

Supporting Information

CO₂-Responsive Fluorescent Hyperbranched Poly(ether amine)s

Bing Yu, Yue Zhao*

Département de chimie, Université de Sherbrooke,

Sherbrooke, QC, Canada J1K 2R1

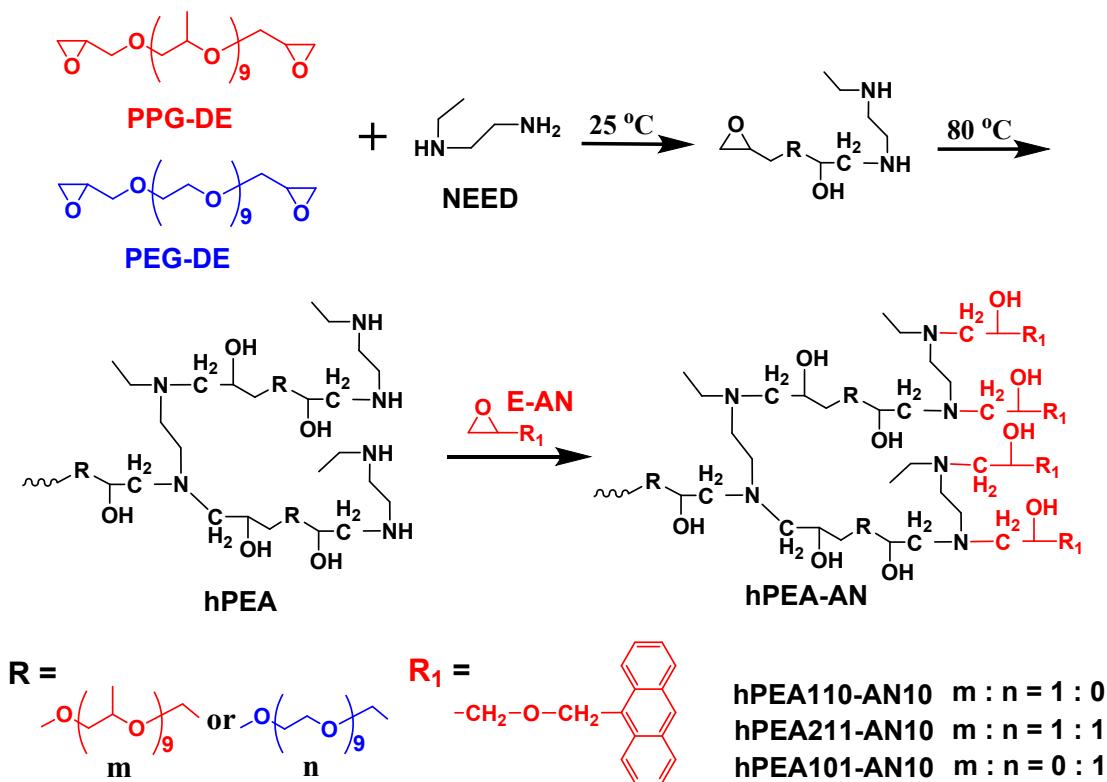


Figure S1. The synthesis process of hPEA-AN

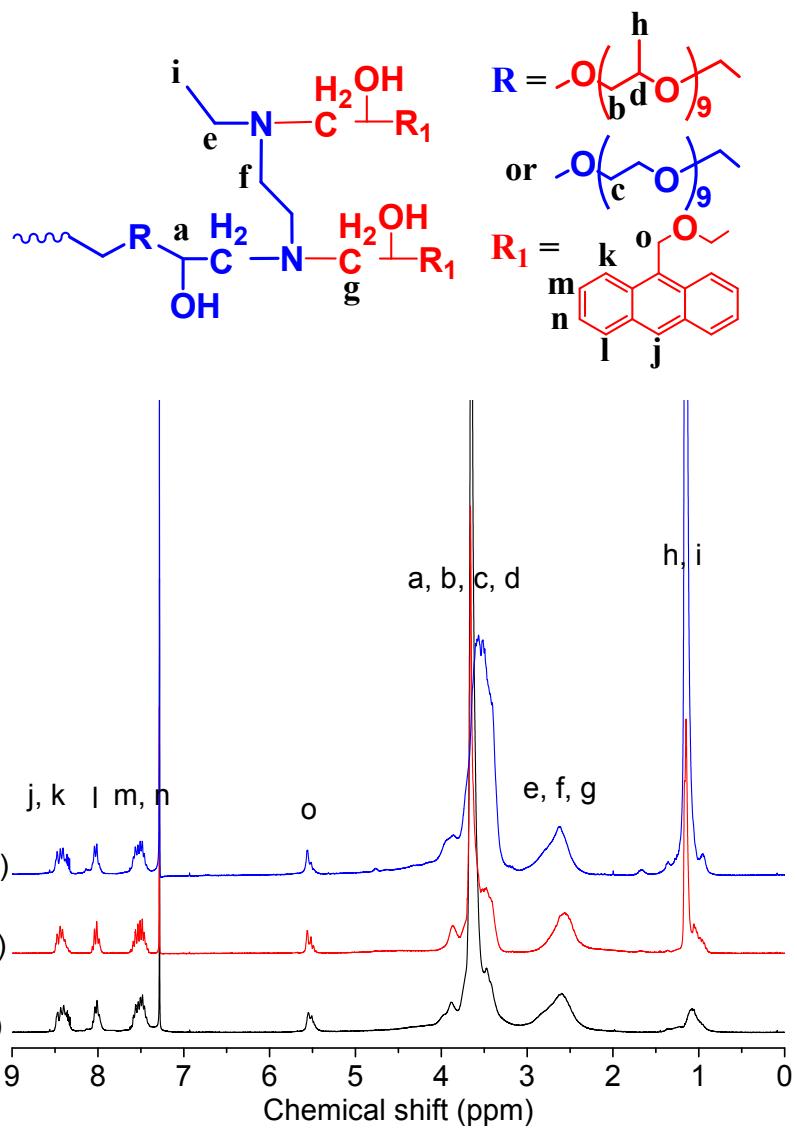


Figure S2. ¹H NMR spectra of (i) hPEA101-AN10, (ii) hPEA211-AN10, and (iii) hPEA110-AN10 in CDCl₃

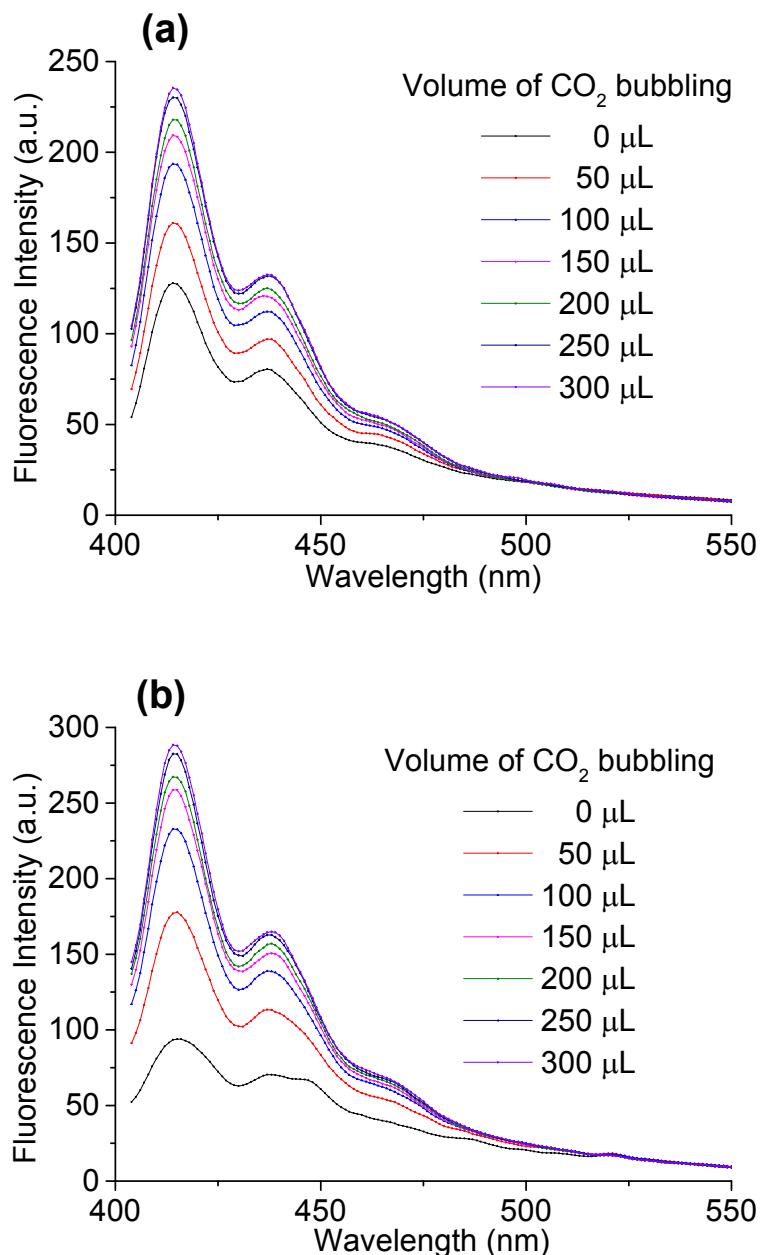


Figure S3. (a) The fluorescence emission spectra of 1 mg/mL hPEA101-AN10 aqueous solution with different amounts of CO_2 bubbling. (b) The fluorescence emission spectra of 1 mg/mL hPEA110-AN10 aqueous solution with different amounts of CO_2 bubbling.

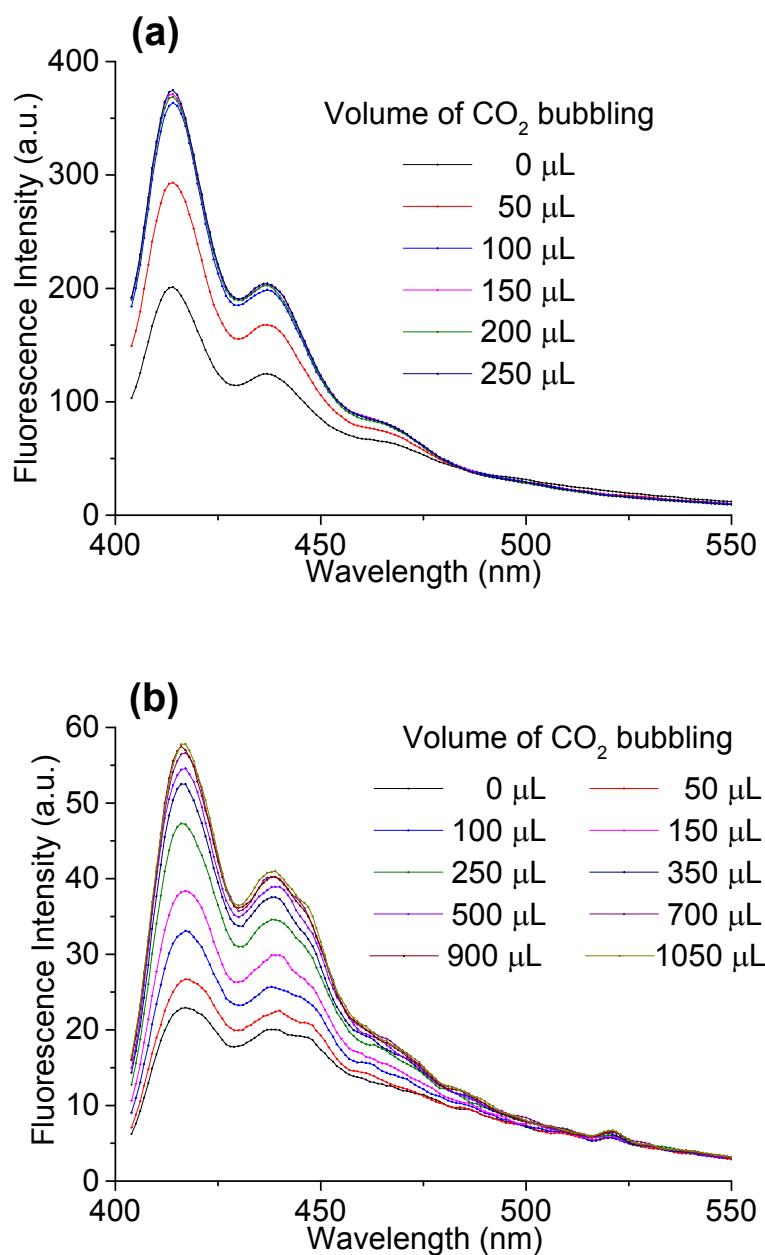


Figure S4. (a) The fluorescence emission spectra of 0.5 mg/mL hPEA211-AN10 aqueous solution with different amounts of CO_2 bubbling. (b) The fluorescence emission spectra of 3 mg/mL hPEA211-AN10 aqueous solution with different amounts of CO_2 bubbling.

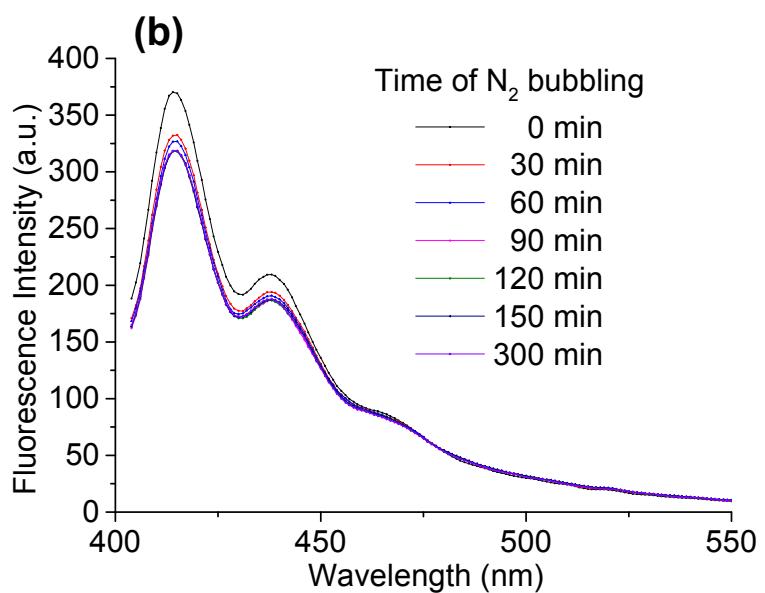
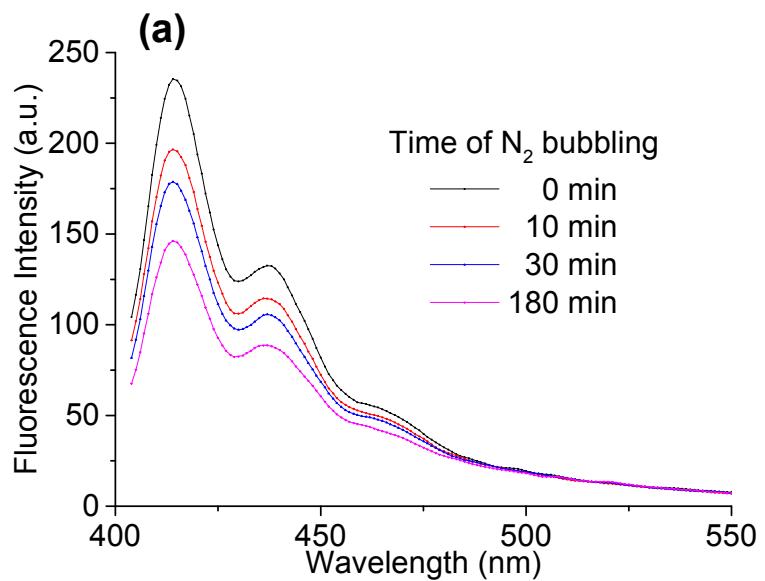


Figure S5. (a) The fluorescence emission spectra of 1 mg/mL hPEA101-AN10 aqueous solution with different time of N_2 bubbling. (b) The fluorescence emission spectra of 1 mg/mL hPEA110-AN10 aqueous solution with different time of N_2 bubbling.

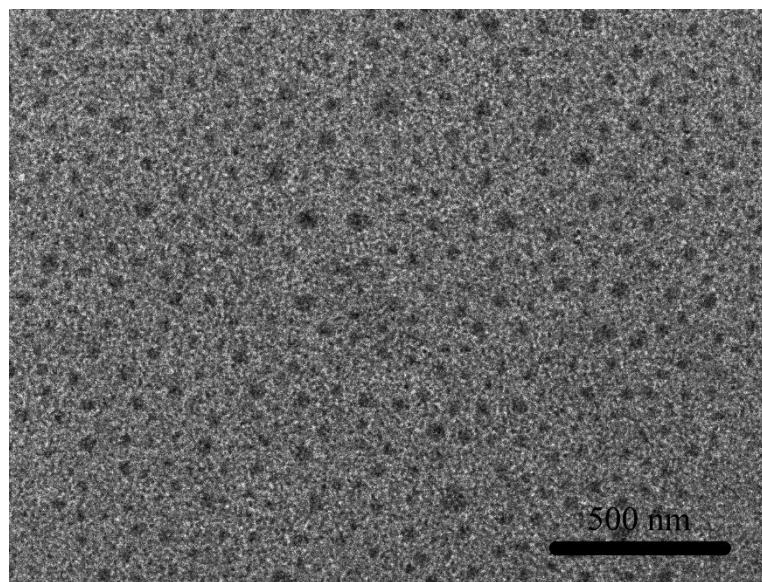


Figure S6. TEM image of hPEA211-AN10 micelles in water.

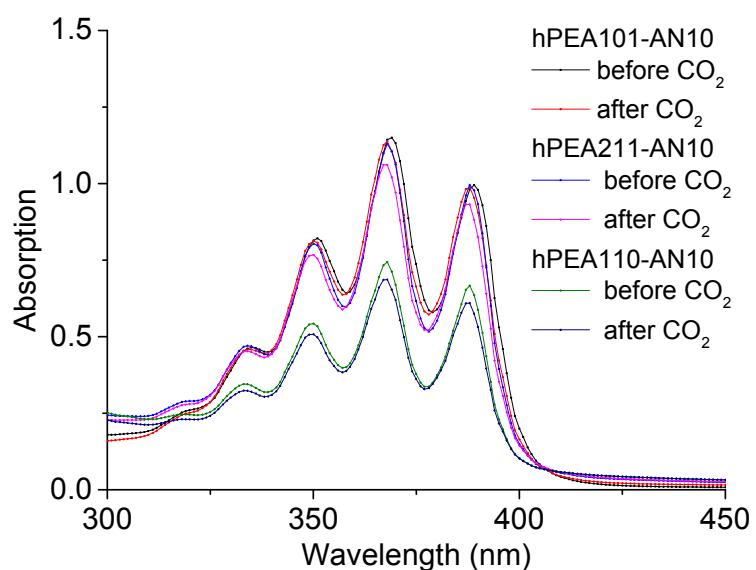


Figure S7. The UV-vis spectra of hPEA-AN aqueous solution before and after CO₂ bubbling at 25 °C.

Table S1. Change in pH value after bubbling CO₂ and, subsequently, N₂ through the aqueous solution of 1 mg/mL hPEA-AN with different compositions

pH	hPEA101-AN10	hPEA211-AN10	hPEA110-AN10
Initial state before CO ₂ injection	9.16	8.91	8.28
CO ₂ injection (400 μL)	5.28	5.11	4.85
N ₂ bubbling (90 min, 1000 mL/min)	8.64	8.44	7.48