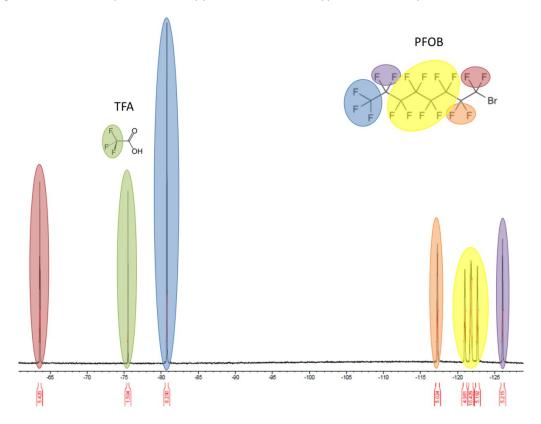
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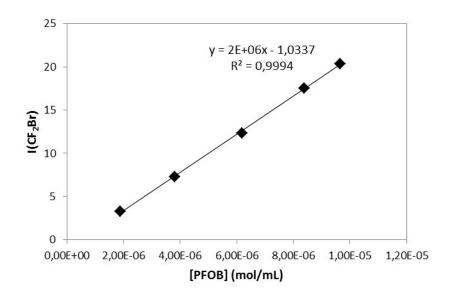
Supporting information

Calculation of PFOB encapsulation efficiency

A calibration curve with five solutions of known concentration of PFOB in chloroform was established. PFOB solutions were collected and introduced into a usual 5mm-NMR sample tube loaded with a stem coaxial insert containing TFA in D_20 as an external standard ([TFA]= 9.4 µmol.mL-1). The same insert was used for all solutions and the integration of the TFA peak at -76.5 ppm was set to 3. A typical ¹⁹F NMR spectrum is shown below.



Integrations of the peak at -64.7 ppm corresponding to the CF_2Br group of PFOB were drawn as a function of PFOB concentration. A typical calibration curve is shown below.



For each NCs sample, the integration of the TFA peak was set to 3 and the resulting integration of the CF₂Br peak was used to calculate the concentration of PFOB in the tube with the equation of the curve, and therefore the amount of encapsulated PFOB n_{PFOB} . Absolute encapsulation efficiency η_{encaps} was then calculated as follows:

$$\eta_{encaps} = \frac{n_{PFOB}}{n_{PFOB}^{max}} \qquad n_{max} = \frac{m_{PFOB}^{feed}}{M_{PFOB}m_{PFOB}^{feed} + m_{polymer}^{feed} + m_{SC}^{feed}}$$

where m_{PFOB}^{feed} , $m_{polymer}^{feed}$ and m_{SC}^{feed} are the initial masses of the components introduced during NCs preparation, m_{NC} corresponds to the mass of NCs recovered after freeze-drying and M_{PFOB} is the molar mass of PFOB (498.96 g/mol).

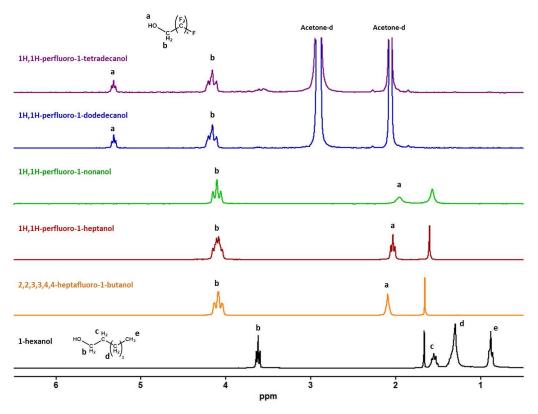


Figure S1: ¹H NMR spectra of the initiators used for polymers synthesis. All initiators were dissolved in CDCl₃ except 1H, H-perfluoro-1tetradecanol and 1H,1H-perfluoro-1-dodecanol that were dissolved in acetone-d

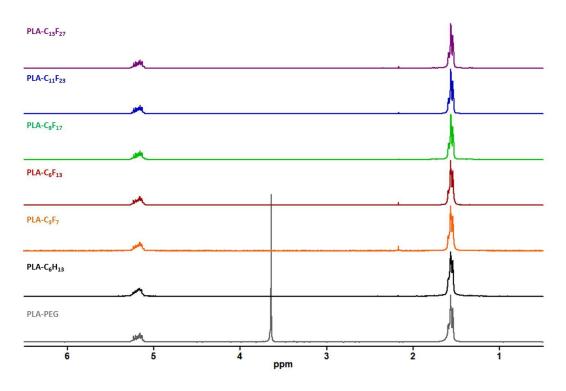


Figure S2: ¹H NMR spectra of all synthesized polymers in CDCl₃

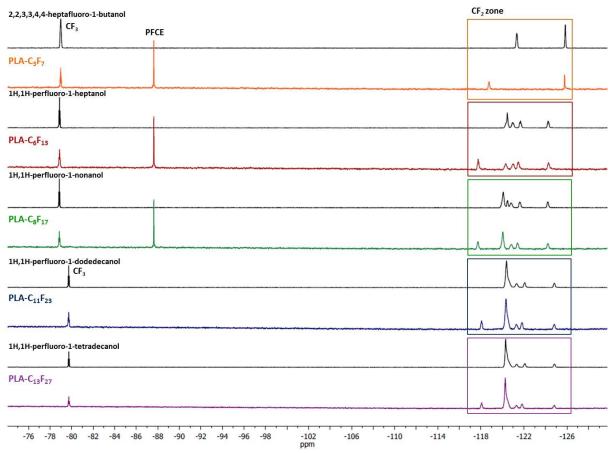


Figure S3: ¹⁹F NMR spectra of the fluorinated polymers (color) and their corresponding fluorinated initiators (black) showing the shifts in the CF₂ zone. PLA-C₃F₇, PLA-C₆F₁₃ and PLA-C₈F₁₇ were dissolved in CDCl₃ with PFCE, their corresponding initiators in CDCl₃ without PFCE. PLA-C₁₁F₂₃ and PLA-C₁₃F₂₇ and their corresponding fluorinated initiators were dissolved in acetone-d

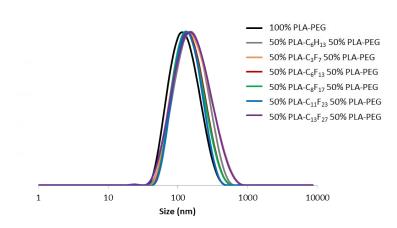


Figure S4: DLS intensity distribution of nanocapsules made from 50 mg of polymer

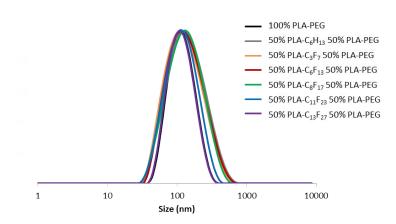


Figure S5: DLS intensity distribution of nanocapsules made from 30 mg of polymer

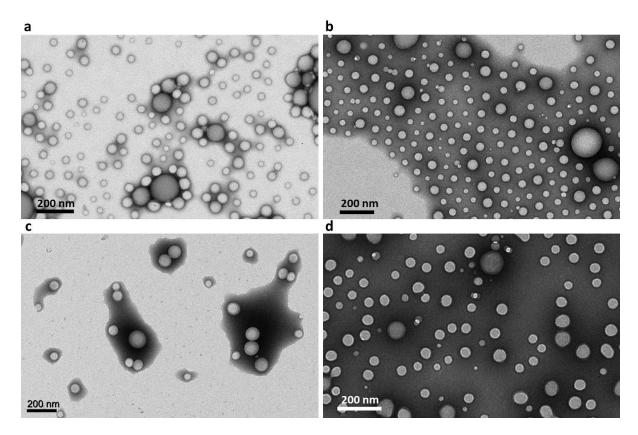


Figure S6: TEM images of NCs made from 50% PLA-C₈ F_{17} / 50% PLA-PEG at 50 mg (a) and 30 mg (b), and NCs made from 100% PLA-PEG at 50 mg (c) and 30 mg (d) with negative staining

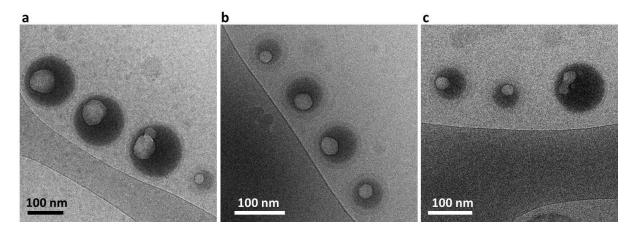
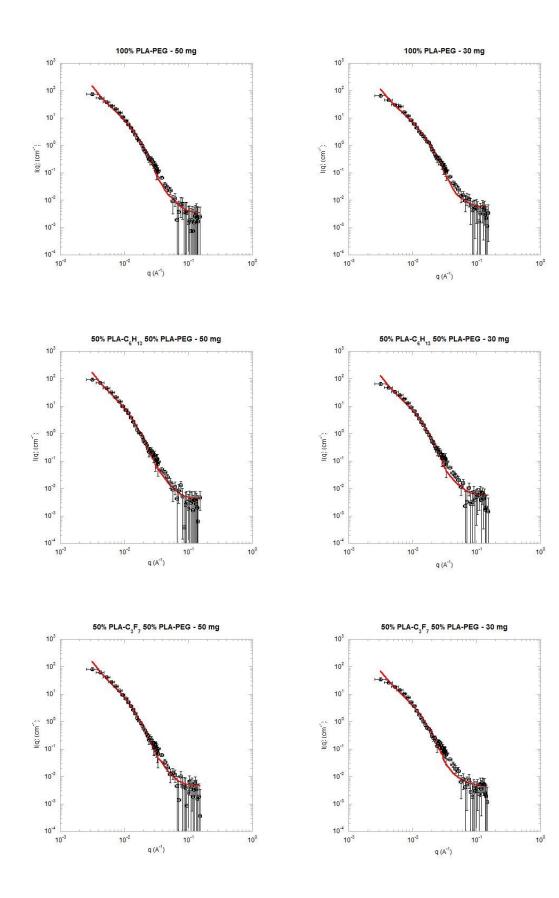


Figure S7: Cryo-TEM images of NCs made from 50% PLA-C₈F₁₇ / 50% PLA-PEG at 50 mg (a) and NCs made from 100% PLA-PEG at 50 mg (b) and 30 mg (c)



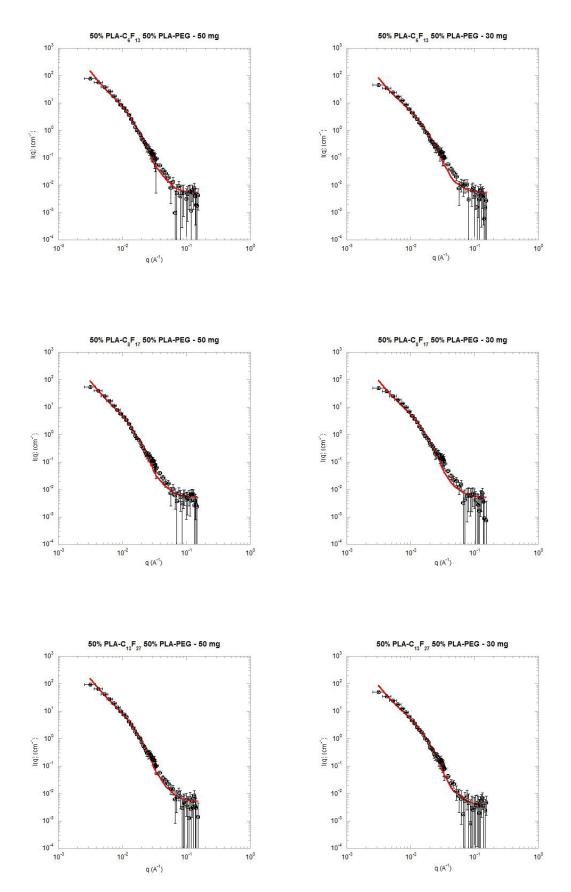
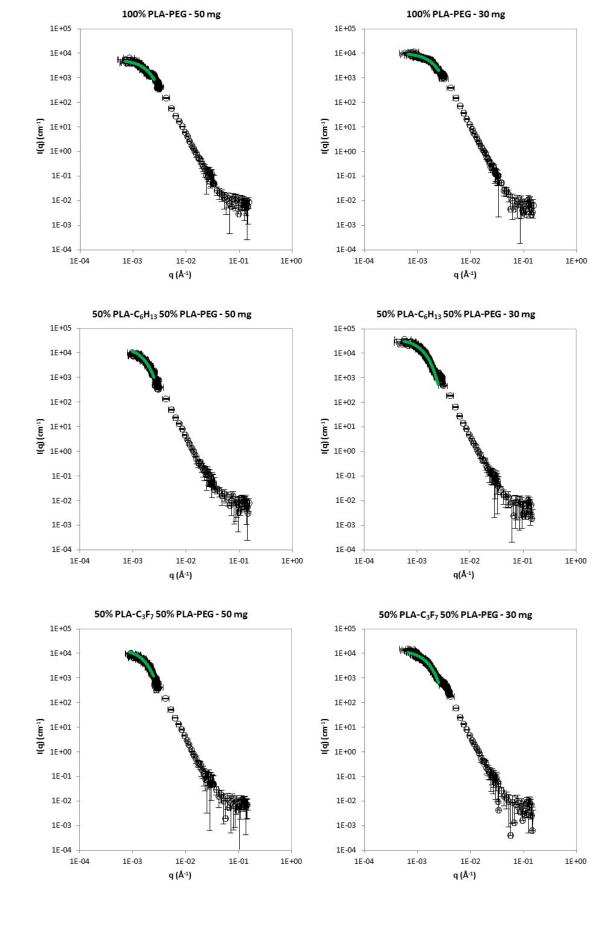


Figure S8: Experimental scattered intensity curves (black circles) in PFOB matching condition fitted with the vesicle model (red line)



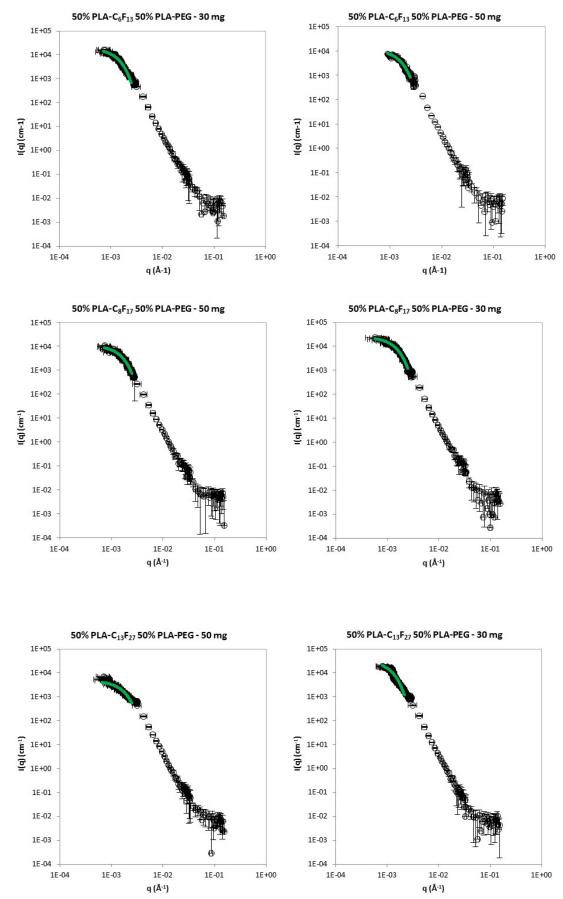


Figure S9: Experimental scattered intensity curves (black circles) in PLA matching condition fitted with the Guinier approximation for q < 0.0025 Å^{-1} (green line)

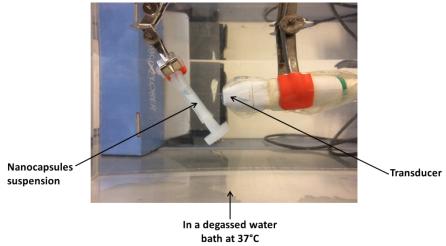


Figure S10 : Picture of the ultrasound imaging set-up