## Supporting Information

Visible-light-induced a tandem oxidative cyclization of 2alkynylanilines with disulfides(diselenides) to 3-sulfenyl- and 3selenylindoles under transition-metal-free and photocatalyst-free conditions<br>Qing Shi, ${ }^{\text {a }}$ Pinhua Li, ${ }^{* a}$ Yan Zhang, ${ }^{\text {a }}$ and Lei Wang*a,b<br>${ }^{a}$ Department of Chemistry, Huaibei Normal University, Huaibei, Anhui 235000, P. R. China, Tel: +86-561-3802-069 Fax: +86-561-3090-518 E-mail: leiwang@chnu.edu.cn<br>${ }^{b}$ State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, P. R. China

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## 1. Optimization of the reaction conditions ${ }^{a}$ (Table S1)

| Entry | $\text { SeSePh } \xrightarrow[\mathrm{N}_{2}, \text { r.t., } 12 \mathrm{~h}]{\text { Solvent }(2 \mathrm{~mL})}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | Oxidant | Solvent | Light Source | Yield ${ }^{\text {b }}$ (\%) |
| 1 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | PhCl | Blue LED | 72 |
| 2 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | PhMe | Blue LED | 75 |
| 3 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | Acetone | Blue LED | 28 |
| 4 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | Blue LED | 97 |
| 5 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | DMC | Blue LED | 33 |
| 6 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | DCE | Blue LED | 54 |
| 7 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | MeOH | Blue LED | n.r. |
| 8 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | EtOH | Blue LED | n.r. |
| 9 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | DMF | Blue LED | n.r. |
| 10 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | DMSO | Blue LED | n.r. |
| 11 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | 1,4-Dioxane | Blue LED | n.r. |
| 12 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | Blue LED | $69^{c}, 88^{d}$ |
| 13 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | Blue LED | $73^{e}, 98^{f}$ |
| 14 | $\mathrm{H}_{2} \mathrm{O}_{2}$ | $\mathrm{CH}_{3} \mathrm{CN}$ | Blue LED | $68^{\text {g }, ~} 83^{h}, 97^{i}$ |

${ }^{a}$ Reaction conditions: 1a ( 0.20 mmol ), $\mathbf{4 a}(0.125 \mathrm{mmol}), \mathrm{H}_{2} \mathrm{O}_{2}(1.0$ equiv, $30 \%$ aqueous solution), solvent ( 2.0 mL ), nitrogen atmosphere at room temperature under 3 W LED irradiation for $12 \mathrm{~h} .{ }^{b}$ isolated yield of the product based on compound 1a; ${ }^{c} \mathrm{H}_{2} \mathrm{O}_{2}$ ( 0.50 equiv, $30 \%$ aqueous solution) was used. ${ }^{d} \mathrm{H}_{2} \mathrm{O}_{2}$ (1.5 equiv, $30 \%$ aqueous solution) was added. ${ }^{e} \mathbf{4 a}(0.10 \mathrm{mmol})$ was used. ${ }^{f} \mathbf{4 a}$ ( 0.15 mmol ) was added. $g 6 \mathrm{~h} .{ }^{h} 9 \mathrm{~h} .{ }^{i} 15 \mathrm{~h}$.

## 2. Reaction mechanism study

### 2.1 Free radical-trapping experiment (I)



4-Methyl- $N$-(2-(phenylethynyl)phenyl)benzenesulfonamide (1a, 0.20 mmol ), diphenyl disulfide ( $\mathbf{2 a}, 0.125 \mathrm{mmol}$ ), $\mathrm{H}_{2} \mathrm{O}_{2}$ ( 1.0 equiv, $30 \%$ aqueous solution) and chlorobenzene ( 2.0 mL ) and (2,2,6,6-tetramethylpiperidine)-1-oxyl (TEMPO, 0.40 mmol, 2.0 equiv) was added to an oven-dried Schlenk tube equipped with magnetic stirring bar, and the reaction tube was irradiated using 3 W blue LED at room temperature under $\mathrm{N}_{2}$ atmosphere for 21 h . After the reaction was completed, no desired product 3a was obtained, and a free radical-trapping adduct of PhS • with TEMPO was observed by HPLC-HRMS analysis of the reaction solution (Figure S1).


Figure S1. HRMS analysis of the adduct of $\mathrm{PhS} \bullet$ with TEMPO

### 2.2 Free radical-trapping experiment (II)



4-Methyl- $N$-(2-(phenylethynyl)phenyl)benzenesulfonamide (1a, 0.20 mmol ), $\mathrm{H}_{2} \mathrm{O}_{2}$ (1.0 equiv, $30 \%$ aqueous solution) and chlorobenzene ( 2.0 mL ) and ( $2,2,6,6-$ tetramethylpiperidine)-1-oxyl (TEMPO, $0.40 \mathrm{mmol}, 2.0$ equiv) was added to an ovendried Schlenk tube equipped with magnetic stirring bar, and the reaction tube was irradiated using 3 W blue LED at room temperature under $\mathrm{N}_{2}$ atmosphere for 21 h . After the reaction was completed, a free radical-trapping adduct (6) of indole radical with TEMPO, which was confirmed by HPLC-HRMS and GC-MS analysis of the reaction solution (Figure S2 and S3).


Figure S2. HRMS analysis of the adduct of adduct (6) of indole radical with TEMPO


Figure S3. GC-MS analysis of the adduct of adduct (6) of indole radical with TEMPO

## 3. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of the products




$-21.518$

$\begin{array}{llllllllllllllllllllllll}190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & & \text { ppm }\end{array}$


3b



3b














3h



 mon $\dot{\sim}$











$3 n$








$\stackrel{n}{\circ}$
$\stackrel{1}{N}$
$\stackrel{1}{2}$

3s




$\begin{array}{lllllllllllllllllll}190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & \mathrm{ppm}\end{array}$


-

3u









$3 x$





N $\qquad$















$\begin{array}{ll}m \\ m & \stackrel{\bullet}{\infty} \\ \dot{m} & \stackrel{\sim}{n} \\ & \stackrel{\sim}{n}\end{array}$





##  <br> 40



5 g













