

New insight into efficient visible-light-driven photocatalytic organic transformation over CdS/TiO₂ photocatalysts

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Table S1. Progress achieved in the photocatalytic Aerobic Oxidation of benzyl alcohol to benzaldehyde under visible light

Sample	Solvent	Benzyl alcohol (mmol)	Time (h)	Conversion (%)/(g.h)	Selectivity (%)	Light source	Year
CdS/TiO ₂ -90	BTF	0.1	4	1407.2	>97.03	>420 nm	This work
Au/BiOCl-OV	Acetonitrile	0.5	8	189	>99	>420 nm	2017 ¹
CdS	BTF	0.1	10	412.5	100	>420 nm	2012 ²
CdS/TiO ₂ (001)	BTF	0.2	3	1055.6	100	>420 nm	2016 ³
Pt@CeO ₂	BTF	0.1	5	925	>98	>420 nm	2011 ⁴
CdS/TiO ₂ nanofiber	BTF	0.1	4	687.5	>99	>420 nm	2015 ⁵
CdS-Uio-66(NH ₂)	BTF	0.1	4	941.2	>99	>420 nm	2013 ⁶
Mpg-C ₃ N ₄	BTF	0.1	3	380	>99	400-800 nm	2010 ⁷
CdS/Graphene	BTF	0.1	4	1375	>99	>420	2011 ⁸
GO-C ₃ N ₃ S ₃	BTF	0.5	8	214.6	100	>420	2014 ⁹

Table S2. BET specific surface area of the samples.

Sample	BET(m^2g^{-1})
TiO ₂	201
CdS/TiO ₂ -30	163
CdS/TiO ₂ -60	155
CdS/TiO ₂ -90	148
CdS/TiO ₂ -120	136
Co-CdS	16

Table S3. Photocatalytic selective oxidation of various benzyl alcohols over CdS/TiO₂-90 under visible light irradiation for 4h.

Light(nm)	R	Atm	Conversion(%)	Selectivity(%)
420-800	F	O ₂	40.04	89.63
420-800	NO ₂	O ₂	30.72	88.01
420-800	CH ₃	O ₂	61.05	> 99
420-800	CH ₃ O	O ₂	79.22	> 99
420-800	H	O ₂	44.2	95.2

Table S4. Photoreduction performances of 4-NA over sample Cd-Ti-90 under different conditions.

Catalyst	Hole scavenger	Atmosphere	Conversion(%)	Selectivity(%)
CdS/TiO ₂ -90	CH ₃ OH	N ₂	-	-
CdS/TiO ₂ -90	(NH ₄) ₂ C ₂ O ₄	N ₂	-	-
CdS/TiO ₂ -90	HCO ₂ NH ₄	N ₂	98.6	95.3
CdS/TiO ₂ -90	HCO ₂ NH ₄	O ₂	-	-
CdS/TiO ₂ -90	-	Air	-	-
-	HCO ₂ NH ₄	N ₂	-	-

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