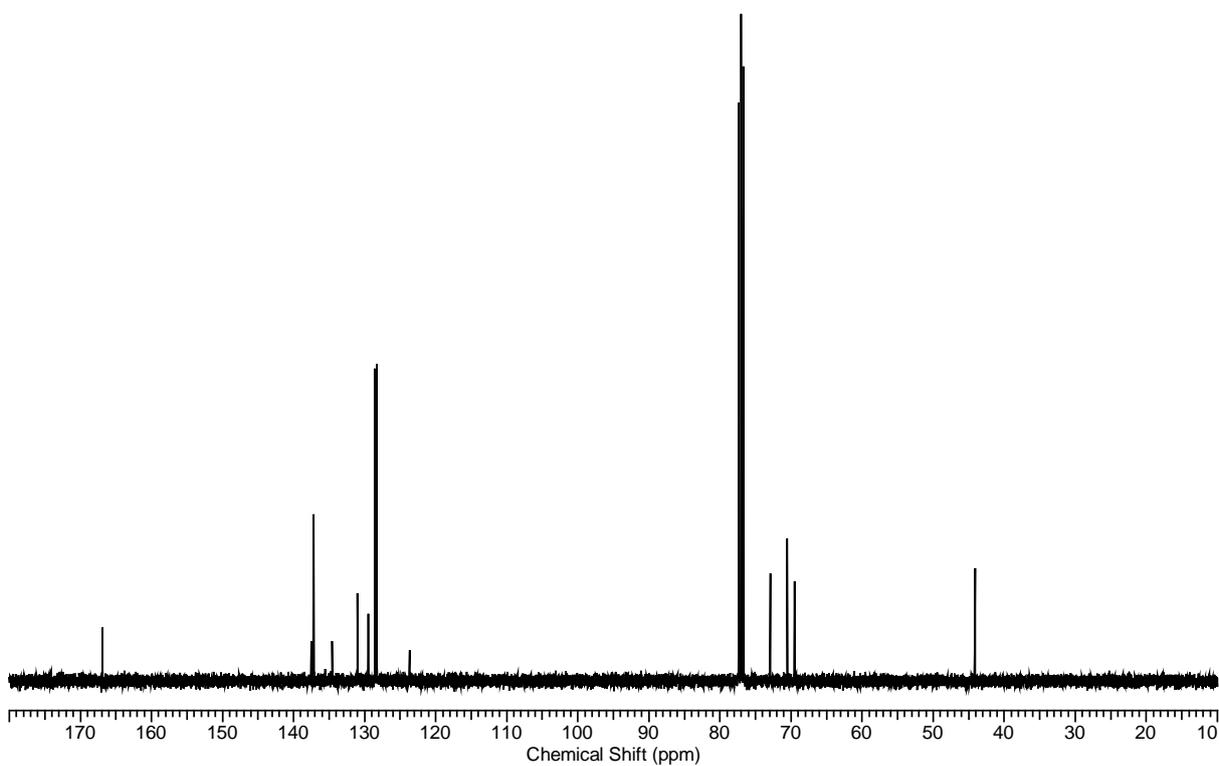
## Part II: Characterisation Spectra

### Macrocycle 1

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)



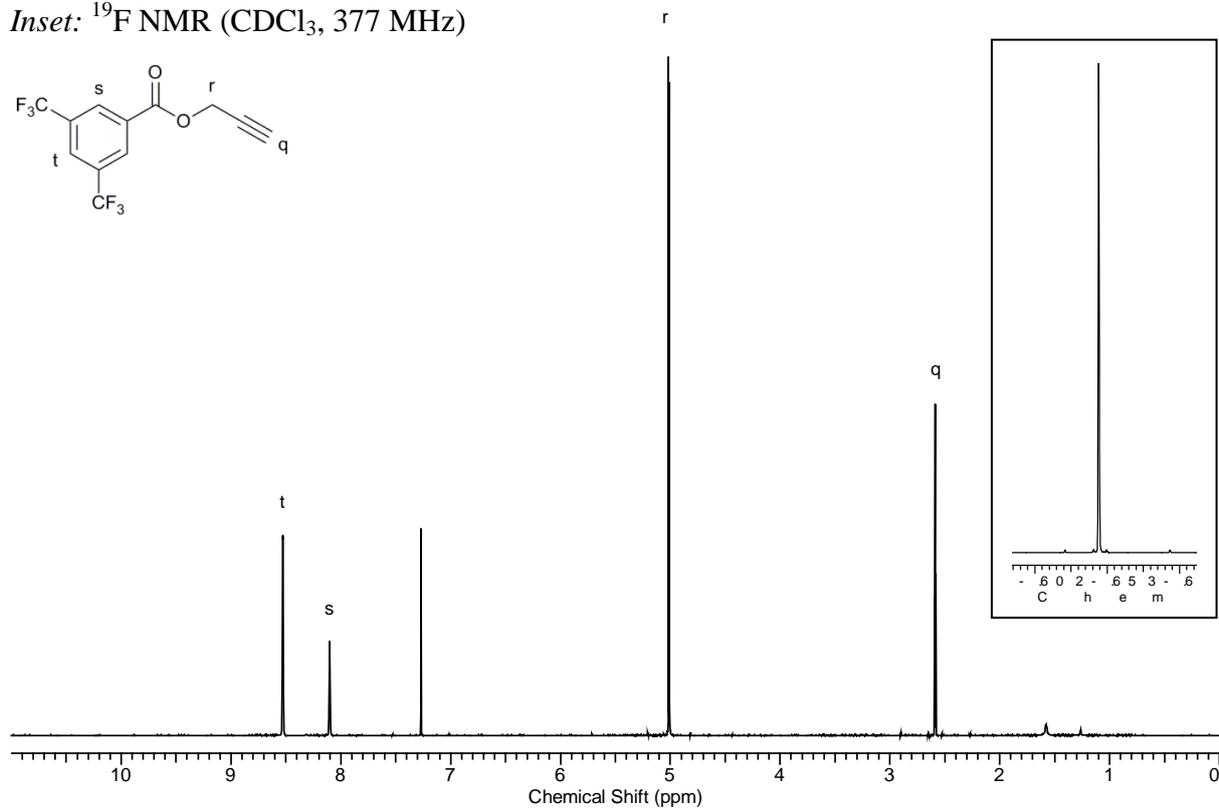
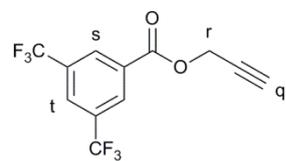
$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



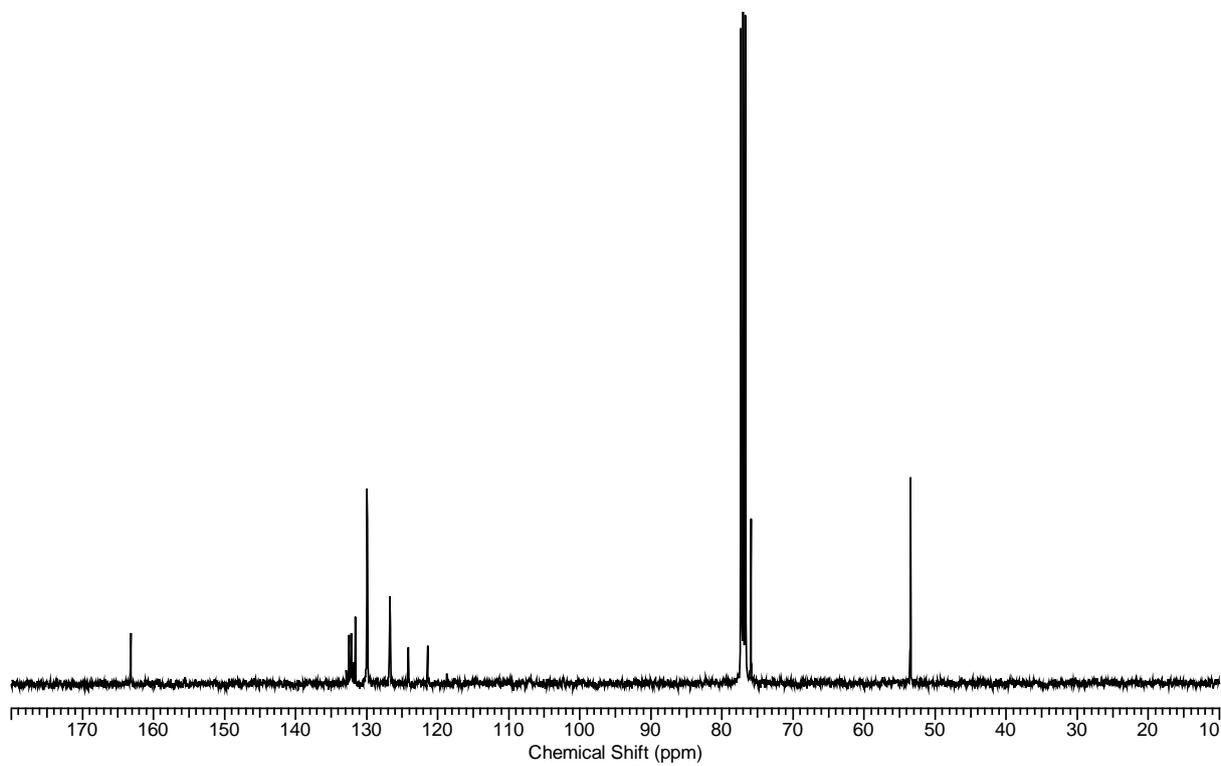
### Alkyne 3

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)

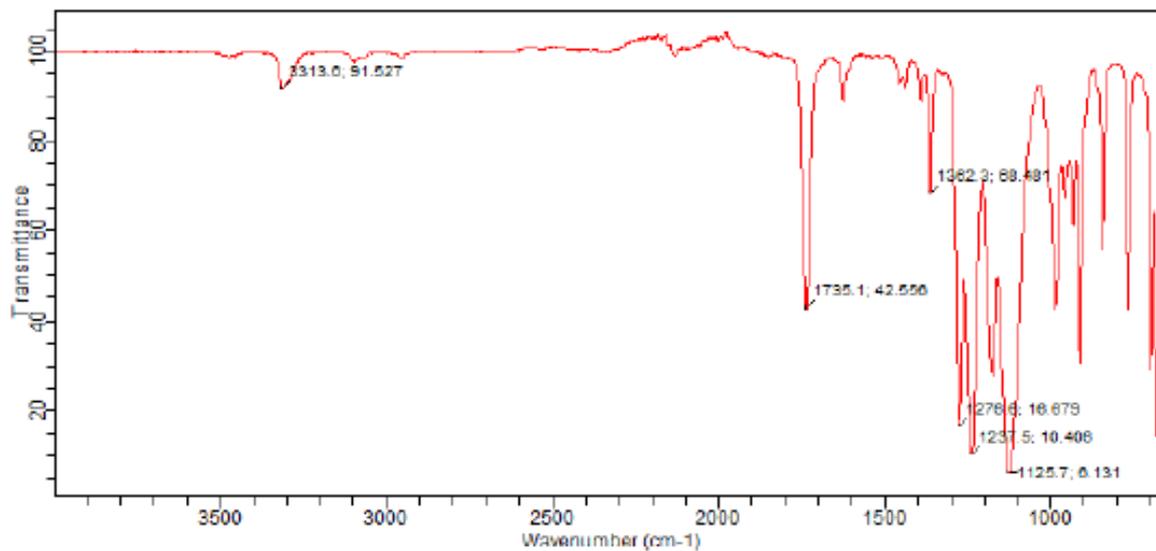


$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)

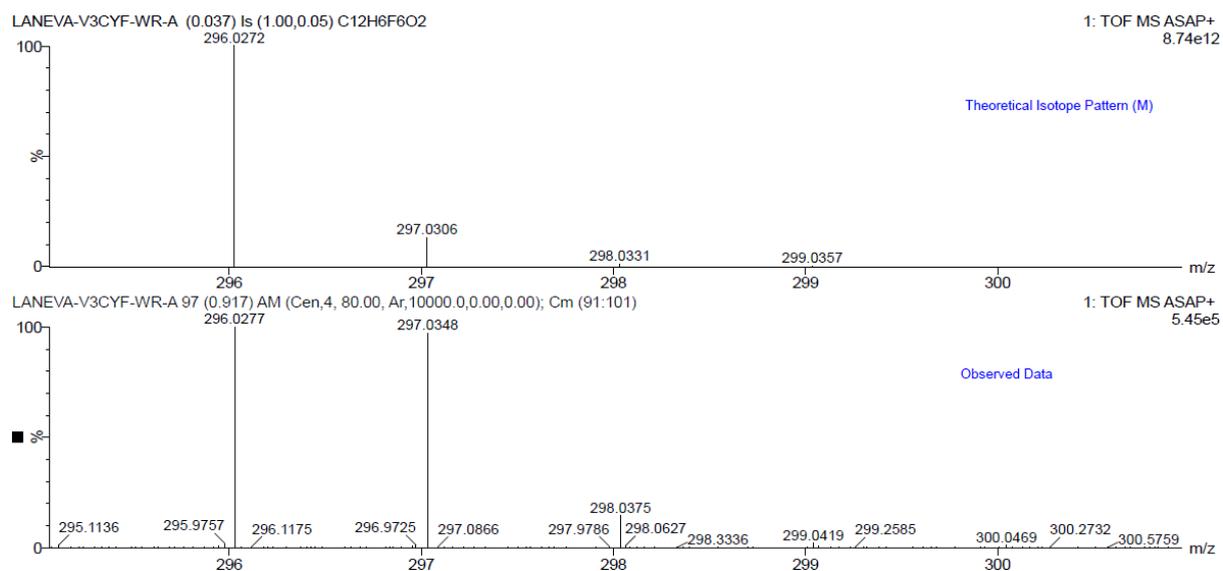


# Alkyne 3

## IR Spectrum (neat)



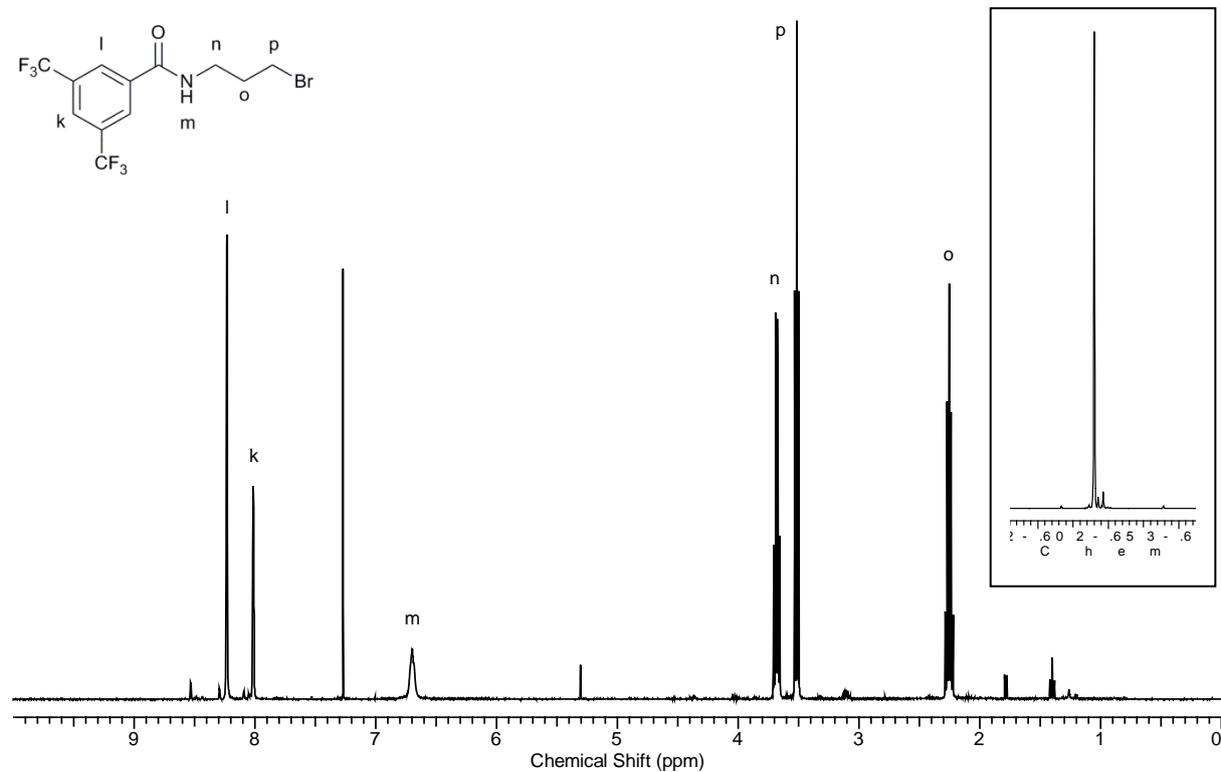
## Mass Spectrum (ASAP +ve)



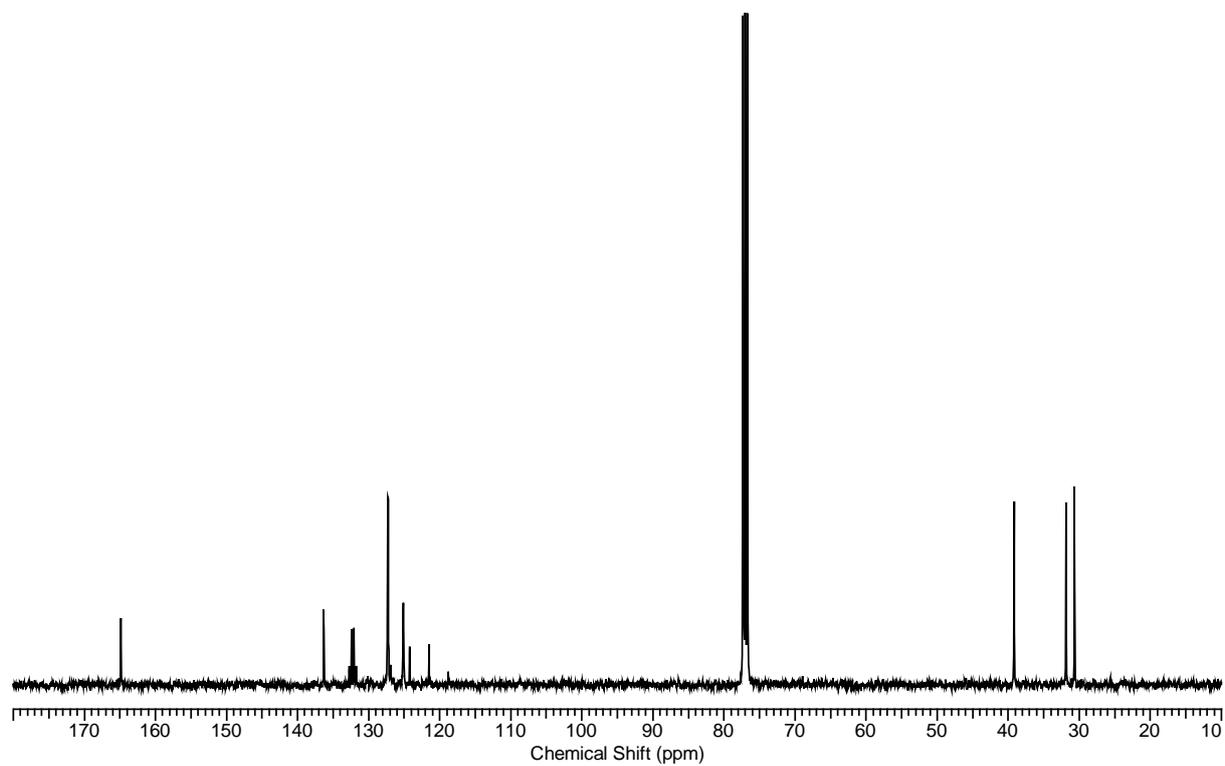
## Bromo-amide 4

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)

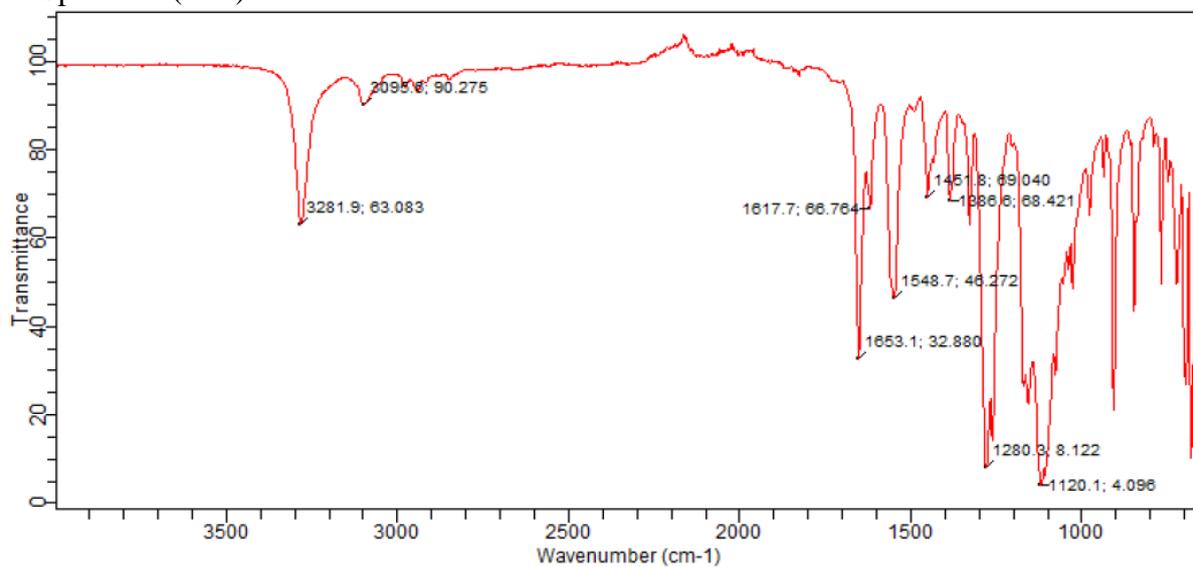


$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)

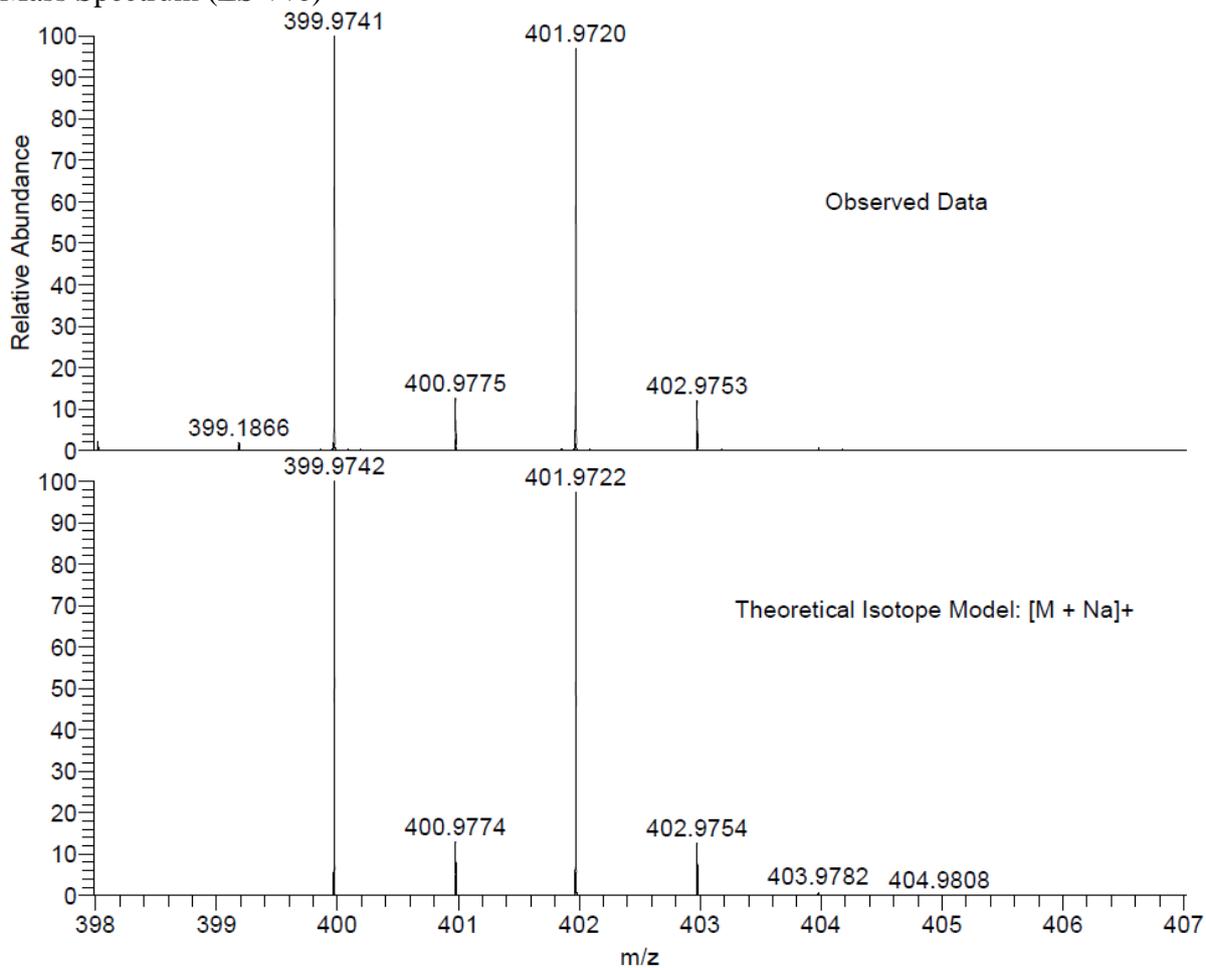


# Bromo-amide 4

## IR Spectrum (neat)



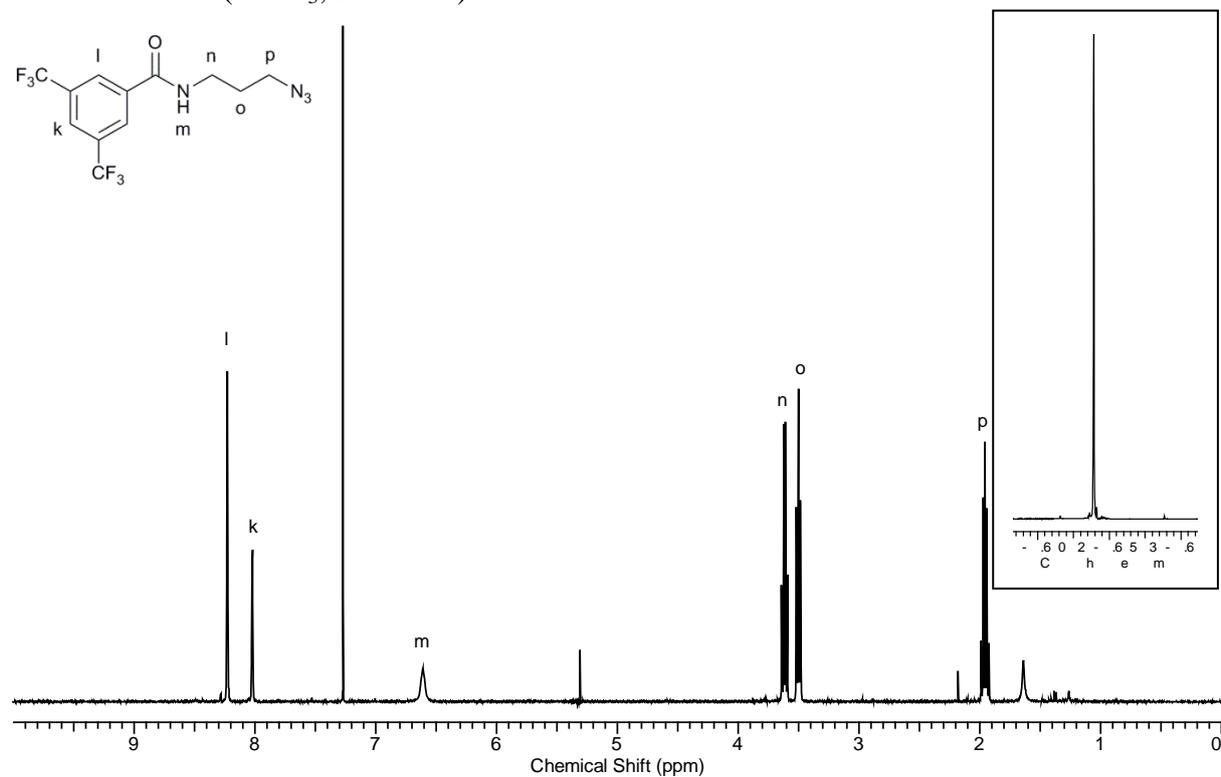
## Mass Spectrum (ES +ve)



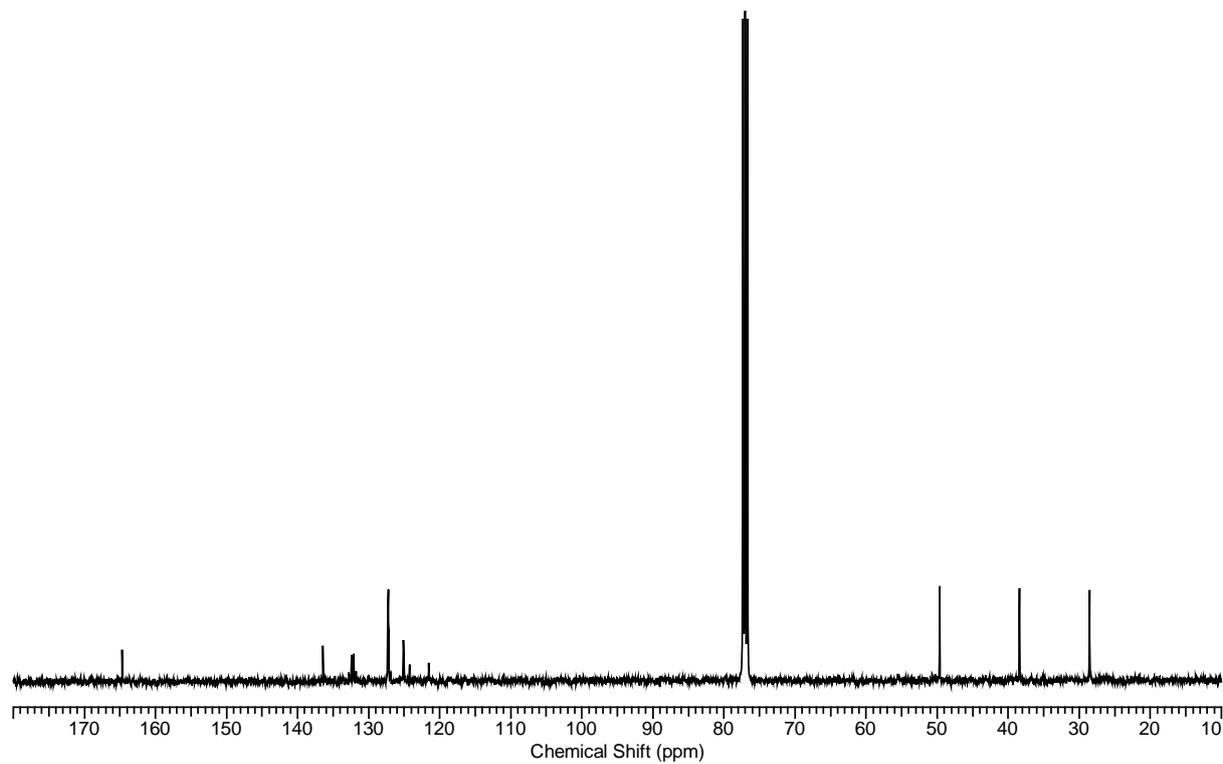
## Azide 5

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)

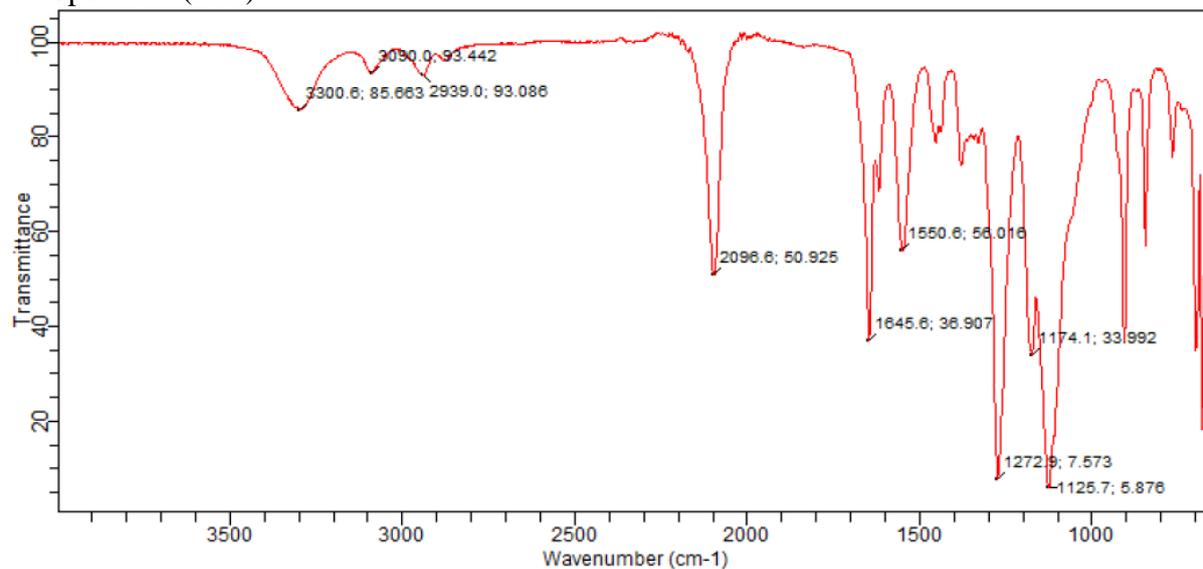


$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



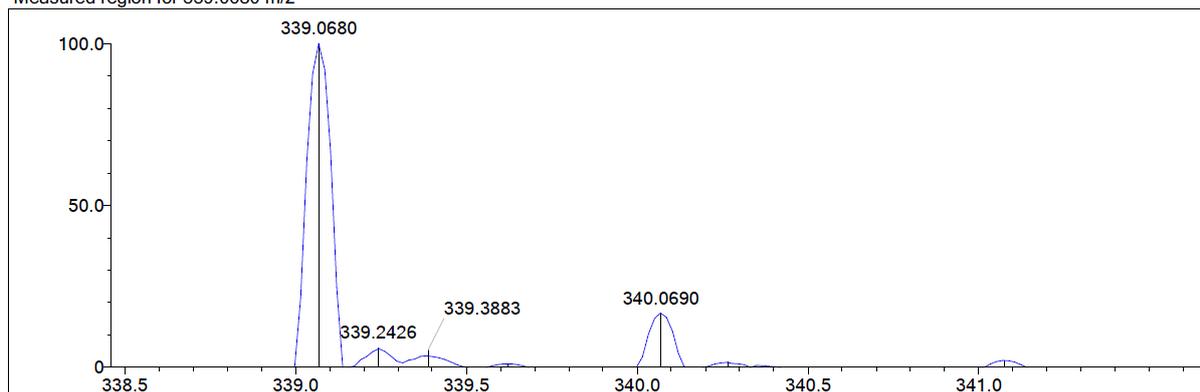
## Azide 5

IR Spectrum (neat)

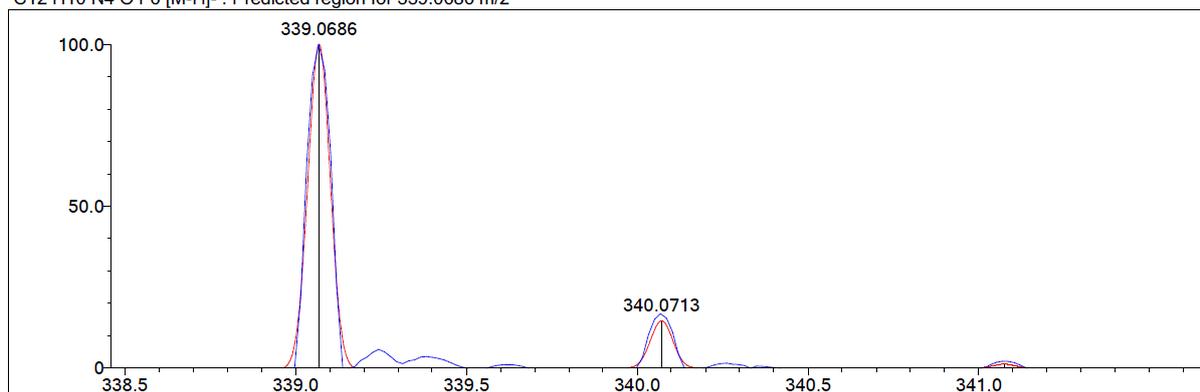


Mass Spectrum (APC -ve)

Measured region for 339.0680 m/z



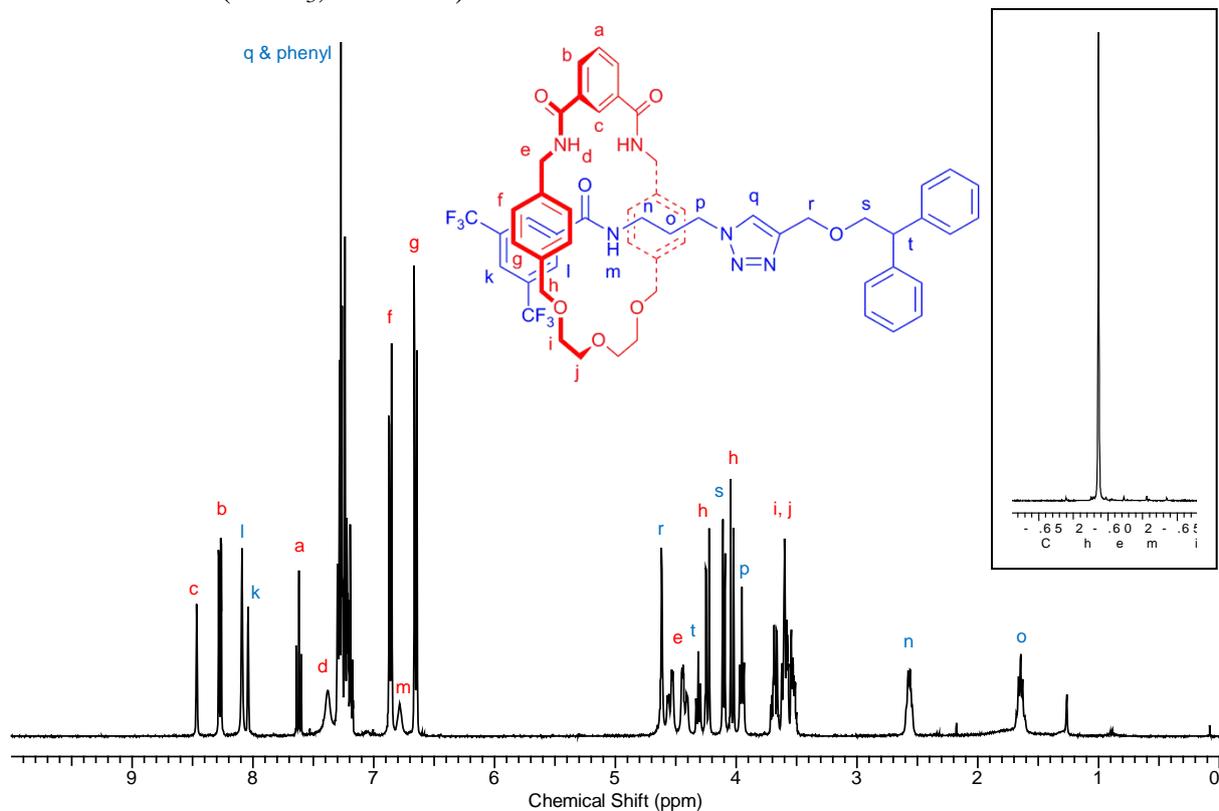
C12 H10 N4 O F6 [M-H]<sup>-</sup> : Predicted region for 339.0686 m/z



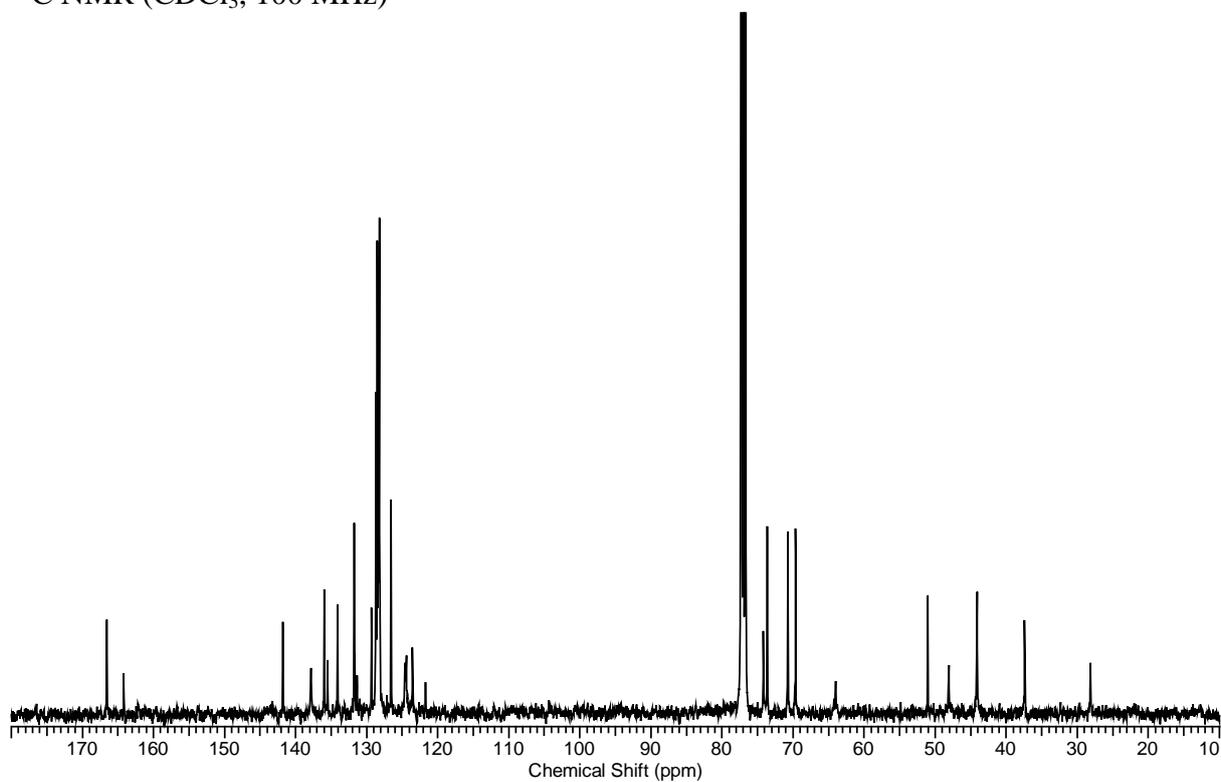
## Rotaxane 6

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)



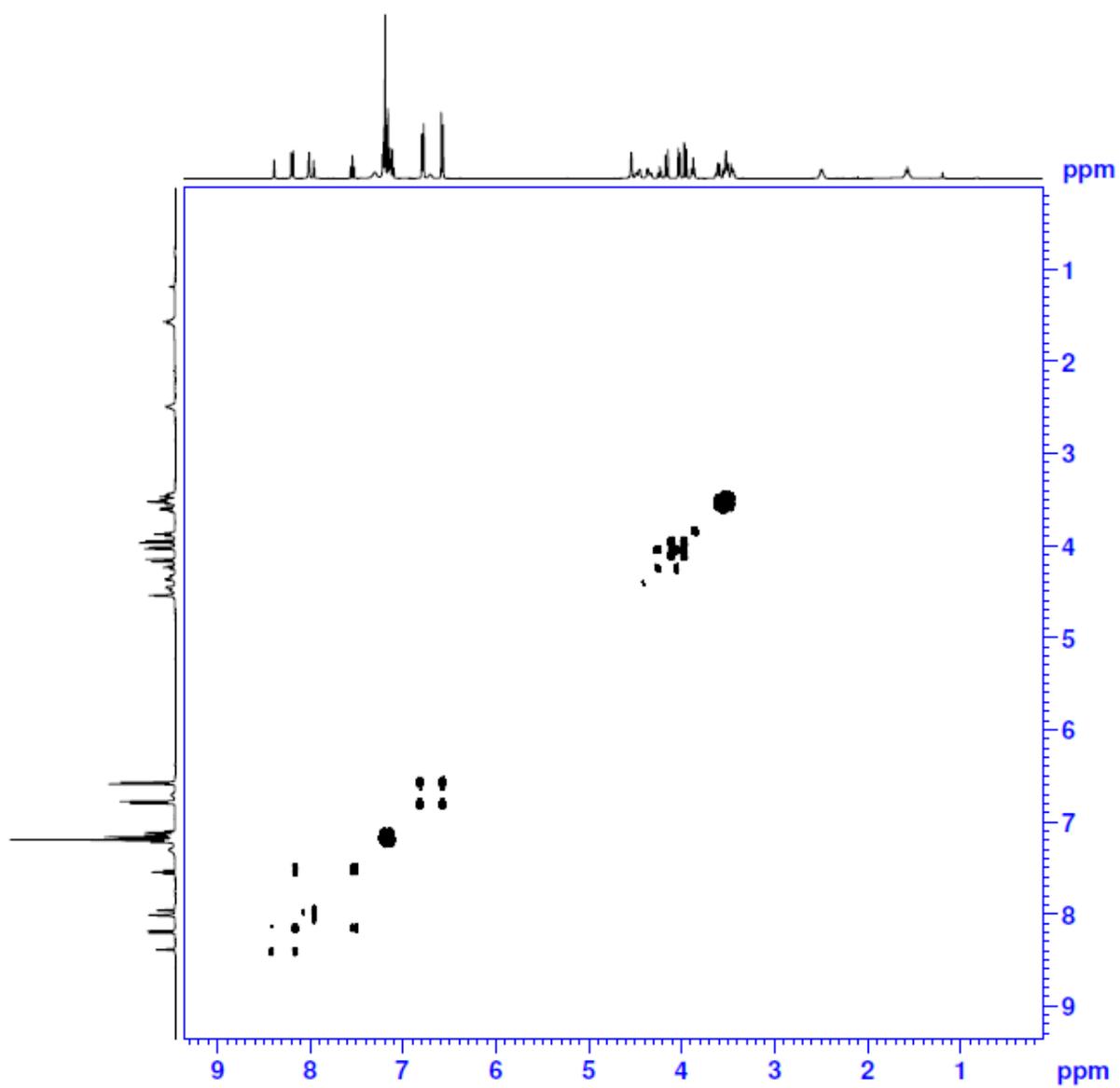
$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



*NB: Quartets arising from  $^1\text{J}$  and  $^2\text{J}$  C-F couplings not well resolved in this  $^{13}\text{C}$  NMR spectrum, and broad peaks in aromatic region provides evidence of coincident resonances.*

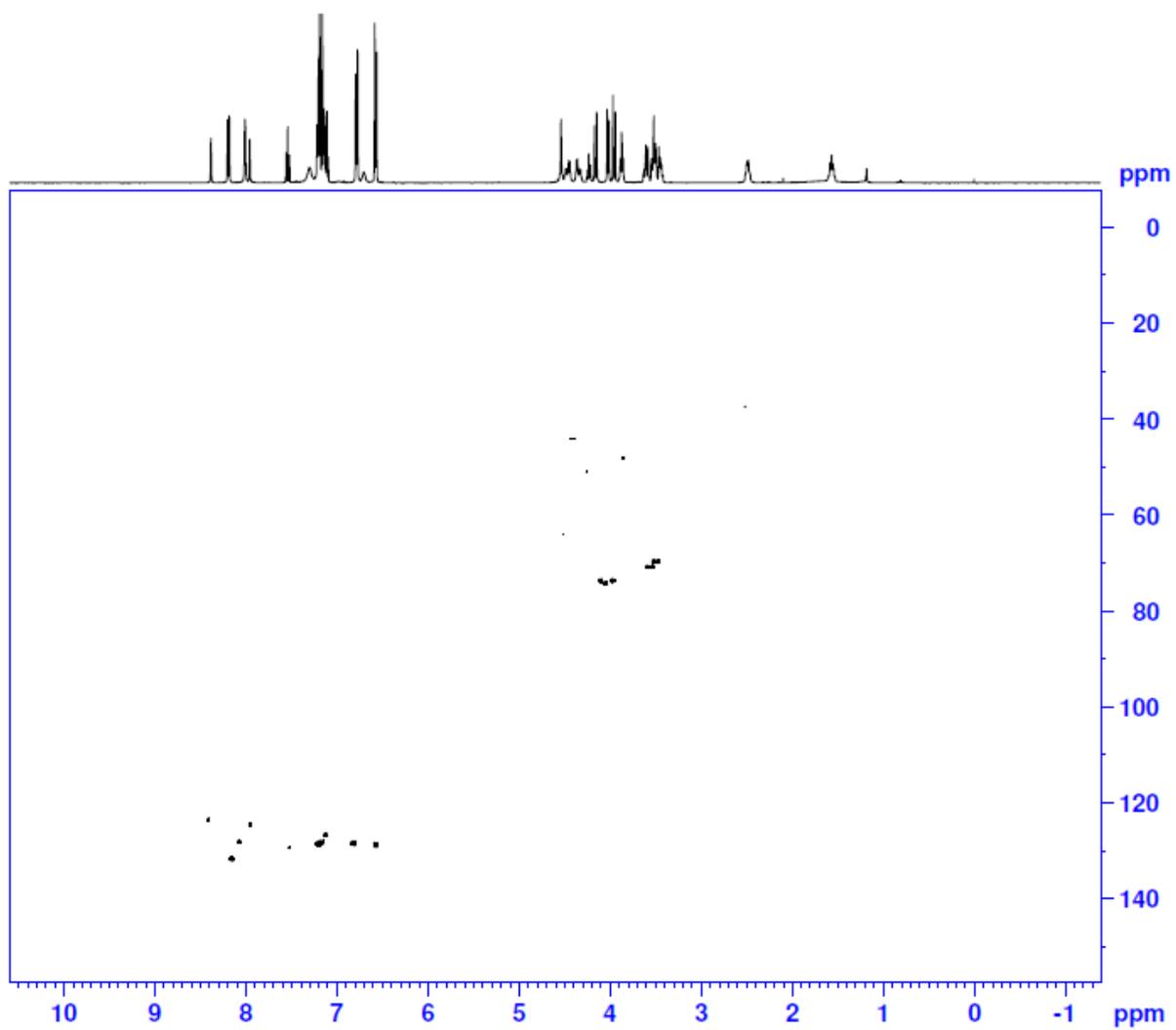
## Rotaxane 6

$^1\text{H}$ - $^1\text{H}$  COSY NMR ( $\text{CDCl}_3$ )



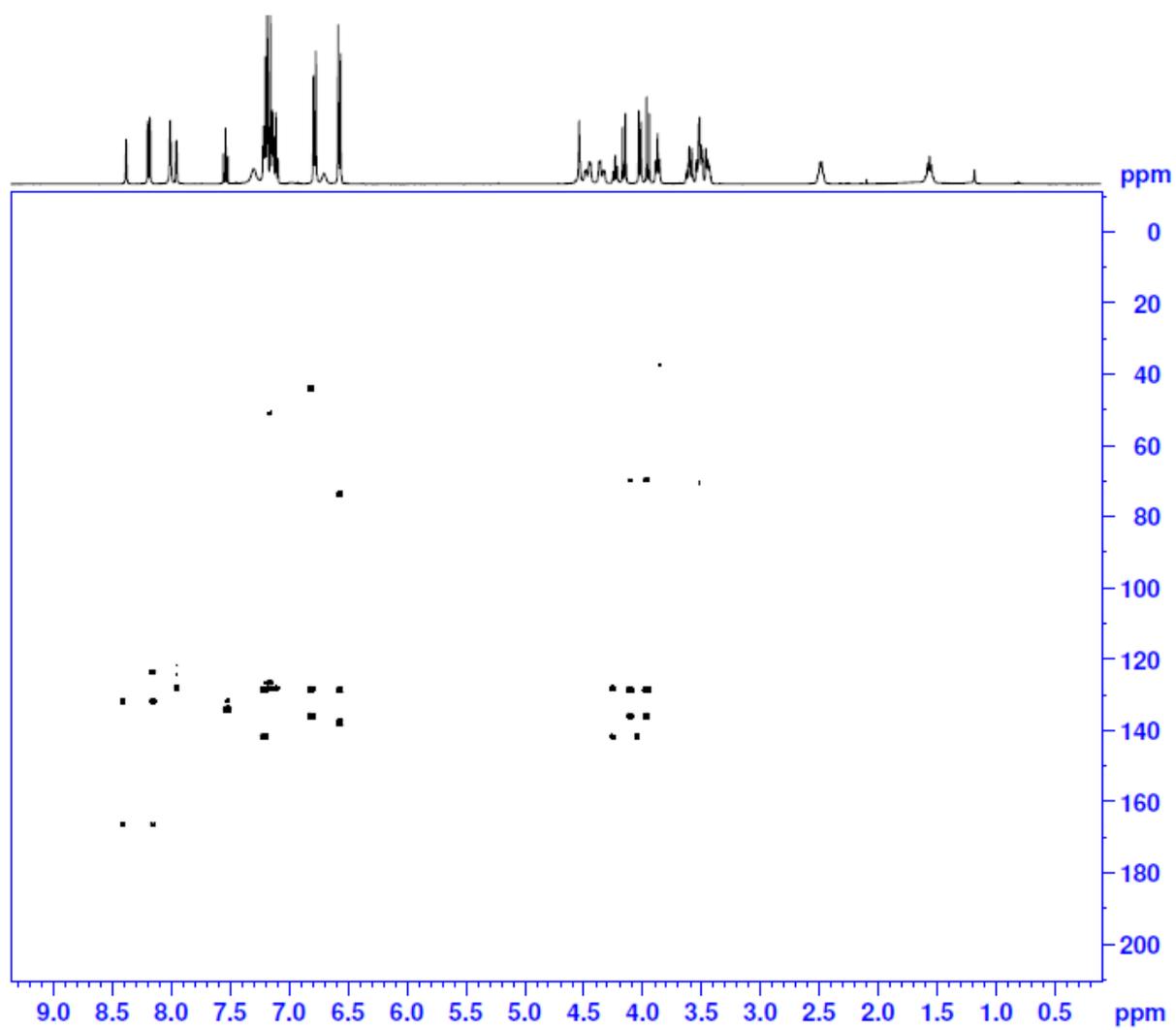
## Rotaxane 6

$^1\text{H}$ - $^{13}\text{C}$  HSQC NMR ( $\text{CDCl}_3$ )



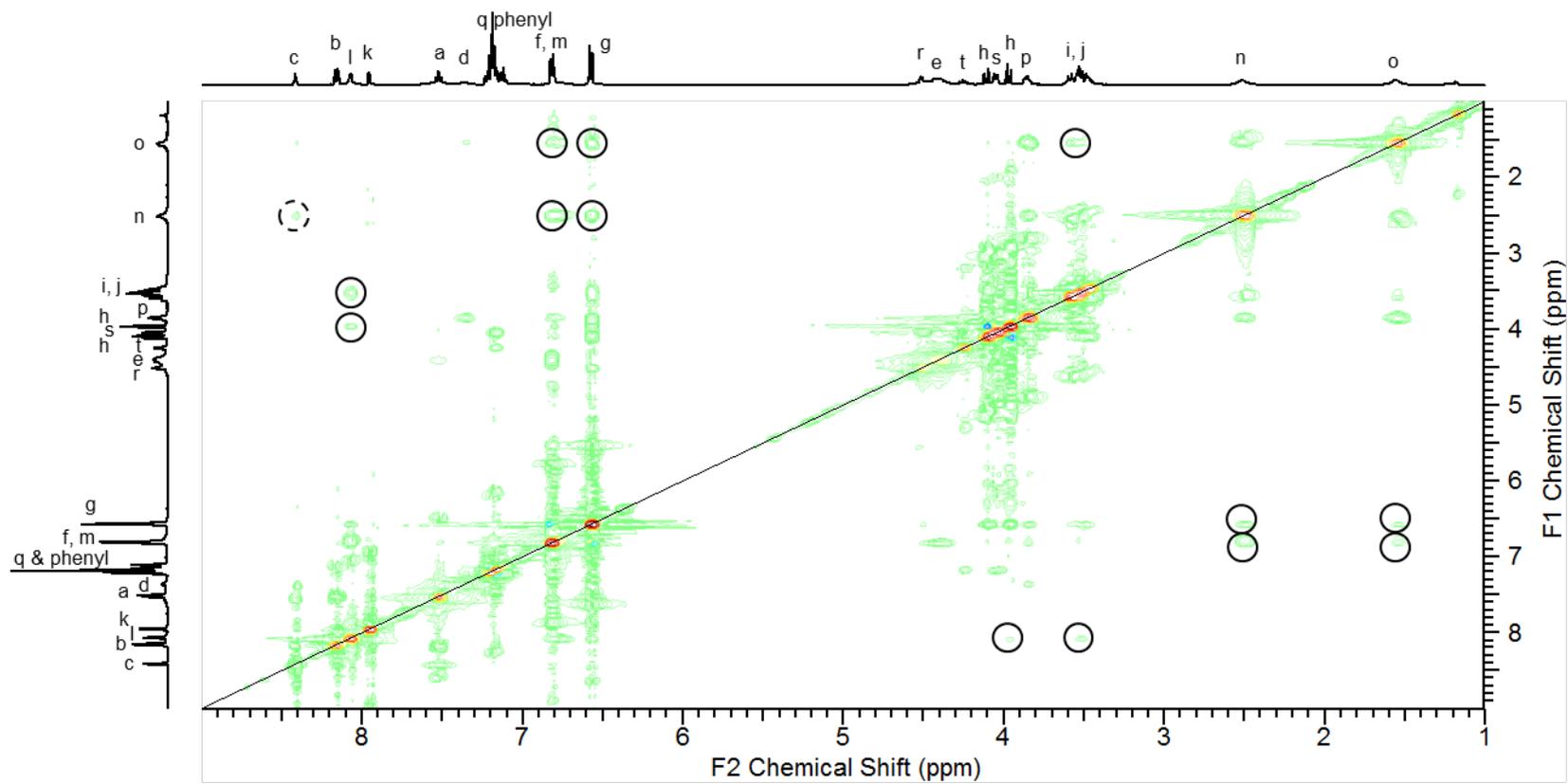
## Rotaxane 6

$^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $\text{CDCl}_3$ )



## Rotaxane 6

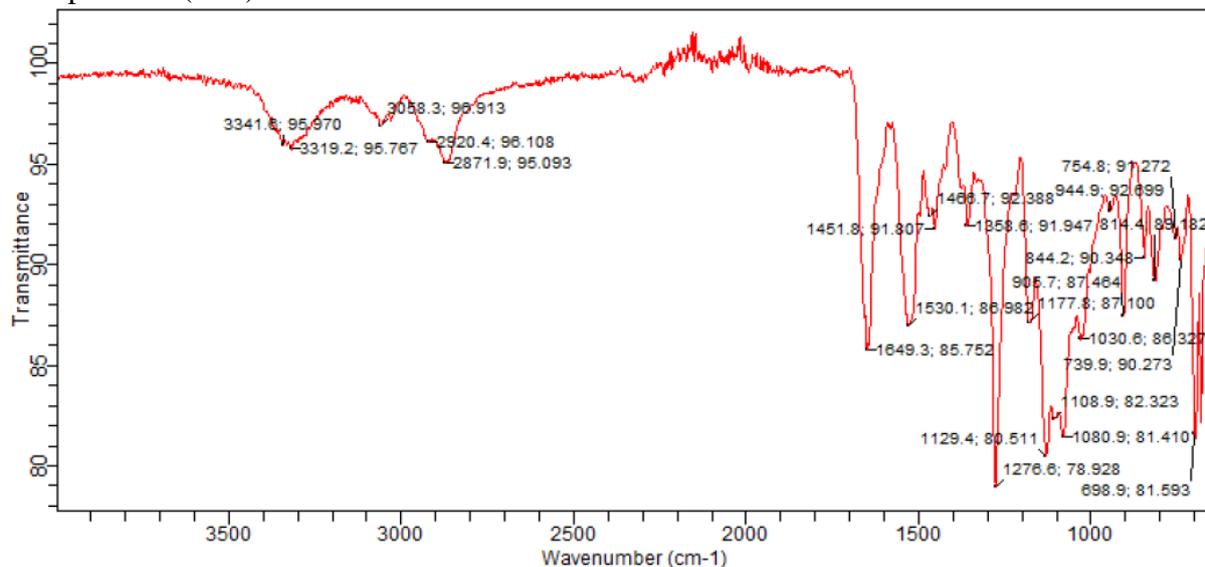
$^1\text{H}$ - $^1\text{H}$  ROESY NMR ( $\text{CDCl}_3$ )



*Inter-component through-space cross-peaks are circled. Cross-peaks that appear on only one side of the diagonal are enclosed by a dashed circle.*

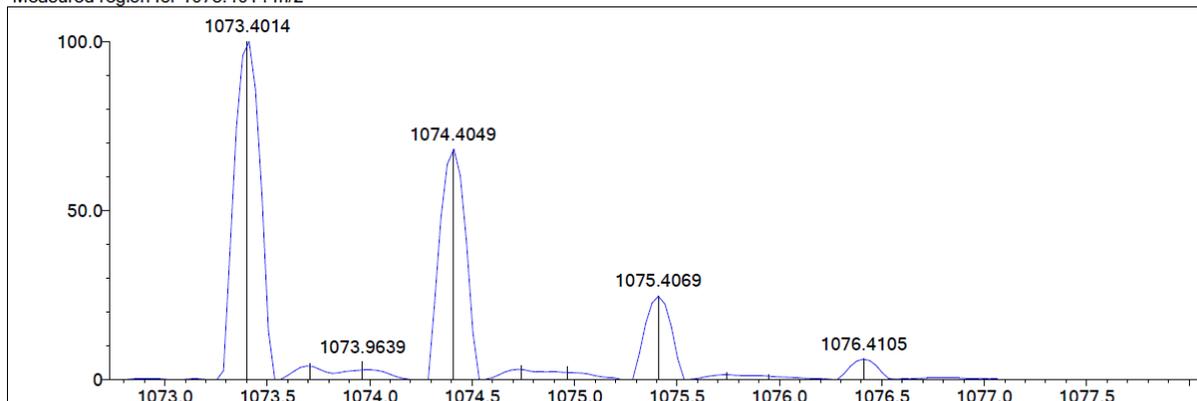
# Rotaxane 6

## IR Spectrum (neat)

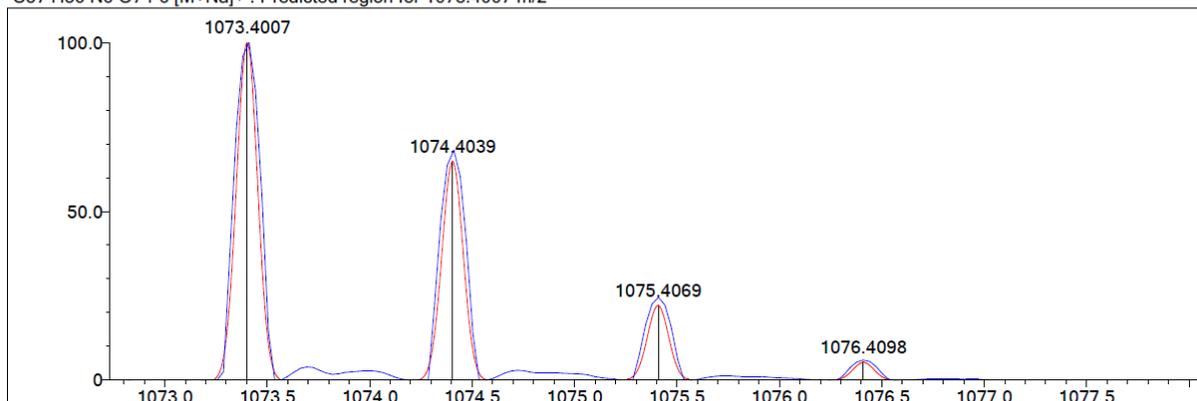


## Mass Spectrum (ES +ve)

Measured region for 1073.4014 m/z



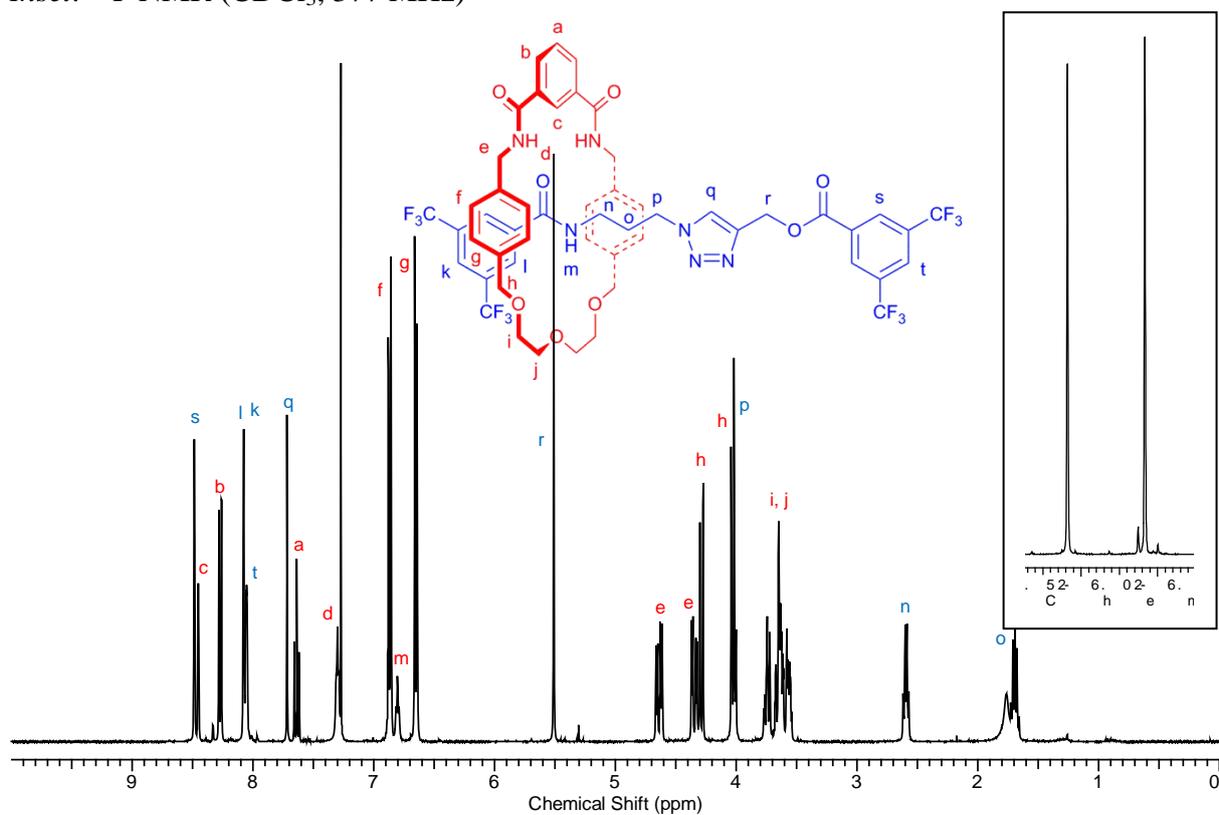
C57 H56 N6 O7 F6 [M+Na]<sup>+</sup> : Predicted region for 1073.4007 m/z



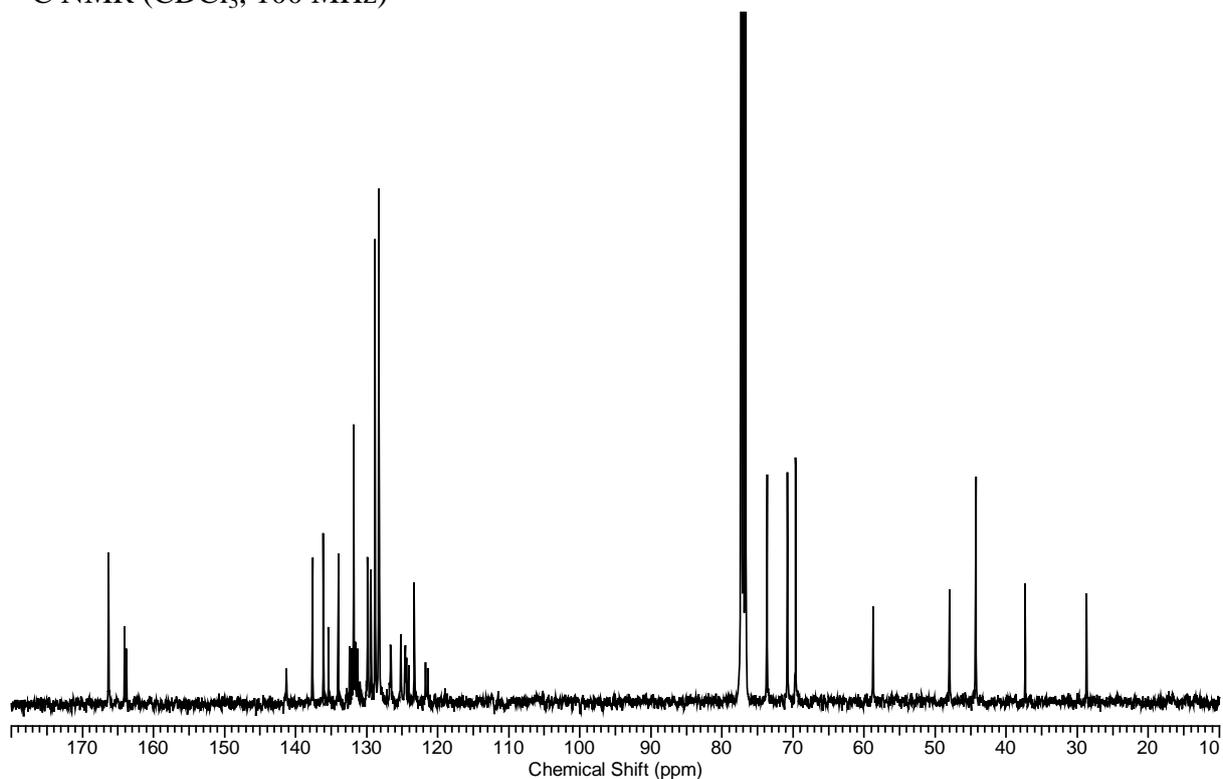
## Rotaxane 7

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)



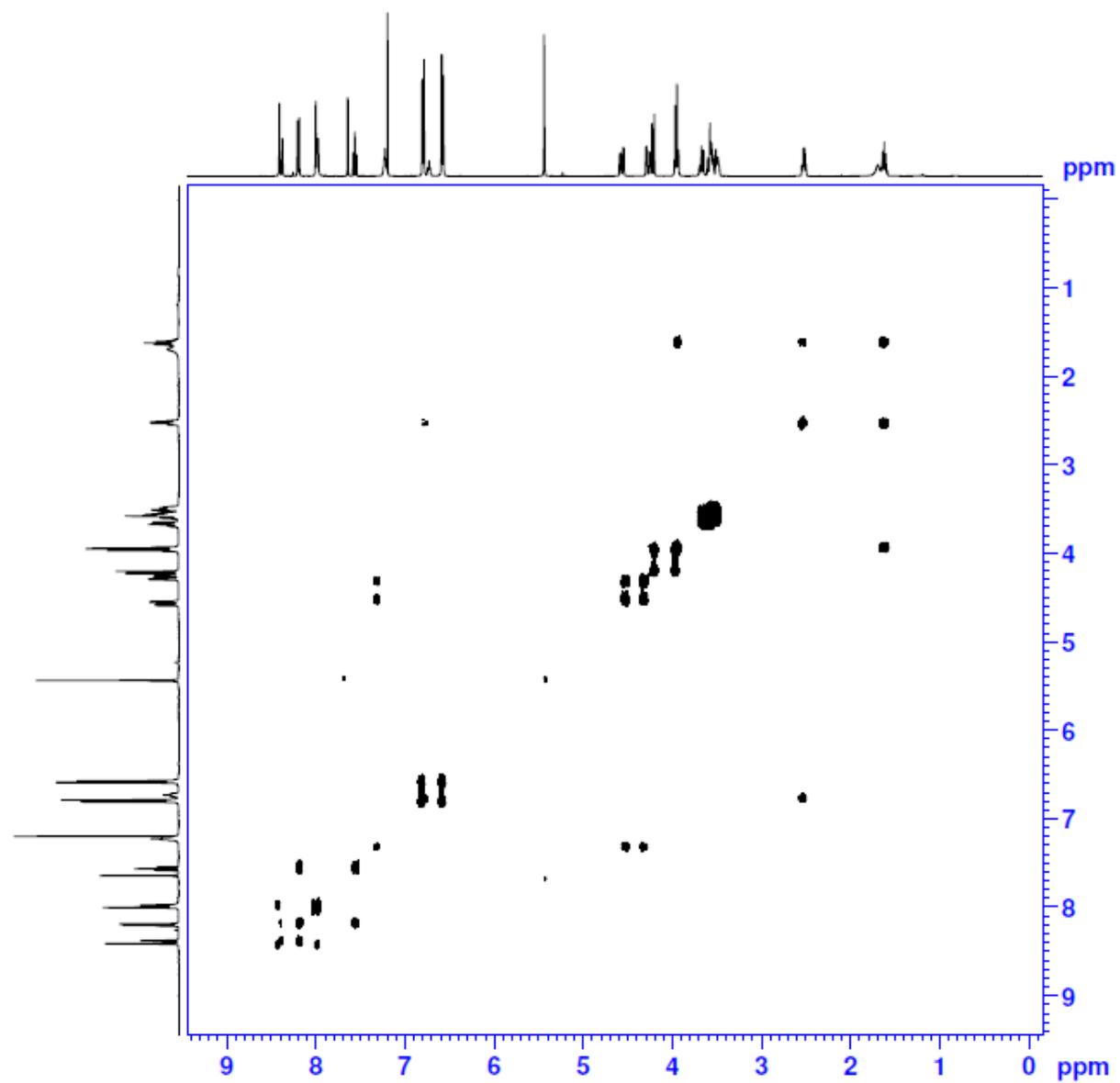
$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



*NB:* In this  $^{13}\text{C}$  NMR spectrum, broad peaks in aromatic region provides evidence of coincident resonances.

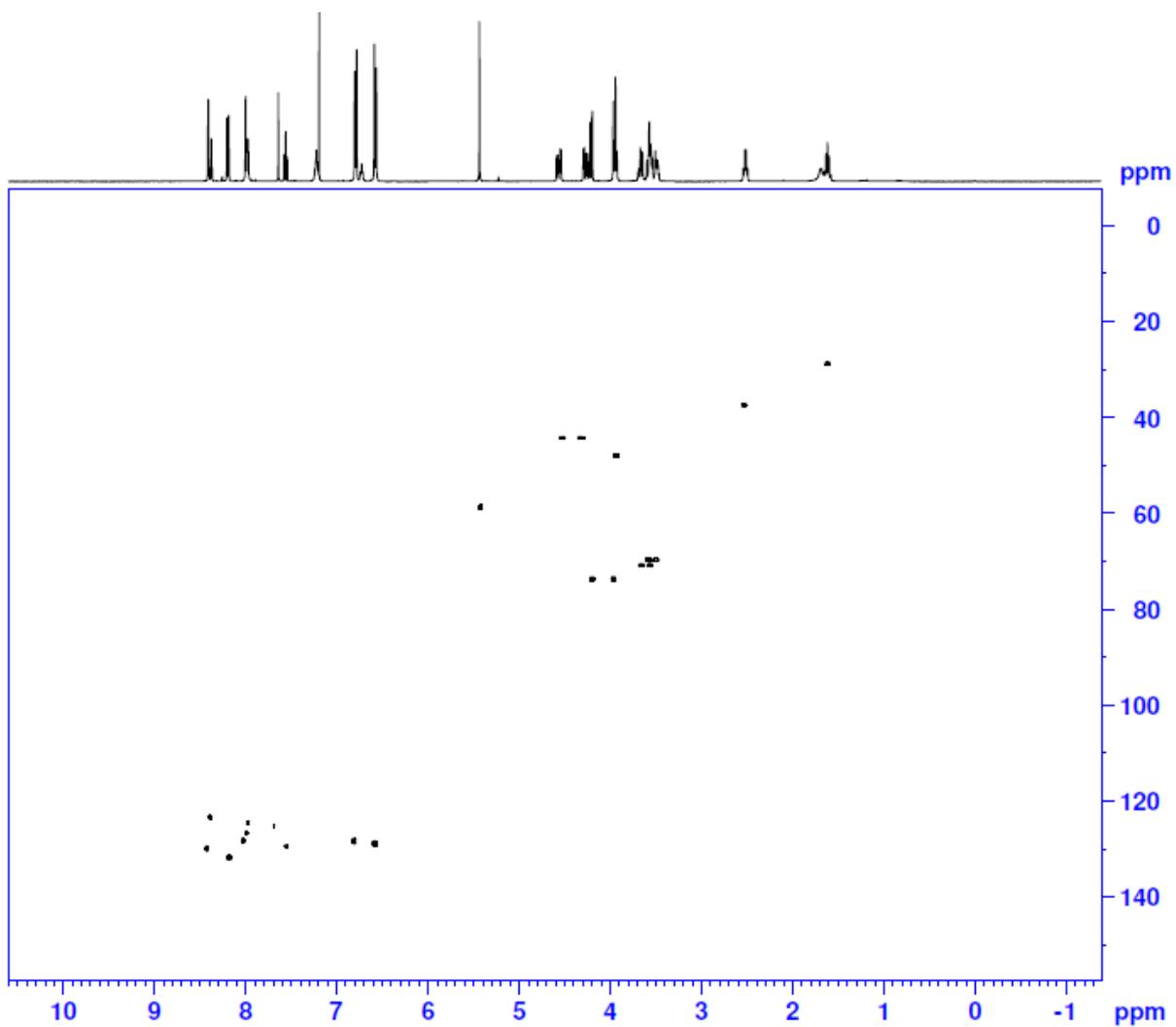
# Rotaxane 7

$^1\text{H}$ - $^1\text{H}$  COSY NMR ( $\text{CDCl}_3$ )



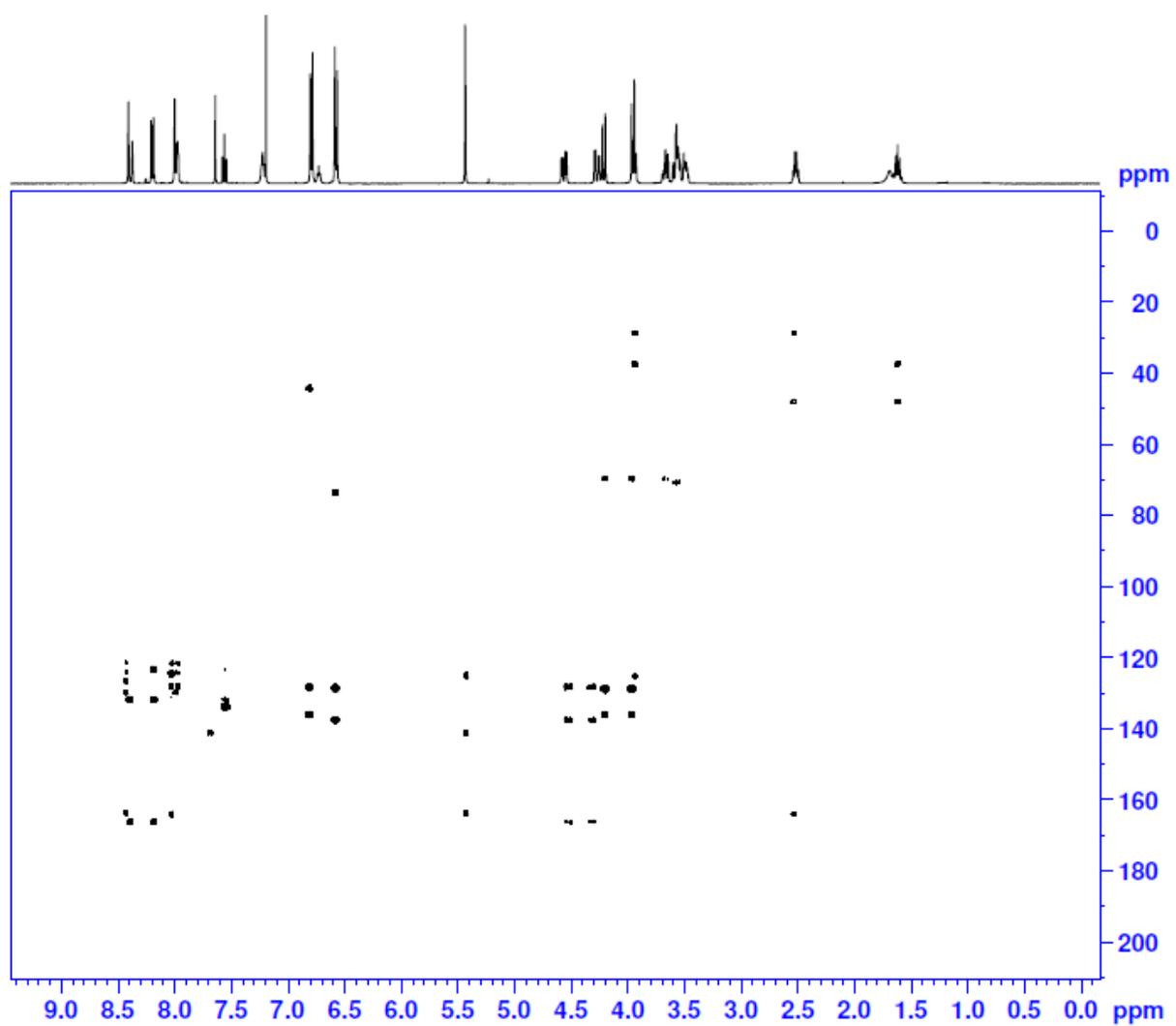
## Rotaxane 7

$^1\text{H}$ - $^{13}\text{C}$  HSQC NMR ( $\text{CDCl}_3$ )



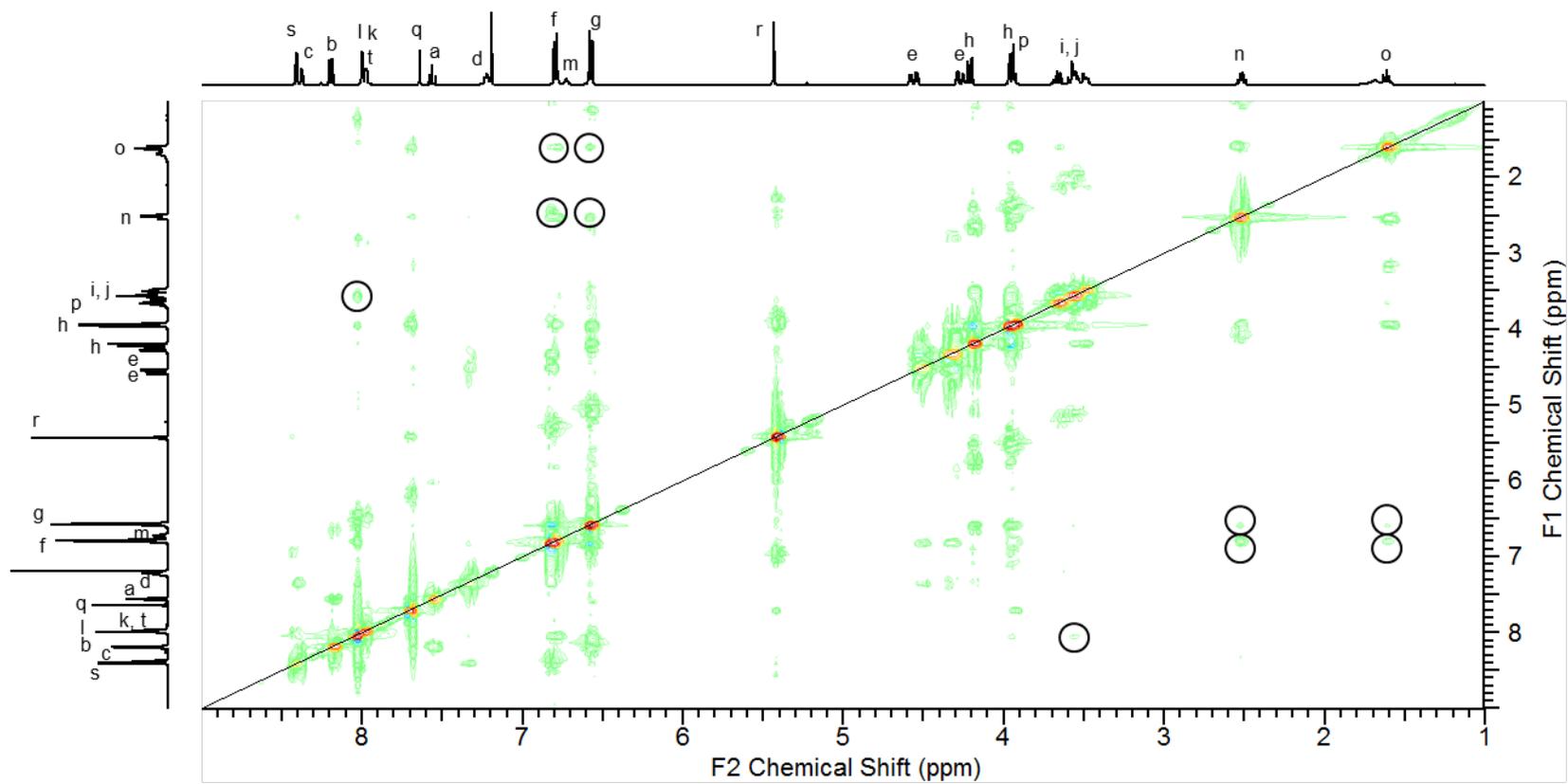
# Rotaxane 7

$^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $\text{CDCl}_3$ )



# Rotaxane 7

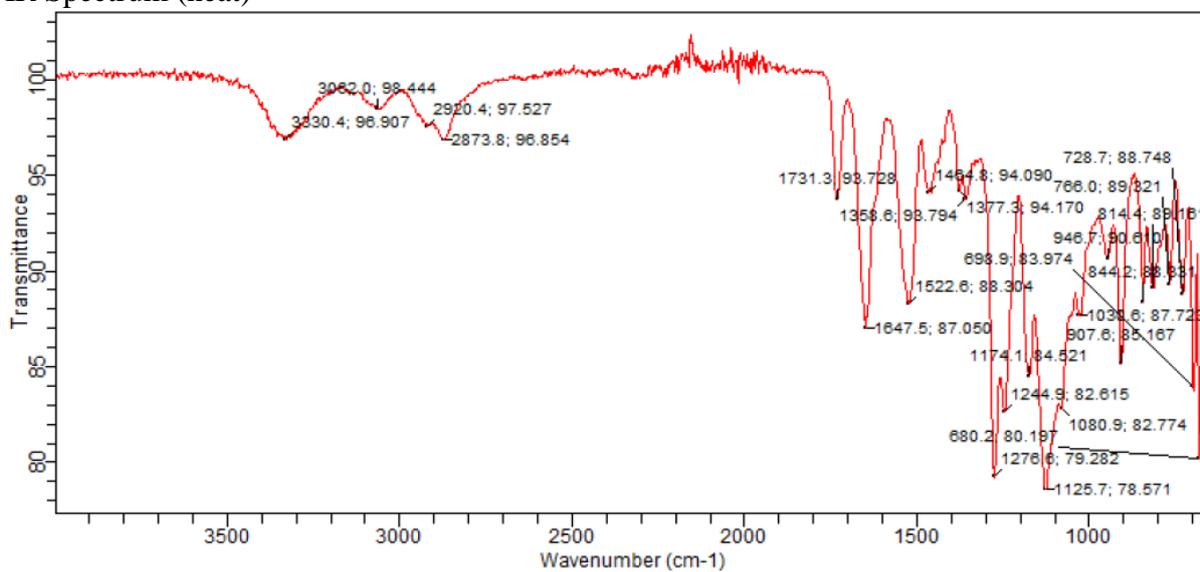
$^1\text{H}$ - $^1\text{H}$  ROESY NMR ( $\text{CDCl}_3$ )



*Inter-component through-space cross-peaks are circled.*

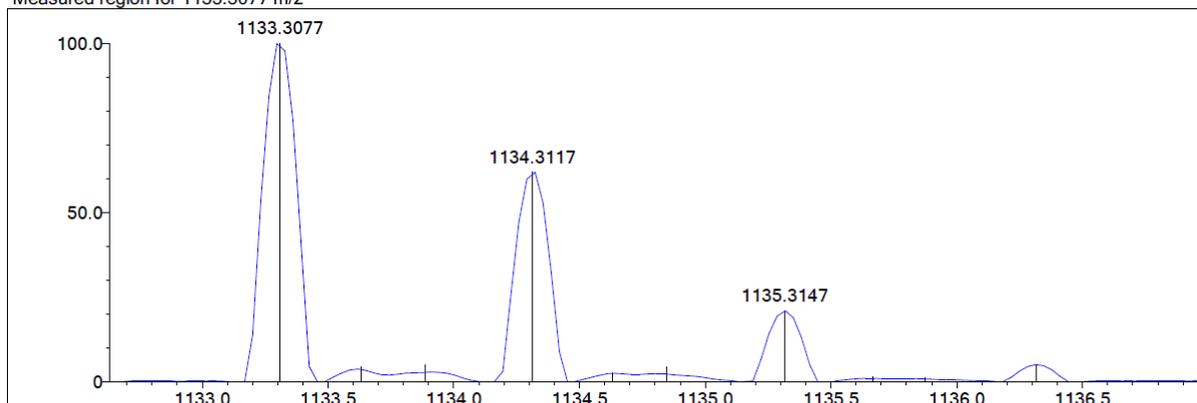
# Rotaxane 7

## IR Spectrum (neat)

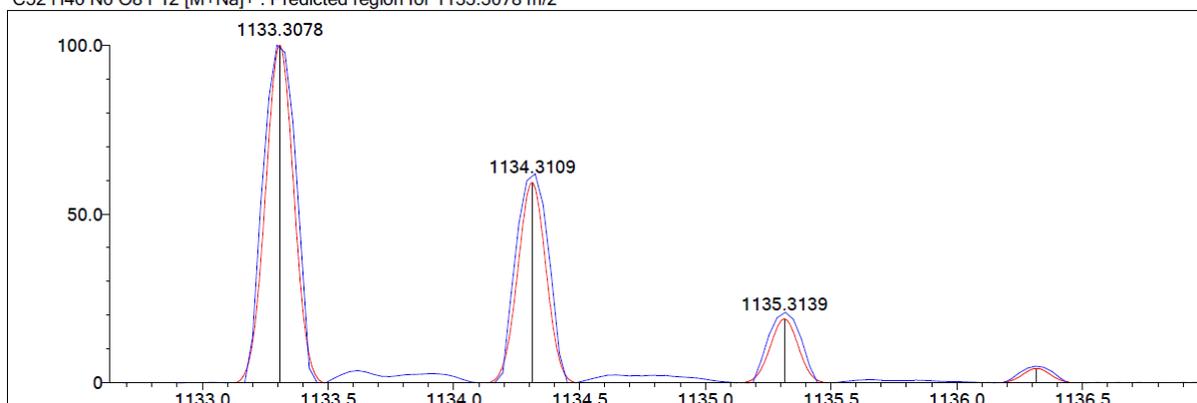


## Mass Spectrum (ES +ve)

Measured region for 1133.3077 m/z



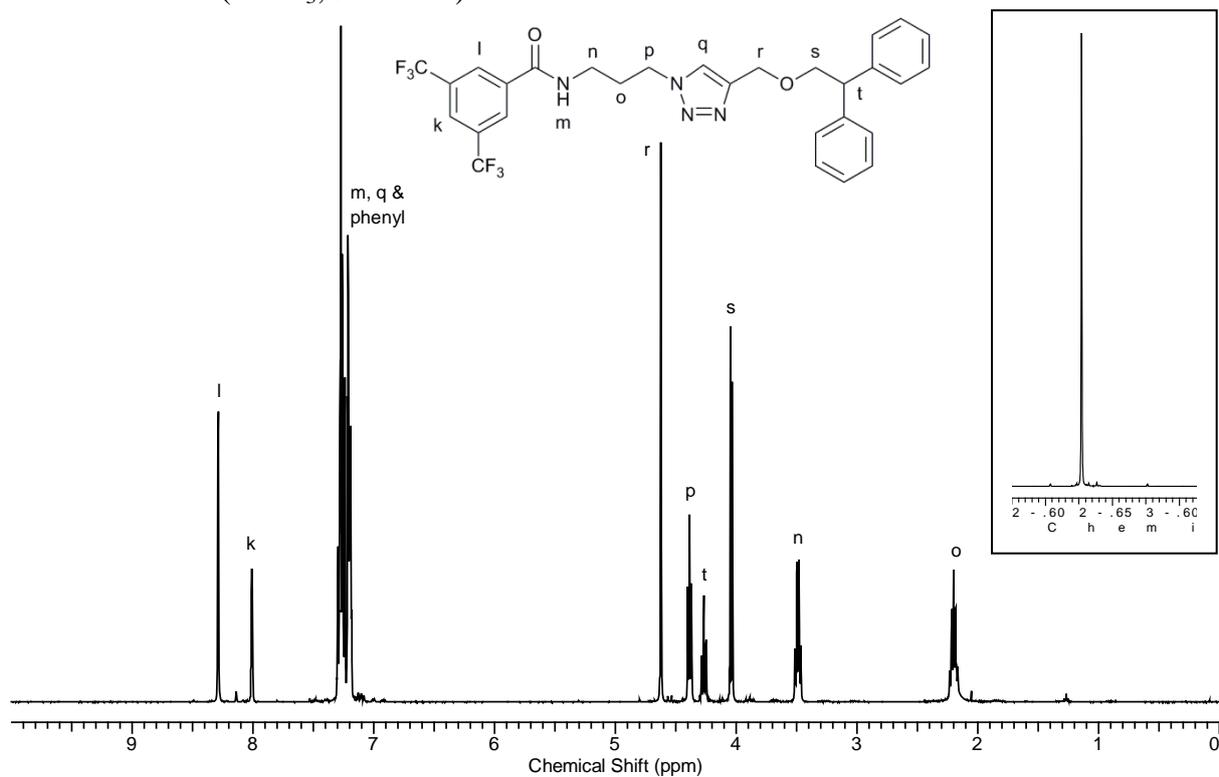
C52 H46 N6 O8 F12 [M+Na]<sup>+</sup> : Predicted region for 1133.3078 m/z



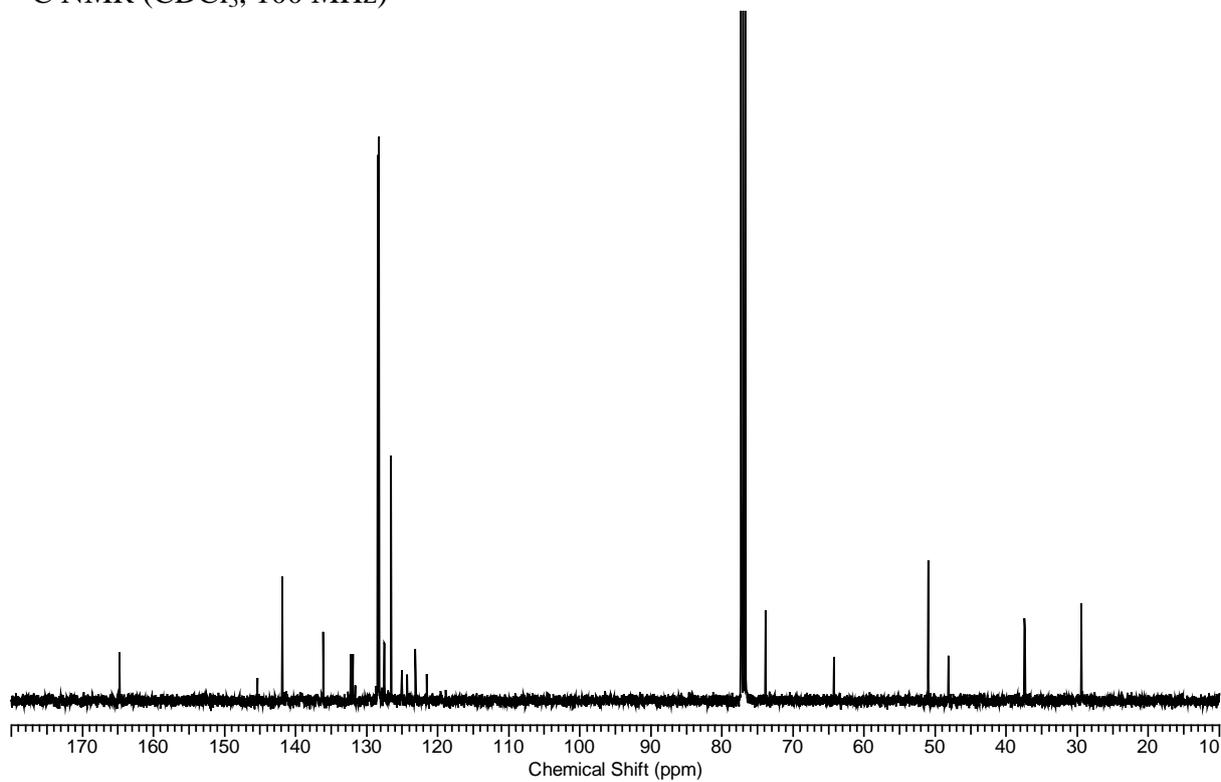
## Axle 8

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)

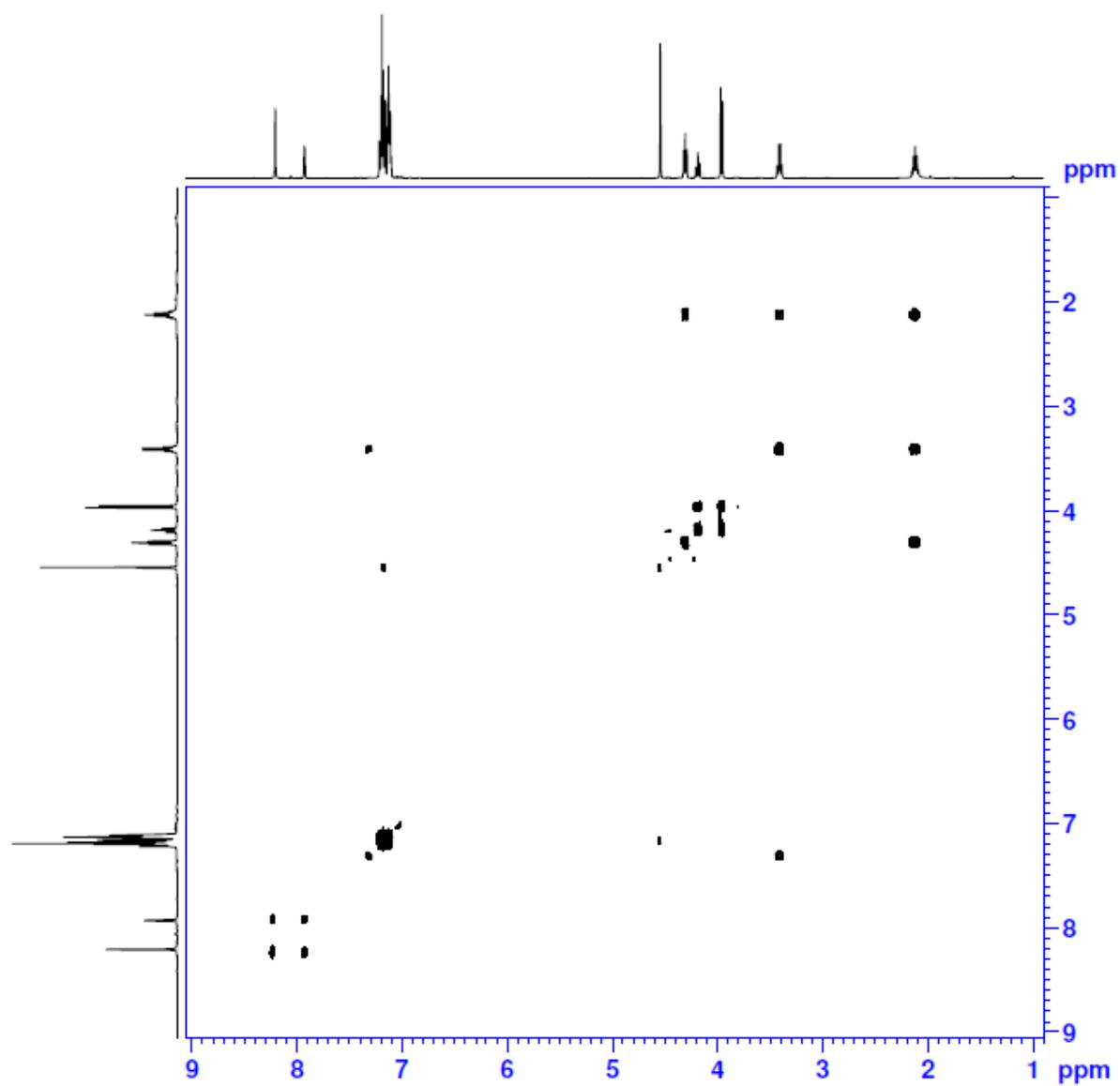


$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



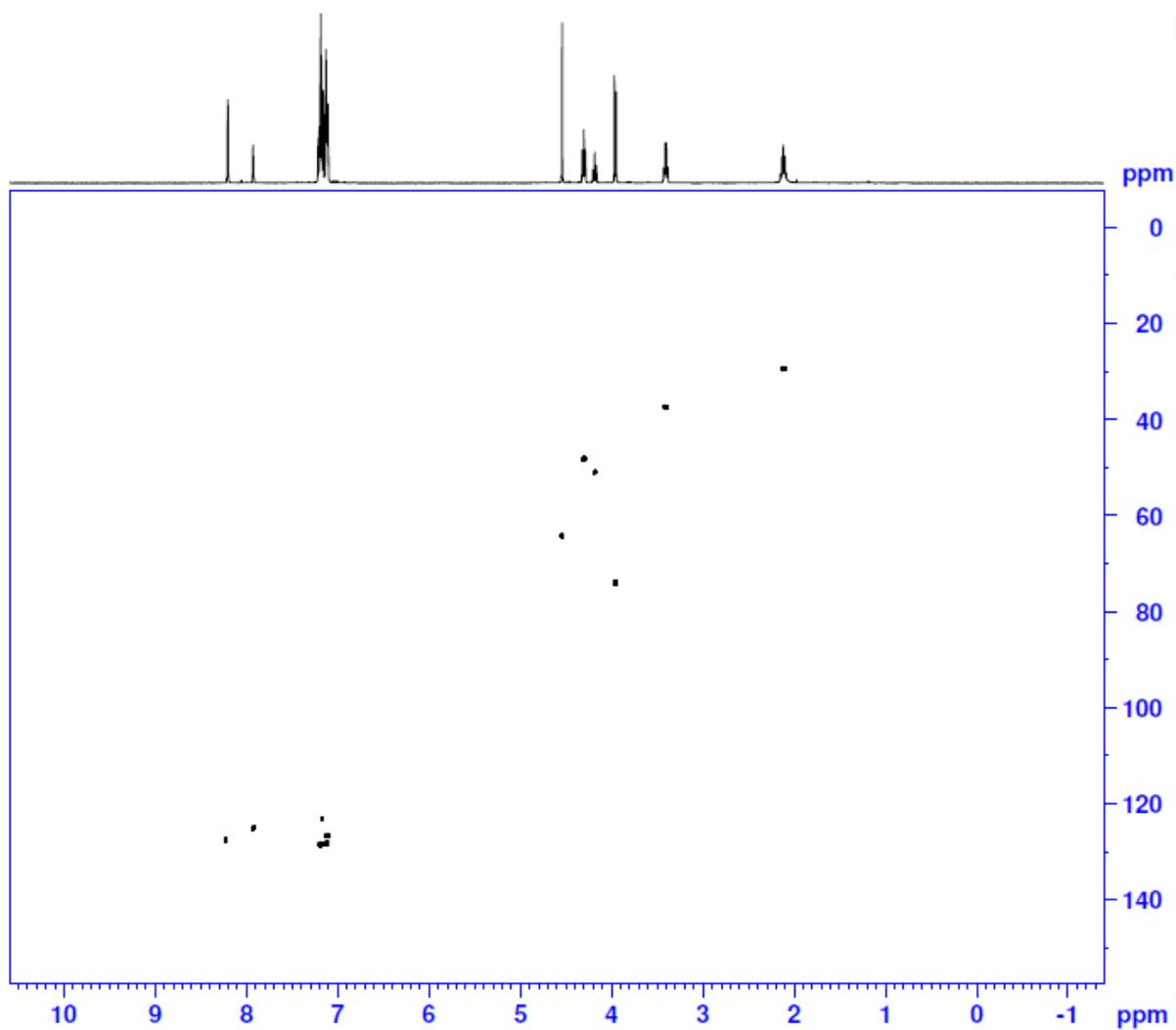
**Axle 8**

$^1\text{H}$ - $^1\text{H}$  COSY NMR ( $\text{CDCl}_3$ )



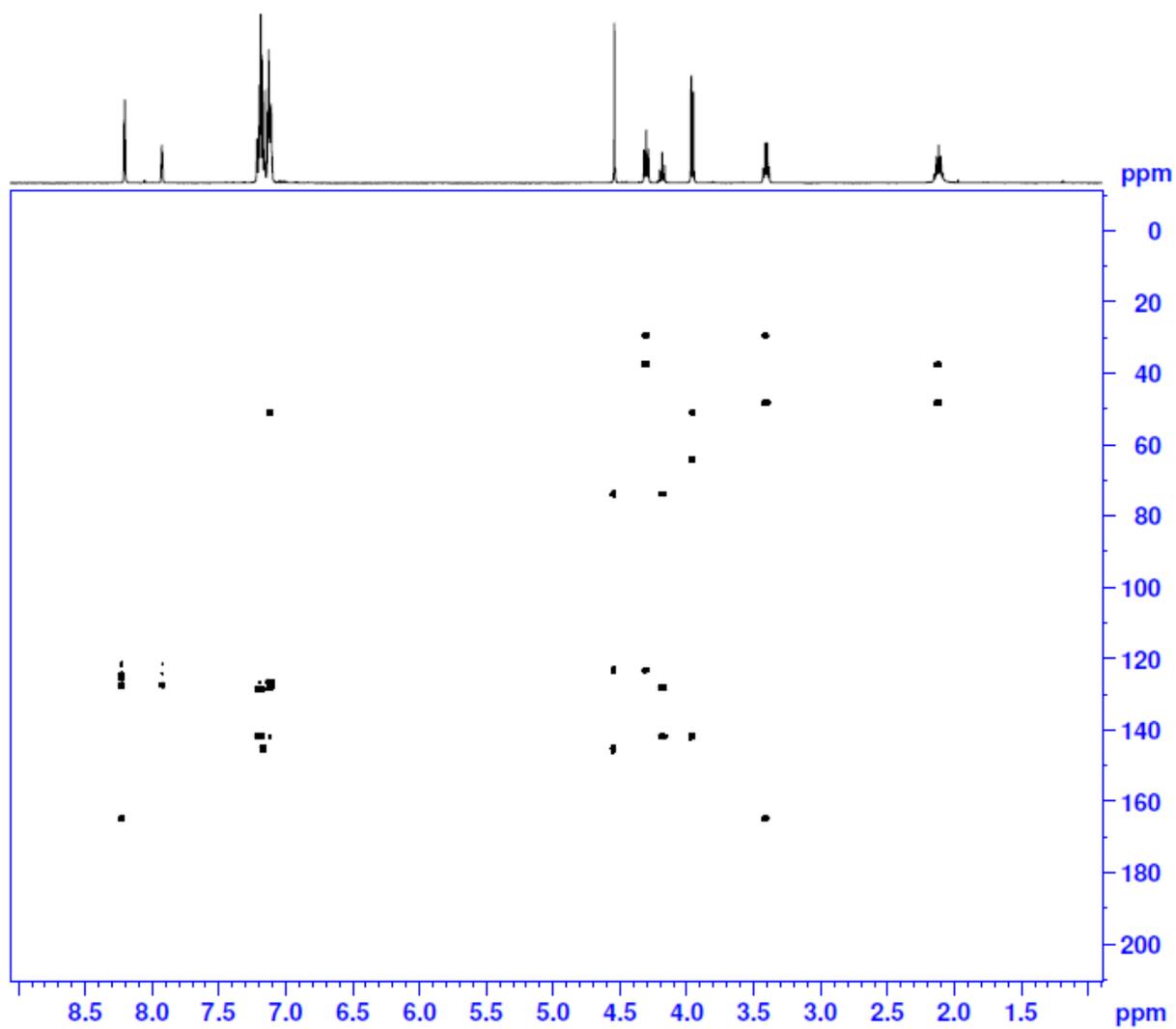
**Axle 8**

$^1\text{H}$ - $^{13}\text{C}$  HSQC NMR ( $\text{CDCl}_3$ )



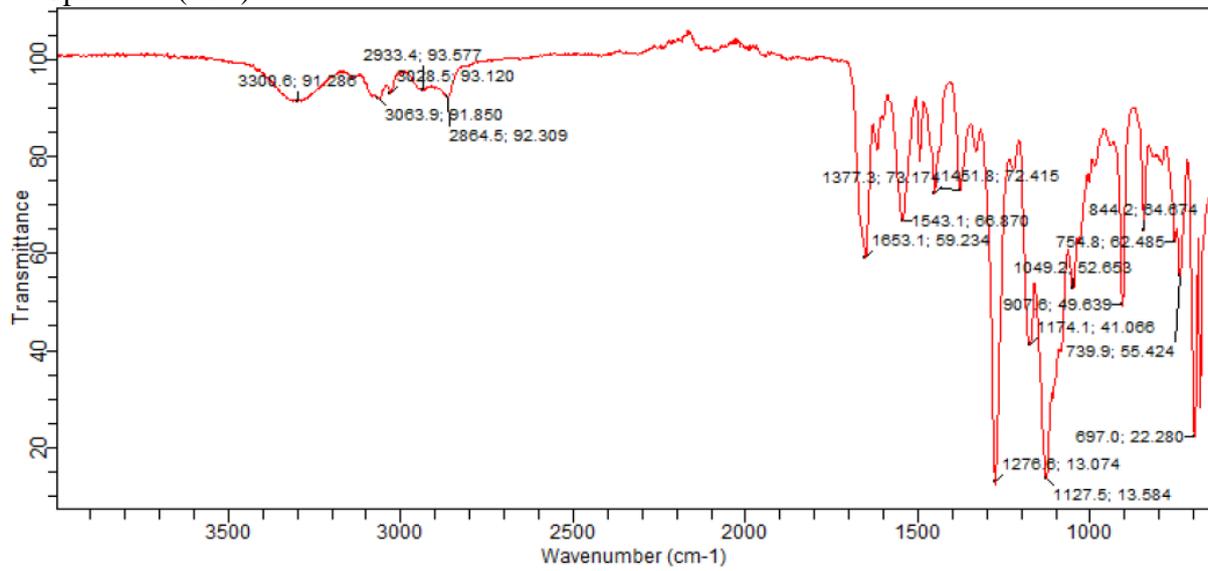
**Axle 8**

$^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $\text{CDCl}_3$ )



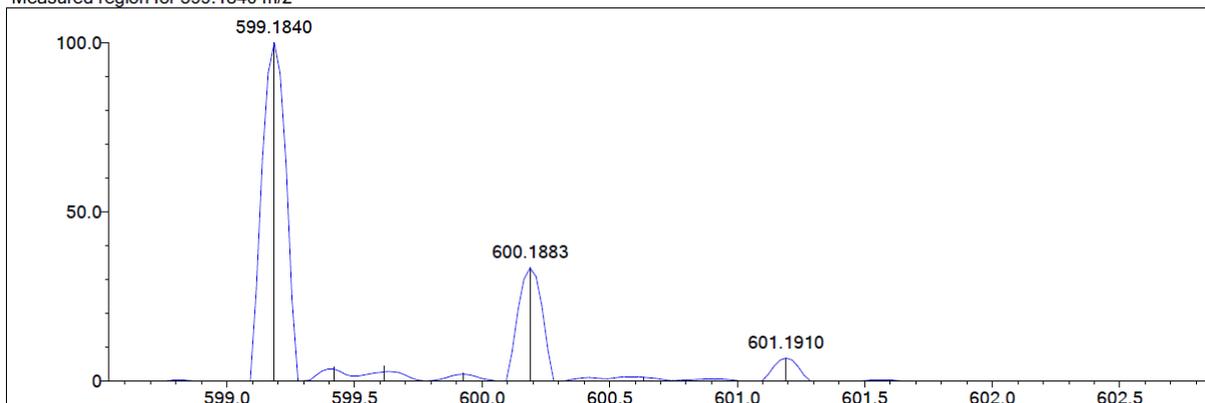
## Axle 8

### IR Spectrum (neat)

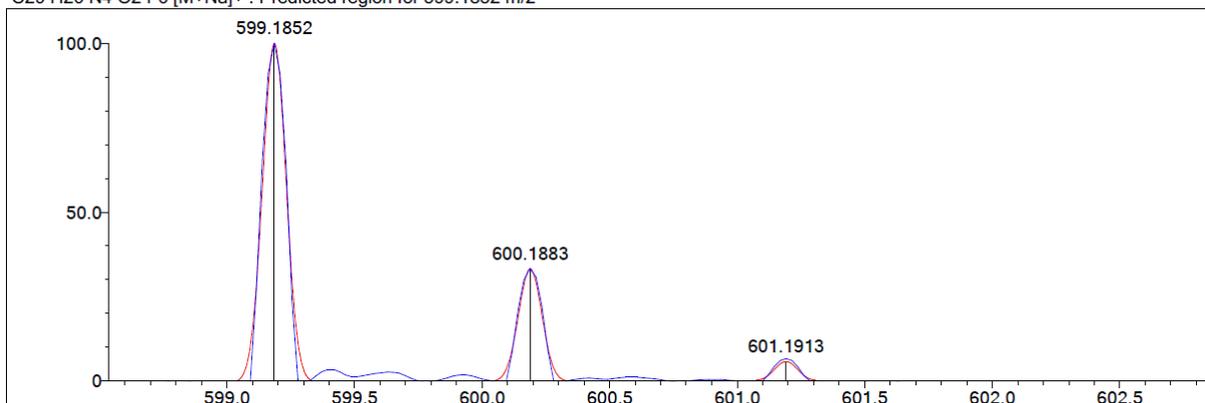


### Mass Spectrum (ES +ve)

Measured region for 599.1840 m/z



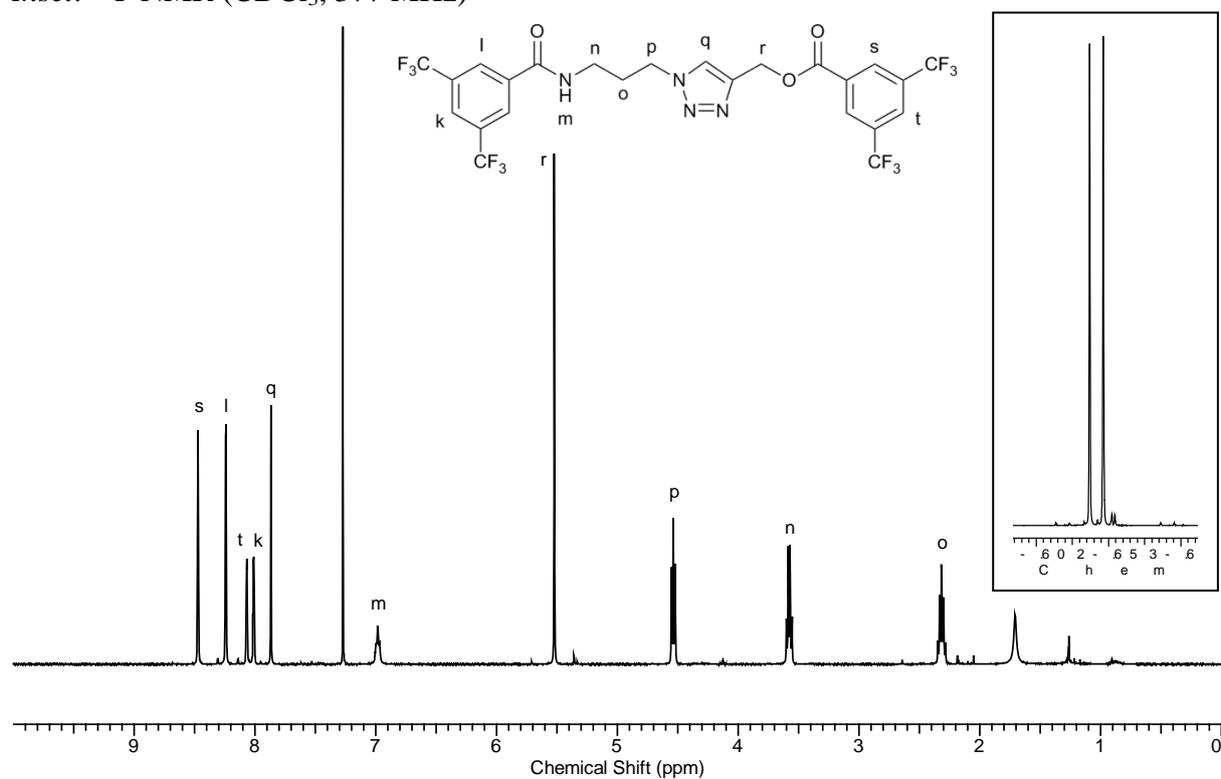
C29 H26 N4 O2 F6 [M+Na]<sup>+</sup> : Predicted region for 599.1852 m/z



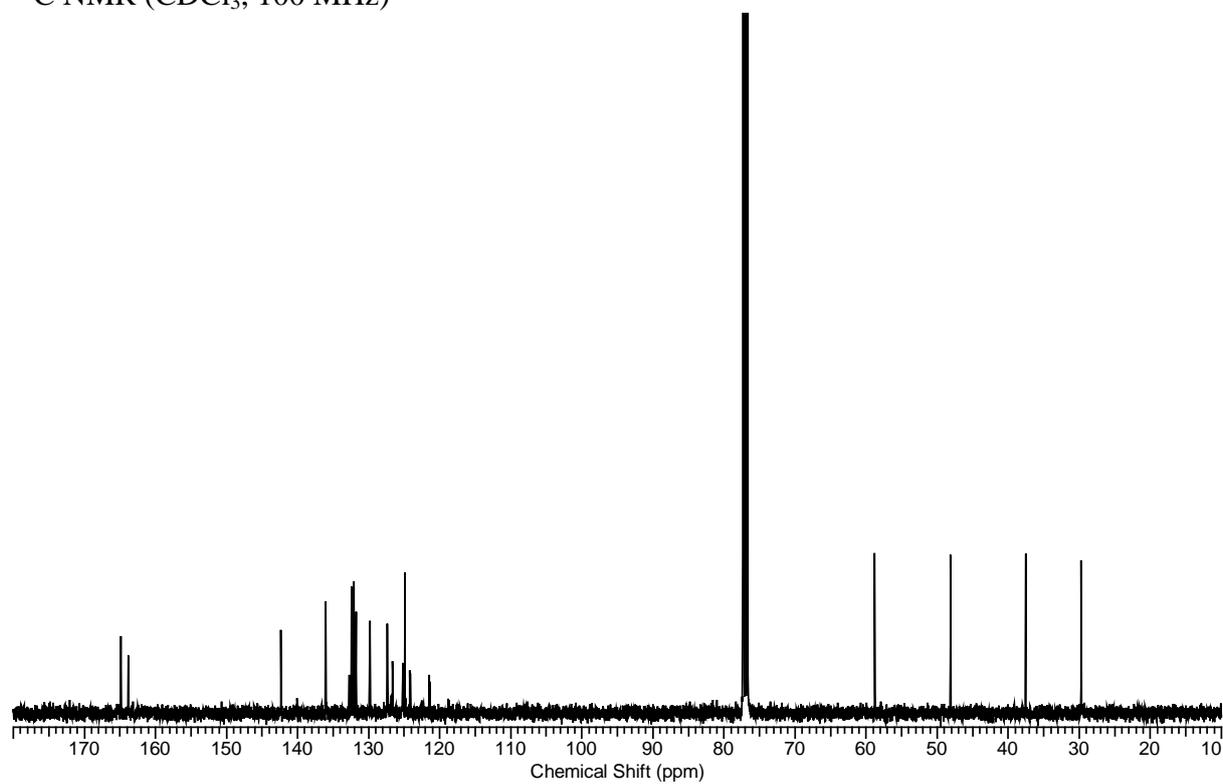
## Axle 9

$^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)

*Inset:*  $^{19}\text{F}$  NMR ( $\text{CDCl}_3$ , 377 MHz)

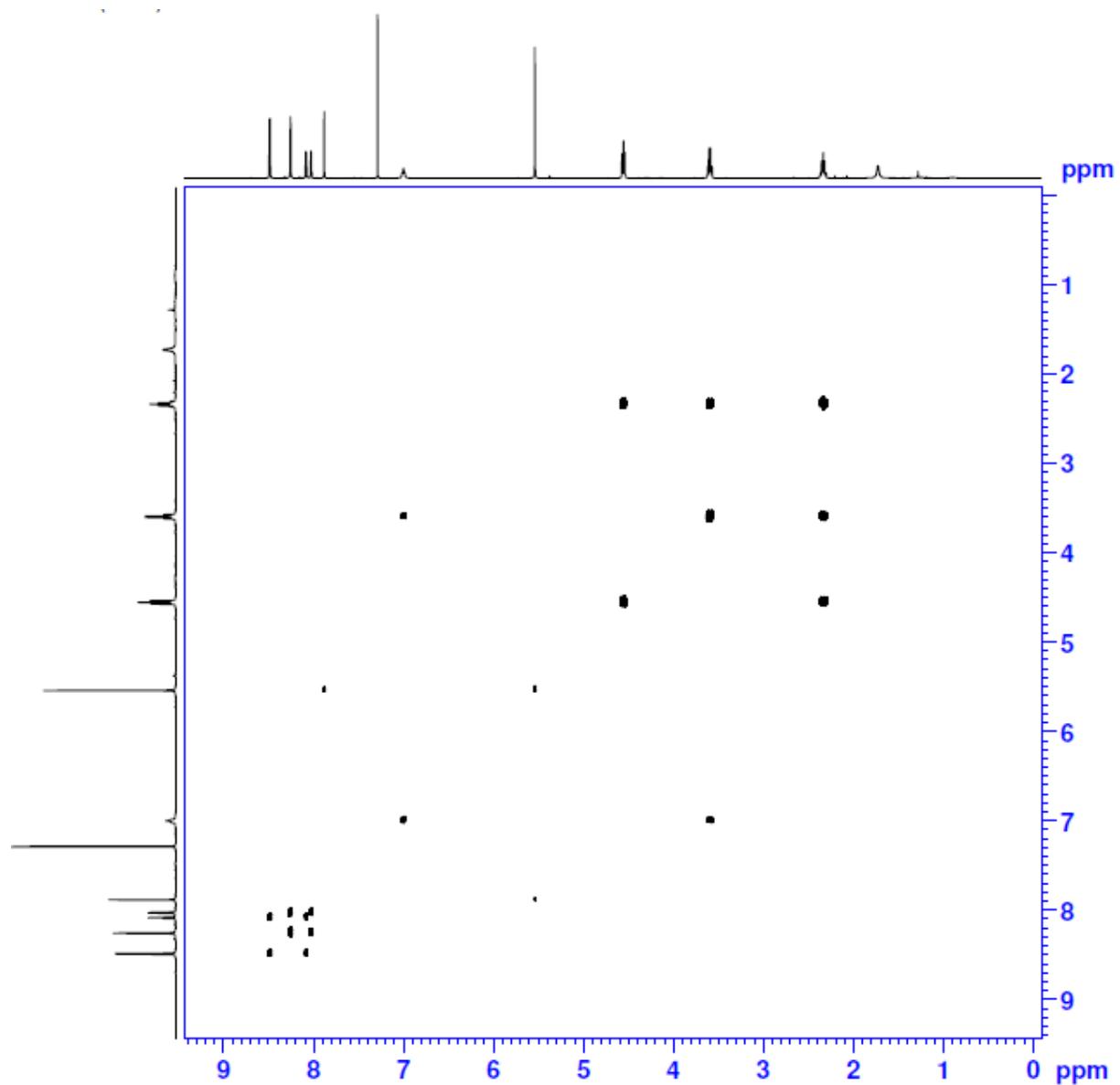


$^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz)



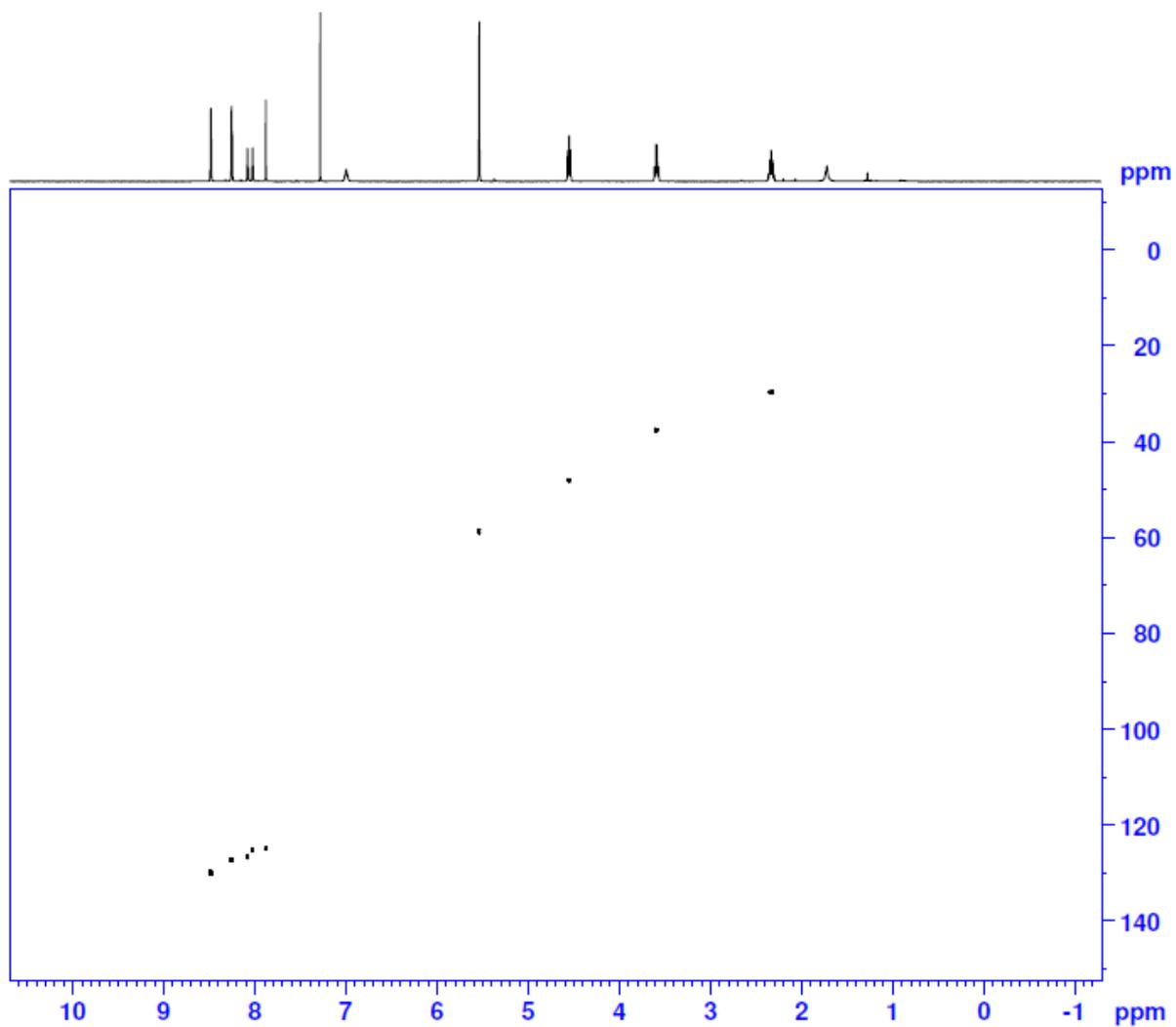
**Axle 9**

$^1\text{H}$ - $^1\text{H}$  COSY NMR ( $\text{CDCl}_3$ )



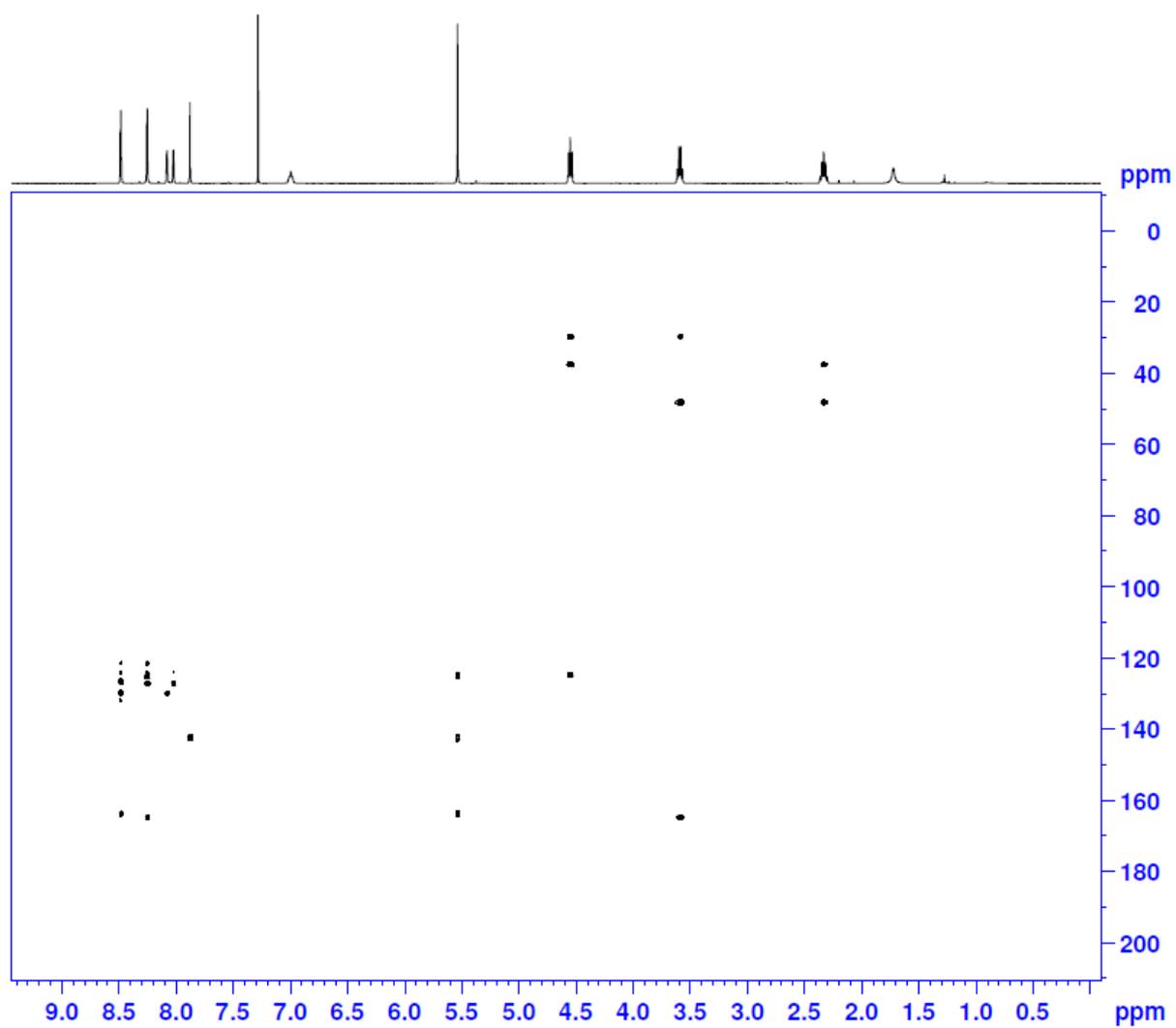
**Axle 9**

$^1\text{H}$ - $^{13}\text{C}$  HSQC NMR ( $\text{CDCl}_3$ )



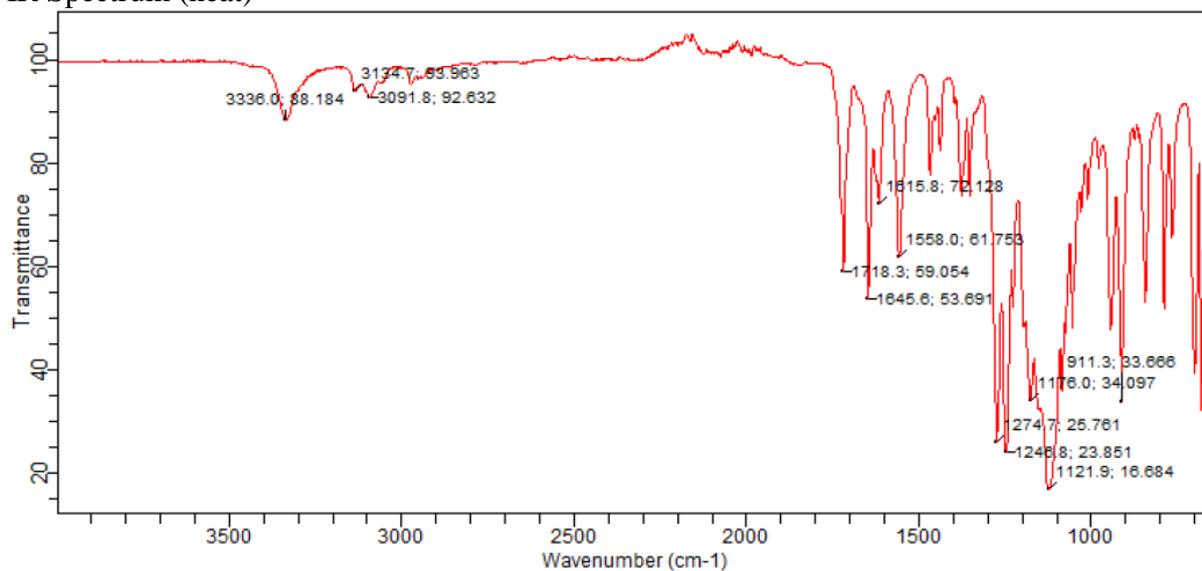
### Axle 9

$^1\text{H}$ - $^{13}\text{C}$  HMBC NMR ( $\text{CDCl}_3$ )



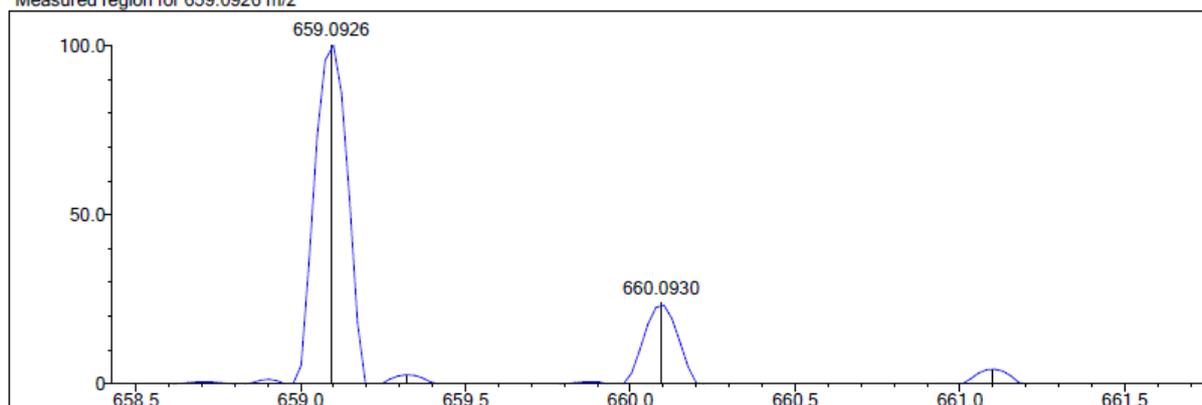
# Axle 9

## IR Spectrum (neat)

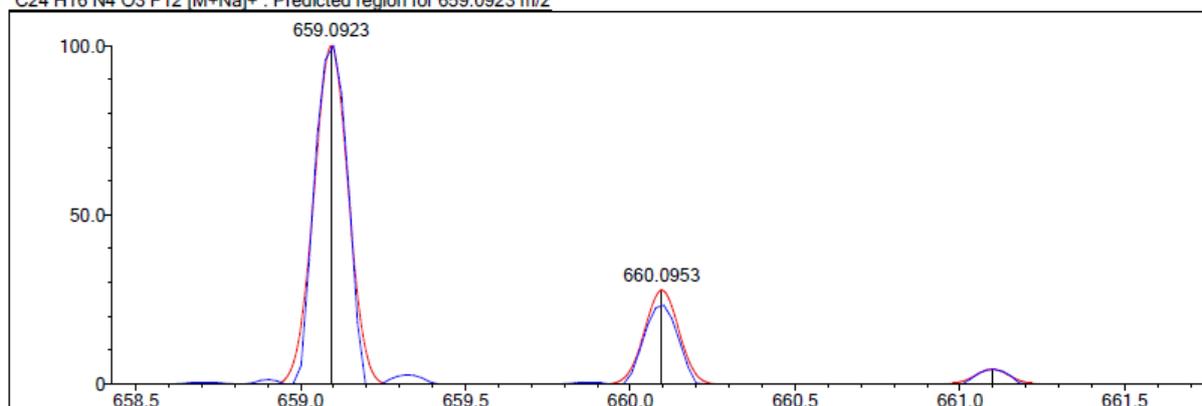


## Mass Spectrum (ES +ve)

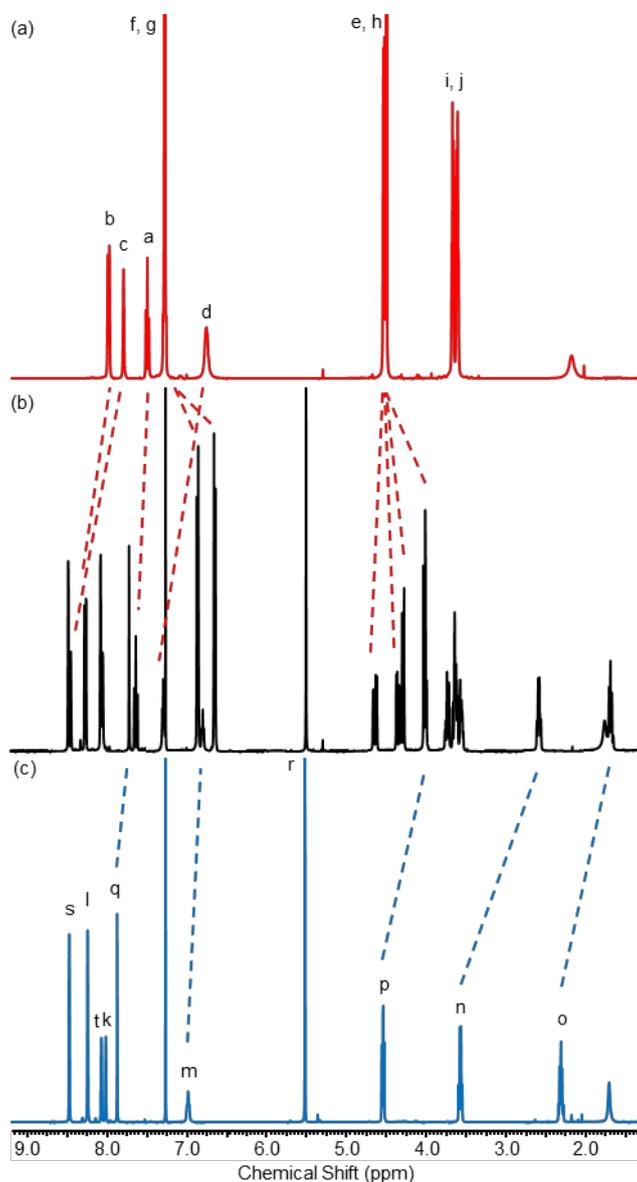
Measured region for 659.0926 m/z



C24 H16 N4 O3 F12 [M+Na]<sup>+</sup> : Predicted region for 659.0923 m/z



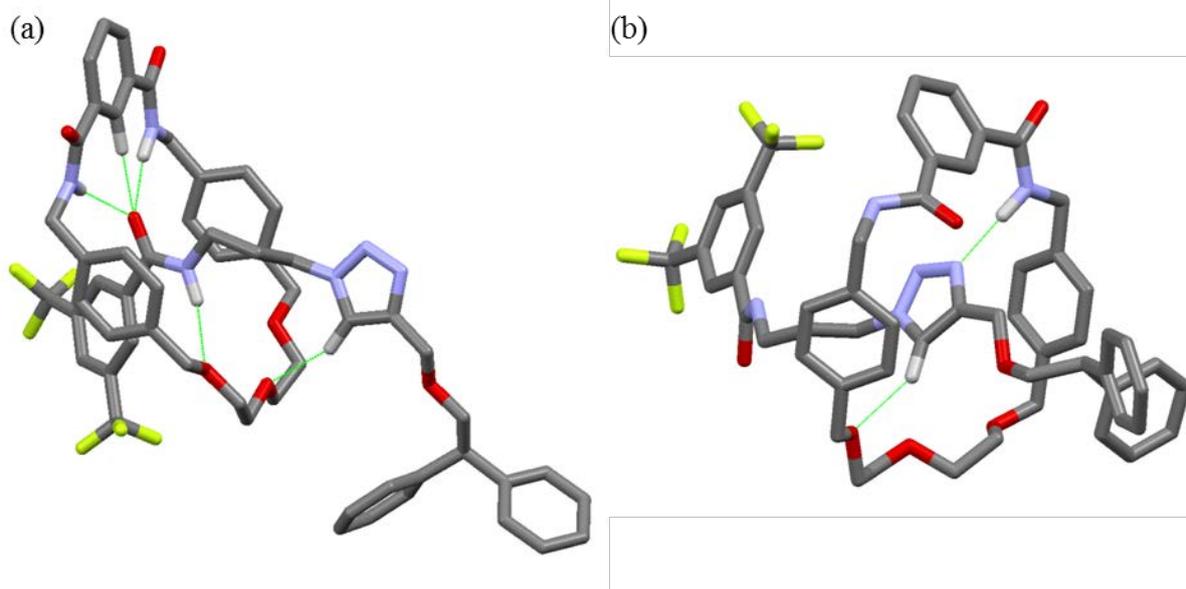
**$^1\text{H}$  NMR Spectral Comparison of (a) Macrocycle **1**, (b) Rotaxane **7** and (c) Axle **9** ( $\text{CDCl}_3$ , 400 MHz)**



While both the axle amide proton *m* and triazole proton *q* are further upfield in rotaxane **7** compared to free axle **9**, it is noted that alkyl proton *n* (adjacent to the amide) is very much further upfield in the rotaxane compared to the free axle, while alkyl proton *r* (adjacent to the triazole) has the same chemical shift in the rotaxane and free axle. These observations support the macrocycle residing over the amide rather than the triazole in rotaxane **7**.

## Part III: Computational Modelling

### Minimum Energy Structures of Rotaxane 6



### Energies of Minimised Structures of Rotaxanes 6 and 7

“Station” occupied	Rotaxane 6	Rotaxane 7
amide	-9609241 kJ mol <sup>-1</sup>	-10863599 kJ mol <sup>-1</sup>
triazole	-9609200 kJ mol <sup>-1</sup>	-10863523 kJ mol <sup>-1</sup>
difference	41 kJ mol <sup>-1</sup>	76 kJ mol <sup>-1</sup>

*NB: The four relevant structure files are supplied as part of the ESI.*

### Hydrogen Bond Distances in Minimised Structures of Rotaxanes 6 and 7

*Macrocycle residing over amide of axle component*

	Rotaxane 6	Rotaxane 7
isophthalamide N-H...O of axle amide	2.134 Å, 2.500 Å	2.195 Å, 2.385 Å
isophthalamide C-H...O of axle amide	2.205 Å	2.195 Å
axle N-H...O polyether	2.629 Å	2.762 Å
triazole C-H...O polyether	2.355 Å	2.316 Å

*Macrocycle residing over triazole of axle component*

	Rotaxane 6	Rotaxane 7
isophthalamide N-H...N of triazole	1.973 Å	1.993 Å
triazole C-H...O polyether	2.376 Å	2.286 Å

#### Part IV: References and Notes

- 1) C. N. Marrs and N. H. Evans, *Org. Biomol. Chem.*, 2015, **13**, 11021-11025.
- 2) L. M. Hancock and P. D. Beer, *Chem. Eur. J.*, 2009, **15**, 42-44.
- 3) A. Vidonne and D. Philp, *Tetrahedron*, 2008, **64**, 8464-8475.
- 4) J. A. Wisner, P. D. Beer and M. G. B. Drew, *Angew. Chem. Int. Edit.*, 2001, **40**, 3606-3609.
- 5) L. M. Hancock, DPhil Thesis, University of Oxford, 2011.