

Supplemental information

March 6, 2018

Table 1: Compounds with the CaAl_2Si_2 structure type.

compound	ref	compound	ref	compound	ref
BaCd_2As_2	[1, 2]	CaZn_2Sb_2	[3]	SrMg_2Bi_2	[4]
BaCd_2P_2	[1, 2]	EuCd_2As_2	[5]	$\text{SrMg}_2\text{N}_2^*$	[6]
BaCd_2Sb_2	[7]	EuCd_2P_2	[8]	SrMg_2Sb_2	[4]
BaMg_2As_2	[1, 4]	EuCd_2Sb_2	[8]	SrMn_2As_2	[9, 10]
BaMg_2Bi_2	[4]	EuMg_2Bi_2	[11]	SrMn_2P_2	[9]
BaMg_2P_2	[1]	EuMg_2Sb_2	[12]	SrMn_2Sb_2	[13]
BaMg_2Sb_2	[4]	EuMn_2As_2	[14]	SrZn_2As_2	[15]
CaCd_2As_2	[16]	EuMn_2P_2	[14]	SrZn_2P_2	[1, 2]
CaCd_2P_2	[16]	EuMn_2Sb_2	[14, 5]	SrZn_2Sb_2	[3]
CaCd_2Sb_2	[3]	EuZn_2As_2	[2]	YbCd_2Sb_2	[17, 8]
CaMg_2As_2	[4]	EuZn_2P_2	[2]	YbMg_2Bi_2	[11]
CaMg_2Bi_2	[4, 11]	EuZn_2Sb_2	[5, 2]	YbMg_2Sb_2	[12]
$\text{CaMg}_2\text{N}_2^*$	[6]	$\text{Mg}(\text{MgMn})_2\text{As}_2$	[18]	YbMn_2As_2	[14]
CaMg_2Sb_2	[4]	MgMg_2As_2	[18]	YbMn_2Sb_2	[5, 19]
CaMn_2As_2	[9, 10]	MgMg_2Bi_2	[20]	YbZn_2As_2	[2, 21]
CaMn_2Bi_2	[13]	MgMg_2Sb_2	[20, 22]	YbZn_2P_2	[1, 23]
CaMn_2P_2	[9]	SrCd_2As_2	[15]	YbZn_2Sb_2	[2, 5, 17]
CaMn_2Sb_2	[13]	SrCd_2P_2	[1, 2]	SmMg_2Bi_2	[24]
CaZn_2As_2	[16]	SrCd_2Sb_2	[3]	SmMg_2Sb_2	[24]
CaZn_2P_2	[16]	SrMg_2As_2	[4]		
compound	ref	compound	ref	compound	ref
CaAl_2Ge_2	[25]	MgAl_2Ge_2	[26]	YbAl_2Ge_2	[27]
CaAl_2Si_2	[25]	MgAl_2Si_2	[28]	YbAl_2Si_2	[27]
EuAl_2Ge_2	[29]	SrAl_2Ge_2	[25]		
EuAl_2Si_2	[30], [27]	SrAl_2Si_2	[25]		
compound	ref	compound	ref	compound	ref
CeAgZnAs_2	[31]	LaAgZnP_2	[32]	SmAgZnAs_2	[31]
CeAgZnP_2	[31]	LaCuZnP_2	[32]	SmAgZnP_2	[32]
CeCuZnAs_2	[31]	LuCuZnP_2	[33]	SmCuZnAs_2	[31]
CeCuZnP_2	[32]	NdAgZnAs_2	[31]	SmCuZnP_2	[32]
DyCuZnAs_2	[31]	NdAgZnP_2	[31]	TbAgZnAs_2	[31]
DyCuZnP_2	[33]	NdCuZnAs_2	[31]	TbCuZnAs_2	[31]
ErCuZnP_2	[33]	NdCuZnP_2	[33]	TbCuZnP_2	[33]
GdAgZnAs_2	[31]	PrAgZnAs_2	[31]	TmCuZnP_2	[33]
GdCuZnP_2	[33]	PrAgZnP_2	[31]	YbCuZnP_2	[34]
HoCuZnP_2	[33]	PrCuZnP_2	[33]	YbMnCuP_2	[15]
LaAgZnAs_2	[31]	ScCuZnP_2	[32]	YCuZnP_2	[32, 2]

Table 2: Compounds with the CaAl_2Si_2 structure type, continued. Compounds with * are not included in phase map in figure 3(d).

compound	ref	compound	ref	compound	ref
CeLi_2As_2	[35]	LaLi_3P_2	[36]	$\text{PrLi}_x\text{Cu}_{2-y}\text{P}_2^*$	[37]
CeLi_3Bi_2	[38]	LaLi_3Sb_2	[36]	SmLi_3Bi_2	[38]
CeLi_3Sb_2	[39]	$\text{LaLi}_x\text{Cu}_{2-y}\text{P}_2^*$	[37]	SmLi_3Sb_2	[39]
DyLi_3Sb_2	[39]	$\text{MdLi}_x\text{Cu}_{2-y}\text{P}_2^*$	[37]	TbLi_3Bi_2	[38]
$\text{ErLi}_x\text{Cu}_{2-y}\text{P}_2^*$	[37]	NdLi_3As_2	[36]	TbLi_3Sb_2	[39]
GdLi_3Bi_2	[38]	NdLi_3Bi_2	[38]	YLi_3Bi_2	[36]
$\text{GdLi}_x\text{Cu}_{2-y}\text{P}_2^*$	[37]	NdLi_3Sb_2	[39]	YLi_3Sb_2	[36]
HoLi_3Sb_2	[39]	PrLi_3Bi_2	[38]		
LaLi_3Bi_2	[38]	PrLi_3Sb_2	[39]		
compound	ref	compound	ref	compound	ref
$\text{NaCd}_{1.5}\text{Sn}_{0.5}\text{As}_2$	[40]	$\text{NaZn}_{1.5}\text{Sn}_{0.5}\text{As}_2$	[40]	$\text{RbCd}_{1.5}\text{Sn}_{0.5}\text{As}_2$	[40]
$\text{NaZn}_{1.5}\text{Ge}_{0.5}\text{As}_2$	[40]	$\text{KZn}_{1.5}\text{Sn}_{0.5}\text{As}_2$	[40]		
$\text{NaZn}_{1.5}\text{Si}_{0.5}\text{As}_2$	[40]	$\text{KCd}_{1.5}\text{Sn}_{0.5}\text{As}_2$	[40]		
compound	ref	compound	ref	compound	ref
$\text{NaFe}_{1.6}\text{S}_2$	[41]				
compound	ref	compound	ref	compound	ref
CeAl_2Ge_2	[42, 43]	LuAl_2Ge_2	[44]	TbAl_2Si_2	[45]
DyAl_2Si_2	[45]	NdAl_2Ge_2	[44]	YAl_2Ge_2	[46, 44]
GdAl_2Ge_2	[44]	PrAl_2Si_2	[47]	YAl_2Si_2	[46, 45]
GdAl_2Si_2	[48]	SmAl_2Si_2	[45]		
LaAl_2Ge_2	[42, 44]	TbAl_2Ge_2	[44]		
GdAlZnGe_2	[45]				

Table 3: Compounds with $\text{VEC}=16$ forming the ThCr_2Si_2 structure type.

compound	ref	compound	ref	compound	ref
BaMn_2As_2	[10]	BaPd_2As_2	[49]	$\text{KZn}_{1.5}\text{Si}_{0.5}\text{As}_2$	[50]
BaMn_2Bi_2	[51]	BaZn_2P_2	[52]	$\text{RbZn}_{1.5}\text{Ge}_{0.5}\text{As}_2$	[50]
BaMn_2P_2	[9]	KFe_2AsSe	[53]		
BaMn_2Sb_2	[7]	$\text{KZn}_{1.5}\text{Ge}_{0.5}\text{As}_2$	[50]		

Table 4: Compounds with $\text{VEC}=16$ forming the BaCu_2S_2 structure type.

compound	ref	compound	ref	compound	ref
BaAl_2Si_2	[54]	$\alpha\text{-BaCu}_2\text{S}_2$	[55]	BaAl_2Ge_2	[56]
BaZn_2As_2	[52]	$\alpha\text{-BaCu}_2\text{Se}_2$	[55]		
BaZn_2Sb_2	[7]	$\alpha\text{-BaCu}_2\text{Te}_2$	[57]		

Table 5: Source data for Figure 8b) and Figure 9. The room temperature Seebeck coefficients, α , electrical conductivity, σ , Hall mobility, μ_H , and total thermal conductivity as κ_{total} , are obtained from the literature. The Lorenz numbers, L , are estimated using an effective mass model. The electronic thermal conductivity, κ_e , and lattice thermal conductivity, κ_L , are calculated from the Wiedemann–Franz law using the corresponding σ . The transverse and longitudinal speed of sound (v_t and v_l) are from the predicted elastic modulus [58] and density. The mean speed of sound, v_s calc, is obtained through $(2v_t+v_l)/3$.

Ref.	Compound	α $\mu V K^{-1}$	$L \times 10^8$ $W \Omega K^{-2}$	σ S/m	κ_e $W/(mK)$	κ_{total} $W/(mK)$	κ_L $W/(mK)$	v_s calc. m/s	μ_H $cm^2/(Vs)$
[59]	Mg ₃ Sb ₂	193	1.58	2.11×10^2	9.8×10^{-4}	1.33	1.3	2790	
[60]	YbCd ₂ Sb ₂	115	1.75	4.56×10^4	2.40×10^{-1}	2.04	1.80	2160	73
	CaCd ₂ Sb ₂	261	1.55	2.50×10^2	1.16×10^{-3}	1.06	1.06	2381	
[61]	YbZn ₂ Sb ₂	48	2.3	3.13×10^5	2.16	4.25	2.09	2379	130
	CaZn ₂ Sb ₂	120	1.81	4.22×10^4	2.29×10^{-1}	2.60	2.37	2480	83
[62]	YbCd ₂ Sb ₂	124	1.82	4.90×10^4	2.68×10^{-1}	2.25	1.99	2117	72
	YbCd _{1.95} Mn _{0.25} Sb ₂	138	1.75	4.25×10^3	2.23×10^{-2}	1.95	1.93	2117	73
	YbMn ₂ Sb ₂	10	2.4	6.25×10^3	4.50×10^{-2}	2.75	2.71	2055	
[63]	CaZn ₂ Sb ₂	115	1.75	8.00×10^4	4.20×10^{-1}	2.91	2.49	2480	
	YbZn ₂ Sb ₂	100	1.9	1.33×10^5	7.58×10^{-1}	2.60	1.84	2379	
[64]	EuMg ₂ Bi ₂	102	1.9	9.35×10^4	5.33×10^{-1}	4.72	4.19	2228	
	CaMg ₂ Bi ₂	288	1.54	5.30×10^3	2.45×10^{-2}	2.93	2.91	2486	143
	YbMg ₂ Bi ₂	209	1.6	1.33×10^4	6.38×10^{-2}	2.68	2.62	2133	119
[65]	SrZn ₂ Sb ₂	159	1.69	2.22×10^4	1.13×10^{-1}	2.05	1.94	2366	
	CaZn ₂ Sb ₂	120	1.81	4.22×10^4	2.29×10^{-1}	2.60	2.37	2480	
	YbZn ₂ Sb ₂	48	2.3	3.13×10^5	2.16	4.25	2.09	2379	
	EuZn ₂ Sb ₂	120	1.81	1.14×10^5	6.17×10^{-1}	2.50	1.88	2401	
[66]	YbCd ₂ Sb ₂	118	1.75	4.60×10^4	2.42×10^{-1}	2.10	1.86	2160	73
	YbZn ₂ Sb ₂	45	2.2	2.16×10^5	1.43	3.80	2.37	2379	119
[67]	CaZn ₂ Sb ₂	120	1.81	3.20×10^4	1.74×10^{-1}	3.89	3.72	2480	63.6
[68]	YbZn ₂ Sb ₂	53	2.19	1.58×10^5	1.04	3.23	2.20	2379	85
[69]	EuZn ₂ Sb ₂	120	1.81	1.13×10^4	6.14×10^{-2}	2.62	2.56	2401	
[70]	EuCd ₂ Sb ₂	229	1.58	1.11×10^4	5.26×10^{-2}	1.40	1.35	2133	
[71]	EuCd ₂ Sb ₂	222	1.59	1.16×10^4	5.53×10^{-2}	1.40	1.35	2133	
	CaCd ₂ Sb ₂	262	1.55	6.67×10^4	3.10×10^{-1}	1.06	0.75	2381	
[72]	SrZn ₂ Sb ₂	167	1.69	3.50×10^3	1.77×10^{-2}	2.20	2.18	2366	
[73]	YbCd ₂ Sb ₂	169	1.69	3.30×10^4	1.67×10^{-1}	2.08	1.91	2160	
[74]	YbZn ₂ Sb ₂	58	2.16	2.10×10^5	1.36	4.13	2.76	2379	
	La _{0.01} Yb _{0.99} Zn ₂ Sb ₂	63	2.14	1.67×10^5	1.07	3.50	2.43	2379	
[75]	YbZn ₂ Sb ₂	130	1.52	4.55×10^4	2.07×10^{-1}	2.90	2.69	2379	115
	Yb _{0.99} Zn ₂ Sb ₂	65	2.14	1.67×10^5	1.07	2.40	1.33	2379	90
[76]	Mg ₃ Sb ₂	313	1.5	9.52×10^2	4.29×10^{-3}	1.40	1.40	2790	28.3
	Mg _{2.995} Ag _{0.005} Sb ₂	162	1.68	1.22×10^4	6.15×10^{-2}	1.45	1.39	2790	50.1
	Mg _{2.990} Ag _{0.010} Sb ₂	177	1.65	1.04×10^4	5.16×10^{-2}	1.35	1.30	2790	47.6
	Mg _{2.985} Ag _{0.015} Sb ₂	177	1.65	1.18×10^4	5.82×10^{-2}	1.25	1.19	2790	47.3
	Mg _{2.980} Ag _{0.020} Sb ₂	165	1.68	1.03×10^4	5.20×10^{-2}	1.30	1.24	2790	48.9
[77]	Mg ₃ Sb ₂	1000	1.5	1.00	4.50×10^{-6}	1.40	1.35	2790	23.2
	Mg _{2.994} Na _{0.006} Sb ₂	110	1.9	1.04×10^4	5.94×10^{-2}	1.60	1.60	2790	16.7
	Mg _{0.975} Na _{0.025} Sb ₂	100	1.9	2.08×10^4	1.19×10^{-1}	1.70	1.64	2790	15.9
[78]	CaMg ₂ Sb ₂						4.55	3207	

Table 6: Source data for figure 8a). Hall carrier concentration, n_H , Seebeck coefficient, α , and Hall mobility, μ_H , for p -type AM_2X_2 compounds.

Compound [Ref.]	$n_H \times 10^{-19}$ carriers/cm ³	α $\mu V K^{-1}$	μ_H $cm^2/(Vs)$	Compound [Ref.]	$n_H \times 10^{19}$ carriers/cm ³	α $\mu V K^{-1}$	μ_H $cm^2/(Vs)$
YbZn _{2-x} Mn _x Sb ₂ [68]				Yb _{1-x} Ca _x Cd ₂ Sb ₂ [60]			
0	11.5	51	85	0	3.9	115	73
0.05	9	85	56	0.2	2.66	128	82
0.1	7.2	69	88	0.4	1.88	149	104
0.15	8.3	924	59	0.5	1.94	149	91
0.2	6.6	64	38	0.6	1.12	161	135
0.3	6	87	54	0.8	0.33	216	114
Ca _{1-x} Eu _x Zn ₂ Sb ₂ [67]				Yb _{1-x} Zn ₂ Sb ₂ [75]			
0	3.04	120	63.6	0.98	2.38	131	120
0.1	3.29	112	117.4	0.99	2.04	121	115
0.3	3.59	110	120.8	1	3.09	118	90
0.7	3.81	111	123.6	1.025	9.06	62	75
0.9	3.83	111	140	1.05	15	40	92
1	2.94	112	186.2	(Eu _{0.5} Yb _{0.5}) _{1-x} Ca _x Mg ₂ Bi ₂ [64]			
YbCd _{2-x} Zn _x Sb ₂ [66]				0	3.99	105	200
0	3.9	116	73	0.4	2.74	126	180
0.4	6	101	118	0.5	2.26	148	166
0.8	7.2	68	93	0.6	1.7	162	171
1	8.6	66	103	0.7	1.41	172	164
1.2	9.1	55	104	EuMg ₂ Bi ₂	4.33	102	202
1.6	10.3	40	123	Eu _{0.5} Ca _{0.5} Mg ₂ Bi ₂	3.07	128	159
2	11.4	45	119	Mg ₃ Sb _{2-x} Bi _x [59]			
Ca _{1-x} Na _x Mg ₂ Bi _{1.98} [79]				0	11	192	30
0	0.35	256	138	0.1	19	275	25
0.0025	1.84	160	144	0.15	23	310	22.2
0.005	2.53	138	147	0.2	29	305	18.6
0.0075	4.4	103	154	0.25	42	237	13.9
Ca _{1-x} Yb _x Mg ₂ Bi ₂ [80]				0.4	150	141	36.9
0	0.24	288	143	Mg _{3-x} Ag _x Sb ₂ [76]			
0.3	0.28	263	153	0	0.18	316	28.3
0.5	0.34	260	138	0.05	1.52	163	50.1
0.7	0.42	247	131	0.02	1.36	167	47.6
1	0.72	210	119	0.025	1.27	174	47.3
YbCd _{2-x} Mn _x Sb ₂ [62]				0.01	1.23	175	48.9
0	4.3	125	72	0.015	1.5	178	43.7
0.05	3.6	139	73	Mg _{3-x} Na _x Sb ₂ [77]			
0.1	2.8	135	63	0.006	3.87	117	16.7
0.15	2.2	162	64	0.02	1.36	167	47.6
0.2	2.1	161	74	0.0125	8.41	89	15.9
0.4	1.7	181	54	0.025	17.1	73	12.6
0.6	1.1	192	53	0.015	1.5	177	43.7
Ca _x Yb _{1-x} Zn ₂ Sb ₂ [61]							
0	15	48	130				
0.25	8.9	58	72				
0.5	6.4	79	76				
0.75	5.7	96	50				
1	3.1	120	83				

Table 7: Hall carrier concentration, Seebeck coefficient and Hall mobility for all n -type AM_2X_2 compounds. Source data for figure 7(b) in the paper. Here, we define Hall carrier concentration as n_H , Seebeck coefficient as α , Hall mobility as μ_H .

Compound [Ref.]	$n_H \times 10^{-19}$ carriers/ cm^3	α $\mu_H V K^{-1}$	n_H $cm^2 / (Vs)$
Mg _{3.07} Sb _{1.5} Bi _{0.5-x} Se _x [81]			
0.02	0.88	276	76
0.03	0.70	280	64
0.04	0.68	281	63
0.05	0.77	279	60
0.06	0.64	296	61
Mg ₃ Sb _{1.5-0.5x} Bi _{0.5-0.5x} Te _x [82]			
0.04	2.21	208	
0.05	2.17	214	
0.08	2.06	221	
0.2	1.81	227	

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