

Electronic Supplementary Information:
Anion binding by *p*-aminoazobenzene derived aromatic amides.
Spectroscopic and electrochemical studies

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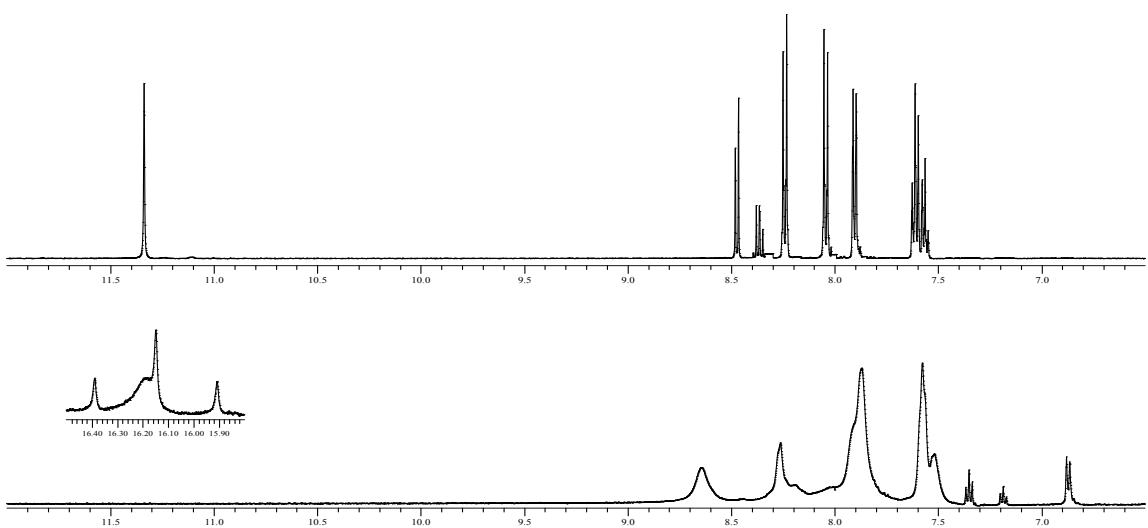


Fig. ESI1 The comparison of ¹H NMR spectra of free ligand 2 (top) and after addition of tetra-n-butylammonium fluoride (2:F- = 1:1, bottom) in DMSO-d₆.

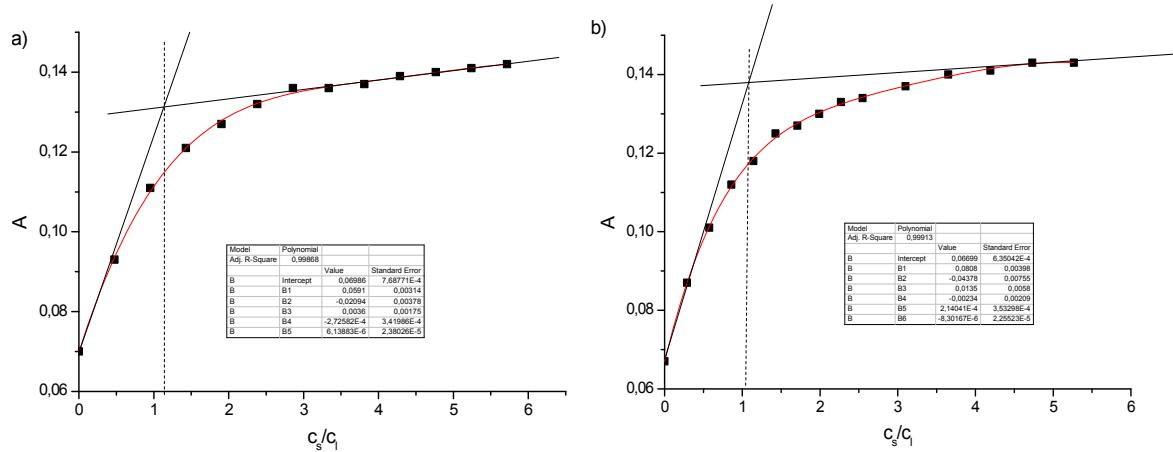


Fig. ESI2 Molar ratio plot obtained from spectrophotometric titration of ligand **4** with tetra-*n*-butylammonium a) benzoate ($c_4 = 1.37 \times 10^{-5}$ M, $0 \leq R \leq 5.72$), b) fluoride ($c_4 = 9.13 \times 10^{-6}$ M, $0 \leq R \leq 3.13$) in chloroform; R= salt/ligand molar ratio.

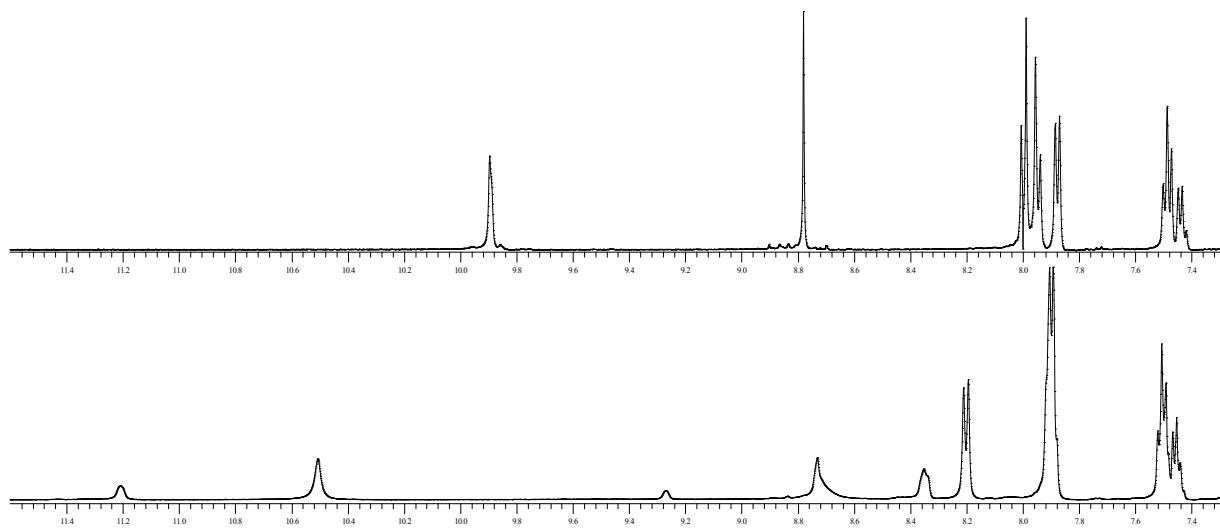


Fig. ESI3 The comparison of ¹H NMR spectra of free ligand **4** (top) and its complex (1:1) with tetra-*n*-butylammonium fluoride (bottom) in chloroform-*d*.

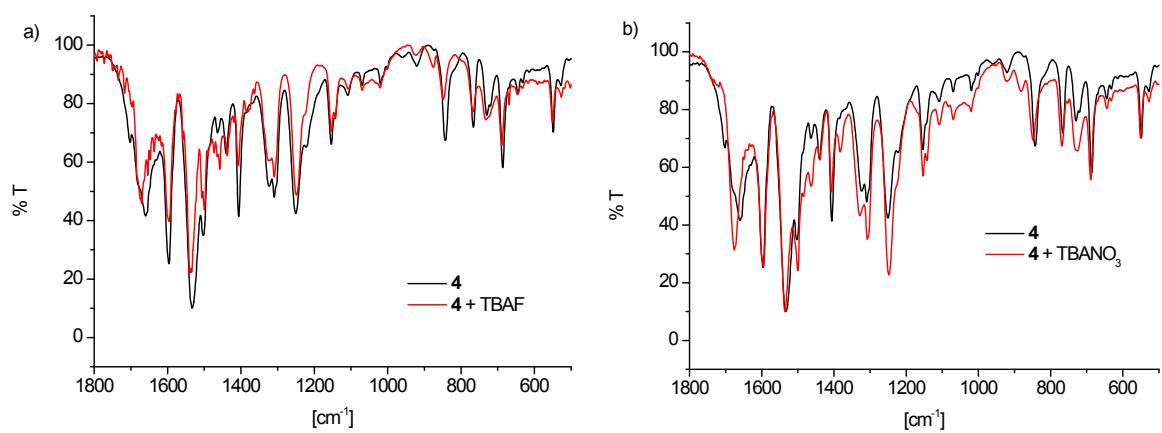


Fig. ESI4 The comparison of FTIR spectra (KBr pellets) of the ligand **4** and its complex with a) tetra-*n*-butylammonium fluoride (1:1); b) tetra-*n*-butylammonium nitrate (1:1).

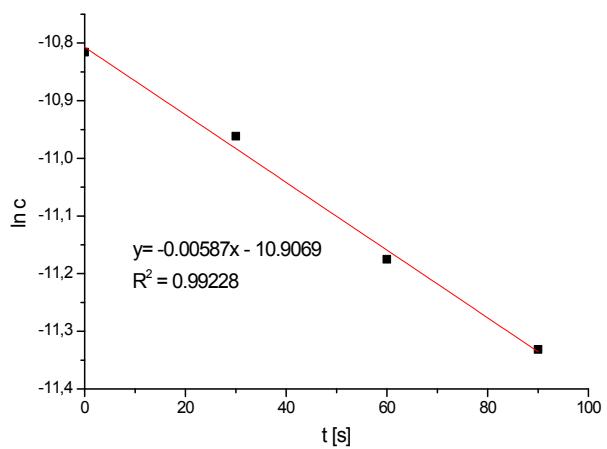


Fig. ESI5 Plot of the first-order reaction determined for *trans* to *cis* isomerization of ligand **4** ($c_4 = 2.01 \times 10^{-5}$ M) solution in chloroform ($\lambda = 354$ nm).

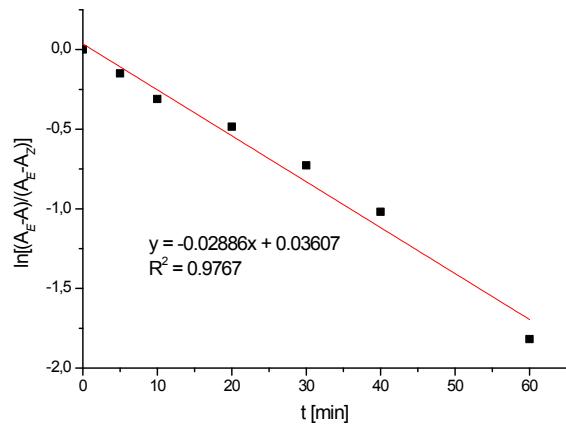


Fig. ESI6 Plot of the first-order reaction determined for thermal *cis* to *trans* isomerization of ligand **4** ($c_{4\text{trans}} = 2.01 \times 10^{-5}$ M) solution in chloroform ($\lambda = 354$ nm).