Electronic Supplementary Material (ESI) for Polymer Chemistry. This journal is © The Royal Society of Chemistry 2017

Electronic Supplementary Information (ESI)

Reaction-Based Conjugated Polymer Fluorescent Probe for Mercury(II): Good Sensing Performance with "Turn-on" Signal Output

Jun Ding,^a Huiyang Li,^a Yujun Xie,^a Qian Peng,^b Qianqian Li,^a and Zhen Li^a*

Department of Chemistry, Hubei Key Lab on Organic and Polymeric Opto-Electronic Materials, Wuhan
University, Wuhan 430072, China

Corresponding author. Phone: 86-27-62254108; Fax: 86-27-68755363;

E-mail: <u>lizhen@whu.edu.cn</u> or <u>lichemlab@163.com</u> .

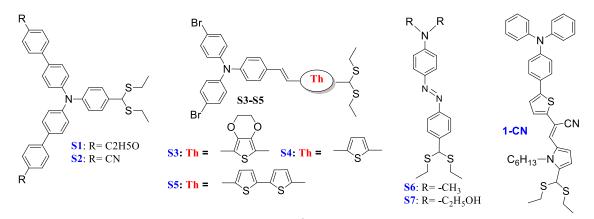


Chart S1. Different chemical Hg²⁺ sensors designed in our group.

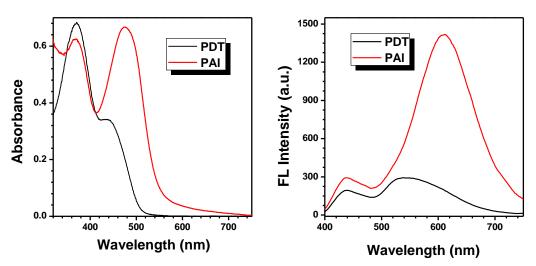


Figure S1. The UV-vis absorption (Left), fluorescent emission (Right) spectra of **PAI** and **PDT** in THF.

^b Key Laboratory of Organic Solids, Beijing National Laboratory For Molecular Science, Institute of Chemistry, Institute of Chemistry, Chinese Academy of Sciences, 100190 Beijing, China.

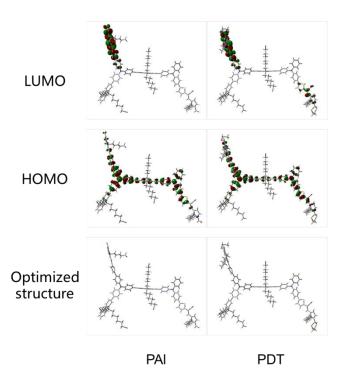


Figure S2. Frontier molecular orbitals optimized at the B3LYP/6-31G* level of theory and electronic cloud distribution in dimers of **PA1** and **PDT**.

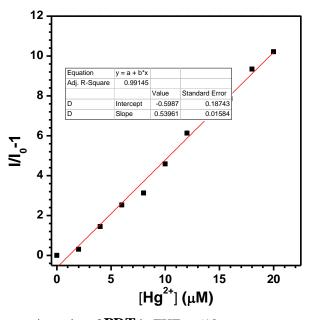


Figure S3. Fluorescence intensity of **PDT** in THF at 610 nm versus concentrations of Hg²⁺.

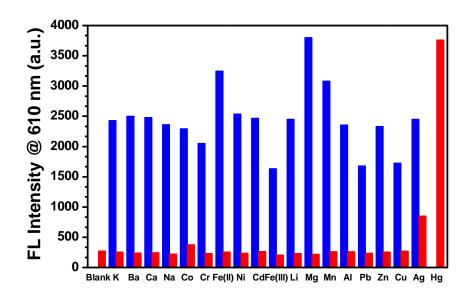


Figure S4. Fluorescence intensity at 610 nm of **PDT**(5 μ M) upon the addition of various metal ions (blue bar: **PDT** with other metals, red bar: **PDT** with Hg²⁺ (10 μ M) and other metals (50 μ M).

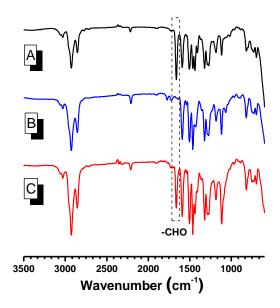


Figure S5. FTIR spectra of A (PAI), B (PDT), and C (PDT in presence of excess Hg²⁺).

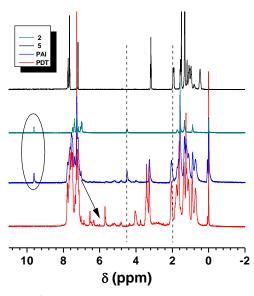


Figure S6. ¹H NMR spectra of 2, 5, PAI and PDT.

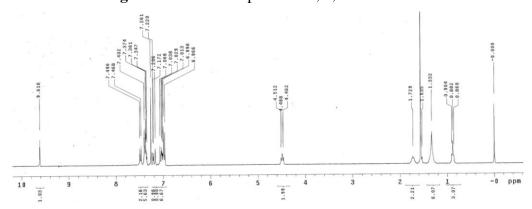


Figure S7. ¹H NMR spectrum of 2

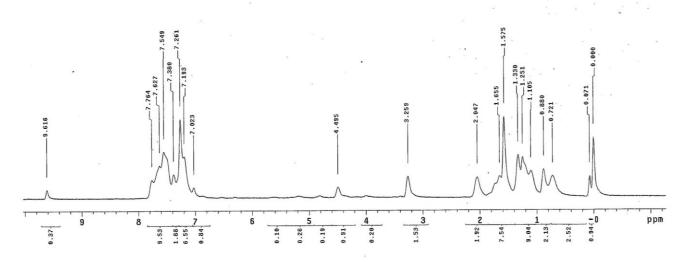


Figure S8. ¹H NMR spectrum of PA1

