

The Diaza-Nazarov Cyclization Involving 2,3-Diaza-pentadienyl Cation for the Synthesis of Polysubstituted Pyrazoles

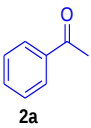
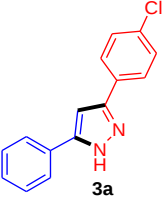
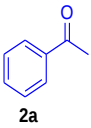
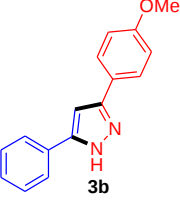
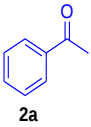
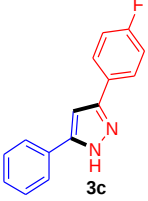
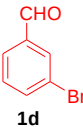
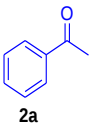
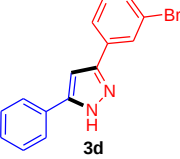
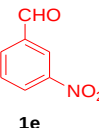
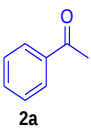
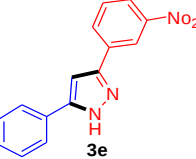
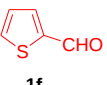
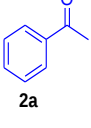
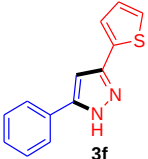
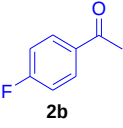
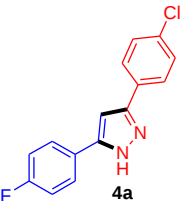
*Balakrishna Aegurla, Rama Krishna Peddinti**

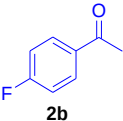
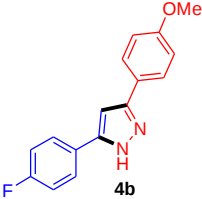
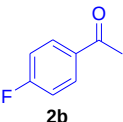
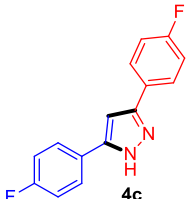
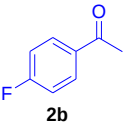
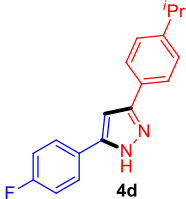
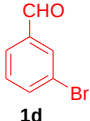
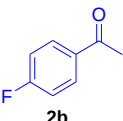
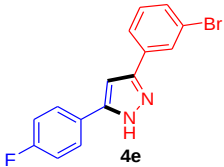
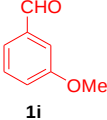
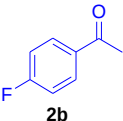
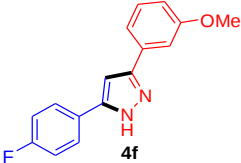
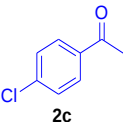
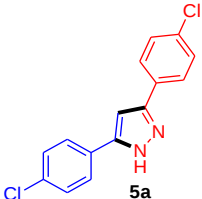
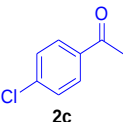
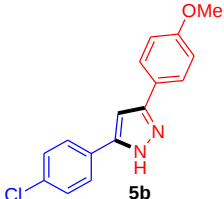
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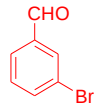
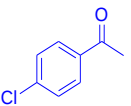
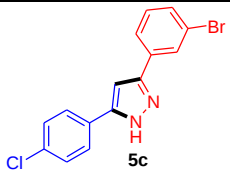
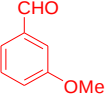
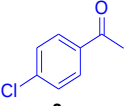
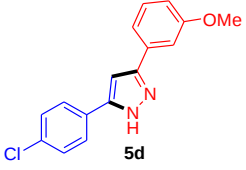
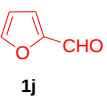
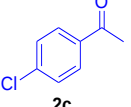
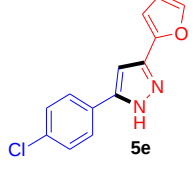
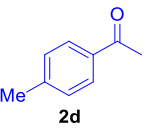
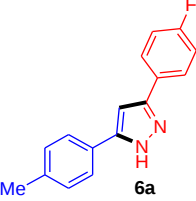
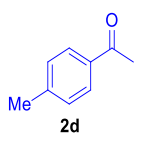
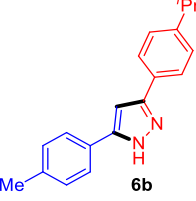
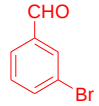
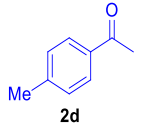
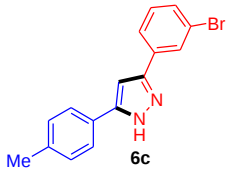
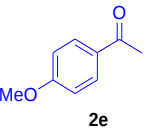
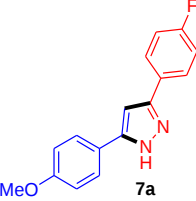
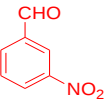
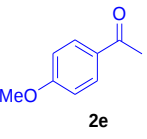
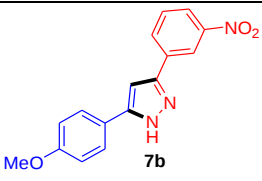
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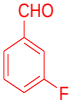
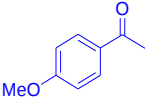
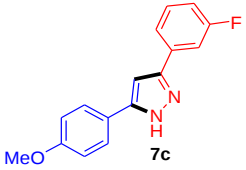
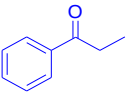
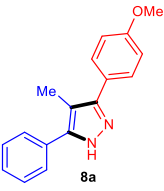
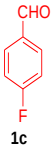
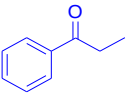
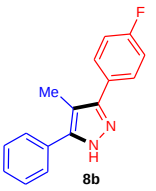
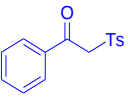
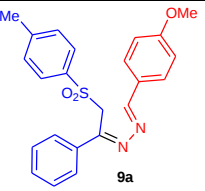
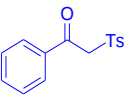
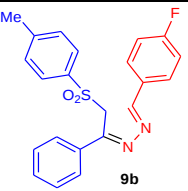
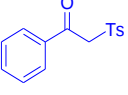
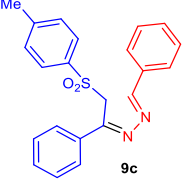
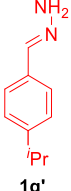
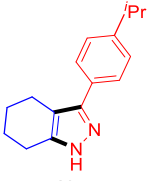
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Table S1. Iodine-mediated one-pot, three-component approach for the synthesis of substituted pyrazoles.

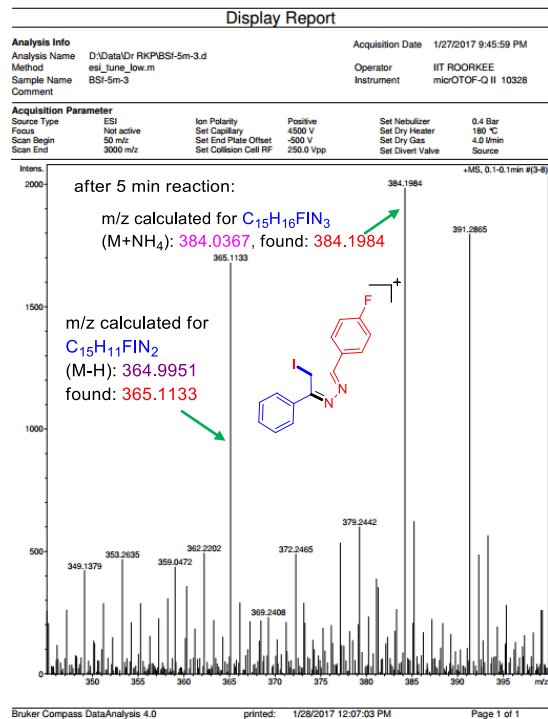
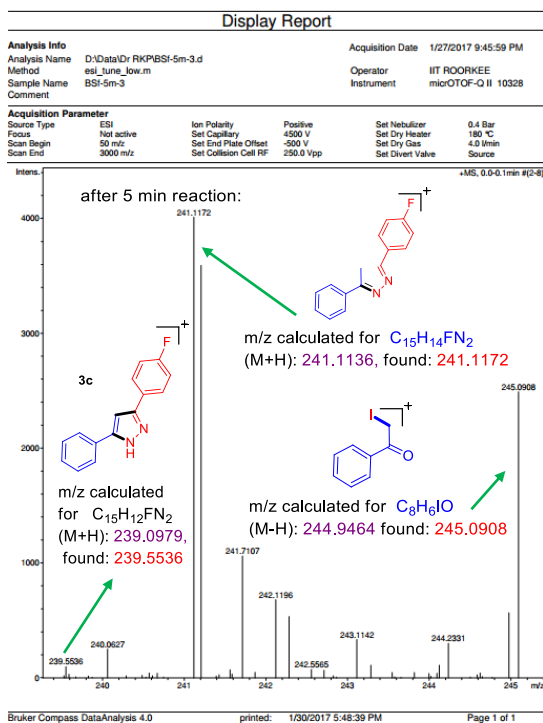
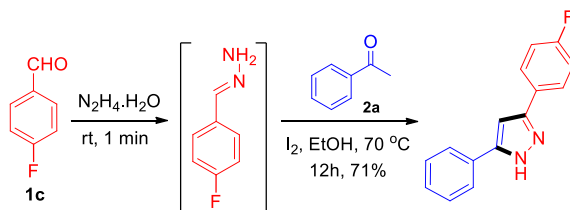
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2	 1b	 2a	 3b	70
3	 1c	 2a	 3c	71
4	 1d	 2a	 3d	67
5	 1e	 2a	 3e	82
6	 1f	 2a	 3f	58
7	 1a	 2b	 4a	70

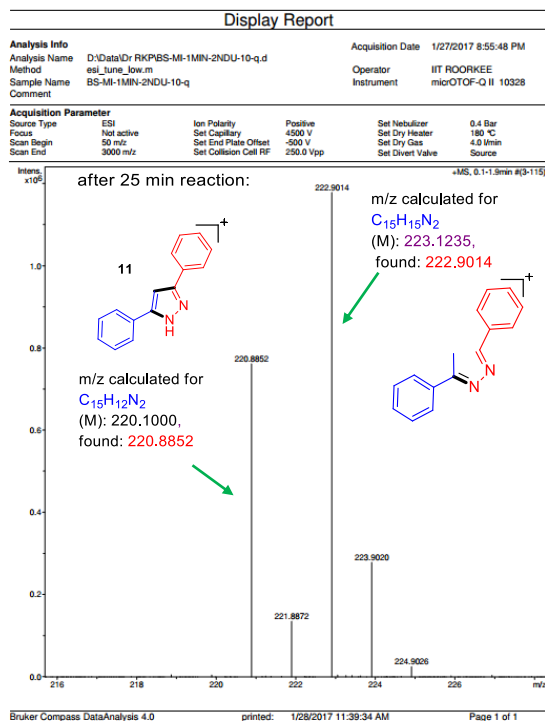
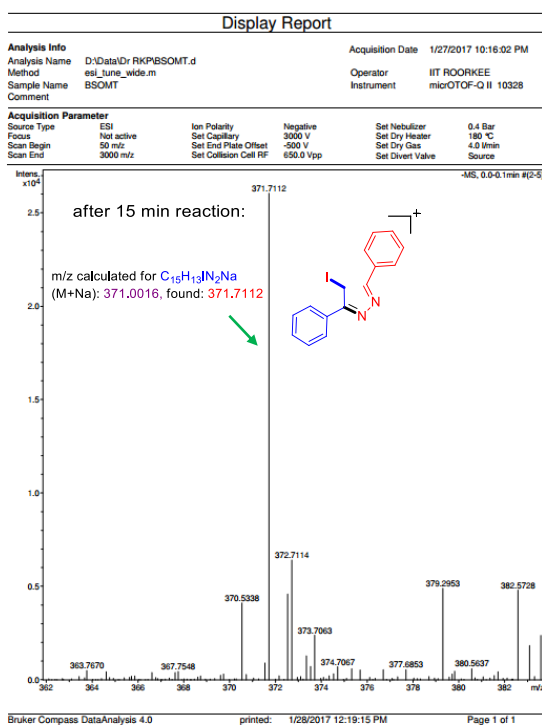
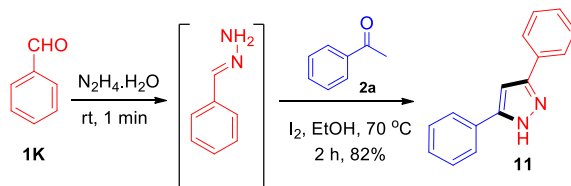
8	 1b	 2b	 4b	66
9	 1c	 2b	 4c	61
10	 1g	 2b	 4d	55
11	 1d	 2b	 4e	67
12	 1i	 2b	 4f	72
13	 1a	 2c	 5a	58
14	 1b	 2c	 5b	61

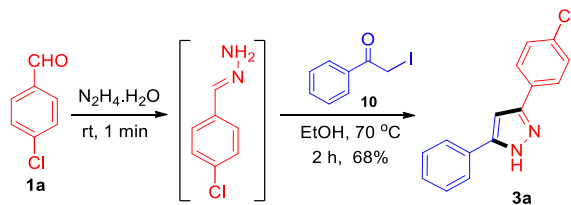
15	 1d	 2c	 5c	66
16	 1i	 2c	 5d	72
23	 1j	 2c	 5e	49
17	 1c	 2d	 6a	67
18	 1g	 2d	 6b	65
19	 1d	 2d	 6c	64
20	 1c	 2e	 7a	50
21	 1e	 2e	 7b	81

22	 1h	 2e	 7c	52
24	 1b	 2f	 8a	52
25	 1c	 2f	 8b	47
26	 1b	 2g	 9a	49
27	 1c	 2g	 9b	50
28	 CHO	 2g	 9c	90
29	 1g'	 2h	 12	24

Monitoring reactions with mass spectral analysis

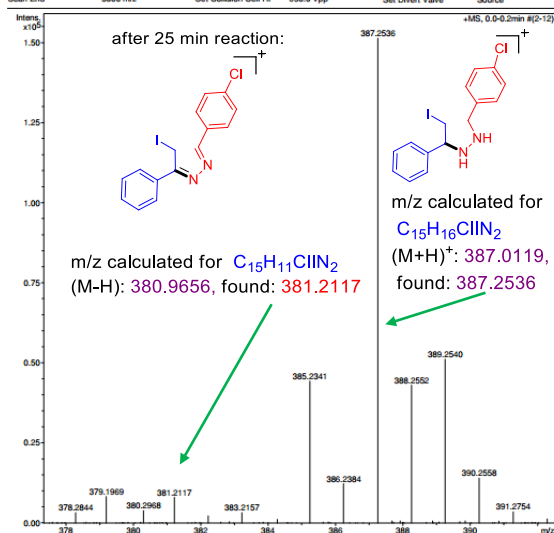






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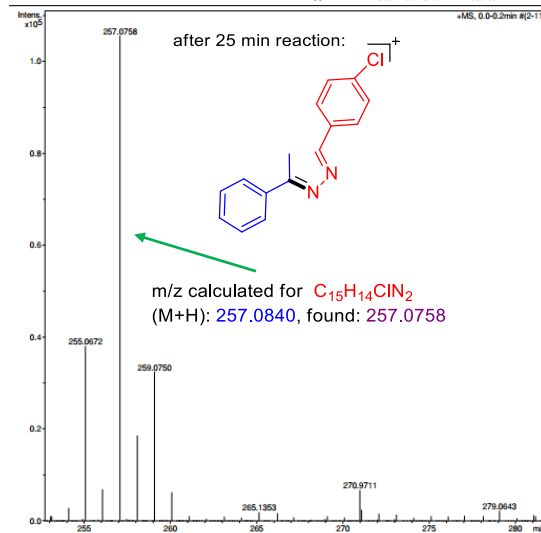
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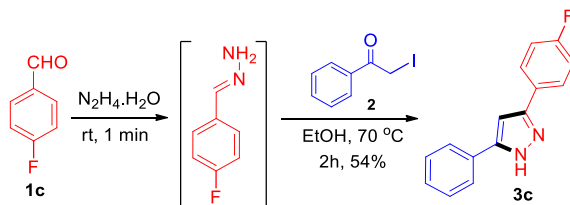
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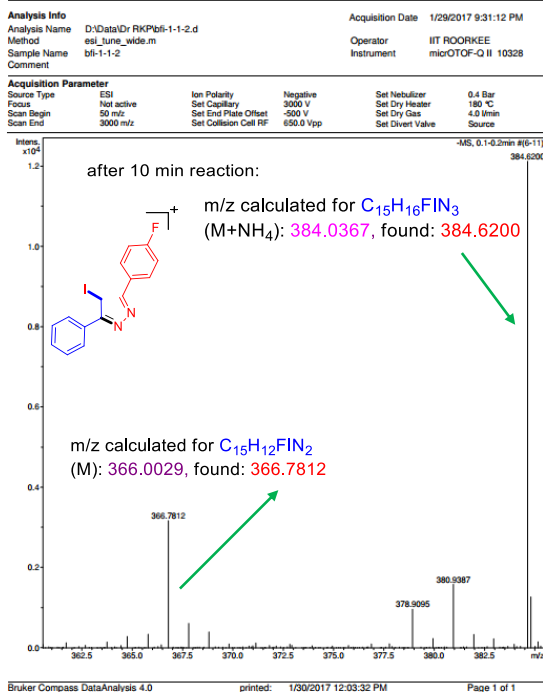
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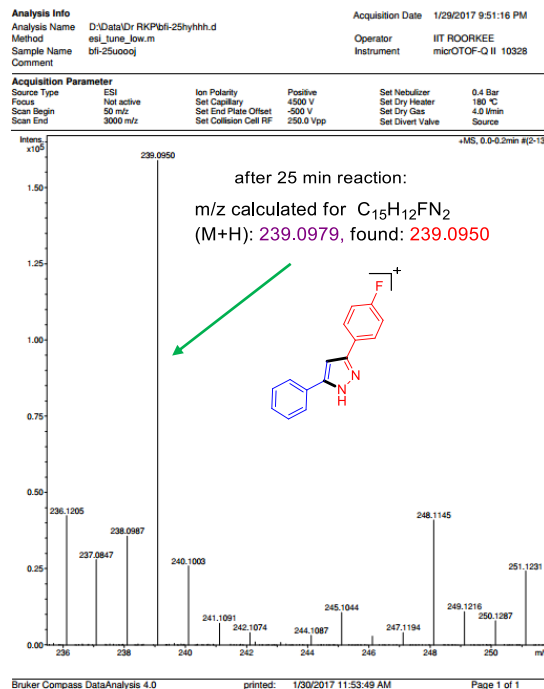
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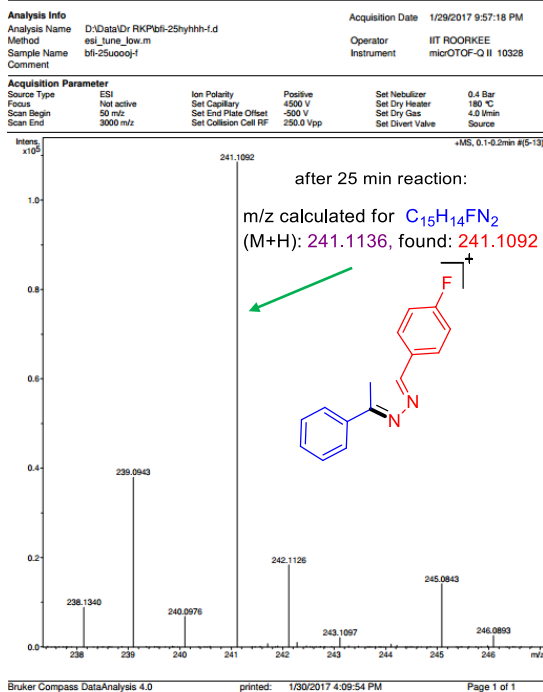
Display Report



Display Report



Display Report



Display Report

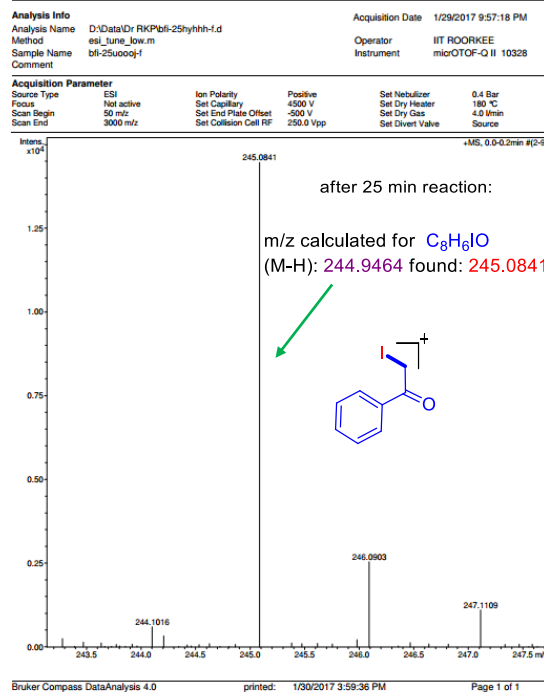
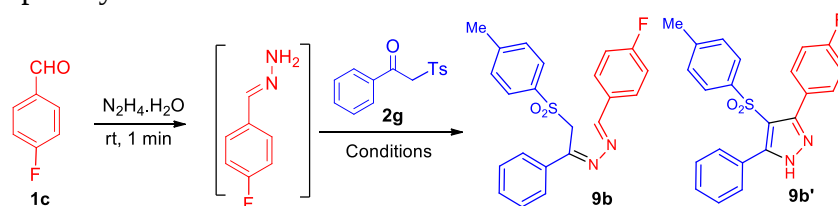


Table S2. Attempted cyclization of heterodienes.



Entry	Conditions	Product 9b Yield (%)	Product 9b' Yield (%)
1	I ₂ /TFA, EtOH, 80 °C, 12 h	52	Not observed
2	I ₂ /TFA, EtOH, 120 °C, 12 h	54	Not observed
3	SnCl ₄ , EtOH, 80 °C, 12 h	46	Not observed
4	SnCl ₄ , EtOH, 120 °C, 12 h	45	Not observed
5 ^a	FeCl ₃ , EtOH, 80 °C, 12 h	0	Not observed
6 ^a	FeCl ₃ , EtOH, 120 °C, 12 h	0	Not observed

Conditions: **1c** (0.5 mmol), Hydrazine hydrate (1 mmol), **2g** (0.6 mmol).

^a Starting materials **1a** and **2g** were recovered as such.

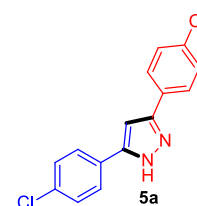
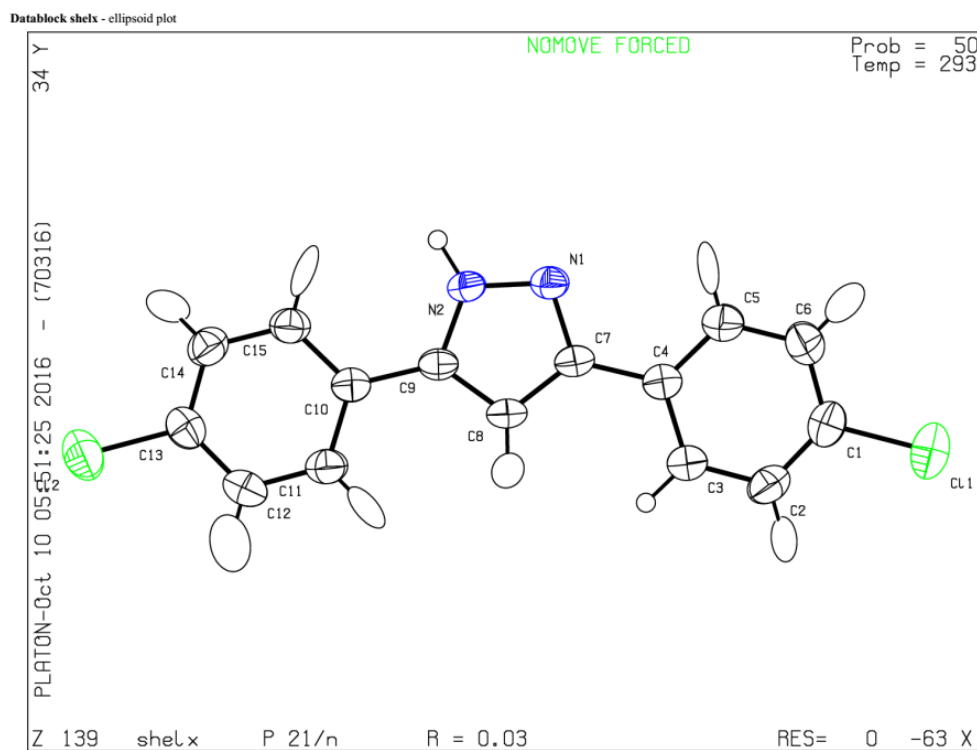
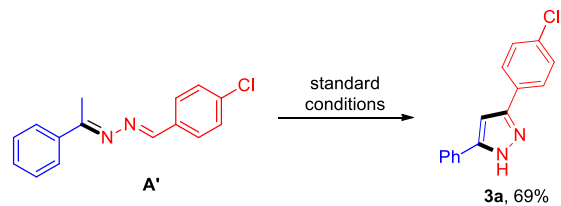
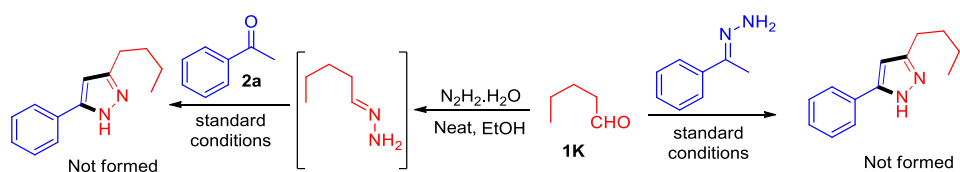


Figure S1. X-Ray crystal structure of the product **5a**.

Scheme S1. Substrate scope of DAN reaction.



Conditions: intermediate **A'** (0.5 mmol), iodine (0.6 mmol), EtOH (3 mL) at 70 °C for 12 h.



Conditions: **1k** (1.0 mmol), **2a** (1.0 mmol) iodine (0.6 mmol), EtOH (3 mL) at 70 °C for 12 h

Computational calculations

Table S3. Calculated electronic energies (in Hartree) for the B3LYP/6-311G** optimized structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction in gas phase.

Pathway	Energy	R	TS	P
1,5-	E	−688.76357468	−688.7450556	−688.7694985
	H	−688.508430	−688.491138	−688.513388
	G	−688.567734	−688.546576	−688.570101
1,6-	E	−688.7701048	−688.7371205	−688.737944
	H	−688.514615	−688.483234	−688.482915
	G	−688.573574	−688.537699	−688.538274
1,7-	E	−688.7632717	−688.7207741	−688.7466207
	H	−688.508123	−688.466286	−688.490507
	G	−688.567709	−688.520559	−688.544822

Table S4. Calculated relative electronic energies (in kcal/mol) for the B3LYP/6-311G** optimized structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction in gas phase.

Pathway	Energy	R	TS	P
1,5-	ΔE	0	11.62	−3.72
	ΔH	0	10.85	−3.11
	ΔG	0	13.28	−1.49
1,6-	ΔE	0	20.70	20.18
	ΔH	0	19.69	19.89
	ΔG	0	22.51	22.15
1,7-	ΔE	0	26.67	10.45
	ΔH	0	26.25	11.05
	ΔG	0	29.59	14.36

Table S5. Calculated electronic energies (in Hartree) for the structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction optimized at B3LYP/6-311G** level of theory using CPCM solvation model, with Ethanol solvent.

Pathway	Energy	R	TS	P
1,5-	E	-688.7635747	-688.7450556	-688.7694985
	H	-688.508430	-688.491138	-688.513388
	G	-688.567734	-688.546576	-688.570101
1,6-	E	-688.7701048	-688.7371204	-688.737944
	H	-688.514615	-688.483234	-688.482915
	G	-688.573574	-688.537700	-688.538274
1,7-	E	-688.7632717	-688.7207743	-688.7466207
	H	-688.508123	-688.466288	-688.490507
	G	-688.567709	-688.520560	-688.544822

Table S6. Calculated electronic energies (in kcal/mol) for the structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction optimized at B3LYP/6-311G** level of theory using CPCM solvation model, with Ethanol solvent.

Pathway	Energy	R	TS	P
1,5-	ΔE	0	11.62	-3.72
	ΔH	0	10.85	-3.11
	ΔG	0	13.28	-1.49
1,6-	ΔE	0	20.69	20.18
	ΔH	0	19.69	19.89
	ΔG	0	22.51	22.15
1,7-	ΔE	0	26.67	10.45
	ΔH	0	26.44	11.25
	ΔG	0	29.60	14.37

Table S7. Single point electronic energy, E (in Hartree) for the structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction calculated at M06/6-311+G(d,p)//B3LYP/6-311G** level of theory using CPCM solvation model, with Ethanol solvent

Pathway	Energy	R	TS	P
1,5-	E	-688.2902521	-688.2709314	-688.3065877
1,6-	E	-688.2960064	-688.2722362	-688.2768907
1,7-	E	-688.290431	-688.2525481	-688.2866835

Table S8. Single point electronic energy, E (in kcal/mol) for the structures of 1,5-, 1,6- and 1,7-cyclization pathways for electro-cyclization step of diaza-Nazarov reaction calculated at M06/6-311+G(d,p)//B3LYP/6-311G** level of theory using CPCM solvation model, with Ethanol solvent.

Pathway	Energy	R	TS	P
1,5-	ΔE	0	12.12	-10.25
1,6-	ΔE	0	14.91	11.99
1,7-	ΔE	0	23.77	2.35

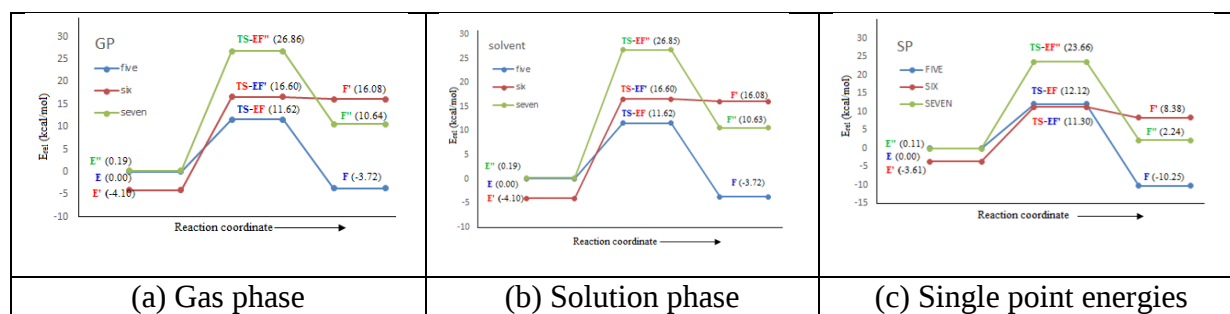


Figure S2. A comparative energy profile diagrams for the 1,5-, 1,6- and 1,7-cyclization pathways of electro-cyclization step of diaza-Nazarov reaction for (a) gas phase (optimized at B3LYP/6-311G** level), (b) Solution phase (optimized at B3LYP/6-311G** level using Ethanol solvent with CPCM model) and (c) Single point energies (at M06/6-311+G(d,p)//B3LYP/6-311G** level using Ethanol solvent with CPCM model).

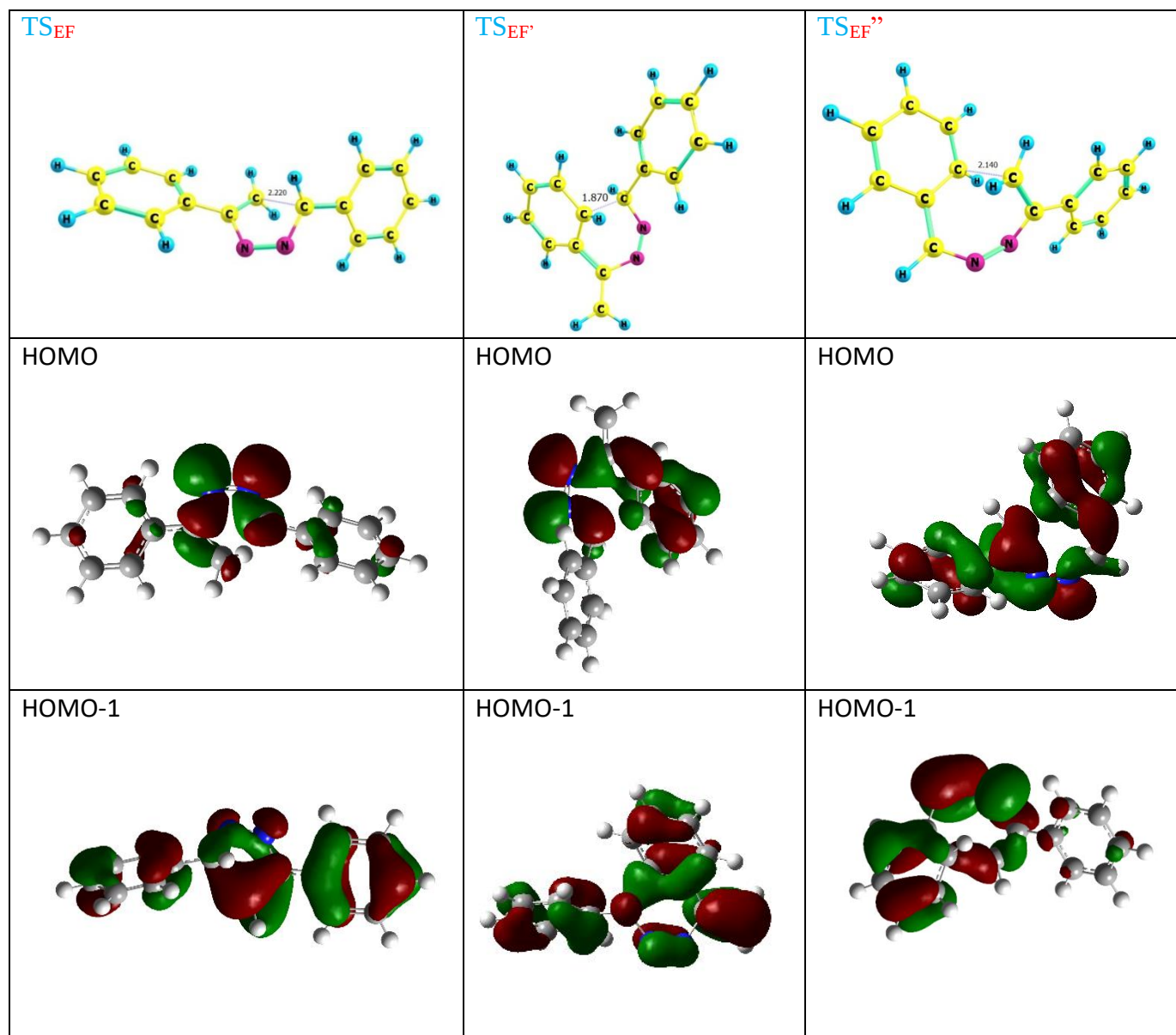
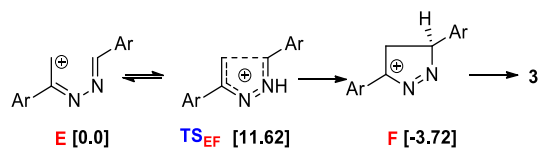


Figure S3. Optimised structures of TSs with HOMO and HOMO-1 molecular diagrams of each TS of the 1,5-, 1,6- and 1,7-cyclization pathways of electrocyclization step of diaza-Nazarov reaction at B3LYP/6-311G** level of theory in gas phase.

Table S9. Cartesian coordinates of B3LYP/6-311G** optimized structures of 1,5-, 1,6- and 1,7-electro-cyclization pathways of electro-cyclization step of Nazarov reaction in gas phase.



E

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TS_{EF}

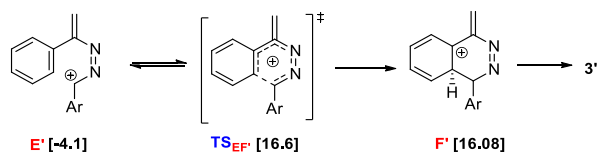
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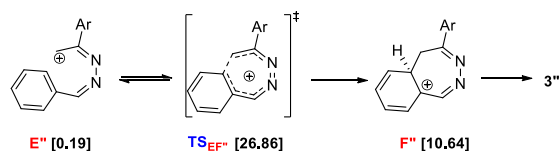
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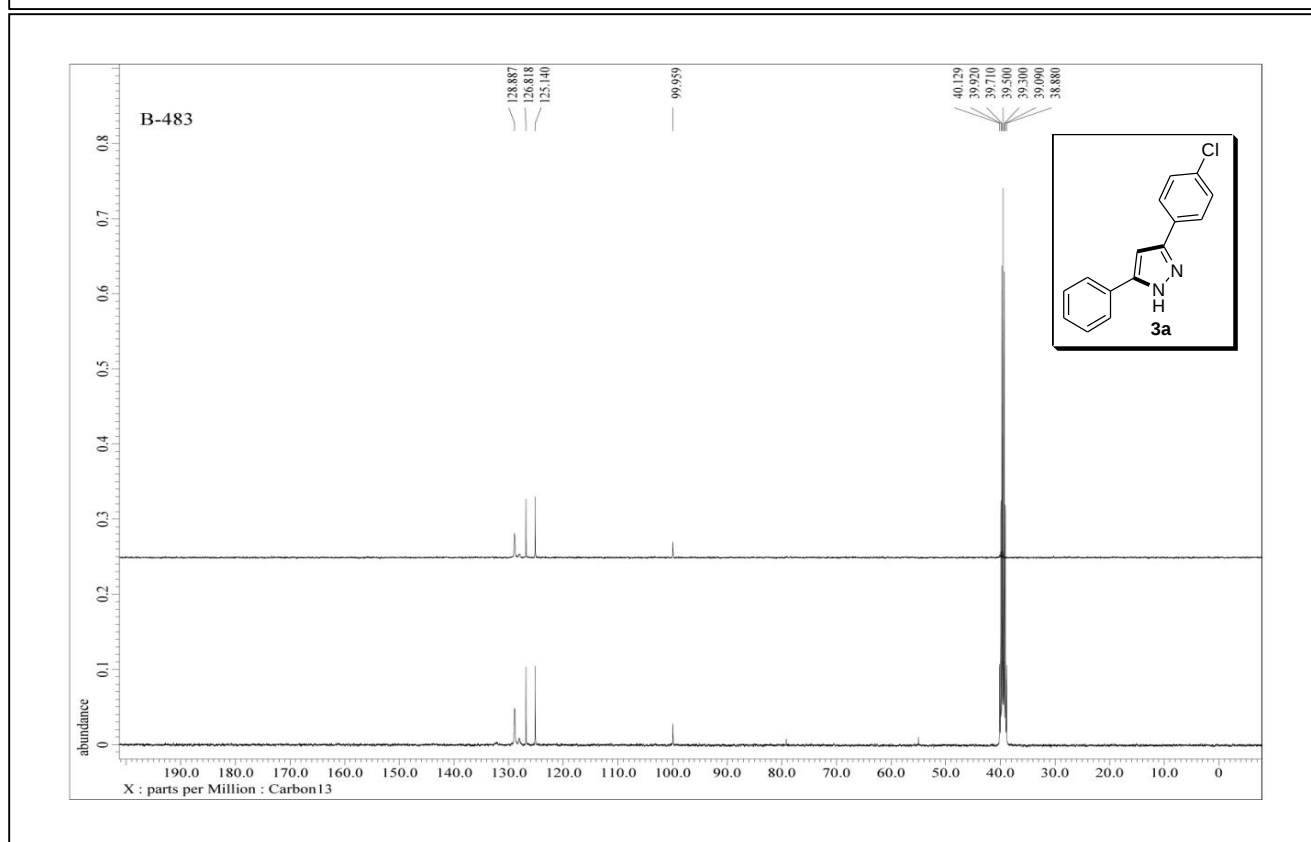
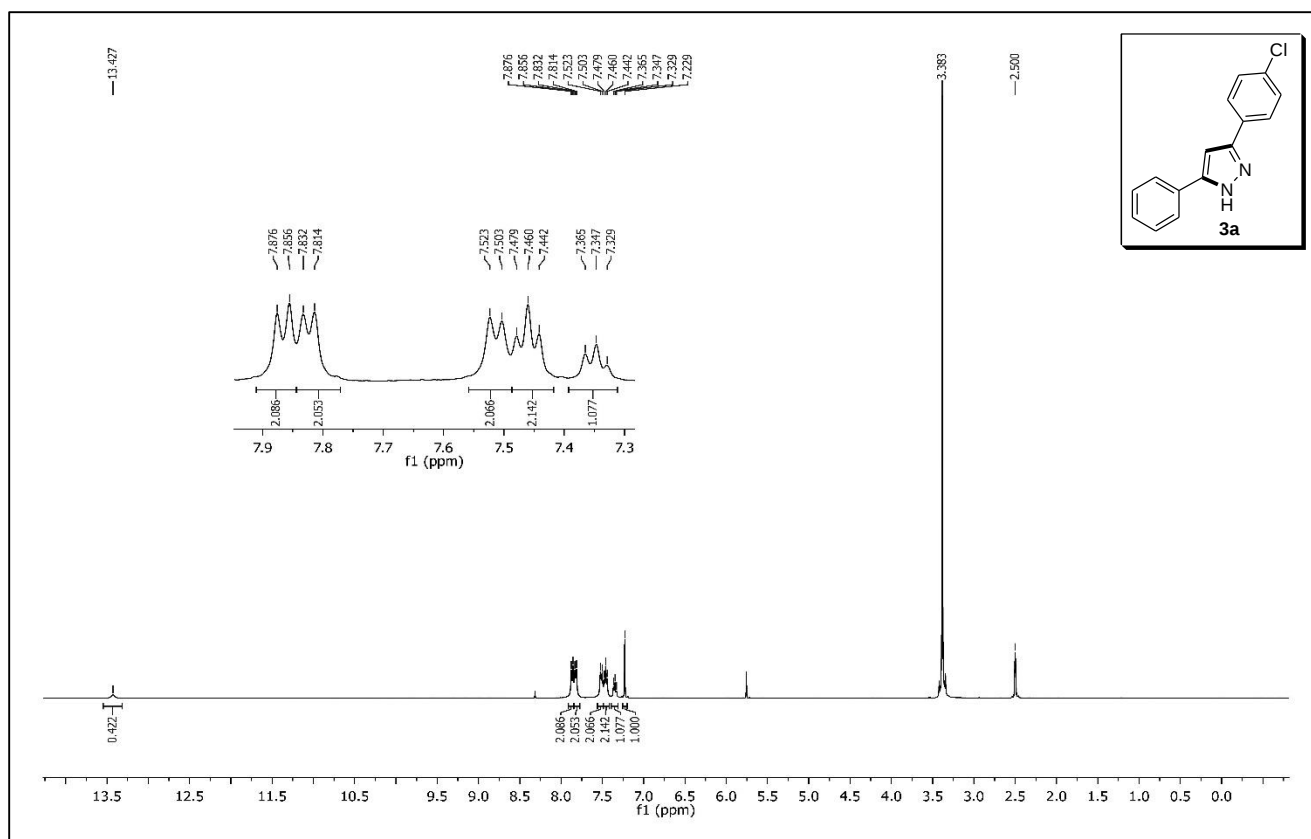
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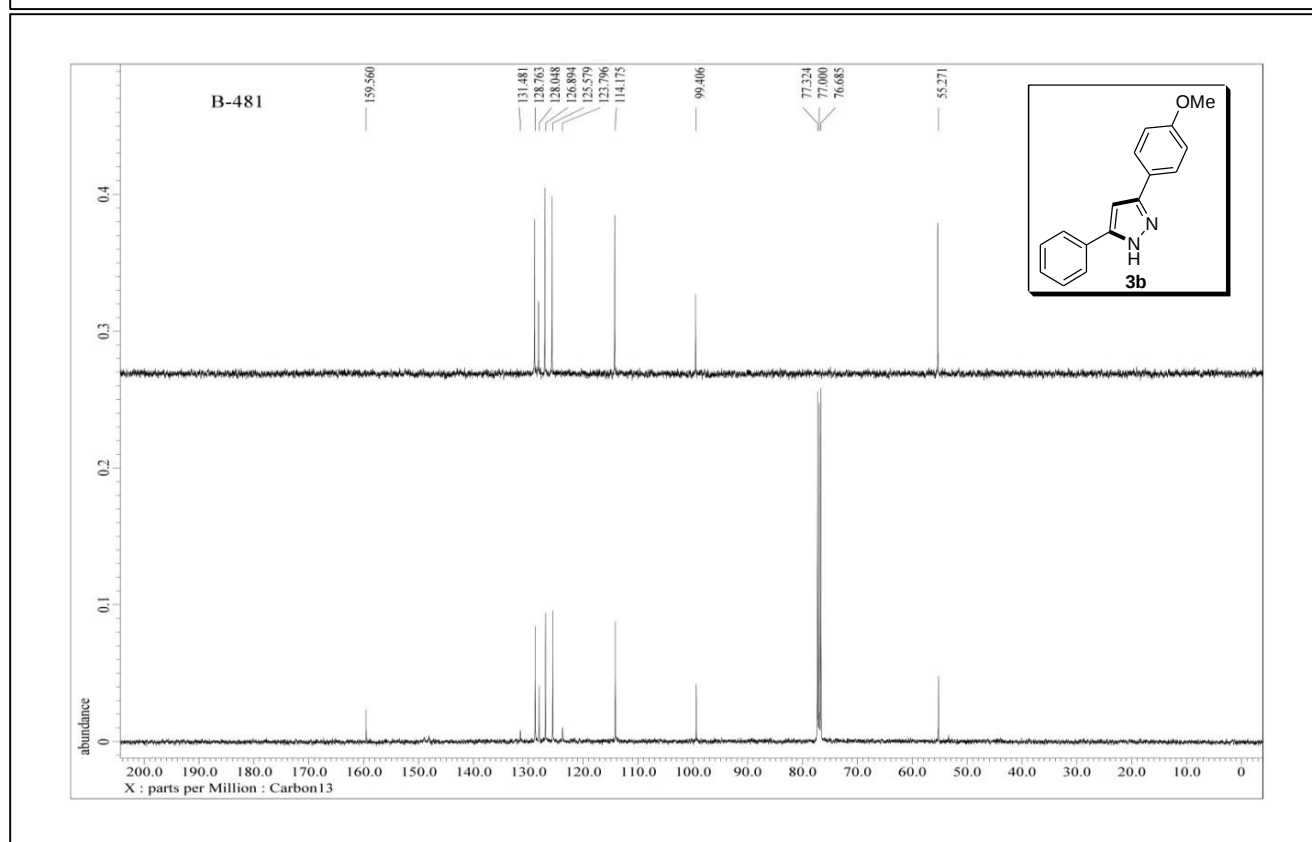
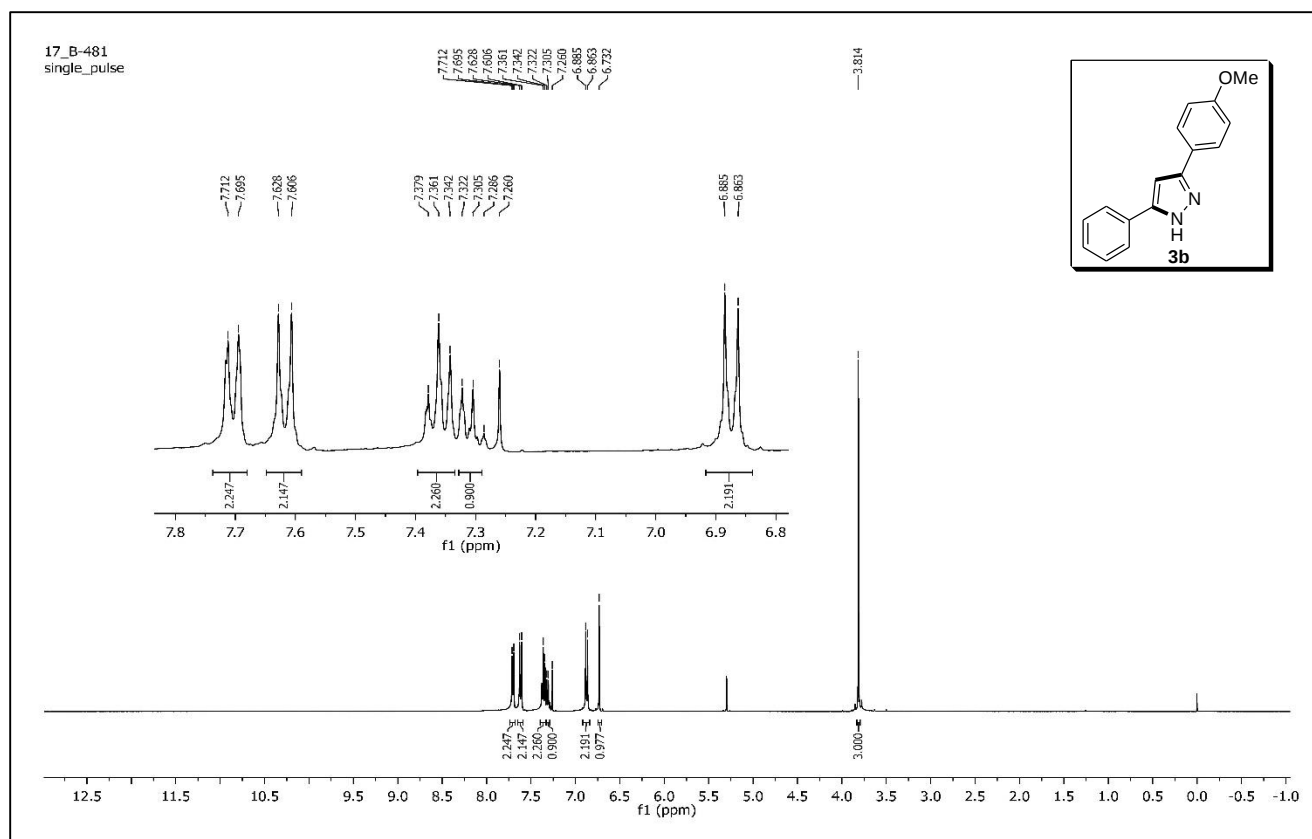
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6	2.522327000	0.675039000	-0.267363000
6	1.631832000	-0.396339000	-0.658975000
6	3.832714000	0.352268000	0.112939000
6	2.153781000	-1.724109000	-0.749463000
6	4.288083000	-0.957514000	0.050781000
6	3.442452000	-2.005377000	-0.366055000
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1	4.501081000	1.138635000	0.443755000
1	1.507138000	-2.508650000	-1.125123000
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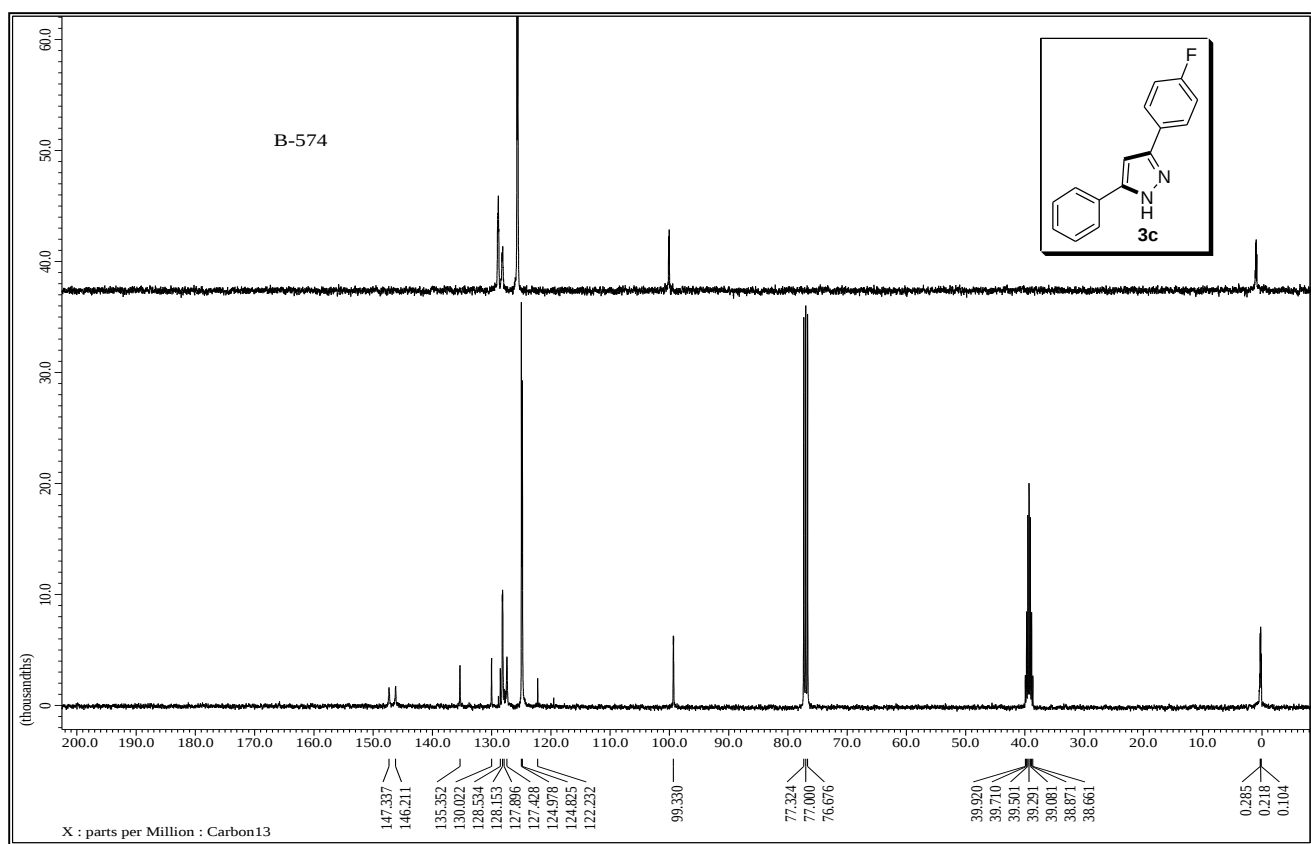
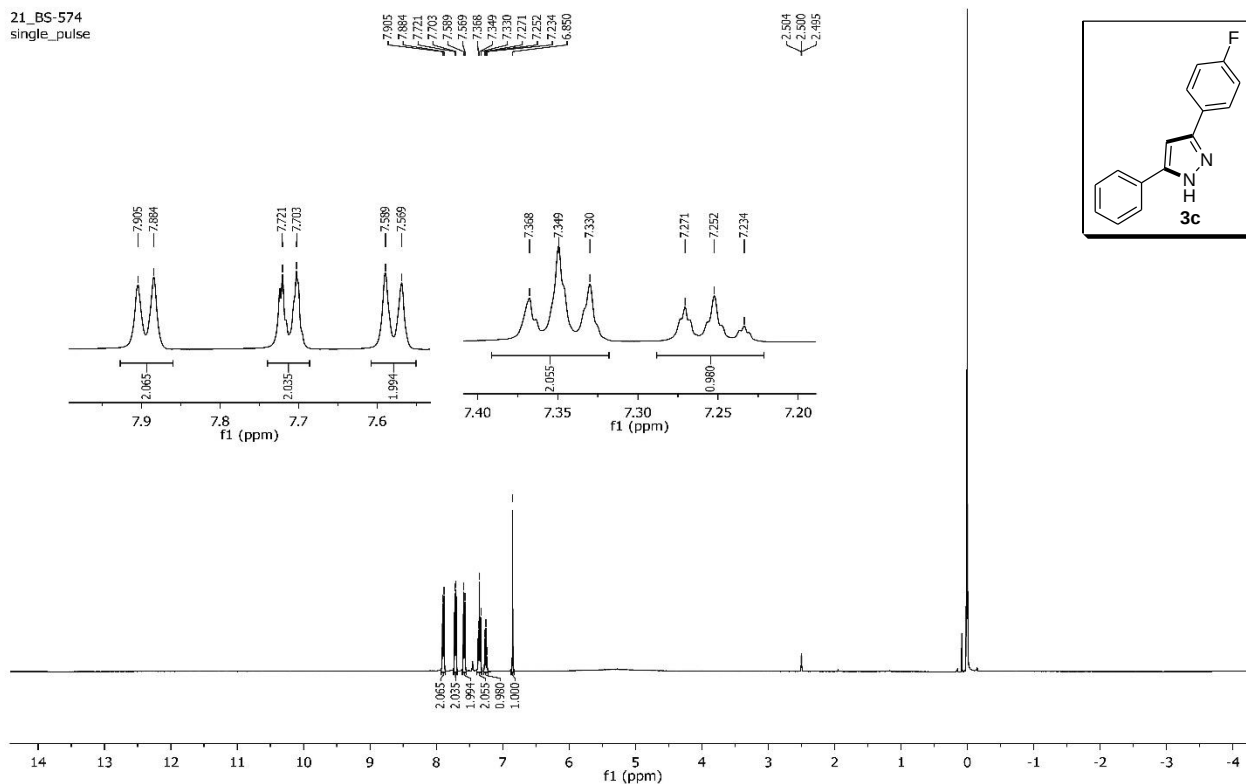
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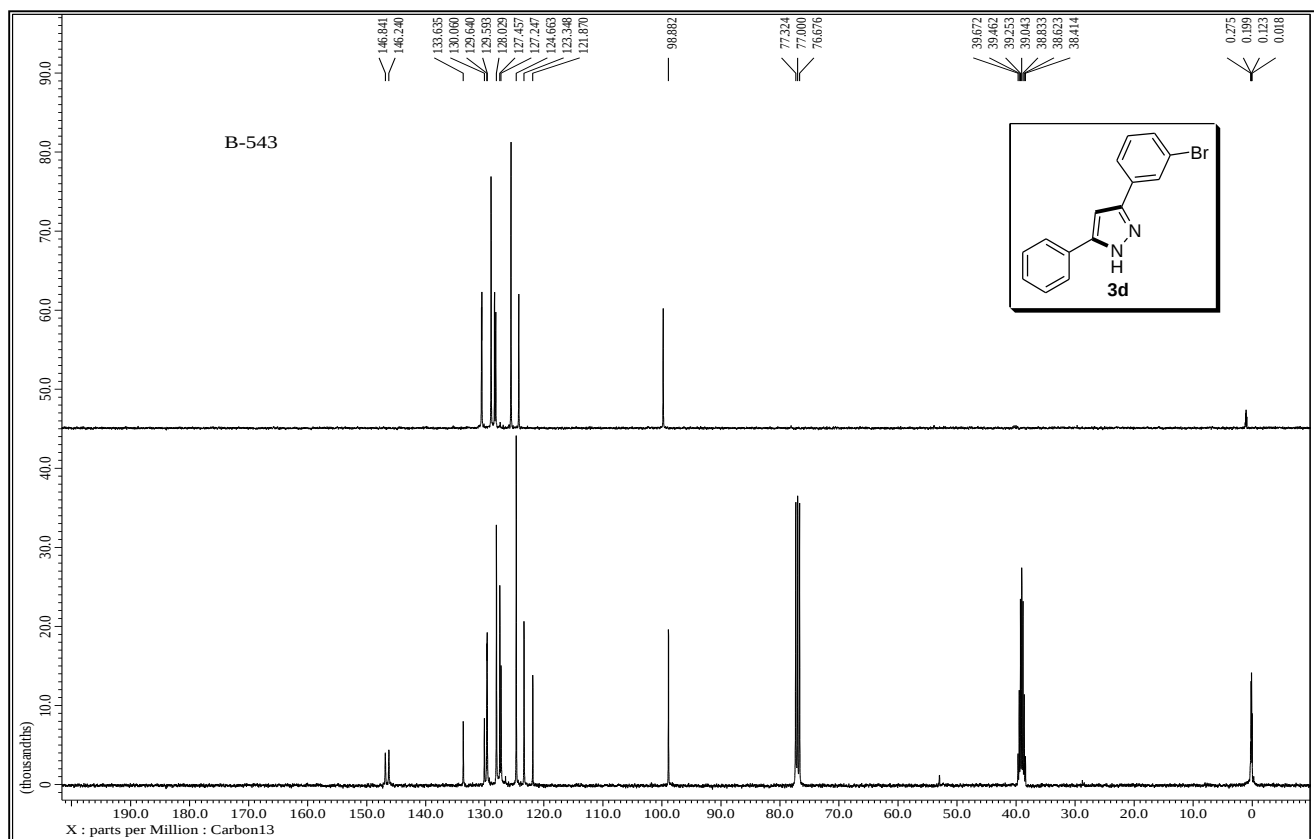
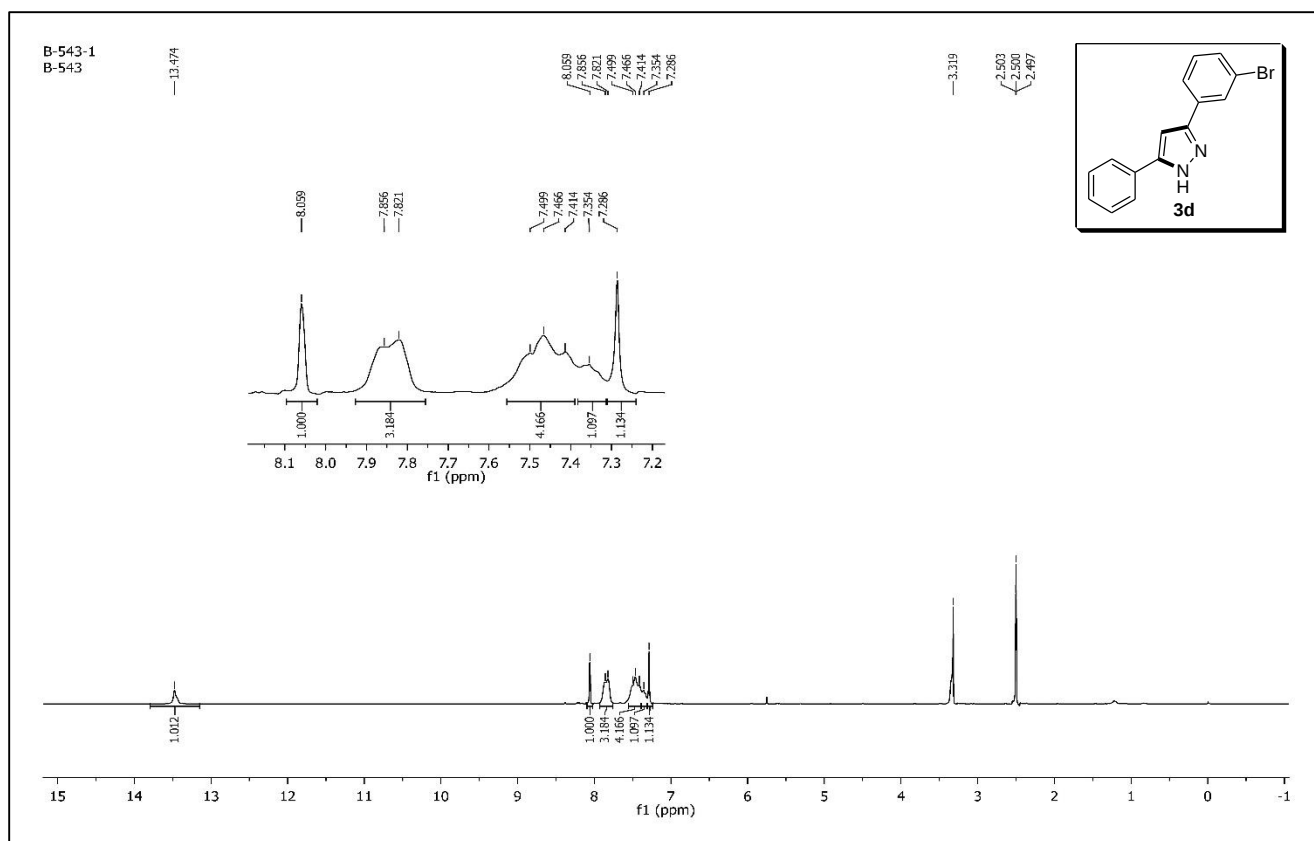
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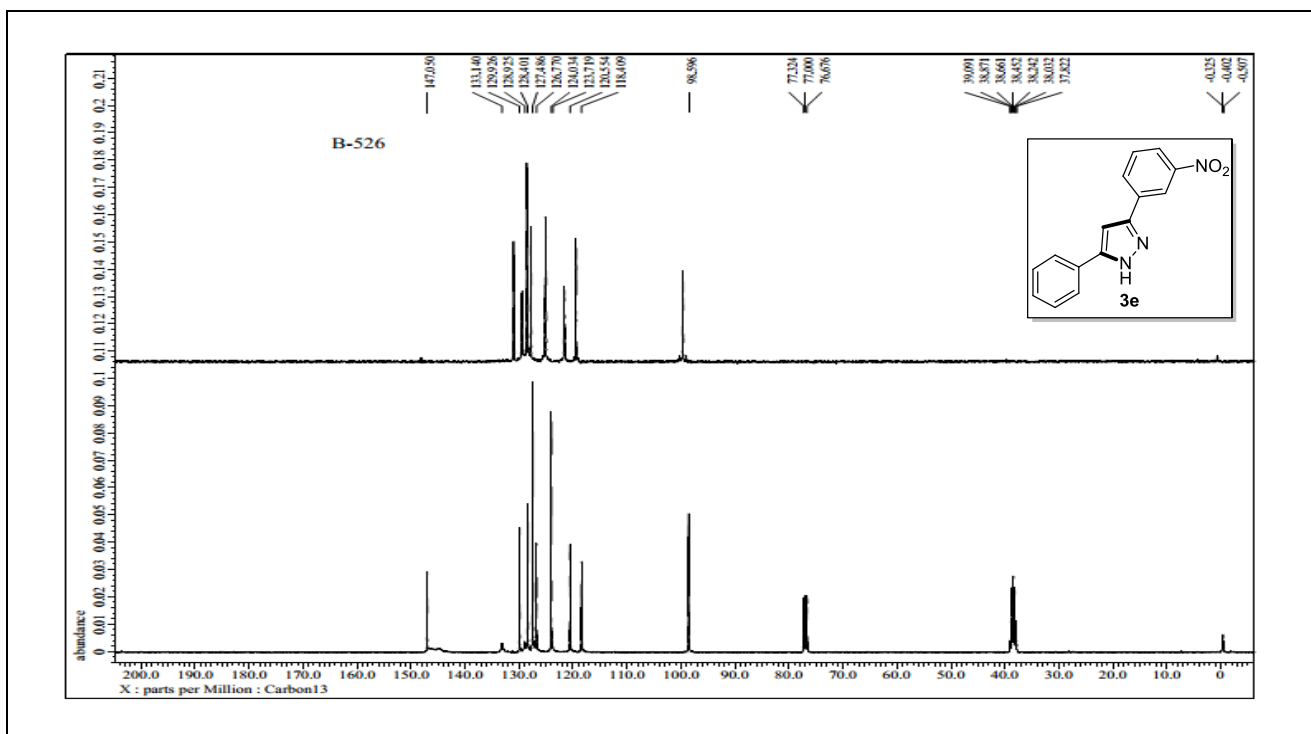
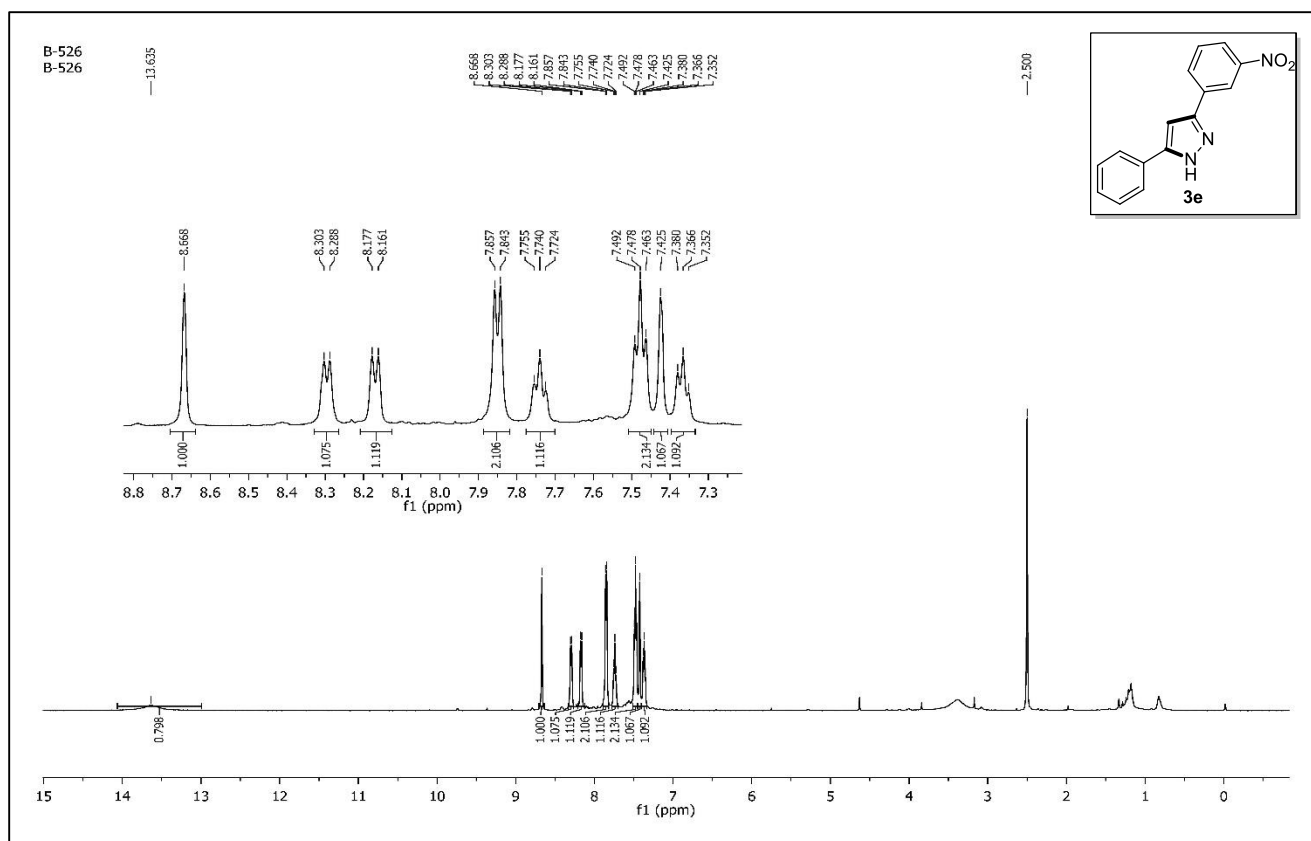


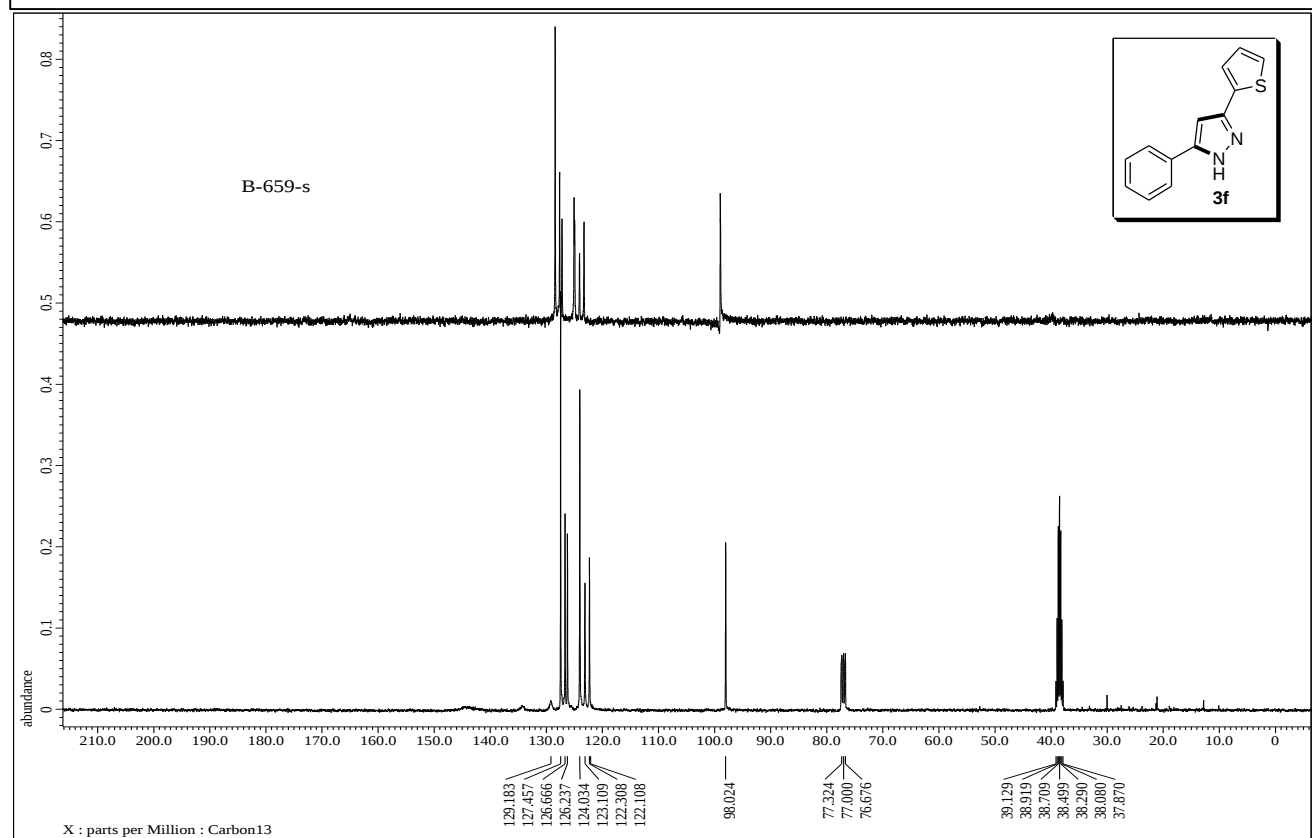
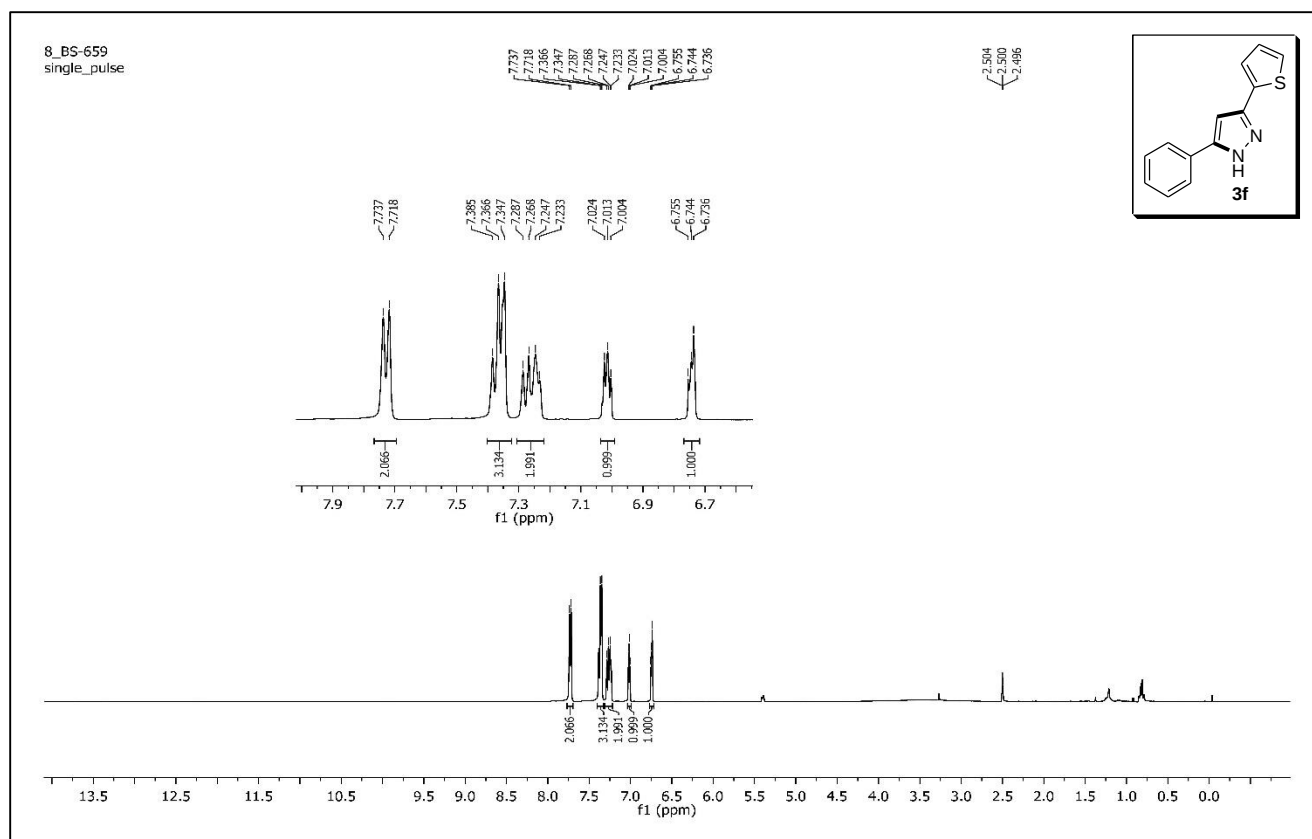


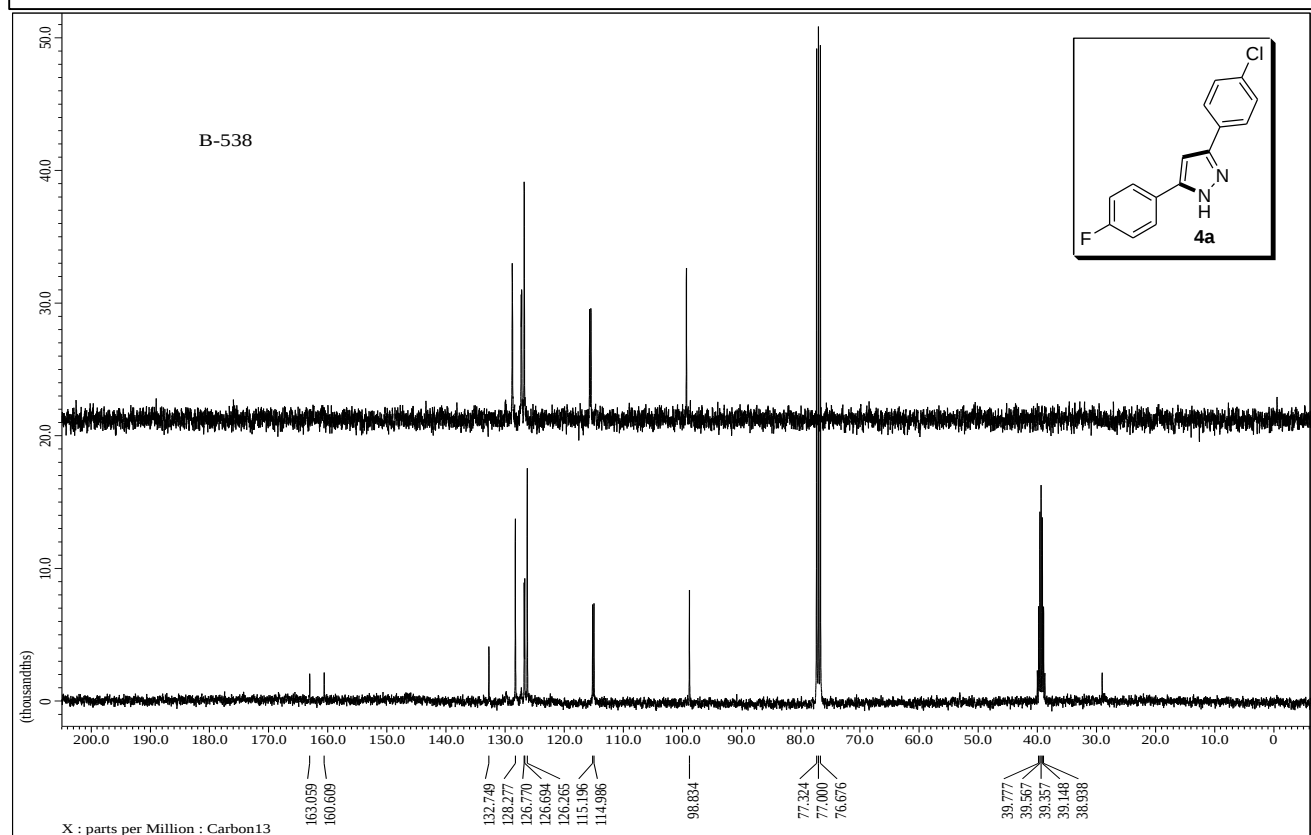
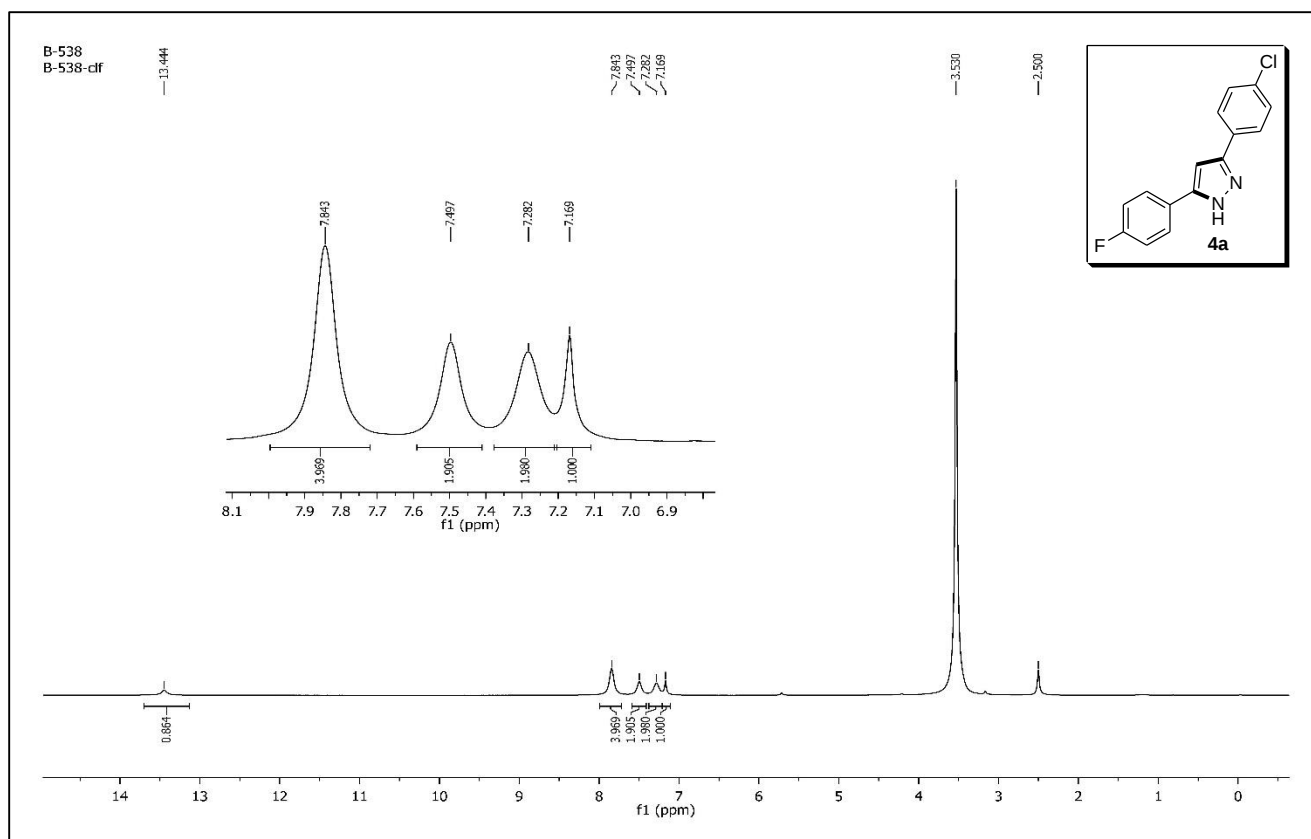
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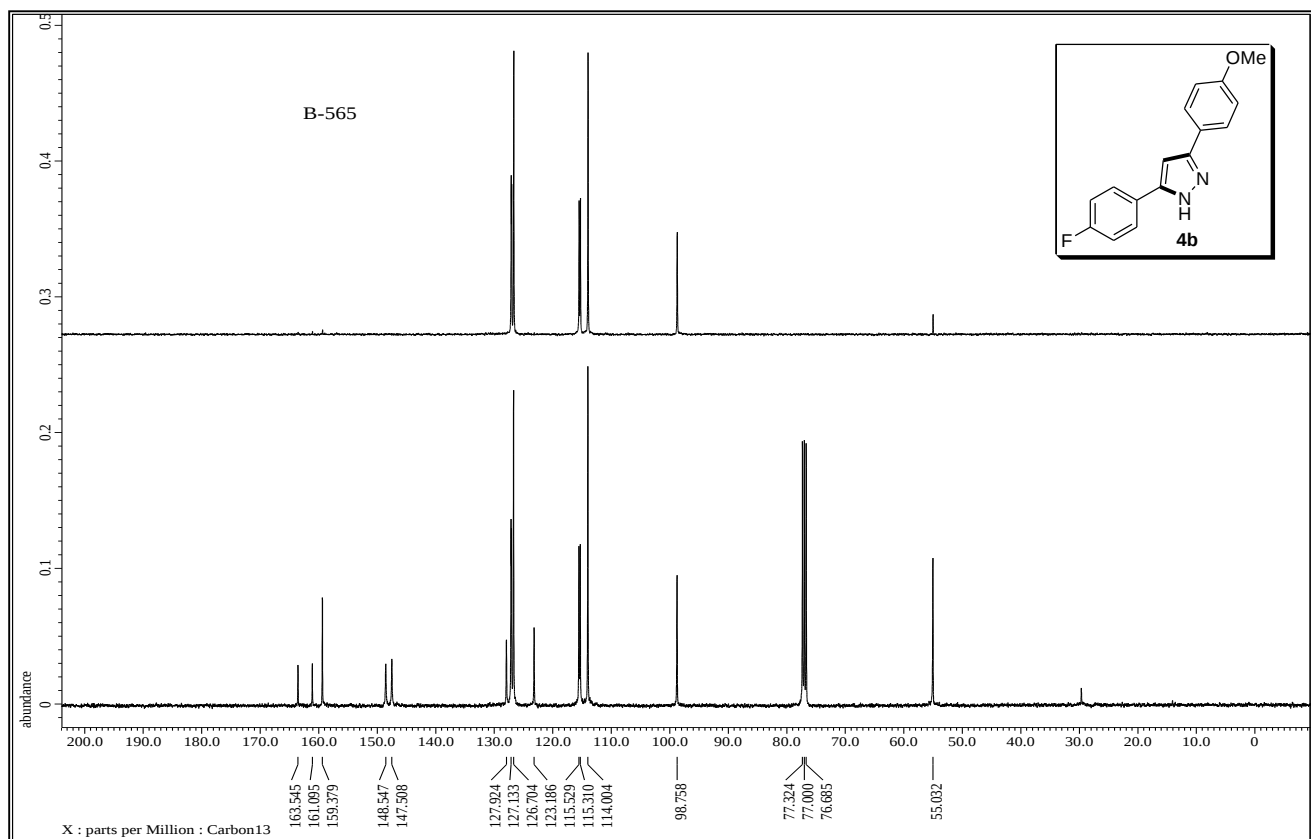
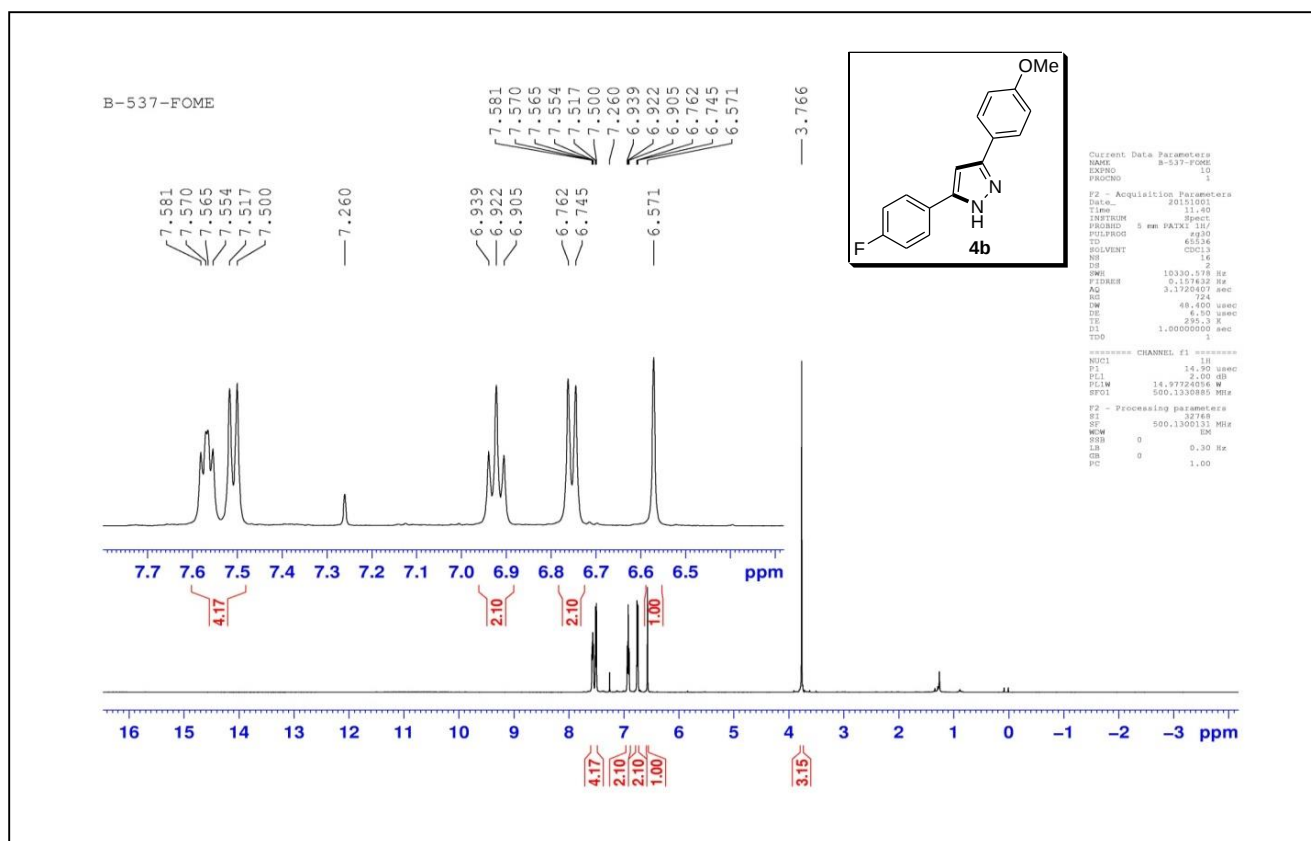


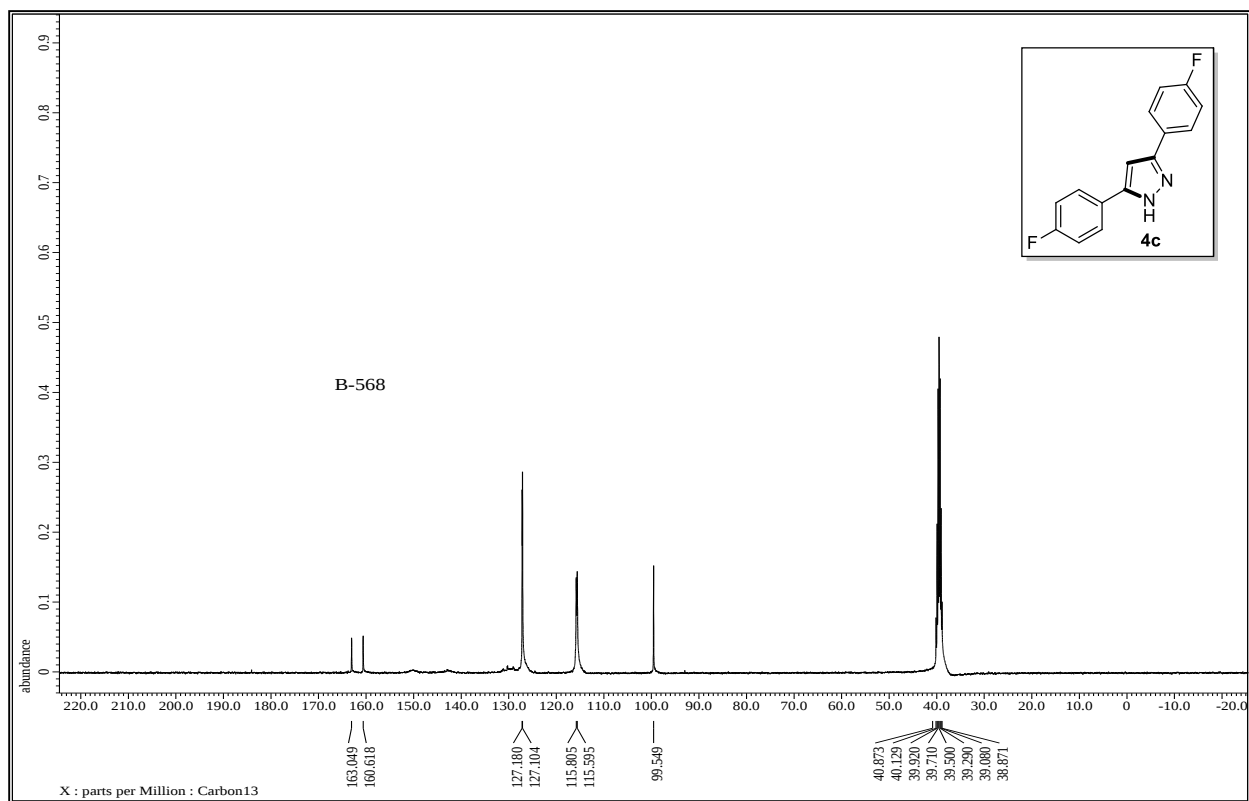
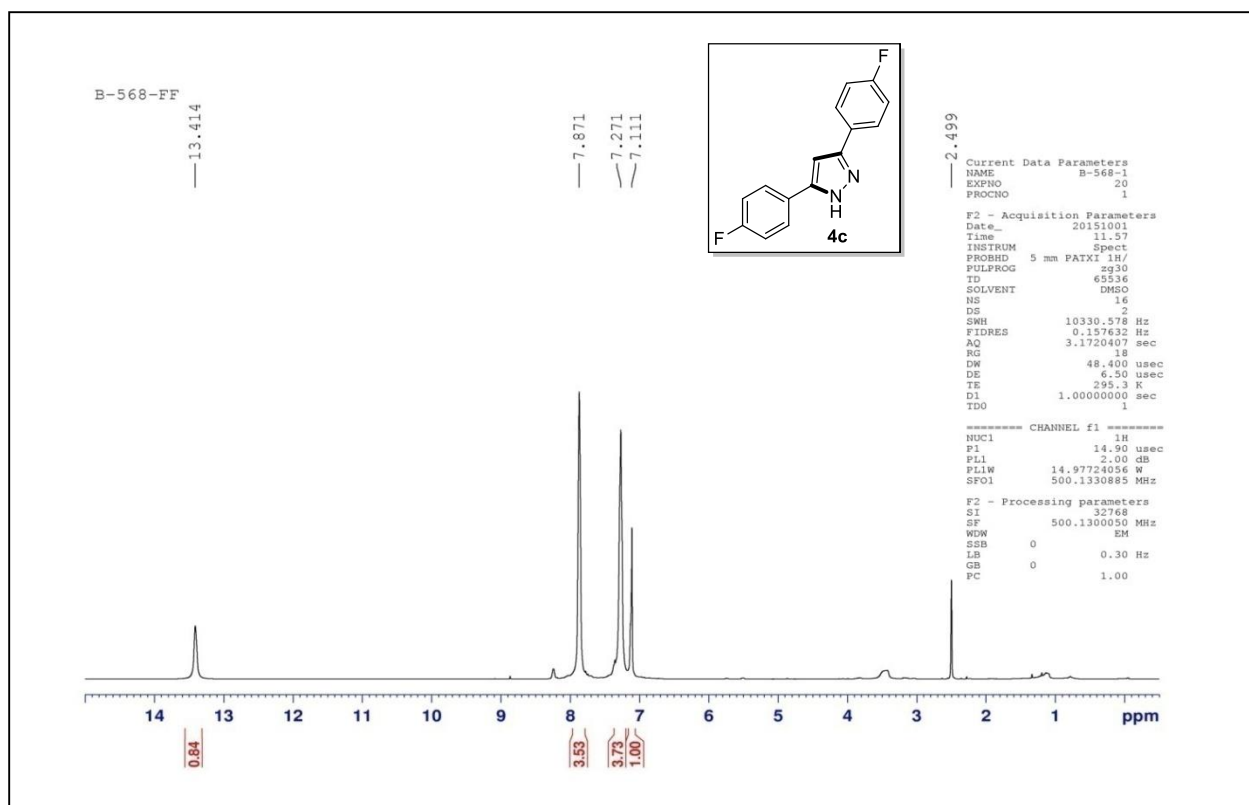


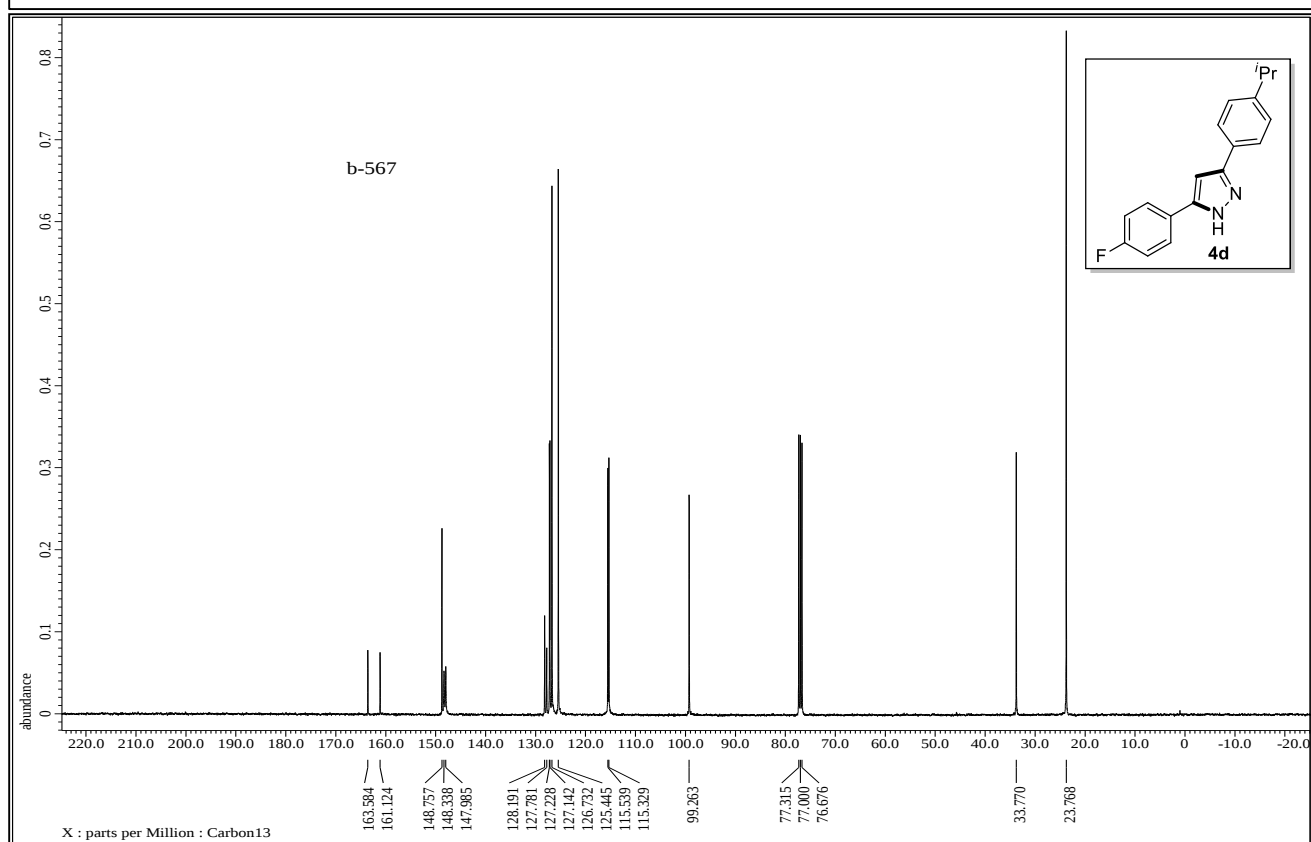
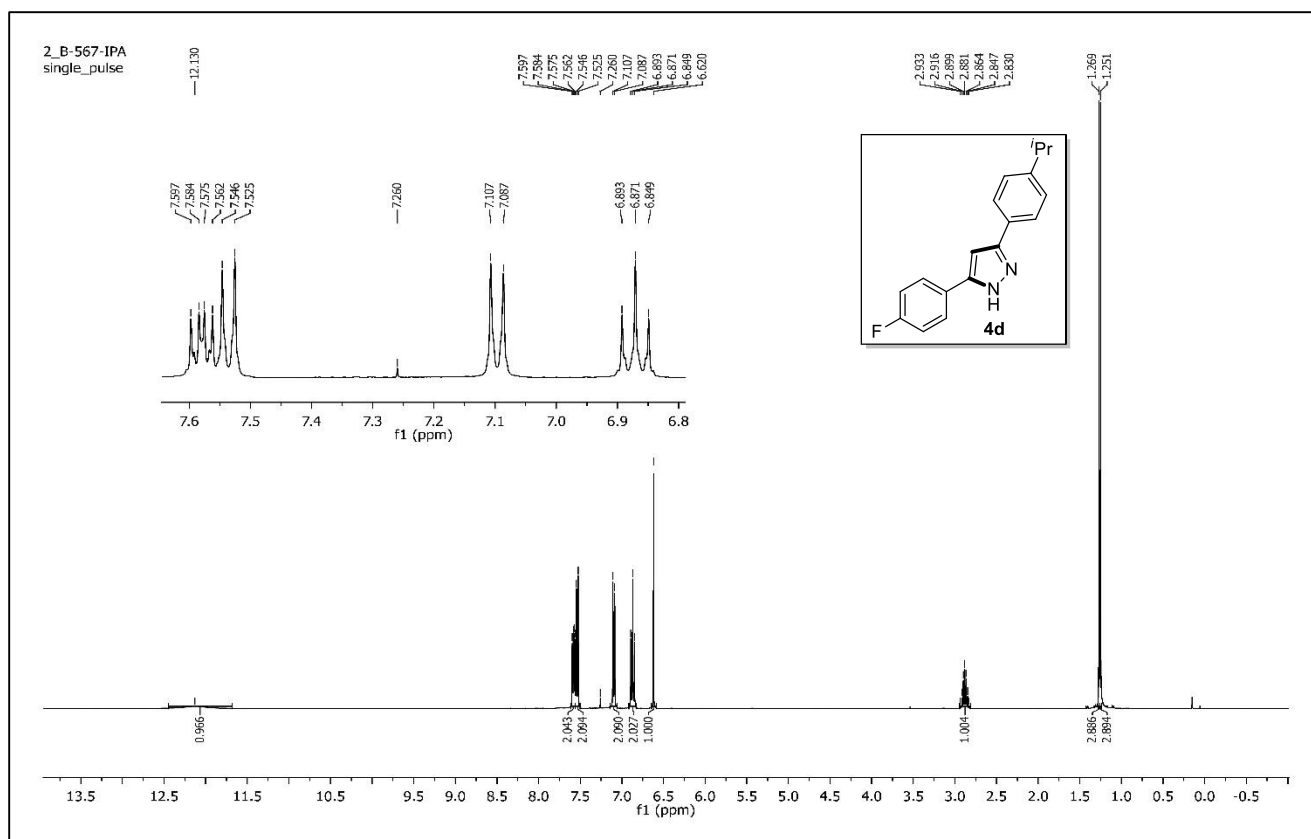


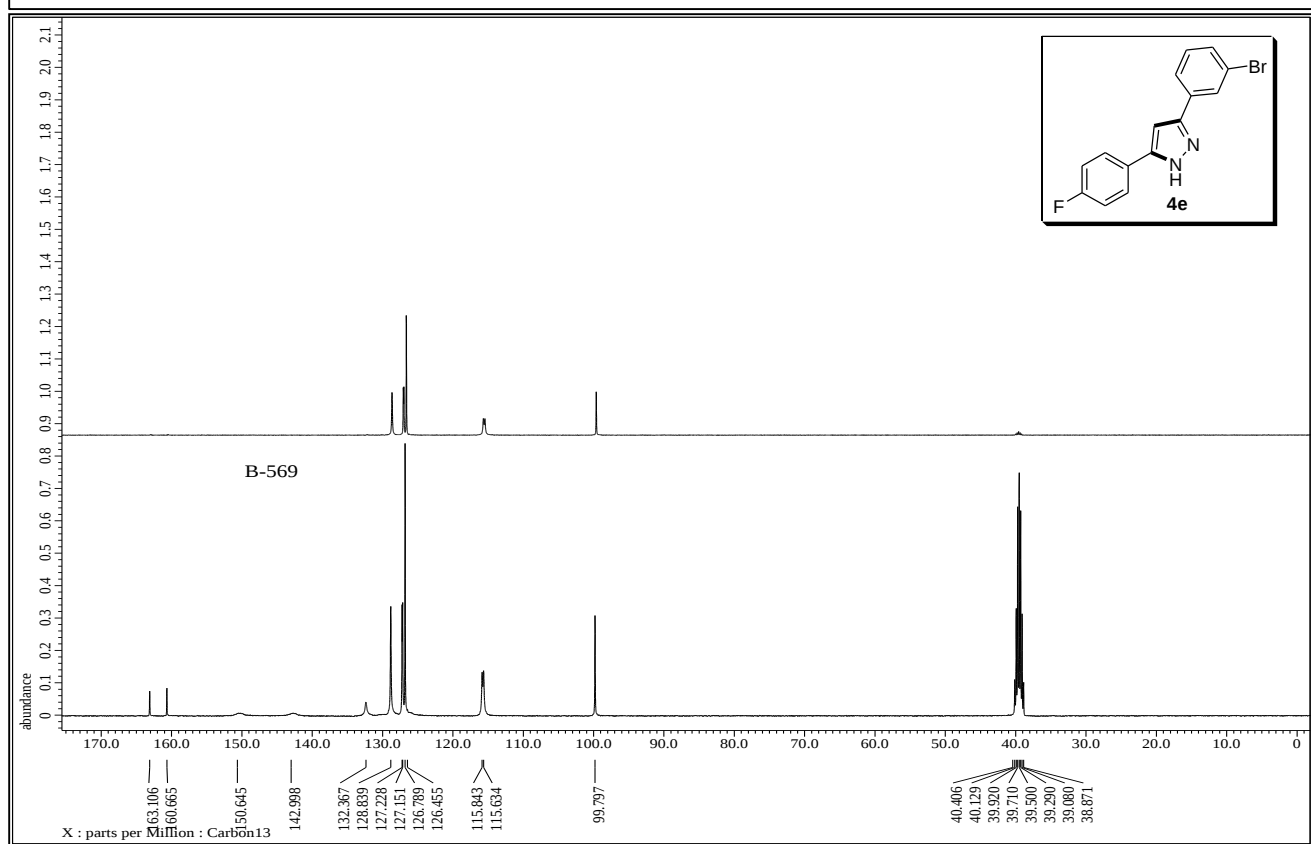
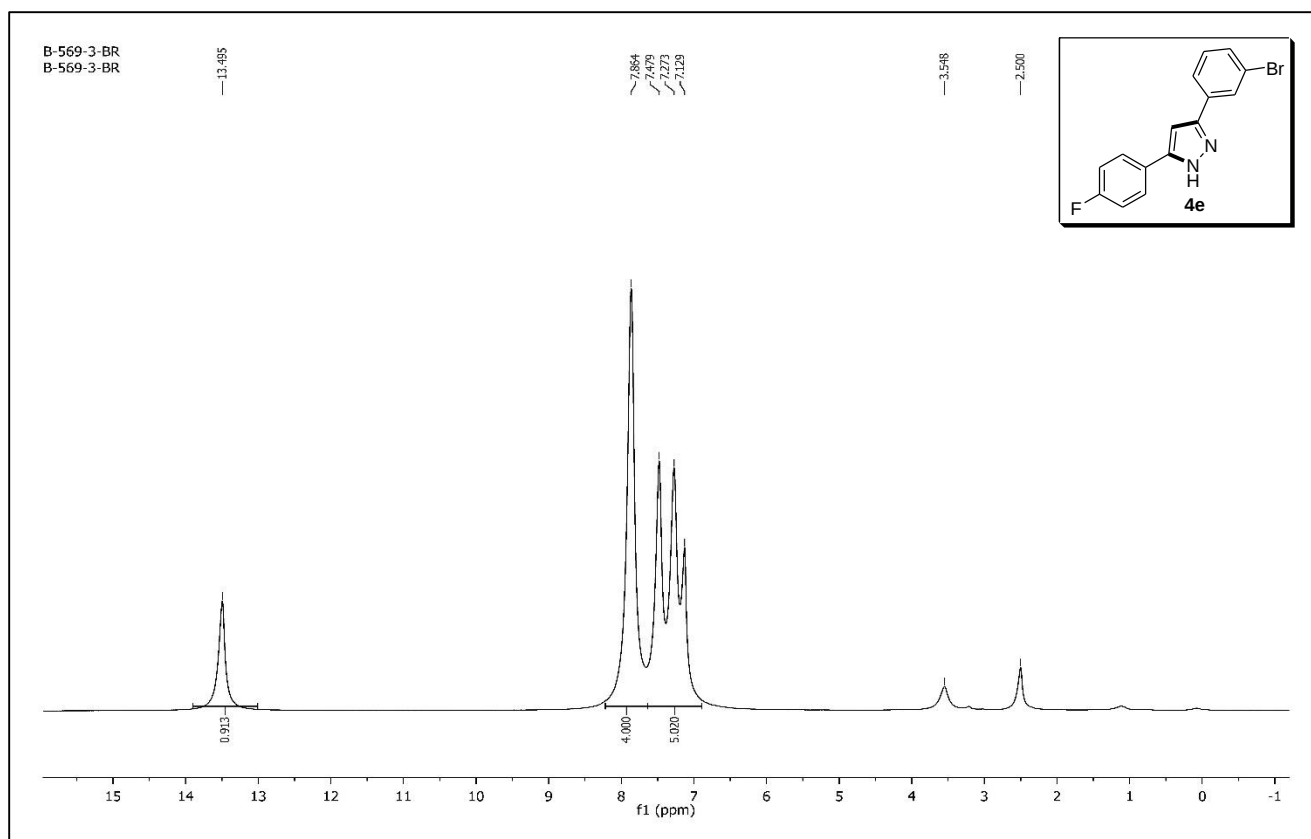


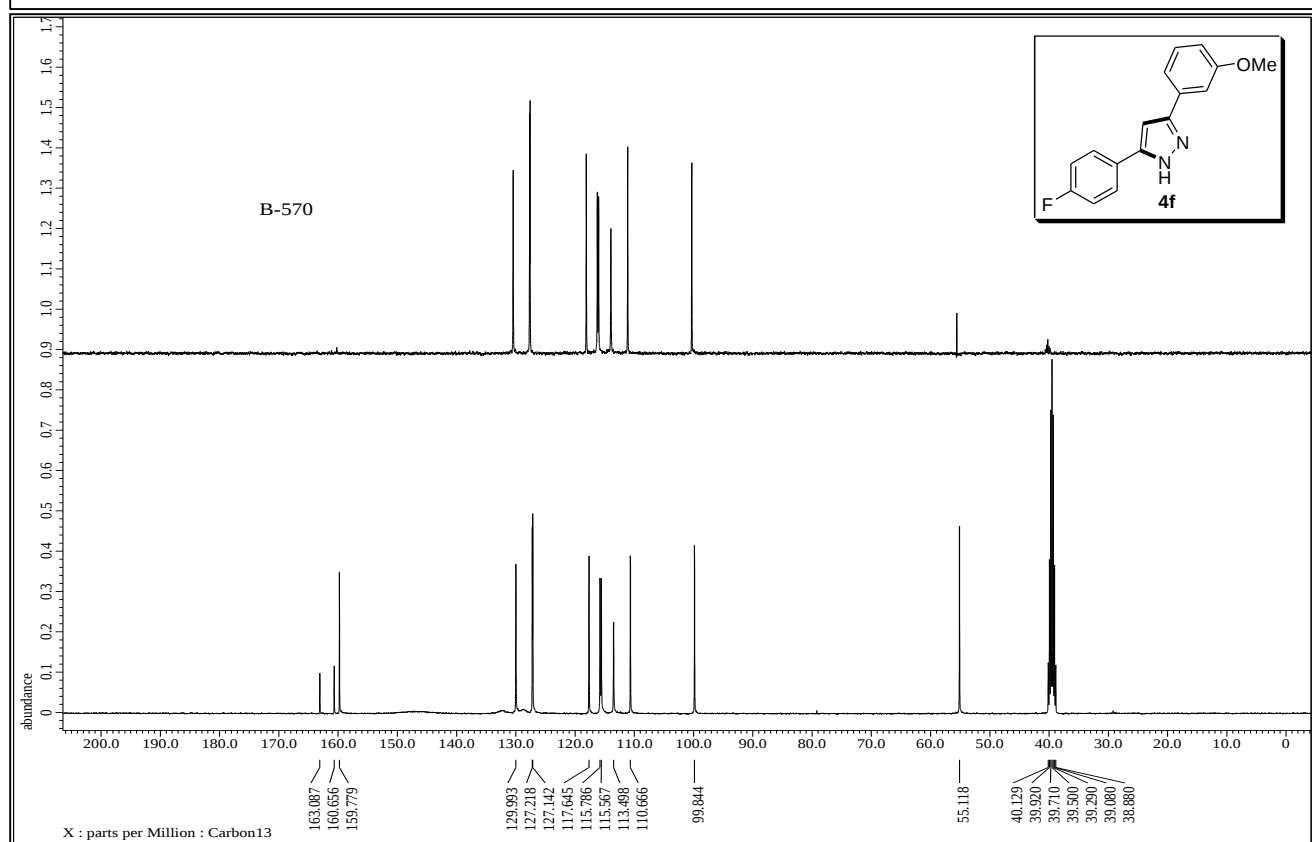
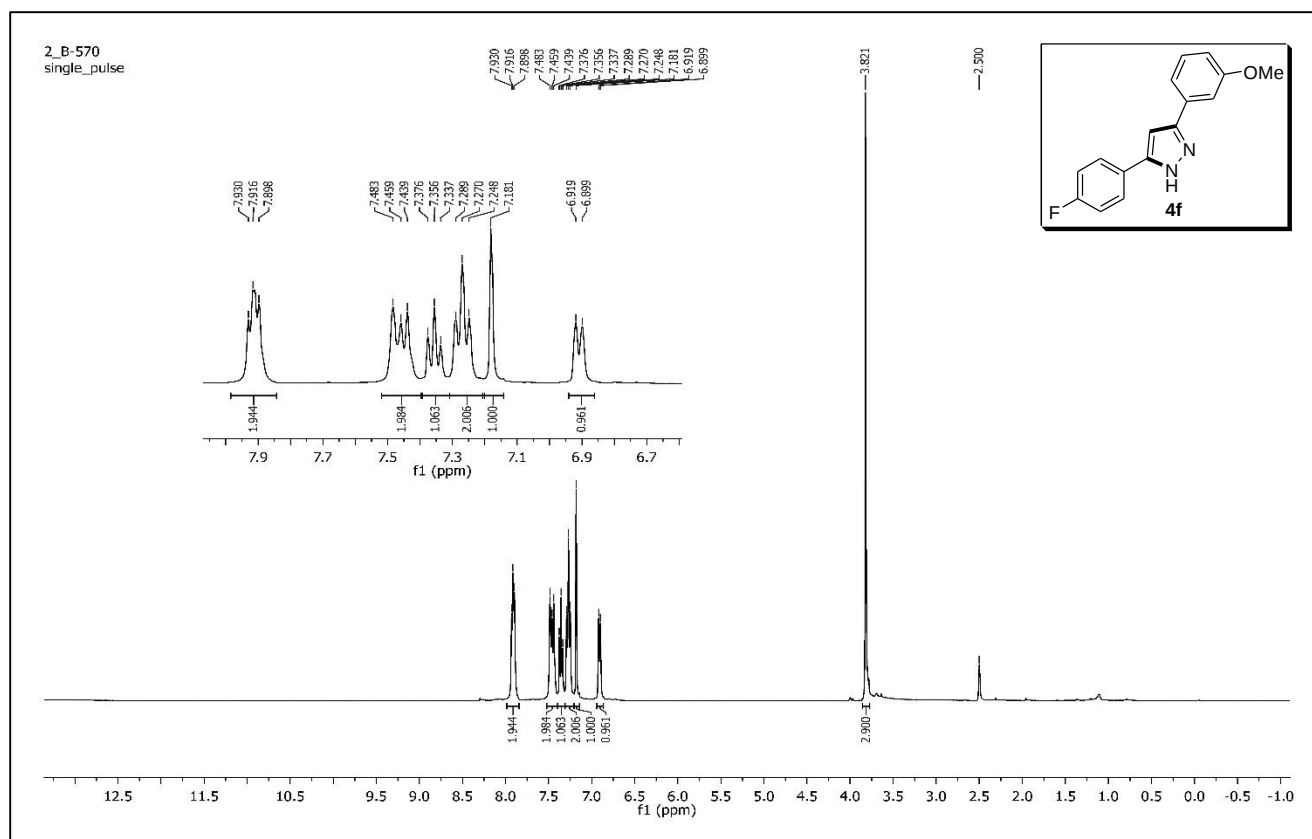


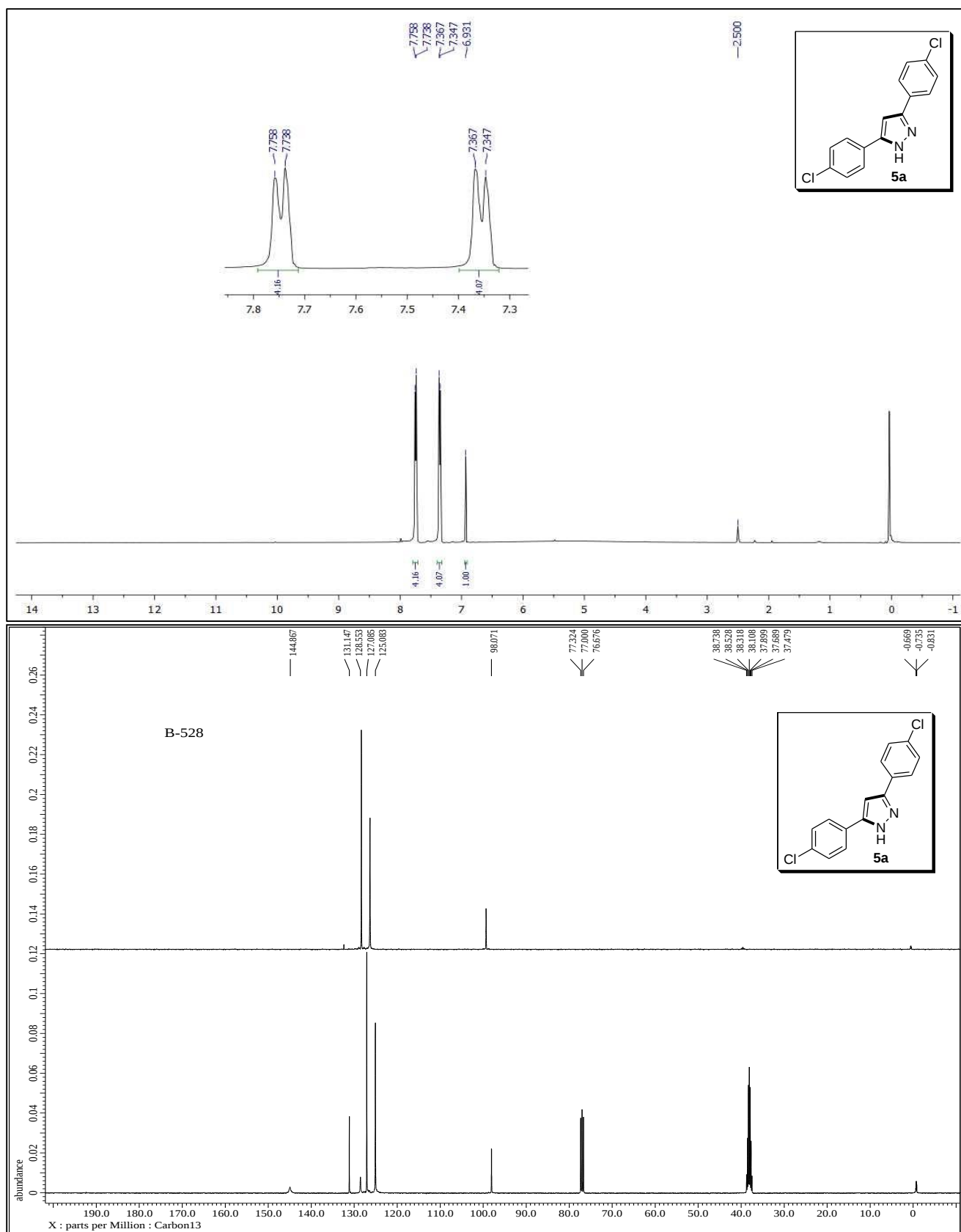












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