

## Electronic Supplementary Information

### **In-situ preparation of uniform SnO<sub>2</sub> nanocrystals anchored within mesoporous carbon network as advanced anode materials**

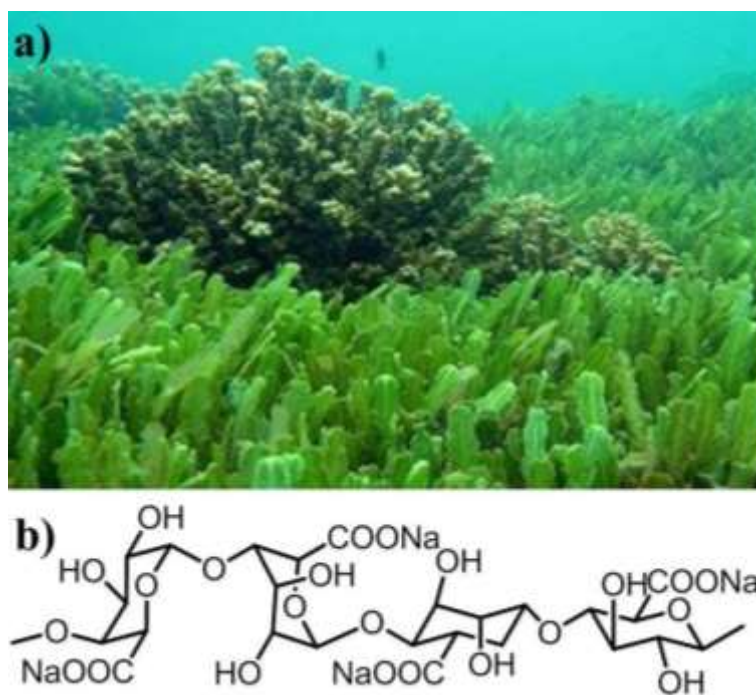
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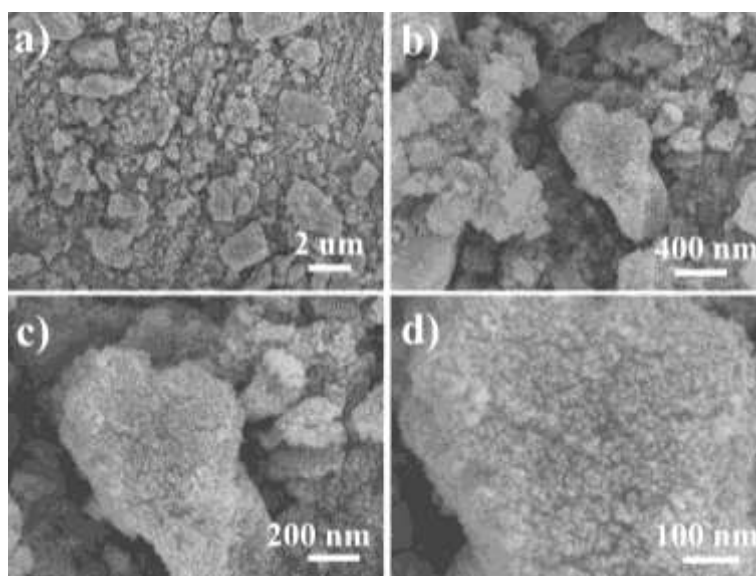
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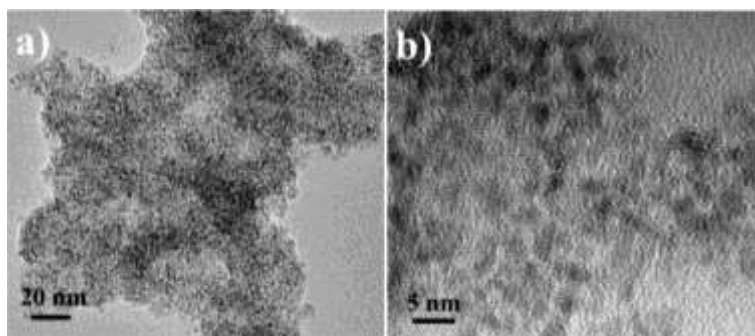
E-mail: weijxu@hnu.edu.cn (W. J. Xu); Fax: +86-731-88821549.



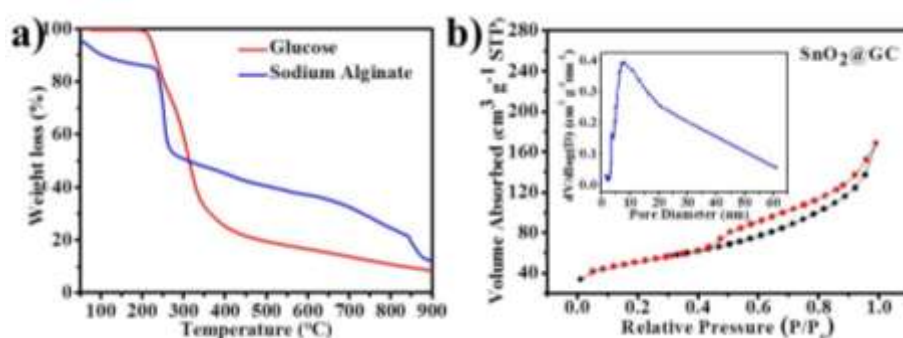
**Fig. S1.** a) Photograph of seaweeds. b) Chemical Structure of sodium alginate.



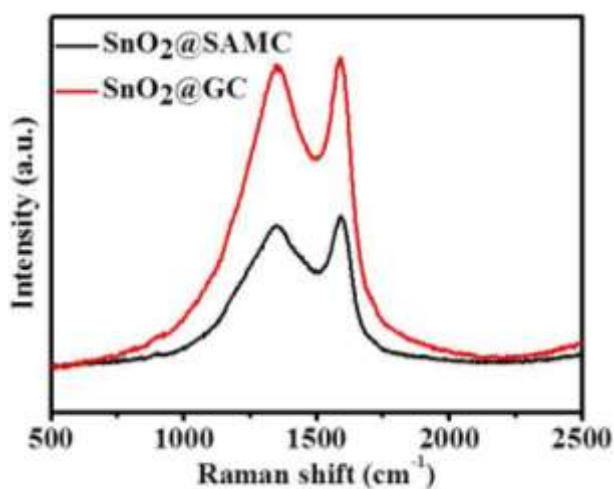
**Fig. S2.** SEM images of the  $\text{SnO}_2\text{@SAMC}$  composite with different magnifications: a) 2  $\mu\text{m}$ . b) 400 nm, c) 200 nm, d) 100 nm.



**Fig. S3.** Higher magnification TEM images of  $\text{SnO}_2\text{@SAMC}$  composite before calcination.

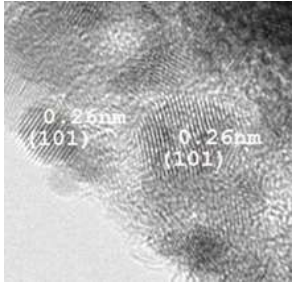
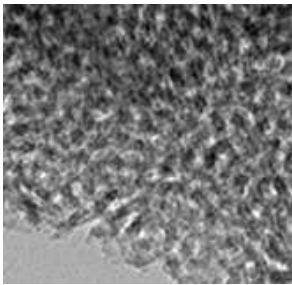
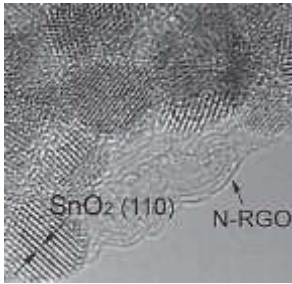
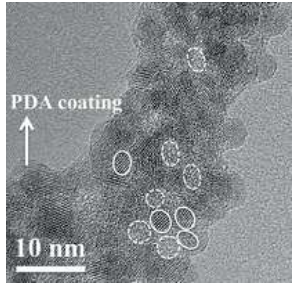
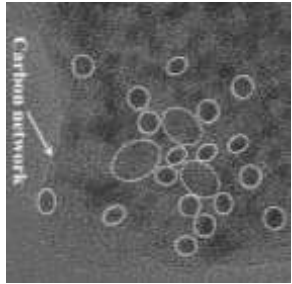


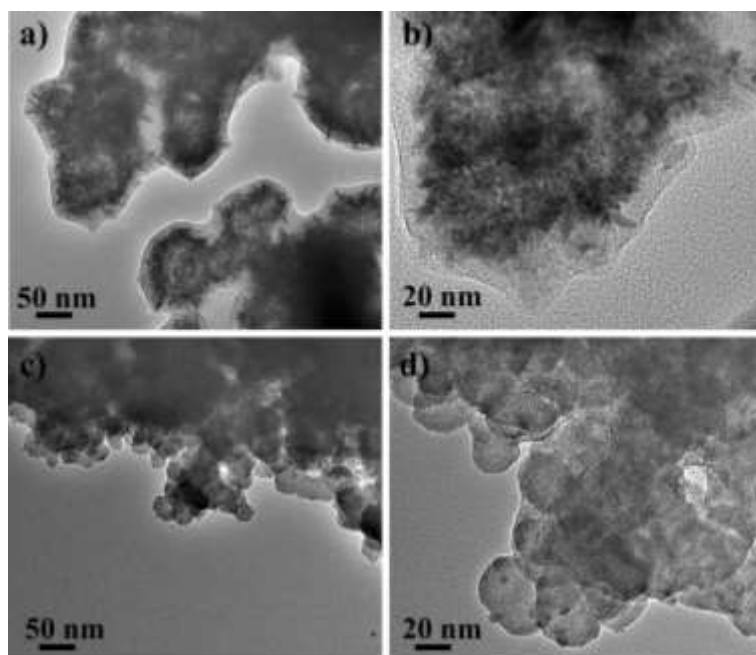
**Fig. S4.** a) Thermogravimetric analysis of sodium alginate and glucose under nitrogen atmosphere. Heating rate:  $10^\circ\text{C min}^{-1}$ . b)  $\text{N}_2$  adsorption and desorption isotherms of  $\text{SnO}_2\text{@GC}$  composite, the inset shows the pore-size distribution.



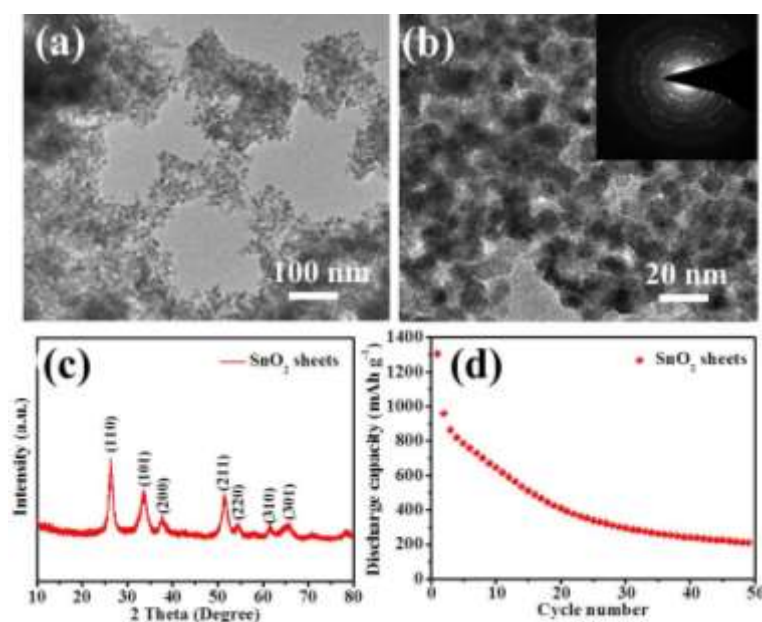
**Fig. S5.** Raman spectra for  $\text{SnO}_2\text{@SAMC}$  and  $\text{SnO}_2\text{@GC}$  composites.

**Table S1.** Comparison of synthesizing methods for confinement of 0D crystalline SnO<sub>2</sub> particles within carbonaceous materials

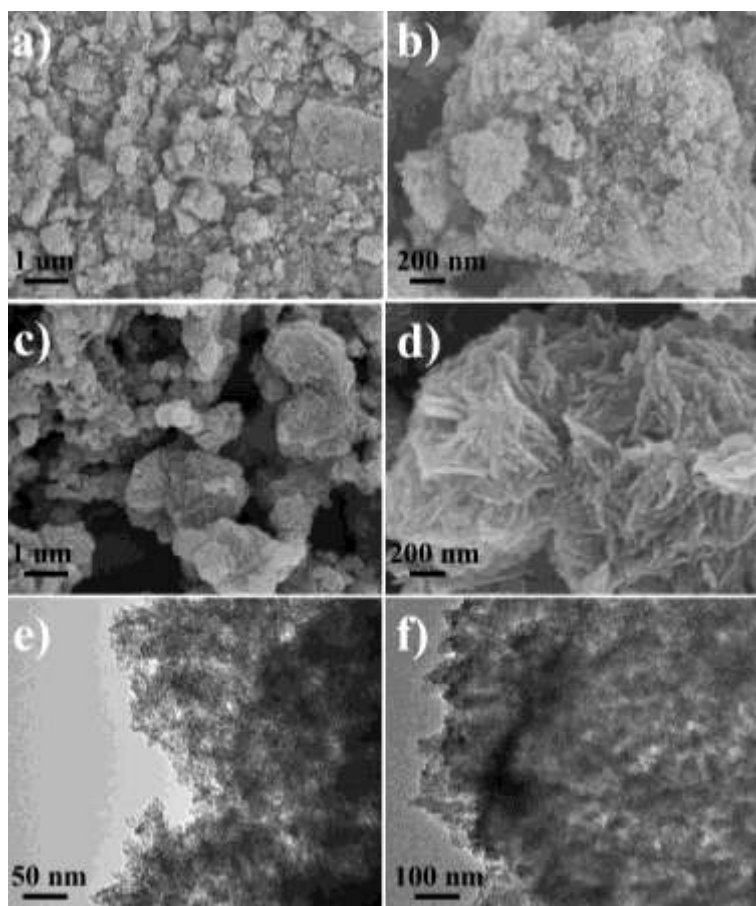
| Materials                                                 | Methods                                                           | Reactants                                                                                                                       | Morphology                                                                           | Particle Sizes                     | References |
|-----------------------------------------------------------|-------------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------|------------------------------------|------------|
| Sn/SnO <sub>2</sub> particles within mesoporous carbon    | hard-template and liquid impregnation                             | mesoporous SBA-15; Pluronic P123; sucrose; HF; H <sub>2</sub> SO <sub>4</sub> ; SnCl <sub>2</sub> ·2H <sub>2</sub> O; etc.      |    | not reported                       | 1          |
| SnO <sub>2</sub> particles within micro/mesoporous carbon | soft-template and liquid impregnation                             | phloroglucinol; Pluronic F127; organic solvent; HCl; H <sub>2</sub> O <sub>2</sub> ; SnCl <sub>4</sub> ·5H <sub>2</sub> O; etc. |    | 0.8-4 nm                           | 2          |
| SnO <sub>2</sub> particles within N-doped graphene sheets | freeze-drying and vapor reduction                                 | SnCl <sub>4</sub> ·5H <sub>2</sub> O; graphite oxide and hydrazine monohydrate.                                                 |   | 4-5 nm                             | 3          |
| Polydopamine-coated SnO <sub>2</sub> particles            | ATRP, hydrolysis, hydrothermal reaction and polydopamine coating; | Hydroxypropyl cellulose, Na <sub>2</sub> SnO <sub>3</sub> ·3H <sub>2</sub> O, dopamine hydrochloride, etc.                      |  | average size of approximately 5 nm | 4          |
| SnO <sub>2</sub> @SAMC                                    | ion exchange, hydrothermal and calcination                        | sodium alginate and SnCl <sub>2</sub> ·2H <sub>2</sub> O                                                                        |  | 2.2-3.8 nm                         | this work  |



**Fig. S6.** TEM images of SnO<sub>2</sub>@GC composite: a, b) Freshly prepared electrode before cycling with different magnification, and c, d) different magnification after cycling at 200 mA g<sup>-1</sup> for 200 cycles.



**Fig. S7.** a and b) TEM images of SnO<sub>2</sub> sheets, inset of b) shows the SAED. c) XRD pattern for SnO<sub>2</sub> sheets, d) Cycle performance at specific current of 200 mA g<sup>-1</sup>.



**Fig. S8.** SEM images of SnO<sub>2</sub>@SAMC composite: a, b) Freshly prepared electrode before cycling with different magnification, and c, d) different magnification after cycling at 200 mA g<sup>-1</sup> for 300 cycles. TEM images of SnO<sub>2</sub>@SAMC composite: e) Freshly prepared electrode before cycling, and f) after cycling at 200 mA g<sup>-1</sup> for 300 cycles.

## References

1. H. M. Hassan , Z. W. Chen , A. P. Yu , Z. Chen and X. C. Xiao, *Electrochim. Acta*, 2013, **87**, 844-852.
2. A. Jahel , C. M. Ghimbeu, L. Monconduit and C. V. Guterl, *Adv. Energy Mater.*, 2014, **4**, 1400025.
3. X. Zhou, L. J. Wan and Y. G. Guo, *Adv. Mater.*, 2013, **25**, 2152-2157.
4. B. B. Jiang, Y. J. He, B. Li, S. Q. Zhao, S. Wang, Y. B. He and Z. Q. Lin, *Angew. Chem. Int. Ed.*, 2017, **56**, 1869-1872.