

SUPPLEMENTARY INFORMATION

Structure-reactivity correlations of the abnormal Beckmann reaction of dihydrolevoglucosenone oxime

Amani Alhifithi, Lars Goergik, Jonathan M. White,* Spencer J. Williams*

School of Chemistry and Bio21 Molecular Science and Biotechnology Institute, University of Melbourne, Parkville, Victoria, Australia, 3010

Table of Contents

<i>Table S1. Bond distances and angles for oximes 1-8 derived from single crystal X-ray structures.</i>	3
<i>Table S2. Natural atomic charges for oximes at PW6B95/def2-TZVP level of theory.</i>	5
<i>Table S3. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the revPBE/def2-TZVP level of theory.</i>	6
<i>Table S4. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the revPBE/def2-TZVP level of theory.</i>	6
<i>Table S5. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the TPSS/def2-TZVP level of theory.</i>	7
<i>Table S6. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the TPSS/def2-TZVP level of theory.</i>	7
<i>Table S7. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the B3LYP/def2-TZVP level of theory.</i>	8
<i>Table S8. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the B3LYP/def2-TZVP level of theory.</i>	8
<i>Table S9. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the PBE0/def2-TZVP level of theory.</i>	9
<i>Table S10. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the PBE0/def2-TZVP level of theory.</i>	9
<i>Table S11. NBO hybrid orbitals for the σ_{C1-C2} bond.</i>	10
<i>Table S12. NBO hybrid orbitals for the σ_{C2-N} and π_{C2-N} bonds.</i>	11
<i>Table S13. NBO hybrid orbitals for the σ_{C1-O5} bond.</i>	12
<i>Table S14. NBO hybrid orbitals for the σ_{C1-O6} bond.</i>	13
<i>Table S15. NBO hybrid orbitals for the σ_{N-O} bonds.</i>	14

<i>Table S16. NBO hybrid orbitals for the σ_{C2-C3} bond.</i>	15
<i>Table S17. NBO hybrid orbitals for the N lone pair.</i>	16
<i>Figure S1. Plots showing variations in a range of geometric parameters for the series of oximes 1-8. (a) Bond distances and (b) angles, as a function of leaving group ability.</i>	17
<i>Figure S2. Plots showing variations in (a) C2 chemical shift and (b) C1 chemical shift, as a function of leaving group ability.</i>	18
<i>Figure S3. Plots showing comparison of $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions at the TPSS and PW6B95 levels of theory, as a function of leaving group ability.</i>	19
<i>Optimized Geometries of Cyrene Oximes</i>	20
(Z)-Dihydrolevoglucosenone oxime	20
(Z)-Dihydrolevoglucosenone O-(phenyl)oxime	21
(Z)-Dihydrolevoglucosenone O-(acetyl)oxime	23
(Z)-Dihydrolevoglucosenone O-(3-nitrobenzoyl)oxime	24
(Z)-Dihydrolevoglucosenone O-(3,5-dinitrobenzoyl)oxime	26
(Z)-Dihydrolevoglucosenone O-(trifluoroacetyl)oxime	28
Cyclohexanone oxime	29
Cyclohexanone O-(phenyl)oxime	30
Cyclohexanone O-(acetyl)oxime	32
Cyclohexanone O-(3-nitrobenzoyl)oxime	33
Cyclohexanone O-(3,5-dinitrobenzoyl)oxime	35
Cyclohexanone O-(trifluoroacetyl)oxime	37
<i>¹H-NMR and ¹³C-NMR spectra for Oximes (1-8):</i>	38
(Z)-Dihydrolevoglucosenone oxime (1).	38
(Z)-Dihydrolevoglucosenone O-(4-bromobenzyl)oxime (2).	40
(Z)-Dihydrolevoglucosenone O-(phenyl)oxime (3).	42
(Z)-Dihydrolevoglucosenone O-(4-nitrophenyl)oxime (4).	44
(Z)-Dihydrolevoglucosenone O-(acetyl)oxime (5).	46
(Z)-Dihydrolevoglucosenone O-(3-nitrobenzoyl)oxime (6).	48
(Z)-Dihydrolevoglucosenone O-(4-nitrobenzoyl)oxime (7).	50
(Z)-Dihydrolevoglucosenone O-(3,5-dinitrobenzoyl)oxime (8).	52
Dihydrolevoglucosenone O-(2-nitrobenzoyl)oxime (11).	56
Nitriles 12 and 13 .	58

Table S1. Bond distances and angles for oximes 1-8 derived from single crystal X-ray structures.

Distances are in Ångstroms, angles are in degrees. pK_a values are from Ref. {Dean, 1999 #7949} Except for 4-bromobenzyl alcohol, for which the pK_a value was predicted using the Advanced Chemistry Development (ACD/Labs) Software V11.02.

pK_a value	N-O	C1-C2	C2-C3	C1-O1	C1-O5	N-C2	C1-C2-N	C2-N-O	O1-C1-O5	C3-C2-N	O1-C1-C2	O5-C1-C2
16	1.405(1)	1.503(2)	1.500(1)	1.416(2)	1.425(1)	1.278(1)	116.4(4)	112.69(9)	105.39(9)	127.9(1)	108.31(9)	109.98(9)
	1.408(1)	1.505(2)	1.500(2)	1.409(1)	1.428(1)	1.280(1)	115.8(1)	112.27(9)	105.45(9)	127.1(1)	109.33(9)	109.11(9)
14.2	1.417(5)	1.505(5)	1.504(6)	1.405(5)	1.420(5)	1.278(5)	115.5(3)	111.5(3)	105.8(3)	128.7(4)	110.1(3)	109.3(3)
	1.410(5)	1.502(5)	1.499(6)	1.415(5)	1.425(5)	1.276(5)	115.5(3)	112.1(3)	105.9(3)	130.0(4)	107.7(3)	110.3(3)
9.9	1.426(2)	1.505(2)	1.449(2)	1.403(2)	1.425(2)	1.276(2)	114.3(1)	110.5(1)	106.0(1)	129.1(1)	109.2(1)	109.0(1)
7.15	1.437(2)	1.513(3)	1.500(3)	1.405(2)	1.425(2)	1.273(2)	115.5(2)	109.7(1)	106.2(1)	129.0(2)	109.0(2)	108.1(2)
4.75	1.441(1)	1.515(2)	1.499(2)	1.403(2)	1.423(2)	1.276(2)	114.4(1)	109.4(1)	106.1(1)	129.9(1)	110.0(1)	107.6(1)
3.46	1.442(1)	1.515(1)	1.501(1)	1.406(1)	1.421(1)	1.275(2)	114.61(9)	109.30(8)	105.94(8)	130,4(1)	107.38(8)	110.03

3.44	1.445(3)	1.516(5)	1.499(5)	1.405(4)	1.411(4)	1.275(4)	114.0(3)	109.1(2)	106.2(3)	129.4(3)	109.3(3)	108.4(3)
	1.444(5)	1.509(5)	1.505((5)	1.413(4)	1.425(4)	1.271(4)	114.7(3)	108.8(3)	105.6(3)	130.7(3)	107.2(3)	110.1(3)
	1.448(4)	1.512(5)	1.498(5)	1.406(5)	1.420(4)	1.276(4)	115.9(3)	108.1(2)	105.8(3)	129.1(3)	108.7(3)	108.7(3)
	1.447(4)	1.510(5)	1.498(5)	1.412(4)	1.419(4)	1.272(5)	115.0(3)	108.0(3)	105.9(3)	130.6(3)	107.7(3)	109.9(3)
2.85	1.445(3)	1.516(5)	1.499(5)	1.405(4)	1.411(4)	1.278(3)	115.1(2)	108.3(2)	106.2(2)	128.9(2)	109.0(2)	108.1(2)

Table S2. Natural atomic charges for oximes at PW6B95/def2-TZVP level of theory.

R	pK _a	Cyrene oximes				Cyclohexanone oximes	
		C1	O5	O6	C2	C1	C-anti
H	16	0.3058 3	-0.48453	-0.49869	0.17671	0.22821	-0.43697
C ₆ H ₅	9.9	0.3054 4	-0.48230	-0.49703	0.20595	0.25918	-0.43845
CH ₃ CO	4.75	0.3046 0	-0.48048	-0.49566	0.21098	0.27141	-0.44008
3-NO ₂ C ₆ H ₄ CO	3.46	0.3044 3	-0.47820	-0.49448	0.21763	0.27124	-0.43989
3,5- (NO ₂) ₂ C ₆ H ₃ C O	2.85	0.3046 1	-0.47741	-0.49314	0.22550	0.28207	-0.44088
CF ₃ CO	0.50	0.3045 0	-0.47660	-0.49214	0.23163	0.28227	-0.44089

Table S3. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{\text{C1-C2}} \rightarrow \sigma_{\text{N-O}}^*$ interactions in Cyrene and cyclohexanone oximes at the revPBE/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	Cyrene oximes		
		$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i
H	16	4.81	0.46	1.96979
CH ₃ CO	4.75	5.42	0.44	1.96598
CF ₃ CO	0.50	5.96	0.42	1.96096

Table S4. Donor-acceptor stabilization energies ($E(2)$) for $n_{\text{O5}} \rightarrow \sigma_{\text{C1-C2}}^*$ and $n_{\text{O6}} \rightarrow \sigma_{\text{C1-C2}}^*$ interactions for Cyrene oximes at the revPBE/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	LP O5			LP O6		
		$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i	$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i
H	16	1.76	0.66	1.88928	1.79	0.65	1.89863
CH ₃ CO	4.75	1.85	0.65	1.88648	1.85	0.64	1.89639
CF ₃ CO	0.50	1.95	0.64	1.88361	1.95	0.64	1.89368

Table S5. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the TPSS/def2-TZVP level of theory.

R	pK_a	Cyrene oximes			Cyclohexanone oximes		
		$E(2)$	ΔE_{ij}	q_i	$E(2)$	ΔE_{ij}	q_i
		(kcal mol ⁻¹)	(au)		(kcal mol ⁻¹)	(au)	
H	16	4.84	0.49	1.97051	4.80	0.47	1.96761
C ₆ H ₅	9.9	5.13	0.49	1.96884	5.12	0.46	1.96591
CH ₃ CO	4.75	5.46	0.47	1.96671	5.36	0.45	1.96436
3-NO ₂ C ₆ H ₄ CO	3.46	5.51	0.47	1.96350	5.42	0.45	1.96400
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	5.69	0.46	1.96364	5.58	0.44	1.96252
CF ₃ CO	0.50	6.01	0.44	1.96171	5.90	0.42	1.95950

Table S6. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the TPSS/def2-TZVP level of theory.

R	pK_a	LP O5			LP O6		
		$E(2)$	ΔE_{ij}	q_i	$E(2)$	ΔE_{ij}	q_i
		(kcal mol ⁻¹)	(au)		(kcal mol ⁻¹)	(au)	
H	16	1.81	0.70	1.89305	1.85	0.69	1.90177
C ₆ H ₅	9.9	1.86	0.70	1.89160	1.85	0.69	1.90050
CH ₃ CO	4.75	1.91	0.69	1.89036	1.92	0.69	1.89964
3-NO ₂ C ₆ H ₄ CO	3.46	1.97	0.69	1.88918	1.94	0.68	1.89829
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	2.02	0.69	1.8820	1.98	0.68	1.89721
CF ₃ CO	0.50	2.02	0.69	1.88752	2.02	0.68	1.89698

Table S7. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the B3LYP/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	Cyrene oximes		
		$E(2)$	ΔE_{ij}	q_i
		(kcal mol ⁻¹)	(au)	
H	16	5.23	0.60	1.97314
CH ₃ CO	4.75	5.94	0.58	1.96958
CF ₃ CO	0.50	6.62	0.55	1.96512

Table S8. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the B3LYP/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	LP O5			LP O6		
		$E(2)$	ΔE_{ij}	q_i	$E(2)$	ΔE_{ij}	q_i
		(kcal mol ⁻¹)	(au)		(kcal mol ⁻¹)	(au)	
H	16	2.77	0.73	1.90473	2.91	0.73	1.91281
CH ₃ CO	4.75	2.90	0.73	1.90272	2.90	0.72	1.91135
CF ₃ CO	0.50	3.06	0.72	1.90046	3.15	0.71	1.90928

Table S9. Donor-acceptor stabilization energies ($E(2)$) for $\sigma_{C1-C2} \rightarrow \sigma^*_{N-O}$ interactions in Cyrene and cyclohexanone oximes at the PBE0/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	Cyrene oximes		
		$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i
H	16	5.79	0.65	1.97346
CH ₃ CO	4.75	6.58	0.63	1.96981
CF ₃ CO	0.50	7.34	0.60	1.96540

Table S10. Donor-acceptor stabilization energies ($E(2)$) for $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions for Cyrene oximes at the PBE0/def2-TZVP level of theory.

Calculations at this level of theory were only performed on a subset of structurally simple oximes.

R	pK_a	LP O5			LP O6		
		$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i	$E(2)$ (kcal mol ⁻¹)	ΔE_{ij} (au)	q_i
H	16	3.40	0.75	1.90483	3.59	0.75	1.91286
CH ₃ CO	4.75	3.55	0.75	1.90287	3.70	0.74	1.91140
CF ₃ CO	0.50	3.72	0.74	1.90068	3.87	0.73	1.90939

Table S11. NBO hybrid orbitals for the $\sigma_{\text{C1-C2}}$ bond.

R	pK_a	bond	atom	h_A	C_A	%s	%p	%d
H	16	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.49}d^{0.01}$	0.7043	28.64	71.20	0.15
			C2	$sp^{2.38}$	0.7099	29.54	70.32	0.10
C_6H_5	9.9	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.51}d^{0.01}$	0.7037	28.46	71.38	0.15
			C2	$sp^{2.41}$	0.7105	29.27	70.59	0.10
CH_3CO	4.75	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.54}d^{0.01}$	0.7029	28.20	71.63	0.15
			C2	$sp^{2.44}$	0.7113	29.06	70.80	0.10
3- $\text{NO}_2\text{C}_6\text{H}_4\text{CO}$	3.46	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.57}d^{0.01}$	0.7018	27.97	71.87	0.15
			C2	$sp^{2.43}$	0.7123	29.09	70.77	0.10
3,5-(NO_2) $_2\text{C}_6\text{H}_3\text{CO}$	2.85	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.60}d^{0.01}$	0.7010	27.77	72.06	0.15
			C2	$sp^{2.44}$	0.7131	29.02	70.84	0.10
CF_3CO	0.50	$\sigma_{\text{C1-C2}}$	C1	$sp^{2.61}d^{0.01}$	0.7007	27.64	72.19	0.15
			C2	$sp^{2.46}$	0.7134	28.85	71.02	0.10

h_A : natural atomic hybrids; C_A : polarization coefficient.

Table S12. NBO hybrid orbitals for the $\sigma_{\text{C2-N}}$ and $\pi_{\text{C2-N}}$ bonds.

R	pK_a	bond	atom	h_A	C_A	%s	%p	%d
H	16	$\sigma_{\text{C2-N}}$	C2	$sp^{1.84}$	0.6486	35.23	64.65	0.07
			N	$sp^{1.47}d^{0.01}$	0.7611	40.36	59.13	0.49
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6561	0.00	99.85	0.10
			N	$sp^{1.00}$	0.7547	0.00	99.69	0.29
C_6H_5	9.9	$\sigma_{\text{C2-N}}$	C2	$sp^{1.82}$	0.6467	35.45	64.43	0.07
			N	$sp^{1.48}d^{0.01}$	0.7627	40.19	59.30	0.50
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6478	0.00	99.85	0.11
			N	$sp^{1.00}$	0.7618	0.00	99.68	0.30
CH_3CO	4.75	$\sigma_{\text{C2-N}}$	C2	$sp^{1.81}$	0.6490	35.51	64.37	0.08
			N	$sp^{1.52}d^{0.01}$	0.7608	39.45	60.01	0.52
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6466	0.00	99.84	0.11
			N	$sp^{1.00}$	0.7628	0.00	99.65	0.33
$3\text{-NO}_2\text{C}_6\text{H}_4\text{CO}$	3.46	$\sigma_{\text{C2-N}}$	C2	$sp^{1.82}$	0.6487	35.44	64.44	0.08
			N	$sp^{1.52}d^{0.01}$	0.7611	39.44	60.01	0.53
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6433	0.00	99.84	0.12
			N	$sp^{1.00}$	0.7656	0.00	99.64	0.34
$3,5\text{-(NO}_2)_2\text{C}_6\text{H}_3\text{CO}$	2.85	$\sigma_{\text{C2-N}}$	C2	$sp^{1.83}$	0.6482	35.35	64.52	0.08
			N	$sp^{1.52}d^{0.01}$	0.7615	39.52	59.94	0.53
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6389	0.00	99.83	0.12
			N	$sp^{1.00}$	0.7693	0.00	99.64	0.34
CF_3CO	0.5	$\sigma_{\text{C2-N}}$	C2	$sp^{1.83}$	0.6481	35.31	64.57	0.08
			N	$sp^{1.50}d^{0.01}$	0.7616	39.71	59.75	0.53
		$\pi_{\text{C2-N}}$	C2	$sp^{1.00}$	0.6351	0.00	99.83	0.13
			N	$sp^{1.00}$	0.7724	0.00	99.64	0.34

h_A : natural atomic hybrids; C_A : polarization coefficient.

Table S13. NBO hybrid orbitals for the $\sigma_{\text{C1-O5}}$ bond.

R	pK_a	bond	atom	h_A	C_A	%s	%p	%d
H	16	σ C1-O5	C1	$sp^{3.52}d^{0.01}$	0.5918	22.10	77.77	0.12
			O5	$sp^{3.24}d^{0.01}$	0.8061	23.52	76.13	0.34
C ₆ H ₅	9.9	σ C1-O5	C1	$sp^{3.51}d^{0.01}$	0.5924	22.17	77.70	0.12
			O5	$sp^{3.23}d^{0.01}$	0.8056	23.56	76.09	0.35
CH ₃ CO	4.75	σ C1-O5	C1	$sp^{3.50}d^{0.01}$	0.5924	22.21	77.65	0.12
			O5	$sp^{3.21}d^{0.01}$	0.8056	23.65	76.00	0.35
3-NO ₂ C ₆ H ₄ CO	3.46	σ C1-O5	C1	$sp^{3.47}d^{0.01}$	0.5931	22.32	77.55	0.12
			O5	$sp^{3.21}d^{0.01}$	0.8051	23.67	75.97	0.35
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	σ C1-O5	C1	$sp^{3.46}d^{0.01}$	0.5936	22.39	77.48	0.12
			O5	$sp^{3.20}d^{0.01}$	0.8048	23.71	75.93	0.35
CF ₃ CO	0.5	σ C1-O5	C1	$sp^{3.46}d^{0.01}$	0.5938	22.40	77.47	0.12
			O5	$sp^{3.21}d^{0.01}$	0.8046	23.68	75.97	0.35

h_A : natural atomic hybrids; C_A : polarization coefficient.

Table S14. NBO hybrid orbitals for the $\sigma_{\text{C1-O6}}$ bond.

R	pK_a	bond	atom	h_A	C_A	%s	%p	%d
H	16	σ C1-O6	C1	$sp^{3.54}d^{0.01}$	0.5905	22.00	77.88	0.11
			O6	$sp^{3.11}d^{0.01}$	0.8071	24.26	75.42	0.30
C ₆ H ₅	9.9	σ C1-O6	C1	$sp^{3.53}d^{0.01}$	0.5911	22.07	77.81	0.11
			O6	$sp^{3.10}d^{0.01}$	0.8066	24.30	75.39	0.31
CH ₃ CO	4.75	σ C1-O6	C1	$sp^{3.52}d^{0.01}$	0.5911	22.10	77.78	0.11
			O6	$sp^{3.10}d^{0.01}$	0.8066	24.32	75.36	0.31
3-NO ₂ C ₆ H ₄ CO	3.46	σ C1-O6	C1	$sp^{3.51}d^{0.01}$	0.5917	22.16	77.72	0.11
			O6	$sp^{3.10}d^{0.01}$	0.8062	24.32	75.36	0.31
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	σ C1-O6	C1	$sp^{3.49}d^{0.01}$	0.5924	22.27	77.61	0.11
			O6	$sp^{3.09}d^{0.01}$	0.8057	24.39	75.30	0.31
CF ₃ CO	0.5	σ C1-O6	C1	$sp^{3.47}d^{0.01}$	0.5927	22.36	77.52	0.11
			O6	$sp^{3.08}d^{0.01}$	0.8055	24.44	75.24	0.31

h_A : natural atomic hybrids; C_A : polarization coefficient.

Table S15. NBO hybrid orbitals for the $\sigma_{\text{N-O}}$ bonds.

R	pK_a	bond	atom	h_A	C_A	%s	%p	%d
H	16	σ N-O	N	$sp^{5.75}d^{0.01}$	0.6392	14.77	84.99	0.22
			O	$sp^{4.38}d^{0.01}$	0.7690	18.54	81.20	0.25
C_6H_5	9.9	σ N-O	N	$sp^{5.85}d^{0.01}$	0.6310	14.58	85.21	0.20
			O	$sp^{4.28}d^{0.01}$	0.7758	18.90	80.92	0.16
CH_3CO	4.75	σ N-O	N	$sp^{6.48}d^{0.02}$	0.6226	13.35	86.43	0.21
			O	$sp^{4.33}d^{0.01}$	0.7826	18.72	81.11	0.16
3- $\text{NO}_2\text{C}_6\text{H}_4\text{CO}$	3.46	σ N-O	N	$sp^{6.72}d^{0.02}$	0.6167	12.92	86.85	0.21
			O	$sp^{4.25}d^{0.01}$	0.7872	19.02	80.82	0.14
3,5- $(\text{NO}_2)_2\text{C}_6\text{H}_3\text{CO}$	2.85	σ N-O	N	$sp^{7.04}d^{0.02}$	0.6122	12.41	87.37	0.21
			O	$sp^{4.26}d^{0.01}$	0.7907	18.97	80.88	0.14
CF_3CO	0.5	σ N-O	N	$sp^{7.56}d^{0.02}$	0.6093	11.65	88.12	0.22
			O	$sp^{4.35}d^{0.01}$	0.7930	18.66	81.18	0.14

h_A : natural atomic hybrids; C_A : polarization coefficient.

Table S16. NBO hybrid orbitals for the σ_{C2-C3} bond.

R	p <i>K</i> _a	bond	atom	<i>h</i> _A	<i>C</i> _A	%s	%p	%d
H	16	σ_{C2-C3}	C2	<i>sp</i> ^{1.83}	0.7050	35.25	64.64	0.05
			C3	<i>sp</i> ^{2.84} <i>d</i> ^{0.01}	0.7092	25.99	73.84	0.15
C ₆ H ₅	9.9	σ_{C2-C3}	C2	<i>sp</i> ^{1.83}	0.7059	35.30	64.60	0.05
			C3	<i>sp</i> ^{2.84} <i>d</i> ^{0.01}	0.7083	26.00	73.84	0.14
CH ₃ CO	4.75	σ_{C2-C3}	C2	<i>sp</i> ^{1.82}	0.7059	35.44	64.46	0.05
			C3	<i>sp</i> ^{2.85} <i>d</i> ^{0.01}	0.7083	25.96	73.87	0.14
3-NO ₂ C ₆ H ₄ CO	3.46	σ_{C2-C3}	C2	<i>sp</i> ^{1.81}	0.7062	35.49	64.41	0.05
			C3	<i>sp</i> ^{2.85} <i>d</i> ^{0.01}	0.7080	25.93	73.91	0.14
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	σ_{C2-C3}	C2	<i>Sp</i> ^{1.80}	0.7074	35.65	64.25	0.05
			C3	<i>Sp</i> ^{2.86} <i>d</i> ^{0.01}	0.7068	25.86	73.97	0.14
CF ₃ CO	0.5	σ_{C2-C3}	C2	<i>Sp</i> ^{1.78}	0.7088	35.88	64.01	0.05
			C3	<i>sp</i> ^{2.88} <i>d</i> ^{0.01}	0.7054	25.75	74.08	0.15

*h*_A: natural atomic hybrids; *C*_A: polarization coefficient.

Table S17. NBO hybrid orbitals for the N lone pair.

R	pK _a	bond	atom	h_A	%s	%p	%d
H	16	nN	LPN	$sp^{1.22}$	45.03	54.84	0.13
C ₆ H ₅	9.9	nN	LPN	$sp^{1.20}$	45.33	54.55	0.12
CH ₃ CO	4.75	nN	LPN	$sp^{1.11}$	47.30	52.55	0.15
3-NO ₂ C ₆ H ₄ CO	3.46	nN	LPN	$sp^{1.09}$	47.71	2.14	0.15
3,5-(NO ₂) ₂ C ₆ H ₃ CO	2.85	nN	LPN	$sp^{1.07}$	48.16	51.70	0.14
CF ₃ CO	0.5	nN	LPN	$sp^{1.05}$	48.72	51.13	0.14

h_A : natural atomic hybrids; C_A : polarization coefficient.

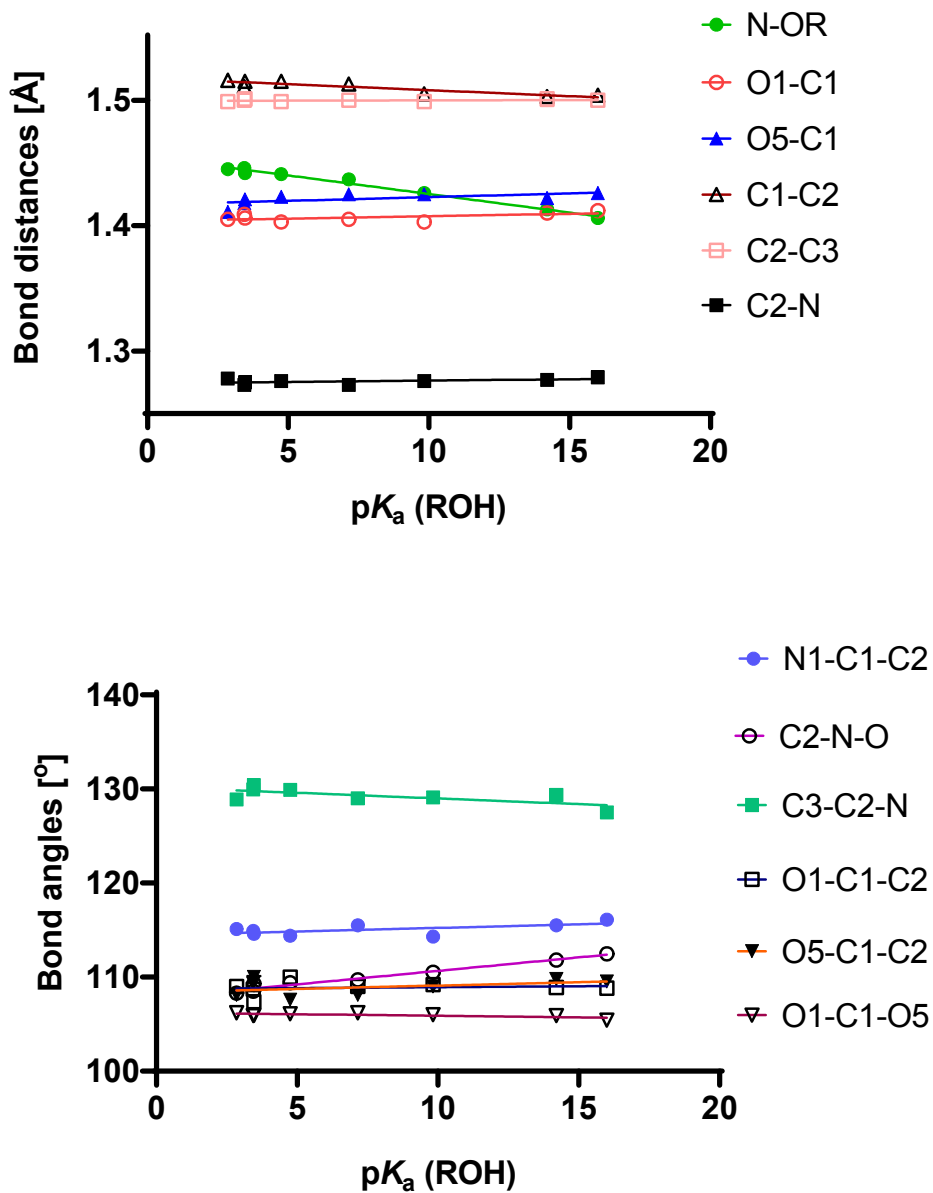


Figure S1. Plots showing variations in a range of geometric parameters for the series of oximes 1-8. (a) Bond distances and (b) angles, as a function of leaving group ability.

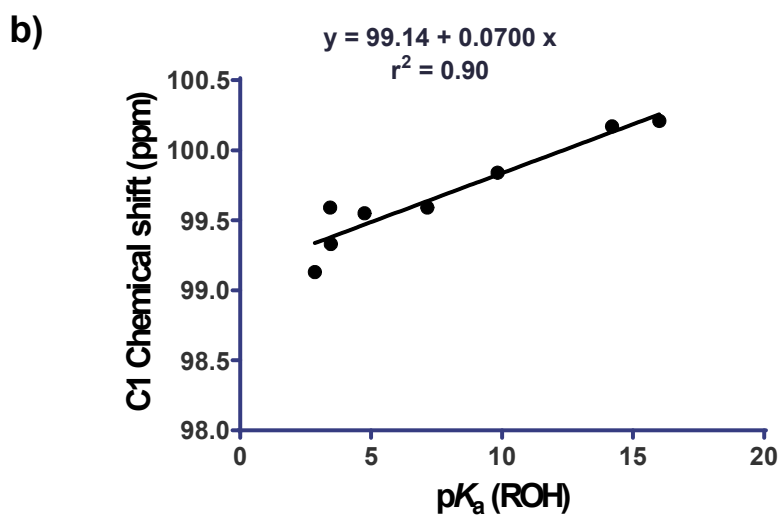
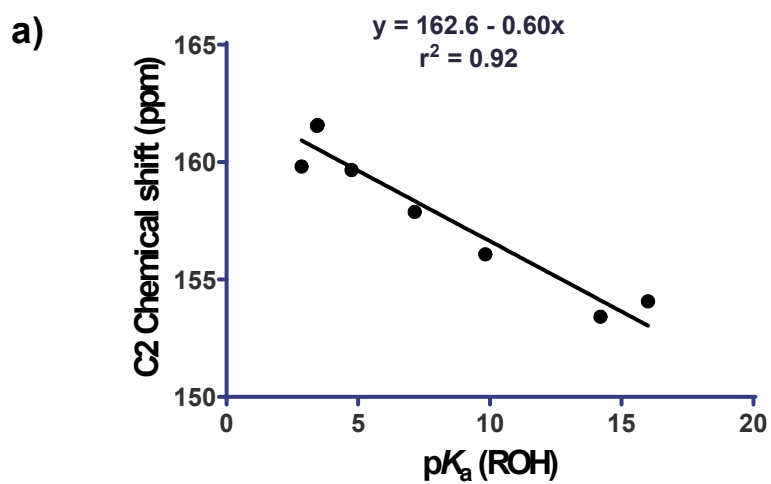


Figure S2. Plots showing variations in (a) C2 chemical shift and (b) C1 chemical shift, as a function of leaving group ability.

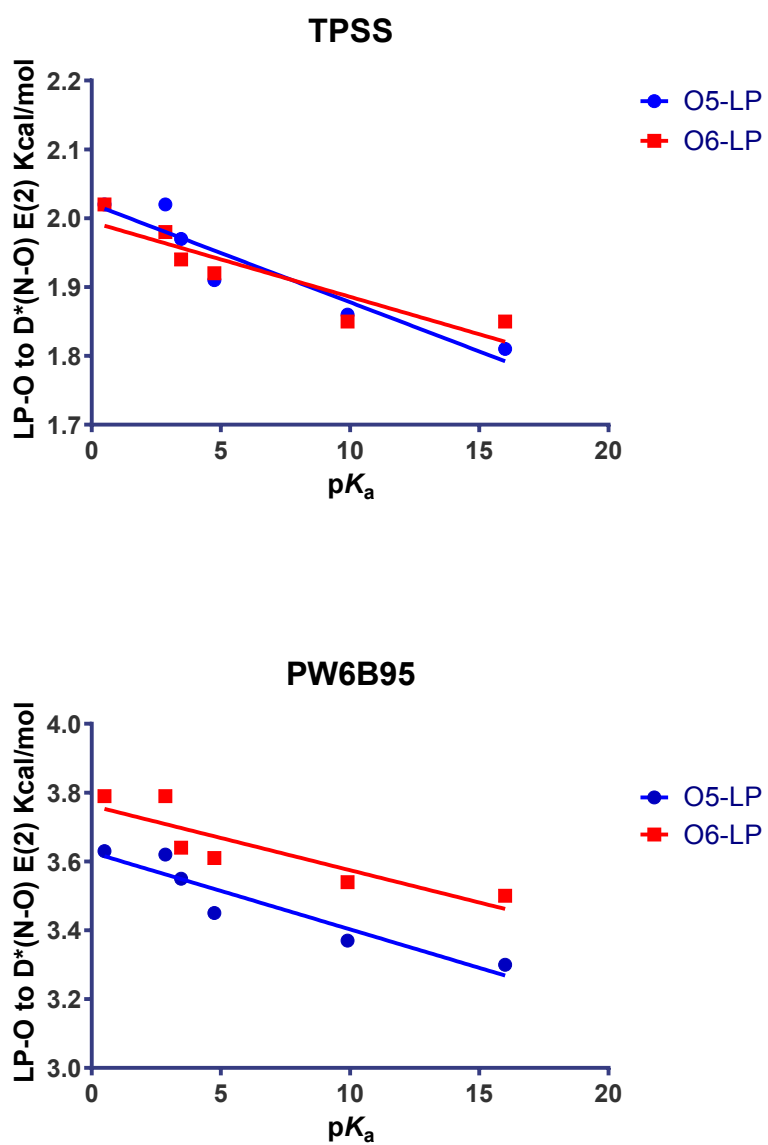


Figure S3. Plots showing comparison of $n_{O5} \rightarrow \sigma^*_{C1-C2}$ and $n_{O6} \rightarrow \sigma^*_{C1-C2}$ interactions at the TPSS and PW6B95 levels of theory, as a function of leaving group ability.

Optimized Geometries of Cyrene Oximes

All geometries were optimized at the TPSS-D3(BJ)/def2-TZVP level of theory and are presented in Cartesian coordinates (Å).

(Z)-Dihydrolevoglucosenone oxime

C	1.530162	5.679199	3.446329
H	2.235403	5.784281	2.744992
C	1.836637	6.527974	4.648167
C	0.856054	6.432445	5.779331
H	0.099521	7.050040	5.614362
H	1.296330	6.704544	6.622884
C	0.325699	4.993938	5.915761
H	0.956936	4.462796	6.461867
H	-0.546092	5.008904	6.384734
C	0.155917	4.334469	4.560138
H	-0.202445	3.405609	4.655977
C	-0.703454	5.155700	3.602139
H	-1.172141	4.569470	2.956145
H	-1.373735	5.690344	4.096758
N	2.899049	7.236322	4.587907
O	1.429960	4.331126	3.867768
O	0.244040	6.011003	2.931270
O	3.095625	8.016953	5.738836
H	3.887288	8.298286	5.640394

Total Energy : -514.60360283 Eh

(Z)-Dihydrolevoglucosenone O-(phenyl)oxime

C	1.004174	3.706666	14.868272
H	0.368505	4.381750	14.493376
C	2.425367	4.054490	14.514096
C	3.471778	3.037281	14.857789
H	3.855530	3.253708	15.744479
H	4.202416	3.085884	14.191233
C	2.910797	1.603347	14.887453
H	3.004171	1.197331	13.989343
H	3.437893	1.059206	15.524524
C	1.447962	1.580407	15.298503
H	1.090987	0.646446	15.287953
C	1.159441	2.256683	16.638262
H	0.377767	1.842544	17.082789
H	1.941929	2.196383	17.241455
C	4.084515	6.589935	12.904912
C	5.307014	7.232976	13.013529
H	5.956544	6.933690	13.638736
C	5.566386	8.323138	12.191653
H	6.398482	8.777044	12.257559
C	4.617536	8.753561	11.275879
H	4.799709	9.498704	10.715888
C	3.402771	8.093174	11.179529
H	2.755346	8.389517	10.550911
C	3.121983	6.998958	11.995607
H	2.290148	6.544371	11.930057
N	2.565166	5.202205	13.974802

O 0.709228 2.422691 14.386521

O 0.884007 3.630389 16.286093

O 3.941979 5.463539 13.708723

Total Energy : -745.71680526 Eh

(Z)-Dihydrolevoglucosenone O-(acetyl)oxime

C	3.097172	8.963653	4.490746
H	3.514514	8.831023	3.591853
C	3.402240	7.796792	5.407153
C	2.752979	7.846332	6.757889
H	3.337597	8.340324	7.386244
H	2.642406	6.924663	7.103474
C	1.377340	8.535506	6.690671
H	0.690088	7.870005	6.437157
H	1.144287	8.889029	7.585845
C	1.368268	9.672267	5.685715
H	0.478582	10.129104	5.666146
C	2.496677	10.676868	5.898623
H	2.235839	11.576118	5.576130
H	2.741638	10.737383	6.855929
C	5.520701	5.103902	5.527599
C	5.865462	4.178471	6.648598
H	5.757724	3.251472	6.351054
H	5.270637	4.348883	7.407769
H	6.794840	4.328503	6.921485
N	4.225582	6.948806	4.926054
O	1.709626	9.137514	4.380476
O	3.593559	10.142900	5.114697
O	4.519819	5.949065	5.921617
O	6.033306	5.127418	4.448086

Total Energy : -667.33248082 Eh

(Z)-Dihydrolevoglucosenone O-(3-nitrobenzoyl)oxime

C	5.182307	13.247564	8.859416
H	5.530629	13.568219	9.841088
C	5.100345	11.740544	8.724775
C	4.651111	11.246734	7.373377
H	3.553126	11.276868	7.368104
H	4.946325	10.206349	7.229857
C	5.203332	12.146394	6.242254
H	6.232712	11.855822	6.008126
H	4.604165	12.013359	5.333939
C	5.215285	13.616120	6.661638
H	5.653718	14.257507	5.894455
C	3.861140	14.123450	7.177771
H	3.758869	15.208085	7.070775
H	3.008299	13.622867	6.706227
C	5.736848	8.855314	10.505592
C	5.410060	7.428824	10.191640
C	5.965128	6.448232	11.018760
H	6.601514	6.725177	11.850779
C	5.683175	5.116905	10.747578
C	4.867972	4.721115	9.690188
H	4.682920	3.666264	9.526101
C	4.314881	5.706864	8.877199
H	3.672369	5.424160	8.049173
C	4.583229	7.052711	9.122741
H	4.147582	7.817438	8.490436
N	5.399559	11.076851	9.777176

N	6.275072	4.068915	11.618004
O	6.041279	13.748981	7.848291
O	3.885064	13.798631	8.587664
O	5.224093	9.662941	9.512795
O	6.362944	9.229877	11.464546
O	6.988011	4.439394	12.551897
O	6.013416	2.893409	11.347164
Total Energy	:		-1064.99930653 Eh

(Z)-Dihydrolevoglucosenone O-(3,5-dinitrobenzoyl)oxime

C	5.297132	2.813160	7.687402
H	5.868419	2.365627	8.500629
C	3.963867	3.376720	8.141265
C	3.121459	3.977912	7.045847
H	3.471029	5.009397	6.904054
H	2.075640	4.032874	7.350133
C	3.276407	3.186657	5.725522
H	2.620569	2.309977	5.740626
H	2.974420	3.814329	4.879341
C	4.711606	2.696053	5.539757
H	4.822425	2.084157	4.642464
C	5.765959	3.805697	5.654526
H	6.690839	3.552319	5.127023
H	5.401123	4.781294	5.315971
C	2.032857	3.736253	10.976759
C	0.726612	4.451756	11.164990
C	0.159438	5.254189	10.169243
H	0.645254	5.400927	9.214155
C	-1.055155	5.880283	10.421738
C	-1.736051	5.743772	11.624022
H	-2.682489	6.238291	11.801341
C	-1.142516	4.939547	12.589094
C	0.071196	4.293560	12.390341
H	0.504641	3.675638	13.168158
N	3.734388	3.293386	9.397431
N	-1.653540	6.729868	9.356981

N	-1.846298	4.764927	13.889827
O	5.059489	1.853938	6.671695
O	6.046884	3.869385	7.073469
O	2.443985	3.905539	9.680555
O	2.600416	3.111424	11.836305
O	-1.037736	6.815671	8.292129
O	-2.721774	7.285860	9.611423
O	-2.919691	5.352383	14.031061
O	-1.305681	4.045691	14.729602
Total Energy	:		-1269.81293791 Eh

(Z)-Dihydrolevoglucosenone O-(trifluoroacetyl)oxime

C	3.137552	9.007999	4.501389
H	3.578735	8.948622	3.506633
C	3.516250	7.832886	5.384364
C	2.912945	7.849123	6.764219
H	3.553058	8.495013	7.380109
H	2.938009	6.854199	7.209804
C	1.475146	8.419911	6.730432
H	0.772072	7.634614	6.434336
H	1.188946	8.764311	7.730791
C	1.357353	9.561979	5.721447
H	0.334749	9.935787	5.642474
C	2.398158	10.673448	5.918266
H	2.064623	11.629860	5.503759
H	2.693265	10.805004	6.964709
C	5.444853	4.995340	5.316115
C	5.647896	3.891018	6.395859
N	4.313913	6.993553	4.843021
O	1.724528	9.057200	4.409015
O	3.540821	10.215935	5.155669
O	4.587615	5.920261	5.805258
O	5.988301	4.948471	4.249738
F	5.073978	4.175465	7.585195
F	6.964125	3.699725	6.607047
F	5.119651	2.730046	5.950779

Total Energy : -966.27892763 Eh

Cyclohexanone oxime

O	5.664252	5.562310	2.337721
H	4.805896	5.147920	2.157509
N	5.339667	6.948099	2.158414
C	6.359034	7.705499	2.339808
C	6.136555	9.183298	2.156674
H	6.300241	9.685881	3.121878
H	5.100555	9.361985	1.856013
C	7.134761	9.752825	1.128244
H	6.900422	9.340375	0.137995
H	7.015381	10.840412	1.062168
C	8.577833	9.390046	1.503201
H	8.835332	9.868096	2.459437
H	9.273646	9.784794	0.753386
C	8.751108	7.871083	1.632923
H	9.777621	7.624863	1.928141
H	8.575208	7.400178	0.657039
C	7.765230	7.272403	2.660013
H	7.834998	6.184483	2.689958
H	8.020909	7.653946	3.659599

Total Energy : -365.77780667 Eh

Cyclohexanone *O*-(phenyl)oxime

O	5.123221	5.675012	2.915194
N	4.953359	7.079924	2.693643
C	6.088238	7.681306	2.676855
C	6.045331	9.165771	2.427935
H	6.425279	9.679774	3.323513
H	5.010524	9.481677	2.271643
C	6.941793	9.535222	1.227794
H	6.502133	9.118099	0.312409
H	6.962696	10.624558	1.109501
C	8.361705	8.983930	1.409960
H	8.829844	9.465671	2.280433
H	8.978551	9.234127	0.538834
C	8.342297	7.464687	1.620399
H	9.357484	7.084454	1.781121
H	7.954479	6.975564	0.717450
C	7.455621	7.068762	2.822059
H	7.387777	5.985961	2.925618
H	7.908113	7.468638	3.741382
C	3.925667	4.984638	2.945707
C	2.668678	5.572032	2.796562
C	4.054432	3.606625	3.145541
C	1.536749	4.757209	2.850371
C	2.913301	2.811945	3.195855
H	5.046076	3.178807	3.258418
H	2.590591	6.640281	2.643413
C	1.646743	3.381518	3.048846

H 0.556257 5.211414 2.735145

H 0.757805 2.759381 3.089051

H 3.016957 1.741514 3.350970

Total Energy : -597.21265974 Eh

Cyclohexanone *O*-(acetyl)oxime

O	5.323610	5.660763	2.793778
N	5.097212	7.078514	2.583283
C	6.208007	7.721085	2.612559
C	6.114154	9.207173	2.380294
H	6.443845	9.719992	3.295918
H	5.073978	9.486451	2.193064
C	7.036310	9.628756	1.216761
H	6.644879	9.208772	0.281053
H	7.016200	10.719589	1.114473
C	8.470341	9.131974	1.439946
H	8.891006	9.621960	2.329700
H	9.103706	9.416633	0.591719
C	8.504403	7.610843	1.633421
H	9.526989	7.268824	1.827203
H	8.169304	7.116505	0.712516
C	7.593698	7.164513	2.799455
H	7.568332	6.079209	2.893500
H	7.994075	7.577930	3.736698
C	4.164948	4.906766	2.809017
C	2.855956	5.632626	2.634225
H	2.729295	6.386004	3.416491
H	2.054796	4.894930	2.677354
H	2.838693	6.166091	1.680071
O	4.295903	3.716145	2.965882

Total Energy : -518.67816269 Eh

Cyclohexanone *O*-(3-nitrobenzoyl)oxime

O	5.227503	5.727545	2.986038
N	5.043523	7.138930	2.750105
C	6.175764	7.741499	2.698788
C	6.116500	9.221879	2.429619
H	6.518625	9.748225	3.307716
H	5.076317	9.530490	2.297877
C	6.978194	9.576239	1.199592
H	6.515164	9.143513	0.303354
H	6.988546	10.663509	1.065133
C	8.405591	9.035576	1.351141
H	8.895780	9.534166	2.199477
H	8.995748	9.273704	0.458716
C	8.401003	7.520024	1.586619
H	9.421626	7.147491	1.728000
H	7.992645	7.013260	0.702763
C	7.549119	7.138706	2.818354
H	7.492575	6.057480	2.942929
H	8.021358	7.558789	3.718212
C	4.036556	5.047192	3.054318
C	2.782313	5.643820	2.934574
C	4.155023	3.666440	3.261861
C	1.668522	4.815060	3.028844
C	3.013198	2.875871	3.349743
H	5.145722	3.231836	3.351379
H	2.671309	6.706031	2.775167
C	1.744335	3.441777	3.233626

N	0.327512	5.439643	2.903210
H	0.837228	2.854729	3.297336
O	0.275995	6.659735	2.724448
O	-0.657563	4.699271	2.984635
H	3.114758	1.806950	3.510115
Total Energy	:	-802.03376042 Eh	

Cyclohexanone *O*-(3,5-dinitrobenzoyl)oxime

O	5.228202	5.731803	2.995125
N	5.040137	7.149431	2.751042
C	6.174089	7.748644	2.699881
C	6.114651	9.227795	2.423553
H	6.516622	9.755932	3.300308
H	5.075035	9.537256	2.289699
C	6.978200	9.574824	1.192455
H	6.515459	9.138578	0.297867
H	6.988181	10.661372	1.053886
C	8.405021	9.034486	1.348191
H	8.895473	9.537980	2.193271
H	8.995003	9.266677	0.454315
C	8.399675	7.520557	1.592070
H	9.419237	7.147980	1.738662
H	7.993069	7.007983	0.710881
C	7.546376	7.146299	2.825417
H	7.491810	6.066027	2.957727
H	8.015710	7.573989	3.722942
C	4.043156	5.056631	3.064593
C	2.784600	5.652261	2.940182
C	4.159489	3.678323	3.277895
C	1.665094	4.831985	3.034450
C	2.999863	2.920853	3.361384
H	5.128877	3.205265	3.376070
H	2.677865	6.714487	2.776176
C	1.724714	3.461271	3.244027

N	0.323619	5.464611	2.902505
H	0.834796	2.851276	3.312065
N	3.132668	1.456919	3.586240
O	0.286027	6.683165	2.720061
O	-0.660093	4.727223	2.984678
O	4.273530	0.999583	3.688744
O	2.093947	0.798635	3.655257
Total Energy	:		-1006.84930733 Eh

Cyclohexanone *O*-(trifluoroacetyl)oxime

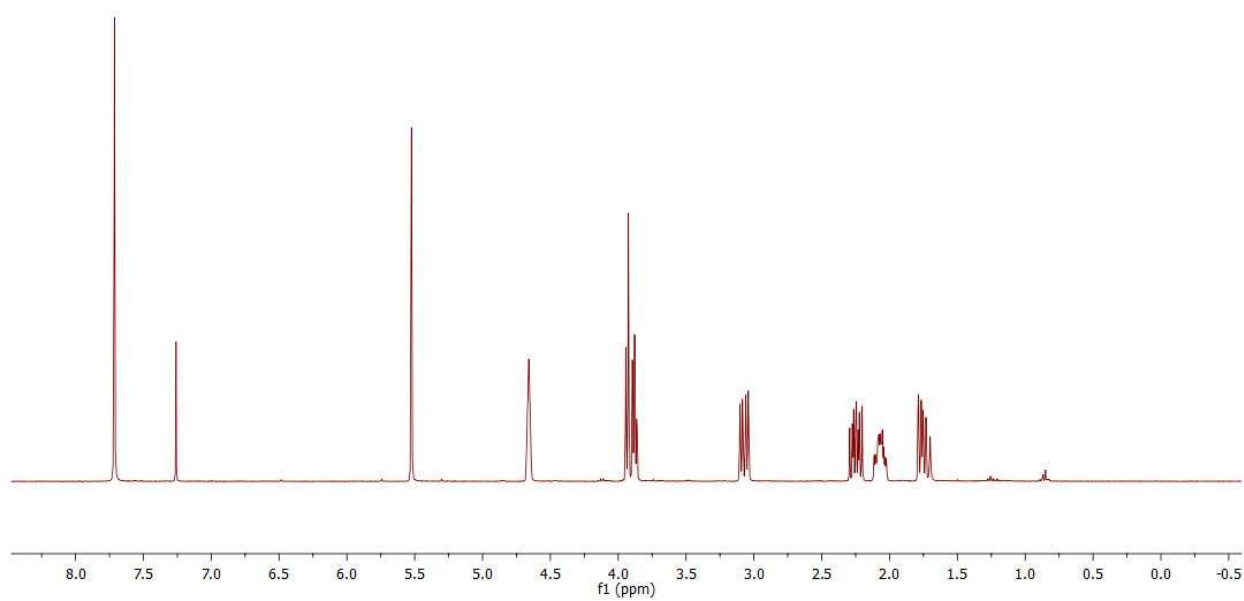
O	5.258111	5.763511	2.877651
N	5.053545	7.207864	2.641476
C	6.187814	7.803329	2.655613
C	6.140033	9.287736	2.395846
H	6.502811	9.800227	3.298451
H	5.108060	9.601156	2.219868
C	7.058598	9.650279	1.209011
H	6.629669	9.236188	0.287627
H	7.082078	10.739367	1.093028
C	8.474896	9.096509	1.410357
H	8.938233	9.584961	2.279256
H	9.098424	9.335594	0.541429
C	8.449476	7.580014	1.635586
H	9.459832	7.195731	1.811407
H	8.066229	7.079764	0.737032
C	7.553826	7.202508	2.836949
H	7.494770	6.122709	2.967698
H	7.985920	7.634223	3.750859
C	4.125287	5.031971	2.941565
C	2.755229	5.768083	2.800367
O	4.180014	3.841914	3.108446
F	1.768039	4.865474	2.931982
F	2.615072	6.349372	1.589264
F	2.578182	6.704014	3.757601

Total Energy : -816.83490193 Eh

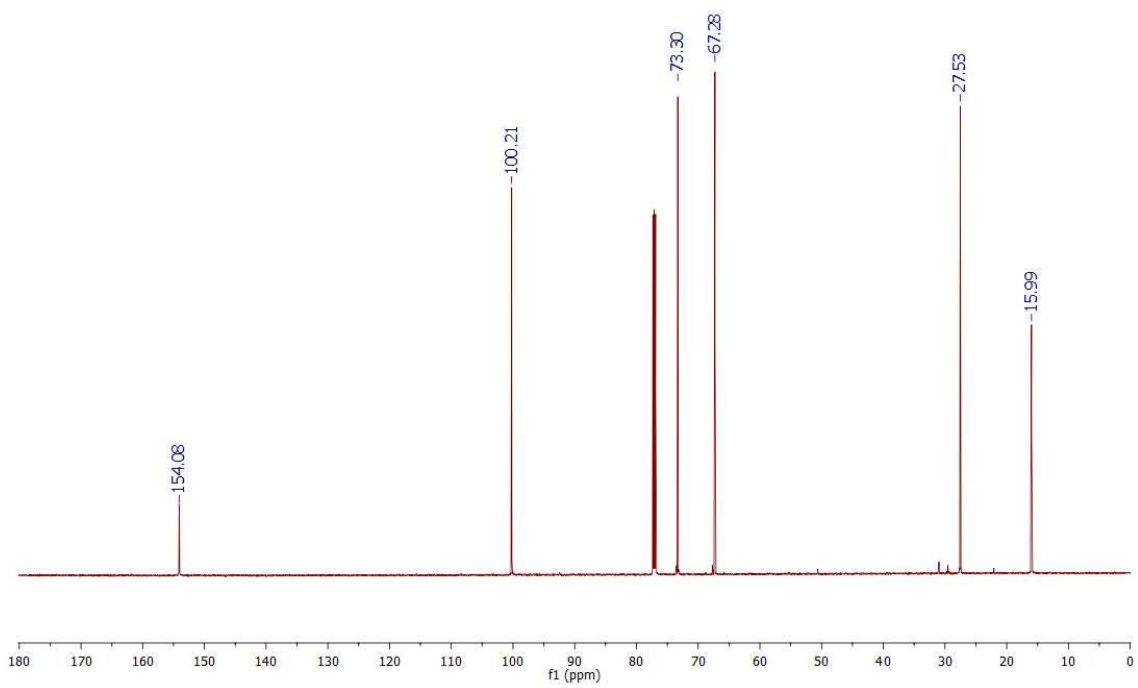
$^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ spectra for Oximes (1-8):

(*Z*)-Dihydrolevoglucosenone oxime (1).

$^1\text{H-NMR}$

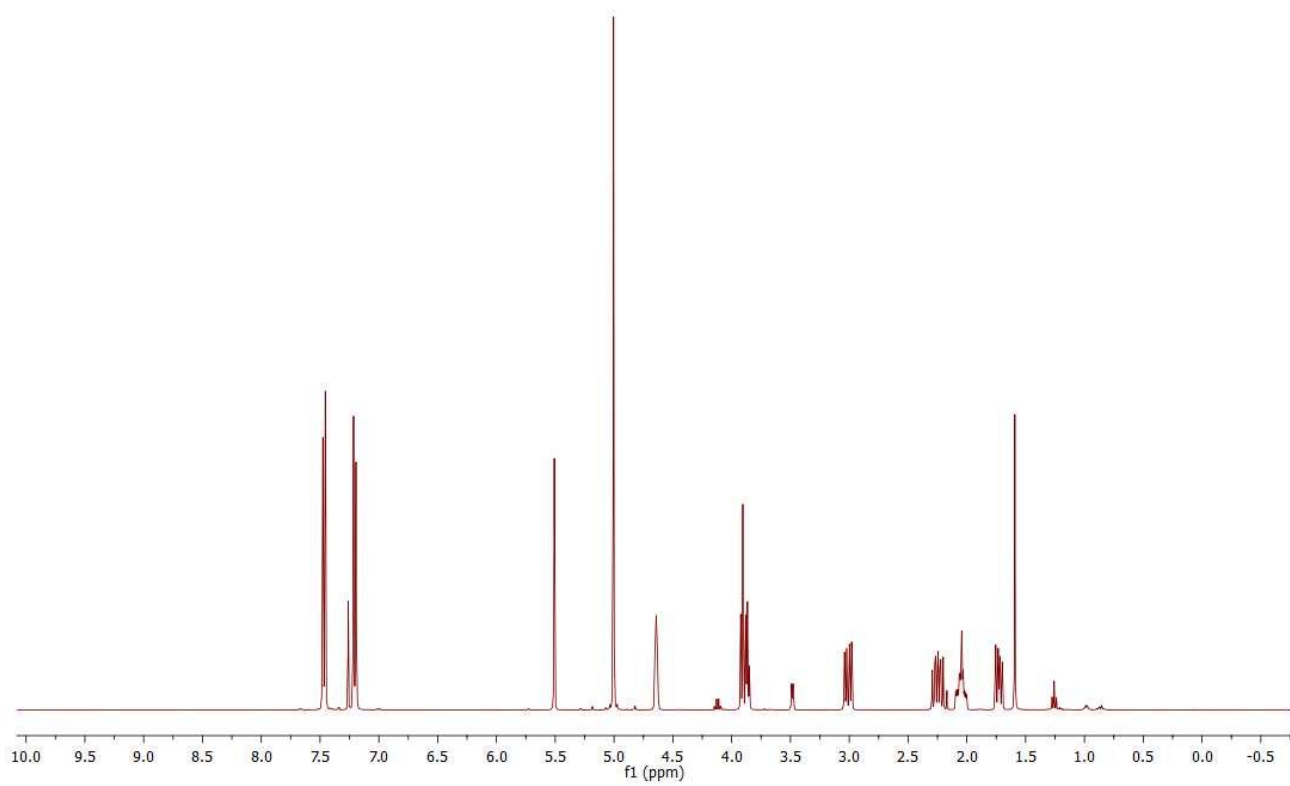


$^{13}\text{C-NMR}$

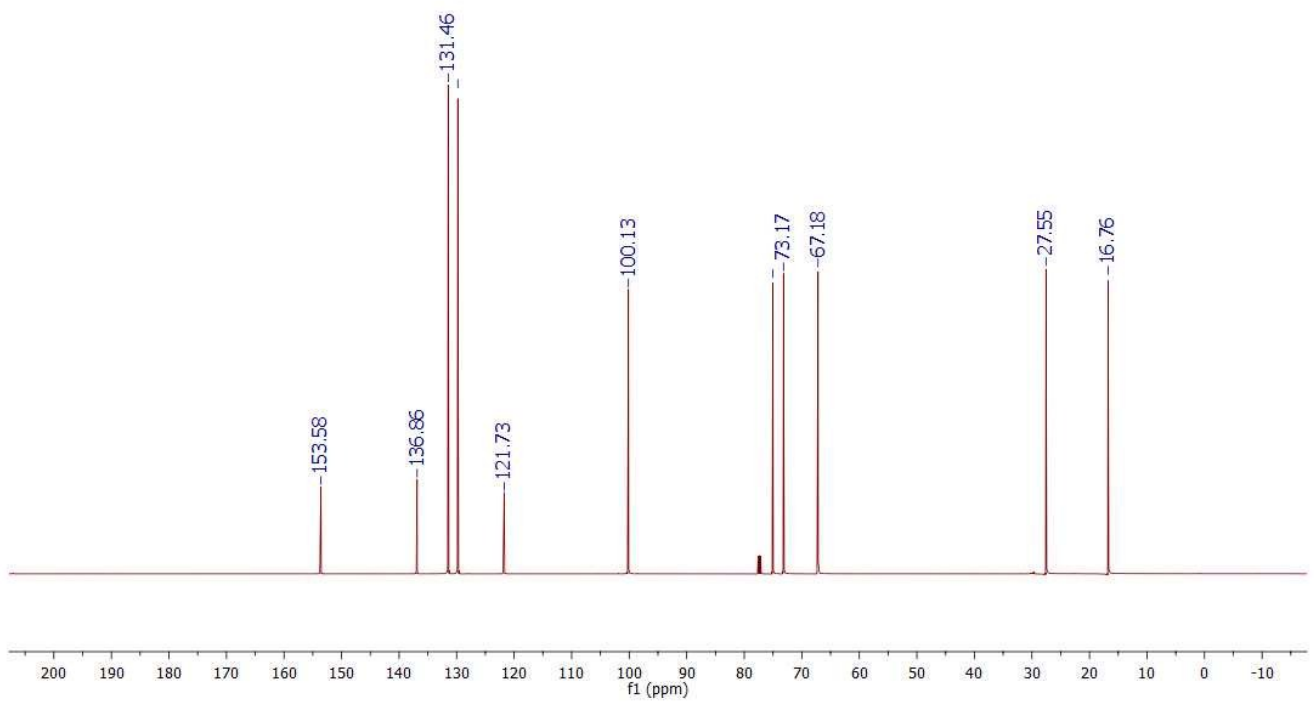


(Z)-Dihydrolevoglucosenone *O*-(4-bromobenzyl)oxime (2).

¹H-NMR

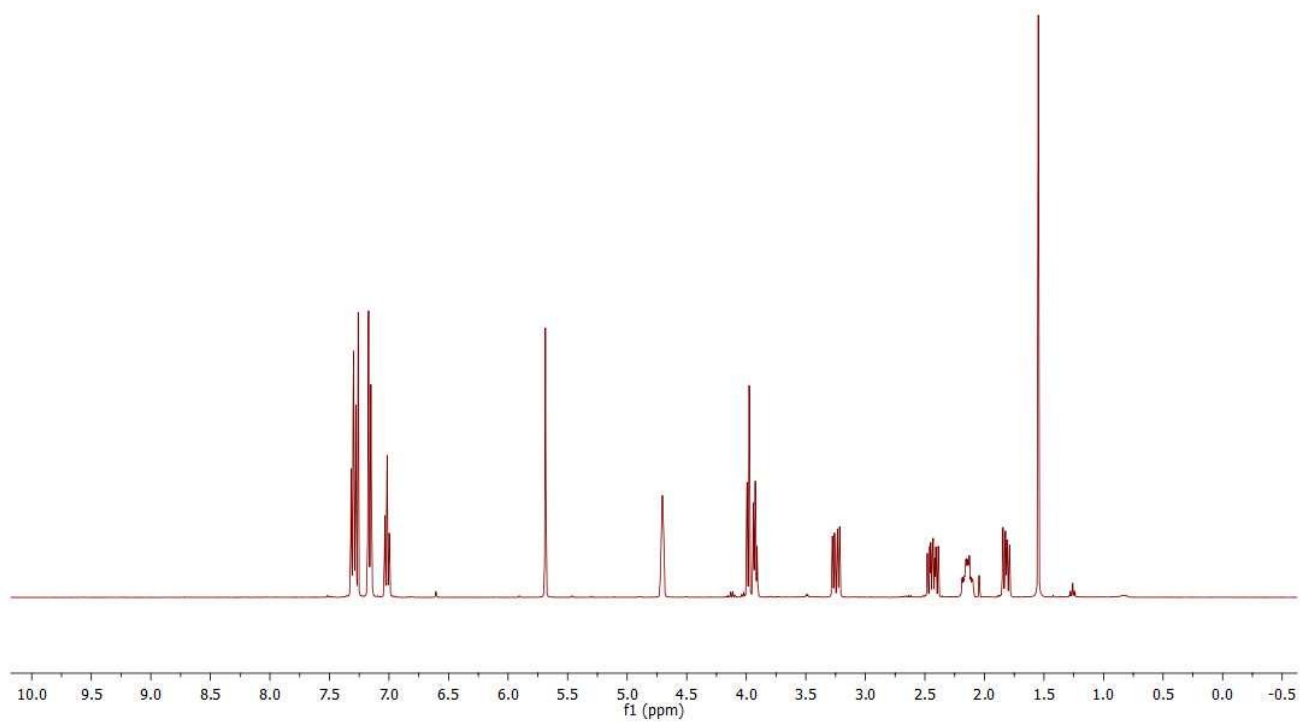


¹³C-NMR

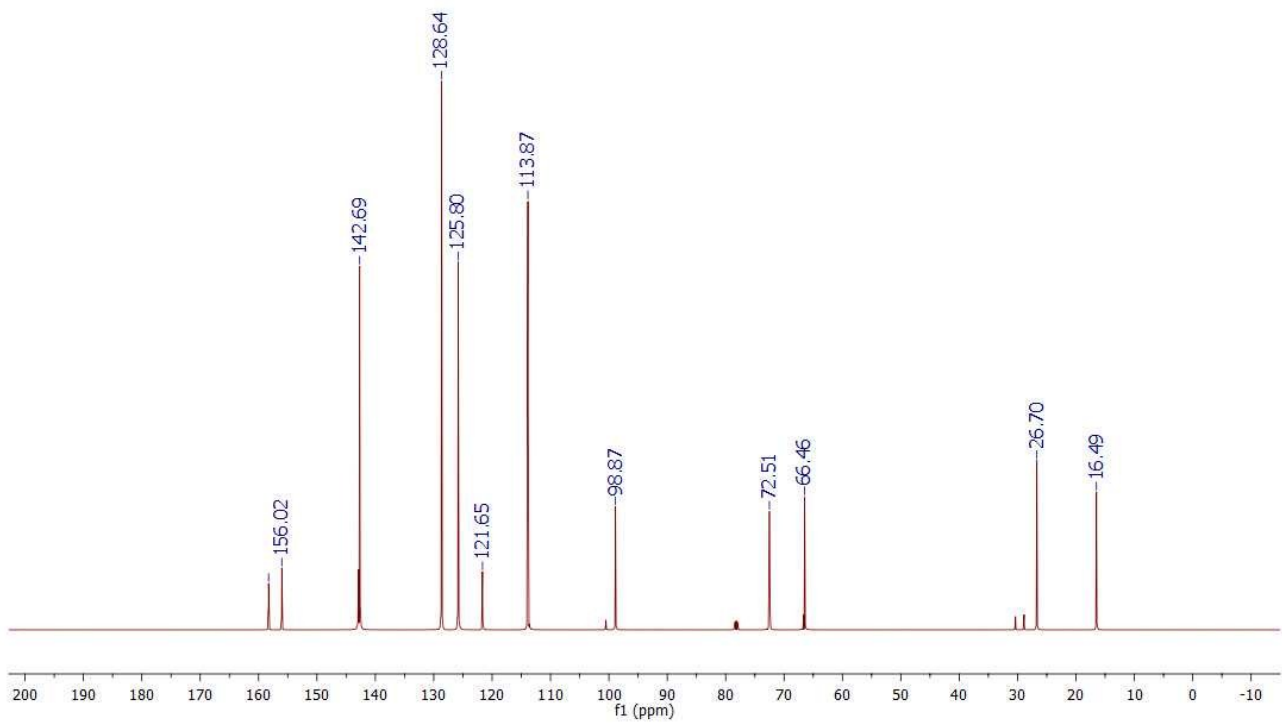


(Z)-Dihydrolevoglucosenone *O*-(phenyl)oxime (3).

$^1\text{H-NMR}$

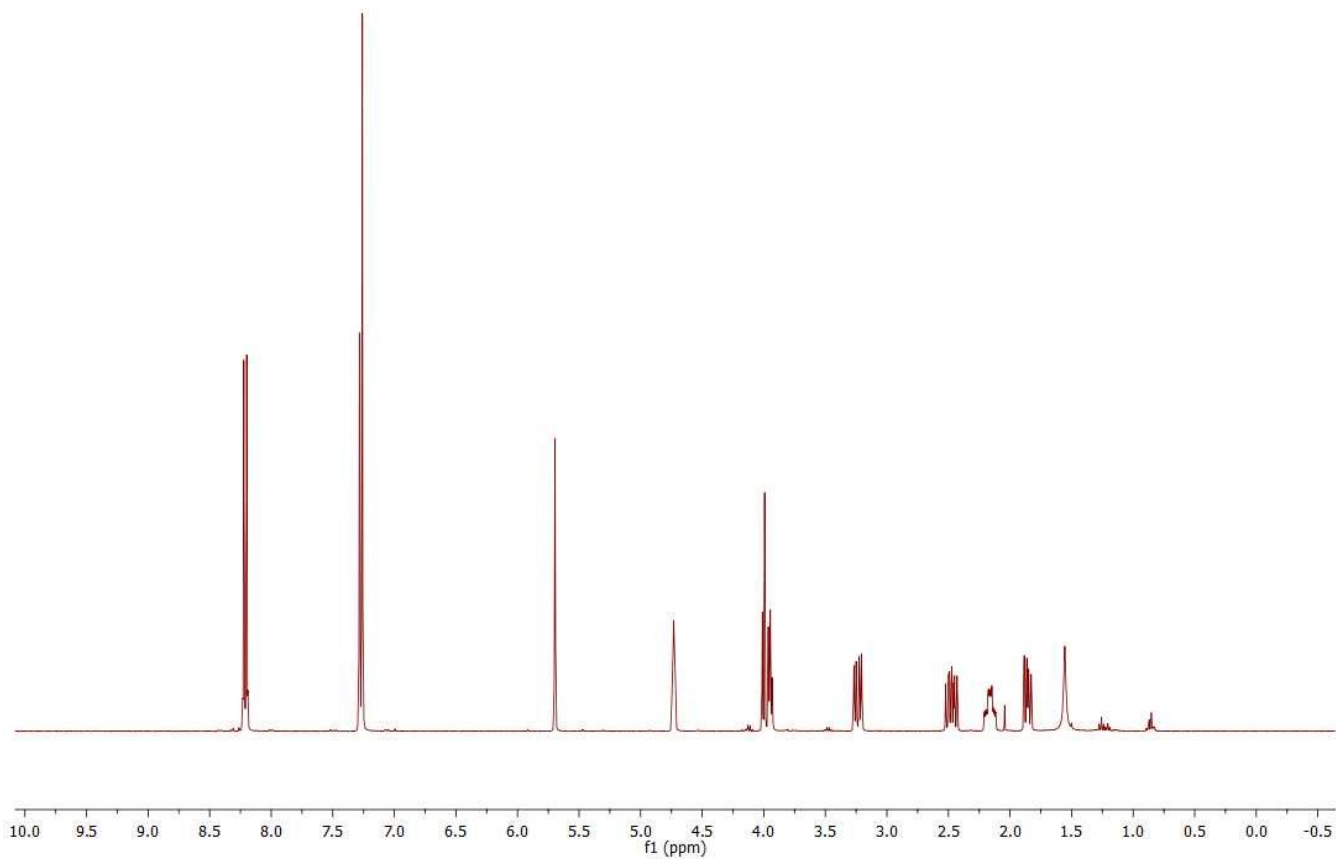


$^{13}\text{C-NMR}$

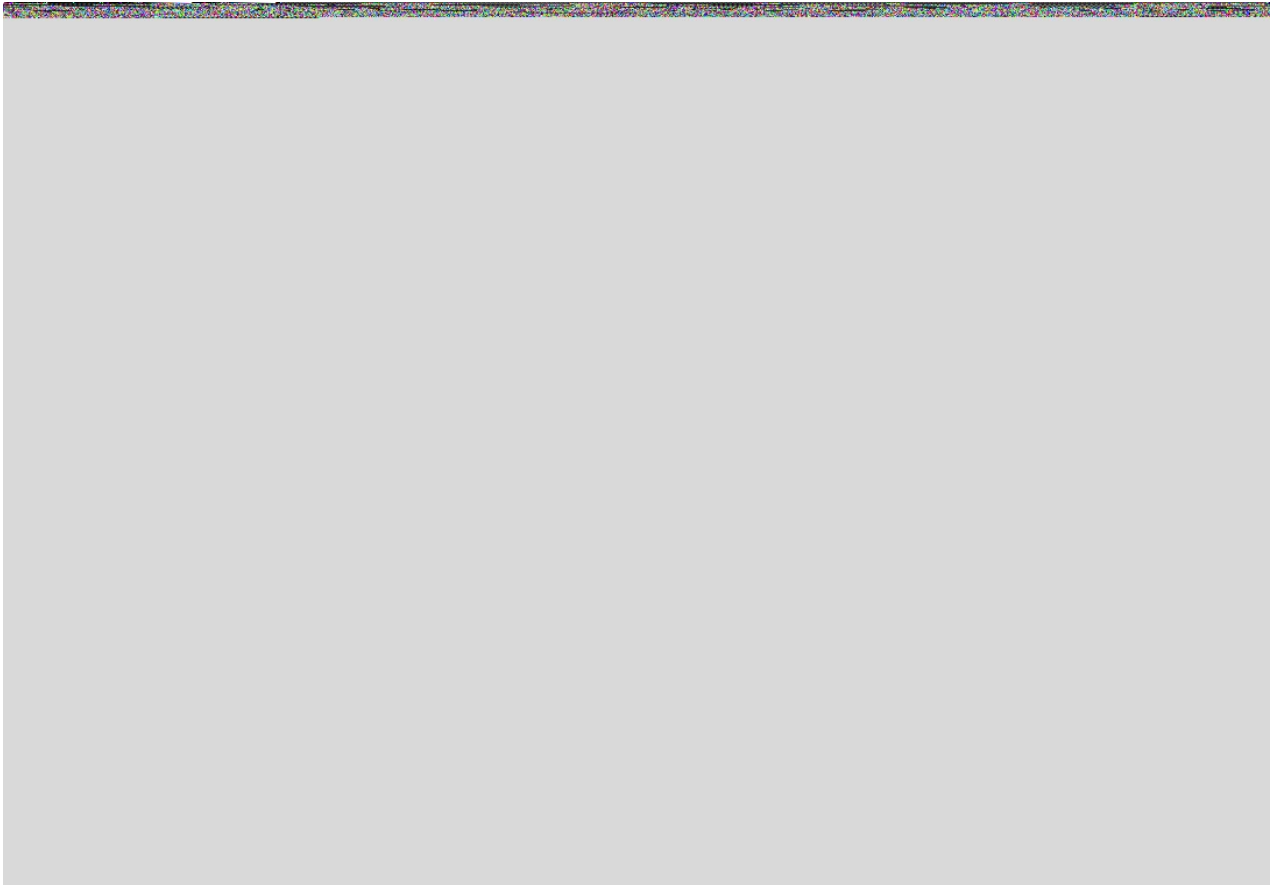


(Z)-Dihydrolevoglucosenone *O*-(4-nitrophenyl)oxime (4).

$^1\text{H-NMR}$

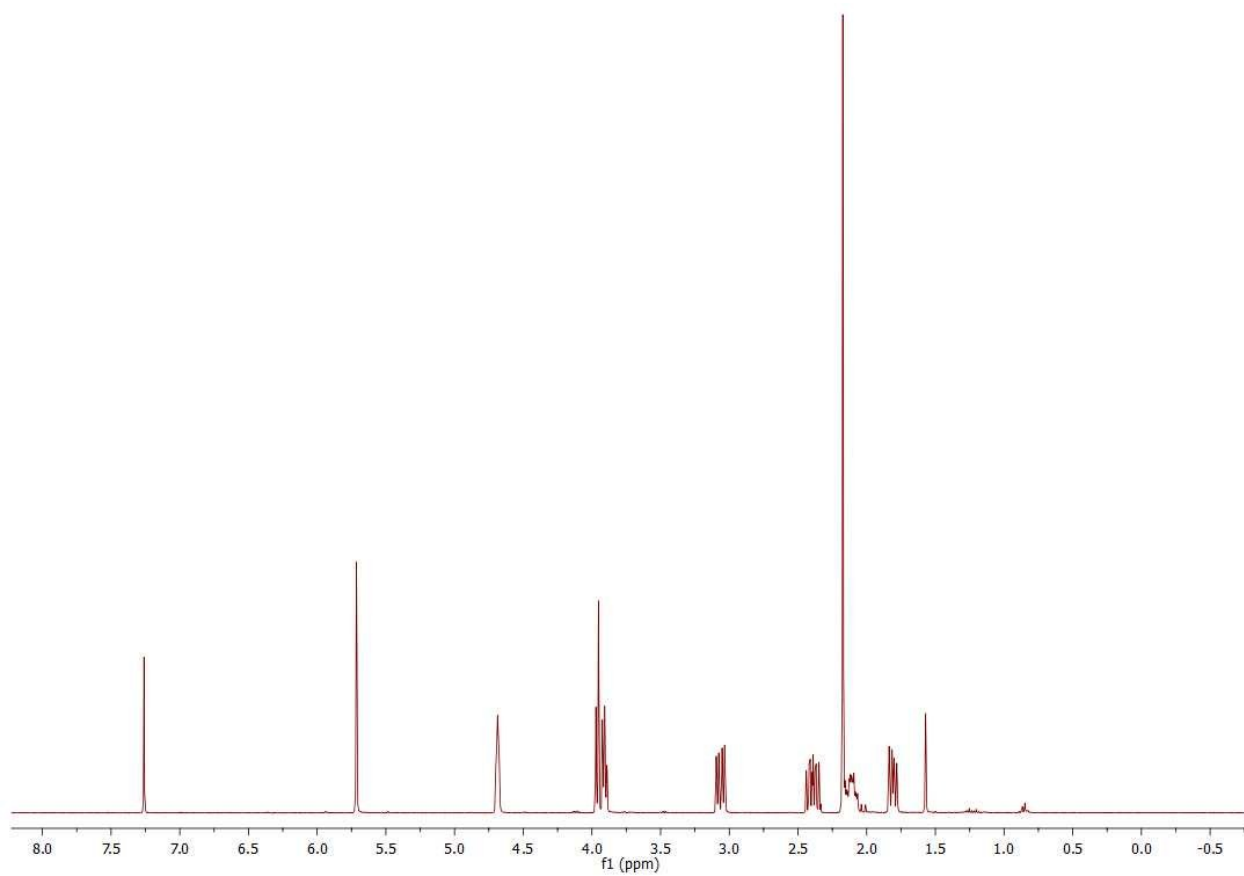


$^{13}\text{C-NMR}$

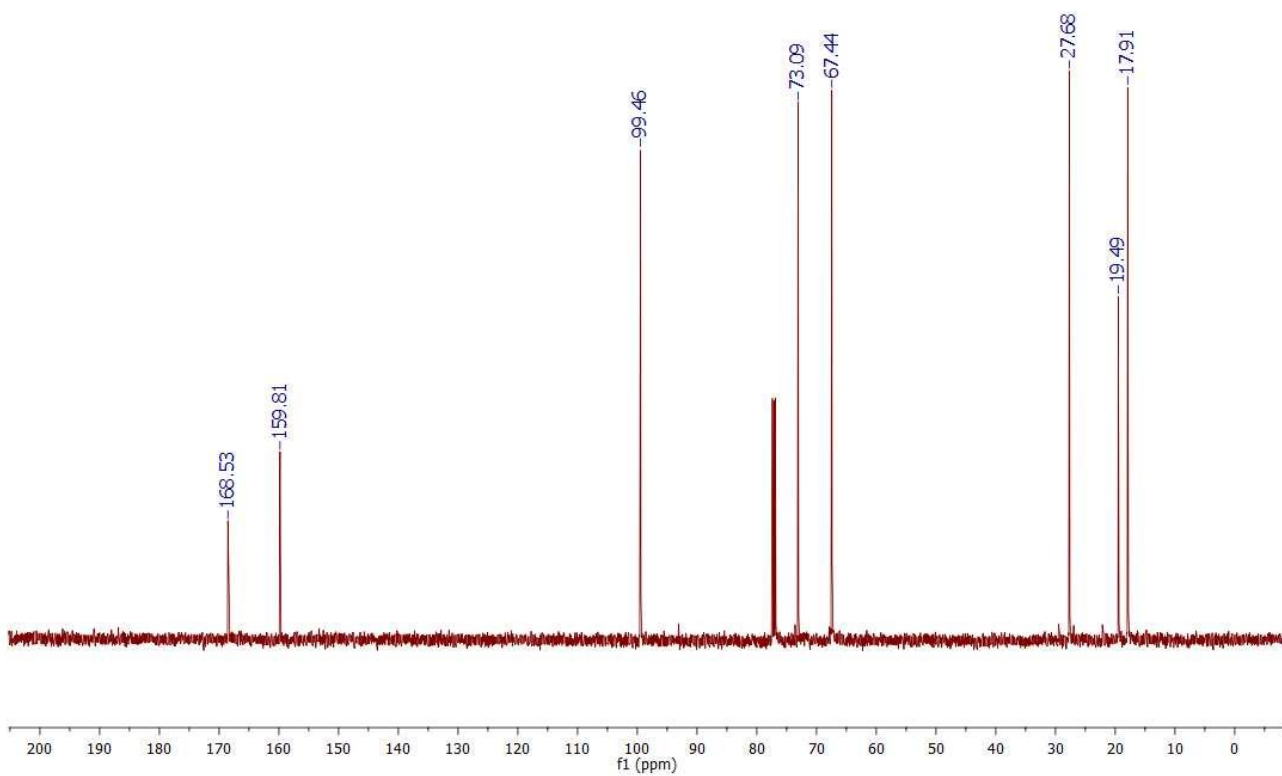


(Z)-Dihydrolevoglucosenone *O*-(acetyl)oxime (5).

$^1\text{H-NMR}$

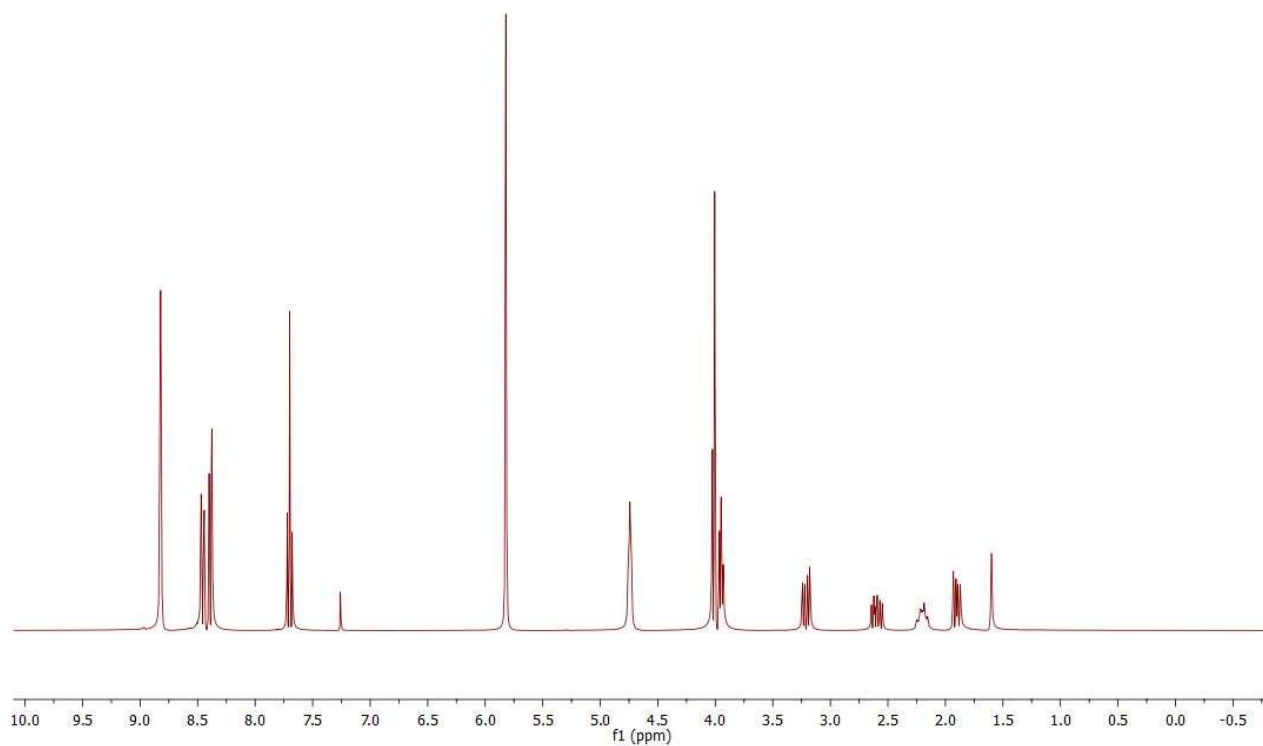


$^{13}\text{C-NMR}$

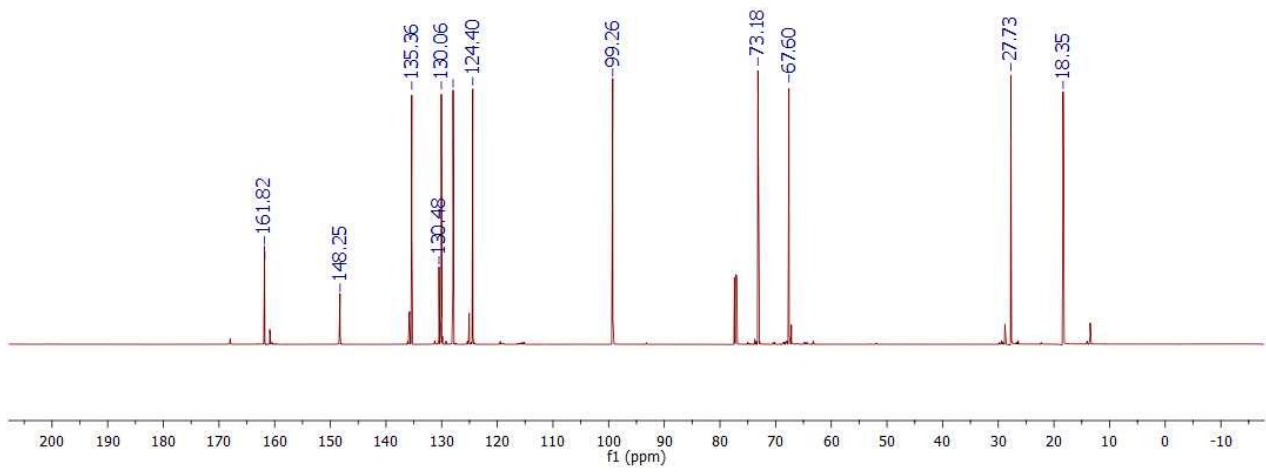


(Z)-Dihydrolevoglucosenone *O*-(3-nitrobenzoyl)oxime (6).

$^1\text{H-NMR}$

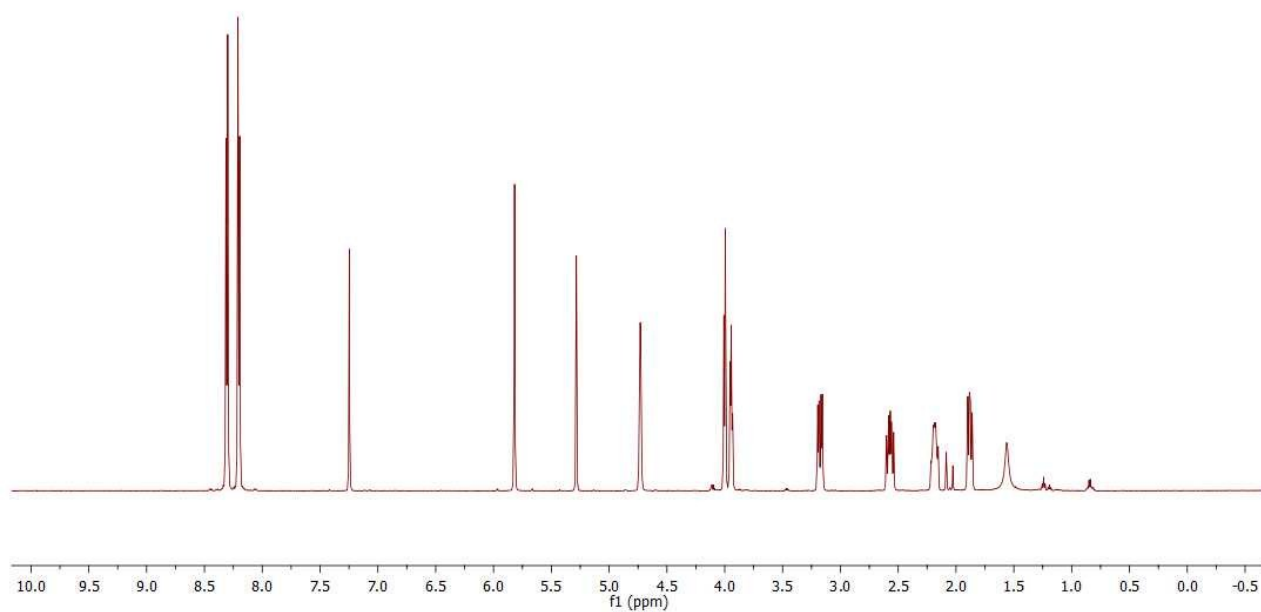


$^{13}\text{C-NMR}$

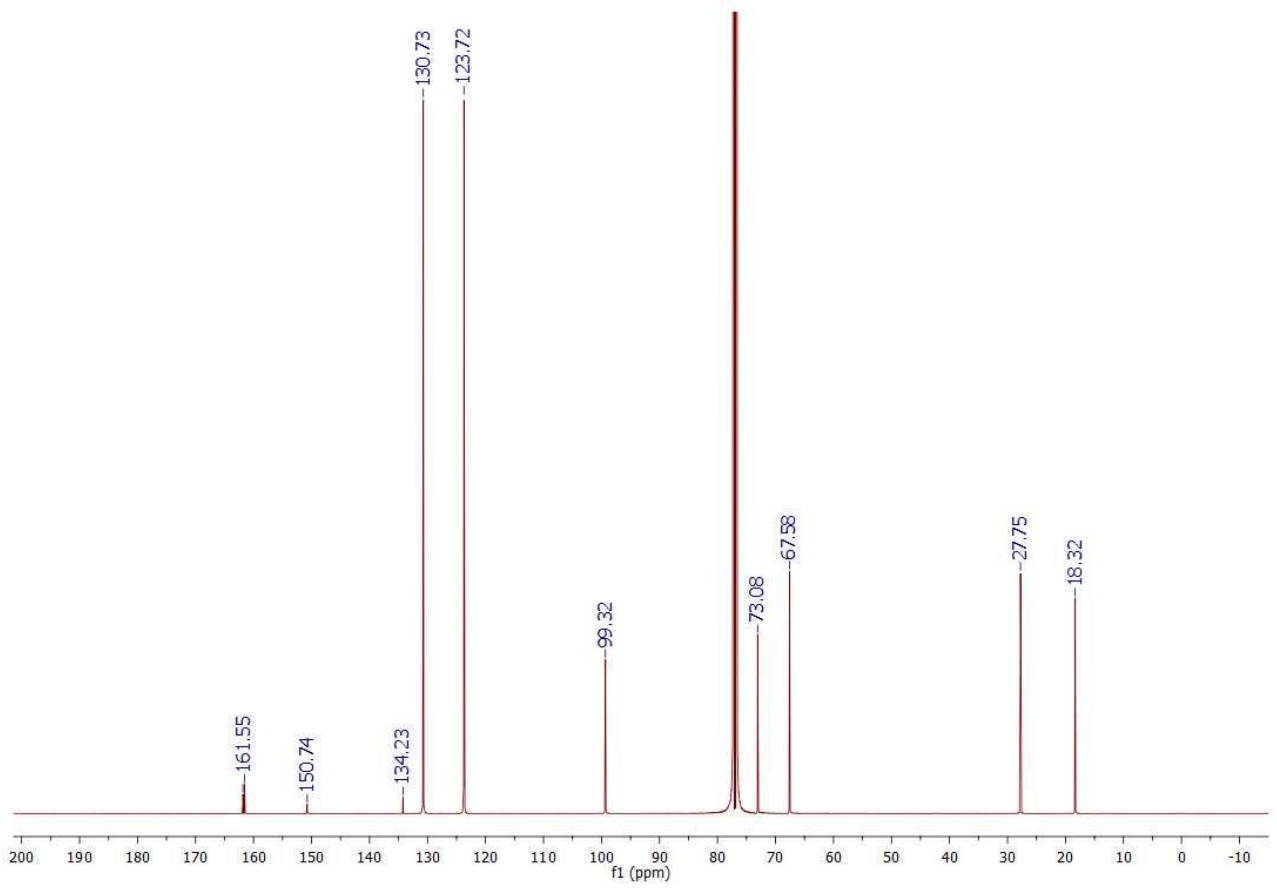


(Z)-Dihydrolevoglucosenone *O*-(4-nitrobenzoyl)oxime (7).

$^1\text{H-NMR}$

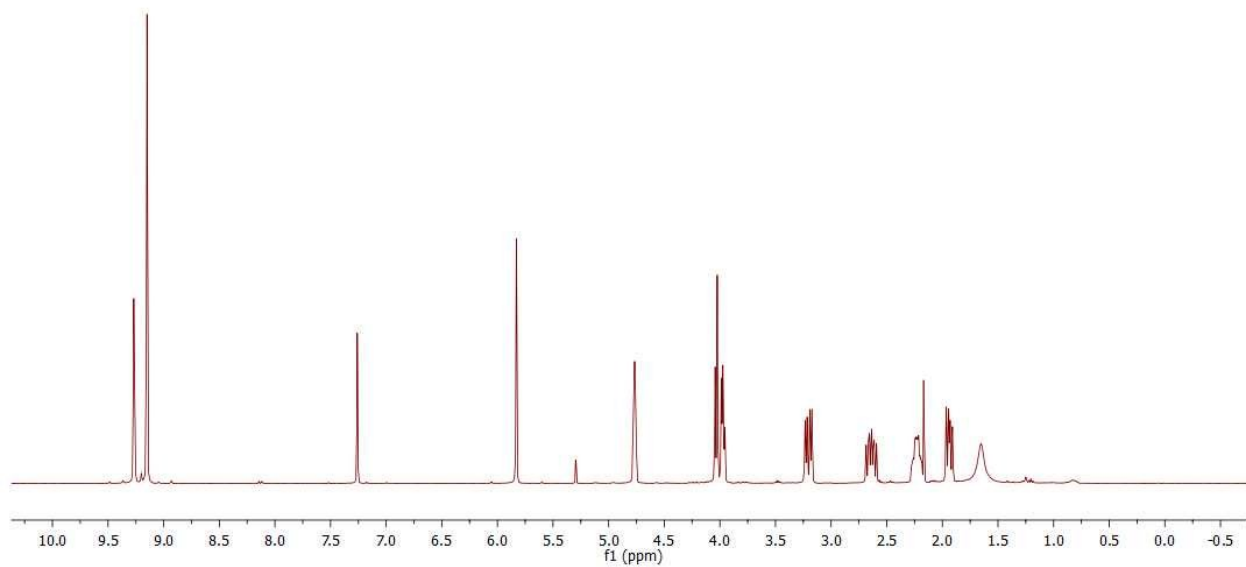


$^{13}\text{C-NMR}$

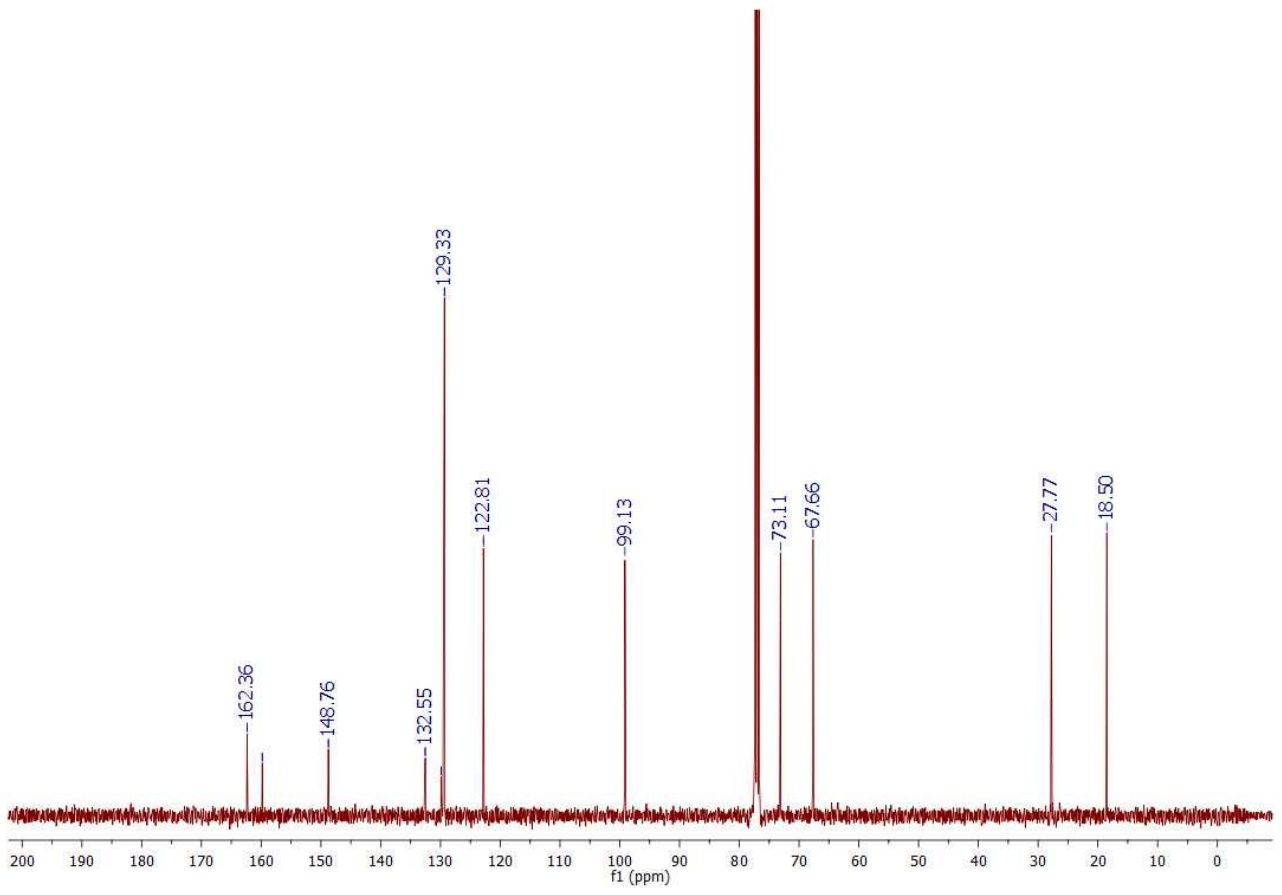


(Z)-Dihydrolevoglucosenone *O*-(3,5-dinitrobenzoyl)oxime (8).

$^1\text{H-NMR}$

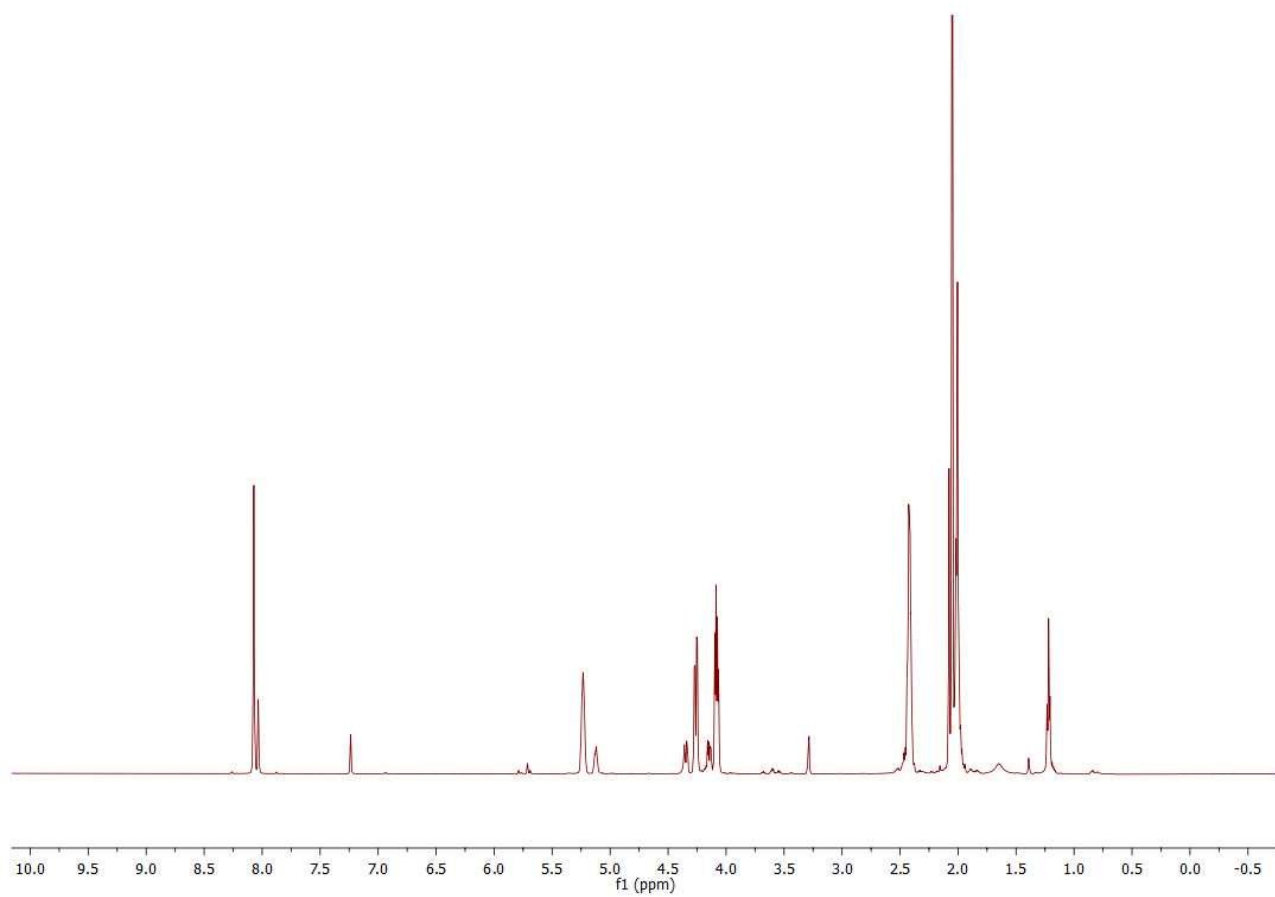


$^{13}\text{C-NMR}$

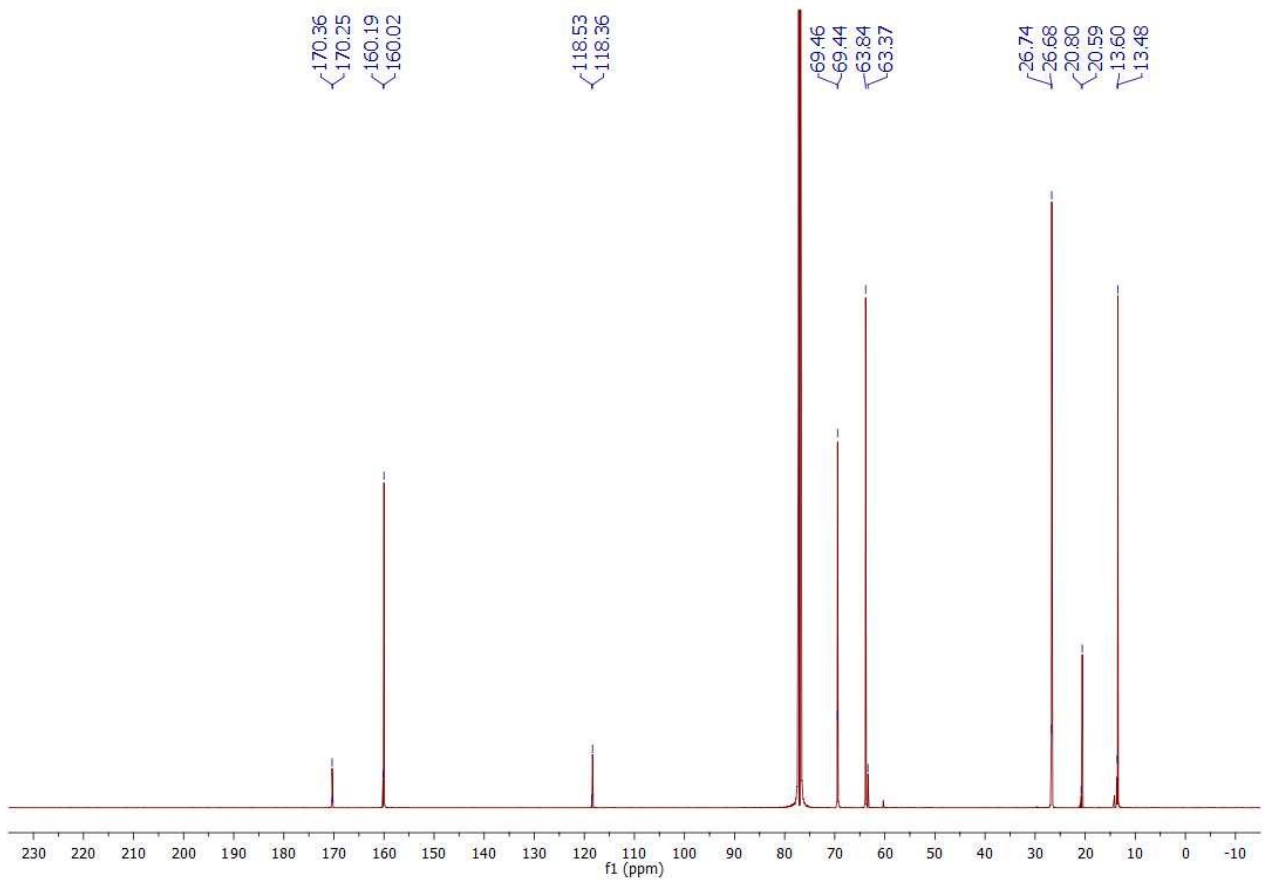


Nitriles 9 and 10.

$^1\text{H-NMR}$

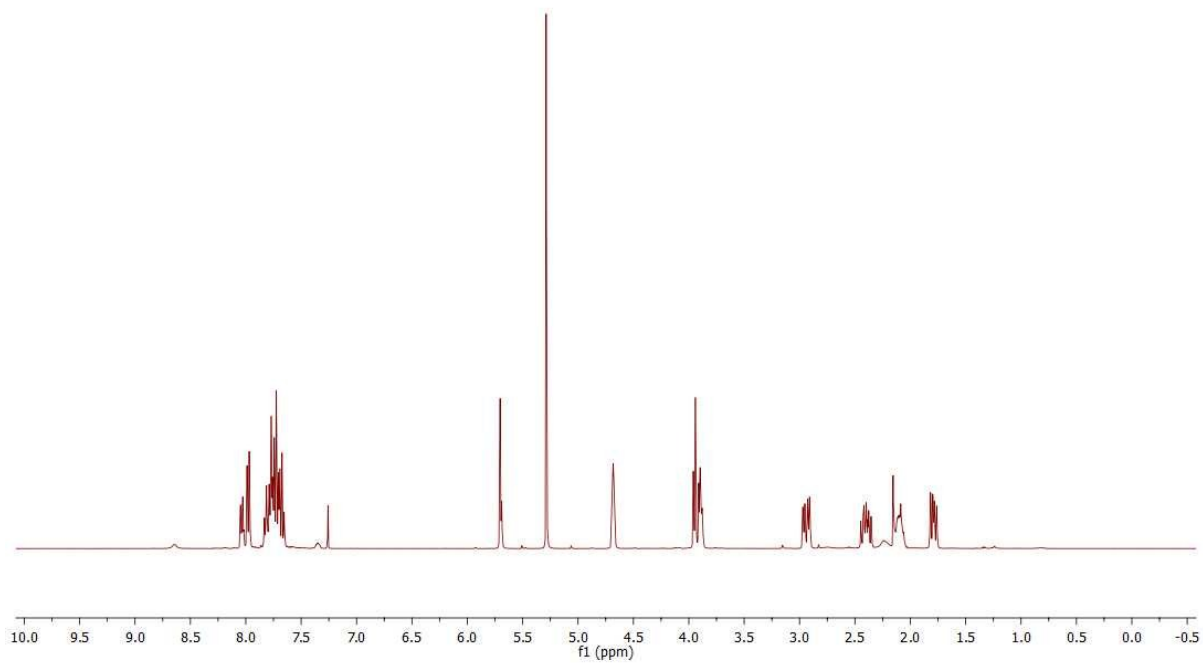


$^{13}\text{C-NMR}$

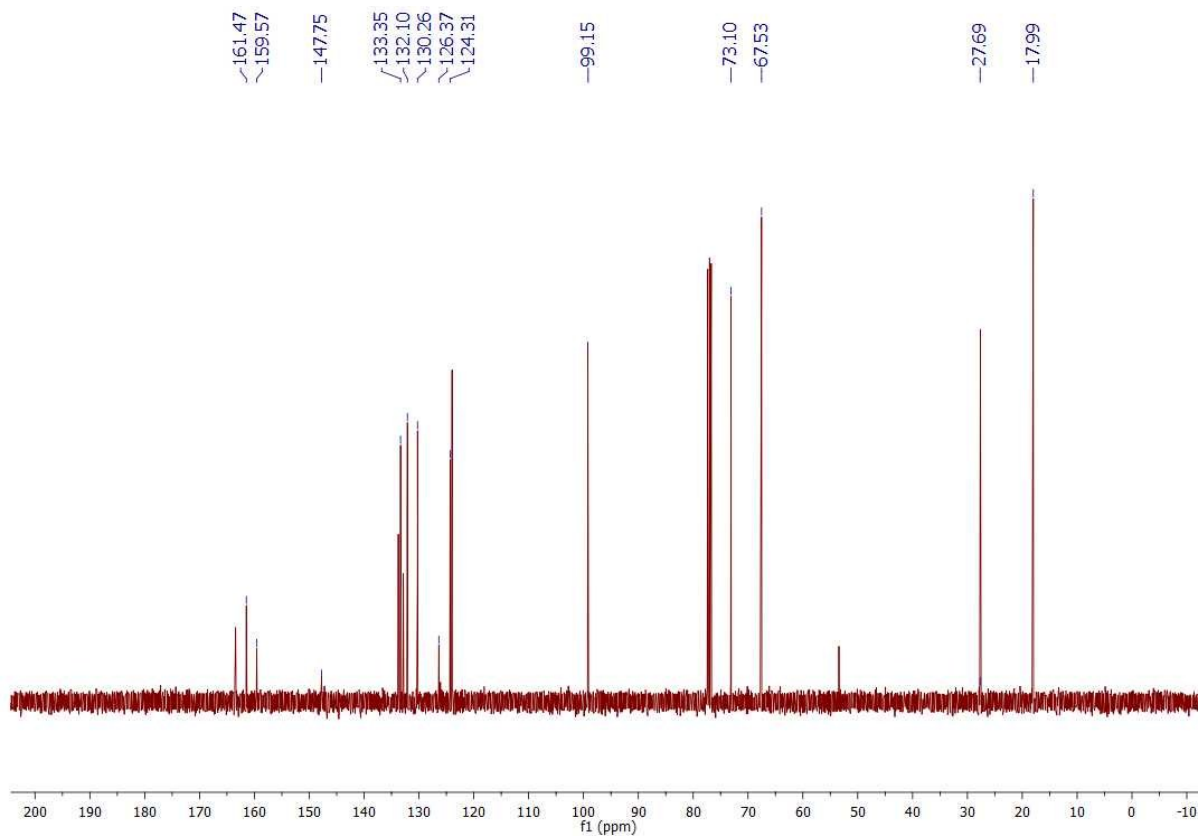


Dihydrolevoglucosenone *O*-(2-nitrobenzoyl)oxime (11).

¹H-NMR

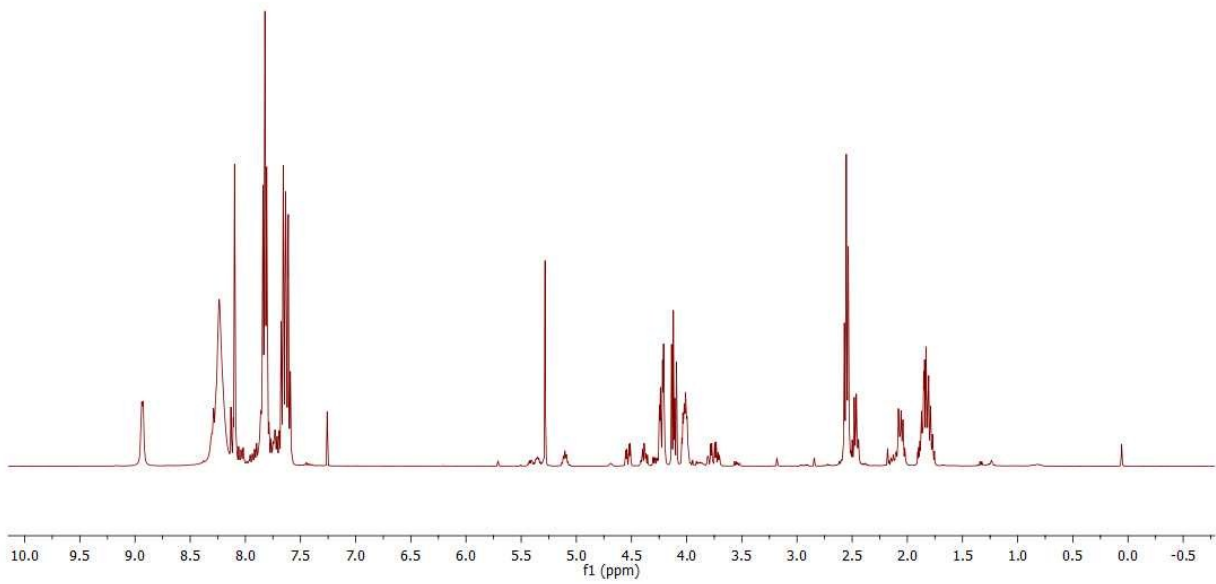


¹³C-NMR



Nitriles 12 and 13

$^1\text{H-NMR}$



$^{13}\text{C-NMR}$

