Supporting Information

Recyclable, Bifunctional Composites of Perovskite Type N-CaTiO₃ and Reduced Graphene Oxide as an Efficient Adsorptive Photocatalyst for Environmental Remediation

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Figure S1. XRD pattern of GO and RGO



Figure S2. Raman spectra of GO and RGO



Figure S3. Elemental mapping of CT.



Figure S4. Elemental mapping of NCT.



Figure S5. Plot of transformed Kubelka-Munk function vs the energy of light: (a) CT (b) NCT



Figure S6. Time dependent absorption spectra of MB degradation: (a) Pure MB (b) CT (c) NCT (d) NCTG01 (e) NCTG03 (f) NCTG05 (g) NCTG10 and (h) NCTG20.



Figure S7. Time dependent absorption spectra of TBZ degradation: (a) Pure TBZ (b) CT (c) NCT and (d) NCTG20.

SI. No.	Photocatalyst	Rate constant k	t _{1/2} (min)	R ²
		(min ⁻¹)		
1	Bare MB	0.0002	2898.14	0.9300
2	СТ	0.0009	766.34	0.9719
3	NCT	0.0016	420.00	0.9899
4	NCTG01	0.0022	308.00	0.9805
5	NCTG03	0.0028	250.18	0.9954
6	NCTG05	0.0036	194.66	0.9808
7	NCTG10	0.0061	114.36	0.9896
8	NCTG20	0.0112	61.88	0.8838

Table S1. Summary of kinetic data of photocatalytic degradation of MB using all preparedphotocatalysts under visible light irradiation



Figure S8. Adsorption/degradation by NCTG20 composite: (a) 1×10^{-5} M (b) 2×10^{-5} M (c) 3×10^{-5} M (d) 4×10^{-5} M (e) 5×10^{-5} M MB and (f) dye adsorption isotherm of MB on NCTG20 composite.

Langmuir model			Freundlich model		
q _m	KL	R ²	K _f	1/n	R ²
16.37	0.00631	0.985	8.234	0.1842	0.936

Table S2. Summary of adsorption parameters



Figure S9. Raman spectra of MB¹ and recovered catalyst after third cycle









Figure S10. Mass spectra of degraded MB solution by NCTG20 composite under visible light

References

1. D. Volpati, P. H. Aoki, C. A. Dantas, F. V. Paulovich, M. C. F. de Oliveira, O. N. Oliveira Jr, A. Riul Jr, R. F. Aroca and C. J. Constantino, *Langmuir*, 2011, 28, 1029-1040.