

Supporting Information

A New Echelon of Precision Polypentenamers: Highly Isotactic Branching on Every Five Carbons

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Mosher acid synthesis for the determination of % enantiomeric excess:¹

To a flame dried 4mL vial equipped with a stir bar, 7.1 mg (84.4 μ mol, 1 eq.) of (S)-1 and 1.00 mL of DCM that was previously run through a plug of silica gel were added. To a separate dry vial, 0.0613 g (261.6 μ mol, 3.1 eq.) of (R)-(+) α -methoxy- α -trifluoromethylphenylacetic was added and dissolved in 0.32 mL of DCM. This solution was then added to (S)-1. To the solution, 0.0545g (261.4 μ mol, 3.1 eq.) of DCC and 0.0323g (261.4 μ mol, 3.1 eq.) of DMAP were added and capped. The mixture was allowed to stir for 3 hours. At this time, the mixture was run through a cotton plug and concentrated. The mixture was purified with a pipet column in 4:1 hexanes:EtOAc. The product was then concentrated and placed on the high-vacuum for 30 minutes. 1 H NMR (400 MHz, CDCl₃) δ (ppm): 7.51 (m, 2H), 7.40 (m, 3H), 6.15 (ddd, J = 5.7, 2.6, 1.6 Hz, 1H), 5.92 (m, 1H), 5.86 (m, 1H), 3.55 (s, 1H), 2.50 (dtdd, J = 15.8, 8.9, 4.2, 2.4 Hz, 1H), 2.37 (m, 2H), 1.95 (m, 1H)

(R, R)-DACH Phenyl Trost ligand:^{2,3}

The Trost ligand was synthesized following previous literature¹ with the following modifications. The brown solid was purified with 97.5:2.5 DCM:MeOH via column chromatography and recrystallized twice from MeCN to yield a white solid. 1 H NMR (600 MHz, CDCl₃) δ (ppm): 7.60 – 7.55 (m, 2H), 7.35 – 7.17 (m, 24H), 6.94 – 6.88 (m, 2H), 6.32 (d, J = 6.9 Hz, 2H), 3.82 – 3.73 (m, 2H), 1.90 – 1.82 (m, 2H), 1.69 – 1.61 (m, 2H), 1.28 – 1.15 (m, 2H), 0.98 (td, J = 15.1, 13.8, 6.2 Hz, 2H). 13 C NMR (150 MHz, CDCl₃) δ (ppm): 169.29, 140.88, 140.72, 137.84, 137.76, 137.70, 136.73, 136.58, 134.27, 133.95, 133.82, 130.16, 128.75, 128.60, 128.54, 128.51, 128.49, 128.45, 128.40, 53.89, 32.00, 24.66. 31 P NMR (243 MHz, CDCl₃) δ -9.78.

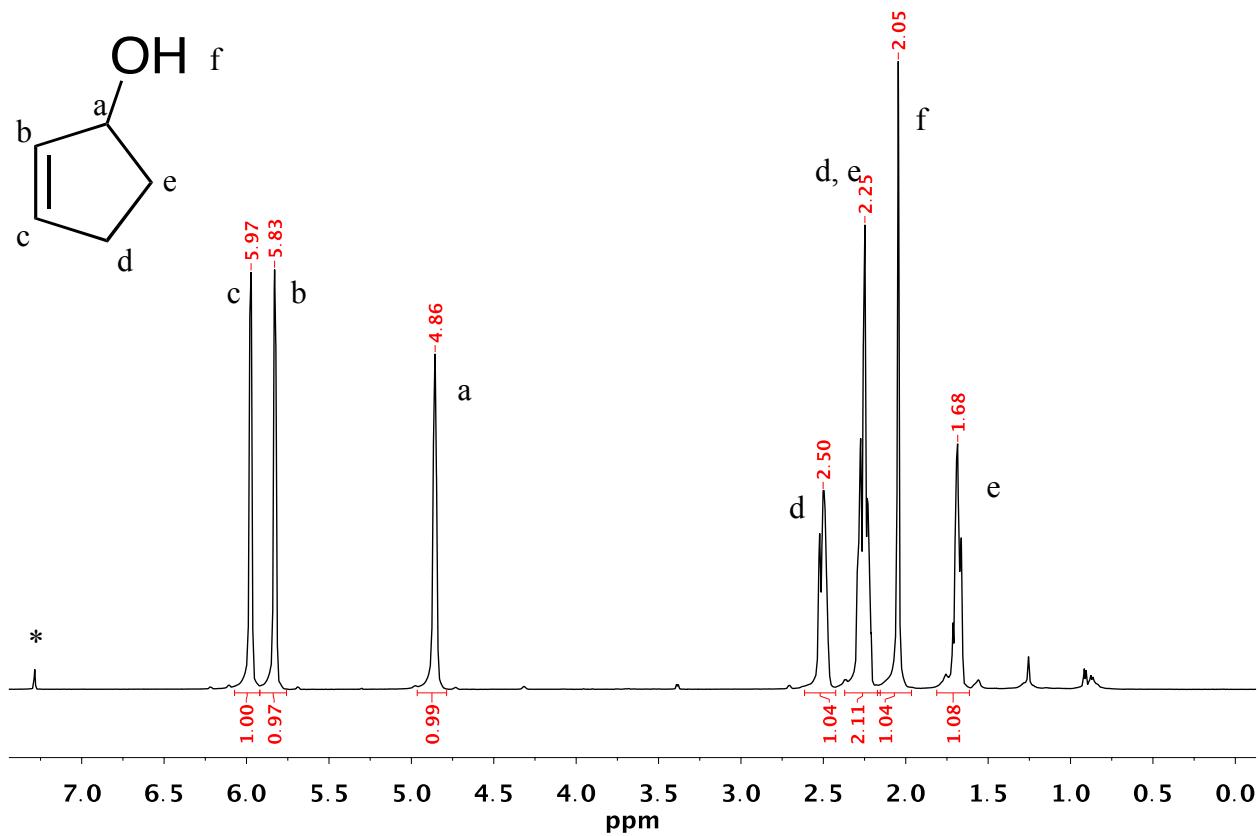


Figure S1. ¹H-NMR spectrum of *rac*-1, (in ³CDCl₃ 400 MHz)

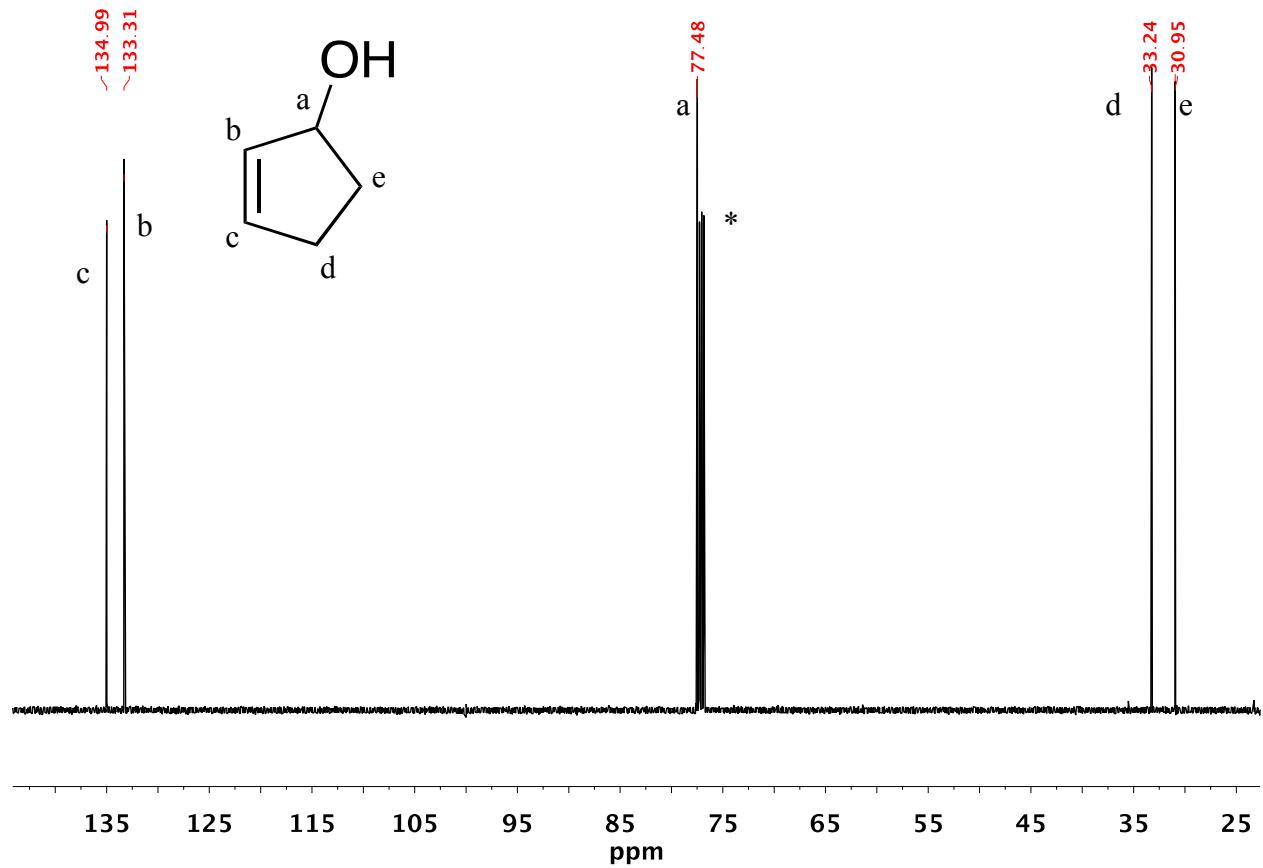


Figure S2. ^{13}C -NMR spectrum of *rac*-1, (in $^*\text{CDCl}_3$ 150 MHz)

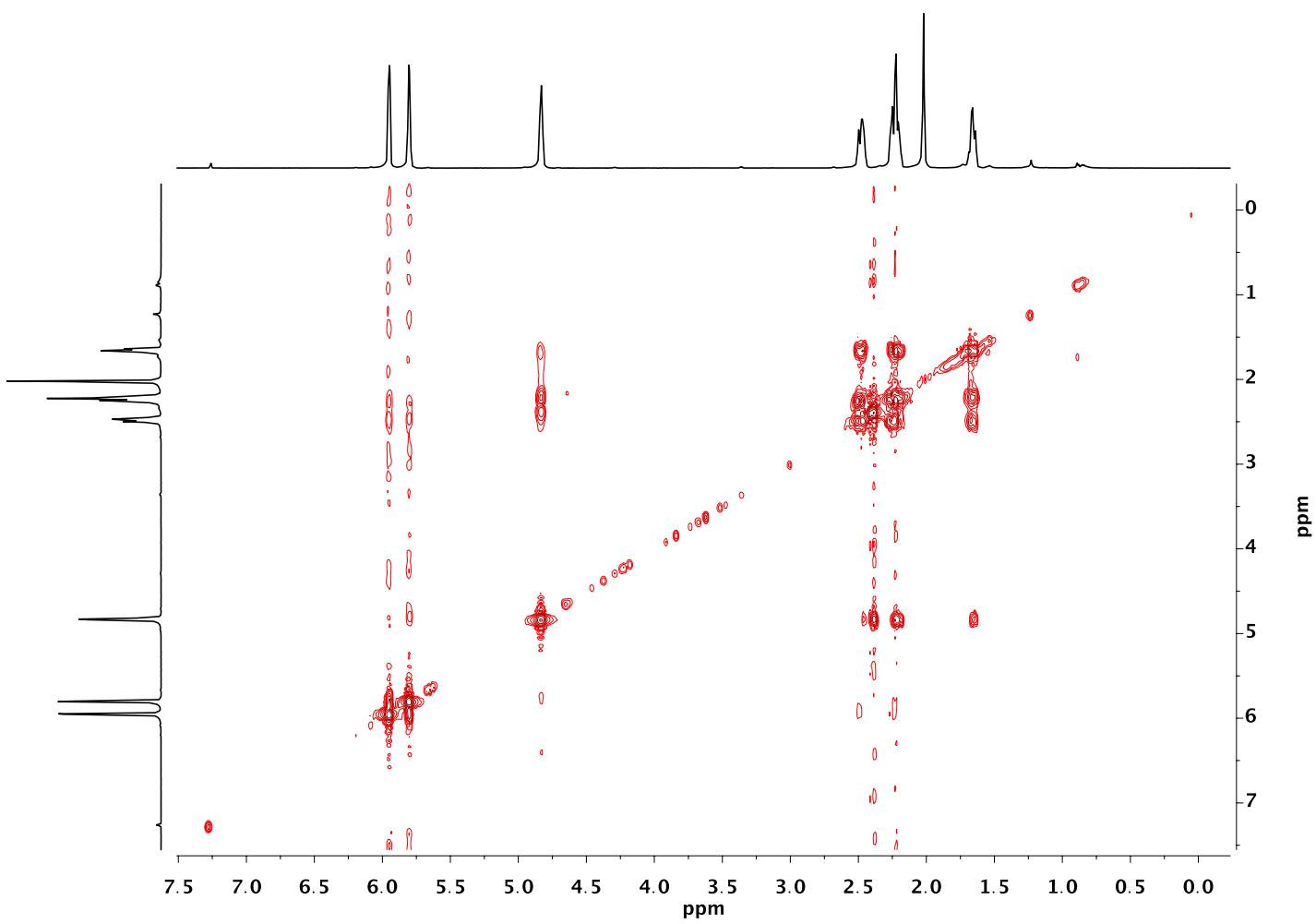


Figure S3. ^1H - ^1H COSY spectrum of *rac*-1, (in CDCl_3 600 MHz)

d:\desktop\...\27279_2-cyc5oh_gcms

11/3/2017 12:23:58 AM

2-cyclopentenol

27279_2-cyc5oh_gcms #261-271 RT: 4.07-4.14 AV: 11 SB: 61 3.81-3.97 , 4.31-4.55 NL: 1.27E7
T: + c Full ms [20.00-600.00]

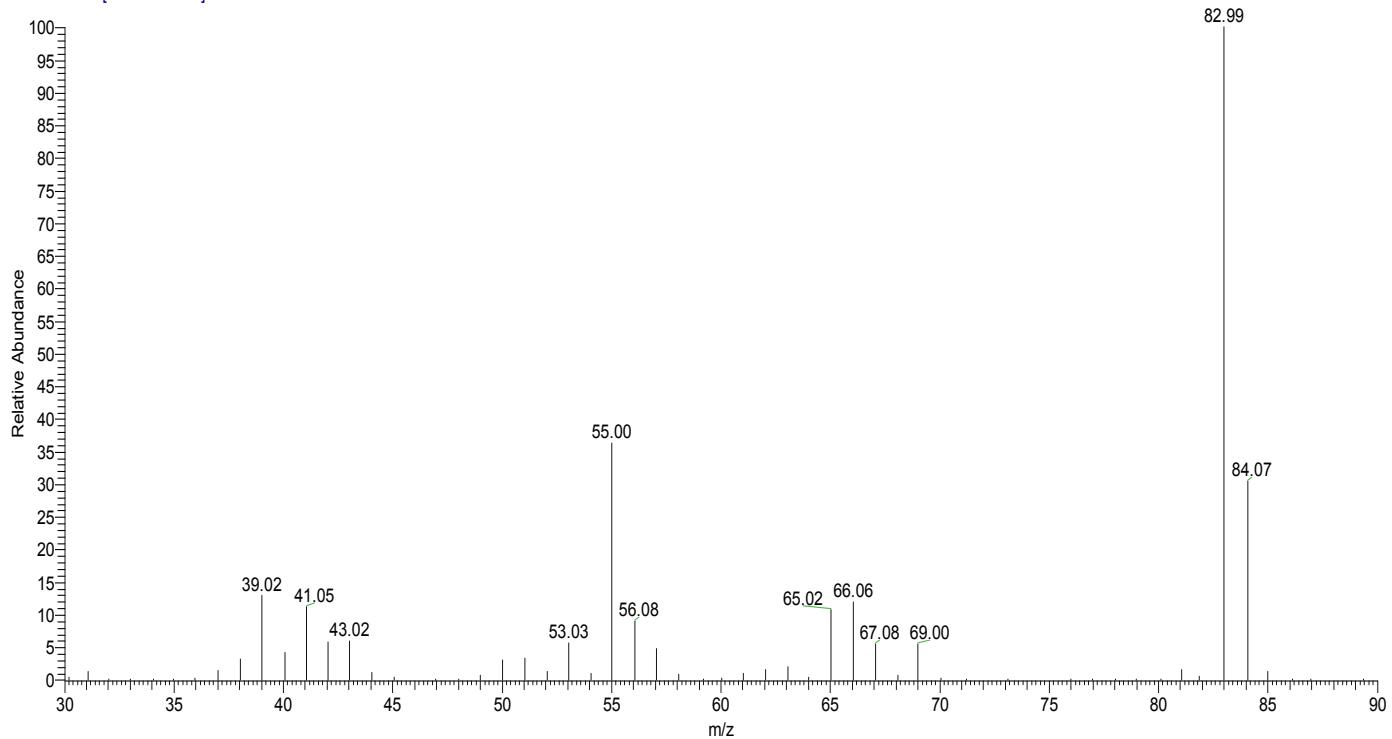


Figure S4. The mass spectrum at RT 4.08-4.12 min for *rac*-1. The molecular peak is observed at m/z 84.07.

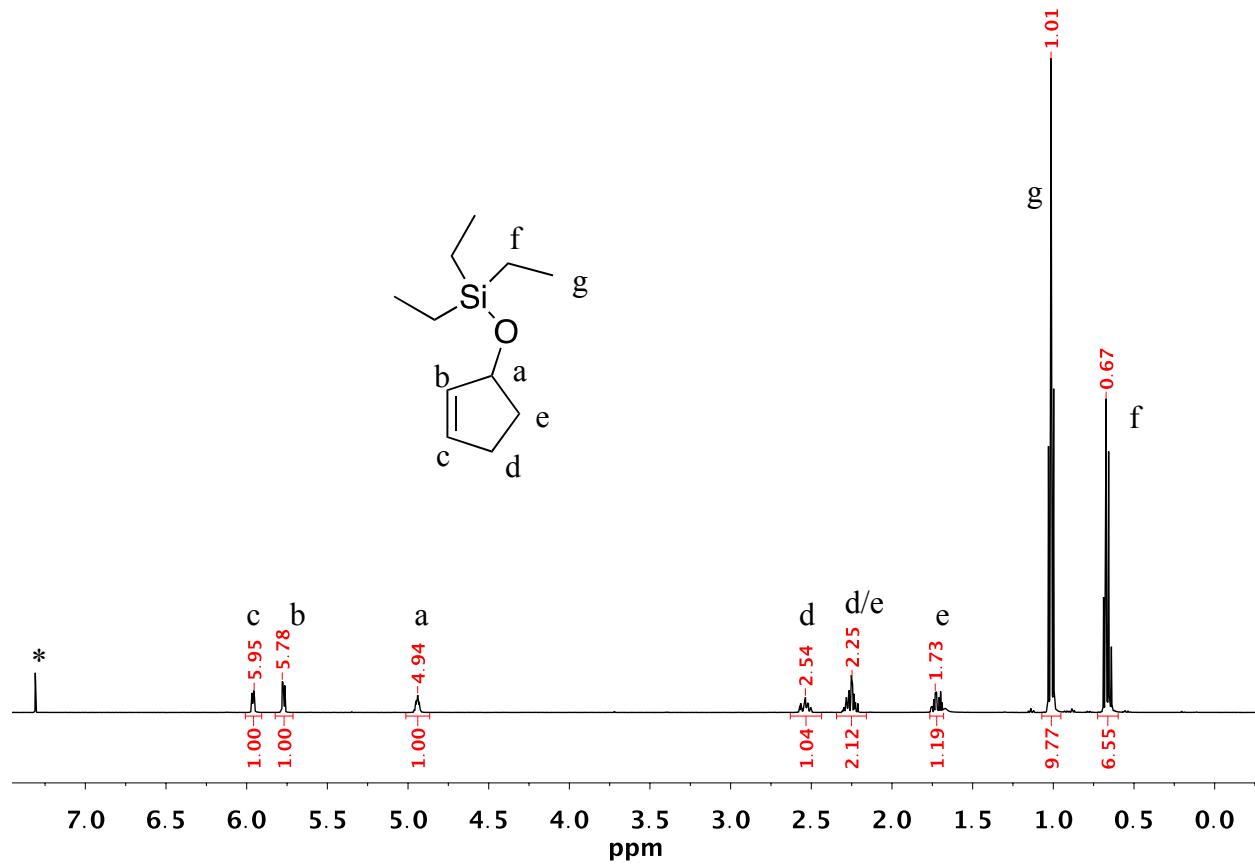


Figure S5. ^1H -NMR spectrum of *rac*-3, (in $^1\text{CDCl}_3$ 500 MHz).

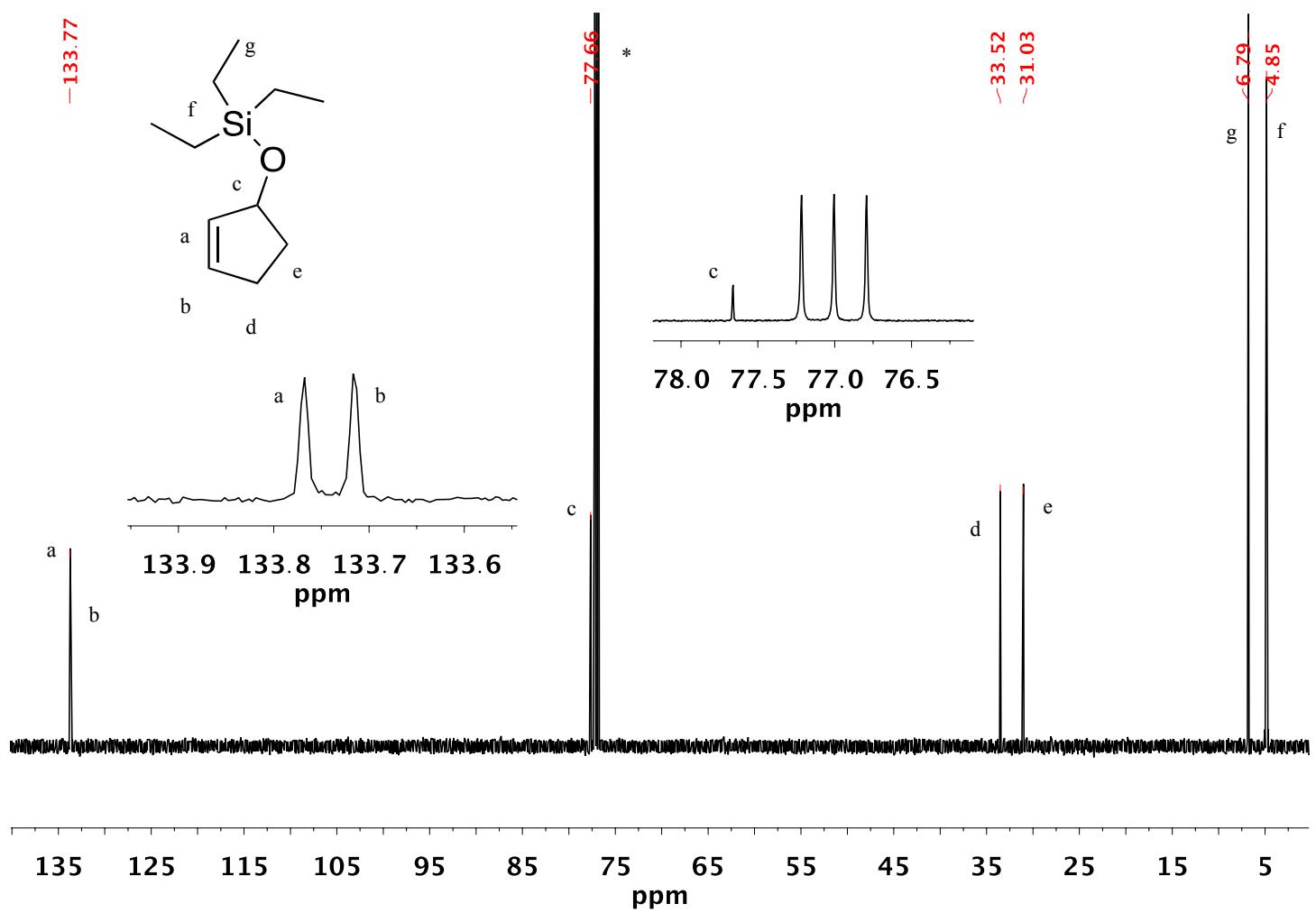


Figure S6. ^{13}C -NMR spectrum of *rac*-3, (in ${}^*\text{CDCl}_3$ 150 MHz)

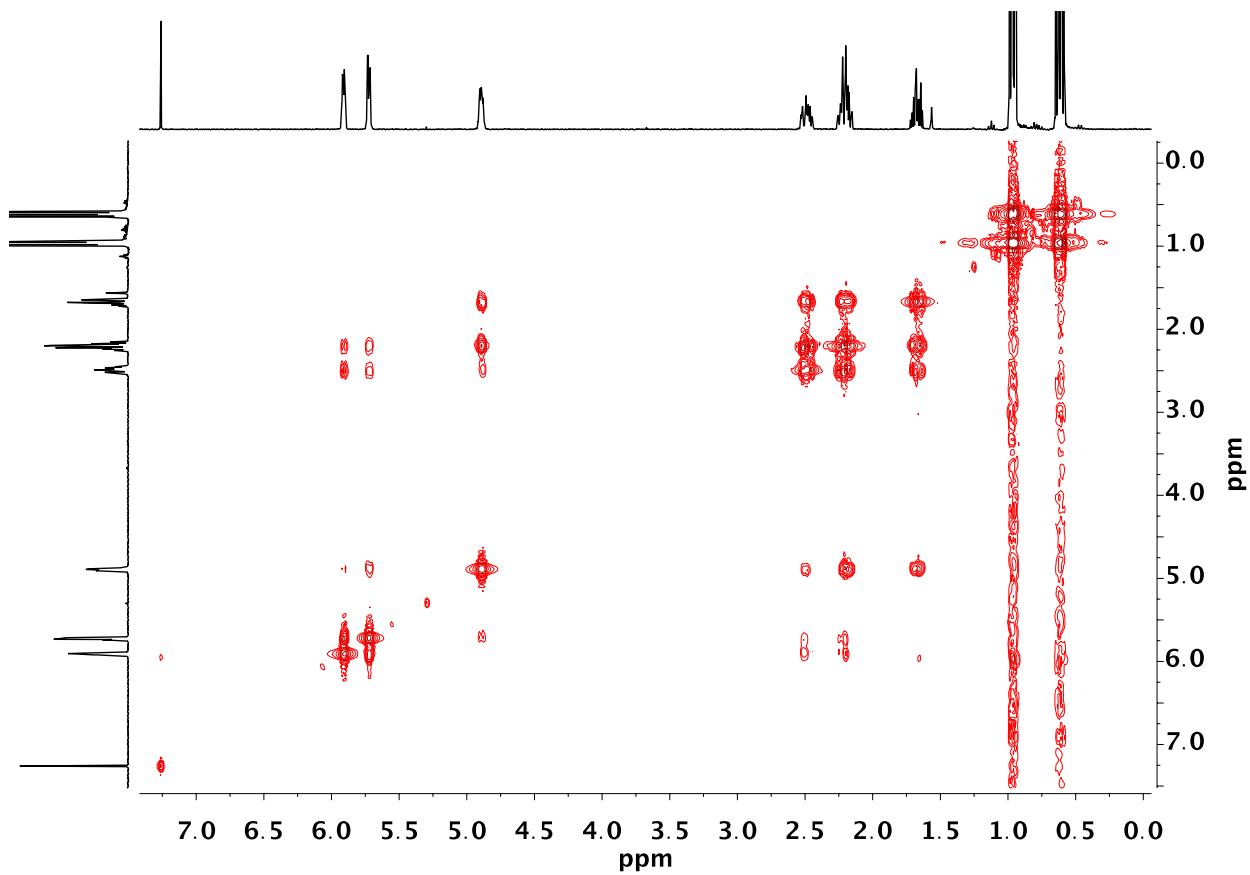


Figure S7. ^1H - ^1H COSY spectrum of *rac*-3, (in CDCl_3 500 MHz).

27278_3-3etsiocyc5ene_gcms #878-886 RT: 8.27-8.32 AV: 9 SB: 93 7.85-8.08 , 8.57-8.96 NL: 5.68E7
T: + c Full ms [20.00-600.00]

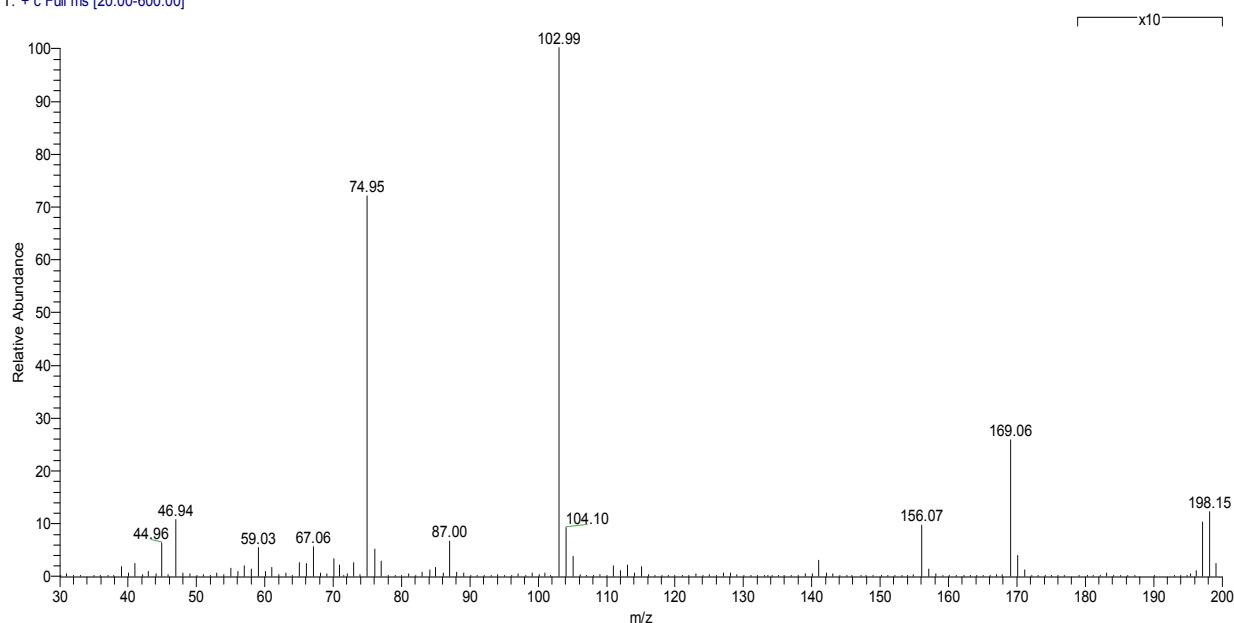


Figure S8: The mass spectrum at RT 8.27-8.32 min for the sample *rac*-3 The molecular peak is observed at m/z 198.15.

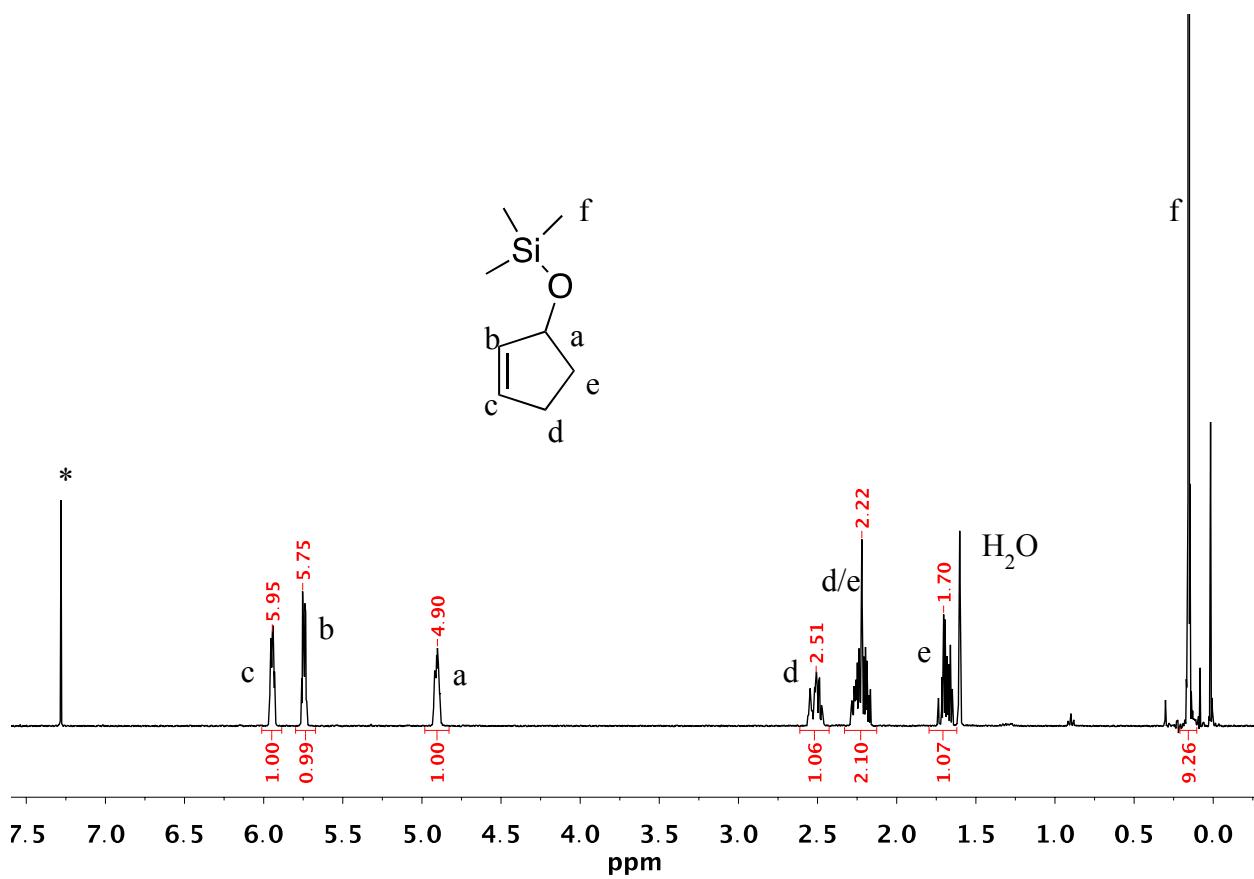


Figure S9. ^1H -NMR spectrum of *rac*-2, (in $^4\text{CDCl}_3$ 400 MHz)

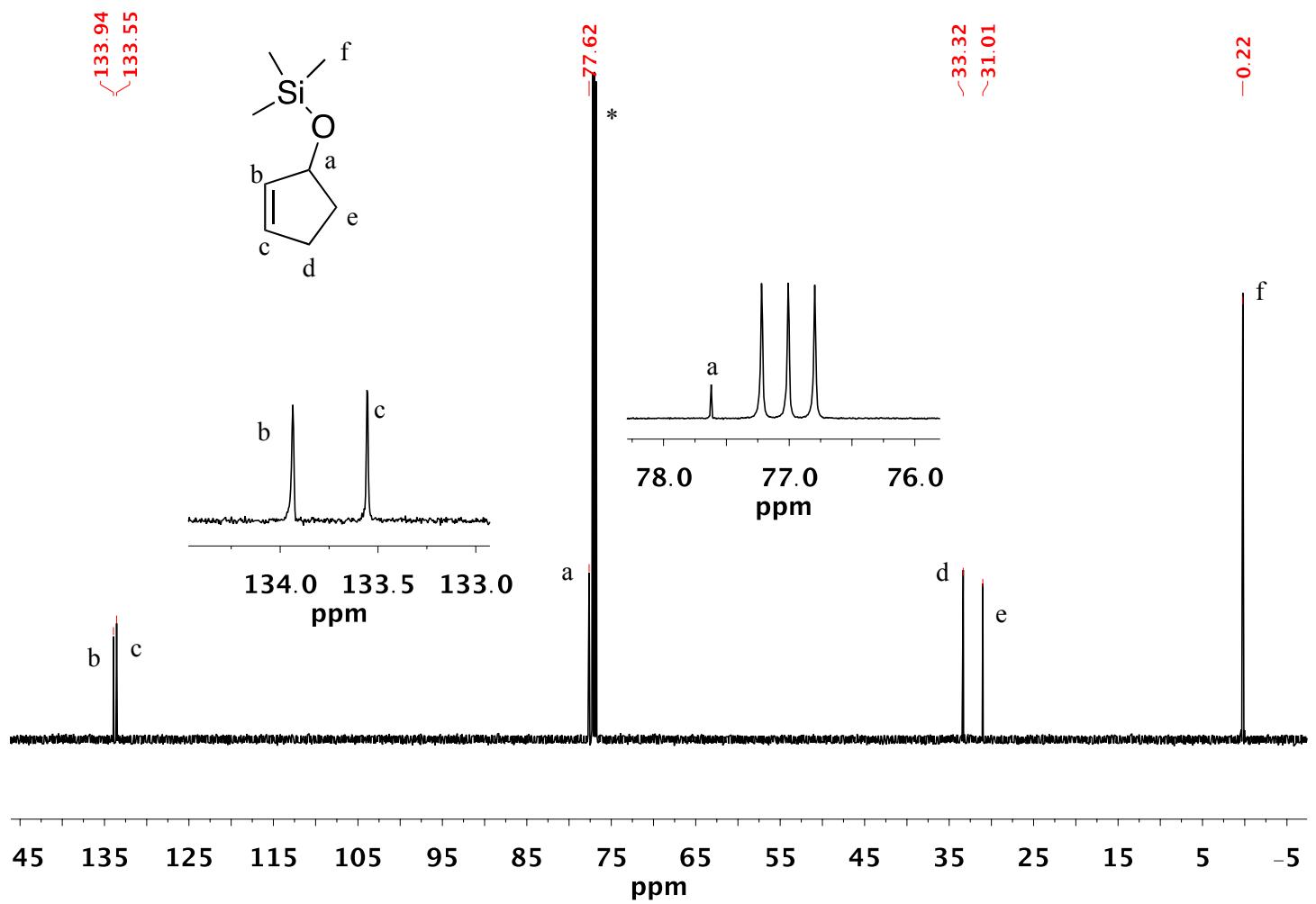


Figure S10. ^{13}C -NMR spectrum of *rac*-2, (in $^4\text{CDCl}_3$ 150 MHz)

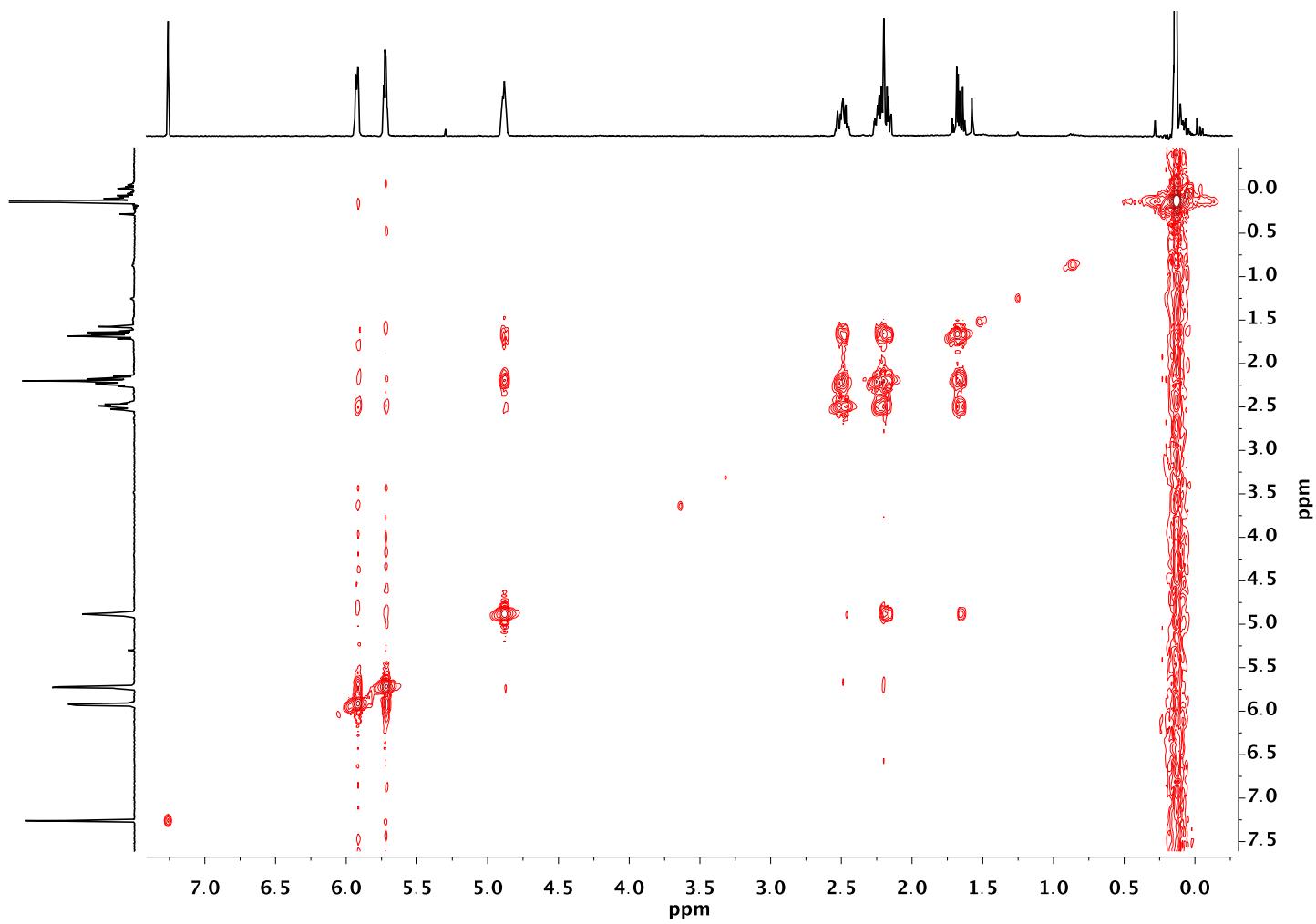


Figure S11. ^1H - ^1H COSY spectrum of *rac*-2, (in CDCl_3 600 MHz)

27277_3-3MeSiOcyc5ene_GCMS #486-493 RT: 5.60-5.65 AV: 8 SB: 74 5.34-5.53 , 5.76-6.06 NL: 1.65E8
T: + c Full ms [20.00-600.00]

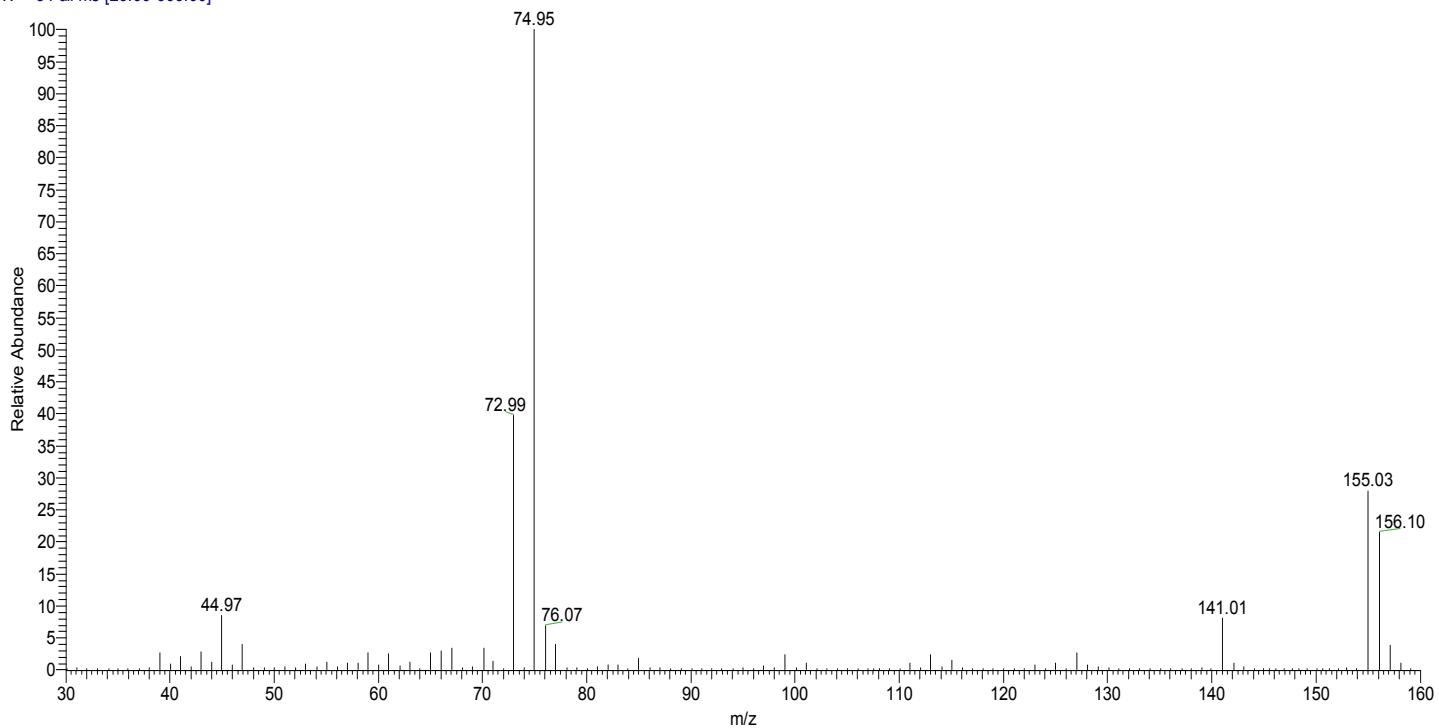


Figure S12. The mass spectrum at RT 5.60-5.65 min for the sample *rac*-2. The molecular peak is observed at m/z 156.10.

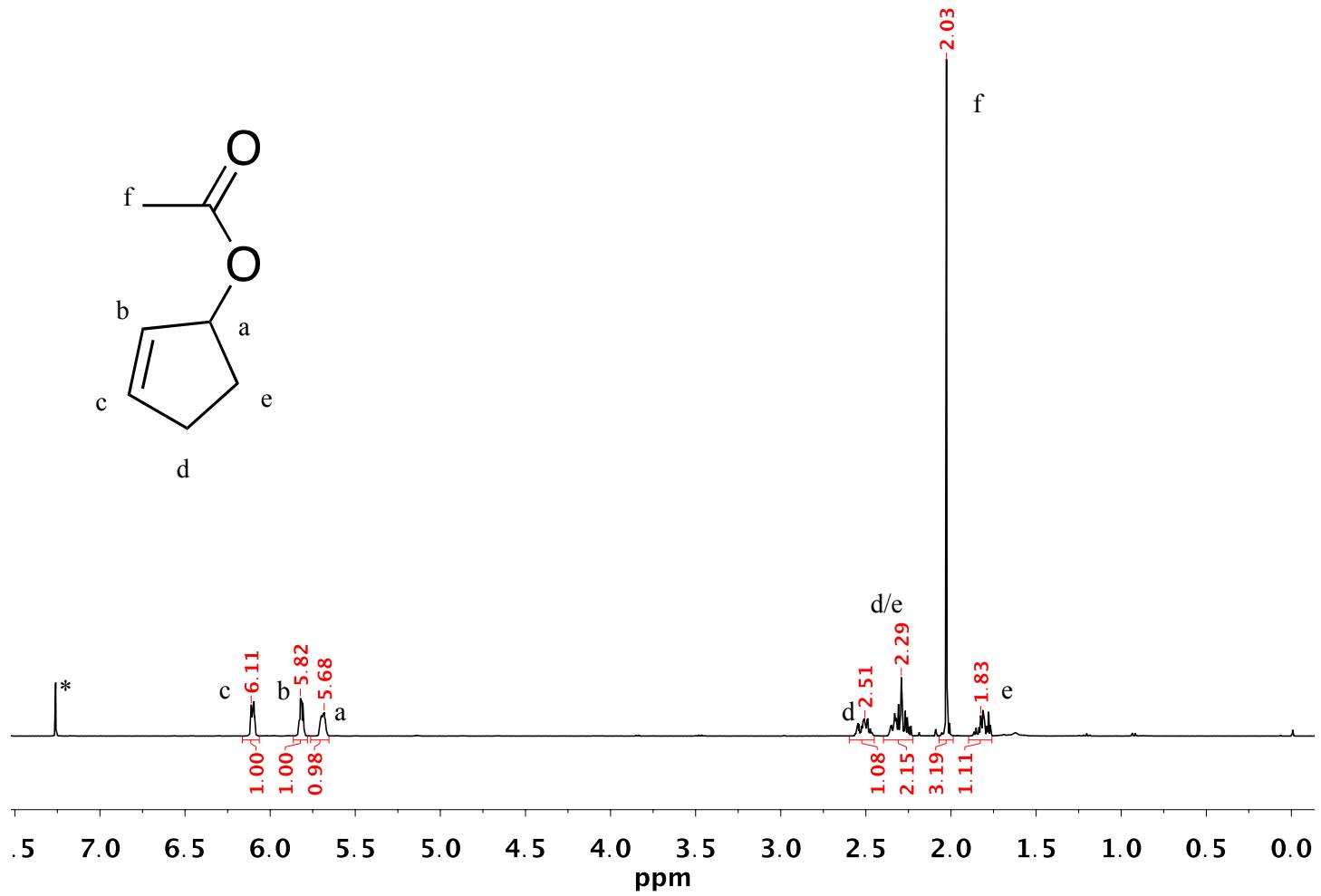


Figure S13. ¹H-NMR spectrum of *rac*-4, (in *CDCl₃ 400 MHz)

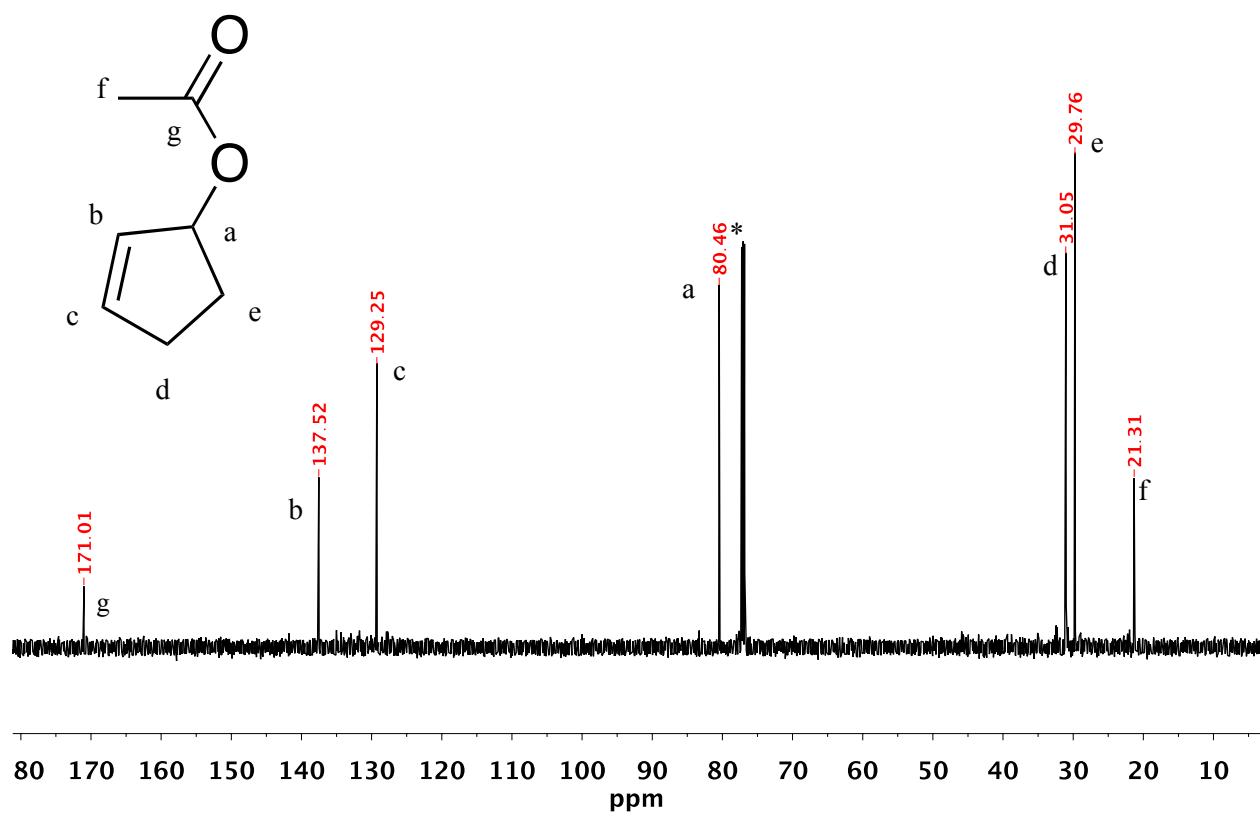


Figure S14. ^{13}C -NMR spectrum of *rac*-4, (in $^*\text{CDCl}_3$, 150 MHz)

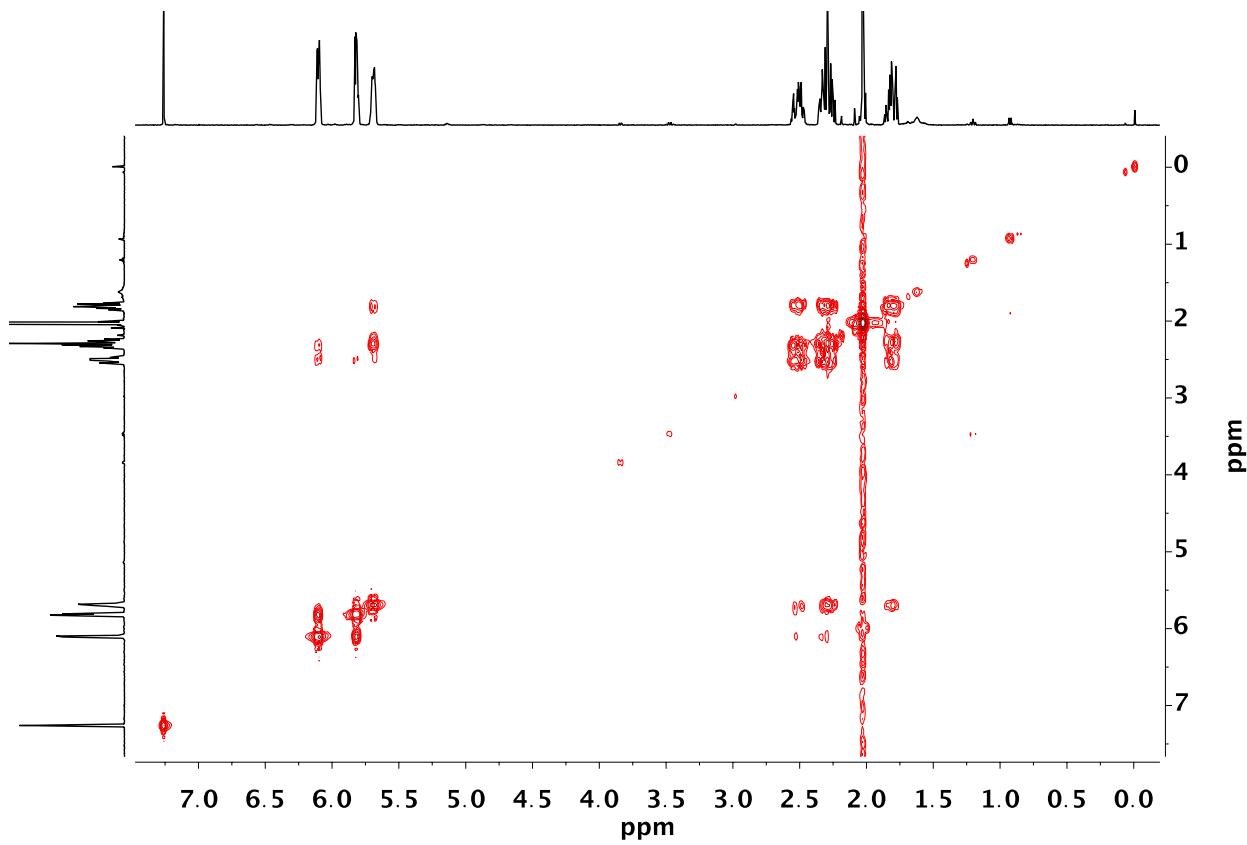


Figure S15. ^1H - ^1H COSY spectrum of *rac*-4, (in CDCl_3 600 MHz)

27281_cyc5ene-3-acetate_gcms #506-517 RT: 5.74-5.81 AV: 12 SB: 69 5.44-5.64 , 5.91-6.16 NL: 5.24E6
T: + c Full ms [20.00-600.00]

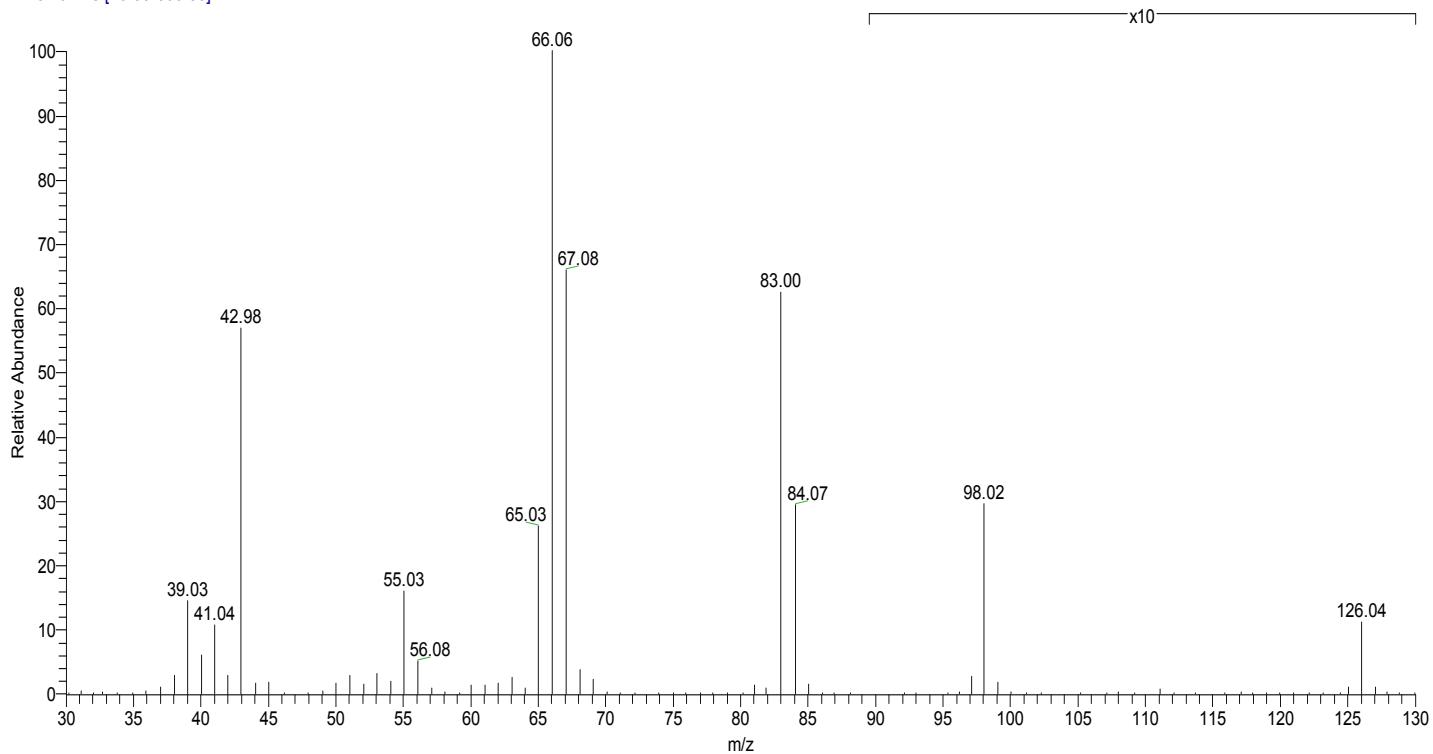


Figure S16: The mass spectrum at RT 5.74-5.81 min of *rac*-4. The molecular peak is observed at m/z 126.04.

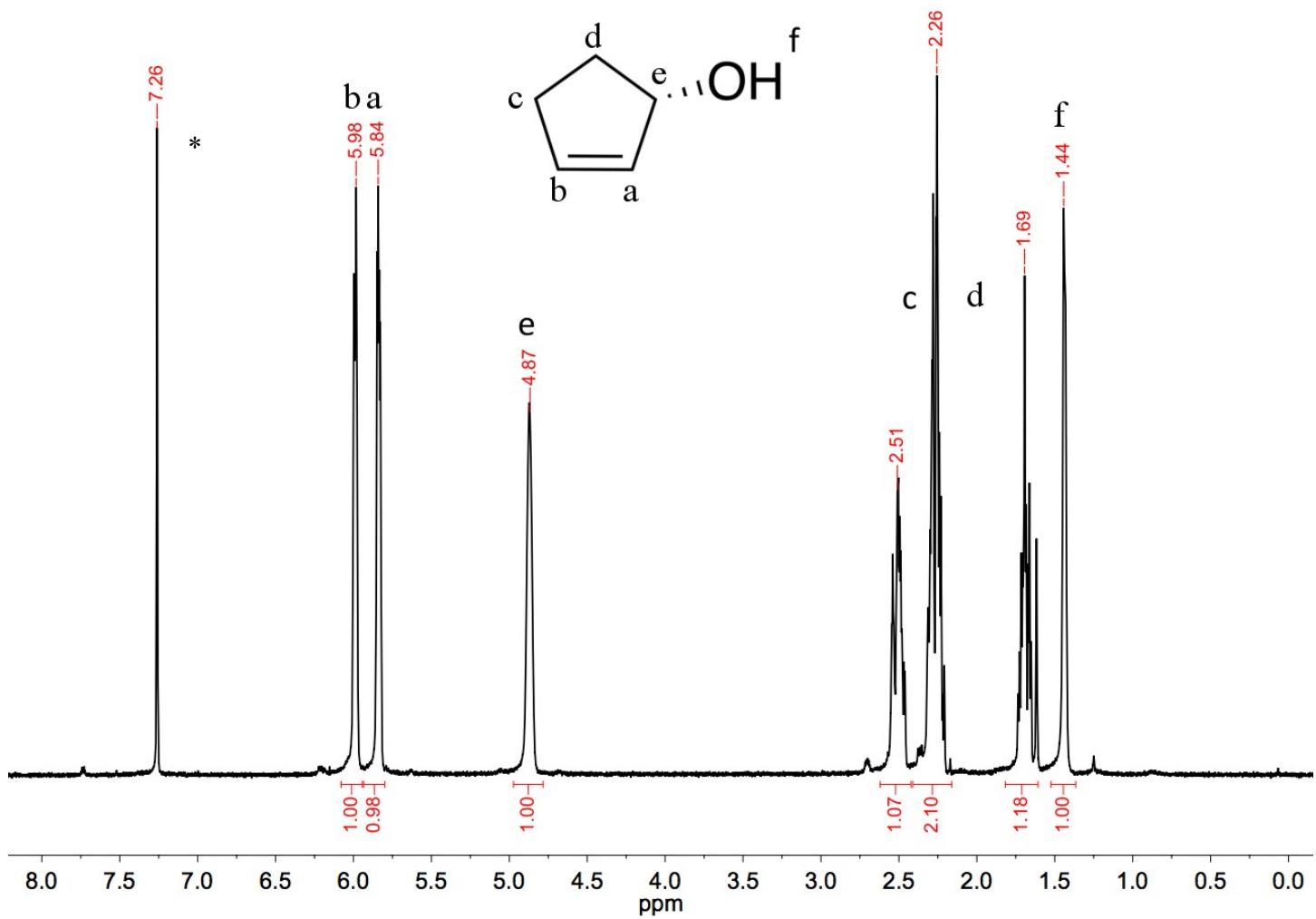


Figure S 17. ^1H -NMR spectrum of (S)-1 (in $^4\text{CDCl}_3$, 400 MHz)

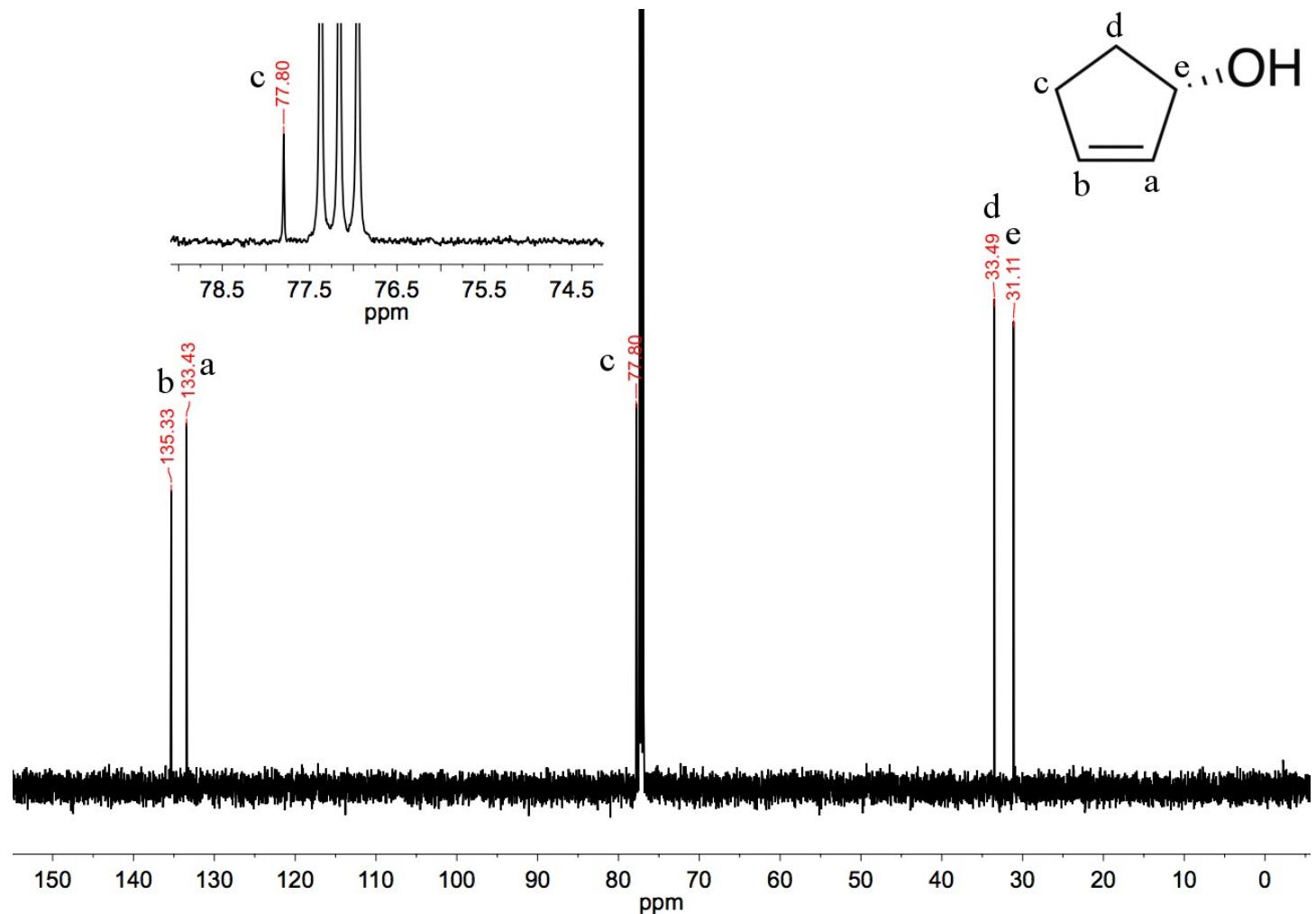


Figure S 18. ^{13}C NMR spectrum of (S)-1 (in CDCl_3 , 150 MHz)

27283_s-2-cyc5oh_gcms #262-268 RT: 4.08-4.12 AV: 7 SB: 115 7.83-8.14 , 8.52-8.98 NL: 2.22E6
T: + c Full ms [20.00-600.00]

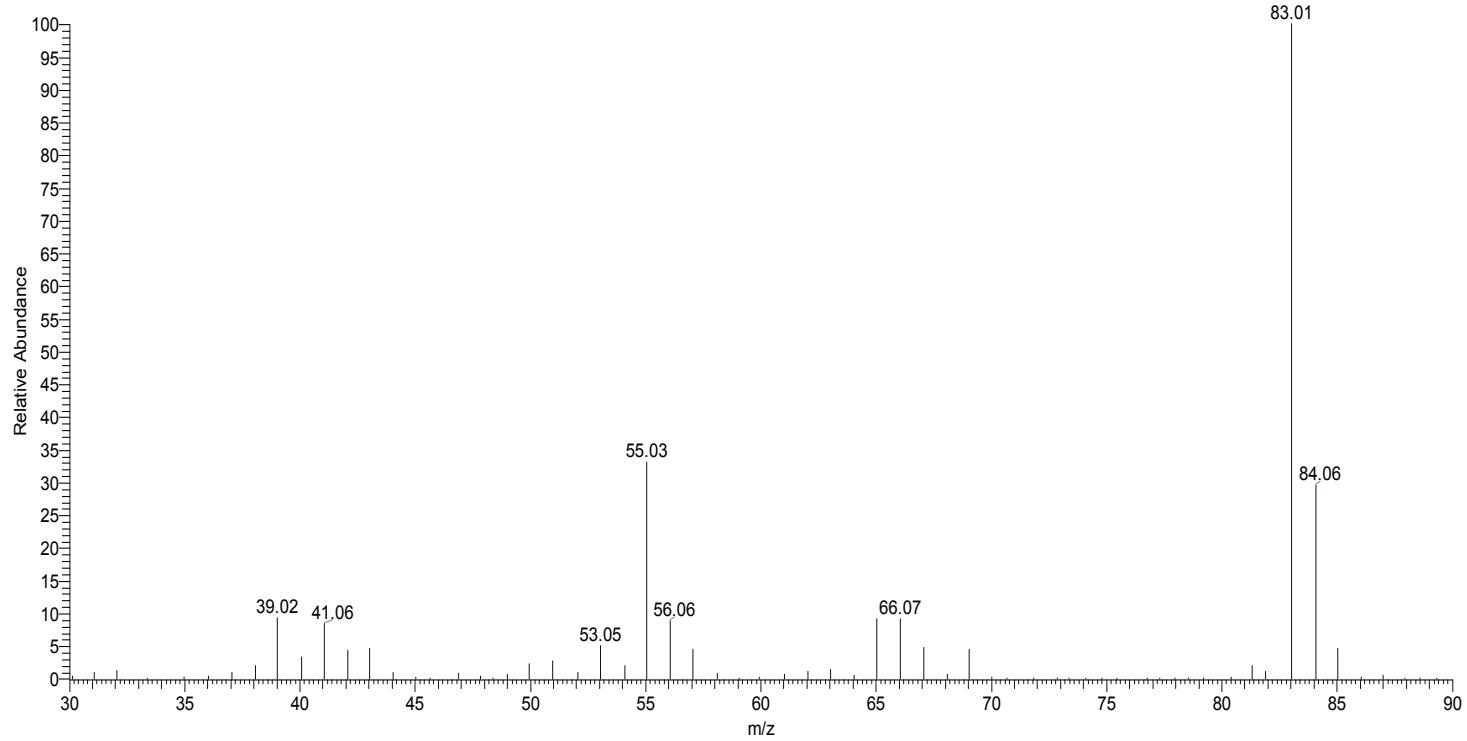


Figure S19: The mass spectrum at RT 4.08-4.12 min for (*S*)-1. The molecular peak is observed at m/z 84.06.

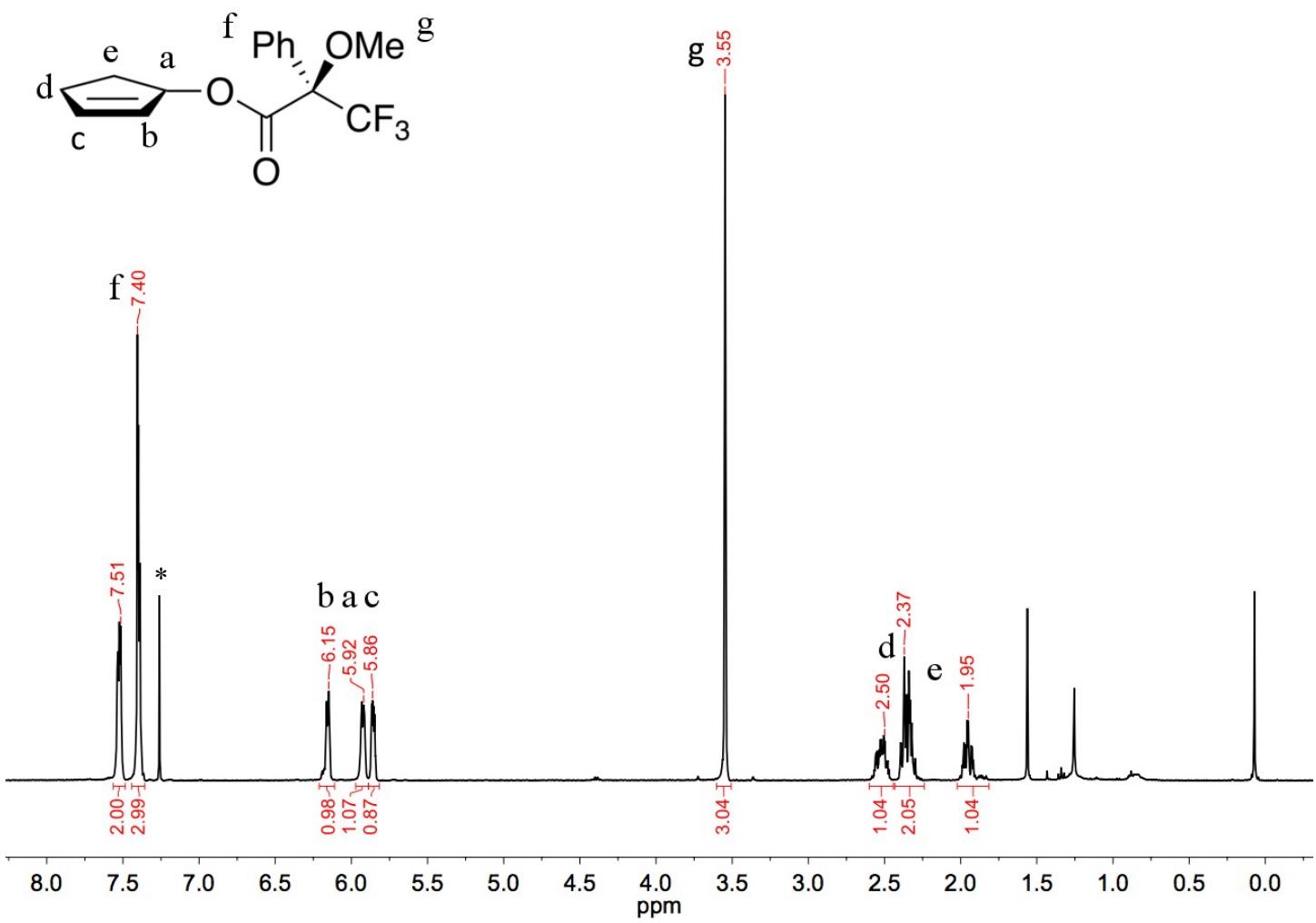


Figure S 20. ¹H-NMR spectrum of (S)-2-MTPACP ester (in *CDCl₃, 400 MHz).

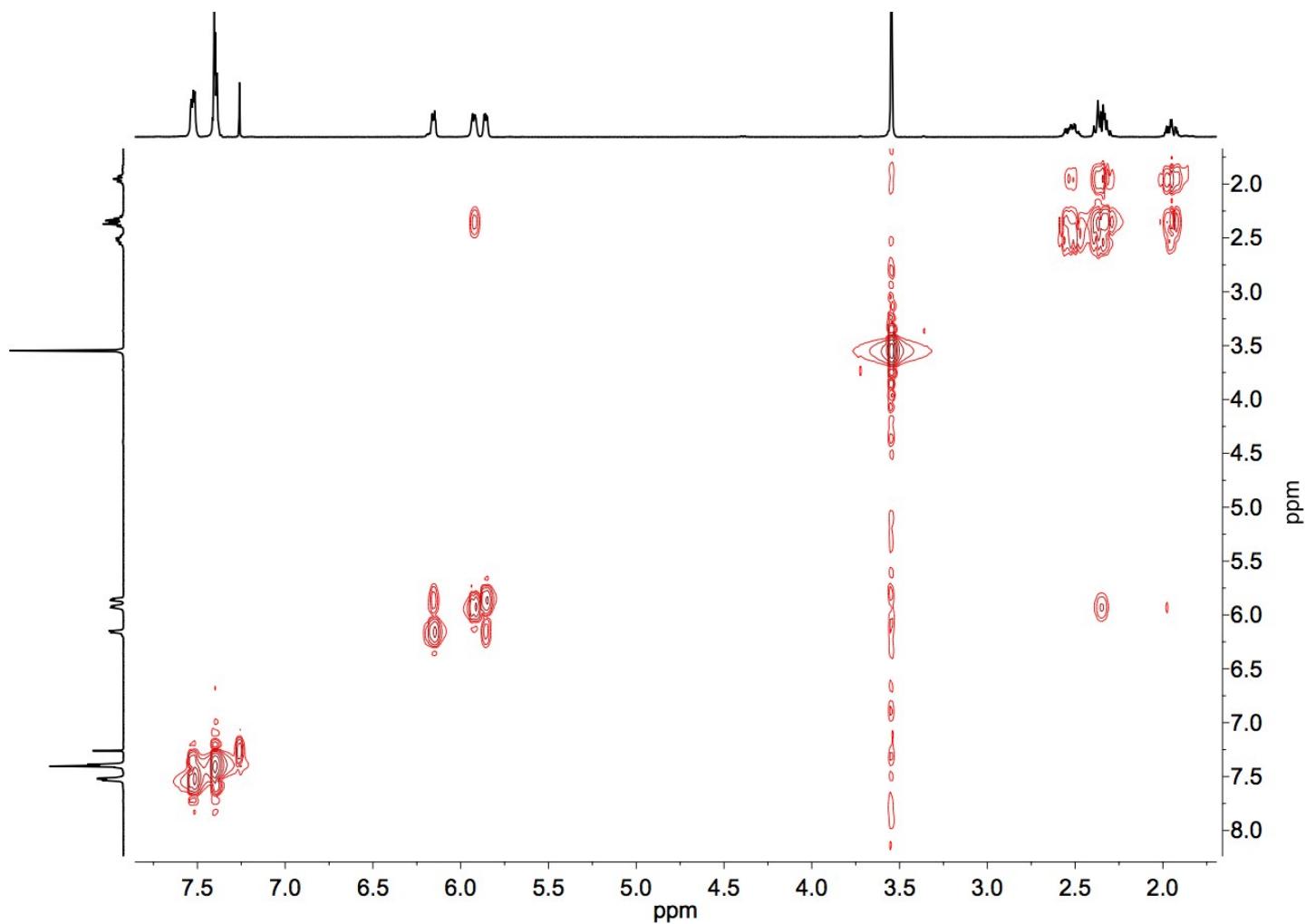


Figure S21. ^1H - ^1H COSY spectrum of (S)-2-MTPACP ester, (in CDCl_3 , 400 MHz,).

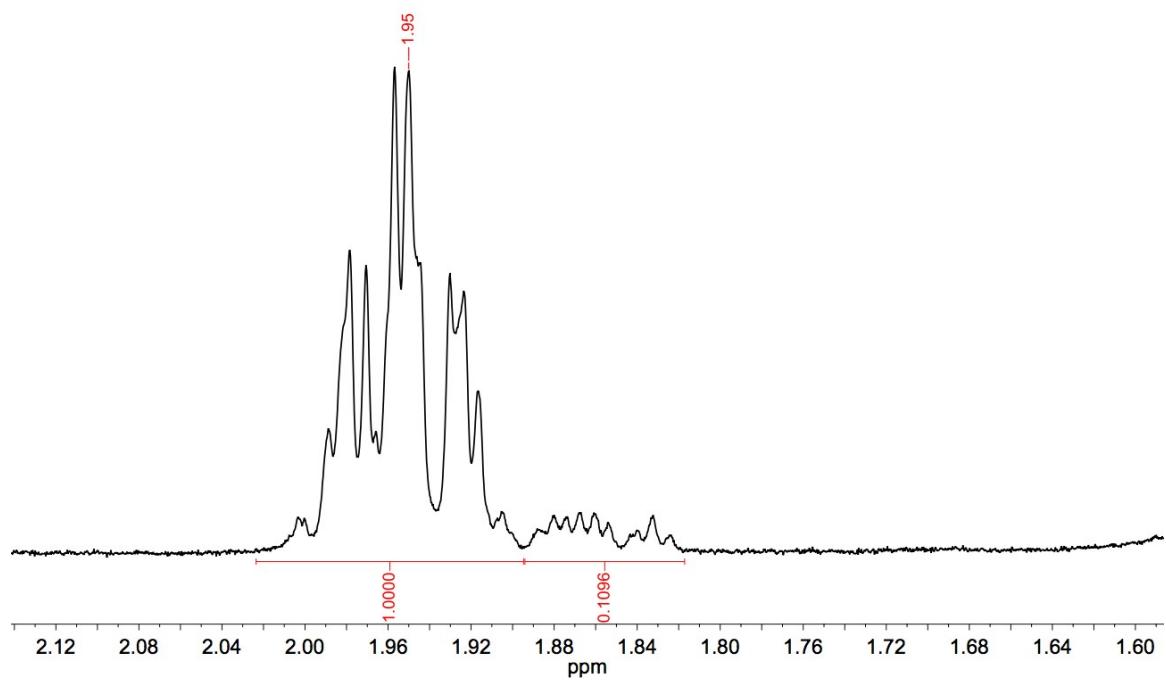


Figure S 22. ^1H -NMR of (*S*)-2-MTPACP ester (in CDCl_3 , 400 MHz,).

$$\% ee = \frac{\text{area major}}{(\text{area major} + \text{area minor})} * 100 = 90.1 \% ee$$

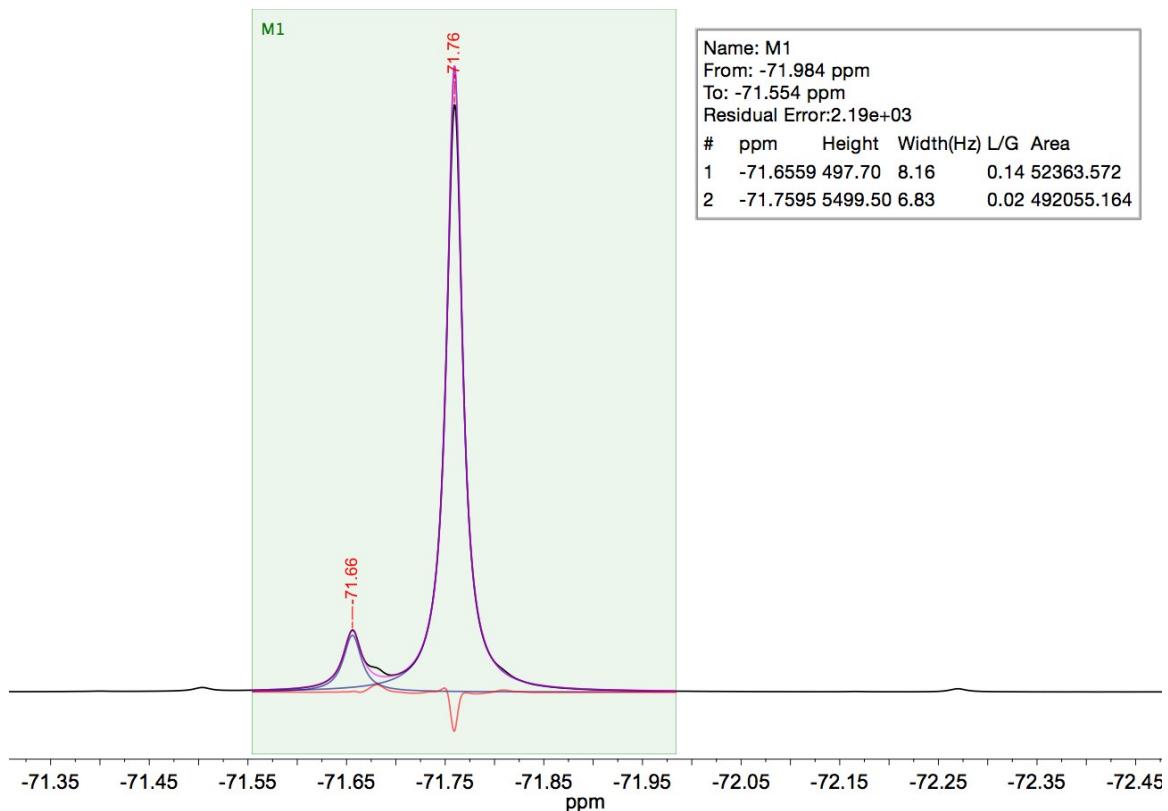


Figure S 23. ^{19}F NMR spectrum of (*S*)-2-MTPACP ester (in CDCl_3 , 376.5 MHz.). Deconvolution of the peaks were used to determine the integration of the two peaks. $\% \text{ ee} = \frac{\text{area major}}{(\text{area major} + \text{area minor})} * 100 = 90.4 \% \text{ ee}$

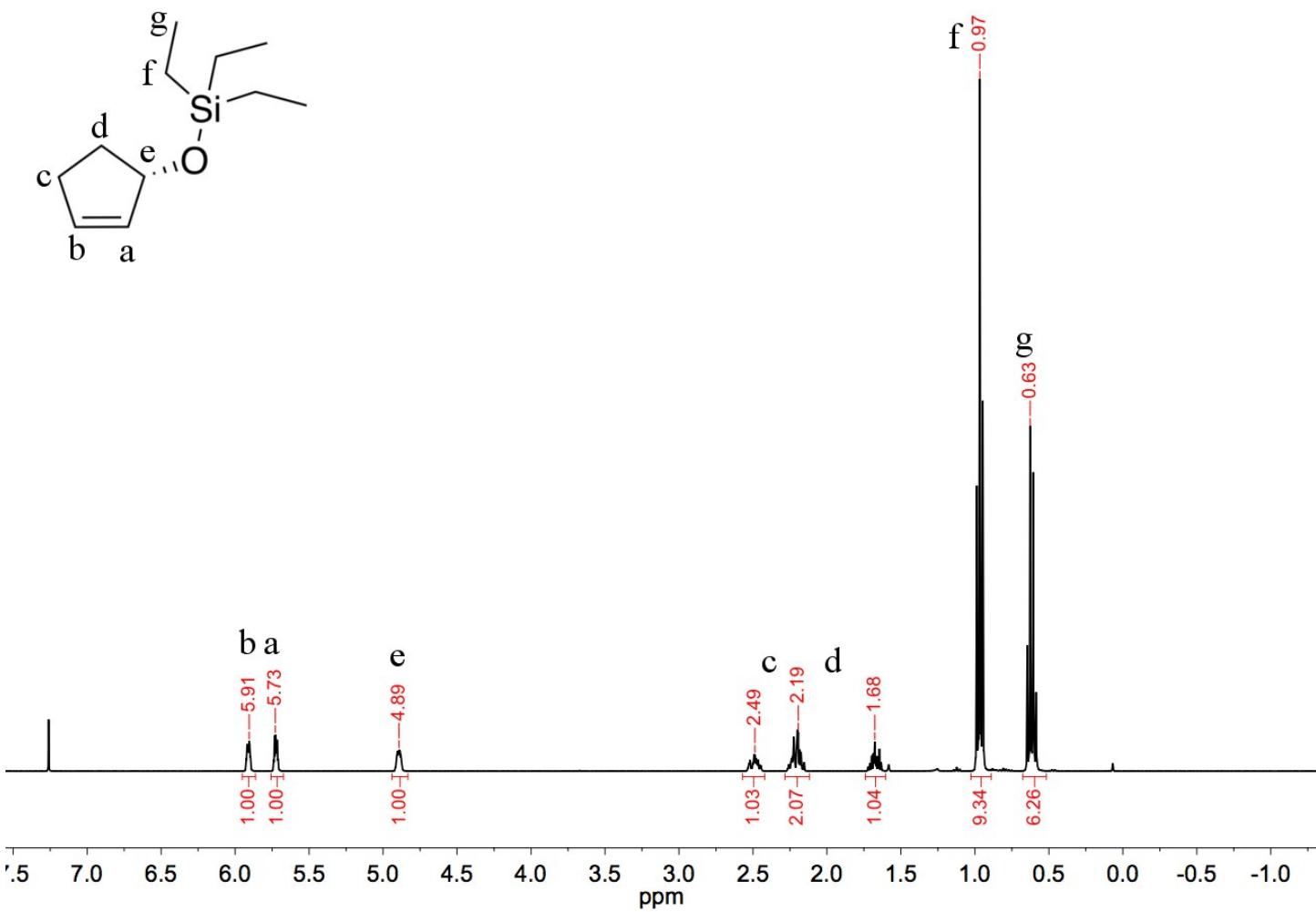


Figure S 24. ^1H NMR spectrum of (S)-3 (in ${}^*\text{CDCl}_3$, 400 MHz).

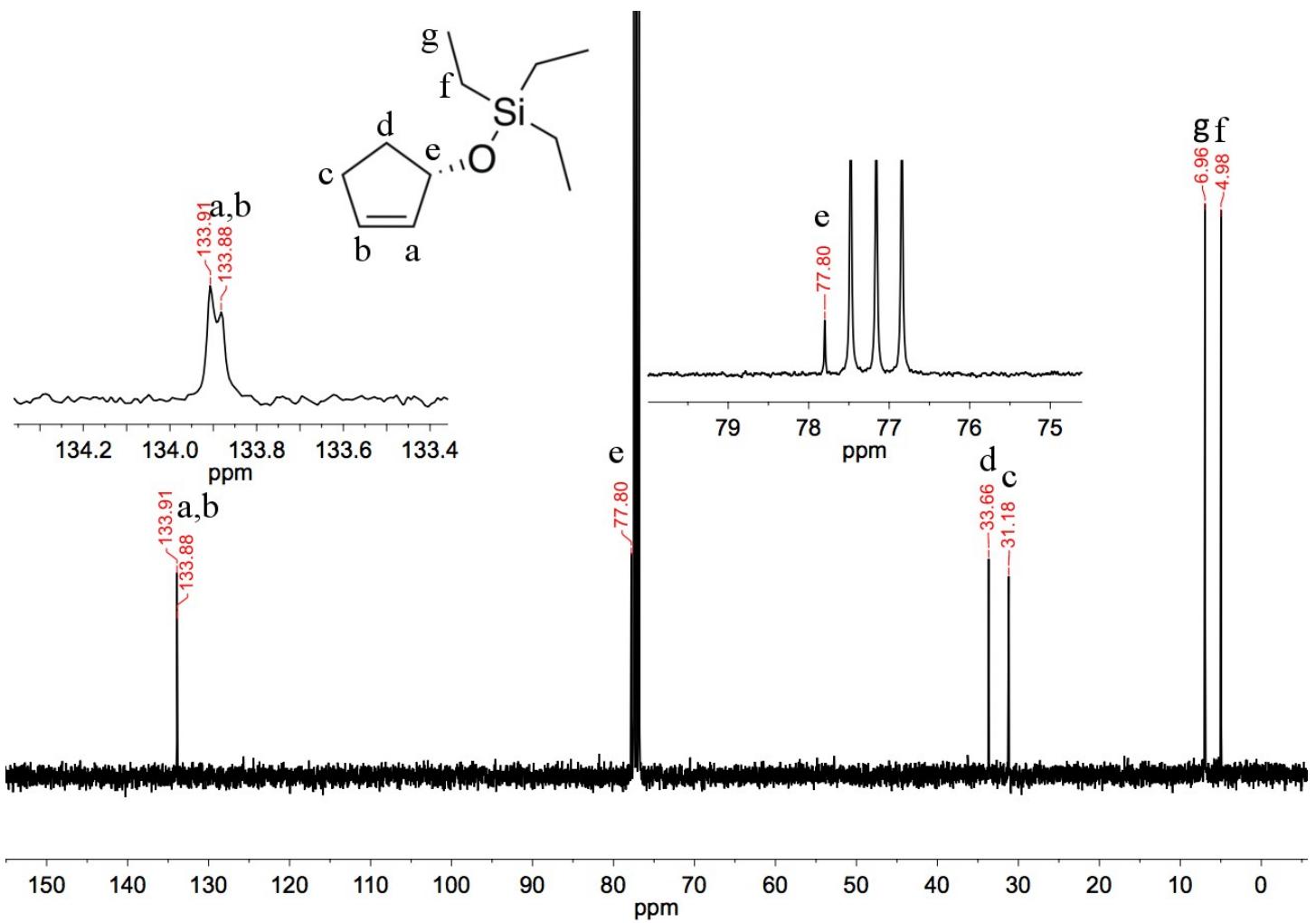


Figure S 25. ^{13}C NMR of (S)-3 (in CDCl_3 , 100 MHz)

27282_s-3-3etsiocyc5ene_gcms #878-889 RT: 8.27-8.34 AV: 12 SB: 116 7.83-8.14 , 8.52-8.98 NL: 4.06E7
T: + c Full ms [20.00-600.00]

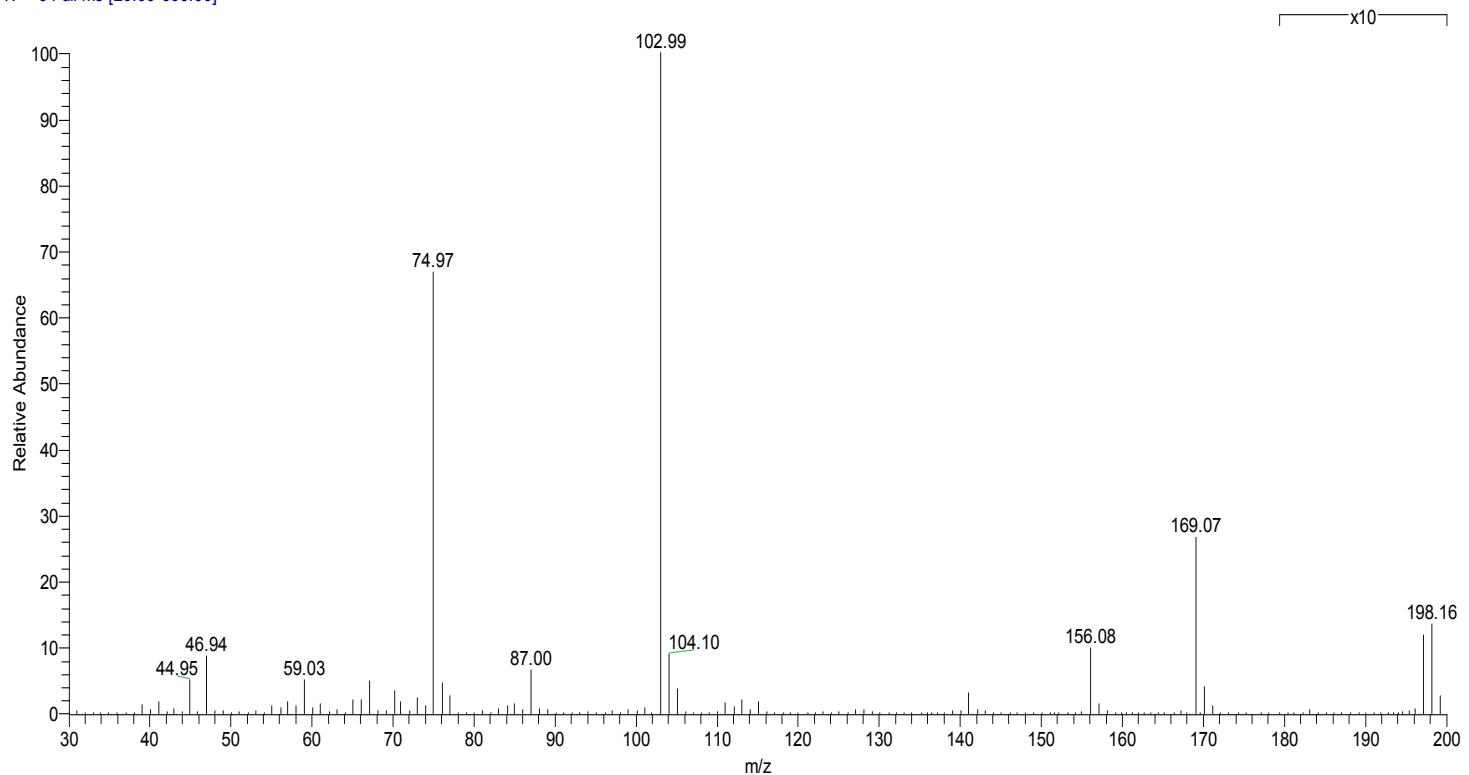


Figure S 26: The mass spectrum at RT 8.27-8.34 min of (S)-3. The molecular peak is observed at m/z 198.16.

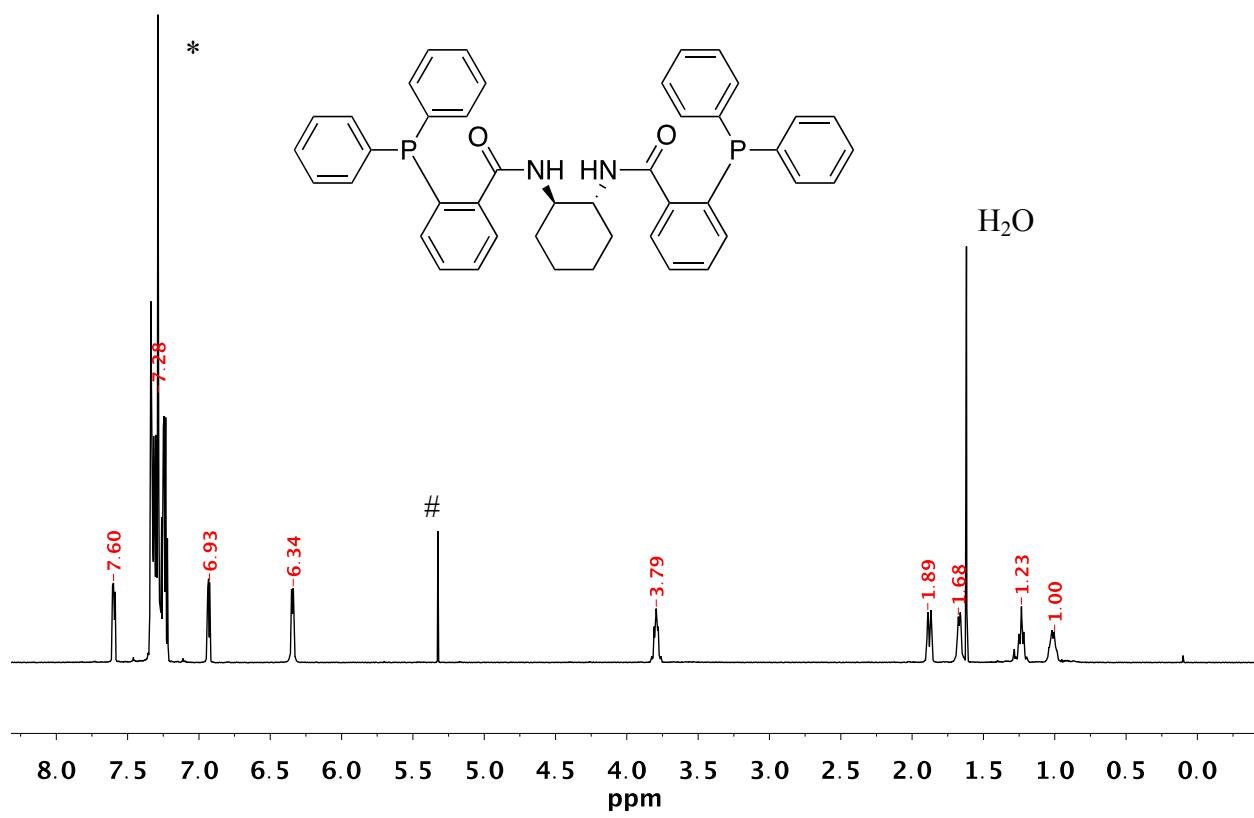


Figure S 27. ¹H NMR spectrum of Trost ligand (in *CDCl₃, 600 MHz) # DCM

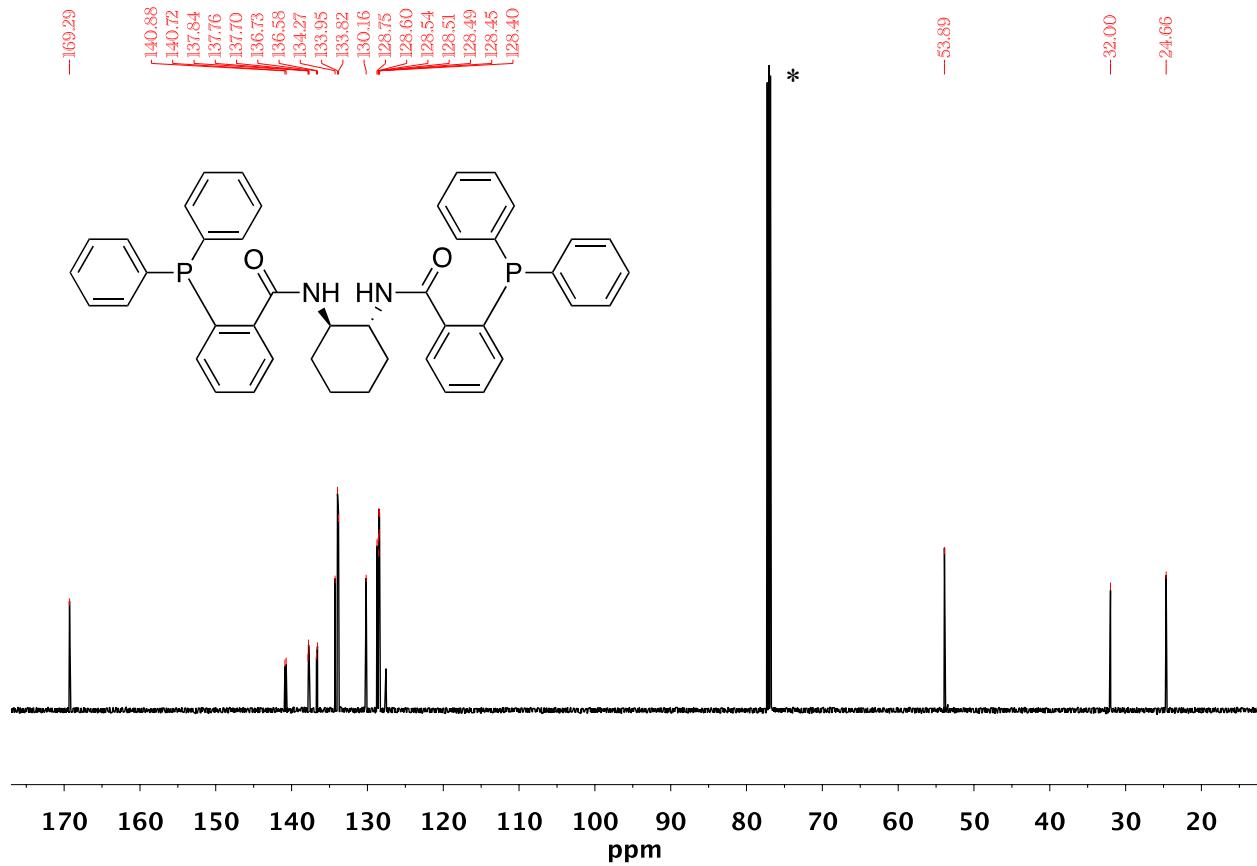


Figure S 28. ¹³C NMR spectra of the TROST Ligand (in CDCl₃, 150 MHz)

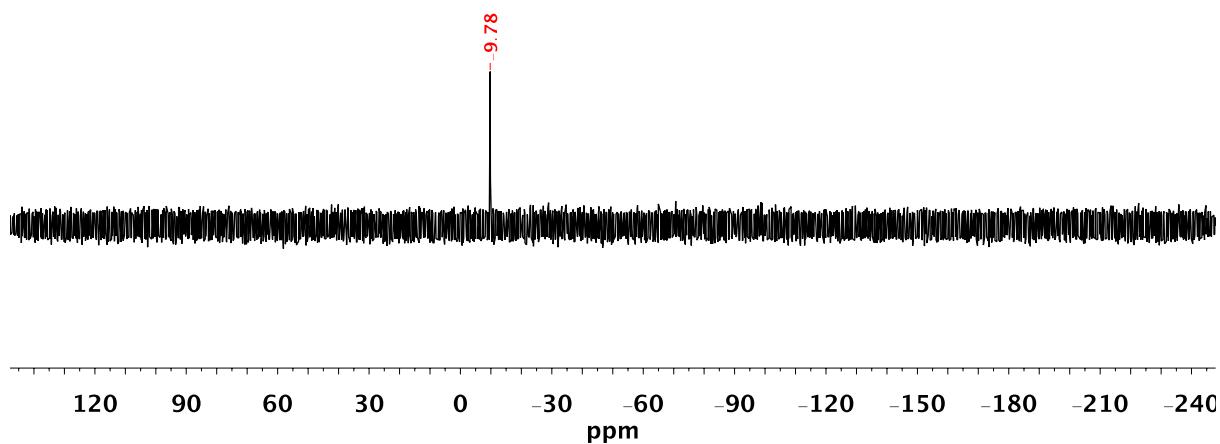
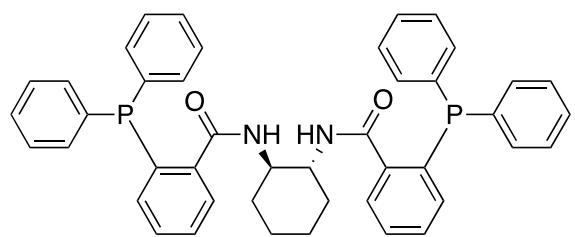


Figure S 29. ^{31}P NMR spectrum of Trost ligand (in CDCl_3 , 243 MHz)

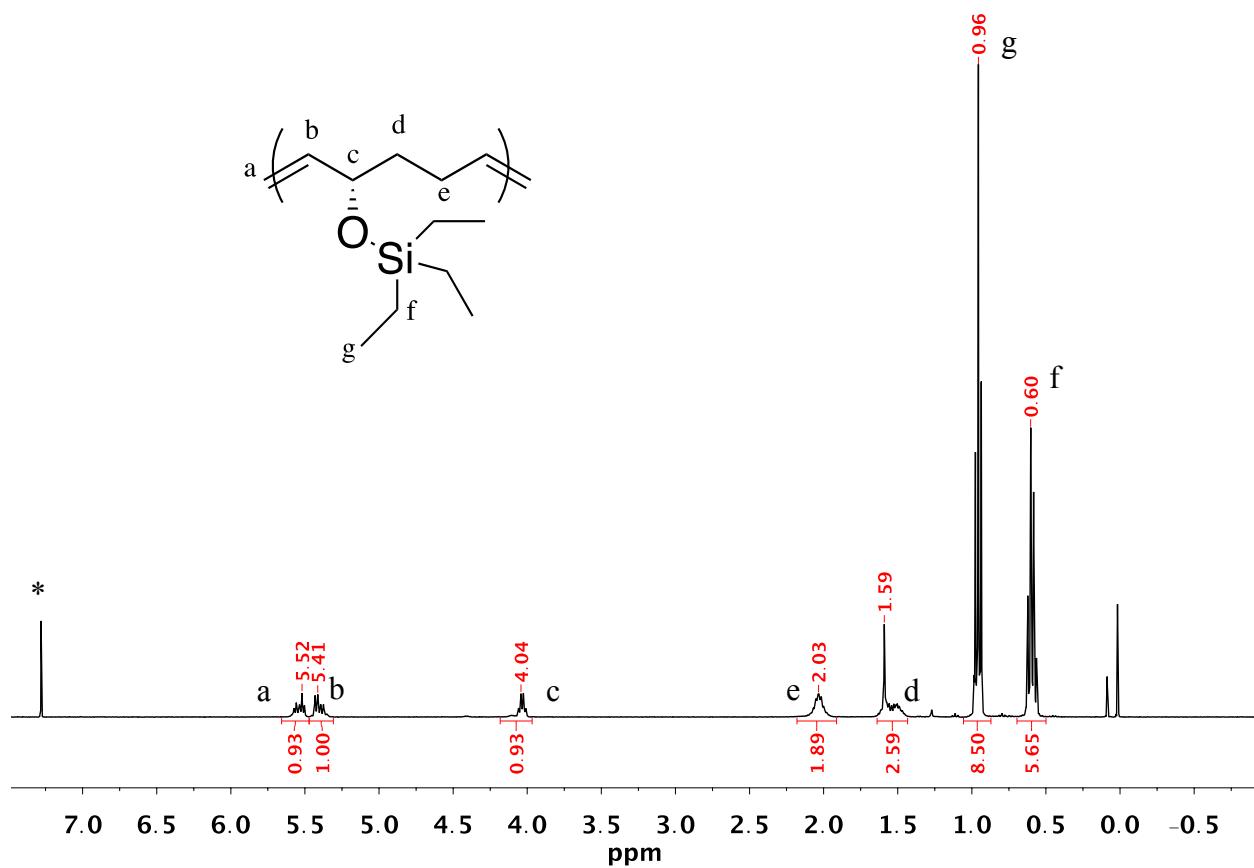


Figure S 30. ^1H NMR spectrum of Poly[(S)-3], (in $^*\text{CDCl}_3$, 400 MHz)

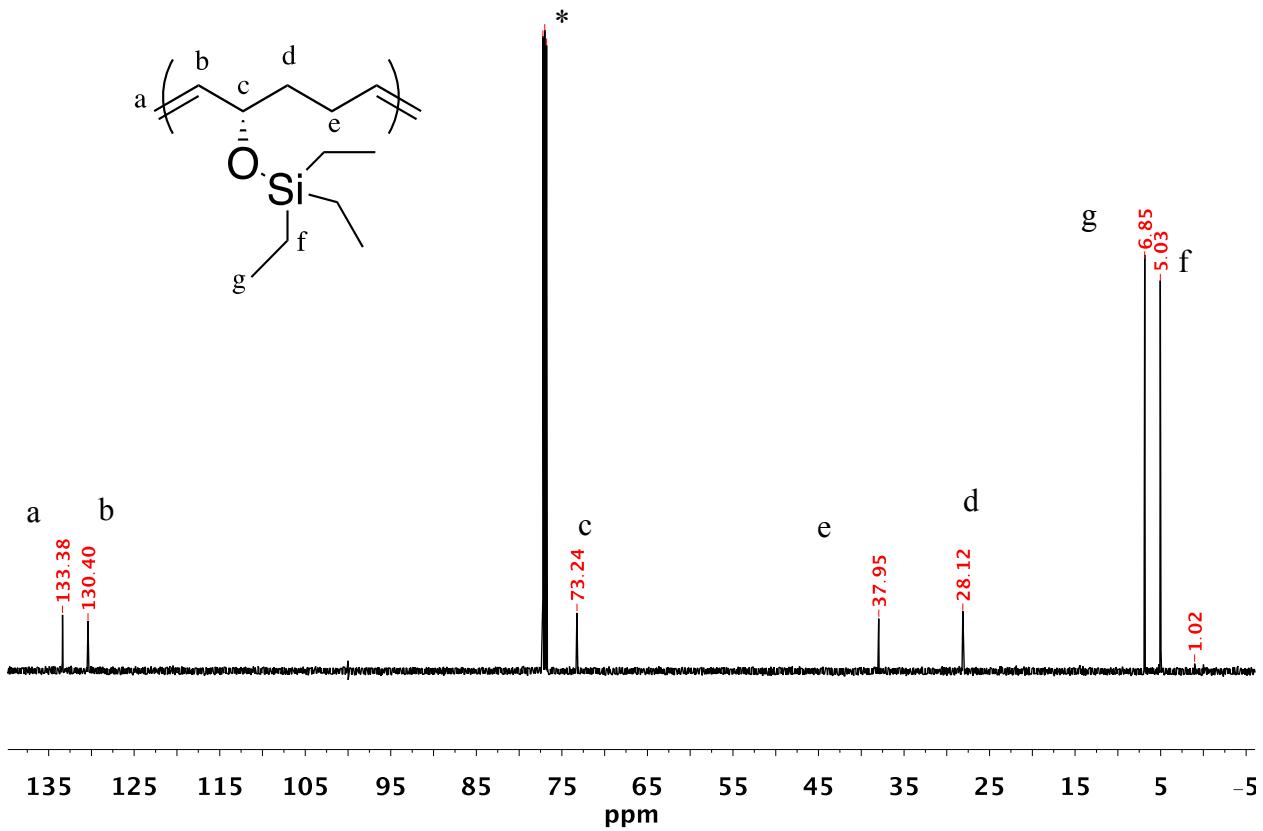


Figure S 31. ^{13}C NMR spectrum of Poly[(S)-3], (in $^*\text{CDCl}_3$, 150 MHz)

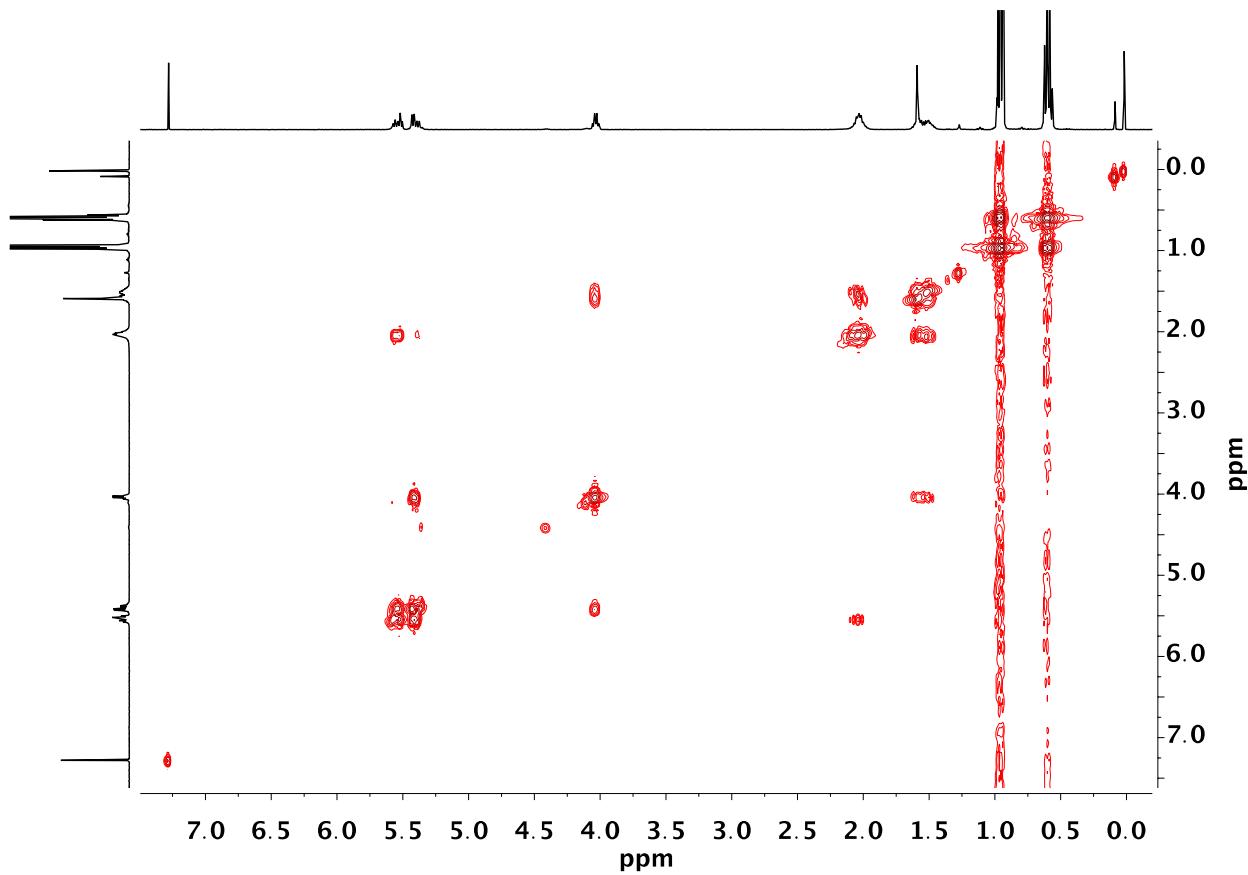


Figure S 32. ^1H - ^1H COSY spectrum of Poly[(S)-3] (CDCl_3 , 150 MHz)

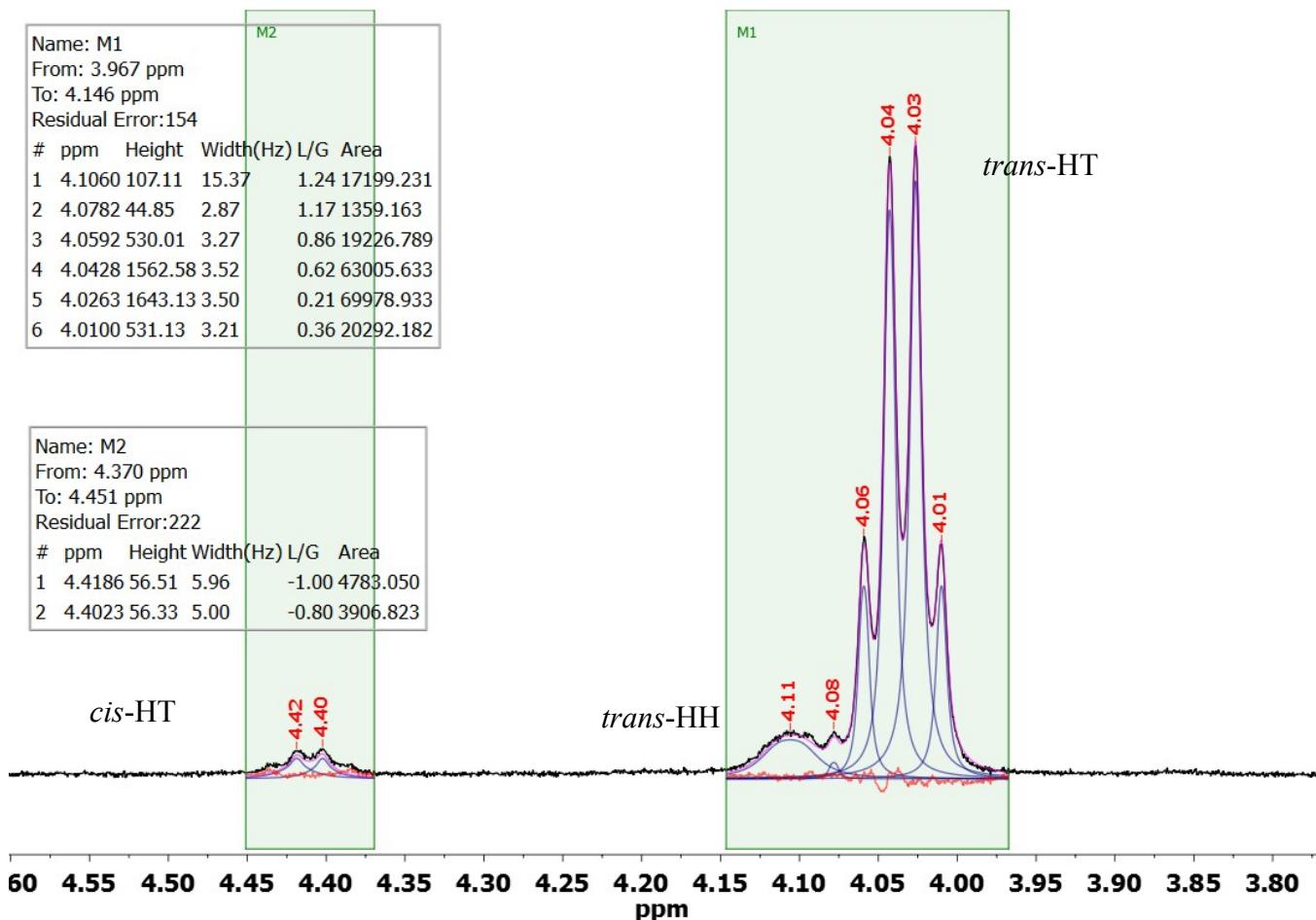


Figure S 33. ¹H-NMR spectrum of Poly[(S)-3] (CDCl₃, 400 MHz). Deconvolution of the peaks were used to determine the %HT and % trans of the polymer.

$$\% HT = \frac{\text{area of } cis - HT + trans - HT}{\text{total area } cis - HT + trans - HH + trans - HT} * 100 = 91.4 \%$$

$$\% trans = \frac{\text{area of } trans - HH + trans - HT}{\text{total area } cis - HT + trans - HH + trans - HT} * 100 = 95.6 \%$$

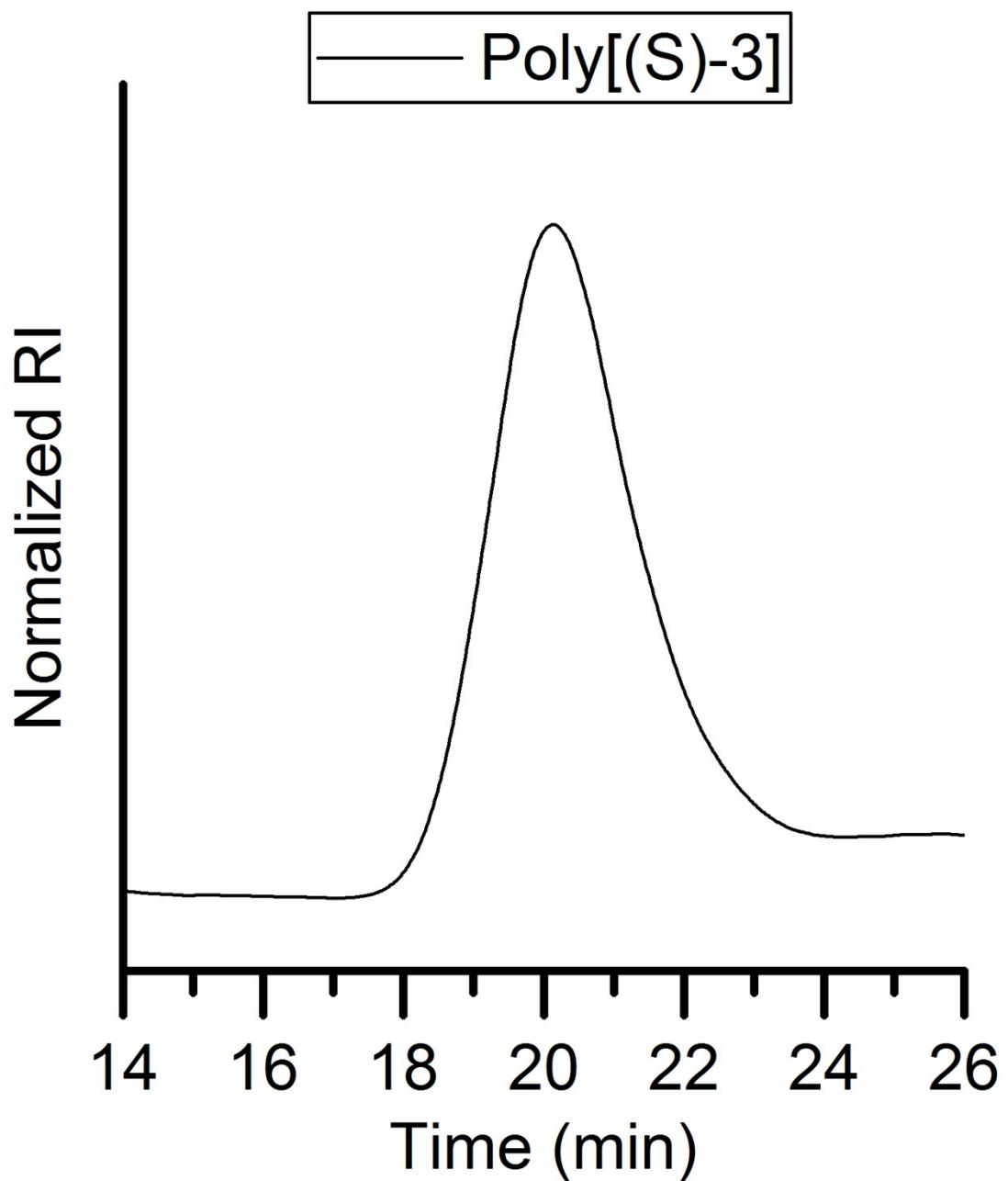


Figure S 34. SEC RI trace of Poly(*S*)-3. ($D = 2.81$, $M_n = 30.2$ kDa)

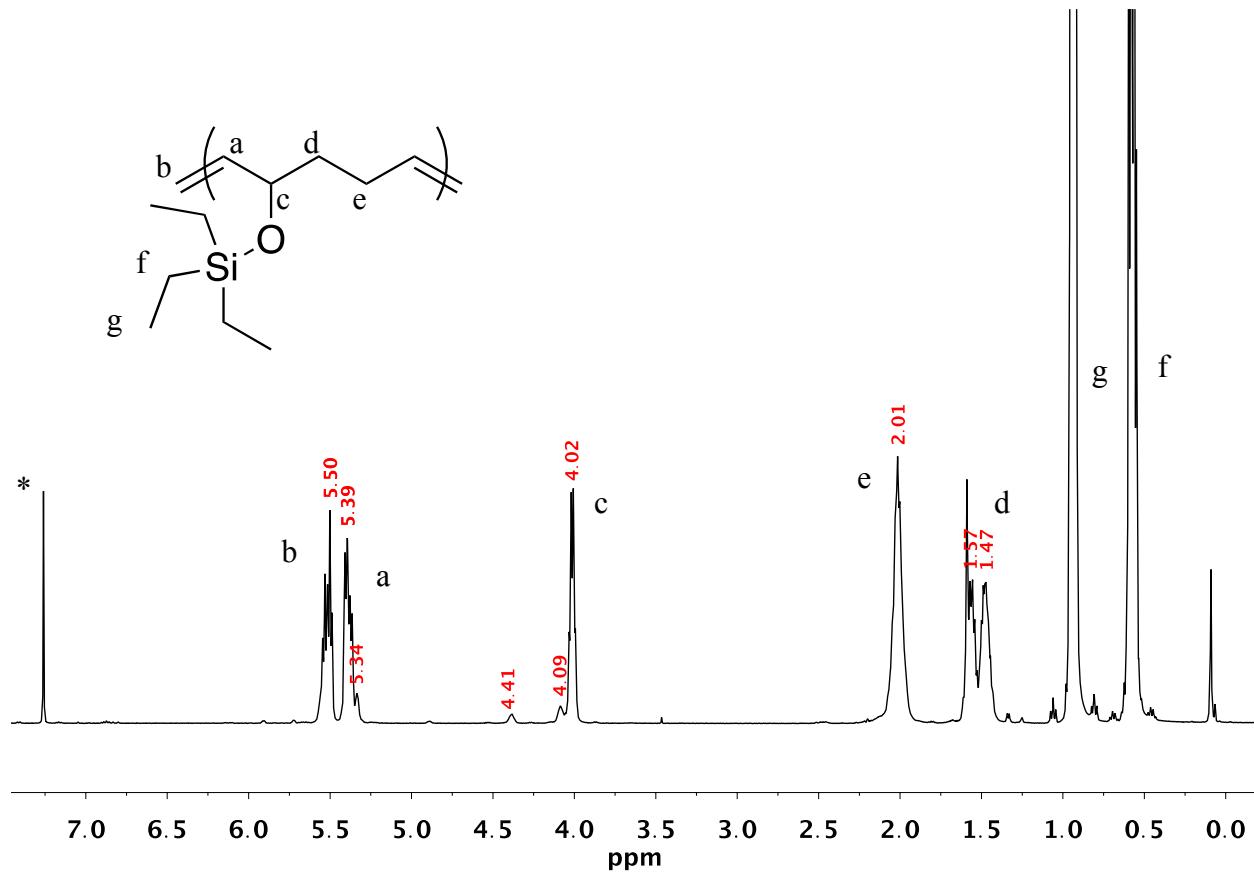


Figure S
35. ^1H
NMR of
spectrum
of
Poly(*rac*-
3), (in
 $^*\text{CDCl}_3$,
500
MHz)

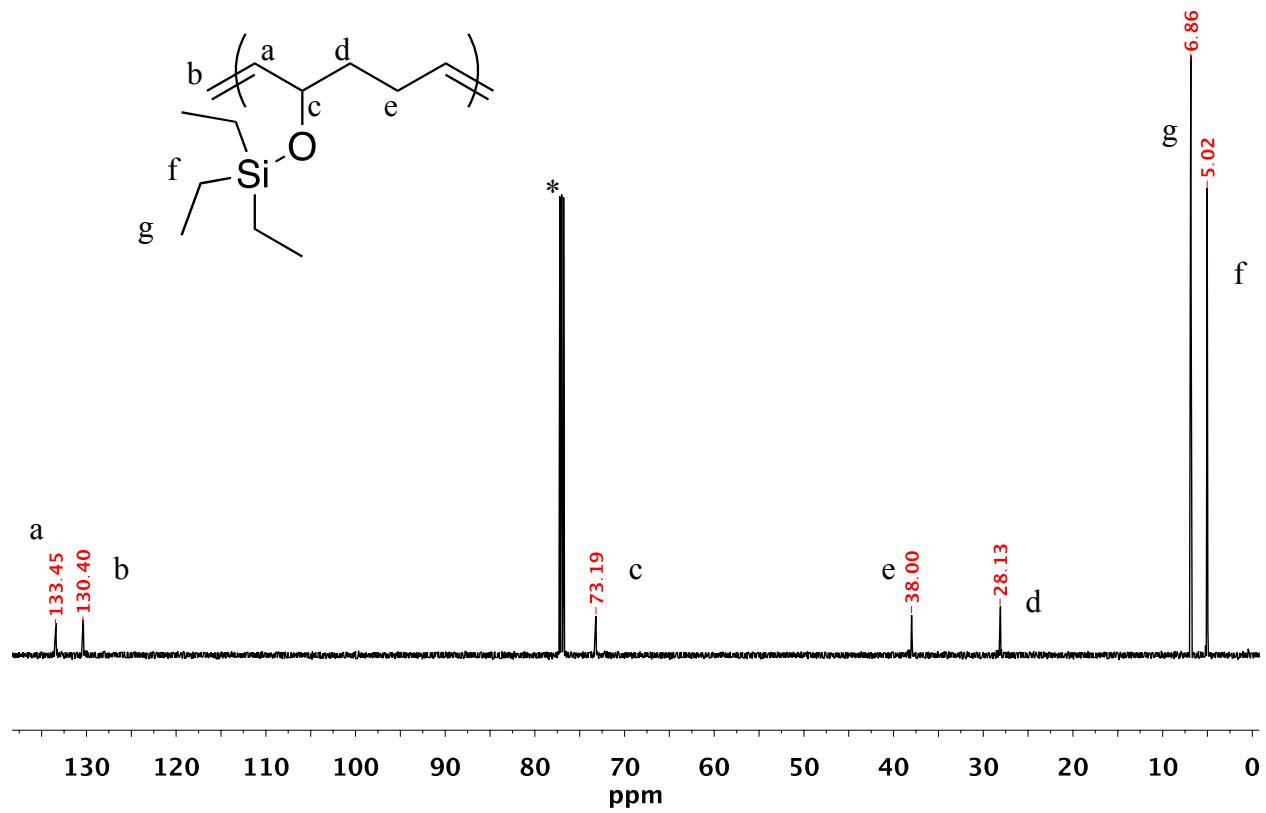


Figure S
36. ^{13}C
NMR of
spectrum of
Poly(*rac*-3),
(in $^*\text{CDCl}_3$,
150 MHz)

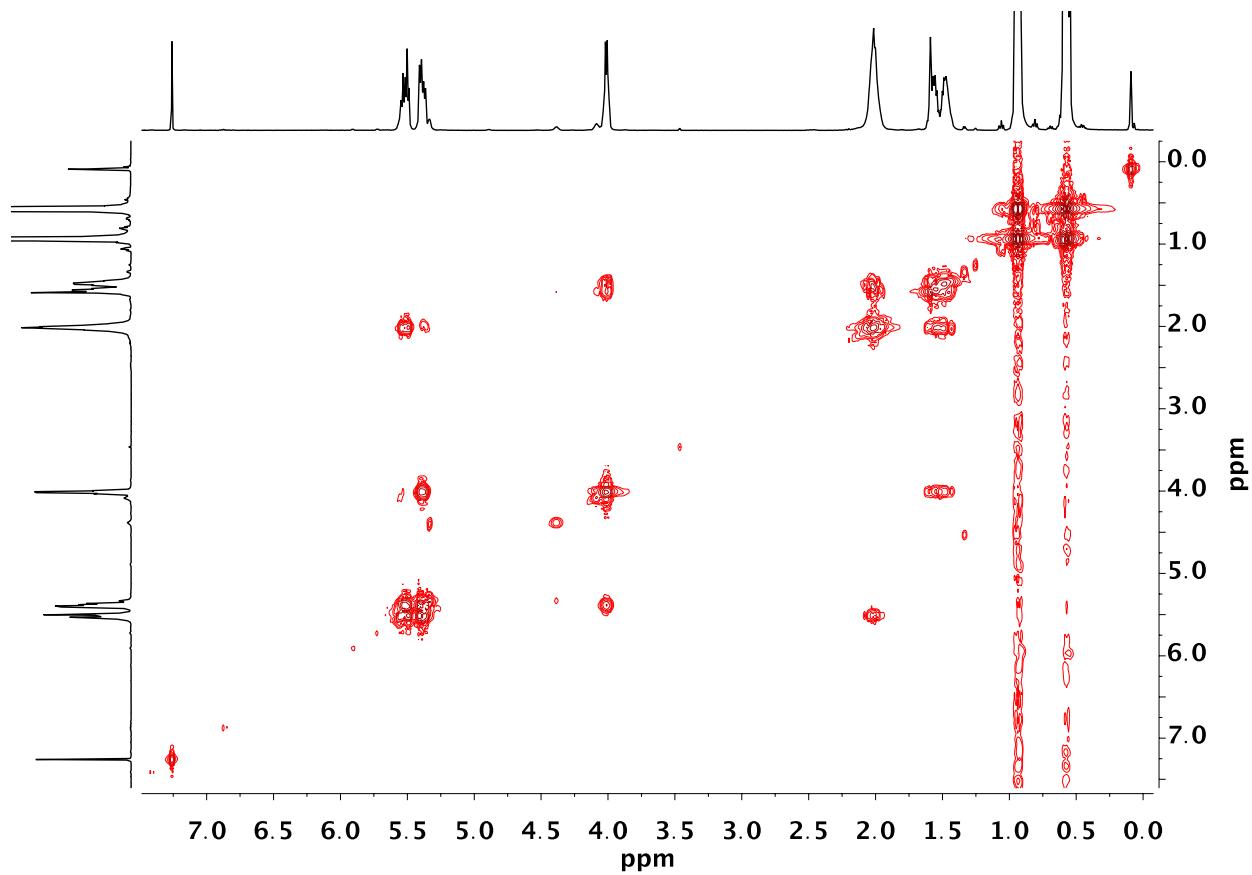


Figure 37. ^1H - ^1H COSY of spectrum of Poly(*rac*-3), (in CDCl_3 , 500 MHz)

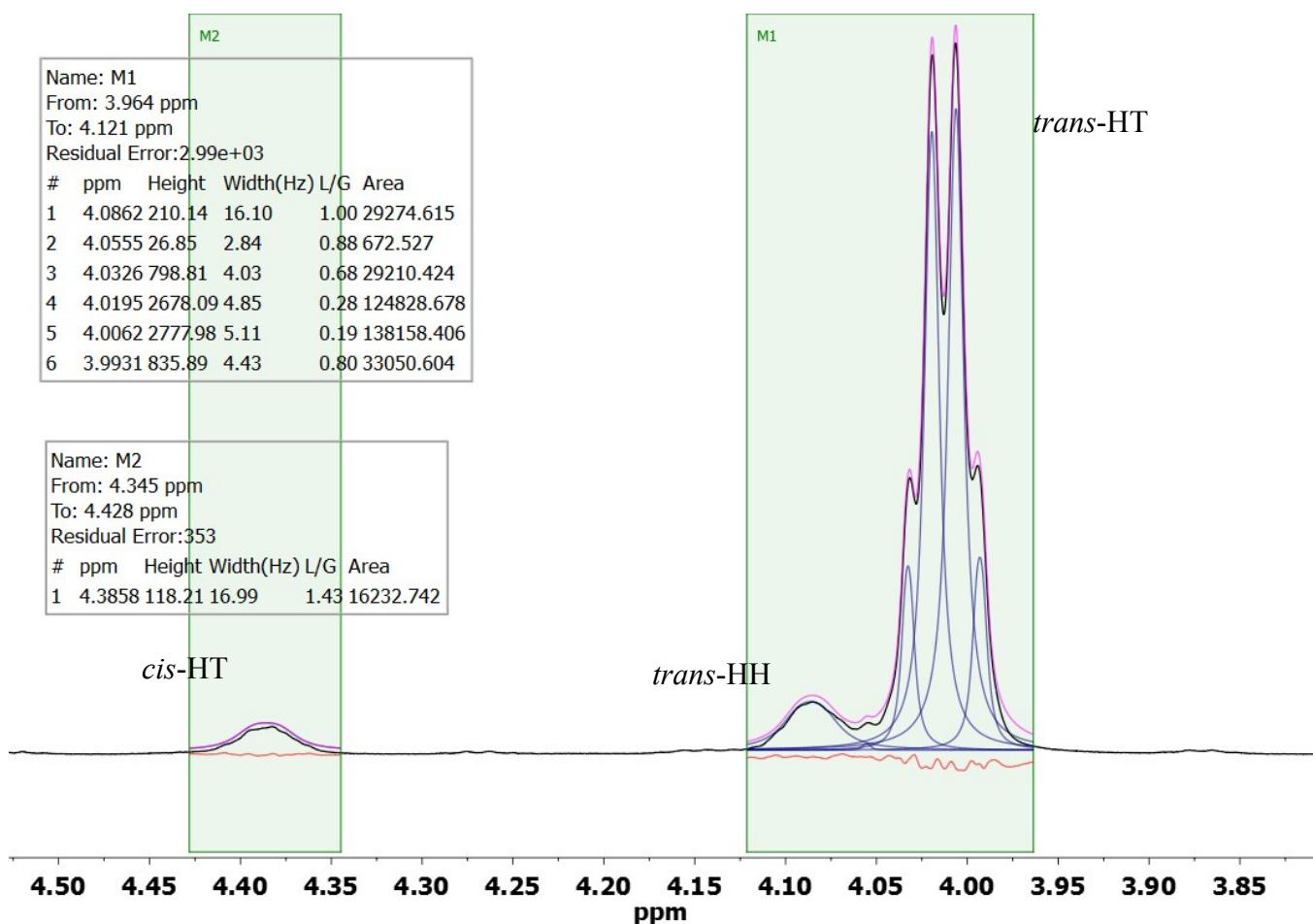


Figure S 38. ¹H-NMR spectrum of Poly(rac-3) (CDCl₃, 400 MHz). Deconvolution of the peaks were used to determine the %HT and % trans of the polymer.

$$\% HT = \frac{\text{area of cis - HT} + \text{trans - HT}}{\text{total area cis - HT} + \text{trans - HH} + \text{trans - HT}} * 100 = 92.1 \%$$

$$\% \text{ trans} = \frac{\text{area of trans - HH} + \text{trans - HT}}{\text{total area cis - HT} + \text{trans - HH} + \text{trans - HT}} * 100 = 95.6 \%$$

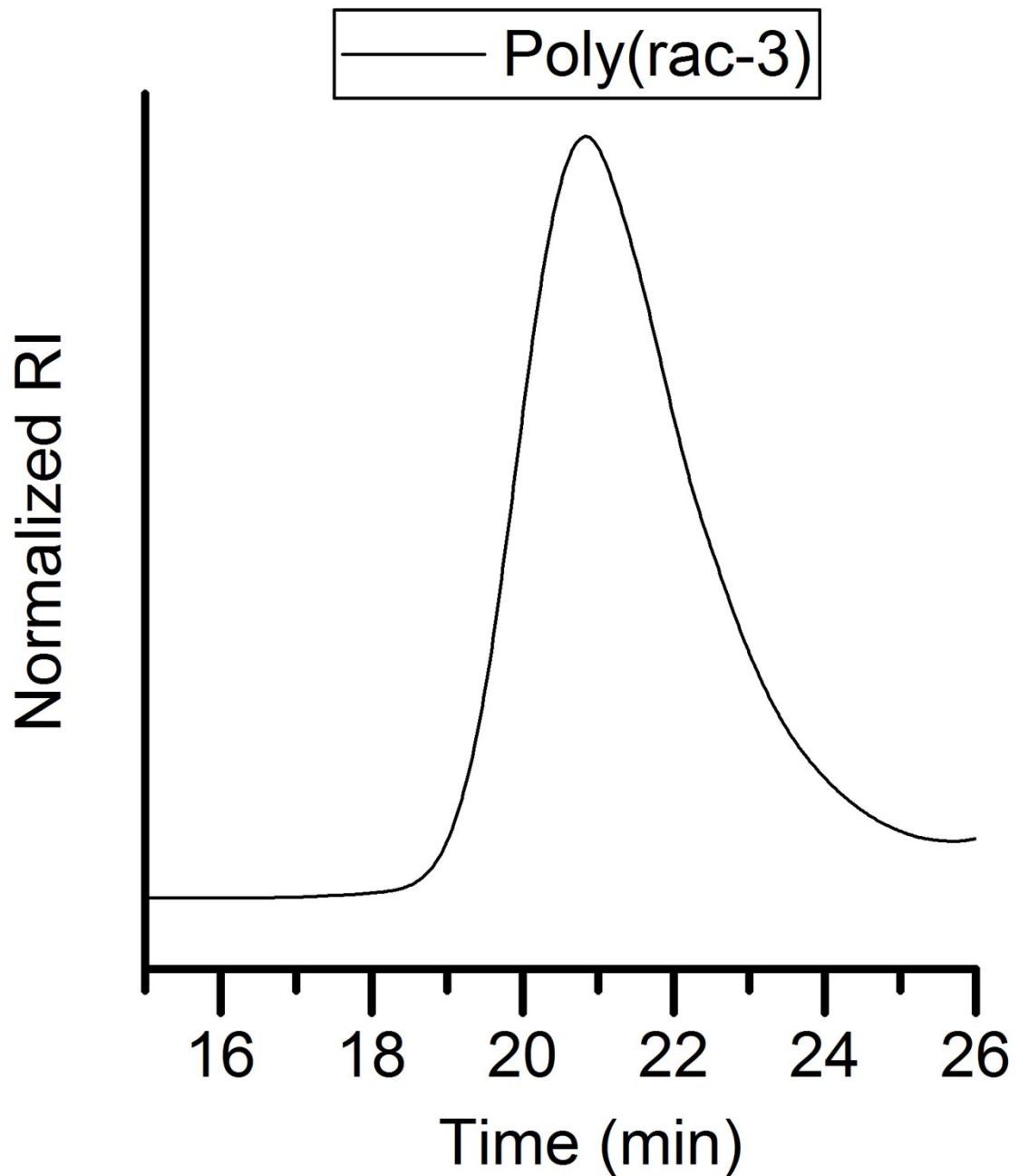


Figure S 39. SEC RI trace of Poly(*rac*-3). ($D = 2.56$, $M_n = 22.3$ kDa)

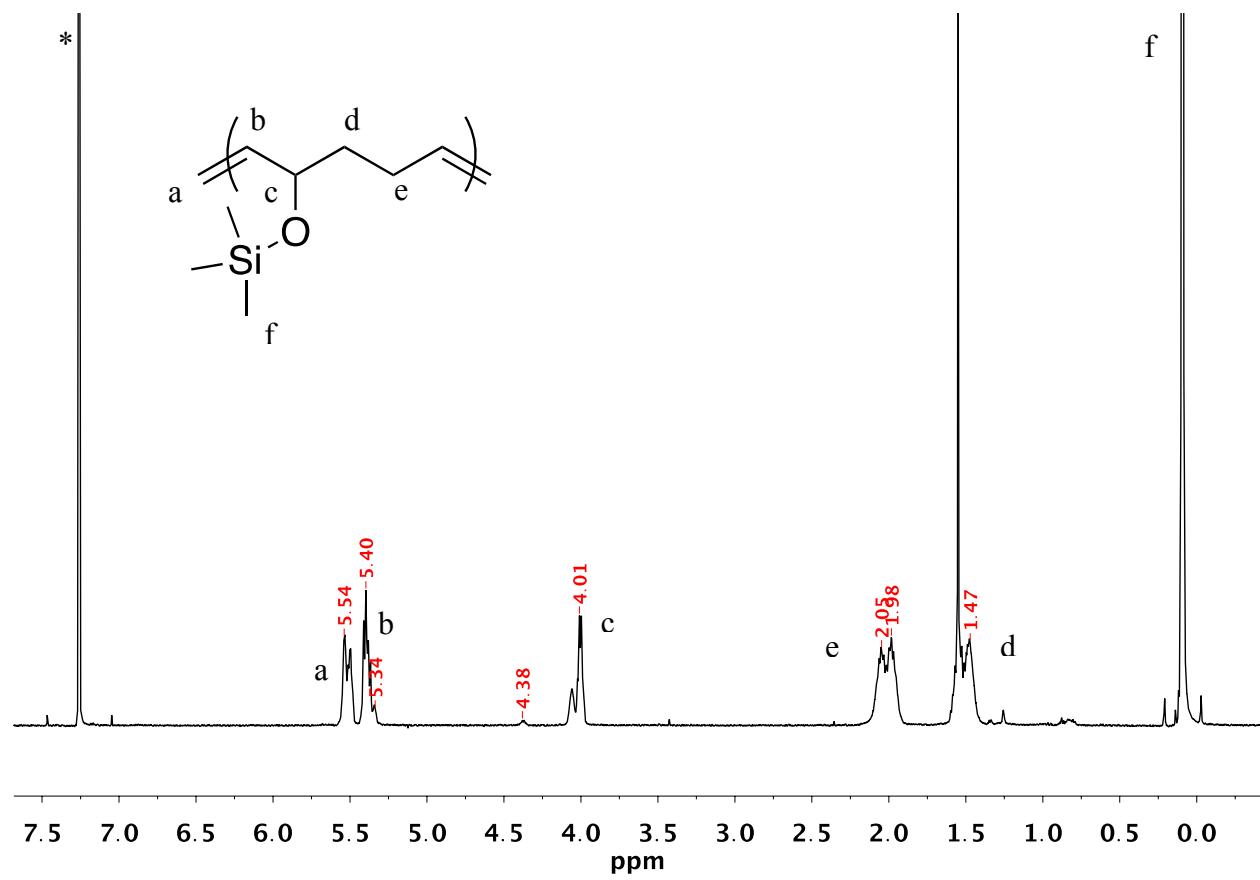


Figure S
40. ^1H
NMR
spectrum of
Poly(*rac*-
2), (in
 $^{\ast}\text{CDCl}_3$,
500 MHz)

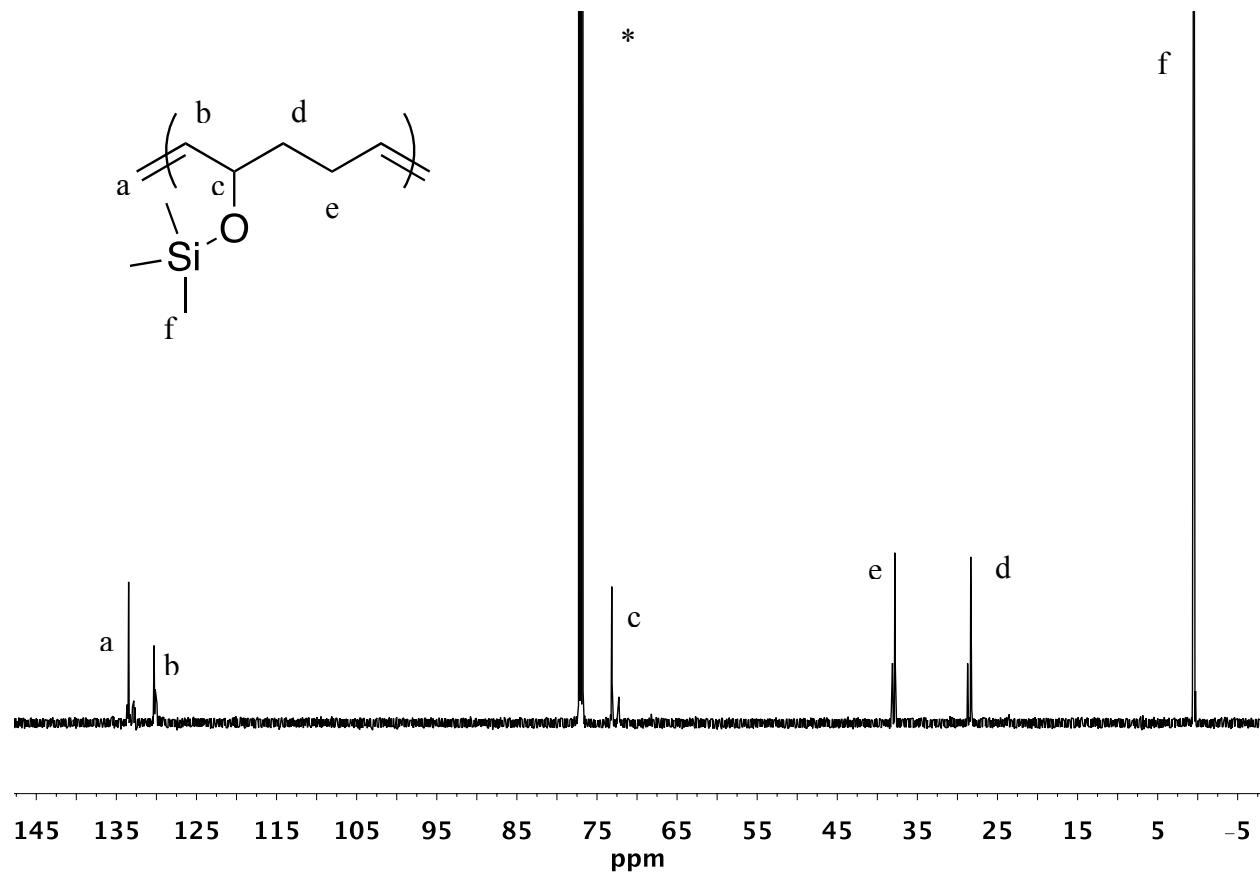


Figure S 41. ^{13}C NMR spectrum of Poly(*rac*-2), (in $^*\text{CDCl}_3$, 150 MHz)

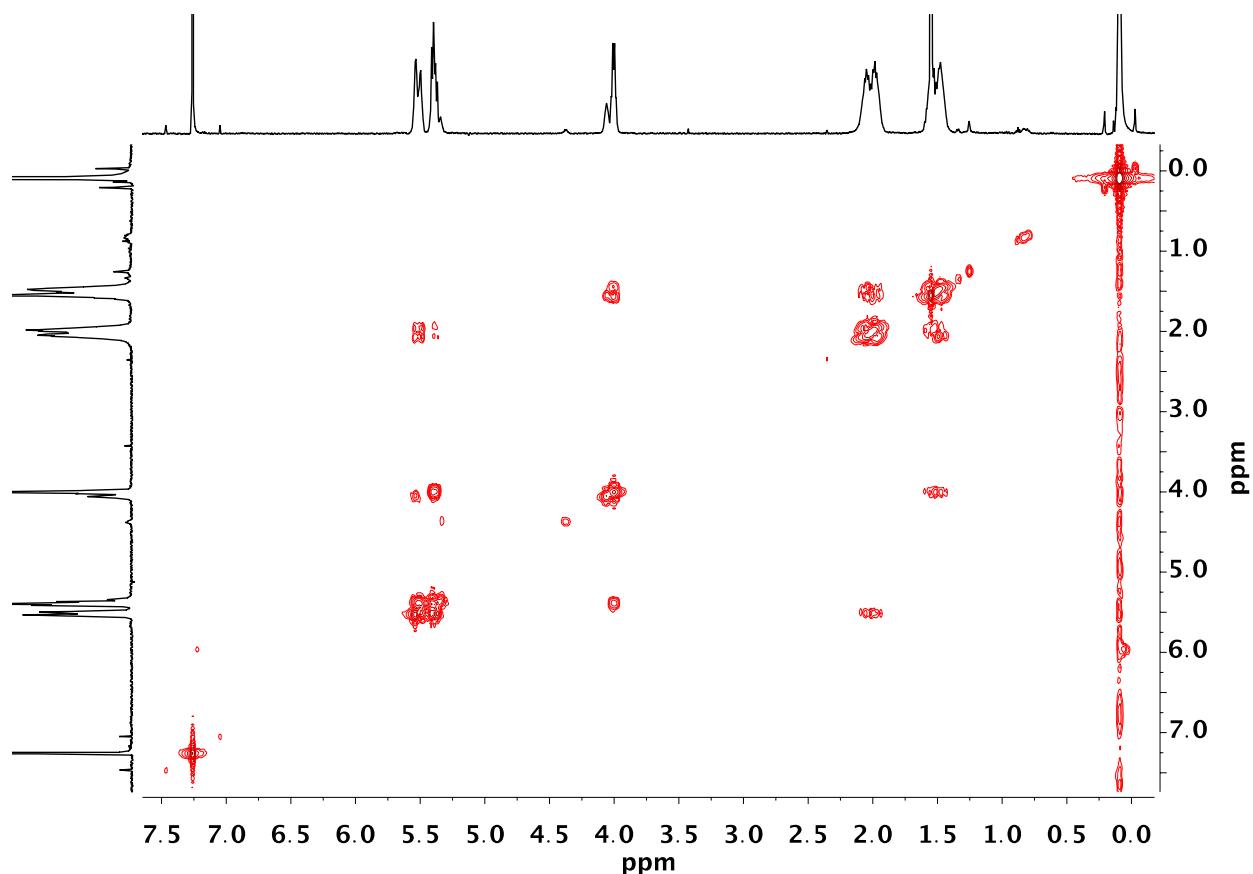


Figure S 42. ^1H - ^1H COSY spectrum of Poly(*rac*-2), (in CDCl_3 , 500 MHz)

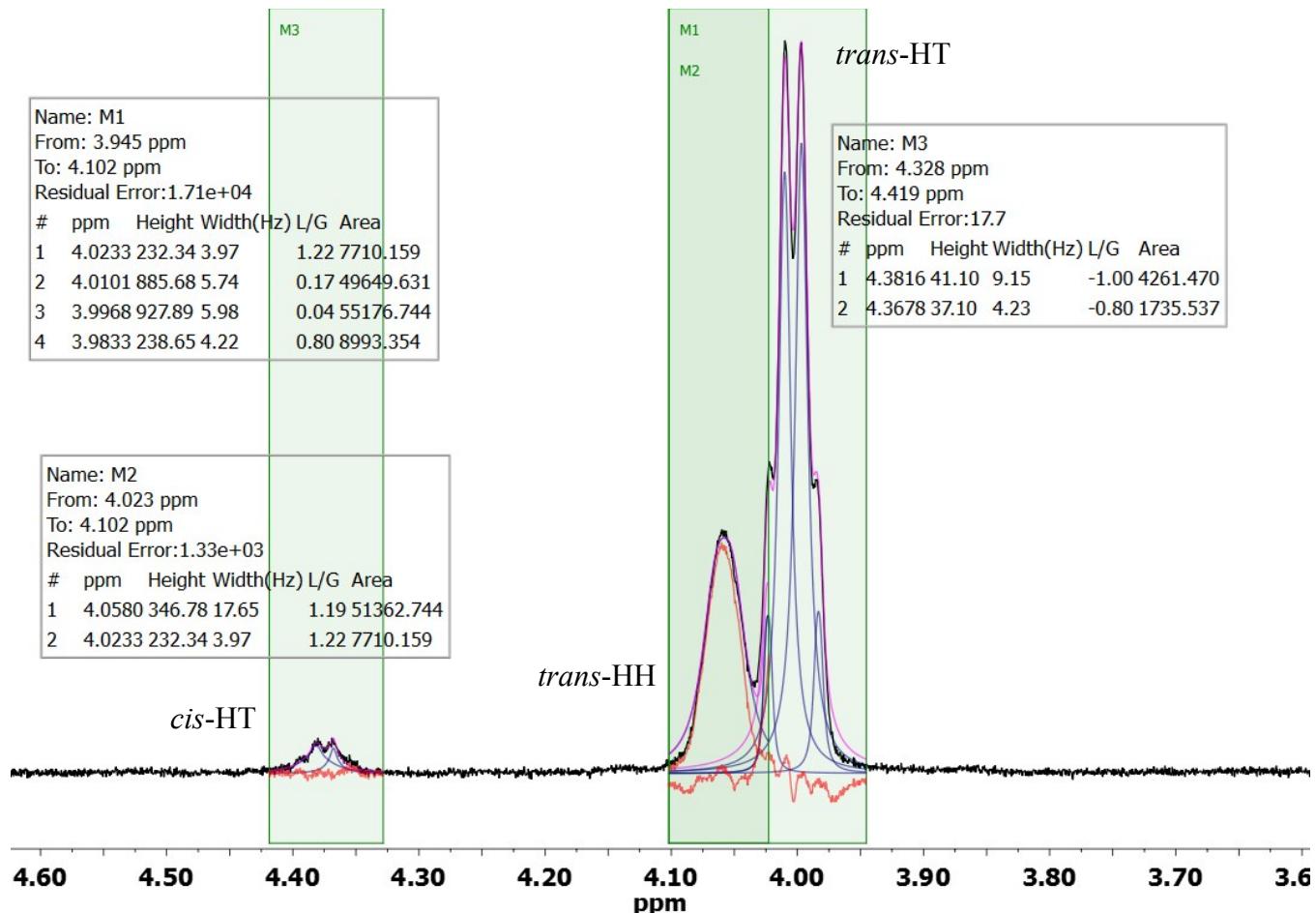


Figure S 43. ¹H-NMR spectrum of Poly(rac-2) (CDCl₃, 400 MHz). Deconvolution of the peaks were used to determine the %HT and % trans of the polymer.

$$\% HT = \frac{\text{area of cis-HT} + \text{trans-HT}}{\text{total area cis-HT} + \text{trans-HH} + \text{trans-HT}} * 100 = 68.3 \%$$

$$\% \text{trans} = \frac{\text{area of trans-HH} + \text{trans-HT}}{\text{total area cis-HT} + \text{trans-HH} + \text{trans-HT}} * 100 = 94.2 \%$$

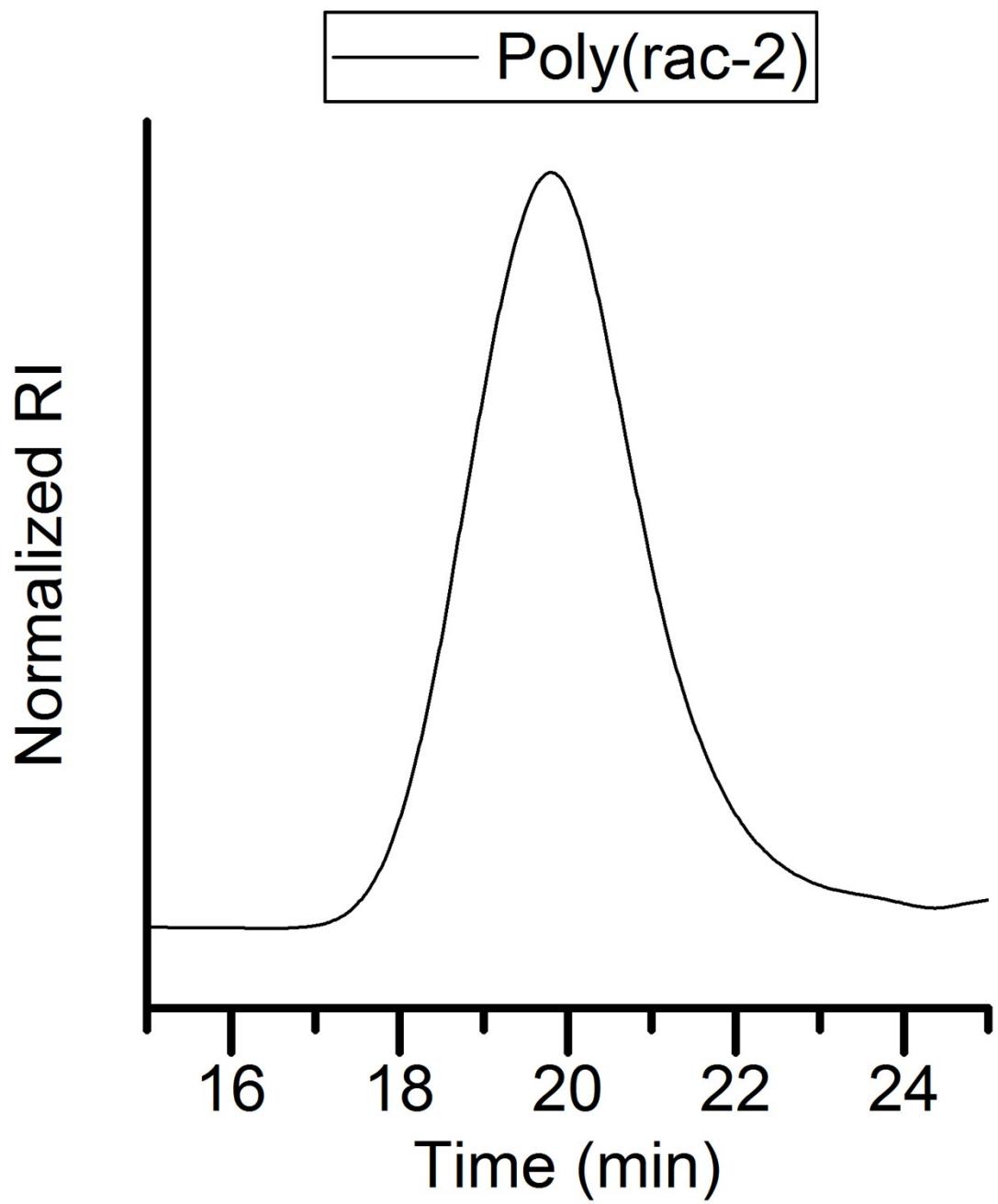


Figure S 44. SEC RI trace of Poly(*rac*-2). ($D = 2.07$, $M_n = 41.6$ kDa)

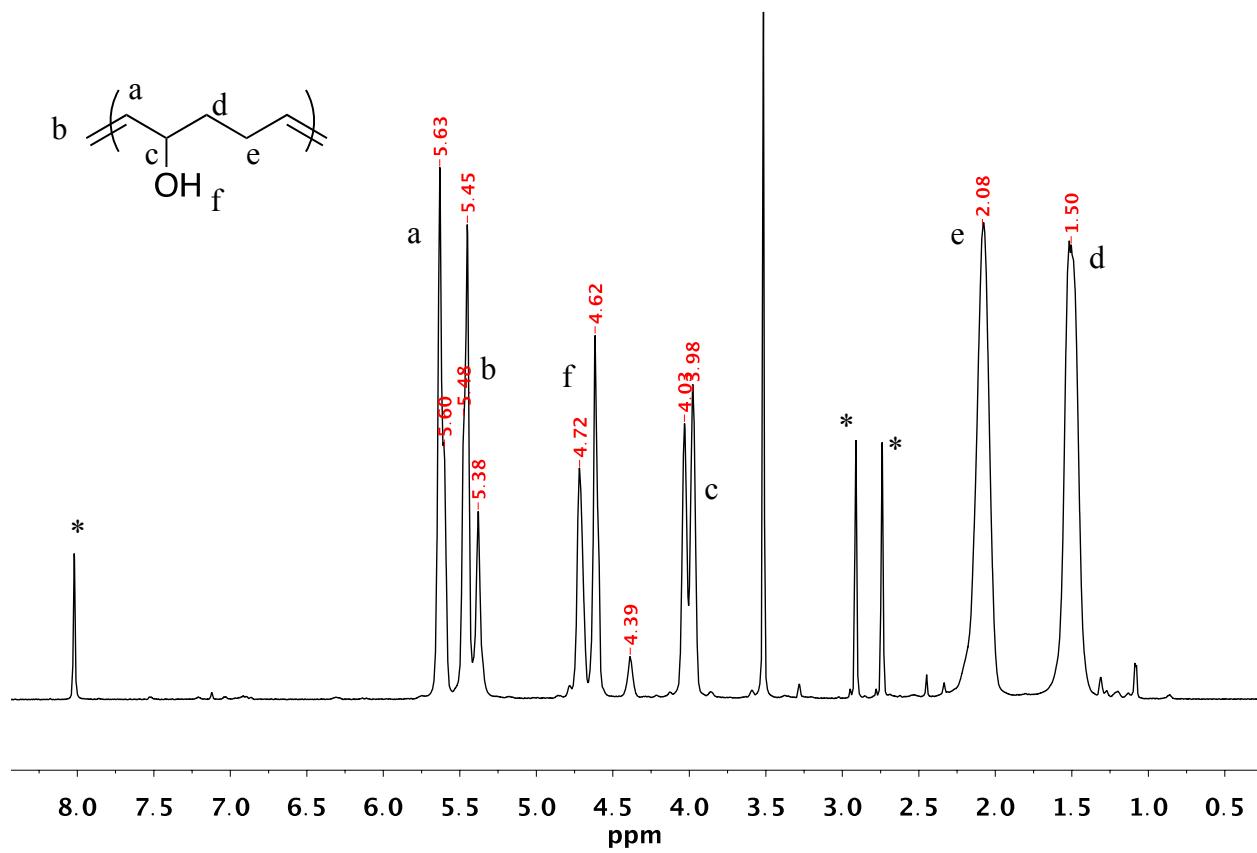


Figure S 45. ^1H NMR spectrum of Poly(*rac*-1), (*DMF- d_7 , 600 MHz)

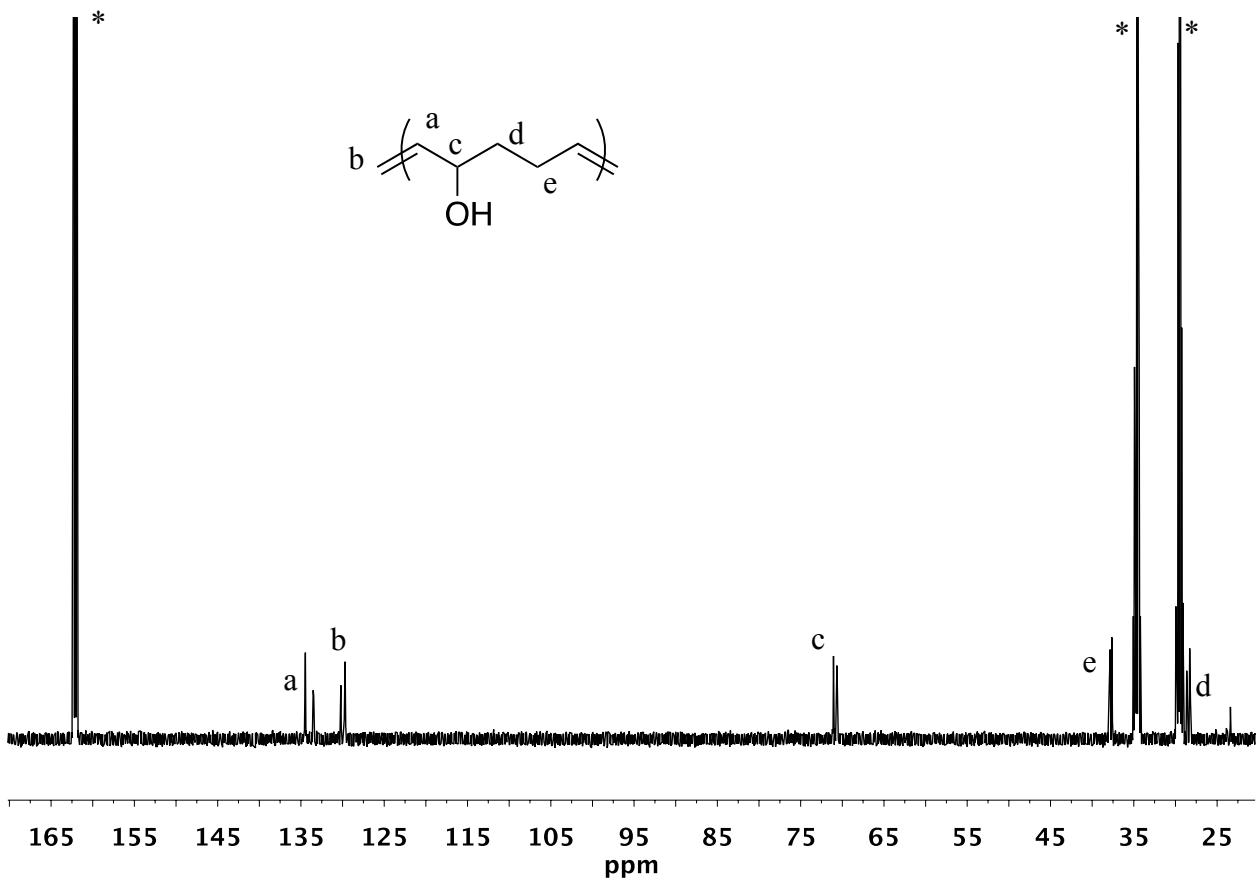


Figure S 46. ^{13}C NMR spectrum of Poly(*rac*-1), (*DMF- γ , 150 MHz)

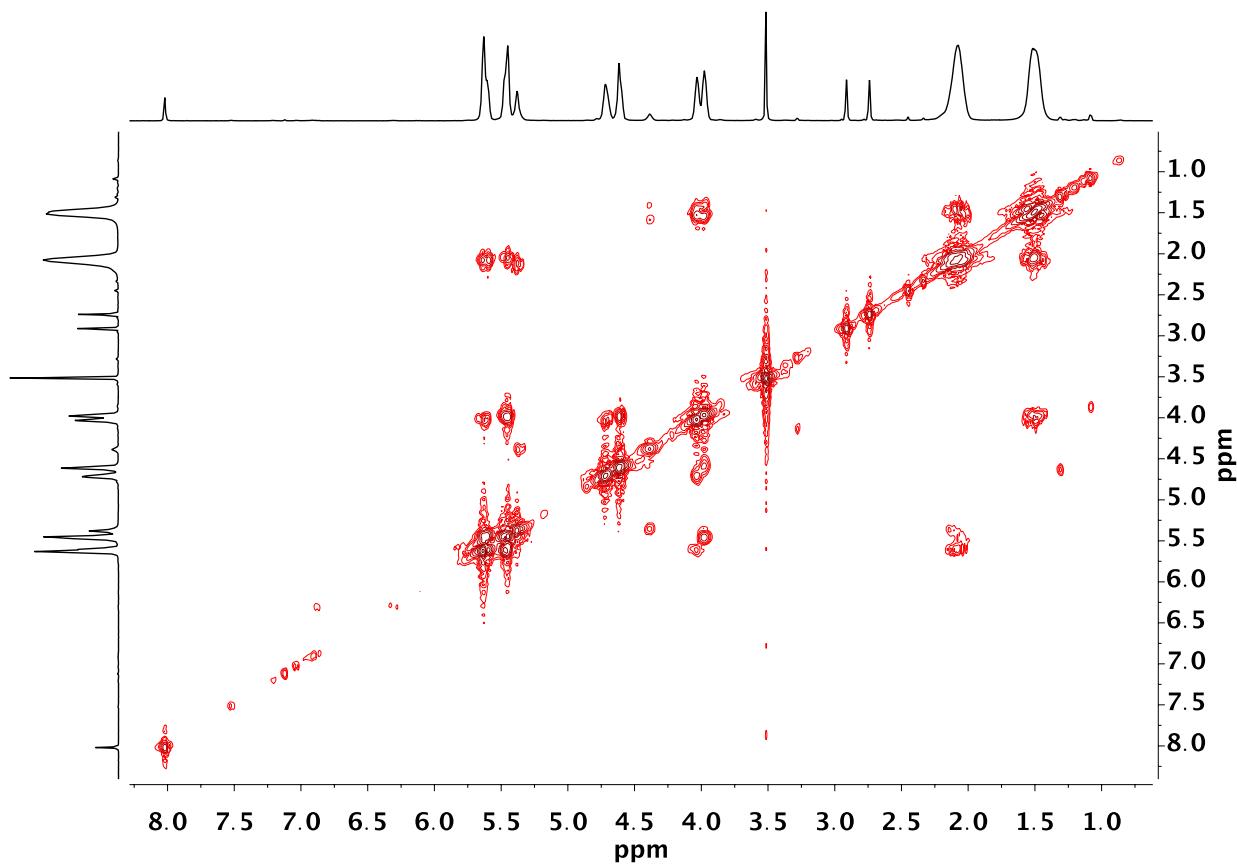


Figure S 47. ^1H - ^1H COSY spectra of Poly(*rac*-1) (CDCl_3 , 600 MHz)

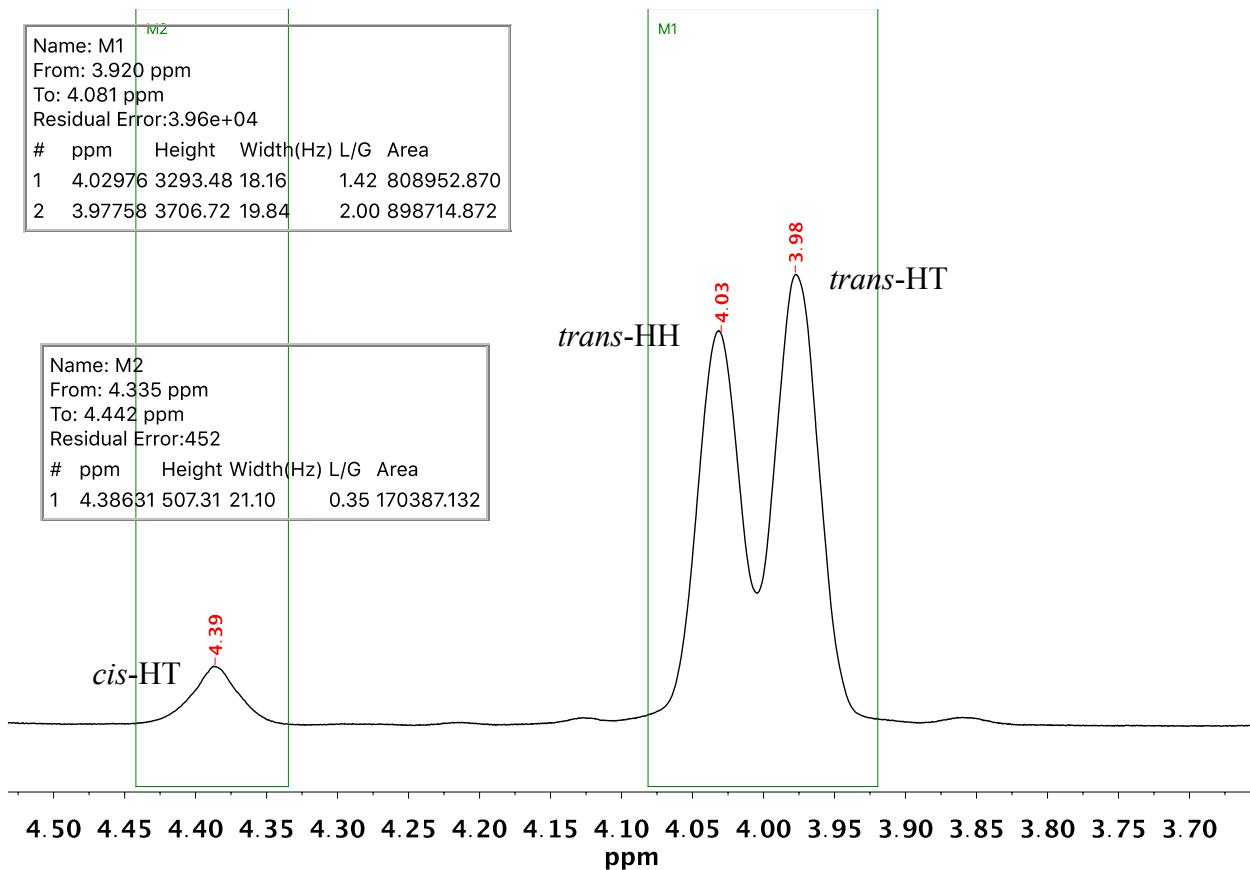


Figure S 48. ^1H -NMR spectrum of Poly(*rac*-1) (CDCl_3 , 400 MHz). Deconvolution of the peaks were used to determine the %HT and % trans of the polymer.

$$\% \text{ HT} = \frac{\text{area of cis-HT} + \text{trans-HT}}{\text{total area cis-HT} + \text{trans-HH} + \text{trans-HT}} * 100 = 57 \text{ \%}.$$

$$\% \text{ trans} = \frac{\text{area of trans-HH} + \text{trans-HT}}{\text{total area cis-HT} + \text{trans-HH} + \text{trans-HT}} * 100 = 91 \text{ \%}.$$

Table S49: Aliquot characterization of Poly(*rac*-3) at 2.5 M with HG2 at -10 °C.

Time (min)	conv ^a %
31.5	4.8
52.33	14.5
90	33.1
116	33.3
157	33.3

^a Determined by ¹H NMR (400 MHz in CDCl₃).

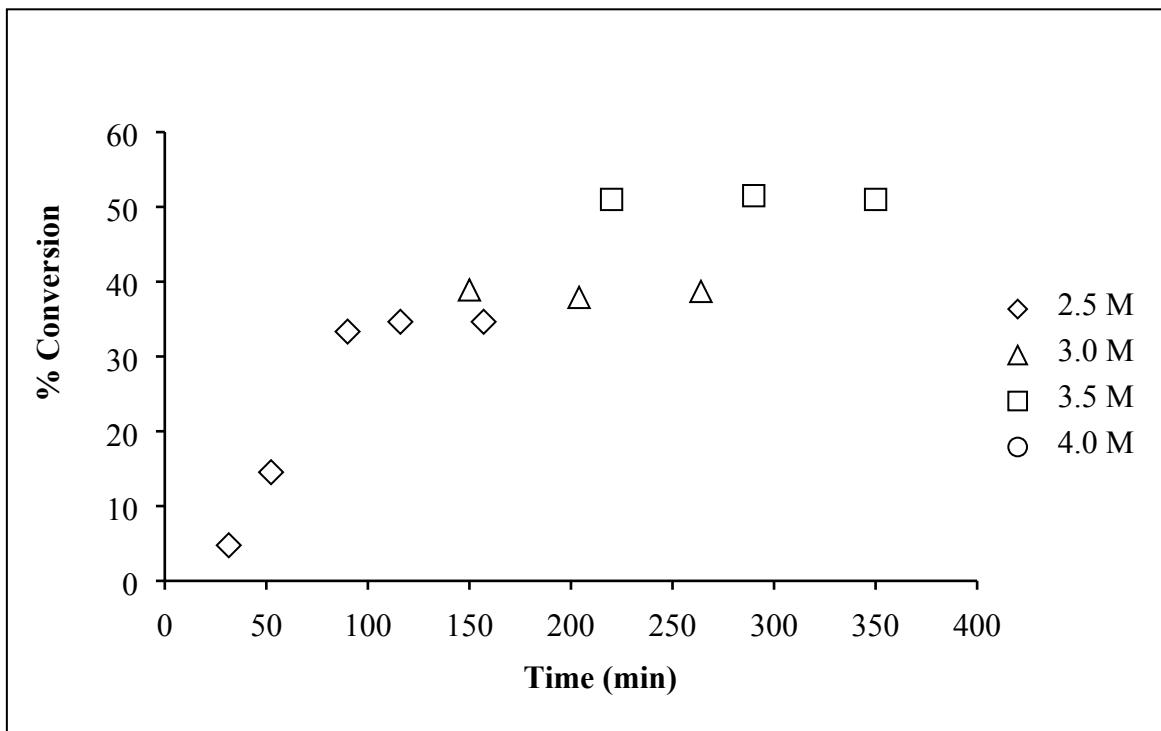


Figure S50. Equilibrium conversion of Poly(*rac*-3) with respect to time.

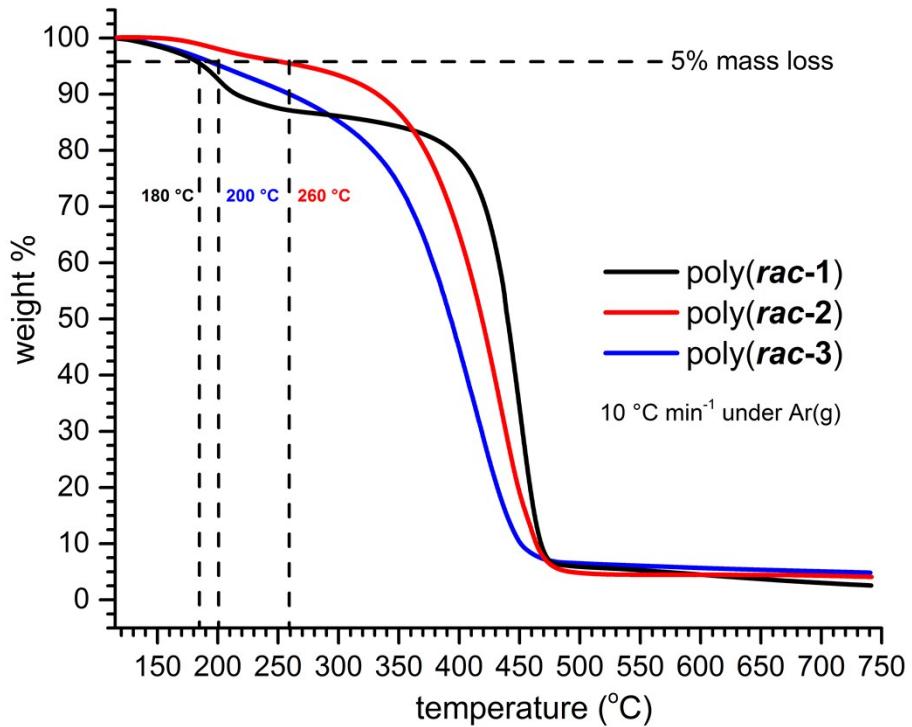


Figure S51. TGA analysis of poly(*rac*-1), poly(*rac*-2) and poly(*rac*-3) after equilibration at 110 °C followed by heating at 10 °C min⁻¹ to 750 °C under argon.

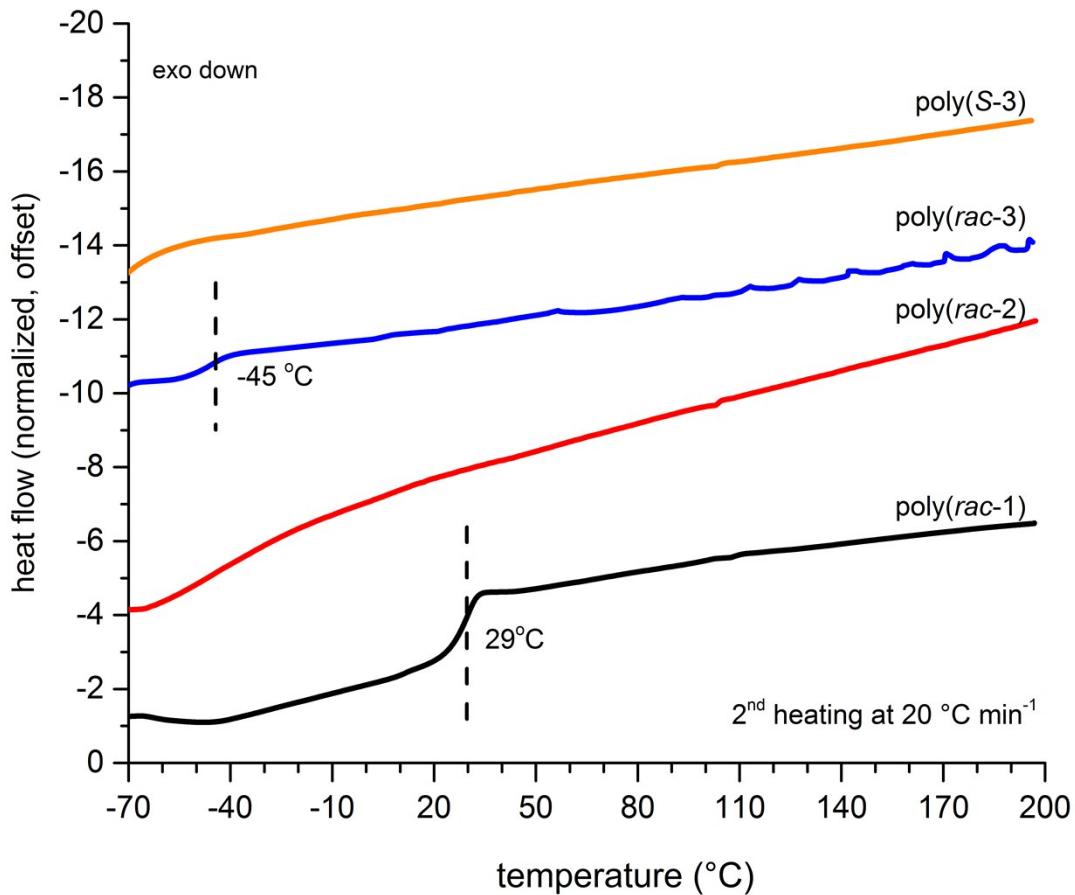


Figure S52. Differential scanning calorimetry thermograms of each polymer sample. The thermal range chosen was based on the minimum temperature for the instrument (-70 °C) and the temperature at which thermal decomposition of the polymer is suspected (~200 °C) based on TGA. Jagged baseline noise is present above 100 °C on some samples and was determined to be an artifact of the instrument by repeated cycles where this noise was found inconsistent.

References:

1. T. R. Hoye, C. S. Jeffrey and F. Shao, *Nature Protocols*, 2007, **2**, 2451.
2. B. M. Trost, D. L. Van Vranken and C. Bingel, *J. Am. Chem. Soc.*, 1992, **114**, 9327-9343.
3. S. Fuchs, V. Berl and J.-P. Lepoittevin, *European Journal of Organic Chemistry*, 2007, **2007**, 1145-1152.