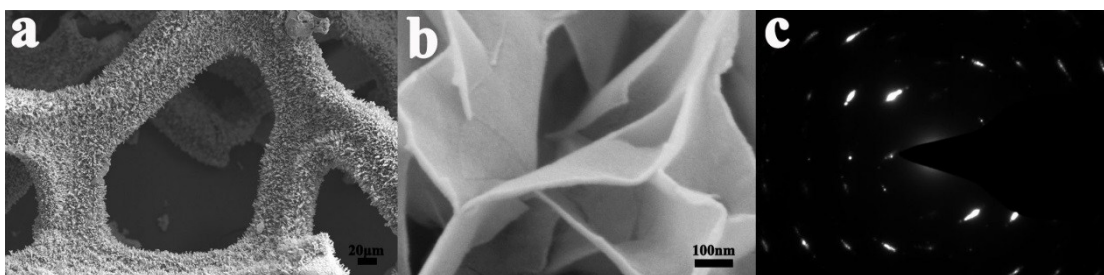


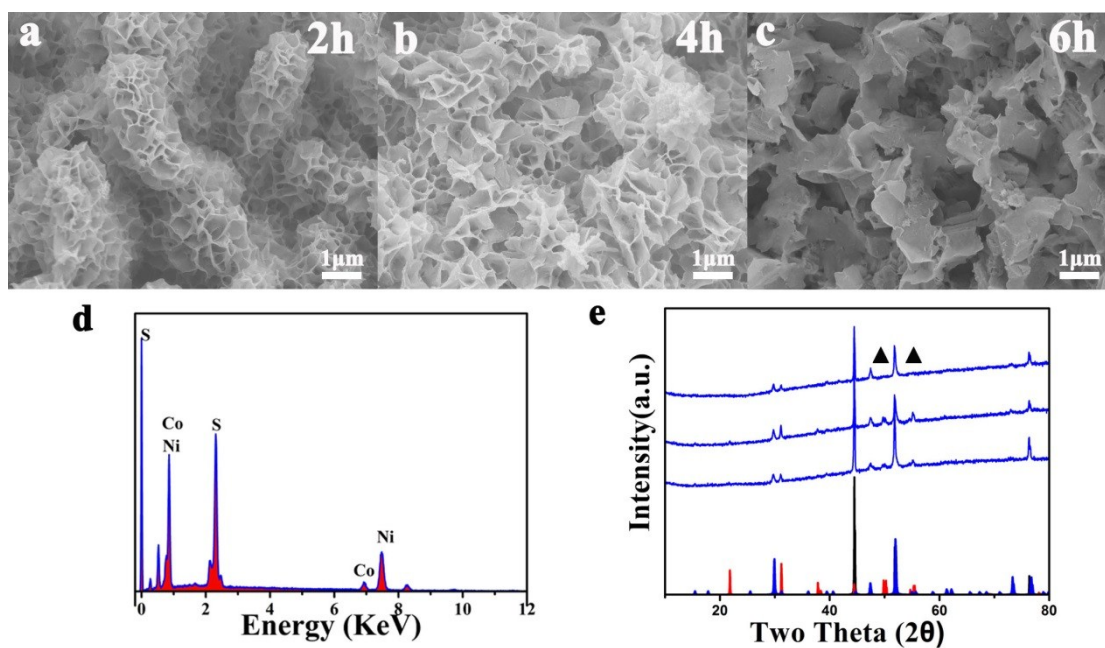
## Supporting Information

### **Sequential Partial Ion Exchange Synthesis of Composite $\text{Ni}_3\text{S}_2/\text{Co}_9\text{S}_8/\text{NiSe}$ Nanoarrays with Lavender-like Hierarchical Morphology**

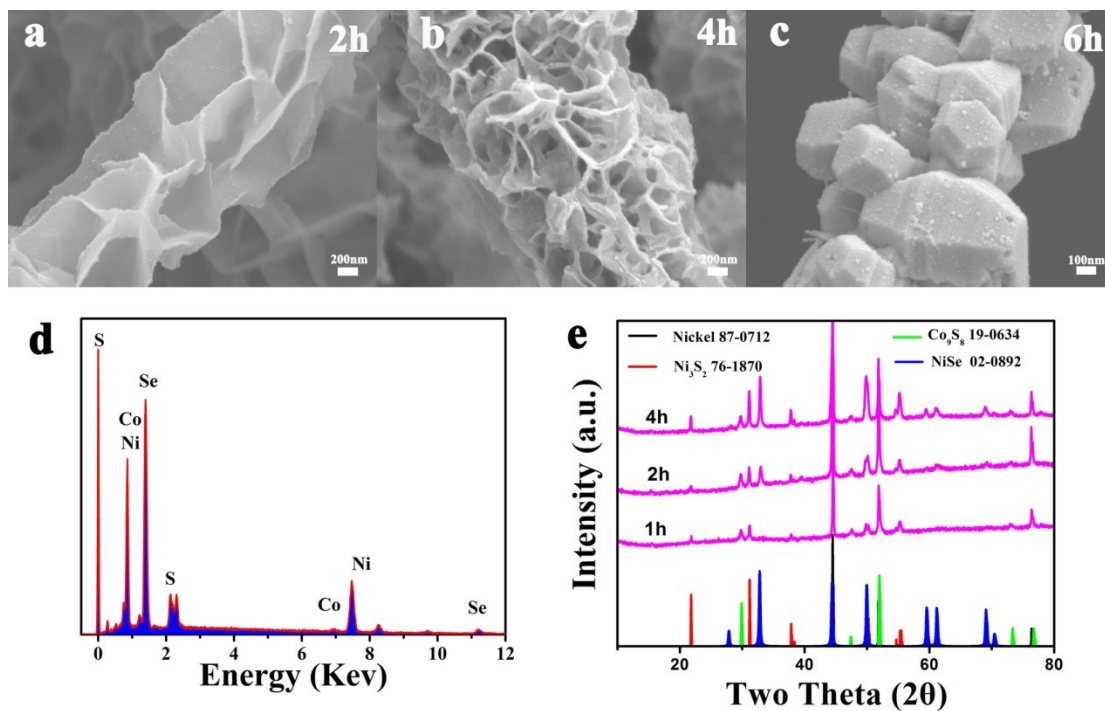
Shaobo Huang,<sup>a</sup> Wangxi Zhang,<sup>a</sup> Shizhong Cui,<sup>a</sup> Weihua Chen<sup>b,\*</sup> and Liwei Mi<sup>a,\*</sup>



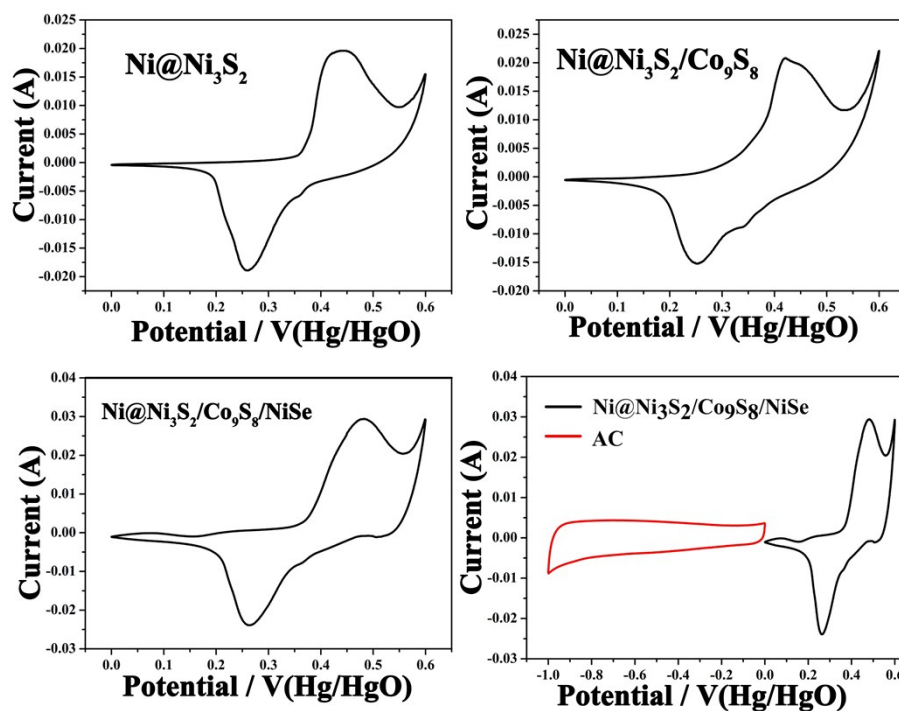
**Figure S1.** (a) Low-magnification skeleton of Ni@Ni<sub>3</sub>S<sub>2</sub> (b) the high magnification of nanoflakes (c) SAED images of Ni@Ni<sub>3</sub>S<sub>2</sub>



**Figure S2.** SEM images of Co-exchange progress at different reaction time from (a-c). (d) EDS results at the Co exchange reaction time of 2h. (e) XRD results at different reaction time.



**Figure S3.** SEM images of Se-exchange progress at different reaction time from (a-c). (d) EDS results at the Se exchange reaction time of 2h. (e) XRD results at different Se exchange reaction time.



**Figure S4.** CV curves of (a) Ni@Ni<sub>3</sub>S<sub>2</sub>, (b) Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub>, (c) Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub>/NiSe at scan rates of 5 mV s<sup>-1</sup> in a three-electrode (d) CV curves of Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub>/NiSe and AC electrodes measured at a scan rate of 5 mV s<sup>-1</sup> in a three-electrode system

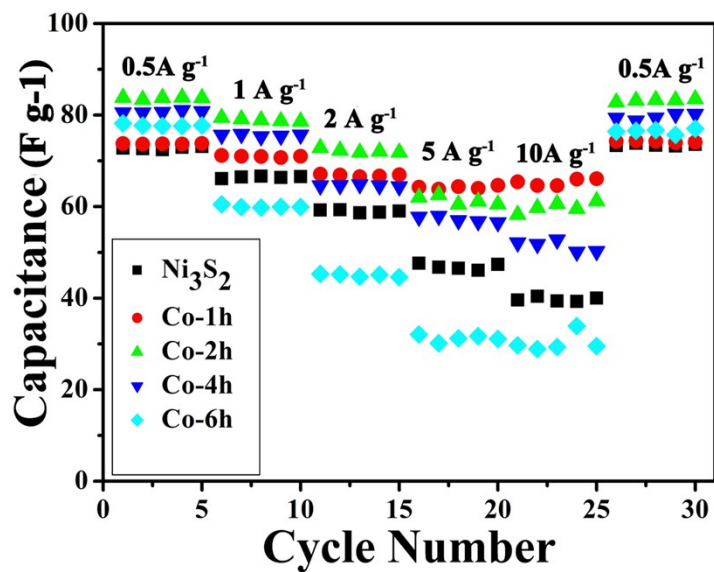


Figure S5. Rate performance of under different reaction time during Co exchange progress at the different current density.

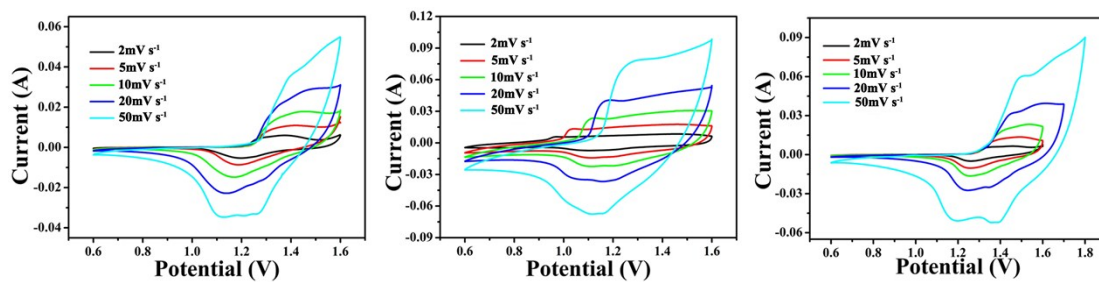


Figure S6. (a) CV curve of Ni@Ni<sub>3</sub>S<sub>2</sub> at different scan speed. (b) CV curve of Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub> at different scan speed

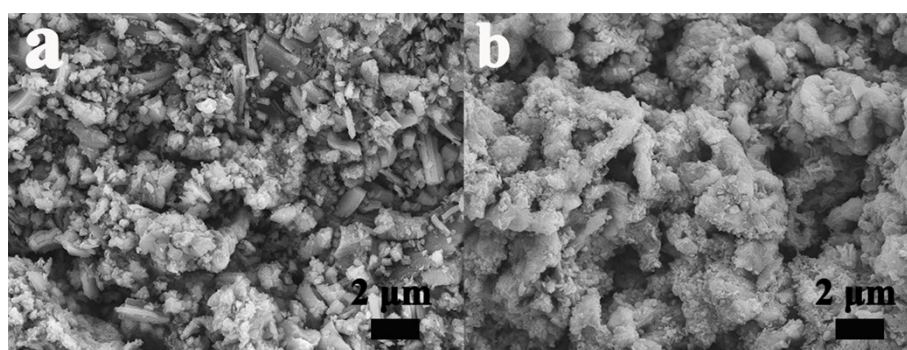


Figure S7. SEM images after 2000th cycles (a) SEM image of Ni@Ni<sub>3</sub>S<sub>2</sub> (b) SEM images of Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub>/NiSe

Table S1: Comparison of related materials about the power density at maximum energy density

<b>Devices</b>	<b>Energy density</b>	<b>Power density</b>	<b>Source</b>
Ni <sub>3</sub> S <sub>2</sub> //AC	17.5 Wh kg <sup>-1</sup>	301 W kg <sup>-1</sup>	<i>Journal of Power Sources</i> , 2016, <b>320</b> , 13-19.
Ni <sub>3</sub> S <sub>2</sub> @CoS//AC	28.24 Wh kg <sup>-1</sup>	134.46 W kg <sup>-1</sup>	<i>Phys. Chem. Chem. Phys.</i> , 2015, <b>17</b> , 16434-16442.
AB-NiCo <sub>2</sub> S <sub>4</sub> //AC	24.7 Wh kg <sup>-1</sup>	428 W kg <sup>-1</sup>	<i>Electrochimica Acta</i> , 2015, <b>186</b> , 562-571.
NiCo <sub>2</sub> O <sub>4</sub> //AC	22.8	160	<i>J. Mater. Chem. A</i> , 2015, <b>3</b> , 12452-12460.
NiCo <sub>2</sub> O <sub>4</sub> @Ni <sub>3</sub> S <sub>2</sub> //AC	1.89 mW h cm <sup>-3</sup>	5.81 W cm <sup>-3</sup>	<i>Nanoscale</i> , 2016, <b>8</b> , 10686-10694.
Co <sub>3</sub> S <sub>4</sub> /NiS//AC	4.18 Wh m <sup>-2</sup>	160 W m <sup>-2</sup>	<i>RSC Adv.</i> , 2016, <b>6</b> , 97482-97490.
Ni(HCO <sub>3</sub> ) <sub>2</sub> //FexCy/C	24.96 Wh kg <sup>-1</sup>	87.75 W kg <sup>-1</sup>	<i>Electrochimica Acta</i> , 2015, <b>180</b> , 330–338.
<b>Ni@Ni<sub>3</sub>S<sub>2</sub>/Co<sub>9</sub>S<sub>8</sub>/NiSe //AC</b>	<b>31.99 Wh kg<sup>-1</sup></b>	<b>105.10 W kg<sup>-1</sup></b>	<b>This work</b>