Supporting Information

Metal-Free Oxidative *para*-Acylation of Unprotected Anilines with Methyl Groups *via* Dual C-H Activation

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General information

All reactions were carried out under O_2 atmosphere (1 atm) using standard Schlenk technique in the parallel synthesizer. All reagent/reactant were commercially available unless other noted (I₂ and salicylic acid were purchased from Energy Chemical). DMSO used was purchased from Aladdin Chemical. Column chromatography was performed using Silica Gel 60 (particle size 37-54 µm). The pure products were obtained by column chromatography using ethyl acetate/petroleum ether/NEt₃ as the eluent and characterized by NMR spectroscopy using d_6 -DMSO as the deuterium reagent. GC analysis was performed on GC 7820A (Shimadzu). GC-MS results were recorded on GC-MS QP2010 (Shimadzu). The ¹H NMR and ¹³C NMR data were acquired on a Brucker ADVANCE III spectrometer (400 MHz for ¹H NMR spectroscopy and 100 MHz for ¹³C NMR spectroscopy). HRMS analysis was conducted at the College of Chemistry and Materials Engineering, Wenzhou University, Wenzhou, 325000, China

Synthetic procedures for 5a, 7a, 1a-CD₃

A 25 mL schlenk tube was charged with PPh₃ (0.03 mmol), NaHCO₃ (0.45 mmol), I₂ (0.45 mmol) and a stir bar, then degassed and refilled with N₂ for 3 times. After addition of 2 mL anhydrous THF and **1a** (0.3 mmol) under N₂ atmosphere with stirring, the mixture was further stirred overnight at 120°C and then cooled to room temperature. The mixture was quenched with 5-10 mL saturated Na₂S₂O₃ solution. Afterwards, the solution was extracted with CH₂Cl₂ (3×3 mL) and washed with brine (3×5 mL). The organic layer was dried over Na₂SO₄ (10-30 min) and concentrated, the residues were passed through a short silica chromatography column (particle size 37–54 µm, petroleum ether/ethyl acetate as eluent) to afford analytically pure product **5a** (yellow solid).

To a 100 mL round bottom flask charged with **6a** (0.383g, 2.44 mmol) and a stir bar, 10 mL anhydrous CH_2Cl_2 was added under N_2 atmosphere. Then aniline (0.227 g, 2.44 mmol) was added dropwise with stirring. The solution was further stirred for 10 h at room temperature. After reaction, the mixture was washed with deionized water and extracted with DCM (2×10 mL), the organic layer was dried over Na_2SO_4 and concentrated to afford analytically pure product **7a** (yellow solid, 0.52g, 92% yield).

A 10 mL schlenk tube was charged with **1a** (0.135 mL, 1mmol), D_2O (1 mL), PhCOOH (9.15 mg, 0.075 mmol) and a stir bar. The mixture was heated at 80 °C for 6 h and cooled to room temperature. The mixture was neutralized with 5 mL saturated NaHCO₃ aqueous solution and extracted with EA (3×5 mL). The organic layer was dried over Na₂SO₄ and concentrated to afford analytically pure product **1a-CD₃** (yellow liquid, D-96%).

KIE experiment procedure

1a-CD₃: An oven-dried Schlenk tube containing a stir bar was charged with I₂ (0.04 mmol, 10.2 mg) and salicylic acid (0.04 mmol, 5.6 mg), the tube was degassed and refilled with O₂ for 3 times. Then 0.2 mL DMSO, **1a-CD₃** d_3 -2-methylquinoline (0.2 mmol, 29.2 mg), **2a** aniline (0.6 mmol, 54 uL) was charged into the tube under O₂ successively. Then the mixture was stirred at 60 °C for 3 h. After reaction, the mixture was diluted with 3-5 mL DCE and then neutralized by saturated NaHCO₃ aqueous solution. The mixture was analyzed by GC using tridecane as the internal standard and the yield of **3a** was 3.4%. The reaction of **1a** with **2a** gave **3a** in 11% yield at 3 h. Thus, the $k_{\rm H}/k_{\rm D} = 3.3$.

 d_5 -PhNH₂: An oven-dried Schlenk tube containing a stir bar was charged with I₂ (0.04 mmol, 10.2 mg) and salicylic acid (0.04 mmol, 5.6 mg), the tube was degassed and refilled with O₂ for 3 times. Then 0.2 mL DMSO, 2-methylquinoline (0.2 mmol, 29.2 mg), **2a** d_5 -aniline (0.6 mmol, 54 uL) was charged into the tube under O₂ successively. Then the mixture was stirred at 60 °C for 3 h. After reaction, the mixture was diluted with 3-5 mL DCE and then neutralized by saturated NaHCO₃ aqueous solution. The mixture was analyzed by GC using tridecane as the internal standard and the yield of **3a**-D₄ was 5.5%. The reaction of **1a** with **2a** gave **3a** in 11% yield at 3 h. the $k_{\rm H}/k_{\rm D} = 2.0$.

Reference

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