Supporting information

Unprecedented 22,26-*seco* physalins from *Physalis angulata* and their anti-inflammatory potential

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Experimental Section

General methods and materials.

Optical rotations were recorded on a Perkin-Elmer 241 polarimeter. UV spectra were measured with a Shimadzu UV 2201 spectrophotometer. ECD spectra were measured with a Bio-Logic Science MOS-450 spectrometer. IR spectra were recorded on a Bruker IFS 55 spectrometer. Bruker AV-600 spectrometer was used to record NMR spectra. Chemical shift values are expressed in δ (ppm) using the peak signals of the solvent DMSO- d_6 (δ_H 2.50 and δ_C 39.5) as reference, and coupling constants (J in Hz) are given in parentheses. HRESIMS data were acquired on an Agilent 6210 TOF mass spectrometer. Silica gel GF254 prepared for TLC was purchased from Qingdao Marine Chemical Factory (Qingdao, China). Silica gel (200–300 mesh, Qingdao Marine Chemical Factory, Qingdao, China) and octadecyl silica gel (Merck Chemical Company Ltd, German) were used for column chromatography (CC). RP-HPLC separations were conducted using an LC-6AD liquid chromatograph and a SPD-20A UV detector (Shimadzu, Kyoto, Japan) with a RP-C18 column (250 × 20 mm, 120 Å, 5 µm, YMC Co. Ltd). Spots were detected on TLC plates under UV light or by heating after spraying with anisaldehyde-H2SO4 reagent.

Plant material.

The stems and leaves of *P. angulata* were collected from Nanning, Guangxi Province, China, in July 2013, and identified by Pharmacist Jia-Fu Wei, Guangxi Institute for Food and Drug Control. A voucher specimen (PA-20130826) has been deposited in the herbarium of the Department of Natural Products Chemistry, Shenyang Pharmaceutical University.

Extraction and isolation.

The dried stems and leaves of P. angulata (9.5 kg) were extracted with 75% EtOH $(110 L \times 2 h \times 2)$. The resulting extracts (1.3 kg) were concentrated *in vacuo*, suspended in H₂O (5 L), and partitioned with petroleum ether, EtOAc, and *n*-BuOH (5 L \times 3), successively. The EtOAc extracts (116 g) were subjected to silica gel CC (10×80 cm) eluted with CH₂Cl₂/MeOH (100:1, 80:1, 60:1, 40:1, 20:1, 10:1, 8:1, 5:1, 3:1, 1:1, and 0:1 v/v) to obtain six fractions (E1–E6). Fraction E3 (35 g) was subjected to silica gel CC (6×80 cm) eluted with petroleum ether/acetone (10:1 to 0:1) to produce seven fractions (E31–E37). Fraction E33 (4.0 g) was separated by ODS CC (3×50 cm) with elution using a gradient of increasing MeOH (10-100%) in H₂O (1:9 to 1:0) to obtain three fractions (E331–E333). Fraction E331 (2 g) was chromatographed over silica gel CC (2 \times 50 cm), eluted with a gradient of increasing acetone (0–100%) in petroleum ether, to obtain four fractions (E3311-E3314). The separation of fraction E3314 (1.1 g) by silica gel CC (2×50 cm) with CHCl₃/MeOH (80:1 to 1:1), preparative TLC (CHCl₃/acetone, 4:1) and preparative HPLC (65% MeOH/H₂O, 6 mL min⁻¹) yielded compound 2 (4 mg). Fraction E4 (15 g) was subjected to silica gel CC (5×70 cm), eluting with CHCl₃/acetone (80:1 to 1:1), to afford five fractions (E41–E45). Fraction E45 (4 g) was separated by ODS column (3×50 cm) eluted with MeOH/H₂O (1:9 to 1:0), preparative TLC (CH₂Cl₂/acetone, 2:1), and preparative HPLC (65% MeOH/H₂O) to obtain compound 1 (2 mg).

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Physalin X (1): amorphous powder; [α]-28.0 (c 0.05, MeOH); UV (MeOH) λ_{max} (log ε) 220 (3.5) nm; IR (KBr) v_{max} 3395, 2921, 2850, 1647, 1469, 1384, 1120

cm⁻¹; ¹H (400 MHz, DMSO- d_6) and ¹³C-NMR (100 MHz, DMSO- d_6) data, see Table 1; HRESIMS *m/z* 593.2234 [M – H]⁻ (calcd for C₂₉H₃₇O₁₃, 593.2234).

Aromaphysalin B (2): amorphous powder; $[\alpha] = -38.0$ (*c* 0.1, MeOH); UV (MeOH) λ_{max} (log ε) 218 (3.9), 277 (3.1) nm; IR (KBr) v_{max} 3395, 3189, 2920, 2849, 1646, 1469, 1420, 1384, 1119, 879, 803, 721 cm⁻¹; ECD ($\Delta \varepsilon$) 222 (+ 0.36), 242 (-2.21), 334 (- 0.80) nm; ¹H (400 MHz, DMSO-*d*₆) and ¹³C-NMR (100 MHz, DMSO-*d*₆) data, see Table1; HRESIMS *m/z* 543.1862 [M – H]⁻, calcd for C₂₈H₃₁O₁₁ 543.1866.

ECD calculation.

In general, conformational analyses were carried out via random searching in the Sybyl-X 2.0 using the MMFF94S force field with an energy cutoff of 2.5 kcal/mol.¹ Due to the limitations of NOESY correlations, the results showed four lowest energy conformers for 6*S*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and only one for 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and only one for 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and only one for 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2**. Subsequently, the conformers were re-optimized using DFT at the B3LYP/6-31+G(d) level in gas phase by the GAUSSIAN 09 program. The energies, oscillator strengths, and rotational strengths (velocity) of the first 60 electronic excitations were calculated using the TDDFT methodology at the B3LYP/6-311++G(d,p) level in vacuum. The ECD spectra were simulated by the overlapping Gaussian function (half the bandwidth at 1/e peak height, $\sigma = 0.25$ eV). To get the final spectra, the simulated spectra of three conformers with the proportion more than one percent were averaged according to the Boltzmann distribution theory and their relative Gibbs free energy (Δ G). Theoretical ECD spectra of 6*S*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2** and 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-**2**.

NO Production Bioassay.

Compounds 1 and 2 were assayed for the inhibition of NO production according to the Griess method.¹⁻³ 1 × 10⁶ Cells/well of RAW 264.7 cells were added into the 96-well plates, and incubated at 37 °C for 24 h by the stimulation of LPS (1 µg/mL) with or without test compounds. After the addition of Griess reagent [0.1% *N*-(1-naphthyl)-ethylenediamine (50 µL); 1% sulfanilamide in 5% H₃PO₄ (50 µL)], absorbance (540 nm) was recorded by using a microplate reader. The standard curve was used to calculate the NO concentrations and inhibitory rates.

Molecular Modeling.

The crystal structure of protein was taken from Protein Data Bank (PDB ID: 5FVX), chain A, HEME, and Zn²⁺ were used for docking according to our previous method.⁴ The protein was modelled with CHARMM 22 force field. The 3D structure of compound was prepared and Gasteiger-Hückel charges was added using Sybyl software. Each ligand was subjected to energy minimization with Tripos force filed parameters. Blind docking was carried out using AutoDock 4 program. The 3D docking grid was sufficiently large to cover the protein. A total 100 conformations of each ligand were searched using Lamarckian generic algorithm, and the final docking conformations were clustered into different number of clusters in terms of the root-mean-square deviations of the ligand within the binding pocket of the receptor.

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Figure S6. HRESIMS spectrum of 1



Figure S7. HPLC chromatogram of 1. The purity of 1 is 94.68% according to the HPLC analysis result.













Figure S13. HRESIMS spectrum of 2



C1



Figure S14. The stable conformers of compound 6S, 11S, 13S, 15S, 16S, 17R, 20S, 22R, 24S-2



Figure S15. The stable conformers of compound 6*R*, 11*S*, 13*S*, 15*S*, 16*S*, 17*R*, 20*S*, 22*R*, 24*S*-2



Figure S16. The molecular docking result of compound 2 with NOS

 Table S1. B3LYP-calculated relative energies (Kcal/mol) and conformational population (%)

 for the most stable conformers of 6S, 11S, 13S, 15S, 16S, 17R, 20S, 22R, 24S-2

conf	$\Delta E_{6-31+G(d)}^{a}$	%ob
C1	0.00	84.75
C2	0.0022904	7.48
C3	0.0025292	5.80
C4	0.0035472	1.97

^aRelative to C1 with $E_{6-311++G(d,p)} = -1953.5332102$ kcal/mol. ^bCalculated using free energy values from Gaussian 03W according to $\Delta G = -RT \ln K$.

 Table S2. B3LYP-calculated relative energies (Kcal/mol) and conformational population (%)

for t	he most st	able conf	formers of	'6R,	11 <i>S</i> ,	13 <i>S</i> , 1	15S, 1	16S, 17	VR, 20S,	22R, 2	4S-2
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conf	$\Delta E_{6-31+G(d)}^{a}$	⁰∕₀ ^b
C1	0.00	93.16
C2	0.0029556	4.06
C3	0.003313	2.78

^aRelative to **C1** with $E_{6-311++G(d,p)} = -1914.22289$ kcal/mol. ^bCalculated using free energy values from Gaussian 03W according to $\Delta G = -RT$ In *K*.