

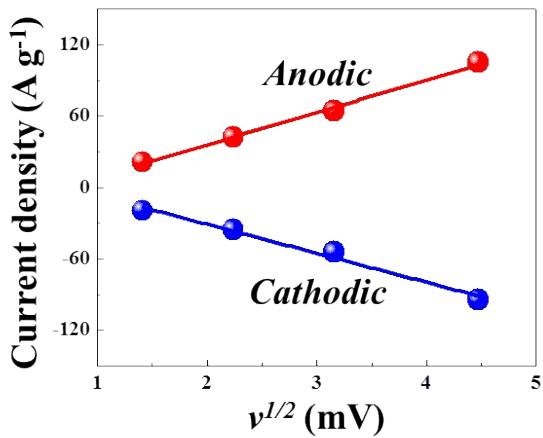
Supplementary Information

**Hierarchical CuCo<sub>2</sub>O<sub>4</sub>@NiMoO<sub>4</sub> core-shell hybrid arrays as the battery-like electrode for supercapacitors**

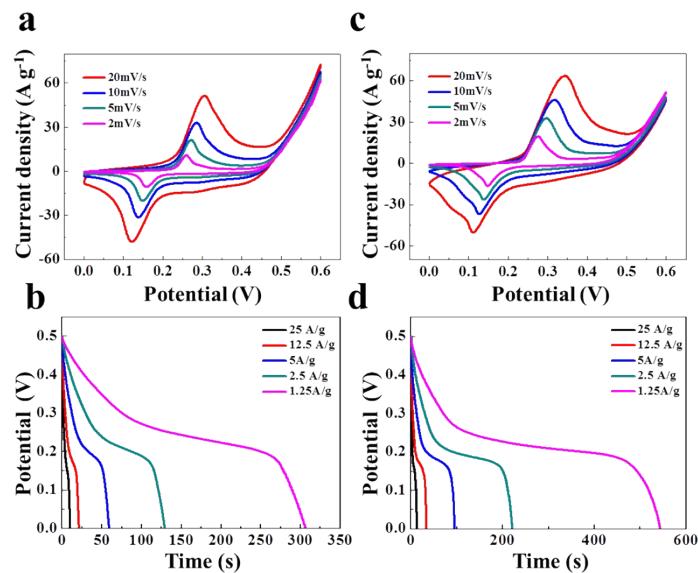
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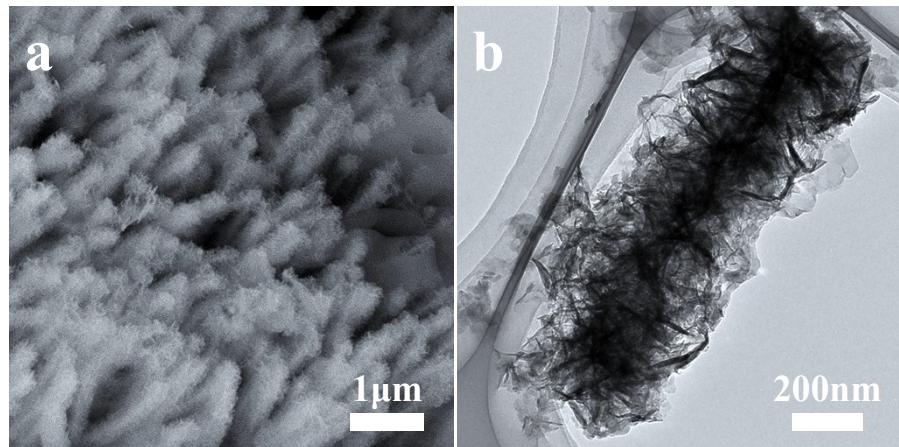
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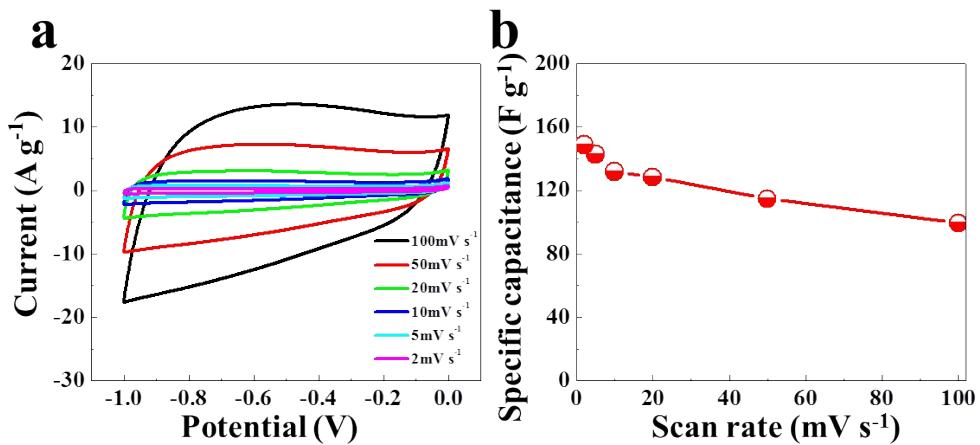
**Fig. S1** The corresponding current density ( $i$ )- $v^{1/2}$  (scant rate $^{1/2}$ ) plots in  $\text{CuCo}_2\text{O}_4@\text{NiMoO}_4$



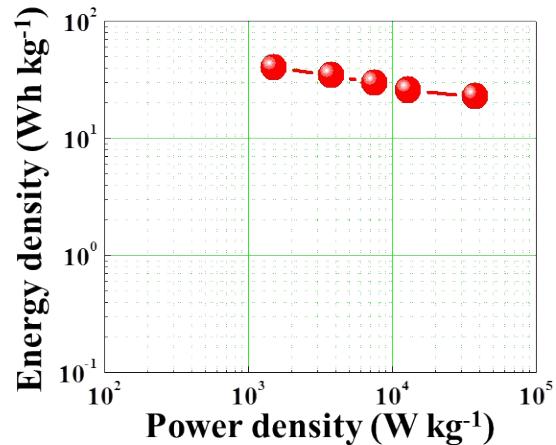
**Fig. S2** CV and galvanostatic discharge curves of (a, b)  $\text{CuCo}_2\text{O}_4$  and (c, d)  $\text{NiMoO}_4$  at various scan rates and current densities



**Fig. S3** (a) Typical SEM and (b) TEM images of  $\text{CuCo}_2\text{O}_4@\text{NiMoO}_4$  after 5000 cycles.



**Fig. S4** (a) CV curves of AC at various scan rate and (b) corresponding specific capacitance.



**Fig. S5** Ragone plots of the  $\text{CuCo}_2\text{O}_4@\text{NiMoO}_4/\text{AC}$  devices

**Table S1.** The specific capacitance of various electrodes in the three-electrode system in references

electrode materials	electrolyte	specific capacitance	reference
NiMoO <sub>4</sub> @MnO <sub>2</sub> core-shell nanorod/nanosheet	2 M KOH	1136 F g <sup>-1</sup>	1
graphitized carbon nanofibers /NiMoO <sub>4</sub> nanoparticles	6 M KOH	1438 F g <sup>-1</sup>	2
CoMoO <sub>4</sub> @NiMoO <sub>4</sub>	2 M KOH	1582 F g <sup>-1</sup>	3
CoMoO <sub>4</sub> @NiMoO <sub>4</sub> nanosheet arrays	2 M KOH	1639.8 F g <sup>-1</sup>	4
one-dimensional NiMoO <sub>4</sub>	1 M KOH	1335 F g <sup>-1</sup>	5
NiMoO <sub>4</sub> @MnO <sub>2</sub> nanosheet arrays	3 M KOH	976 F g <sup>-1</sup>	6
CuCo <sub>2</sub> O <sub>4</sub> nanoflowers	2 M KOH	1075.2 F g <sup>-1</sup>	7
CoMoO <sub>4</sub> –NiMoO <sub>4</sub> nanosheet-nanotubes	3 M KOH	751 F g <sup>-1</sup>	8
NiMoO <sub>4</sub> nanosheet	1 M KOH	1483 F g <sup>-1</sup>	9
MnO <sub>2</sub> @NiMoO <sub>4</sub>	2 M KOH	582.2 F g <sup>-1</sup>	10
CuCo <sub>2</sub> O <sub>4</sub> @CuCo <sub>2</sub> O <sub>4</sub>	2 M KOH	888.9 F g <sup>-1</sup>	11
flower-like CuCo <sub>2</sub> O <sub>4</sub> nanosheets	6 M KOH	1131 F g <sup>-1</sup>	12
CuCo <sub>2</sub> O <sub>4</sub> /CuO	1 M KOH	781 F g <sup>-1</sup>	13
CuCo <sub>2</sub> O <sub>4</sub> /MnCo <sub>2</sub> O <sub>4</sub>	2 M KOH	1434 F g <sup>-1</sup>	14
Ni-Co@Ni-Co layered double hydroxide	1 M NaOH	2200 F g <sup>-1</sup>	15
Co <sub>3</sub> O <sub>4</sub> @Ni(OH) <sub>2</sub>	3 M KOH	1306F g <sup>-1</sup>	16
NiCo <sub>2</sub> O <sub>4</sub> @NiWO <sub>4</sub>	6 M KOH	1384 F g <sup>-1</sup>	17
CoO@MnO <sub>2</sub>	6 M KOH	1835 F g <sup>-1</sup>	18
Co <sub>3</sub> O <sub>4</sub> @CoMoO <sub>4</sub>	2 M KOH	1902 F g <sup>-1</sup>	19
MnCo <sub>2</sub> O <sub>4</sub> @Ni(OH) <sub>2</sub>	2 M KOH	2154 F g <sup>-1</sup>	20
CuCo <sub>2</sub> O <sub>4</sub> @NiMoO <sub>4</sub>	6M KOH	2207 F g <sup>-1</sup>	This work

## Notes and references

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