Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry. This journal is © The Royal Society of Chemistry 2017

Supporting Information

Stereodivergent Mannich reaction of bis(trimethylsilyl)ketenes with *N*-tert-butanesulfinyl imines by Lewis acid or Lewis base activation, a one pot protocol to obtain chiral β-amino acids.

Margarita Cantú-Reyes, Isabel Alvarado-Beltrán, Ricardo Ballinas-Indilí, Cecilio Álvarez-Toledano, Marcos Hernández-Rodríguez*

Instituto de Química. Universidad Nacional Autónoma de México

Circuito Exterior, Ciudad Universitaria, Del. Coyoacán, 04510, Cd. Mx Mexico

Table of contents

1.	Experimental procedure and characterization of β -aminoacids hydrochlorides 5 , 6	.S2
2.	References	S5
3.	Spectra of the β -aminoacids hydrochlorides 5 , 6	S5
4.	Spectra of the crude Mannich reaction of Tables 3, 4 and Figure 4	S19

1. Experimental procedure and characterization of β -amino acids hydrochlorides 5.

General remarks

The synthesis of sulfinyl imines¹ and bistrimethylsilylketenes² were done following reported procedures. The THF was distilled from sodium benzophenone ketyl and DCM from P₂O₅. ¹H and ¹³C NMR were recorded at 300 or 400 MHz and 75 or 100 MHz respectively. Chemical shifts (δ) were reported in ppm and coupling constants are in Hertz. Mass spectra were obtained by DART and TOF mass spectrometer.

General procedure for the one-pot Mannich reaction.

In a 4 mL vial it were added the corresponding sulfinyl imine (0.48 mmol, 1 equiv.) and 26 mg (0.048 mmol, 0.1 equiv.) of TBAT. It was purged with N₂ and the solids were dissolved in: Method A 1.6 mL of DCM and to the solution added 0.2 mL (1.4 mmol, 3 equiv.) of BF₃·OEt₂. Method B 1.6 mL of anhydrous THF. The reaction mixture was cooled with a ice-water and the corresponding bistrimethylsilylketene acetal (0.96 mmol, 2 equiv.) was added. After the addition, the reaction was warmed in a water bath at 20 °C and kept for 24 h. with stirring.

An aliquot of the reaction was taken to obtain the d.r. of the reaction and poured into a 1:1 mixture AcOEt/HCl 0.2 M, the aqueous layer was washed with AcOEt twice and the combined organic layers dried with Na₂SO₄ (anh.) and concentrated. The residue was dissolved in CDCl₃ and analyzed by ¹H NMR.

It was added to the reaction mixture 0.36 mL of MeOH and 0.24 mL (0.096 mmol, 2 equiv) of HCl 4M in dioxane and stirred for 24 h at room temperature. The next day was observed a solid and added AcOEt to induce more precipitation. The suspension was transferred to eppendorf tubes, centrifuged for 2 min. The liquid was removed with a Pasteur pipette and the solid was suspended again in AcOEt. It was put inside an ultrasonic bath (2 min.), centrifuged (2 min.) and again removed again the liquid. This washing of the solid was repeated twice. The solid that corresponds to the aminoacid hydrochloride was dried under high vacuum and weighed to obtain the yield showed in Tables 3, 4 and Figure 4.

The ESI contains the physical and spectroscopic data of the compounds obtained by Method B (only TBAT). The enantiomer obtained by method A have the same data only with opposite sign of optical rotation and lower magnitude because a lower diastereoselectivity in the Mannich reaction

(S)-3-Amino-2, 2-dimethyl-3-phenylpropanoic acid hydrochloride (5a).

White solid, $[\alpha]_D^{25}$ -29.3 (*c* 1, NaOH 1M). Lit.³ $[\alpha]_D^{25}$ -31.6 (*c* 1.1, HCl 1M). ¹H NMR (300 MHz, D₂O): δ = 7.54-7.35 (m, 5H), 4.30 (s, 1H), 1.29 (s, 3H), 1.01 (s, 3H). ¹³C NMR (75 MHz, D₂O): δ = 183.5, 135.3, 129.2, 129.0, 128.0, 62.0, 45.4, 24.9, 22.1. HRMS (DART/TOF): [M+H]⁺ Calcd. for C₁₁H₁₆NO₂ 194.1181; found 194.1183.

(*S*)-3-Amino-2,2-dimethyl-3-(4-nitrophenyl)propanoic acid hydrochloride (**5b**).

Yellow solid, $[\alpha]_D^{25}$ -27 (*c* 0.2, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 8.30 (s, 2H), 7.64 (s, 2H), 4.72 (s, 1H), 1.33 (s, 3H), 1.20 (s, 3H). ¹³C NMR (75 MHz, D₂O): δ = 178.9, 148.2, 140.7, 129.4, 124.2, 60.4, 45.2, 23.3, 20.9. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₁H₁₅N₂O₄ 239.1031; found 239.1028.

(*S*)-3-Amino-3-(4-(methoxycarbonyl)phenyl)-2,2-dimethylpropanoic acid hydrochloride (**5c**).

White solid, $[\alpha]_D^{25}$ -19.2 (*c* 0.26 in NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 8.04 (s, 2H), 7.51 (s, 2H), 4.64 (s, 1H), 3.91 (s, 3H), 1.32 (s, 3H), 1.19 (s, 3H). ¹³C NMR (75 MHz, D₂O): δ = 178.8, 168.5, 138.7, 130.4, 129.8, 128.2, 60.6, 52.7, 45, 23.1, 20.7. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₃H₁₈NO₄ 252.1235; found 252.1244.

(S)-3-Amino-2,2-dimethyl-3-(4-(trifluoromethyl)phenyl)propanoic acid hydrochloride (5d).

White solid, $[\alpha]_D^{25}$ -14.8 (*c* 0.31, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.79 (d, *J* = 7.5 Hz, 2H), 7.56 (d, *J* = 7.5 Hz, 2H), 4.65 (s, 1H), 1.32 (s, 3H), 1.19 (s, 3H). ¹³C NMR (100 MHz, D₂O): δ = 178.8, 137.5, 130.8 (q, *J* = 32.5 Hz), 128.5, 125.9 (q, *J* = 3.4 Hz), 123.8 (q, *J* = 271.9 Hz), 60.6, 45, 23.2, 20.7. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₂H₁₅F₃NO₂ 262.1054; found 262.1048.

(*S*)-3-Amino-3-(3,5-bis(trifluoromethyl)phenyl)-2,2-dimethylpropanoic acid hydrochloride (**5e**).

White solid, $[\alpha]_D^{25}$ -45.2 (*c* 0.25, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 8.24 (s, 1H), 8.05 (s, 2H), 1.38 (s, 3H), 1.24 (s, 3H). ¹³C NMR (100 MHz, D₂O): δ = 178.7, 135.8, 131.6 (q, *J* = 33.6 Hz), 128.70 – 128.46 (m), 123.9 (m), 123 (q, 202.7 Hz), 60.25, 45.12, 23.07, 20.52. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₃H₁₄F₆NO₂ 330.0928; found 330.0917.

(S)-3-Amino-2,2-dimethyl-3-(p-tolyl)propanoic acid hydrochloride (5f).

White solid, $[\alpha]_D^{25}$ -3 (*c* 0.36, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.28 (s, 4H), 4.49 (s, 1H), 2.34 (s, 3H), 1.31 (s, 3H), 1.17 (s, 3H). ¹³C NMR (75 MHz, D₂O): δ = 179.6, 140, 130.7, 129.6, 128.0, 61.1, 45.3, 23.4, 21.0, 20.3. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₂H₁₈NO₂ 208.1337; found 208.1337.

(S)-3-Amino-3-(4-methoxyphenyl)-2,2-dimethylpropanoic acid hydrochloride (5g).

White solid, $[\alpha]_D^{25}$ +16.0 (*c* 0.4, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.22 (d, *J* = 8.6 Hz, 2H), 6.91 (d, *J* = 8.6 Hz, 2H), 4.39 (s, 1H), 3.71 (s, 3H), 1.19 (s, 3H), 1.06 (s, 3H). ¹³C NMR (75 MHz, Deuterium Oxide) δ 179, 159.5, 129.4, 126, 114.3, 60.6, 55.4, 45.2, 23.2, 20.8. HRMS (DART/TOF): M+H]⁺ Calcd for C₁₂H₁₈NO₃ 224.1286; found 224.1292.

(*R*)-3-Amino-2,2-dimethyl-3-(thiophen-2-yl)propanoic acid hydrochloride (5h).

White solid, $[\alpha]_D^{25}$ +29.5 (*c* 0.2, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.52 (d, *J* = 5.1 Hz, 1H), 7.23 (d, *J* = 3.6 Hz, 1H), 7.11 (t, *J* = 4.3 Hz, 1H), 4.87 (s, 1H), 1.37 (s, 3H), 1.23 (s, 3H). ¹³C NMR (100 MHz, D₂O): δ = 179, 134.7, 129.4, 127.7, 127.3, 57.1, 45.2, 23.1, 21. HRMS (DART/TOF): [M+H]⁺ Calcd for C₉H₁₄NO₂S 200.0745; found 200.0743.

(S)-3-Amino-2,2-dimethyl-3-(naphthalen-1-yl)propanoic acid hydrochloride (5i).

White solid, ¹H NMR (300 MHz, D₂O): δ = 8.34 – 8.16 (m, 1H), 8.03 (s, 1H), 7.81 – 7.40 (m, 2H), 5.55 (s, 1H), 1.36 (s, 1H), 1.16 (s, 7H). ¹³C NMR (75 MHz, D₂O): δ = 179.9, 133.6, 131.4, 130.6, 130.2, 129.3, 127.4, 126.7, 125.5, 124.7, 122.9, 54.7, 46.3, 24.1, 20.7. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₅H₁₈NO₂ 244.1337; found 244.1327.

(*S*)-1-(1-Amino-1-phenylmethyl)cyclobutane-1-carboxylic acid hydrochloride (**6a**).

White solid, $[\alpha]_D^{25}$ +10.9 (*c* 0.11, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.44 – 7.37 (m, 3H), 7.37 – 7.31 (m, 2H), 4.60 (s, 1H), 2.34 – 2.17 (m, 3H), 1.96 – 1.82 (m, 3H). ¹³C NMR (75 MHz, D₂O): δ = 182.4, 134.3, 129.2, 129.1, 127.3, 60.1, 50.9, 30.2, 27, 14.8. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₂H₁₆NO₂ 206.1181; found 206.1183.

(*S*)-1-(1-Amino-1-phenylmethyl)cyclopentane-1-carboxylic acid hydrochloride (**6b**).

White solid, $[\alpha]_D^{25}$ -10.9 (*c* 0.11, NaOH 1M). ¹H NMR (400 MHz, D₂O): δ 7.45 – 7.38 (m, 3H), 7.38 – 7.31 (m, 2H), 4.48 (s, 1H), 2.23 – 2.11 (m, 1H), 1.93 – 1.76 (m, 2H), 1.74 – 1.49 (m, 5H). ¹³C NMR (100 MHz, D₂O): δ

= 178.7, 134.2, 129.5, 129.1, 127.3, 60.7, 57.3, 35.8, 32.9, 24.0, 23.7. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₃H₁₈NO₂ 220.1337; found 220.1337.

(S)-1-(1-Amino-1-phenylmethyl)cyclohexane-1-carboxylic acid (6c).

White solid, $[\alpha]_D^{25}$ -12.3 (*c* 0.26, NaOH 1M). ¹H NMR (300 MHz, D₂O): δ = 7.39 (s, 3H), 7.28 (s, 2H), 4.24 (s, 1H), 1.61 – 1.04 (m, 10H). ¹³C NMR (75 MHz, D₂O): δ = 181.2, 134.6, 129.2, 128.8, 127.6, 60.9, 50.1, 33.8, 31.2, 25.2, 22.8, 22.2. HRMS (DART/TOF): [M+H]⁺ Calcd for C₁₄H₂₀NO₂ 234.1494; found 234.1490.

2. References

- 1. G. Liu, D. A. Cogan, T. D. Owens, T. P. Tang, J. A. Ellman, J. Org. Chem., 1999, 64, 1278-1284.
- 2. C. Ainsworth, Y. N. Kuo, J. Organomet. Chem., 1976, 46, 73-87.
- 3. V. H. Kunz, D. Schanzenbach, Angew. Chem. 1989, 101, 1042-1043.

3. Spectra of the β -amino acids hydrochlorides 5 and 6





















Figure S7. ¹H-NMR spectrum of 5d







Figure S9. ¹H-NMR spectrum of 5e



































Figure S21. ¹H-NMR spectrum of 6b



Figure S23. ¹H-NMR spectrum of 6c







Figure S26. ¹³C-NMR spectrum of **3a** recrystallized







Figure S28. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5a** by method A (promoted by BF_3OEt_2) in Table 3.



Figure S29. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5a** by method B (promoted by TBAT) in Table 4.



Figure S30. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5b** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S31. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5b** by method B (promoted by TBAT) in Table 4.



Figure S32. ¹H-NMR spectrum of the Mannich addition to obtain (R)-**5c** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S33. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5c** by method B (promoted by TBAT) in Table 4.



Figure S34. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5d** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S35. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5d** by method B (promoted by TBAT) in Table 4.



Figure S36. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5e** by method A (promoted by BF_3OEt_2) in Table 3.



Figure S37. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5e** by method B (promoted by TBAT) in Table 4.



Figure S38. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5f** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S39. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5f** by method B (promoted by TBAT) in Table 4.



Figure S40.¹H-NMR spectrum of the Mannich addition to obtain (R)-**5g** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S41. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5***g* by method B (promoted by TBAT) in Table 4.



Figure S42. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5h** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S43. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5h** by method B (promoted by TBAT) in Table 4.



Figure S44. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**5i** by method A (promoted by BF₃OEt₂) in Table 3.



Figure S45. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**5i** by method B (promoted by TBAT) in Table 4.



Figure S46. ¹H-NMR spectrum of the Mannich addition to obtain (R)-**6a** by method A (promoted by BF₃OEt₂) in Figure 4



Figure S47. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**6a** by method B (promoted by TBAT) in Figure 4.



Figure S48. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**6b** by method A (promoted by BF₃OEt₂) in Figure 4.



Figure S49. ¹H-NMR spectrum of the Mannich addition to obtain (*S*)-**6b** by method B (promoted by TBAT) in Figure 4.



Figure S50. ¹H-NMR spectrum of the Mannich addition to obtain (*R*)-**6c** by method A (promoted by BF₃OEt₂) in Figure 4.



Figure S51. ¹H-NMR spectrum of the Mannich addition to obtain (S)-**6c** by method B (promoted by TBAT) in Figure 4.