

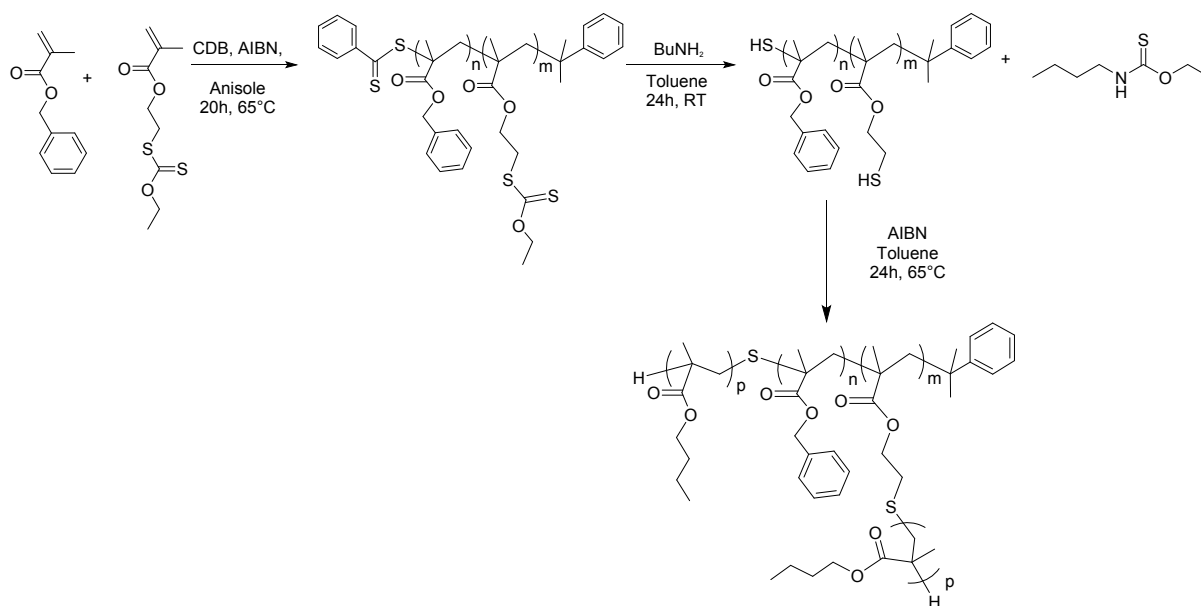
Synthesis of molecular brushes by telomerization

Coralie Teulère and Renaud Nicolay*

Matière Molle et Chimie, École Supérieure de Physique et de Chimie Industrielles de la Ville de Paris (ESPCI)–CNRS, UMR-7167, Paris Sciences et Lettres (PSL) Research University, 10 rue Vauquelin, 75005 Paris, France.

* Corresponding author: email renaud.nicolay@espci.fr

Supporting Information



Scheme S1 : Synthesis of PBMA brushes by telomerization via a two-pot two-step aminolysis and side-chain growth sequence

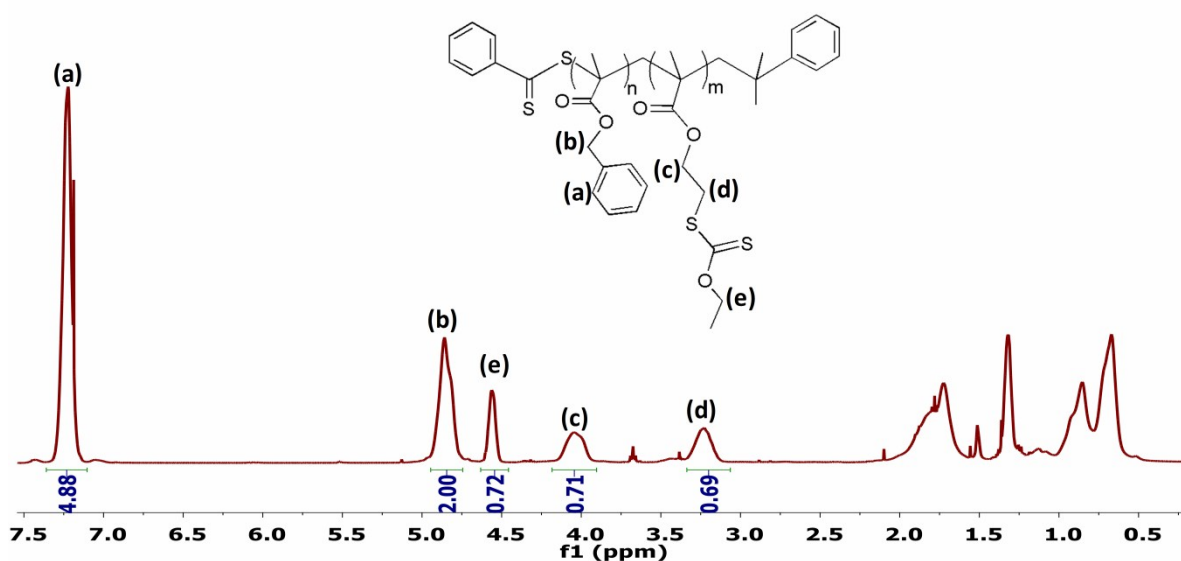


Figure S1: ^1H NMR spectrum of polymer **P1**. Experimental conditions: $[\text{BenzMA}]_0/[\text{XEMA}]_0/[\text{CDB}]_0/[\text{AIBN}]_0 = 2.3/1/0.02/0.002$ in anisole 40 vol%, 20 h at 65 °C

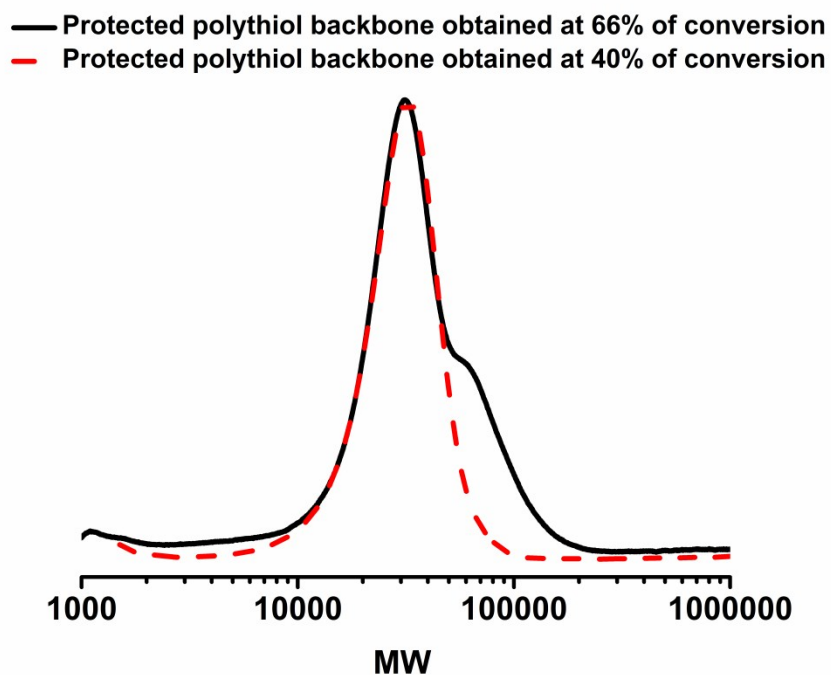


Figure S2: SEC traces of protected polythiol backbones. Black solid line experimental conditions: $[\text{BenzMA}]_0/[\text{XEMA}]_0/[\text{CDB}]_0/[\text{AIBN}]_0 = 2.3/1/0.02/0.002$ in anisole 40 vol%, 20 h at 65 °C; monomer conversion = 66%. Red dashed line experimental conditions: $[\text{BenzMA}]_0/[\text{XEMA}]_0/[\text{CDB}]_0/[\text{AIBN}]_0 = 2.3/1/0.013/0.0013$ in anisole 40 vol%, 20 h at 65 °C ; monomer conversion = 40%.

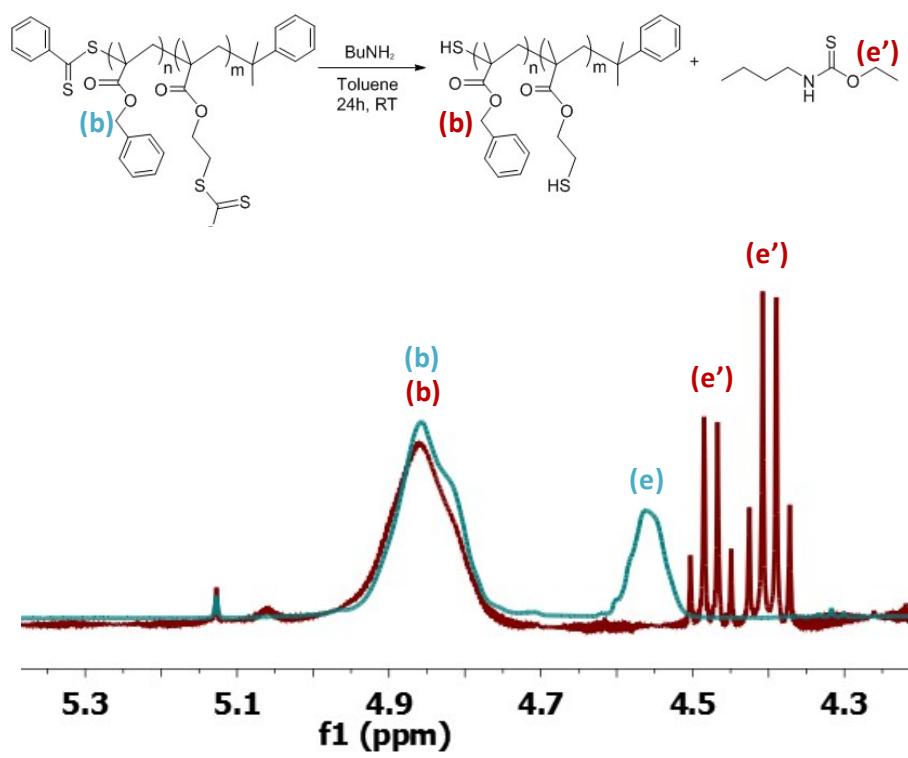


Figure S3: ^1H NMR spectra before (blue) and after (red) polythiol deprotection

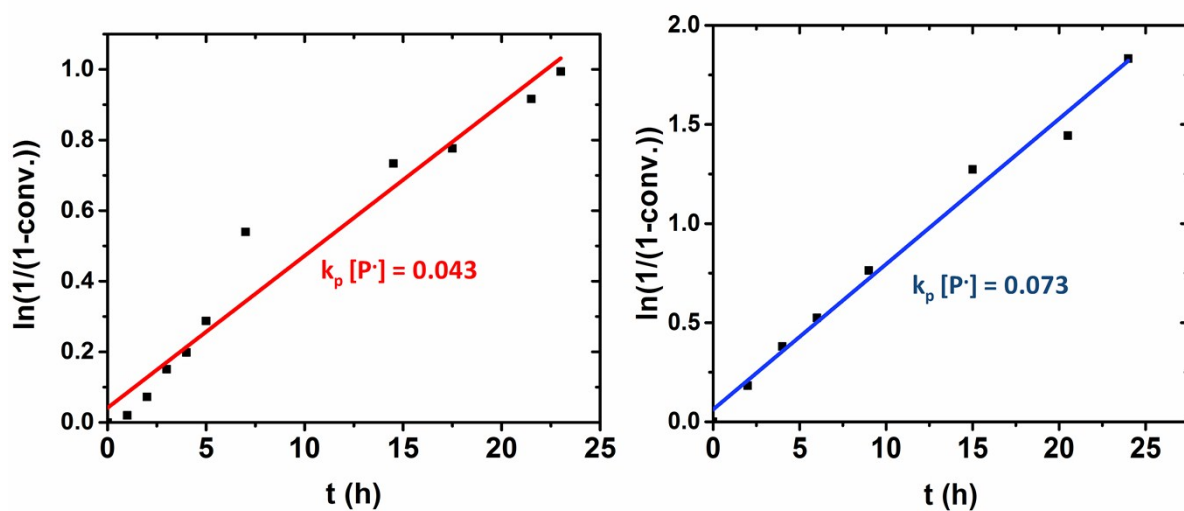
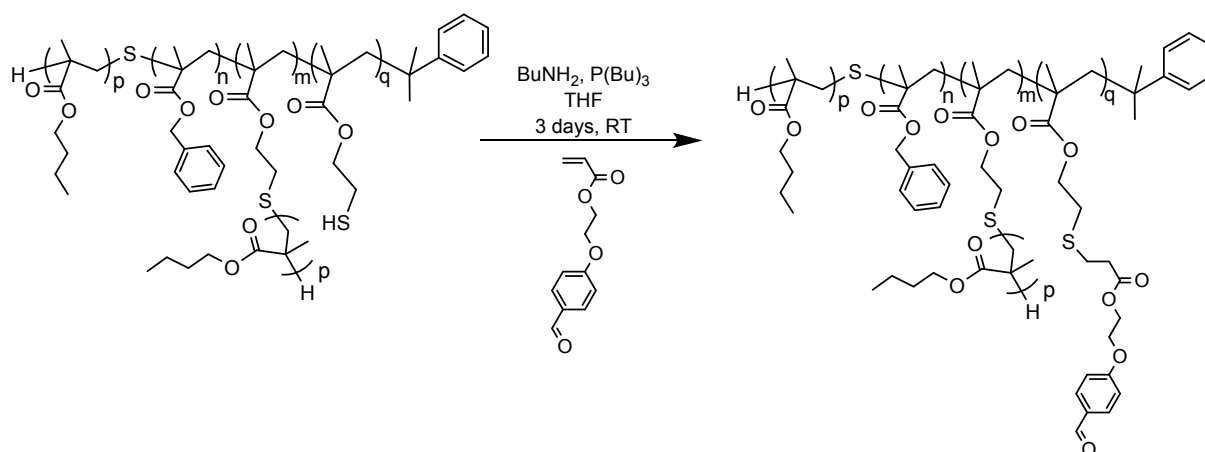


Figure S4: Kinetic plot for (a) **P5** polymerization in 50 % toluene solution at 65 °C with a ratio of $[\text{BMA}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 60/1/0.01$ and (b) **P7** polymerization in 50 % toluene solution at 65 °C with a ratio of $[\text{BMA}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 40/1/0.01$



Scheme S2: General procedure for thiol titration on bottlebrush polymers

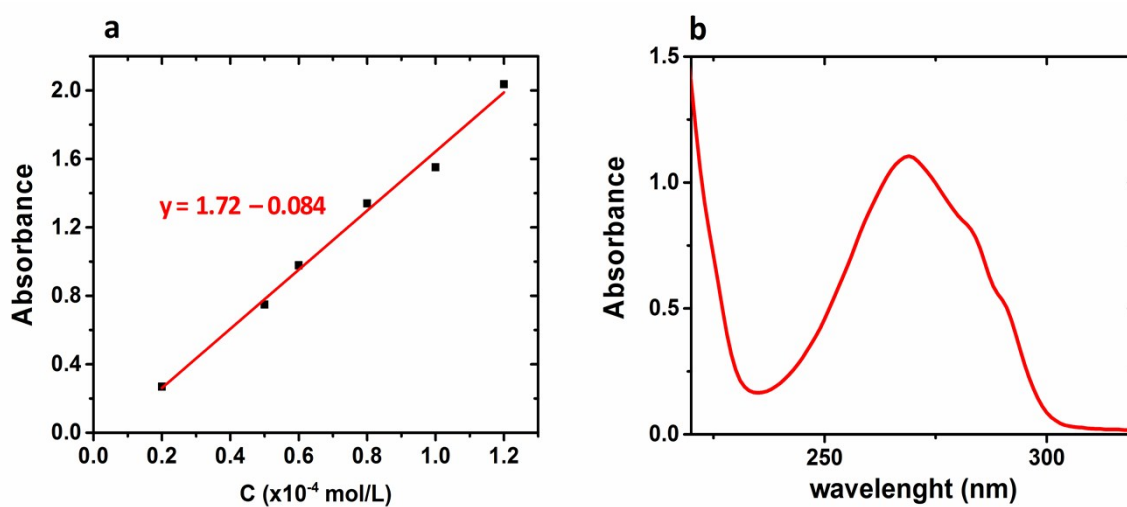


Figure S5: (a) Calibration curve of 4-(2-acryloxyethoxy)benzaldehyde at 270 nm – $\epsilon = 17\,200\text{ M}^{-1}\cdot\text{cm}^{-1}$ (b) UV spectrum of polythiol **P1** after deprotection and functionalization with 4-(2-acryloxyethoxy)benzaldehyde.

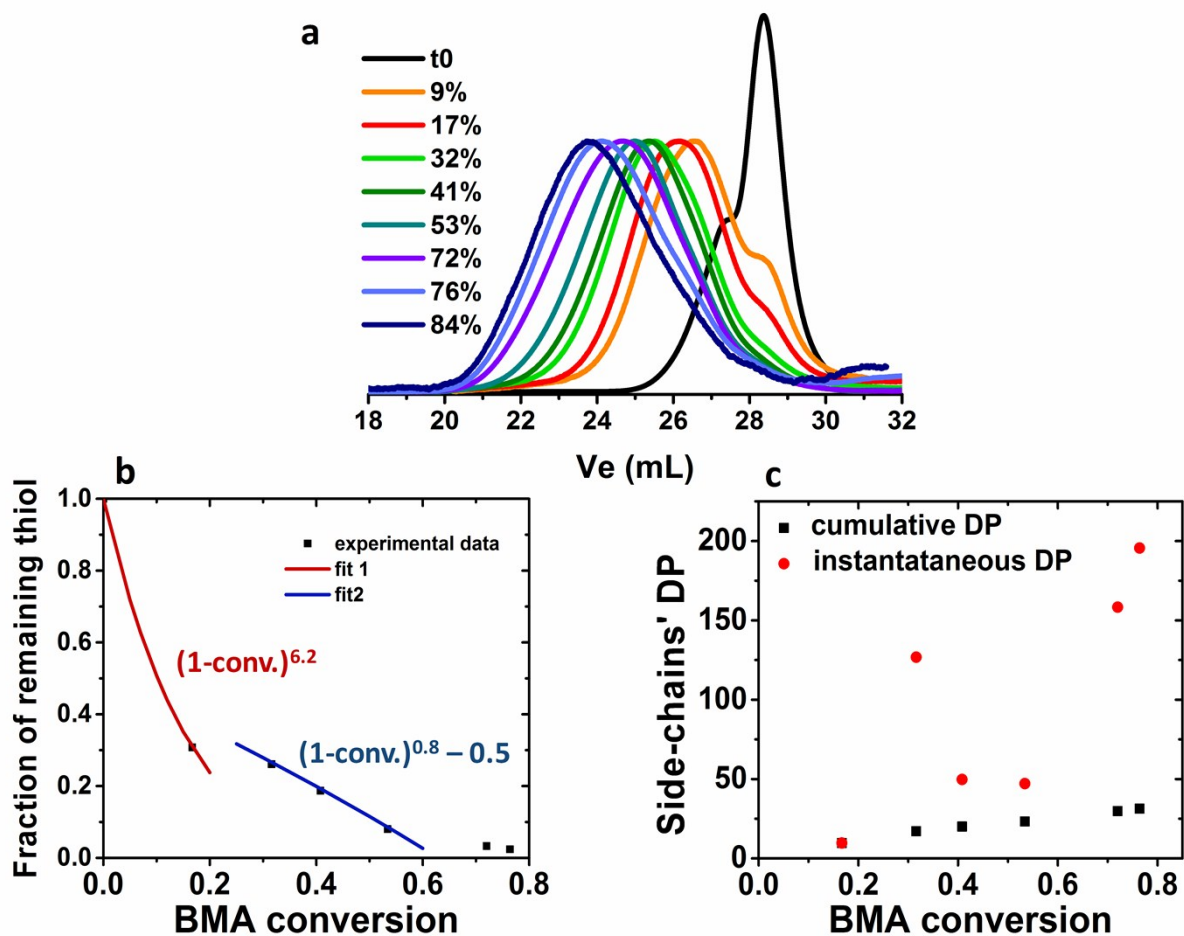


Figure S6: (a) Evolution of SEC traces of the crude reaction medium during the synthesis of brush **P7**. Polymerization conditions: $[\text{BMA}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 40/1/0.01$ in toluene 50 vol%, 24 h at 65 °C. (b) Plot of the remaining fraction of thiol vs BMA conversion during the synthesis of brush **P7** (c) Plot of side-chains cumulated DP (black squares) $([\text{M}]_0 - [\text{M}]_t)/([\text{S}]_0 - [\text{S}]_t)$ and instantaneous DP (red circles) $([\text{M}]_{t-1} - [\text{M}]_t)/([\text{S}]_{t-1} - [\text{S}]_t)$ vs BMA conversion during the synthesis of brush **P7**.

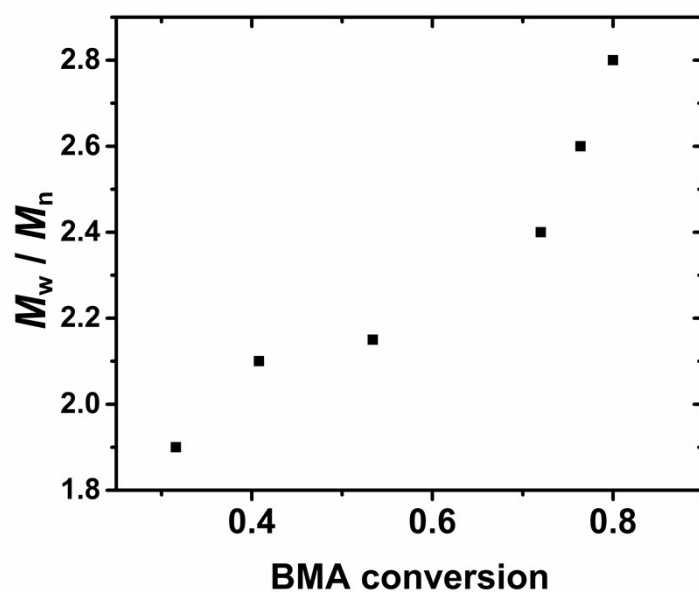


Figure S7: Plot of M_w/M_n vs conversion during brush **P7** polymerization. Polymerization conditions: $[\text{BMA}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 40/1/0.01$ in toluene 50 vol%, 24 h at 65 °C.

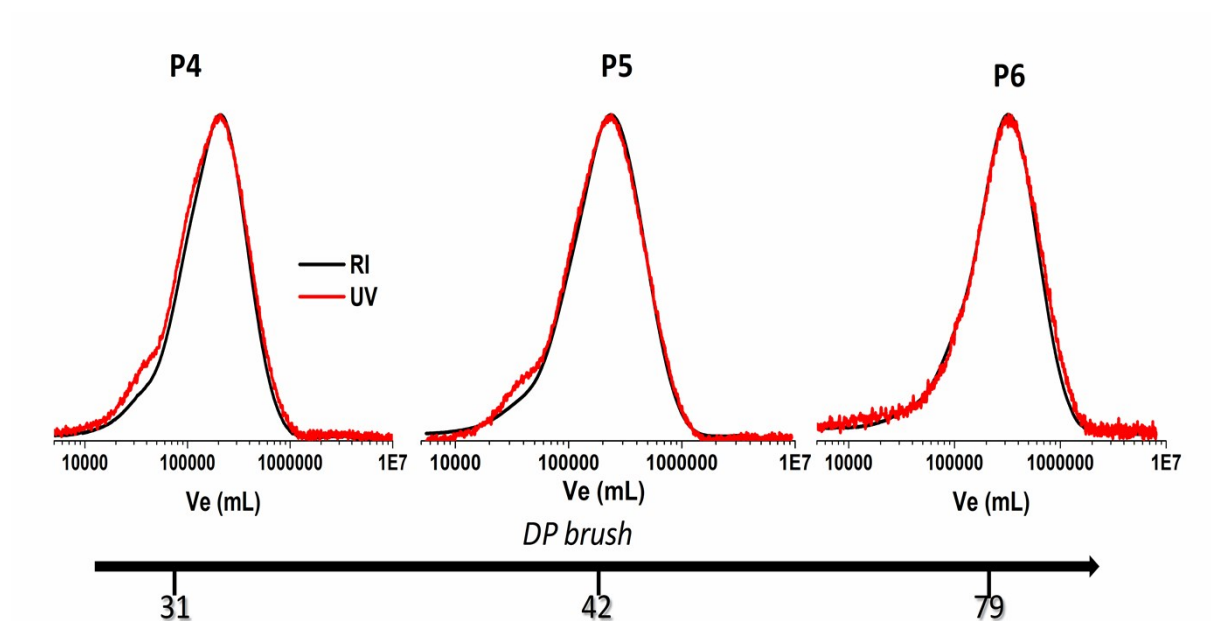


Figure S8: SEC traces of the crude reaction media at the end of the syntheses of brushes with different side-chain lengths.

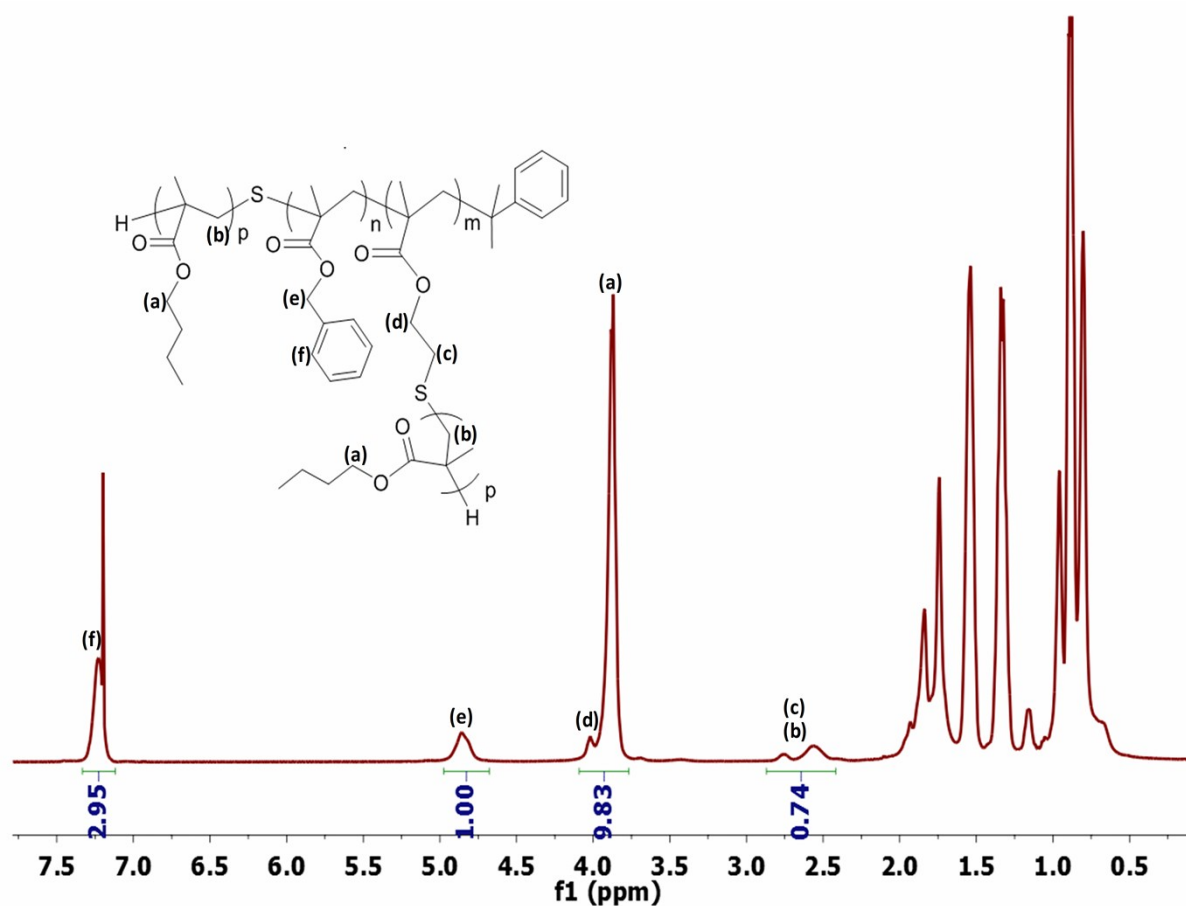


Figure S9: ^1H NMR spectrum of bottlebrush polymer **P4**. Polymerization conditions: $[\text{BMA}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 40/1/0.01$ in toluene 50 vol%, 24 h at 65 °C.

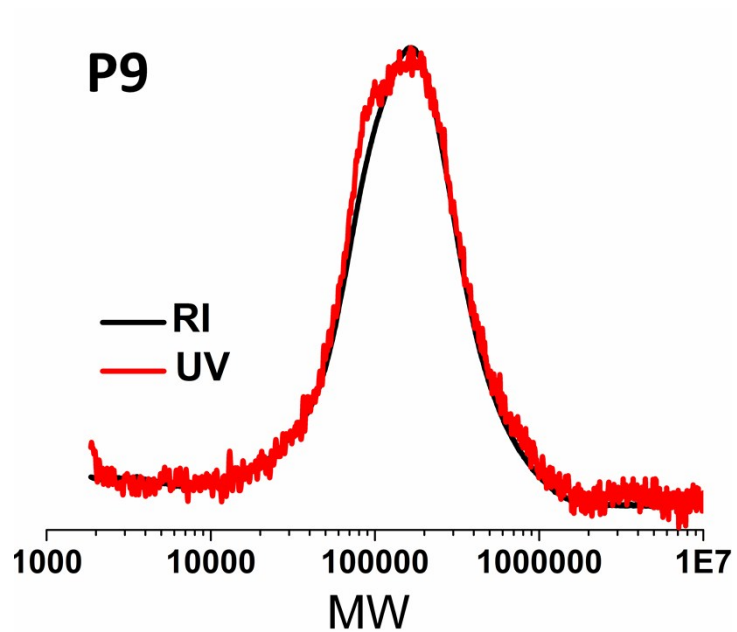


Figure S10: SEC traces of brush **P9** synthesized via a two-step process. Experimental conditions: $[\text{NIPAM}]_0/[\text{SH}]_0/[\text{AIBN}]_0 = 40/1/0.01$ in DMF 50 vol%, 1 h30 at 65 °C. SEC traces of the crude reaction medium.