

Supplementary Information

S1. Differential absorption spectral shape between HPO and $\text{Eu(hfa)}_3(\text{HPO})_2$

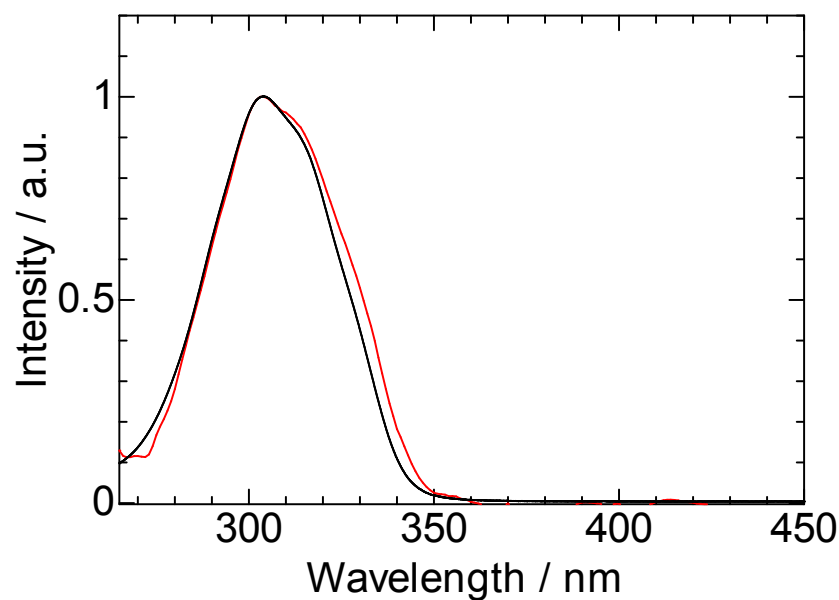


Figure S1. The differential electronic absorption spectrum between HPO and Eu(hfa)(HPO)_2 in CHCl_3 (red line), and electronic absorption spectrum of $\text{Eu(hfa)(H}_2\text{O)}_2$ in CH_3OH (black line). Normalized by intensity maxima.

S2. Electronic absorption spectra of HPO and Eu(hfa)₃(HPO)₂

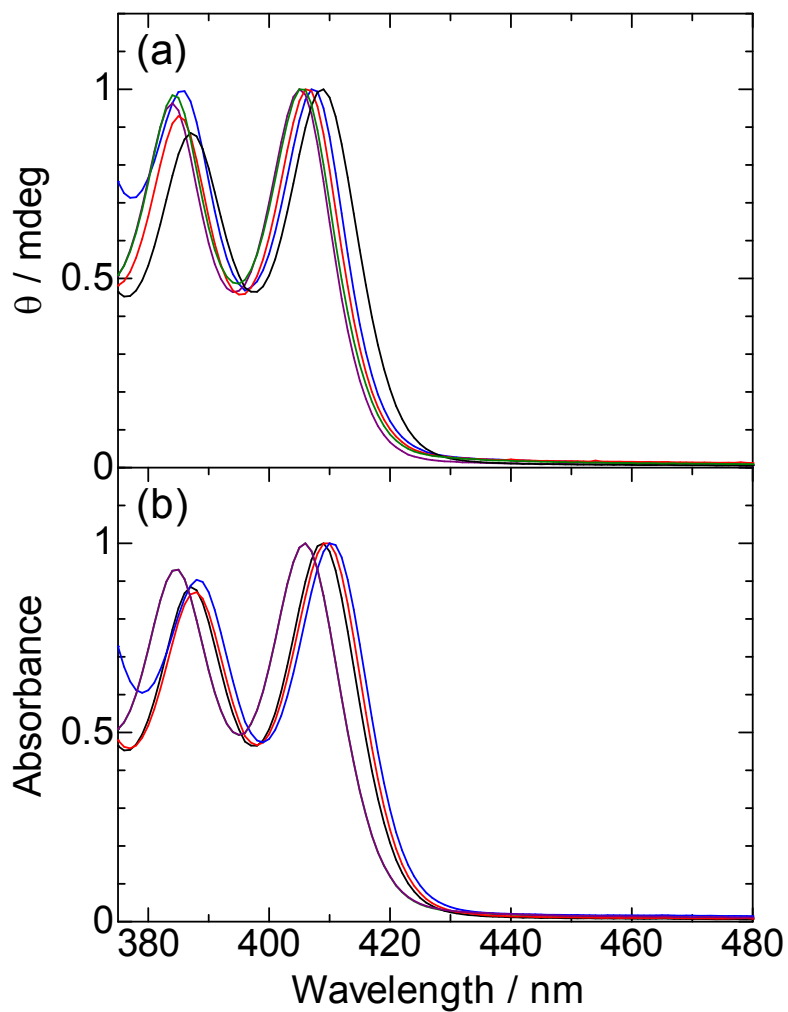


Figure S2. The electronic absorption spectra of HPO (a, 1.0×10^{-4} M) and Eu(hfa)₃(HPO)₂ (b, 1.0×10^{-4} M) (black line: CHCl₃, red line: C₂H₅Br, blue line: C₂H₅I, green line: CH₃COCH₃, purple line: CD₃COCD₃). Normalized by intensity maxima.

S3. Fluorescence spectra of HPO ligand

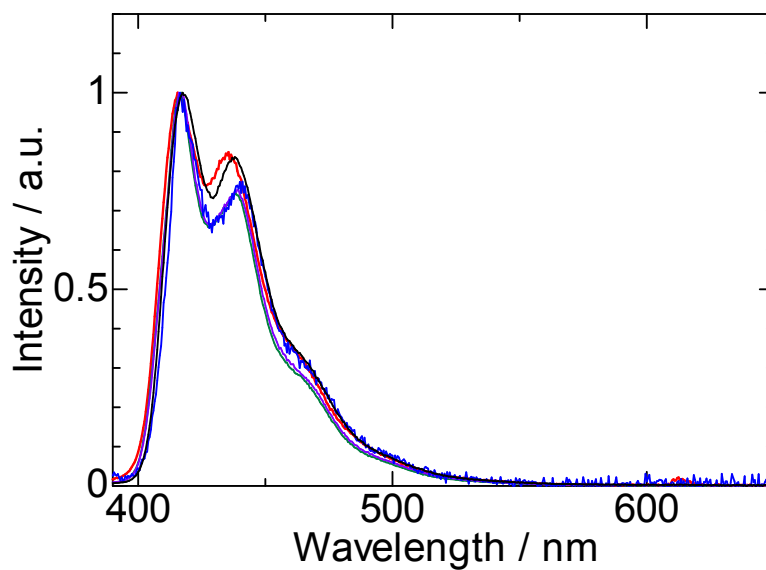


Figure S3. The fluorescence spectra of HPO ($\lambda_{\text{ex}} = 370 \text{ nm}$, $1.0 \times 10^{-4} \text{ M}$) (black line: CHCl_3 , red line: $\text{C}_2\text{H}_5\text{Br}$, blue line: $\text{C}_2\text{H}_5\text{I}$, green line: CH_3COCH_3 , purple line: CD_3COCD_3). Normalized by intensity maxima.

S4. Excitation spectra of Eu(hfa)₃(HPO)₂ ligand

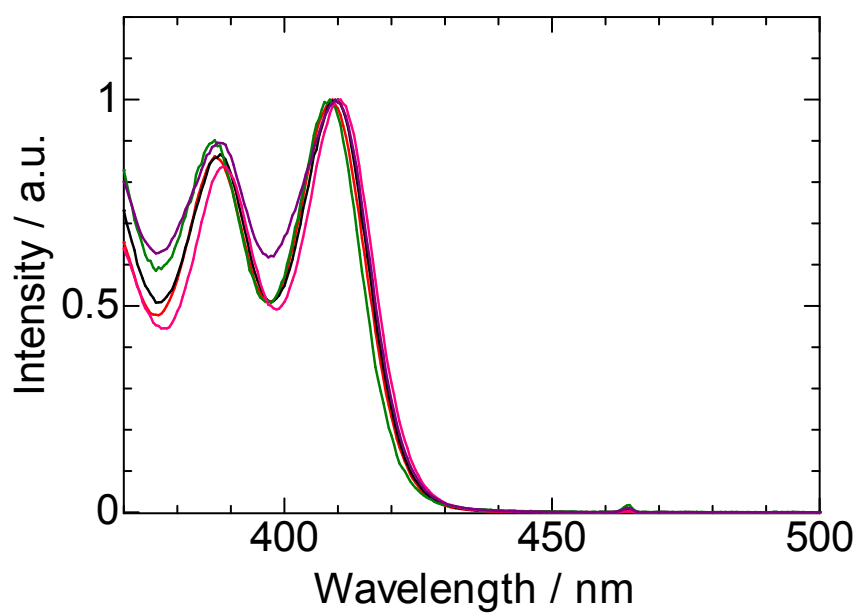


Figure S4. Excitation spectra ($\lambda_{\text{ex}} = 613 \text{ nm}$, $1.0 \times 10^{-4} \text{ M}$) of Eu(hfa)₃(HPO)₂ (black line: CHCl₃, red line: C₂H₅Br, blue line: C₂H₅I, green line: CH₃COCH₃, purple line: CD₃COCD₃). Normalized by intensity maxima.

S5. Luminescence spectra of HPO

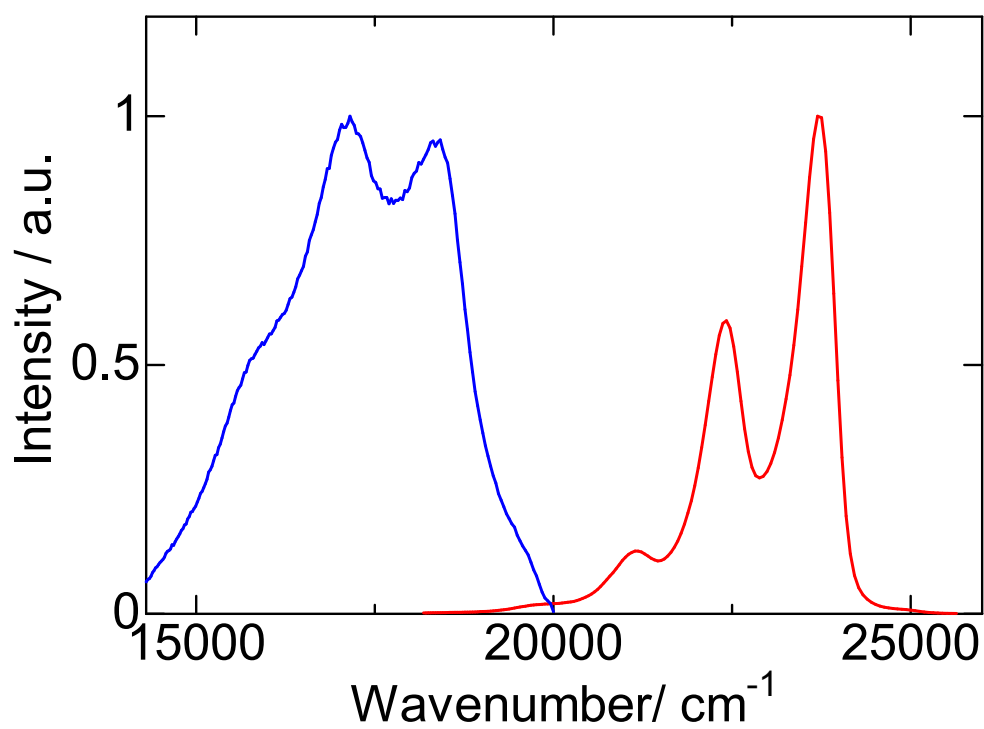


Figure S5. Fluorescence (red line) and phosphorescence (blue line) spectra of HPO in solutions.²⁹ Normalized by intensity maxima.

S6. DFT calculation results

Table S1. TD-DFT calculation (B3LYP/6-31G(d)) for HPO ligand.

	$\lambda_{\text{abs}}/\text{nm}$	f	main contribution
S ₁	369	0.0009	H-1→L (53 %), H→L+1 (39 %)
S ₂	349	0.0871	H→L (71 %), H-1→L+1 (21 %)
S ₃	320	0.3935	H→L (16 %), H-1→L+1 (59 %)
S ₄	315	0.0377	H-1→L (28 %), H→L+1 (31 %), H-2→L (23 %)