

*Supporting Information for*

**Synthesis of dibromo- and tetrabromo-bipyrrolines and their corresponding  
2,6-diazasemibuvallene derivatives**

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## 1) General information

$^1\text{H}$  (300 MHz) and  $^{13}\text{C}$  (75 MHz) NMR spectra were measured in  $\text{CDCl}_3$  and  $^1\text{H}$  (400 MHz) and  $^{13}\text{C}$  (100 MHz) NMR spectra were measured in  $\text{CDCl}_3$  or  $\text{C}_6\text{D}_6$ . Chemical shifts ( $\delta$ ) are reported in ppm downfield from tetramethylsilane. Single crystal X-ray data for **4a** and **6** were collected at a temperatures of 180 K, respectively, using monochromated Mo  $\text{K}\alpha$  radiation. The structures were solved by direct method and refined by full-matrix least-squares on F2 for all data using Olex227 and SHELXTL software. Crystallographic data for the structures have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication CCDC 1056585 (**4a**) and CCDC 1056586 (**6**).

## 2) General procedure for the preparation of 2-6

**General Procedure for the Preparation of  $\alpha,\alpha'$ -Dibromo- $\Delta^1$ -bipyrrolines 2.** NBS (4.4 mmol, 783 mg) was added to a solution of  $\Delta^1$ -bipyrrolines **1** (2.0 mmol) in 20 ml  $\text{CCl}_4$  in a 50 ml Schlenk tube, the mixture was stirred at 80 °C for 12 h. The reaction mixture was then quenched with water and extracted with acetic ether. The extract was washed with brine and dried over  $\text{MgSO}_4$ . The solvent was evaporated in vacuo to give yellow solid, which was purified by column chromatography (Petrol Ether: Ethyl Acetate = 50:1) to afford products  $\alpha,\alpha'$ -dibromo- $\Delta^1$ -bipyrrolines **2a-f**.

**2a**, colorless solid, yield: 48% (452 mg);  $^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta$  = 1.46-1.78 (m, 4H,  $\text{CH}_2$ ), 2.01-2.07 (m, 2H,  $\text{CH}_2$ ), 2.73-2.81 (m, 2H,  $\text{CH}_2$ ), 5.56 (s, 2H, CH), 7.27-7.35 (m, 6H, CH), 7.74-7.77 (m, 4H, CH);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta$  = 22.36 (2  $\text{CH}_2$ ), 33.44 (2  $\text{CH}_2$ ), 56.03 (2 CH), 81.14 (2 quat. C), 128.45 (4 CH), 128.49 (4 CH), 131.27 (2 CH), 131.47 (2 quat. C), 168.09 (2 quat. C); HRMS calcd. for  $\text{C}_{22}\text{H}_{21}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 471.0066, found 471.0058.

**2b**, yellow solid, yield: 55% (644 mg);  $^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta$  = 1.18-2.06 (m, 6H, CH; 30H,  $\text{CH}_2$ ), 2.51-2.57 (m, 2H,  $\text{CH}_2$ ), 4.94 (s, 2H, CH);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta$  = 22.60 (2  $\text{CH}_2$ ), 28.23 (6 CH), 32.48 (2  $\text{CH}_2$ ), 36.46 (6  $\text{CH}_2$ ), 37.66 (2 quat. C), 40.04 (6  $\text{CH}_2$ ), 56.22 (2 CH), 79.36 (2 quat. C), 179.26 (2 quat. C); HRMS calcd. for  $\text{C}_{30}\text{H}_{41}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 587.1631, found 587.1630.

**2c**, yellow solid, yield: 49% (488 mg);  $^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta$  = 1.55-1.79 (m, 4H,  $\text{CH}_2$ ), 2.08-2.15 (m, 2H,  $\text{CH}_2$ ), 2.36 (s, 6H,  $\text{CH}_3$ ), 2.82-2.86 (m, 2H,  $\text{CH}_2$ ), 5.62 (s, 2H,

CH), 7.20 (d,  $J = 8.4$  Hz, 4H, CH), 7.73 (d,  $J = 8.4$  Hz, 4H, CH);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 21.38$  (2  $\text{CH}_3$ ), 22.29 (2  $\text{CH}_2$ ), 33.41 (2  $\text{CH}_2$ ), 56.21 (2 CH), 80.99 (2 quat. C), 128.54 (4 CH), 128.85 (2 quat. C), 129.26 (4 CH), 141.84 (2 quat. C), 168.15 (2 quat. C); HRMS calcd. for  $\text{C}_{24}\text{H}_{25}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 499.0379, found 499.0379.

**2d**, yellow solid, yield: 42% (360 mg);  $^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 1.15$  (s, 18H,  $\text{CH}_3$ ), 1.32-1.39 (m, 4H,  $\text{CH}_2$ ), 2.00-2.07 (m, 2H,  $\text{CH}_2$ ), 2.52-2.56 (m, 2H,  $\text{CH}_2$ ), 4.97 (s, 2H, CH);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 22.54$  (2  $\text{CH}_2$ ), 29.48 (6  $\text{CH}_3$ ), 32.33 (2 CH), 35.58 (2 quat. C), 56.68 (2 CH), 80.12 (2 quat. C), 179.55 (2 quat. C); HRMS calcd. for  $\text{C}_{18}\text{H}_{29}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 431.0692, found 431.0688.

**2e**, yellow solid, yield: 75%(606 mg);  $^1\text{H}$  NMR (300MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 1.22$  (s, 18H,  $\text{CH}_3$ ), 1.54 (s, 6H,  $\text{CH}_3$ ), 4.95 (s, 2H, CH);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 19.24$  (2  $\text{CH}_3$ ), 29.43 (6  $\text{CH}_3$ ), 35.78 (2 quat. C), 56.10(2 CH), 83.01 (2 quat. C), 181.13 (2 quat. C); HRMS calcd. for  $\text{C}_{16}\text{H}_{27}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 405.0463, found 405.0470.

**2f**, yellow solid, yield: 73%(712 mg);  $^1\text{H}$  NMR (400MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 0.93$  (t,  $J = 7.2$  Hz, 6H,  $\text{CH}_3$ ), 1.21 (s, 18H,  $\text{CH}_3$ ), 1.32-1.37 (m, 2H,  $\text{CH}_2$ ), 1.52-1.60 (m, 2H,  $\text{CH}_2$ ), 1.67-1.74 (m, 2H,  $\text{CH}_2$ ), 1.90-1.98 (m, 2H,  $\text{CH}_2$ );  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 14.03$  (2  $\text{CH}_3$ ), 23.37 (2  $\text{CH}_2$ ), 26.97 (2  $\text{CH}_2$ ), 29.48 (6  $\text{CH}_3$ ), 32.33 (2  $\text{CH}_2$ ), 35.88 (2 quat. C), 54.51 (2 CH), 86.00 (2 quat. C), 180.41 (2 quat. C); HRMS calcd. for  $\text{C}_{22}\text{H}_{39}\text{Br}_2\text{N}_2$   $[\text{M}+\text{H}]^+$ : 489.1402, found 489.1497.

**General Procedure for the Preparation of 2,6-Diazasemibuvallenes 3.** Li (1.1 mmol, 8.4 mg) was added to a solution of  $\alpha,\alpha'$ -dibromo- $\Delta^1$ -bipyrrolines **2** (0.5 mmol) in 5 ml THF in a 25 ml round-bottom flask in glove box, the mixture was stirred at room temperature for 2-4 h. The solvent was evaporated in vacuo to give brown solid. This solid was dissolved in THF- $d_8$  and monitored by NMR to confirm  $\alpha,\alpha'$ -dibromo- $\Delta^1$ -bipyrroline was totally disappeared. The THF and THF- $d_8$  was evaporated in vacuo to give brown solid again. Then the 2,6-Diazasemibuvallene **3** was extracted by mixture solvent (Hexane:Et<sub>2</sub>O = 3:1) diethyl ether from brown solid, and the salt (LiBr) was removed.

**3a**, yellow solid, yield: 81%(126mg);  $^1\text{H}$  NMR (300MHz, THF- $d_8$ , 25 °C):  $\delta = 1.29$ -1.36 (m, 4H,  $\text{CH}_2$ ), 1.68-1.78 (m, 2H,  $\text{CH}_2$ ), 2.16-2.23 (m, 2H,  $\text{CH}_2$ ), 5.55 (s, 2H, CH), 7.19-7.26 (m, 6H, CH), 7.71-7.74 (m, 4H, CH);  $^{13}\text{C}$  NMR (75 MHz, THF- $d_8$ ,  $\text{Me}_4\text{Si}$ , 25 °C):  $\delta = 22.49$  (2  $\text{CH}_2$ ), 28.94 (2

CH<sub>2</sub>), 79.47 (2 quat. C), 100.74 (2 CH), 127.56 (4 CH), 128.88 (4 CH), 129.51 (2 CH), 135.81 (2 quat. C), 153.14 (2 quat. C). Elemental Analysis Calcd (%) for C<sub>22</sub>H<sub>20</sub>N<sub>2</sub>: C, 84.58; H, 6.45; N, 8.97; Found: C, 83.94; H, 6.28; N, 8.38.

**3b** (CAS 1387574-71-1), yellow solid, yield: 75%(160 mg); <sup>1</sup>H NMR (300MHz, THF-d<sub>8</sub>, 25 °C): δ = 1.12-2.01 (m, 6H, CH; 32H, CH<sub>2</sub>), 4.73 (s, 2H, CH); <sup>13</sup>C NMR (75 MHz, THF-d<sub>8</sub>, 25 °C): δ = 22.29 (2 CH<sub>2</sub>), 28.94 (2 CH<sub>2</sub>), 29.41 (6 CH), 36.94 (2 quat. C), 37.71 (6 CH<sub>2</sub>), 41.46 (6 CH<sub>2</sub>), 79.00 (2 quat. C), 99.26 (2 CH), 163.22 (2 quat. C).

**3c**, yellow solid, yield: 79%(134mg); <sup>1</sup>H NMR (300MHz, THF-d<sub>8</sub>, 25 °C): δ = 1.28-1.35 (m, 2H, CH<sub>2</sub>), 1.61-1.72 (m, 4H, CH<sub>2</sub>), 2.11-2.36 (m, 2H, CH<sub>2</sub>), 2.23 (s, 6H, CH<sub>3</sub>), 5.47 (s, 2H, CH), 7.03 (d, J = 8.1 Hz, 4H, CH), 7.60 (d, J = 8.1 Hz, 4H, CH); <sup>13</sup>C NMR (75 MHz, THF-d<sub>8</sub>, 25 °C): δ = 21.18 (2 CH<sub>3</sub>), 22.42 (2 CH<sub>2</sub>), 28.89 (2 CH<sub>2</sub>), 79.17 (2 quat. C), 100.21 (2 CH), 127.59 (4 CH), 129.57 (4 CH), 133.26 (2 quat. C), 139.45 (2 quat. C), 153.00 (2 quat. C). Elemental Analysis Calcd (%) for C<sub>24</sub>H<sub>24</sub>N<sub>2</sub>: C, 84.67; H, 7.11; N, 8.23; Found: C, 83.84; H, 6.78; N, 7.95.

**3d** (CAS 1387574-88-0), yellow solid, yield: 92%(125 mg); <sup>1</sup>H NMR (300MHz, THF-d<sub>8</sub>, 25 °C): δ = 1.05 (s, 18H, CH<sub>3</sub>), 1.16-1.27 (m, 4H, CH<sub>2</sub>), 1.48-1.56 (m, 2H, CH<sub>2</sub>), 1.91-1.96 (m, 2H, CH<sub>2</sub>), 4.77 (s, 2H, CH); <sup>13</sup>C NMR (75 MHz, THF-d<sub>8</sub>, 25 °C): δ = 22.26 (2 CH<sub>2</sub>), 28.90 (2 CH<sub>2</sub>), 28.96 (6 CH<sub>3</sub>), 34.75 (2 quat. C), 79.56 (2 quat. C), 99.56 (2 CH), 163.25 (2 quat. C).

**3e** (CAS 1387574-76-6), yellow solid, yield: 88%(106 mg); <sup>1</sup>H NMR (300MHz, THF-d<sub>8</sub>, 25 °C): δ = 1.04 (s, 18H, CH<sub>3</sub>), 1.14 (s, 6H, CH<sub>3</sub>), 4.70 (s, 2H, CH); <sup>13</sup>C NMR (75 MHz, THF-d<sub>8</sub>, 25 °C): δ = 15.62 (2 CH<sub>3</sub>), 28.91 (6 CH<sub>3</sub>), 34.78 (2 quat. C), 80.37 (2 quat. C), 98.44 (2 CH), 162.98 (2 quat. C).

**3f** (CAS 1387574-77-7), yellow solid, yield: 94%(155 mg); <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C): δ = 0.85 (t, J = 7.2 Hz, 6H, CH<sub>3</sub>), 1.02 (s, 18H, CH<sub>3</sub>), 1.25-1.30 (m, 8H, CH<sub>2</sub>), 1.49-1.52 (m, 4H, CH<sub>2</sub>), 4.75 (s, 2H, CH); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C): δ = 14.40 (2 CH<sub>3</sub>), 24.05 (2 CH<sub>2</sub>), 28.34 (2 CH<sub>2</sub>), 28.94 (6 CH<sub>3</sub>), 30.01 (2 CH<sub>2</sub>), 34.90 (2 quat. C), 83.48 (2 quat. C), 96.56 (2 CH), 162.89 (2 quat. C).

**General Procedure for the Preparation of  $\alpha,\alpha'$ -Dibromo- $\Delta^1$ -bipyrrolines **4**.** NBS (20.0 mmol, 3.56 g) was added to a solution of  $\Delta^1$ -bipyrrolines **1** (2.0 mmol) in 20 ml CCl<sub>4</sub> in a 50 ml Schlenk tube, the mixture was stirred at 80 °C for 48 h. The reaction mixture was then quenched with water and extracted with acetic ether. The extract was washed with brine and dried over

MgSO<sub>4</sub>. The solvent was evaporated in vacuo to give yellow solid, which was purified by column chromatography (Petrol Ether: Ethyl Acetate = 50:1) to afford products  $\alpha,\alpha,\alpha',\alpha'$ -tetrabromo- $\Delta^1$ -bipyrrolines **4**.

**4a**, yellow solid, yield: 81% (1.012 g). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 1.61-1.69 (m, 2H, CH<sub>2</sub>), 1.78-1.81 (m, 2H, CH<sub>2</sub>), 2.53-2.57 (m, 2H, CH<sub>2</sub>), 2.87-2.91 (m, 2H, CH<sub>2</sub>), 7.37-7.46 (m, 6H, C<sub>6</sub>H<sub>5</sub>), 8.29-8.31 (m, 4H, C<sub>6</sub>H<sub>5</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, TMS, 25 °C):  $\delta$  = 23.1 (2 CH<sub>2</sub>), 32.1 (2 CH<sub>2</sub>), 69.6 (2 quat. C), 80.3 (2 quat. C), 128.9 (4 CH), 131.0 (2 CH), 131.1 (4 CH), 132.3 (2 quat. C), 166.0 (2 C=N). HRMS:  $m/z$ : calcd for C<sub>22</sub>H<sub>19</sub>Br<sub>4</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 626.8202, found: 626.8199.

**4b**, yellow solid, yield: 78% (1.068g). <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 1.64-1.72 (m, 2H, CH<sub>2</sub>), 1.79-1.81 (m, 2H, CH<sub>2</sub>), 2.49-2.54 (m, 2H, CH<sub>2</sub>), 2.90-2.93 (m, 2H, CH<sub>2</sub>), 3.83 (s, 6H, CH<sub>3</sub>), 6.90 (d,  $J$  = 8 Hz, 4H, C<sub>6</sub>H<sub>5</sub>), 8.22 (d,  $J$  = 8 Hz, 4H, C<sub>6</sub>H<sub>5</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 22.2 (2 CH<sub>2</sub>), 31.1 (2 CH<sub>2</sub>), 55.5 (2 CH<sub>3</sub>O), 69.5 (2 quat. C), 79.2 (2 quat. C), 113.6 (4 CH), 122.6 (2 quat. C), 131.9 (4 CH), 162.3 (2 quat. C), 165.1 (2 C=N). HRMS:  $m/z$ : calcd for C<sub>24</sub>H<sub>23</sub>Br<sub>4</sub>O<sub>2</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 686.8416, found: 686.8419.

**4c**, yellow solid, yield: 63% (823 mg). <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 1.68-1.71 (m, 2H, CH<sub>2</sub>), 1.81-1.83 (m, 2H, CH<sub>2</sub>), 2.37 (s, 6H, CH<sub>3</sub>), 2.54-2.58 (m, 2H, CH<sub>2</sub>), 2.92-2.95 (m, 2H, CH<sub>2</sub>), 7.25-7.33 (m, 4H, C<sub>6</sub>H<sub>5</sub>), 7.97 (s, 2H, C<sub>6</sub>H<sub>5</sub>), 8.10-8.12 (m, 2H, C<sub>6</sub>H<sub>5</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 21.5 (2 CH<sub>2</sub>), 22.1 (2 CH<sub>2</sub>), 30.8 (2 CH<sub>3</sub>), 69.0 (2 quat. C), 79.3 (2 quat. C), 127.1 (2 CH), 127.9 (2 CH), 130.0 (2 quat. C), 130.5 (2 CH), 132.4 (2 CH), 137.9 (2 quat. C), 166.0 (2 C=N). HRMS:  $m/z$ : calcd for C<sub>24</sub>H<sub>23</sub>Br<sub>4</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 654.8518, found: 654.8522.

**4d**, yellow solid, yield: 83% (971 mg). <sup>1</sup>H NMR (400MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 1.40-1.43 (m, 2H, CH<sub>2</sub>), 1.44 (s, 18H, CH<sub>3</sub>), 1.71-1.74 (m, 2H, CH<sub>2</sub>), 2.45-2.52 (m, 2H, CH<sub>2</sub>), 2.65-2.69 (m, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>, TMS, 25 °C):  $\delta$  = 22.3 (2 CH<sub>2</sub>), 29.7 (2 CH<sub>2</sub>), 29.9 (6 CH<sub>3</sub>), 37.8 (2 quat. C), 70.2 (2 quat. C), 77.8 (2 quat. C) 173.8 (2 C=N). HRMS:  $m/z$ : calcd for C<sub>18</sub>H<sub>27</sub>Br<sub>4</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 586.8829, found: 586.8825.

**4e**, yellow solid, yield: 76% (977 mg). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 0.93 (t,  $J$  = 8Hz, 6H, CH<sub>3</sub>), 1.29-1.37 (m, 8H, CH<sub>2</sub>), 1.48 (s, 18H, CH<sub>3</sub>), 1.90-1.94 (m, 2H, CH<sub>2</sub>), 2.24-2.31 (m, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 14.3 (2 CH<sub>3</sub>), 24.4 (2 CH<sub>2</sub>), 26.9 (2

CH<sub>2</sub>), 30.7 (6 CH<sub>3</sub>), 35.2 (2 CH<sub>2</sub>), 39.1 (2 quat. C), 68.9 (2 quat. C), 84.5 (2 quat. C), 176.4 (2 C=N). HRMS: *m/z*: calcd for C<sub>22</sub>H<sub>37</sub>N<sub>2</sub> [M+H]<sup>+</sup>: 644.9614, found: 644.9617.

**General Procedure for the Preparation of 2,6-Diazasemibuvallenes 5.** Li (1.1 mmol, 8.4 mg) was added to a solution of  $\alpha, \alpha, \alpha', \alpha'$ -dibromo- $\Delta^1$ -bipyrrolines **4** (0.5 mmol) in 5 ml THF in a 25 ml round-bottom flask in glove box, the mixture was stirred at room temperature for 2-4 h. The solvent was evaporated in vacuo to give brown solid. This solid was dissolved in THF-d<sub>8</sub> and monitored by NMR to confirm  $\alpha, \alpha, \alpha', \alpha'$ -dibromo- $\Delta^1$ -bipyrroline was totally disappeared. The THF and THF-d<sub>8</sub> was evaporated in vacuo to give brown solid again. Then the 4,8-dibromo-2,6-diazasemibuvallenes were extracted by mixture solvent (Hexane:Et<sub>2</sub>O = 4:1) diethyl ether from brown solid, and the salt (LiBr) was removed.

**5a**, yellow solid, yield: 83% (195 mg). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 1.27-1.36 (m, 2H, CH<sub>2</sub>), 1.76-1.78 (m, 4H, CH<sub>2</sub>), 2.39-2.42 (m, 2H, CH<sub>2</sub>), 7.32-7.33 (m, 6H, C<sub>6</sub>H<sub>5</sub>), 7.78-7.80 (m, 4H, C<sub>6</sub>H<sub>5</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 22.0 (2 CH<sub>2</sub>), 26.9 (2 CH<sub>2</sub>), 79.4 (2 quat. C), 99.0 (2 quat. C), 128.9 (4 CH), 129.0 (4 CH), 130.5 (2 CH), 133.9 (2 quat. C), 151.8 (2 C=N). Elemental Analysis Calcd (%) for C<sub>22</sub>H<sub>18</sub>Br<sub>2</sub>N<sub>2</sub>: C, 56.20; H, 3.86; N, 5.96; Found: C, 56.09; H, 3.71; N, 5.85.

**5b**, yellow solid, yield: 73% (193 mg). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 0.86-0.88 (m, 2H, CH<sub>2</sub>), 1.27-1.36 (m, 4H, CH<sub>2</sub>), 1.73-1.77 (m, 2H, CH<sub>2</sub>), 3.76 (s, 6H, OCH<sub>3</sub>), 6.85 (d, *J* = 8 Hz, 4H, C<sub>6</sub>H<sub>5</sub>), 7.75 (d, *J* = 8 Hz, 4H, C<sub>6</sub>H<sub>5</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 22.2 (2 CH<sub>2</sub>), 27.1 (2 CH<sub>2</sub>), 55.8 (2 CH<sub>3</sub>O), 79.0 (2 quat. C), 114.4 (4 CH), 126.6 (2 quat. C), 130.7 (4 CH), 151.1 (2 quat. C), 162.0 (2 C=N). Elemental Analysis Calcd (%) for C<sub>24</sub>H<sub>22</sub>Br<sub>2</sub>N<sub>2</sub>O<sub>2</sub>: C, 54.36; H, 4.18; N, 5.28; Found: C, 54.09; H, 4.03; N, 5.11.

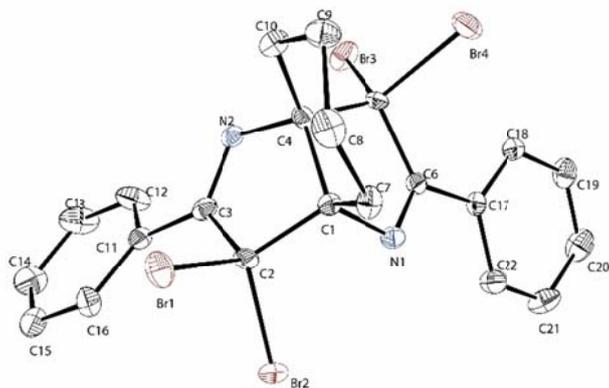
**5c**, yellow solid, yield: 89% (190 mg). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 1.22 (s, 18H, CH<sub>3</sub>), 1.29-1.31 (m, 2H, CH<sub>2</sub>), 1.37-1.41 (m, 4H, CH<sub>2</sub>), 2.16-2.20 (m, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C):  $\delta$  = 21.9 (2 CH<sub>2</sub>), 26.7 (2 CH<sub>2</sub>), 28.7 (6 CH<sub>3</sub>), 35.8 (2 quat. C), 78.5 (2 quat. C), 98.8 (2 quat. C), 159.4 (2 C=N). Elemental Analysis Calcd (%) for C<sub>18</sub>H<sub>26</sub>Br<sub>2</sub>N<sub>2</sub>: C, 50.25; H, 6.09; N, 6.51; Found: C, 50.04; H, 5.98; N, 6.38.

**Procedure for the Preparation of Compound 6.** 4,8-dibromo-2,6-diazasemibuvallene **5e** (0.3 mmol) was dissolved in THF-d<sub>8</sub> at room temperature in glove box. This solution was monitored by NMR. Almost 30 days later, 4,8-dibromo-2,6-diazasemibuvallene **5e** was totally transformed to compound **6**. At the same time, another part of this THF-d<sub>8</sub> solution of **5e** was shined by light, and compound **6** was formed in 3 days. The solvent was evaporated in vacuo to give yellow solid, which was purified by column chromatography (Petrol Ether: Ethyl Acetate = 100:1) to afford pure product **6** (105 mg) in 82% yield. Single crystals of **6** suitable for X-ray structural analysis were grown in hexane at room temperature.

**6**, yellow solid, yield: 82% (105 mg). <sup>1</sup>H NMR (400MHz, THF-d<sub>8</sub>, 25 °C): δ = 1.23 (brs, 1H, CH<sub>2</sub>), 1.29 (brs, 1H, CH<sub>2</sub>), 1.33 (s, 9H, CH<sub>3</sub>), 1.45 (s, 9H, CH<sub>3</sub>), 1.55-1.68 (m, 4H, CH<sub>2</sub>), 2.38-2.45 (m, 1H, CH<sub>2</sub>), 2.79-2.84 (m, 1H, CH<sub>2</sub>); <sup>13</sup>C NMR (100 MHz, THF-d<sub>8</sub>, 25 °C): δ = 22.8 (1 CH<sub>2</sub>), 25.8 (1 CH<sub>2</sub>), 27.8 (3 CH<sub>3</sub>), 28.6 (3 CH<sub>3</sub>), 34.7 (1 CH<sub>2</sub>), 36.5 (1 quat. C), 37.7 (1 quat. C), 41.6 (1 CH<sub>2</sub>), 86.0 (1 quat. C), 95.3 (1 quat. C), 107.9 (1 quat. C), 178.9 (1 quat. C), 181.0 (1 C=N), 181.3 (1 C=N). HRMS: *m/z*: calcd for C<sub>18</sub>H<sub>27</sub>Br<sub>2</sub>N<sub>2</sub>[M+H]<sup>+</sup>: 429.0464, found: 429.0473.

### 3) X-ray crystallographic studies

The single crystals of **4a**, and **6** suitable for X-ray analysis were grown as shown in the experimental section. Data collections for **4a** and **6** were performed at 180 K on SuperNova diffractometer, using monochromated Mo K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ). The determination of crystal class and unit cell parameters was carried out by the CrystalClear (Rigaku Inc. 2007) program package for **4a** and **6**. The raw frame data were processed using CrystalClear (Rigaku Inc. 2007) for **4a** and **6** to yield the reflection data file. The structures of **4a** and **6** were solved by use of SHELXTL program. Refinement was performed on  $F^2$  anisotropically for all the non-hydrogen atoms by the full-matrix least-squares method. The hydrogen atoms were placed at the calculated positions and were included in the structure calculation without further refinement of the parameters. Crystal data, data collection and processing parameters for compounds **4a** and **6** are summarized in Table S1 - Table S4. Crystallographic data (excluding structure factors) have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication nos. CCDC 1056585 (**4a**), CCDC 1056586 (**6**). Copies of these data can be obtained free of charge from the Cambridge Crystallographic Data Centre via

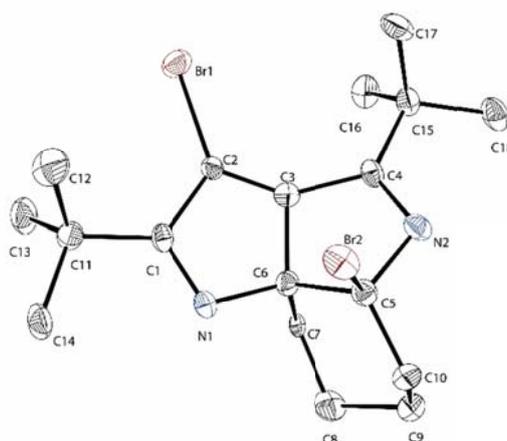


**Figure S1.** ORTEP drawing of **4a** with 30% probability thermal ellipsoids.

**Table S1. Crystal data and structure refinement for 4a.**

Identification code	<b>4a</b>
Empirical formula	C <sub>22</sub> H <sub>18</sub> Br <sub>4</sub> N <sub>2</sub>
Formula weight	630.02
Temperature/K	180.00(10)
Crystal system	orthorhombic
Space group	Pbca
a/Å	10.7971(8)
b/Å	15.2395(13)
c/Å	26.1285(19)
$\alpha$ /°	90
$\beta$ /°	90
$\gamma$ /°	90
Volume/Å <sup>3</sup>	4299.2(6)
Z	8
$\rho_{\text{calc}}/\text{cm}^3$	1.947
$\mu/\text{mm}^{-1}$	7.500
F(000)	2432.0

Crystal size/mm <sup>3</sup>	0.2 × 0.2 × 0.2
Radiation	Mo K $\alpha$ ( $\lambda$ = 0.7107)
2 $\theta$ range for data collection/ $^{\circ}$	6.728 to 52.044
Index ranges	-13 $\leq$ h $\leq$ 9, -18 $\leq$ k $\leq$ 11, -24 $\leq$ l $\leq$ 32
Reflections collected	10810
Independent reflections	4215 [ $R_{\text{int}}$ = 0.0564, $R_{\text{sigma}}$ = 0.0792]
Data/restraints/parameters	4215/0/253
Goodness-of-fit on $F^2$	1.065
Final R indexes [ $I \geq 2\sigma(I)$ ]	$R_1$ = 0.0529, $wR_2$ = 0.0992
Final R indexes [all data]	$R_1$ = 0.0826, $wR_2$ = 0.1106
Largest diff. peak/hole / e $\text{\AA}^{-3}$	1.81/-0.62



**Figure S2.** ORTEP drawing of **6** with 30% probability thermal ellipsoids. Hydrogen atoms have been omitted for clarity.

**Table S2. Crystal data and structure refinement for 6.**

Identification code	<b>6</b>
Empirical formula	$\text{C}_{18}\text{H}_{26}\text{Br}_2\text{N}_2$
Formula weight	430.23
Temperature/K	180.01(10)
Crystal system	monoclinic

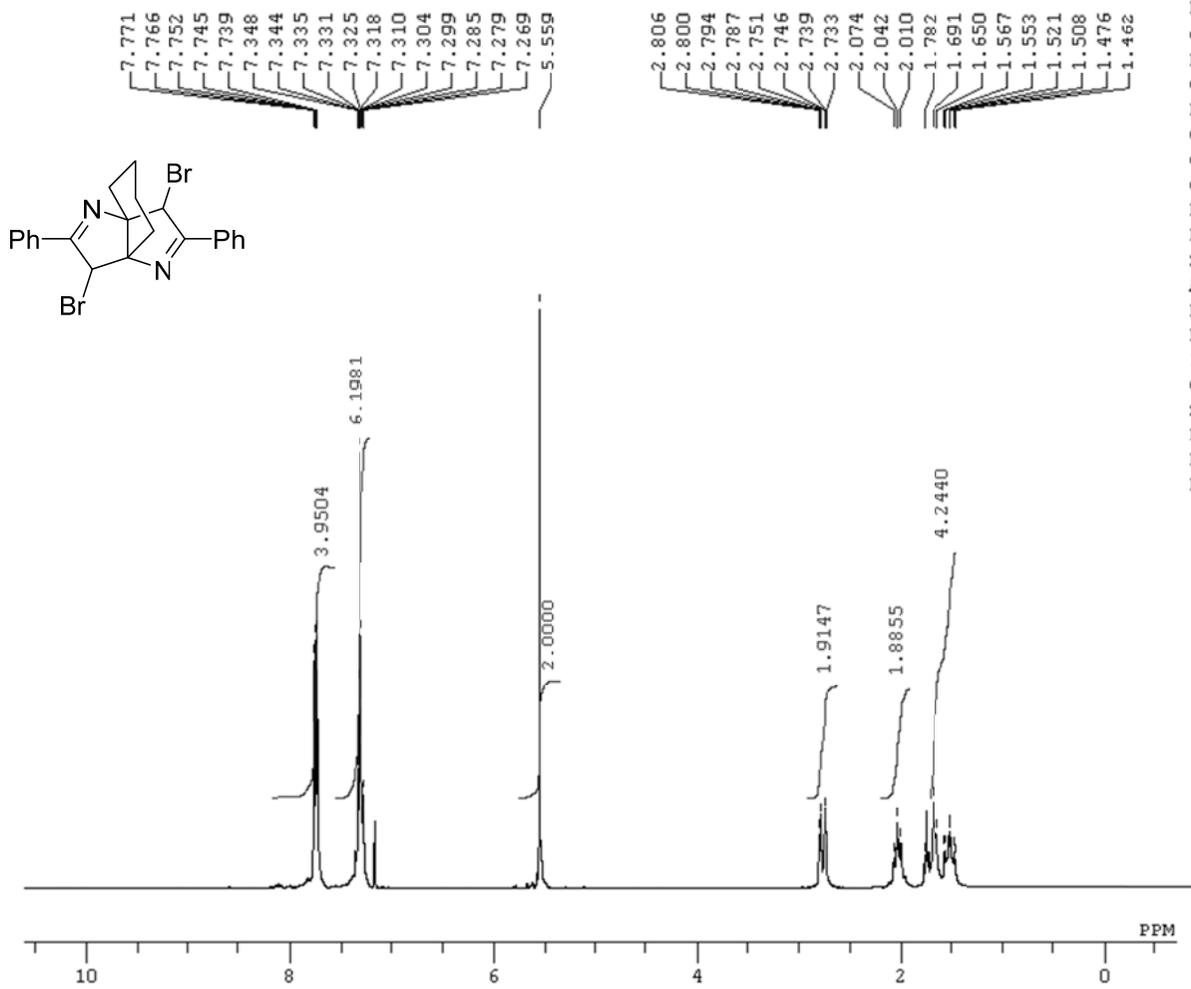
Space group	P2 <sub>1</sub> /c
a/Å	6.6642(4)
b/Å	19.8278(16)
c/Å	14.8224(10)
$\alpha$ /°	90
$\beta$ /°	101.924(7)
$\gamma$ /°	90
Volume/Å <sup>3</sup>	1916.3(2)
Z	4
$\rho_{\text{calc}}$ /cm <sup>3</sup>	1.491
$\mu$ /mm <sup>-1</sup>	4.229
F(000)	872.0
Crystal size/mm <sup>3</sup>	0.1 × 0.03 × 0.02
Radiation	MoK $\alpha$ ( $\lambda$ = 0.71073)
2 $\theta$ range for data collection/°	5.618 to 52.042
Index ranges	-8 ≤ h ≤ 8, -22 ≤ k ≤ 24, -18 ≤ l ≤ 18
Reflections collected	11160
Independent reflections	3773 [R <sub>int</sub> = 0.0696, R <sub>sigma</sub> = 0.0903]
Data/restraints/parameters	3773/0/205
Goodness-of-fit on F <sup>2</sup>	1.028
Final R indexes [I ≥ 2 $\sigma$ (I)]	R <sub>1</sub> = 0.0537, wR <sub>2</sub> = 0.1172
Final R indexes [all data]	R <sub>1</sub> = 0.0978, wR <sub>2</sub> = 0.1320
Largest diff. peak/hole / e Å <sup>-3</sup>	0.97/-0.63

#### 4) References

Sheldrick, G. M. SHELXTL 5.10 for Windows NT: *Structure Determination Software Programs*; Bruker Analytical X-ray Systems, Inc.: (Madison, WI, 1997).

3) Copies of <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra of all new compounds

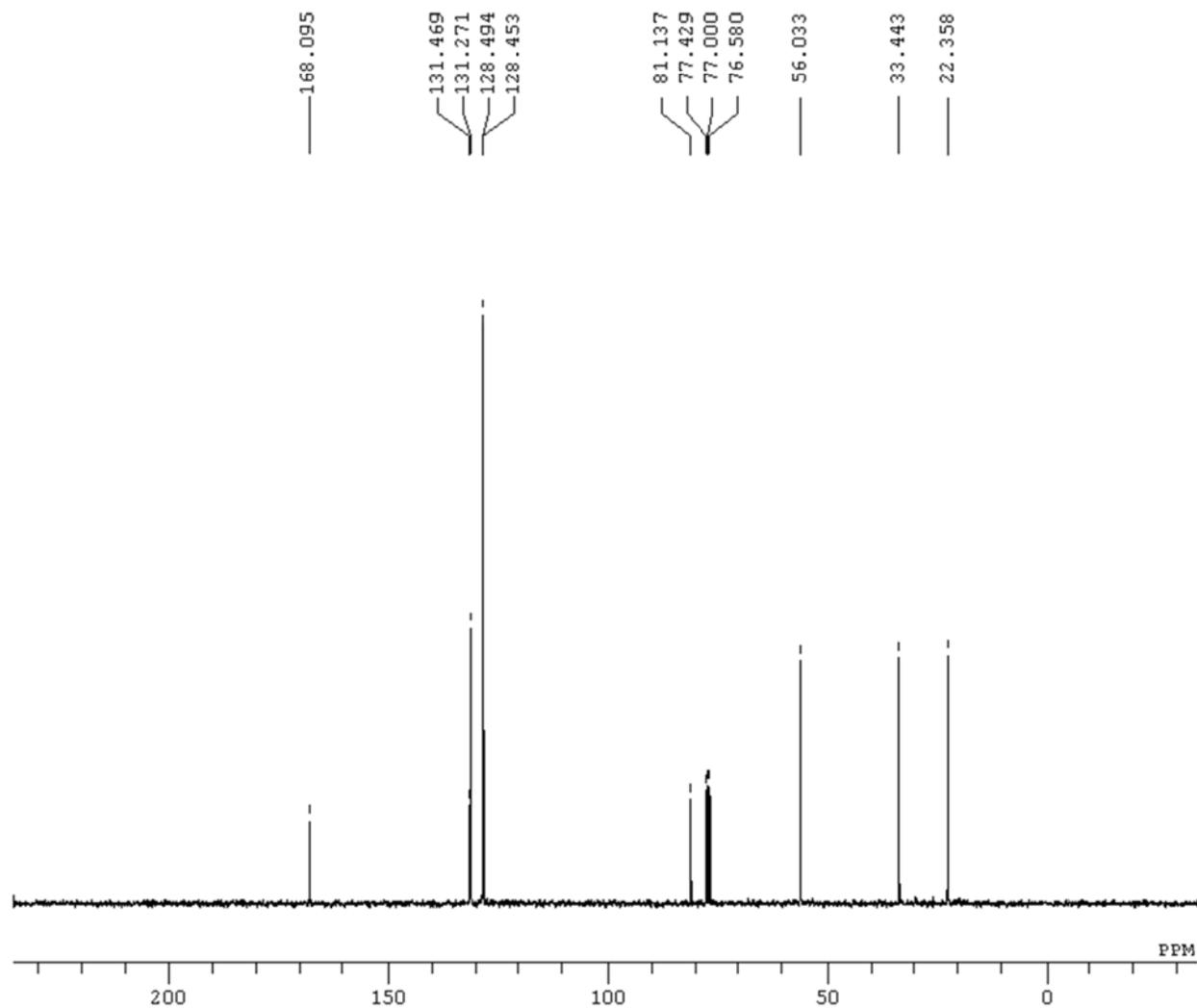
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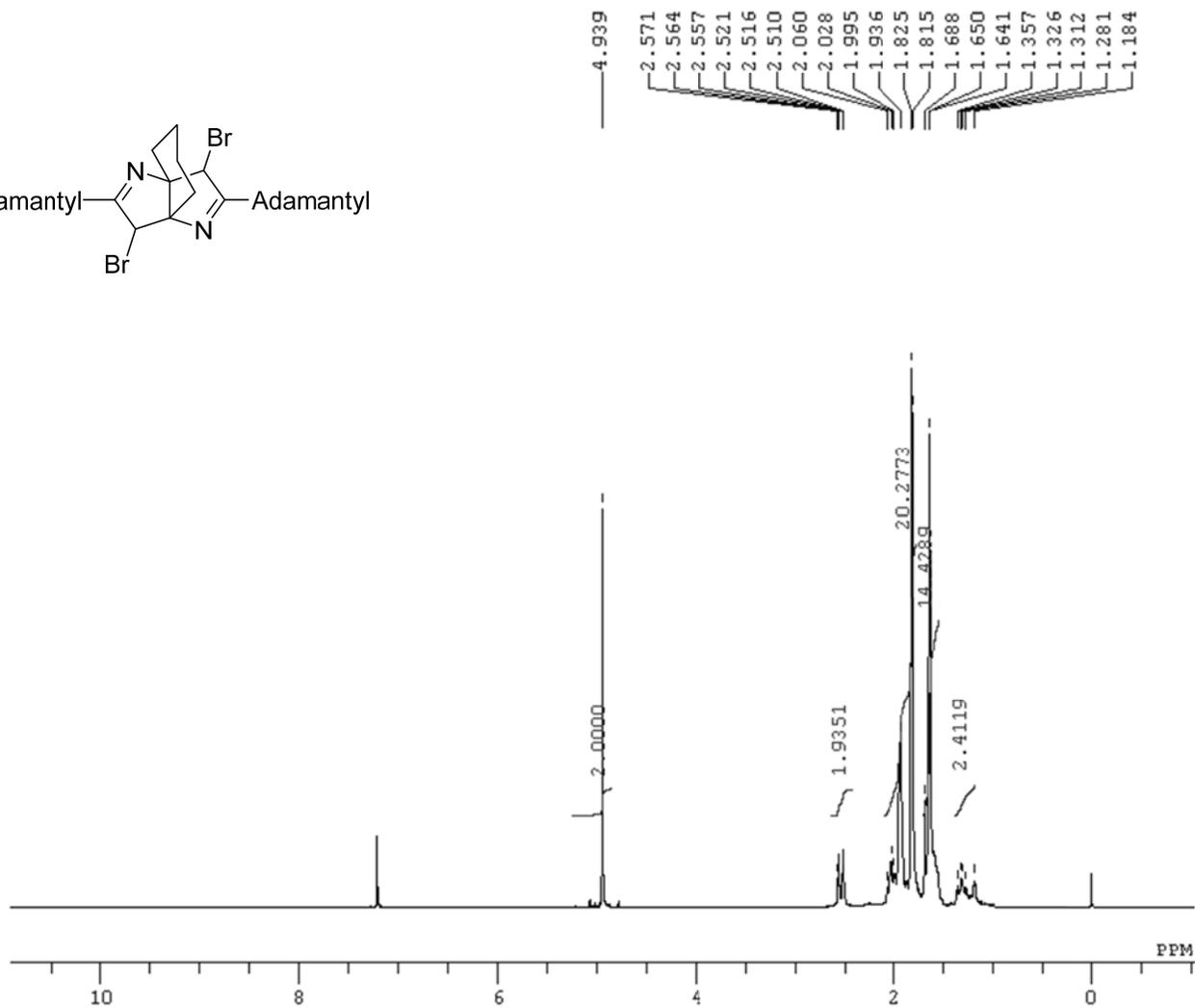
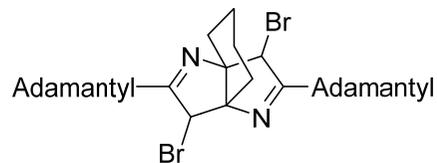
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2a-<sup>13</sup>C NMR



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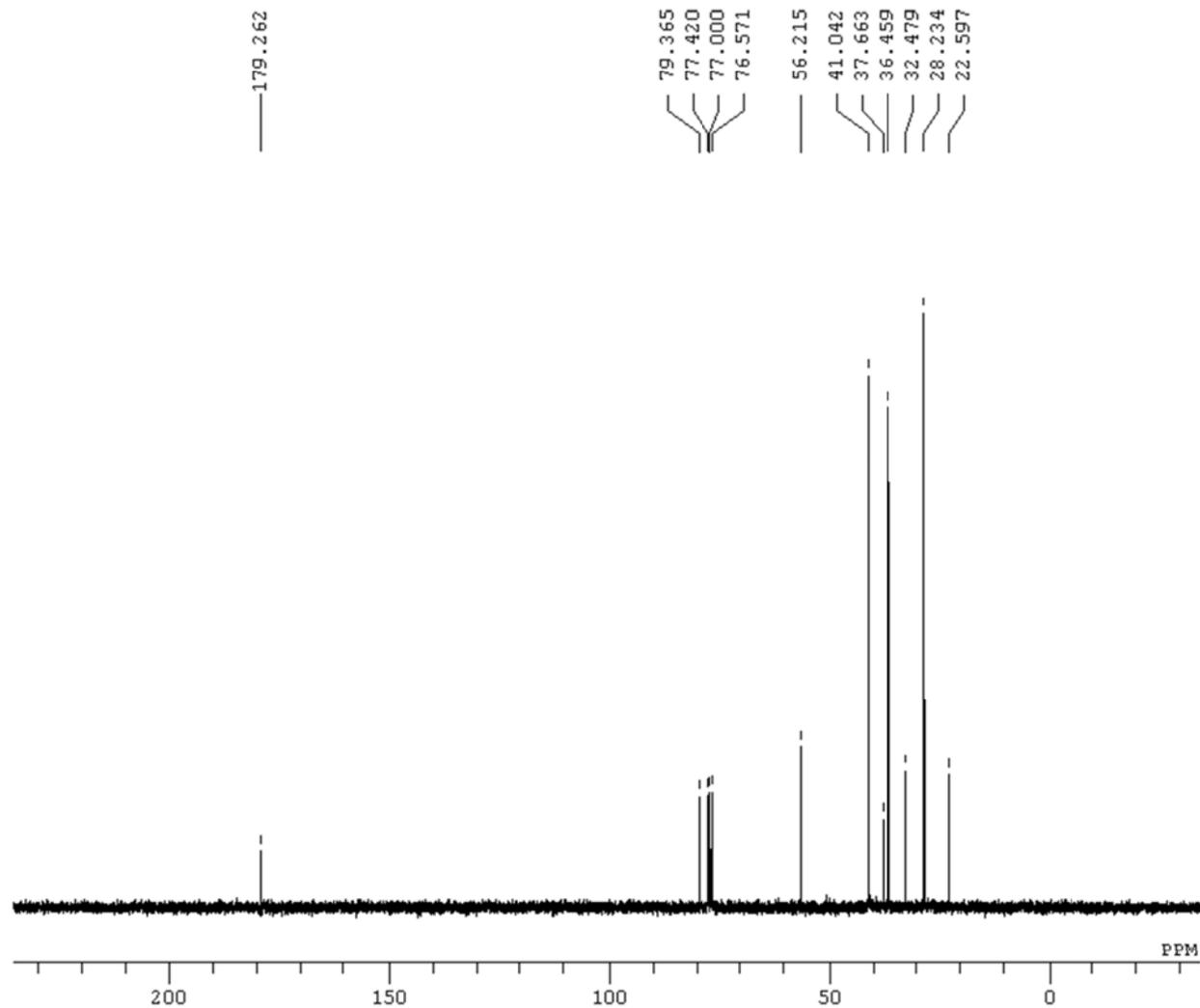
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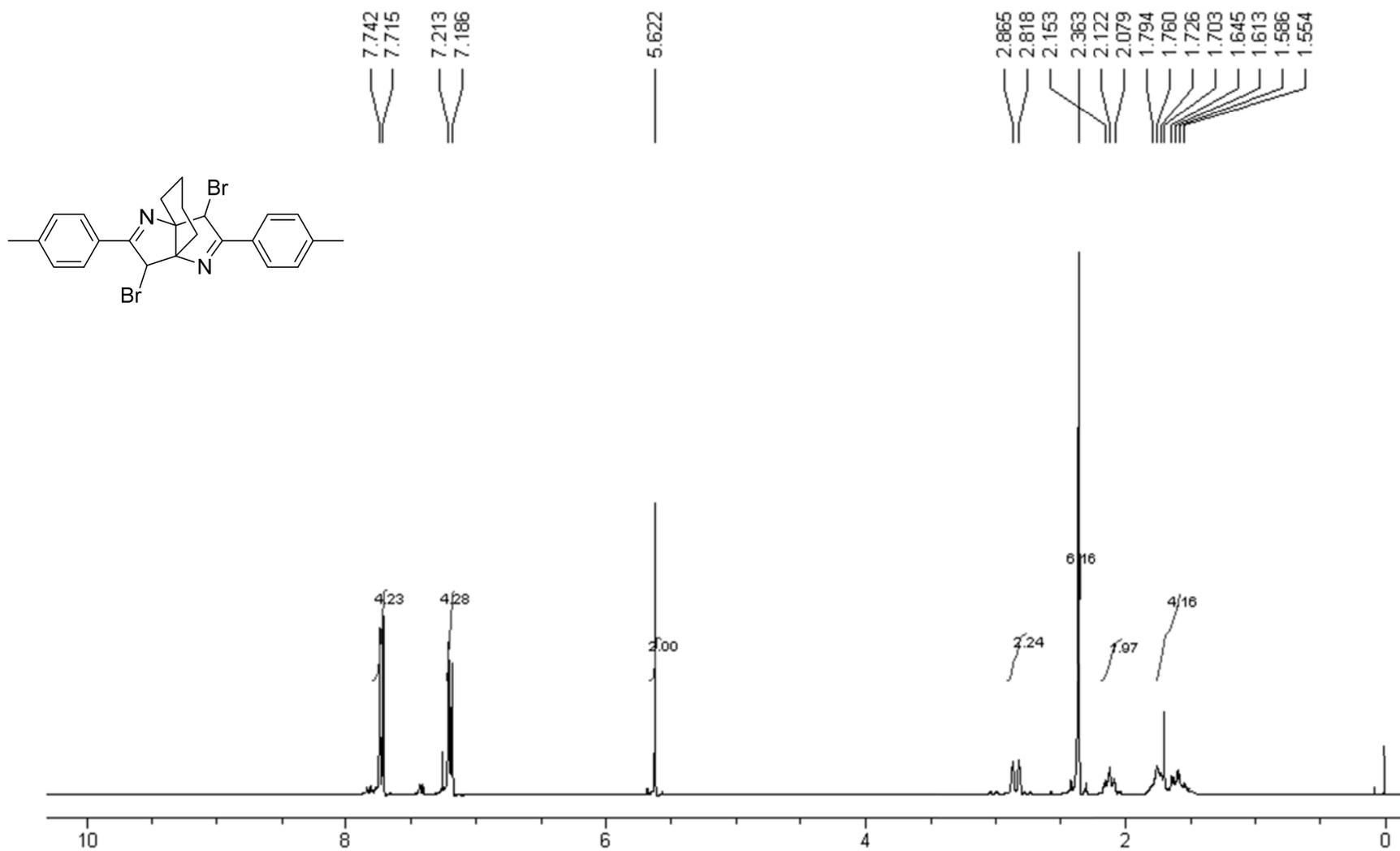
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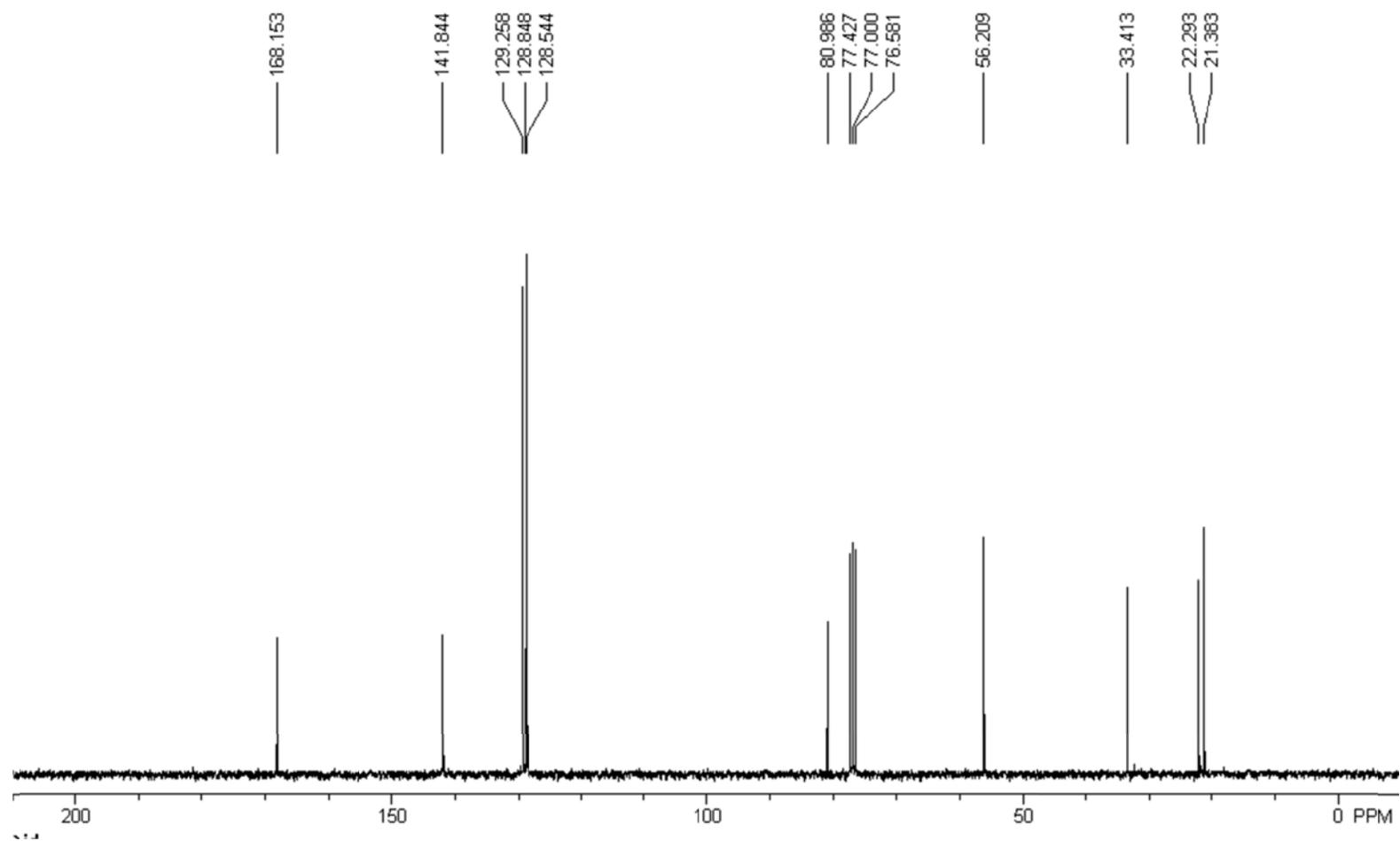


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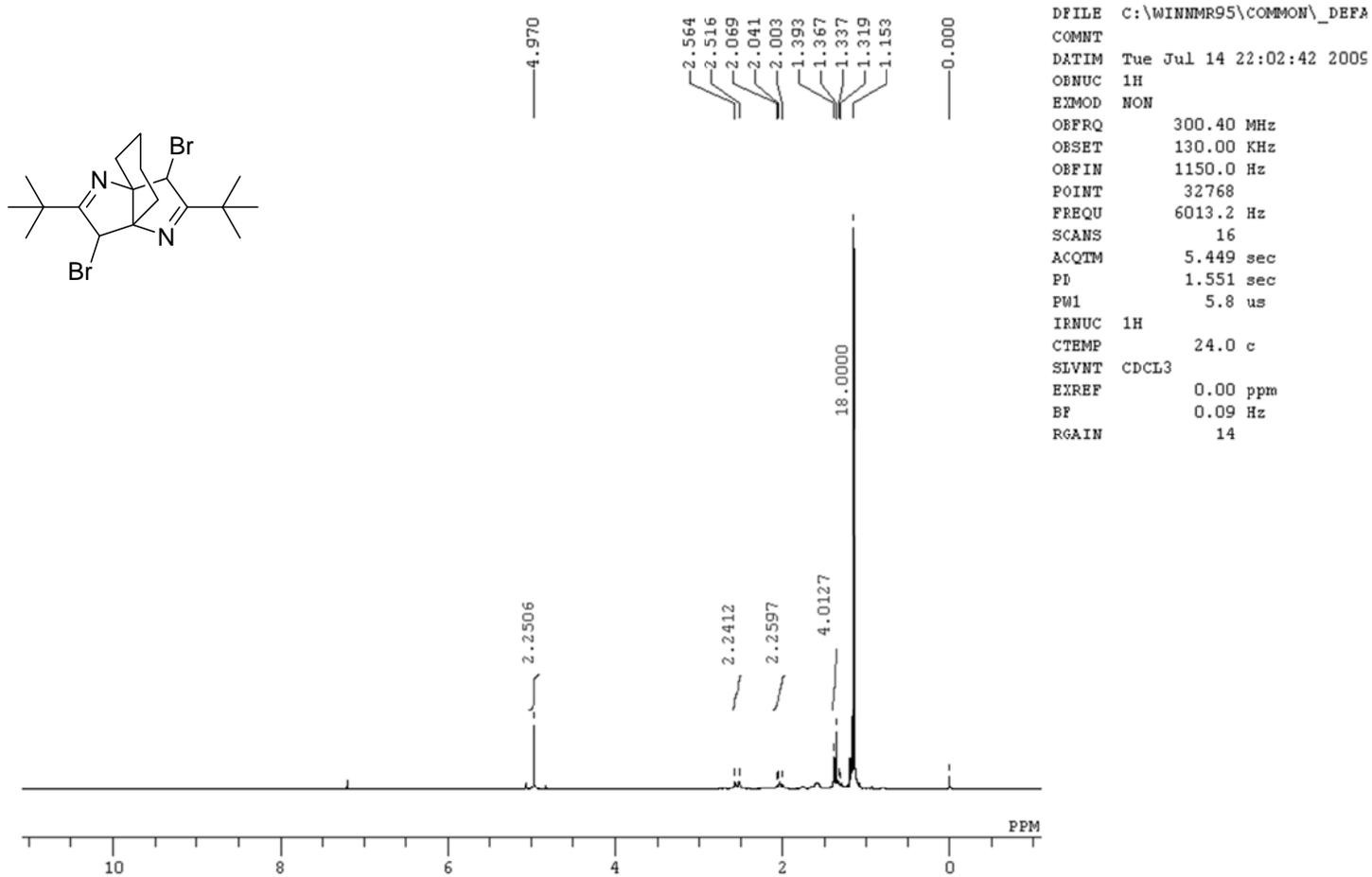
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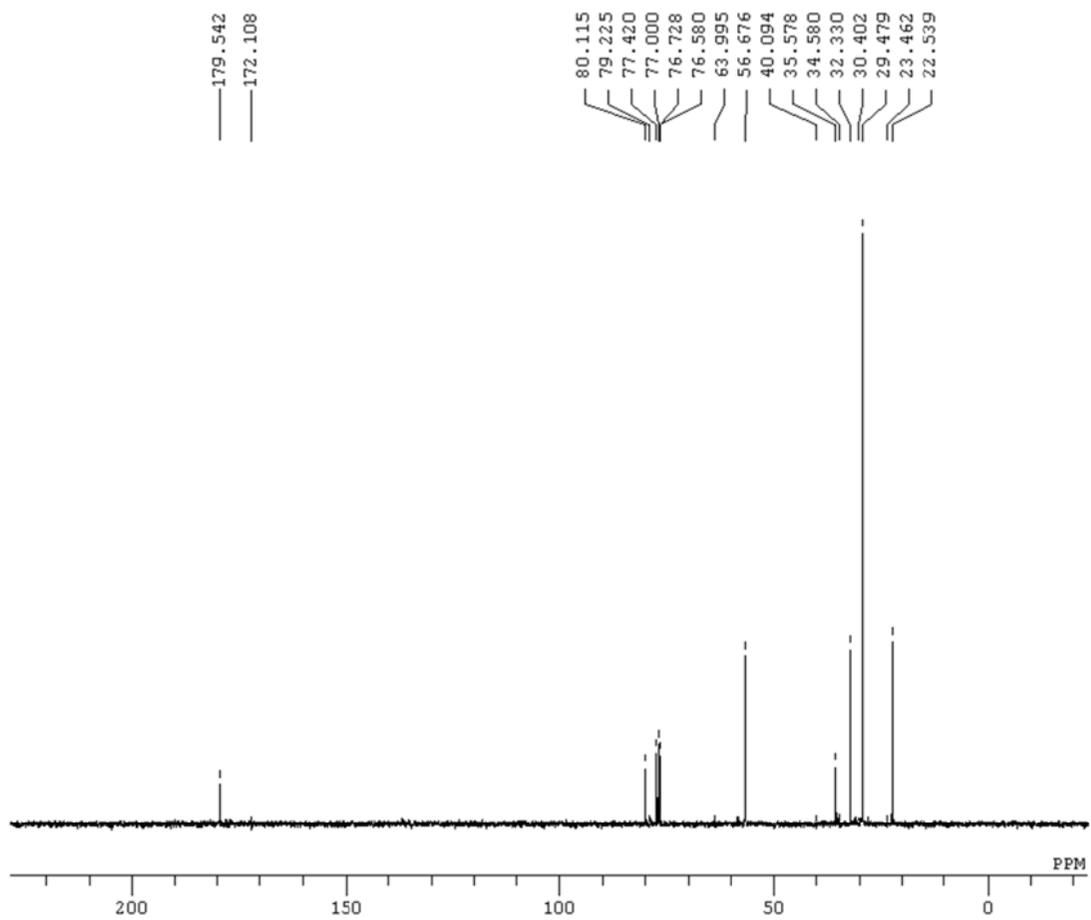
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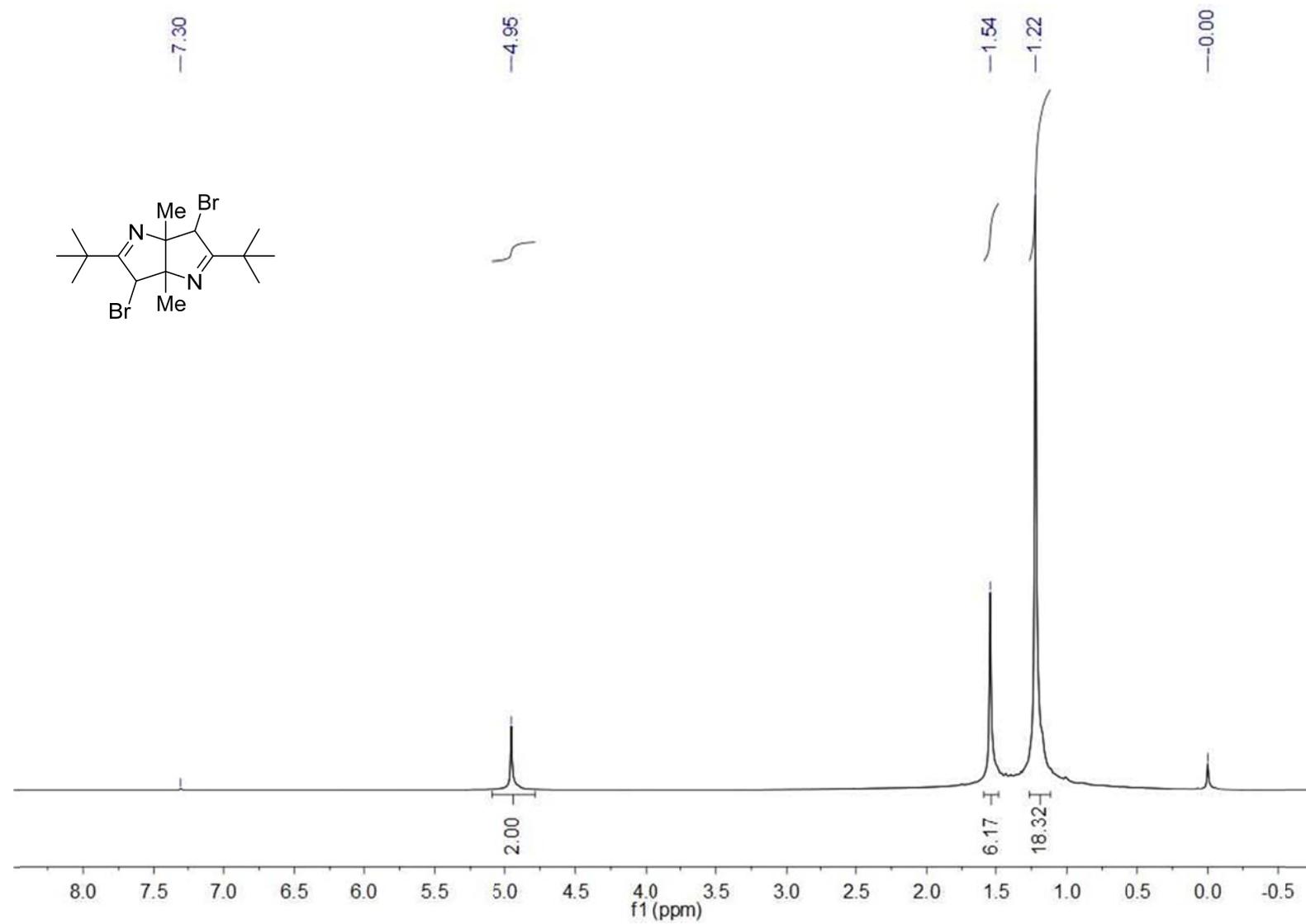


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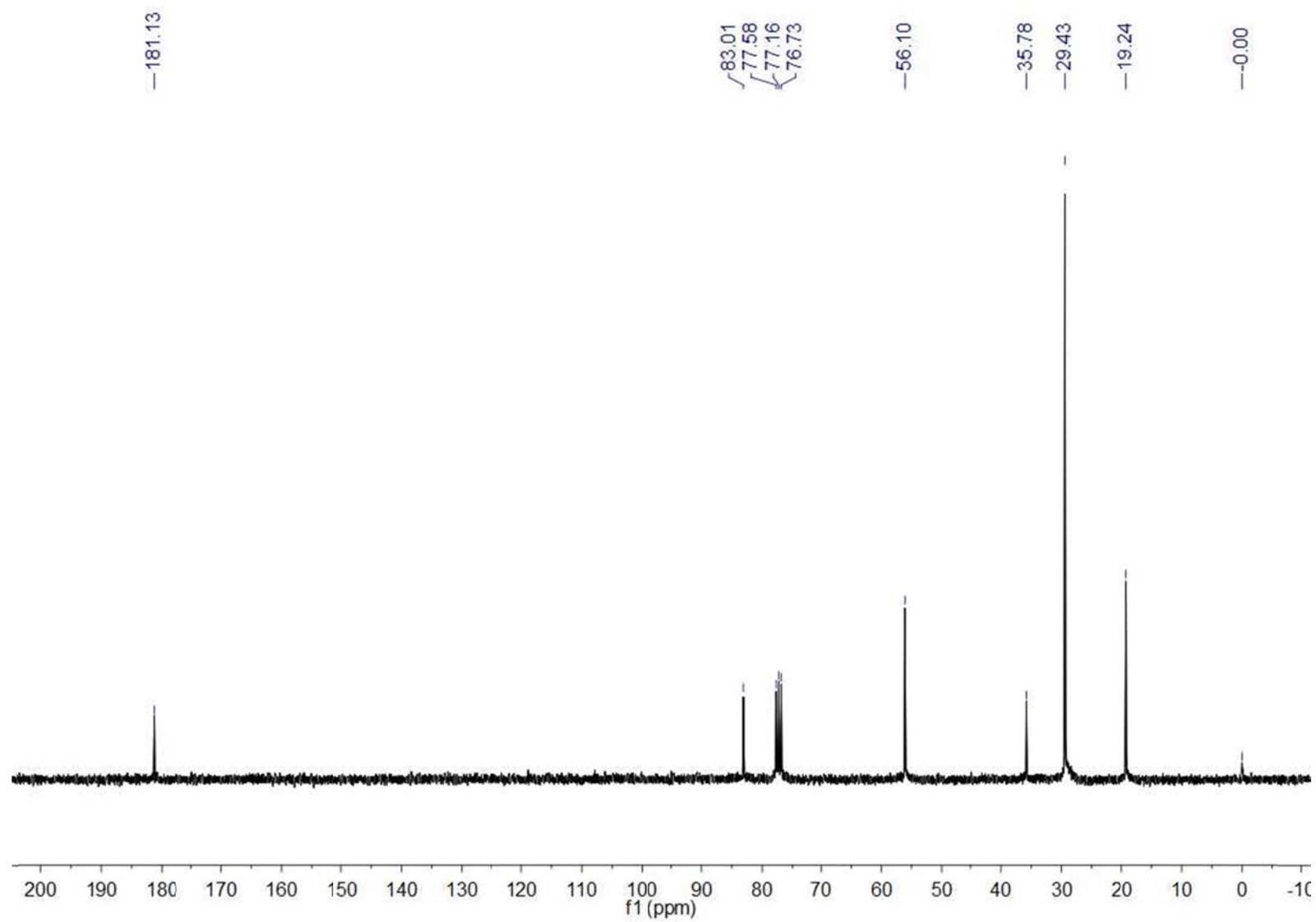


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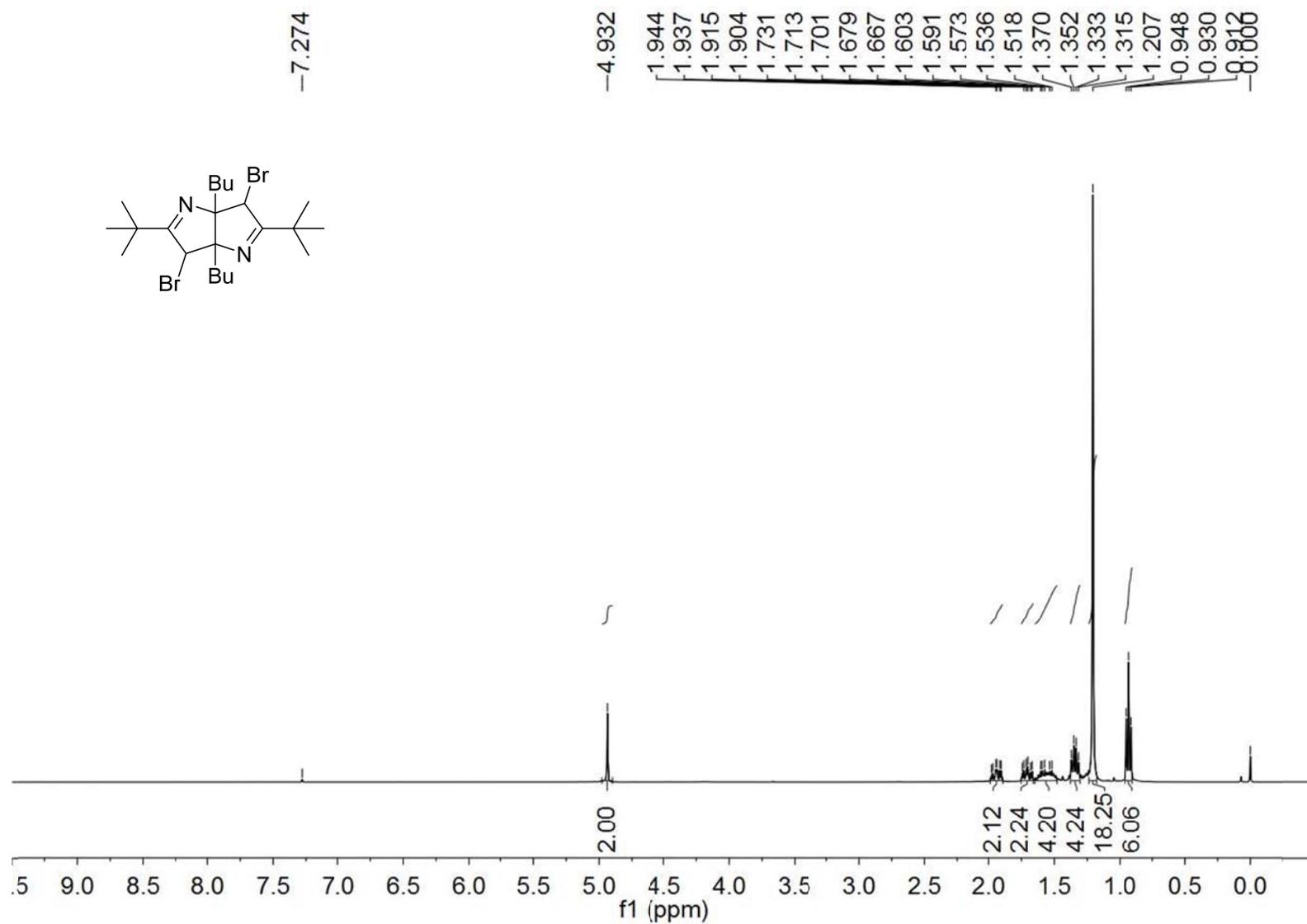
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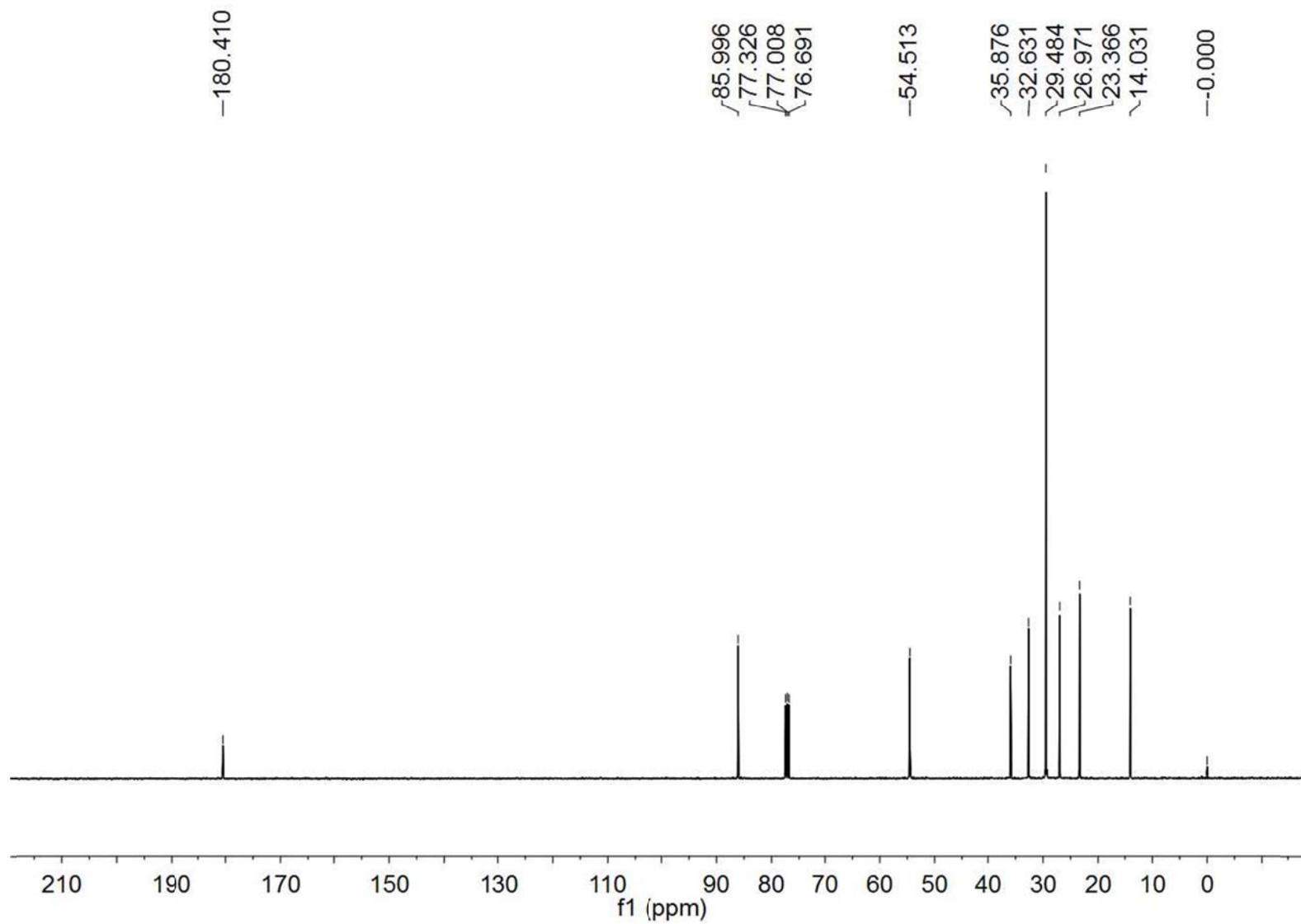
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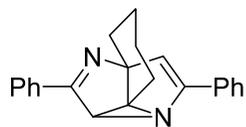
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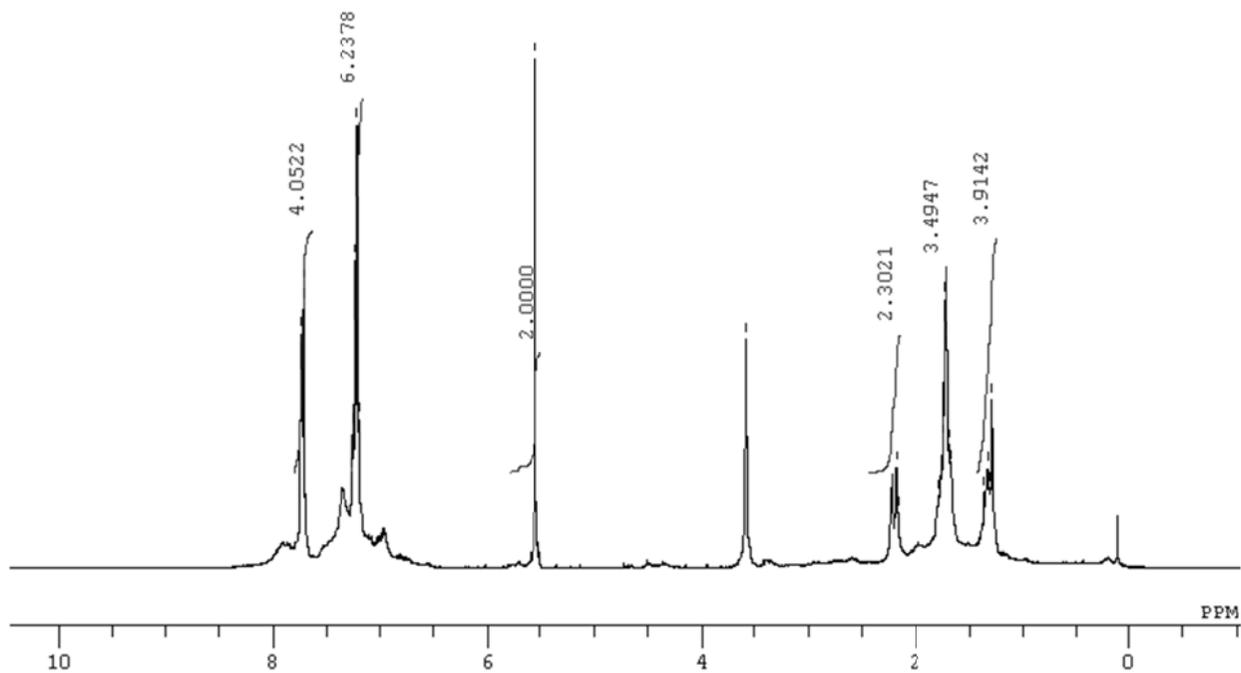


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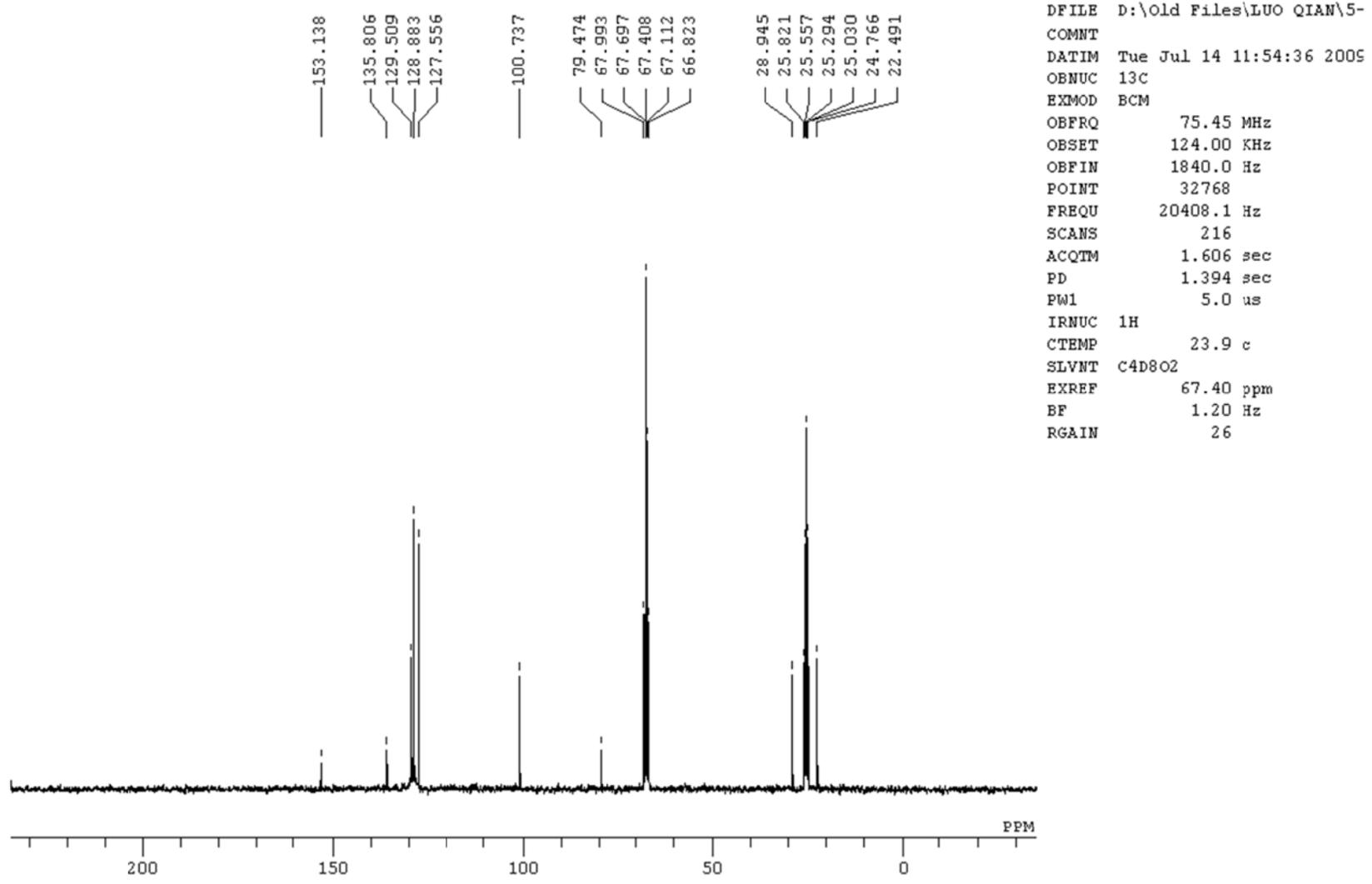
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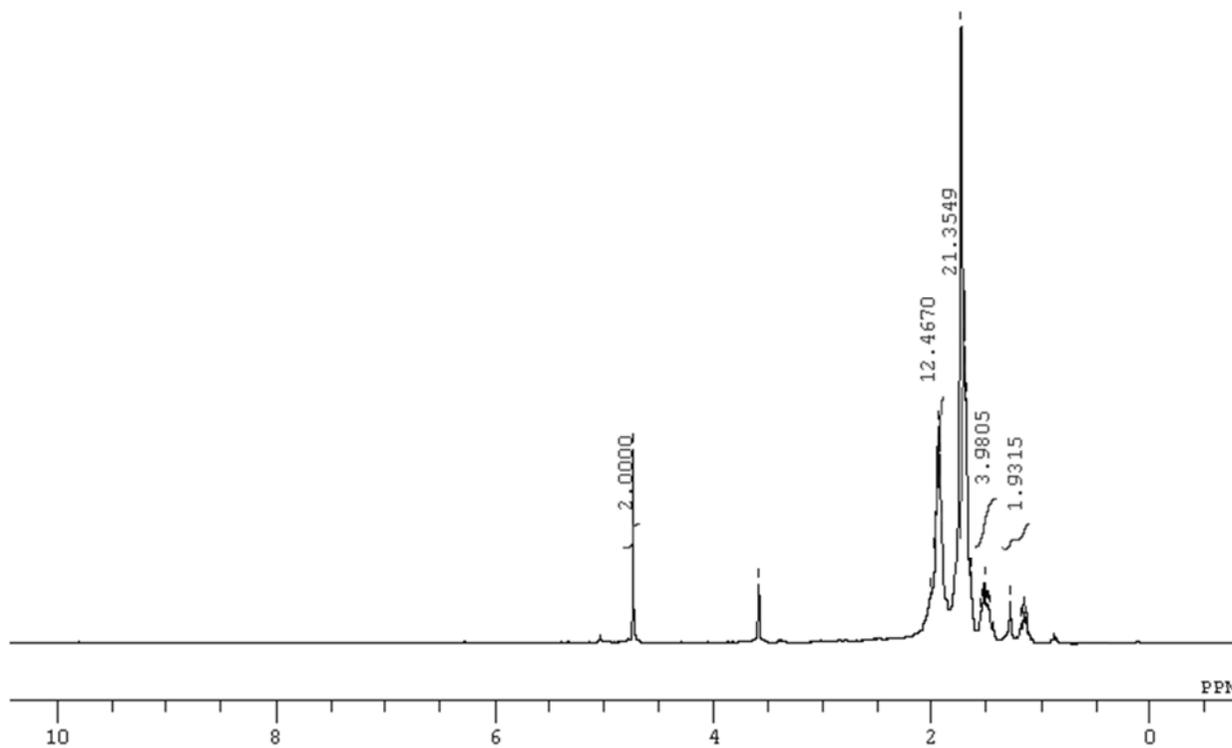
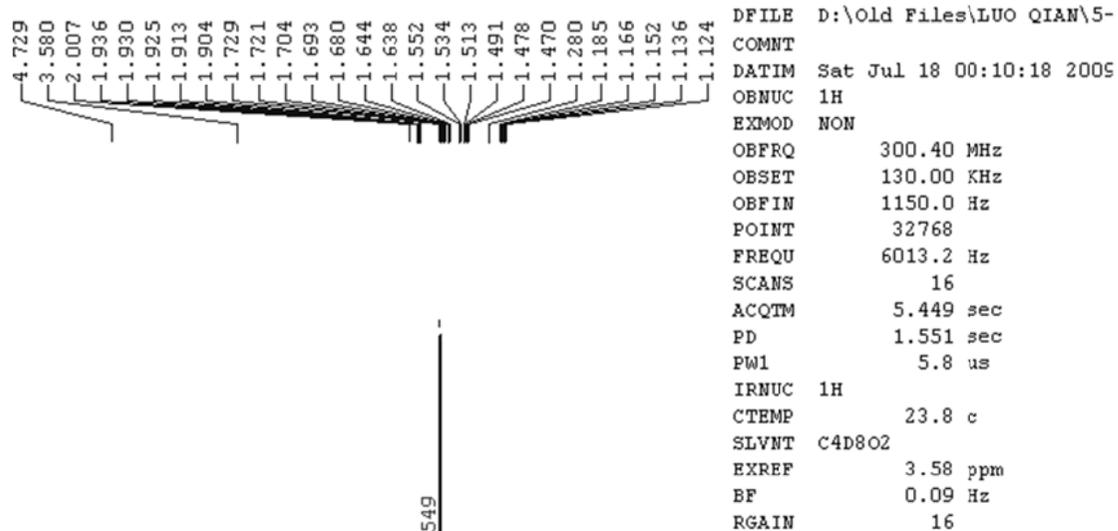
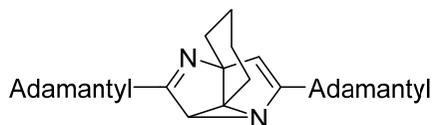


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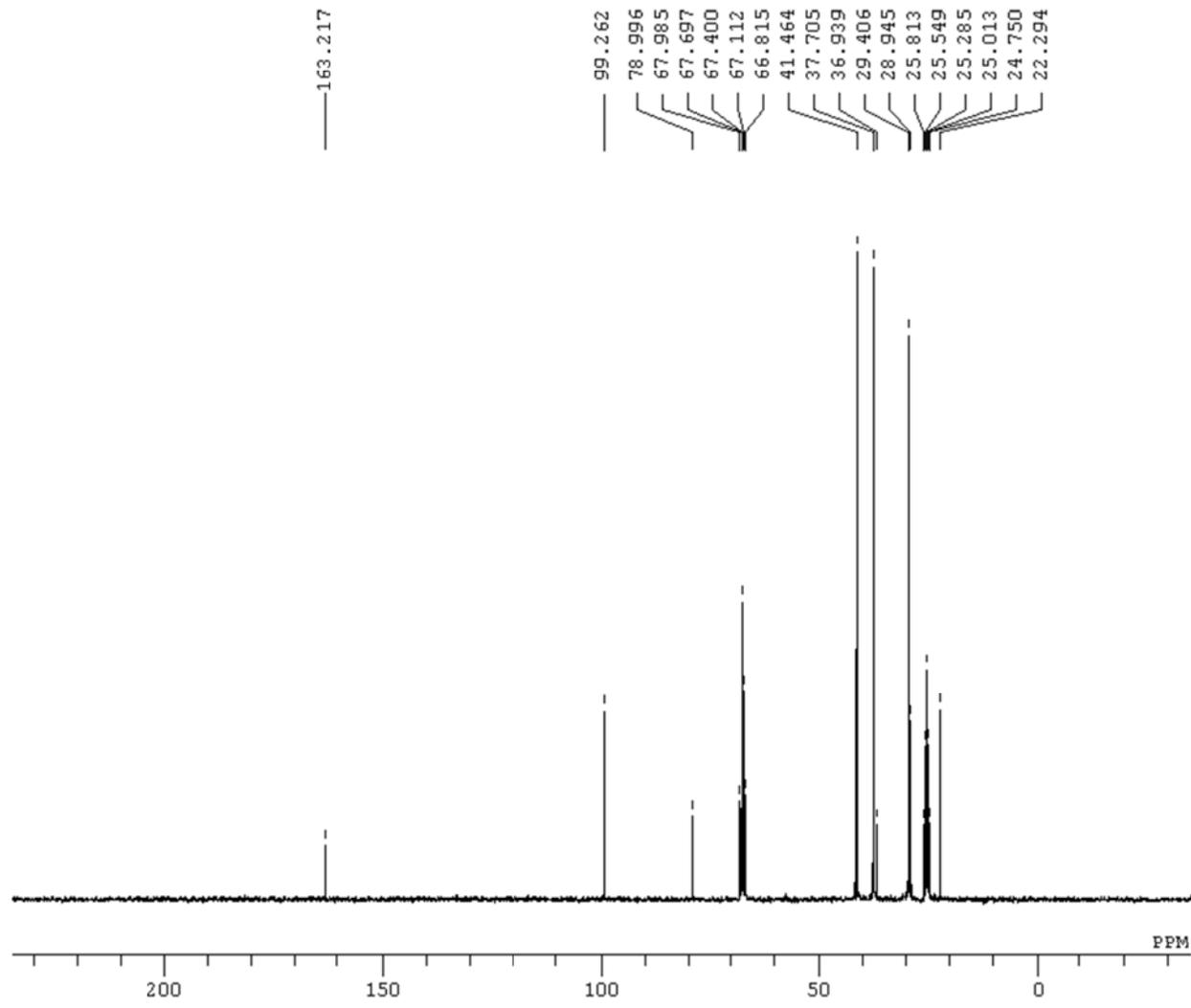
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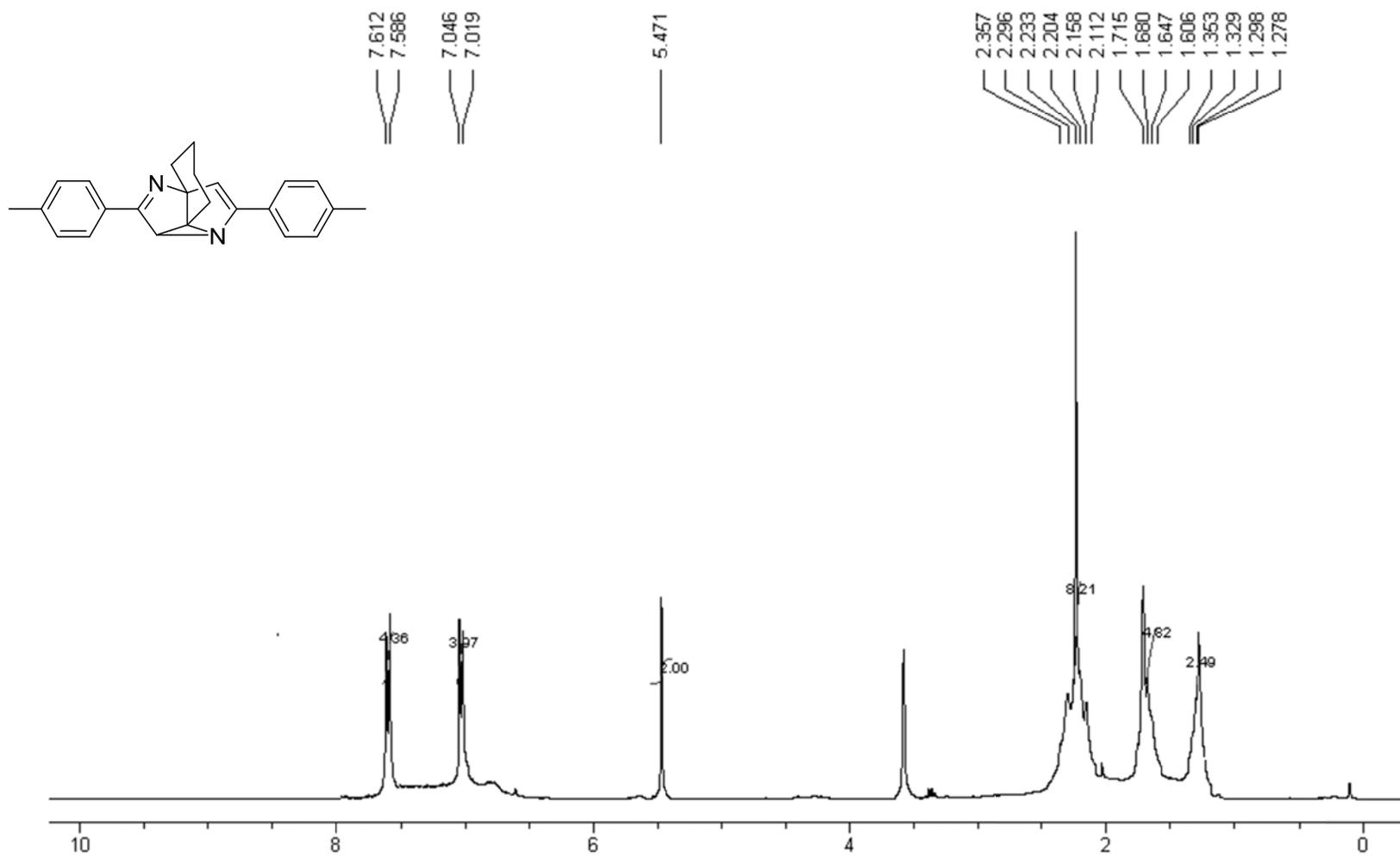


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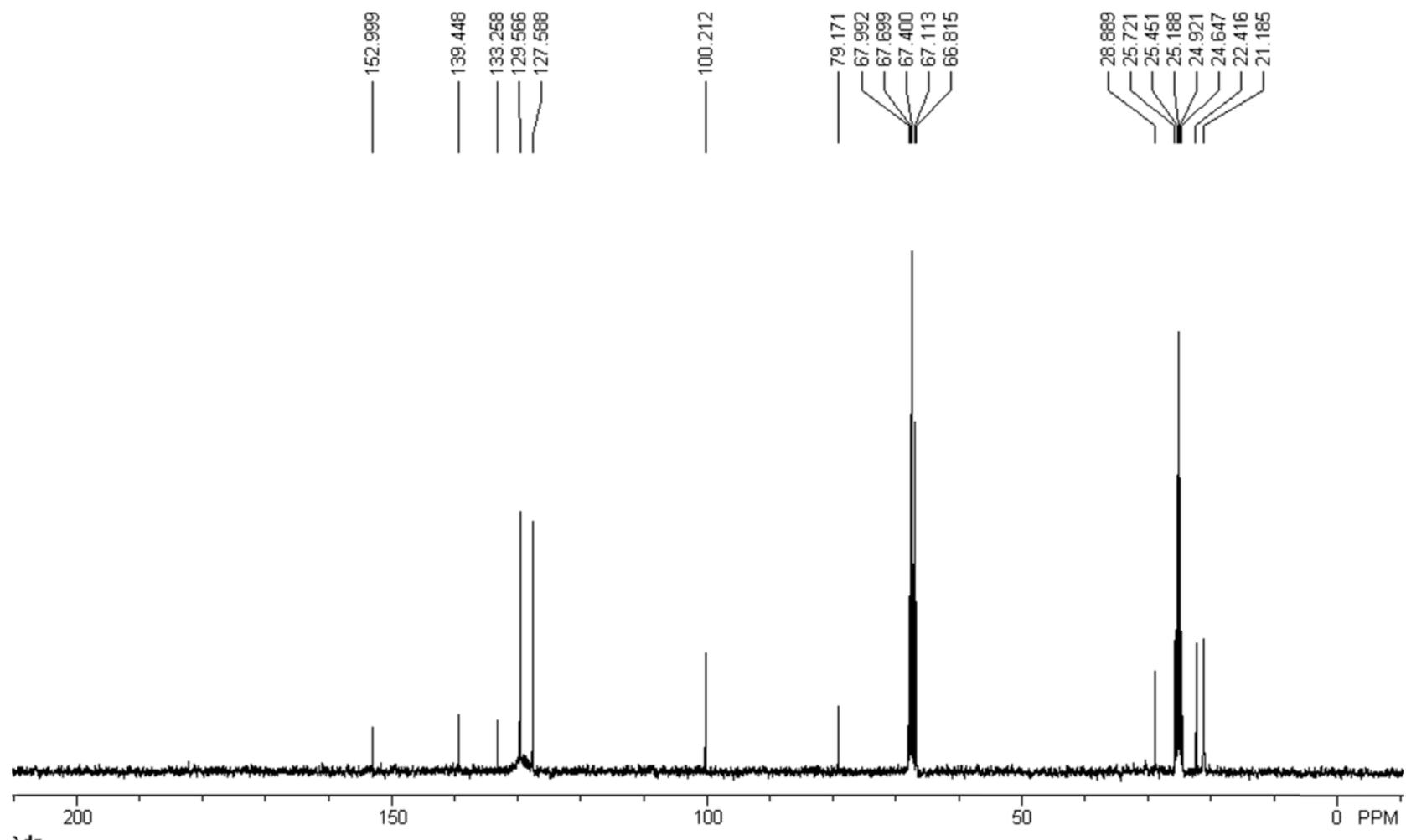


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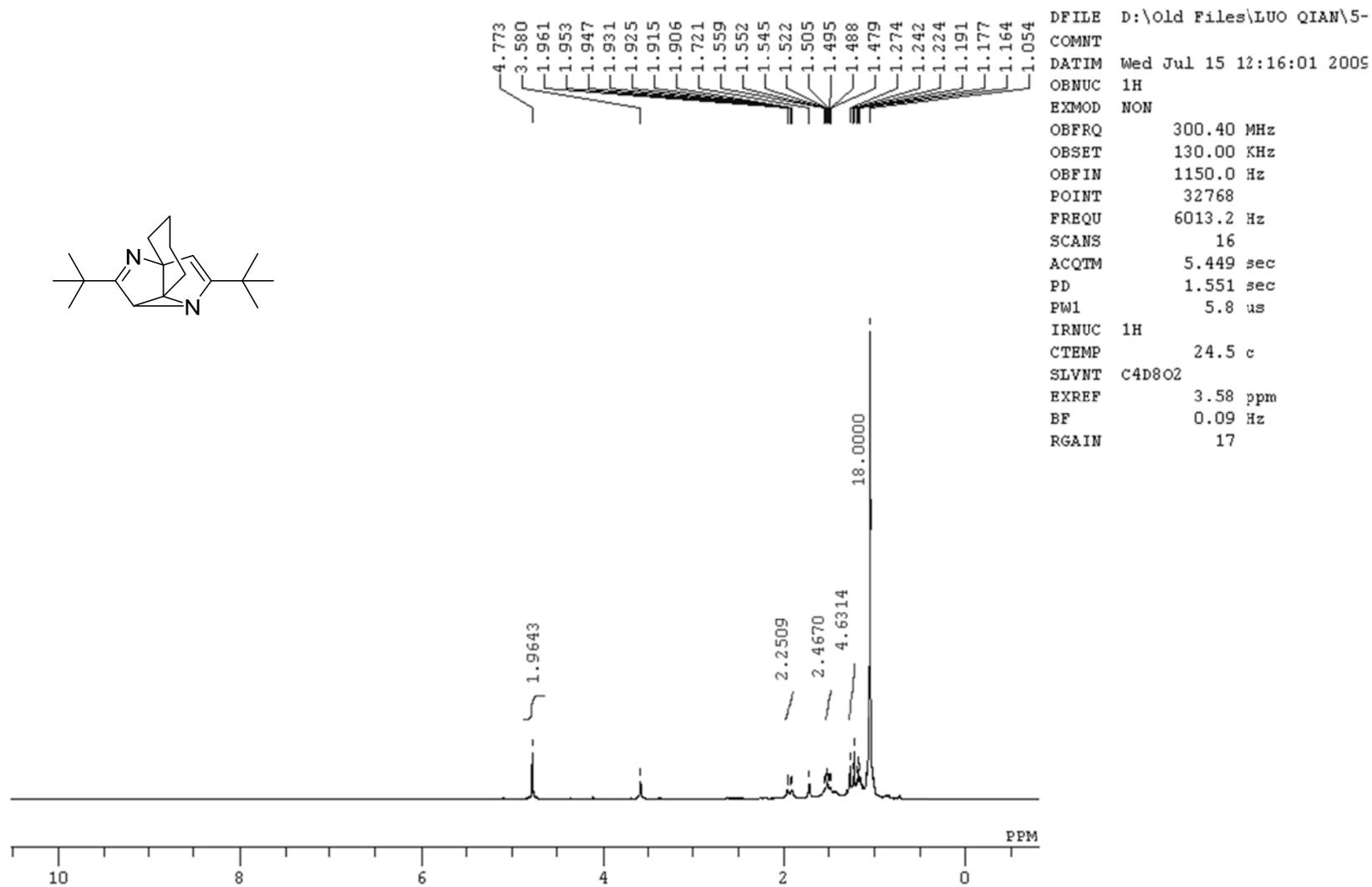
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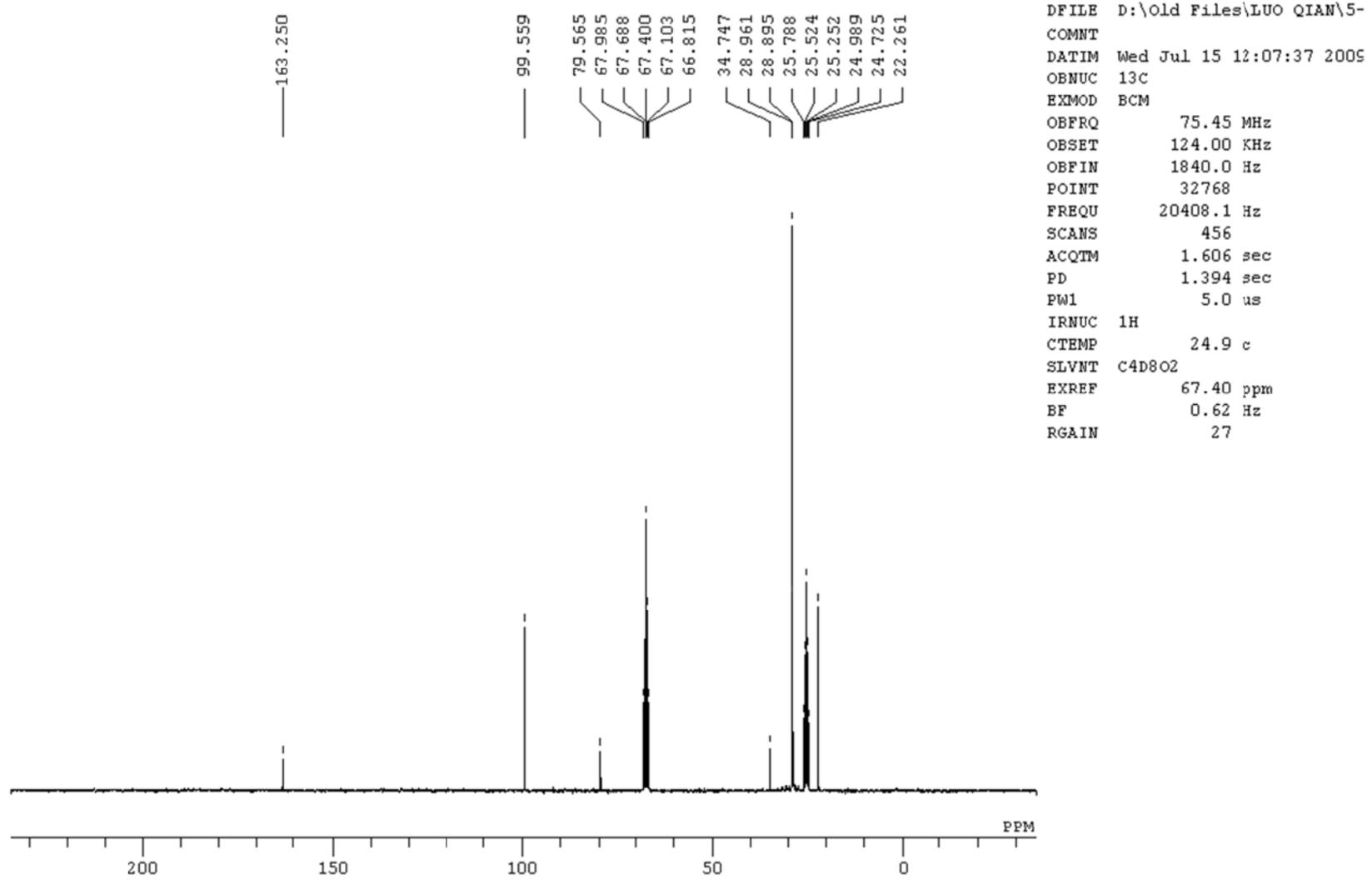
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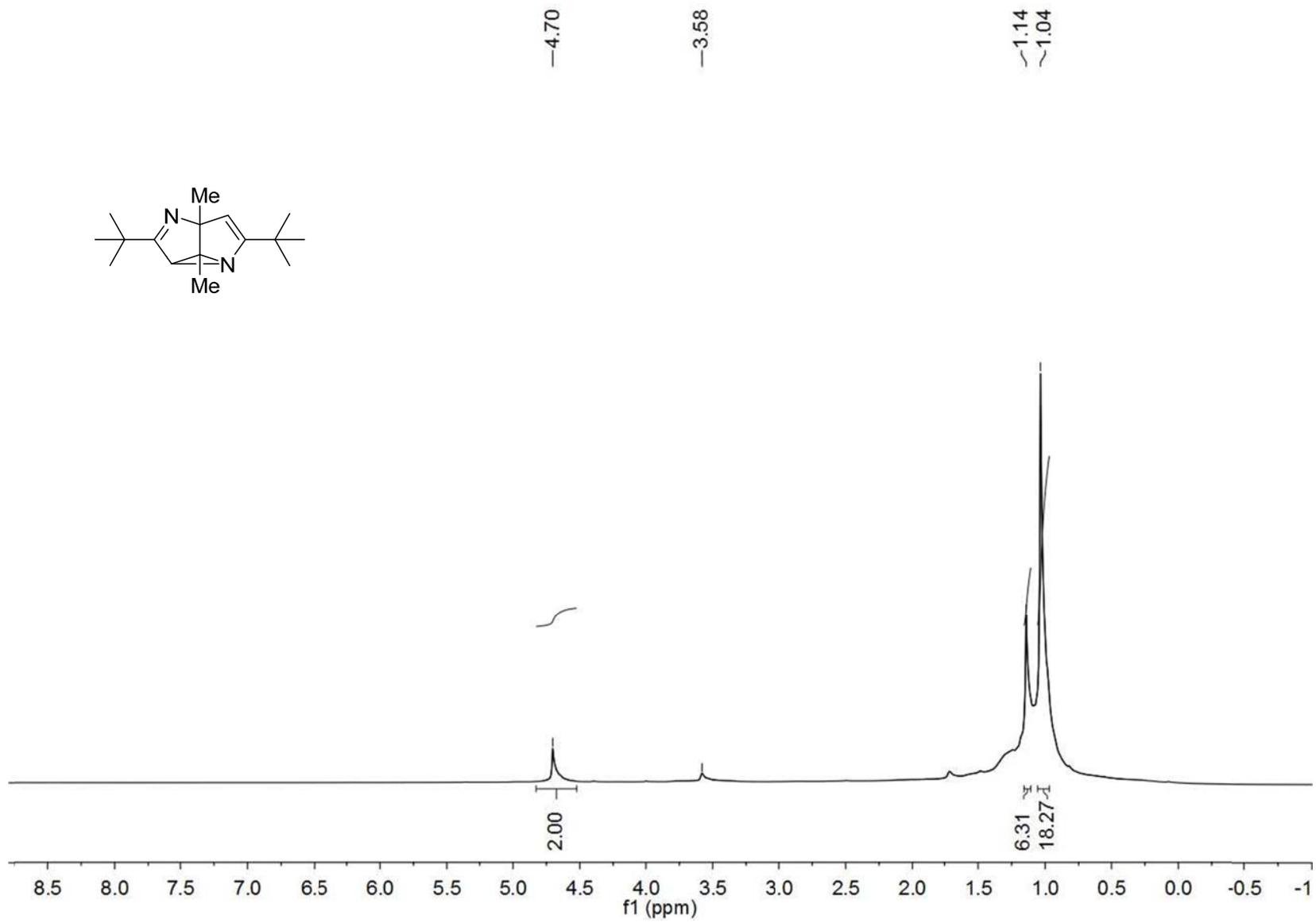
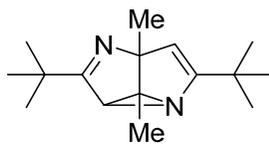
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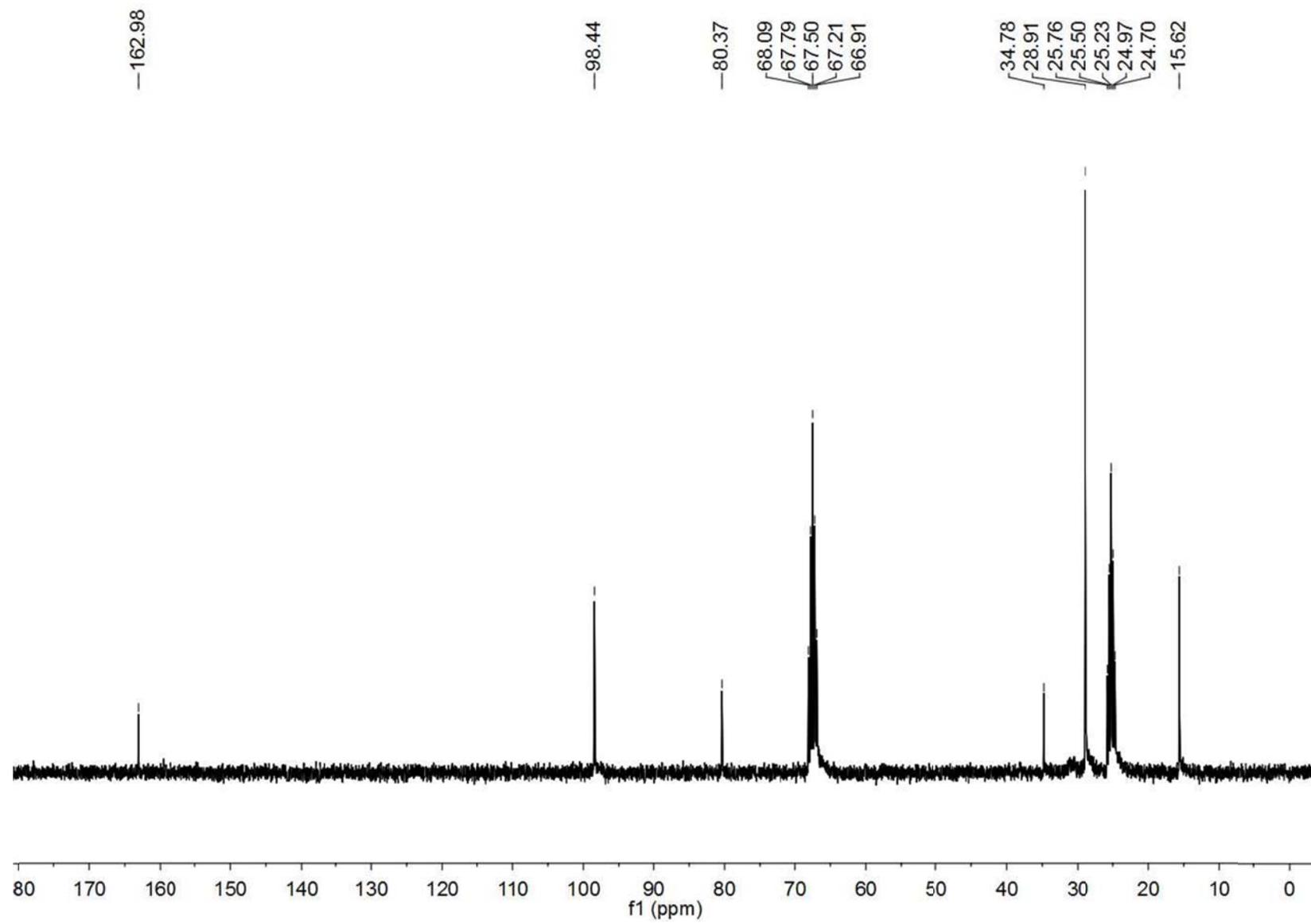
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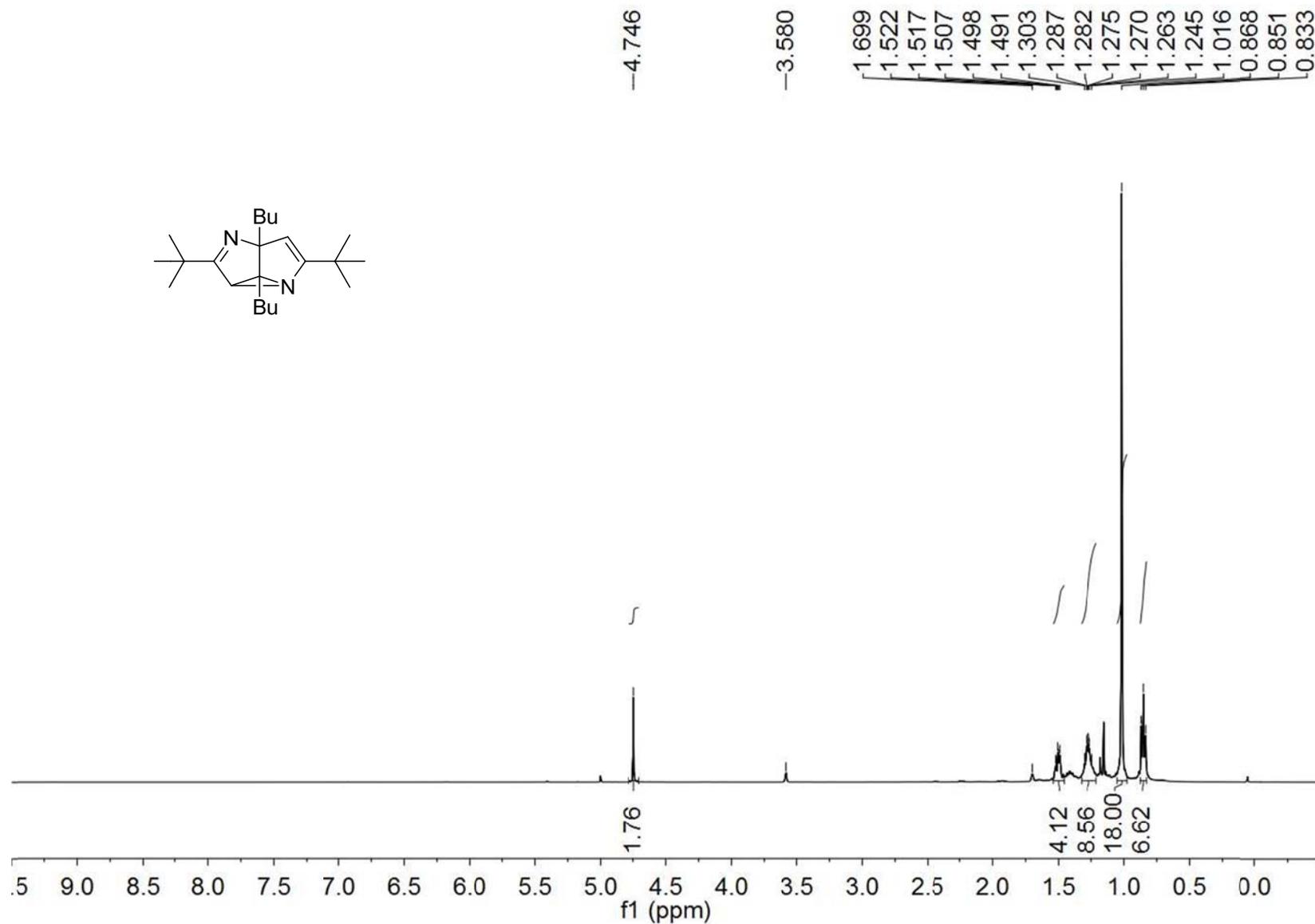
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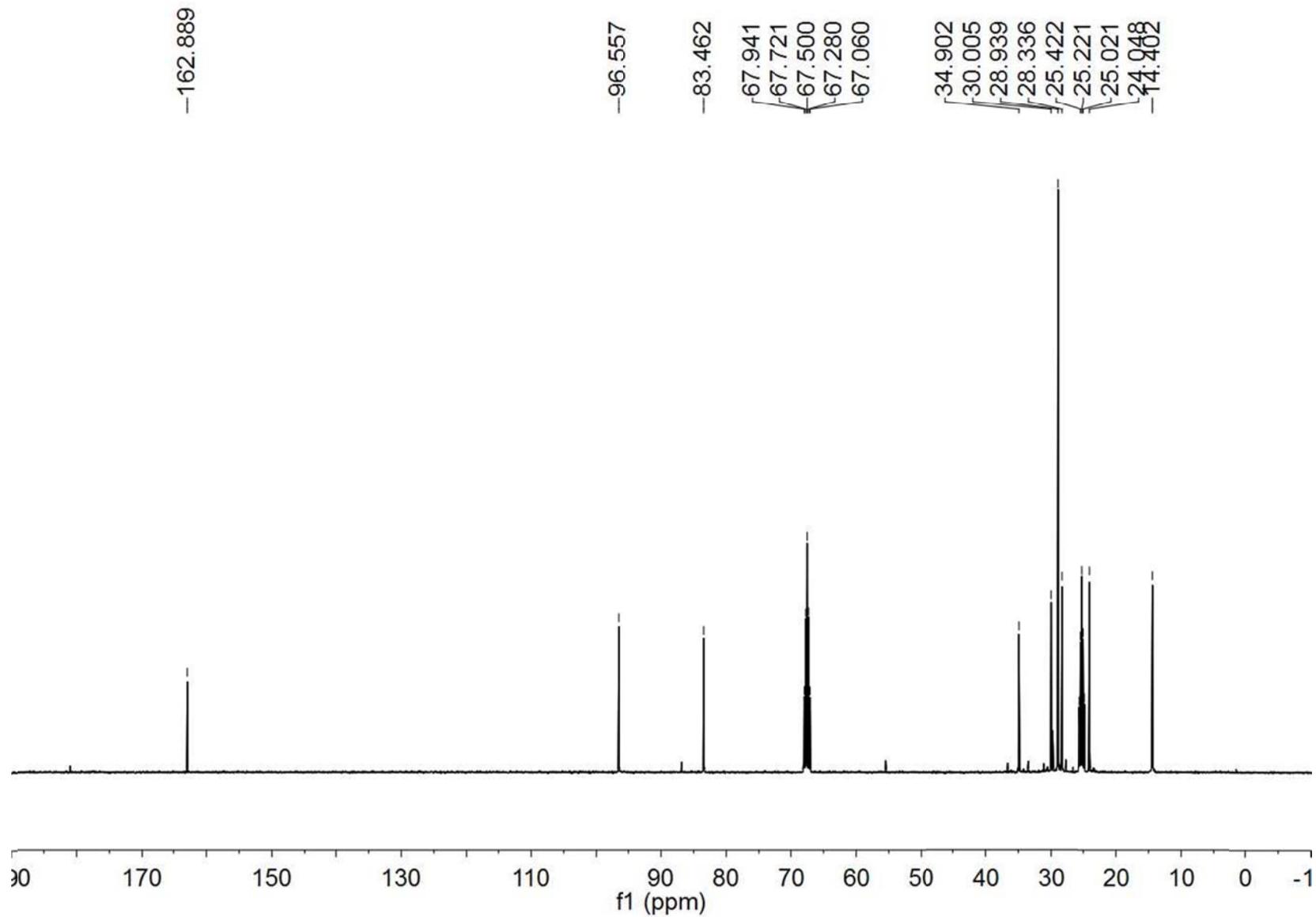
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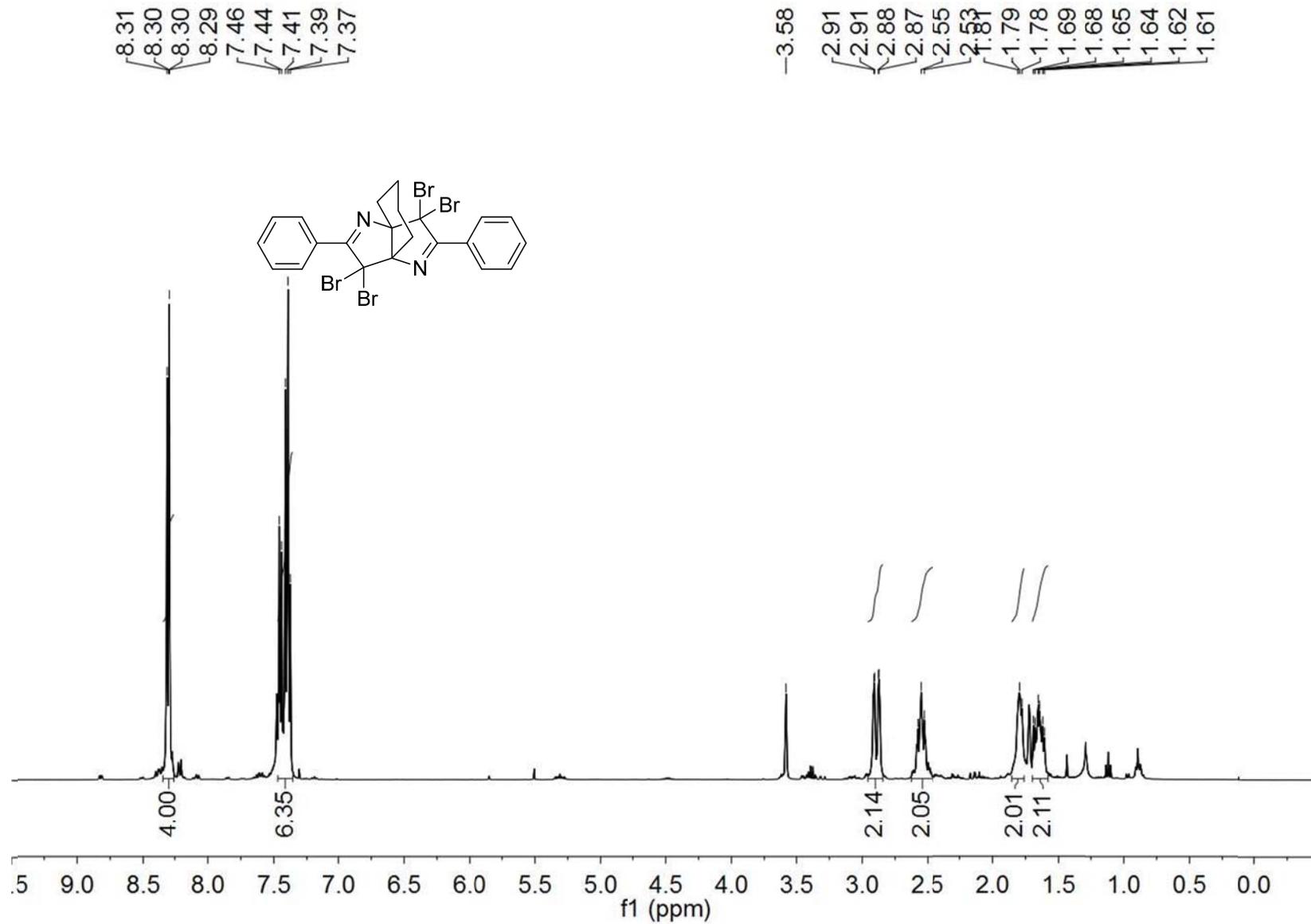
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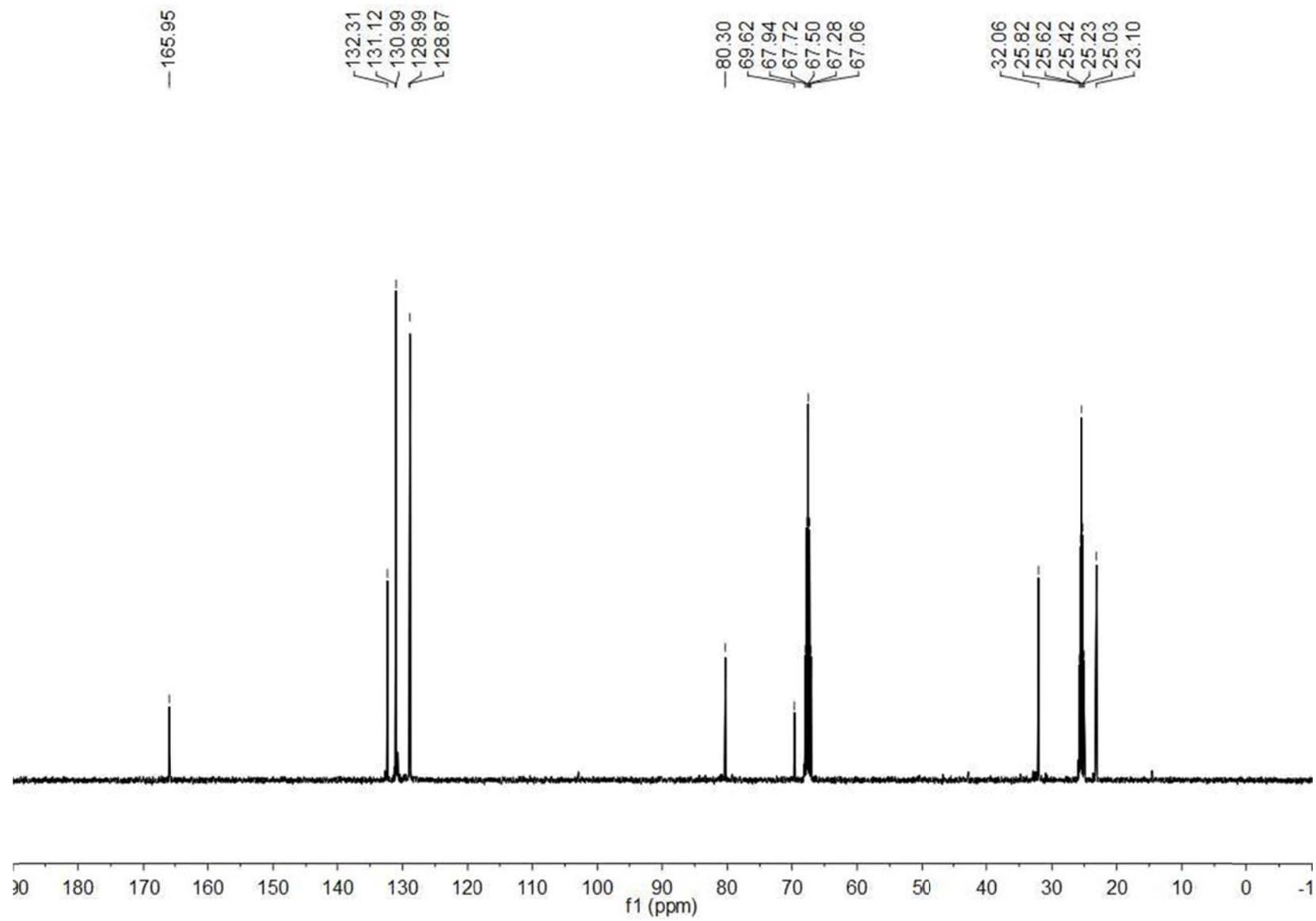
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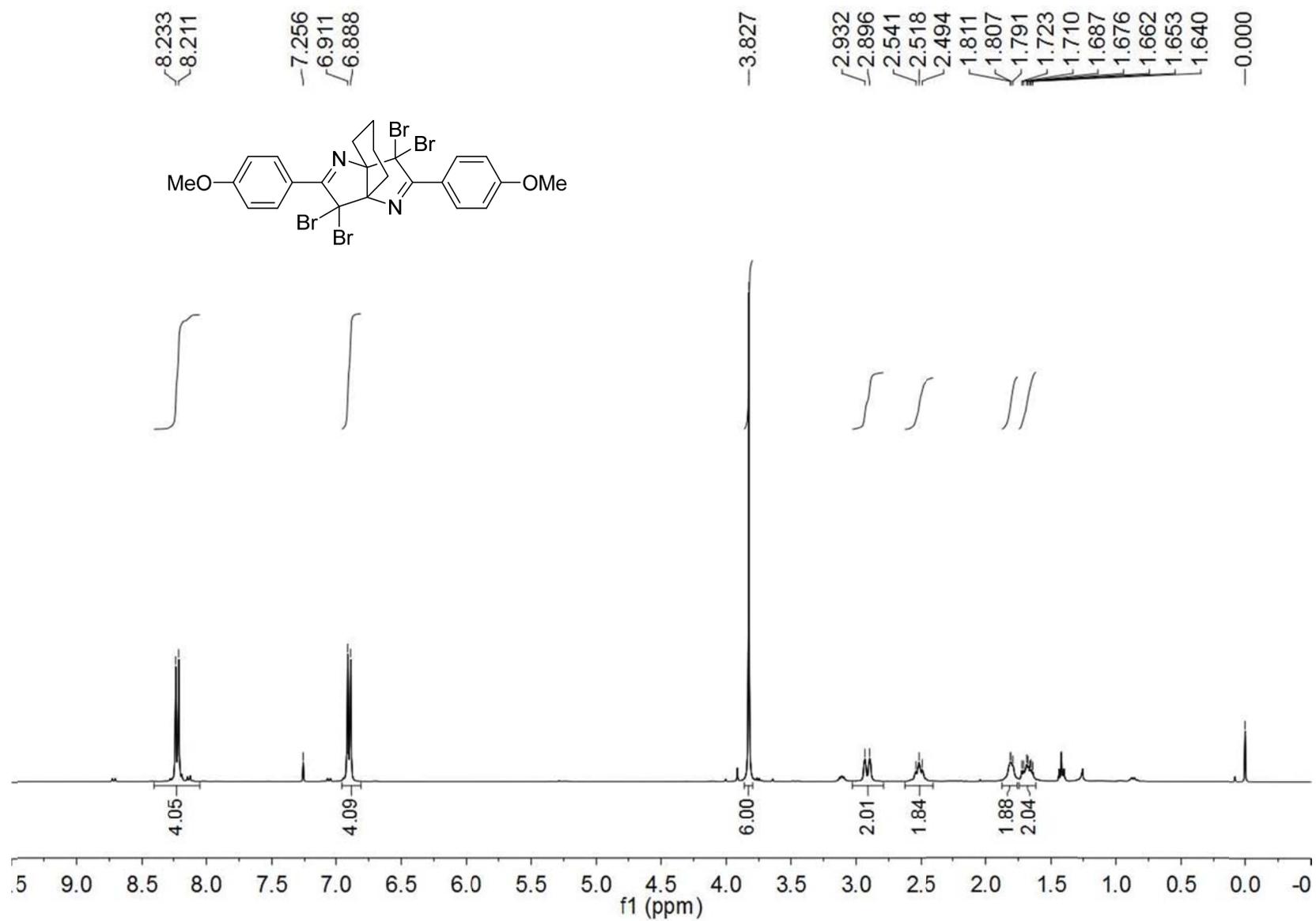
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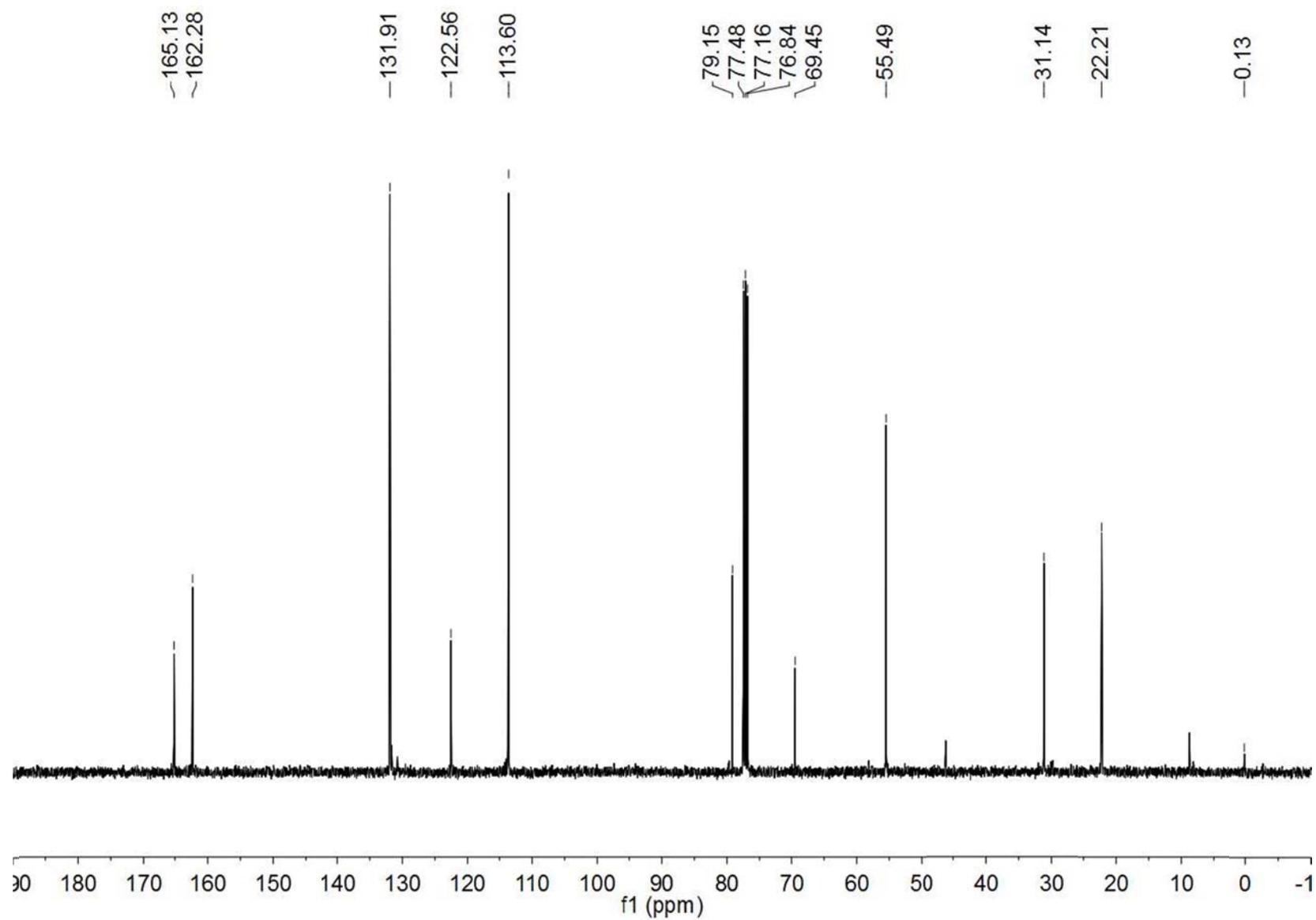
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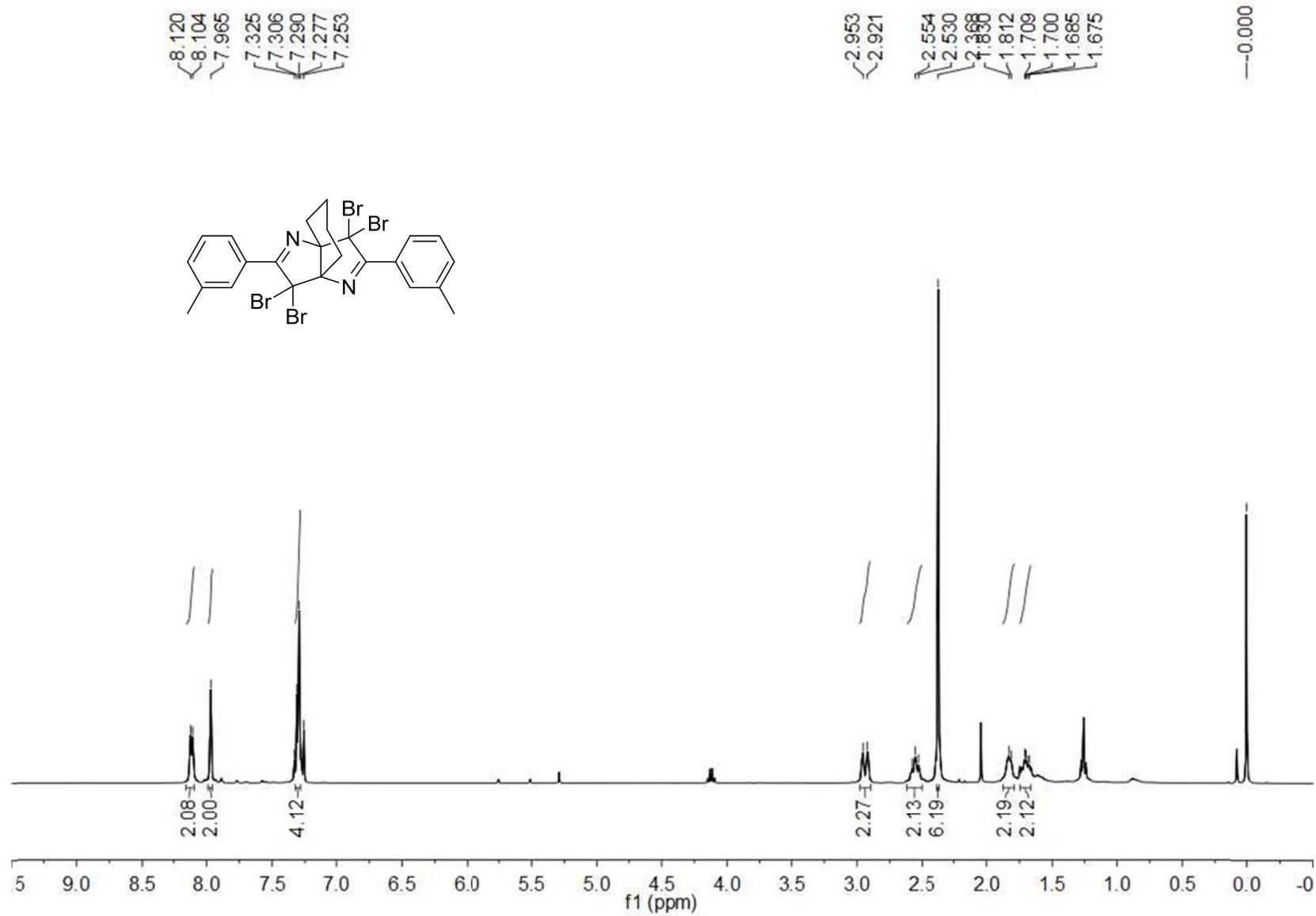
4b-<sup>1</sup>H NMR



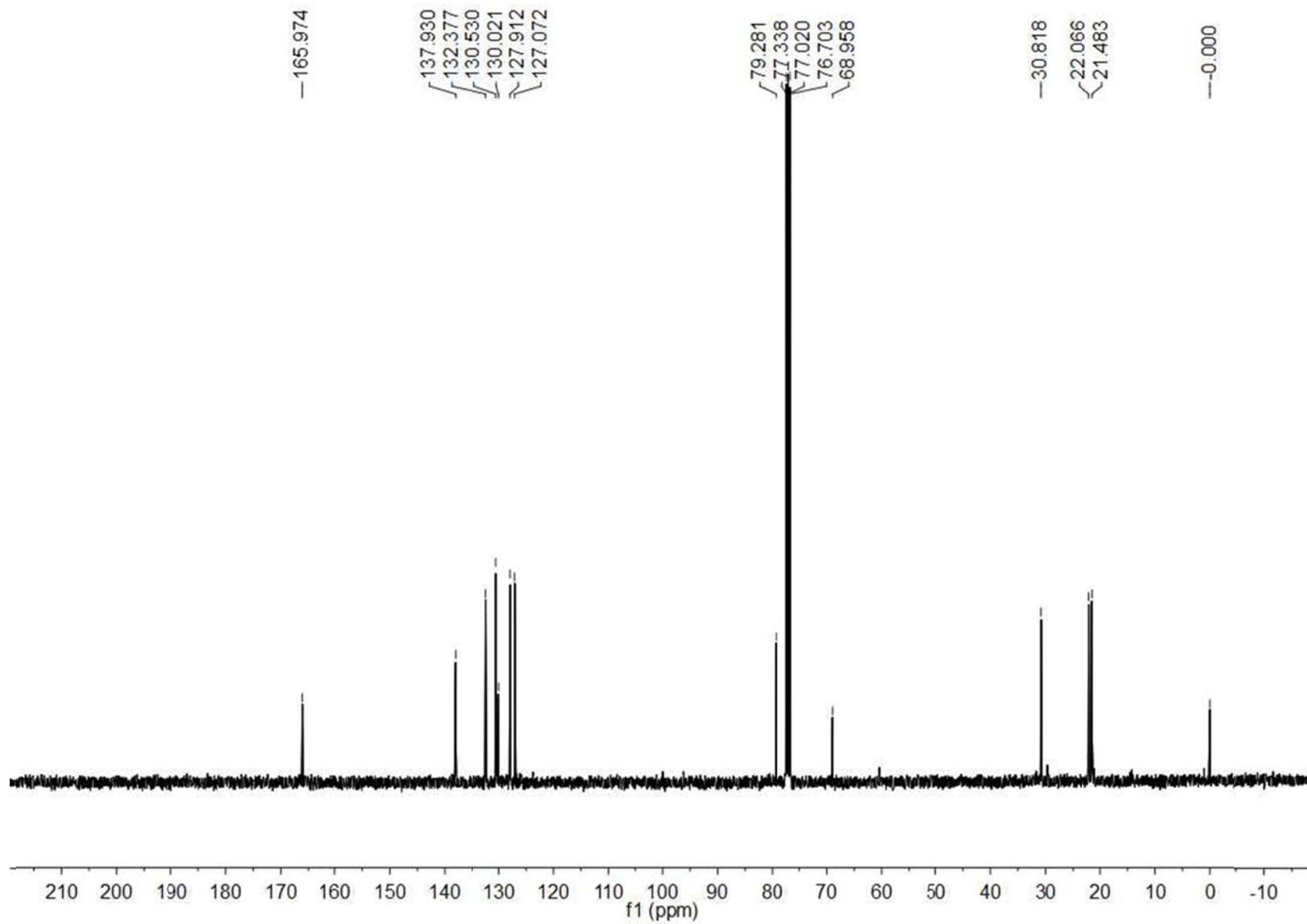
4b-<sup>13</sup>C NMR



4c-<sup>1</sup>H NMR

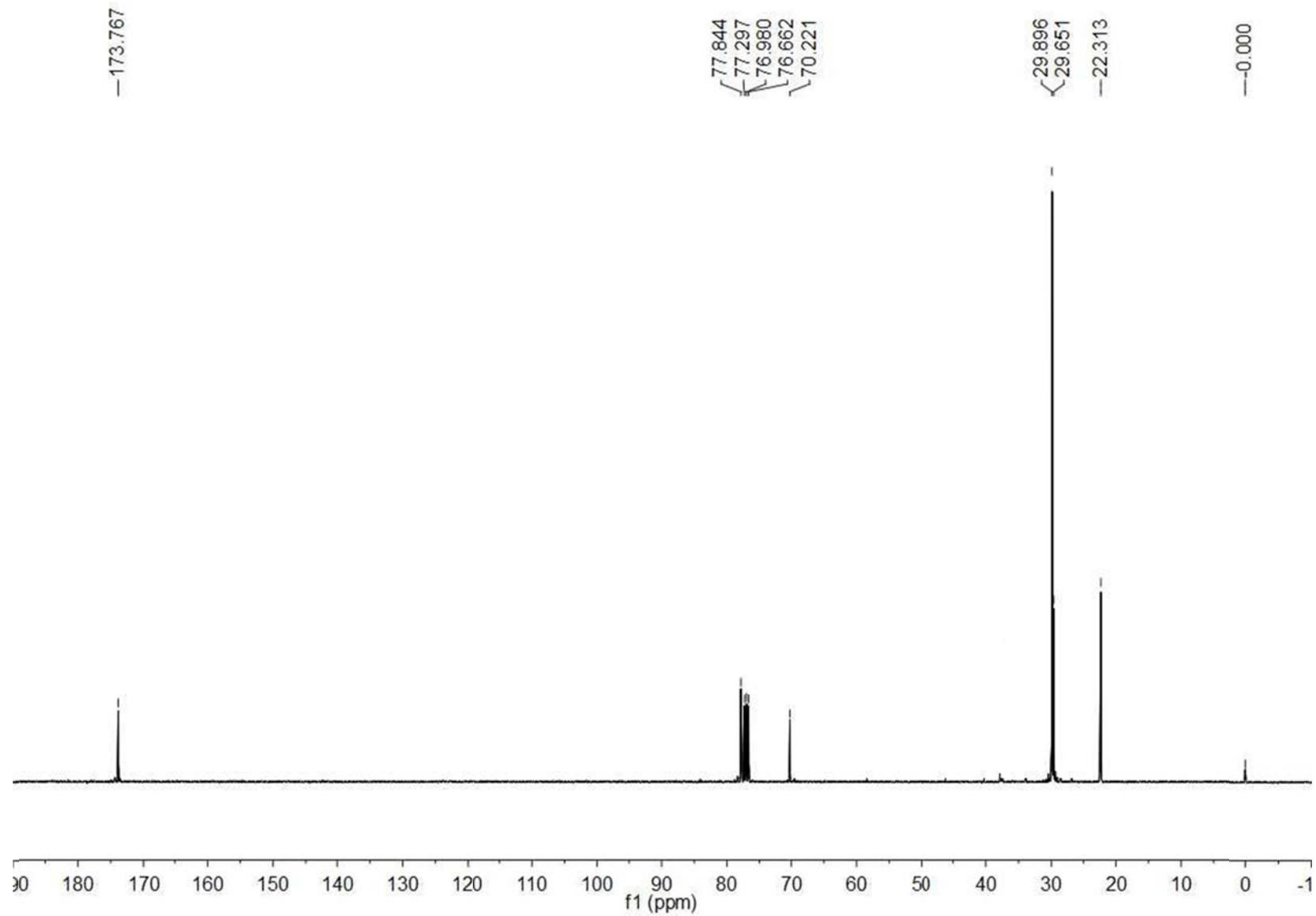


4c-<sup>13</sup>C NMR

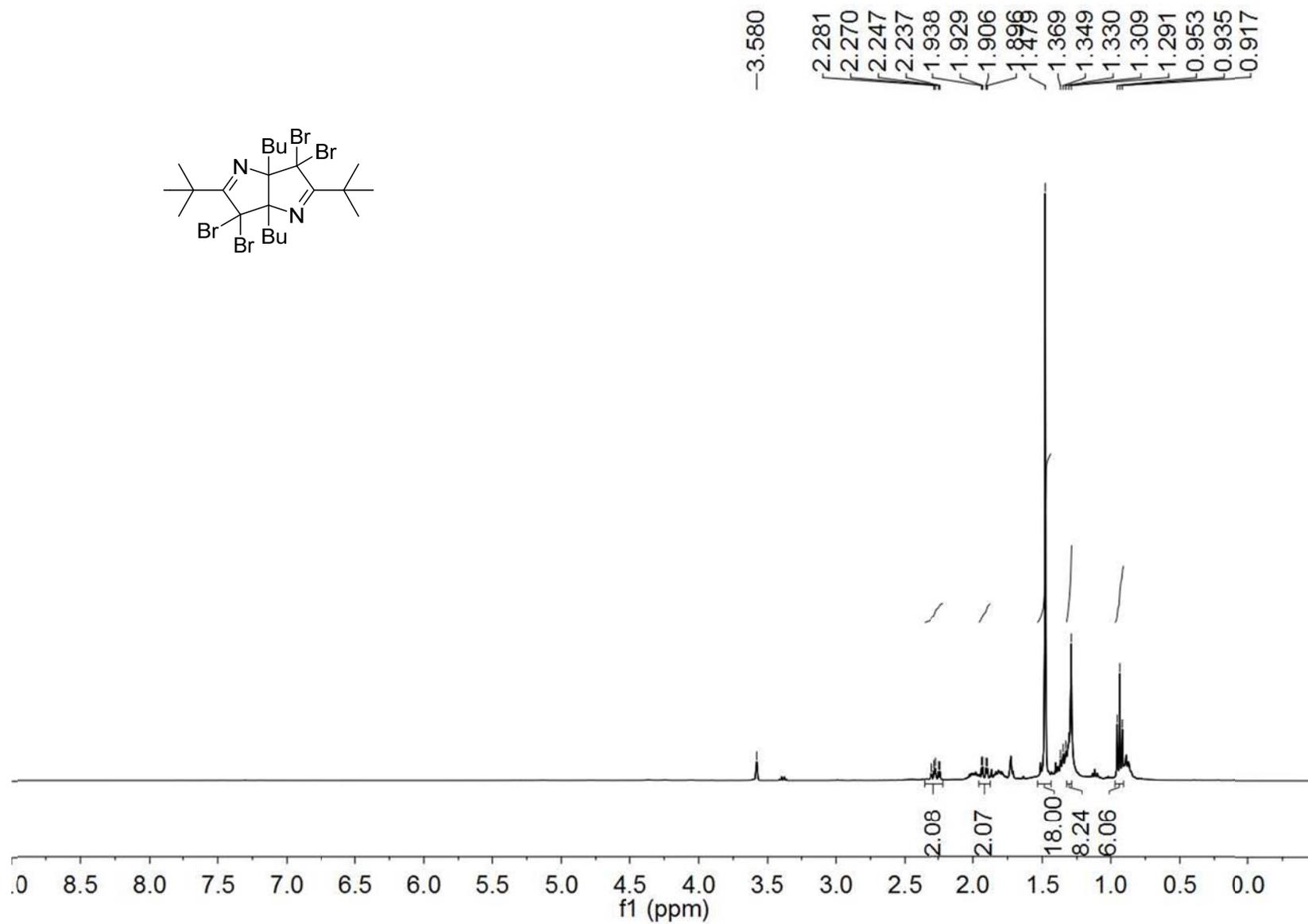
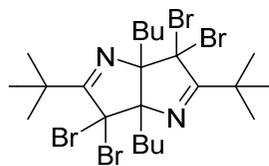




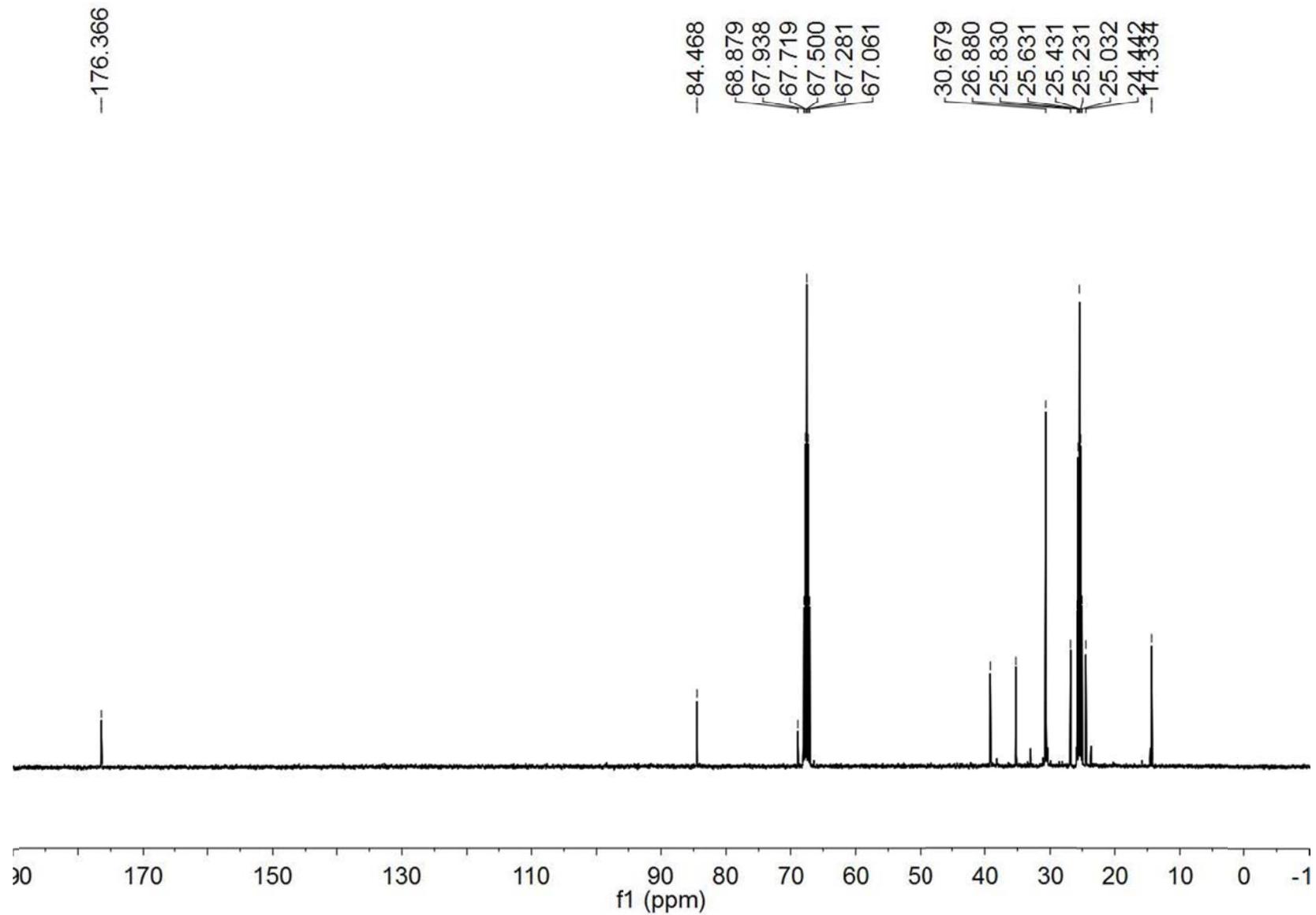
4d-<sup>13</sup>C NMR



4e-<sup>1</sup>H NMR



4e-<sup>13</sup>C NMR



5a-<sup>1</sup>H NMR

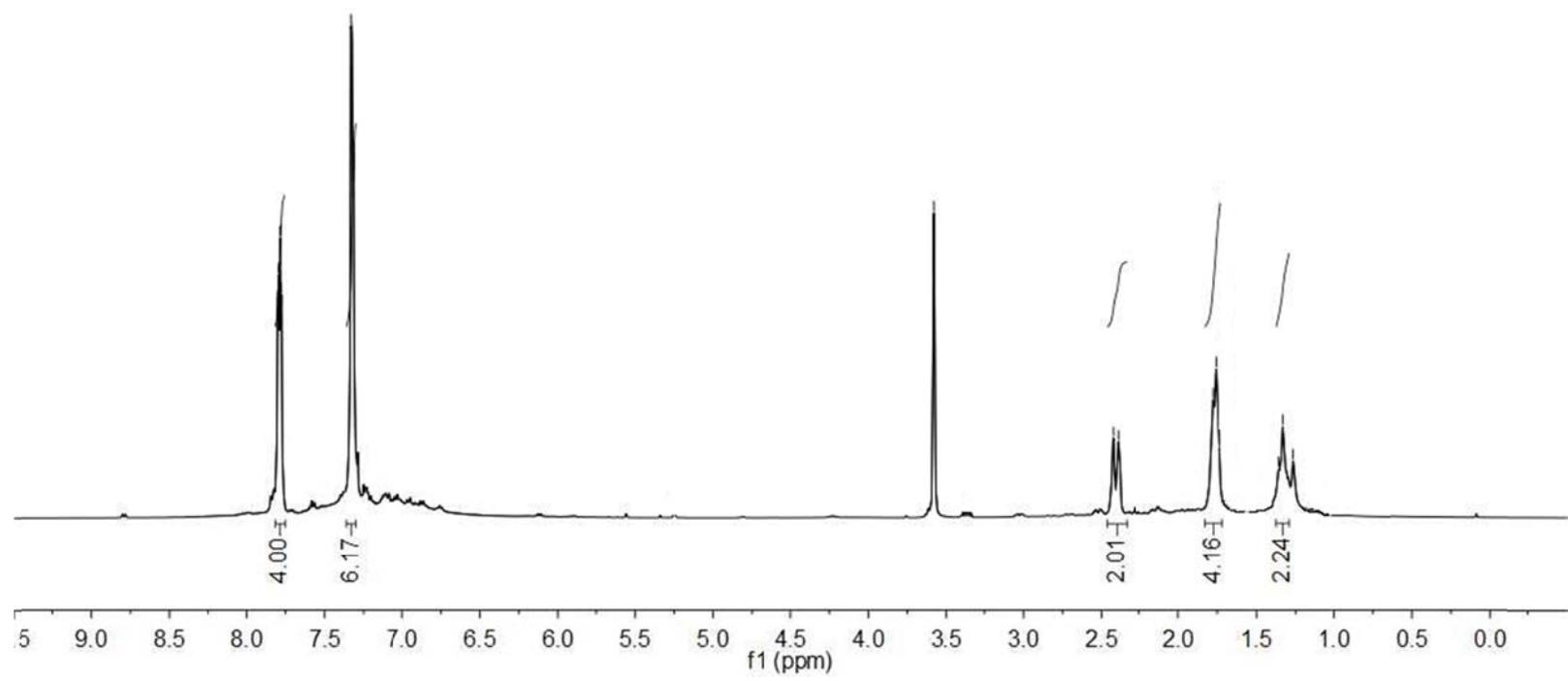
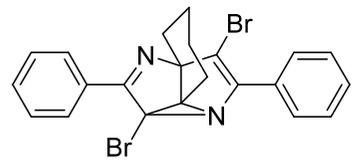
7.80  
7.79  
7.79  
7.78  
7.33  
7.32  
7.32

3.58

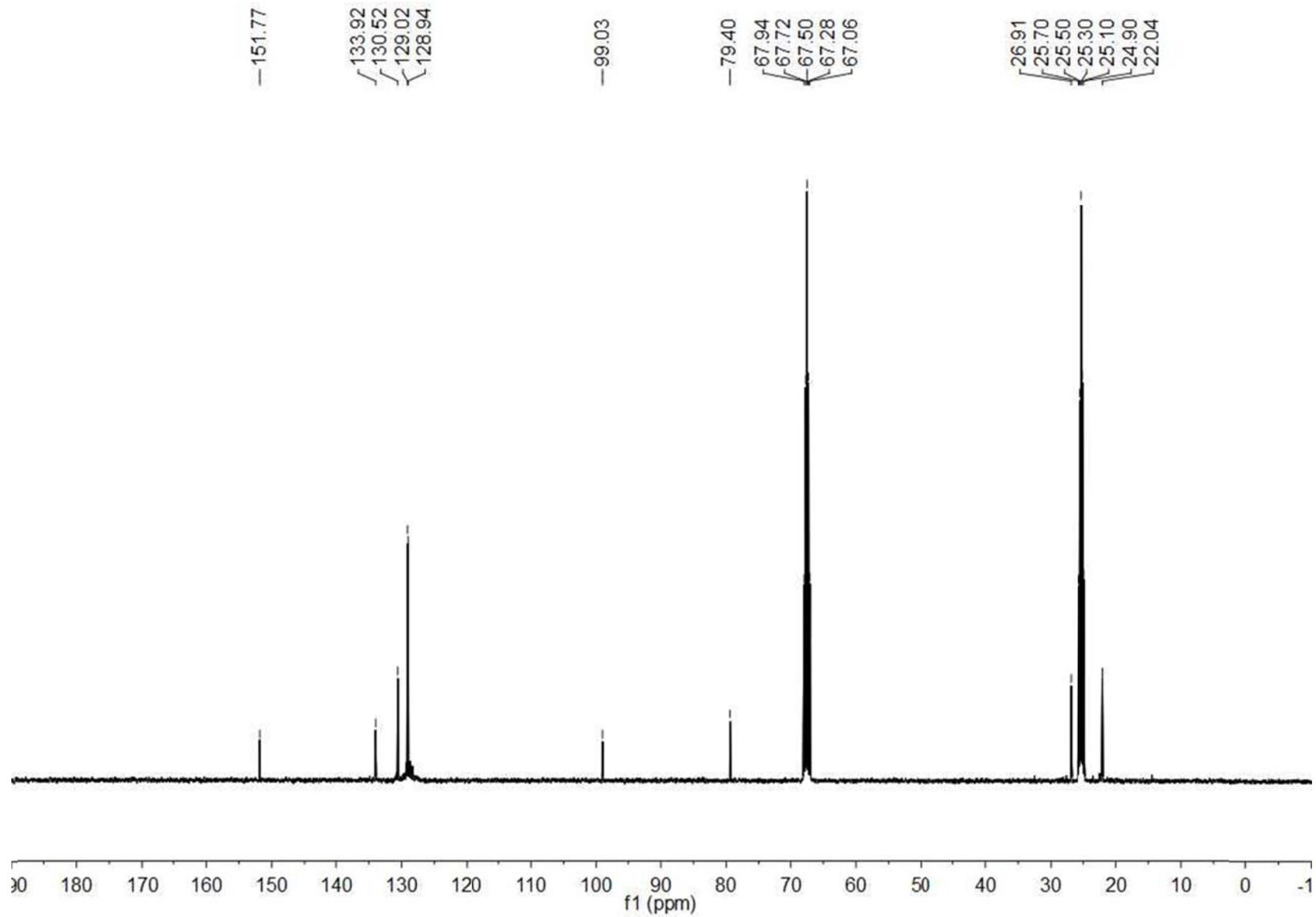
2.42  
2.39

1.78  
1.76

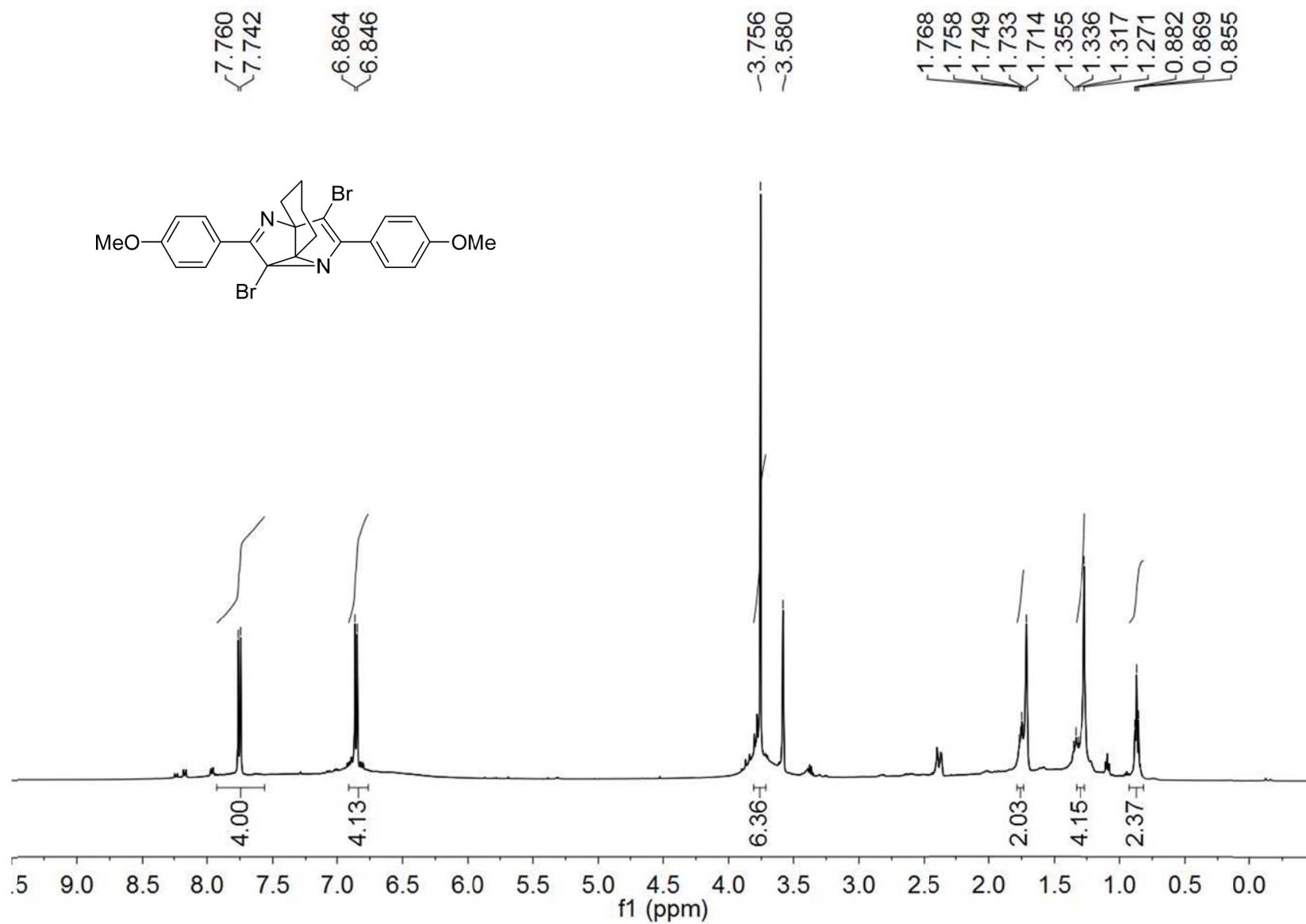
1.36  
1.33  
1.27



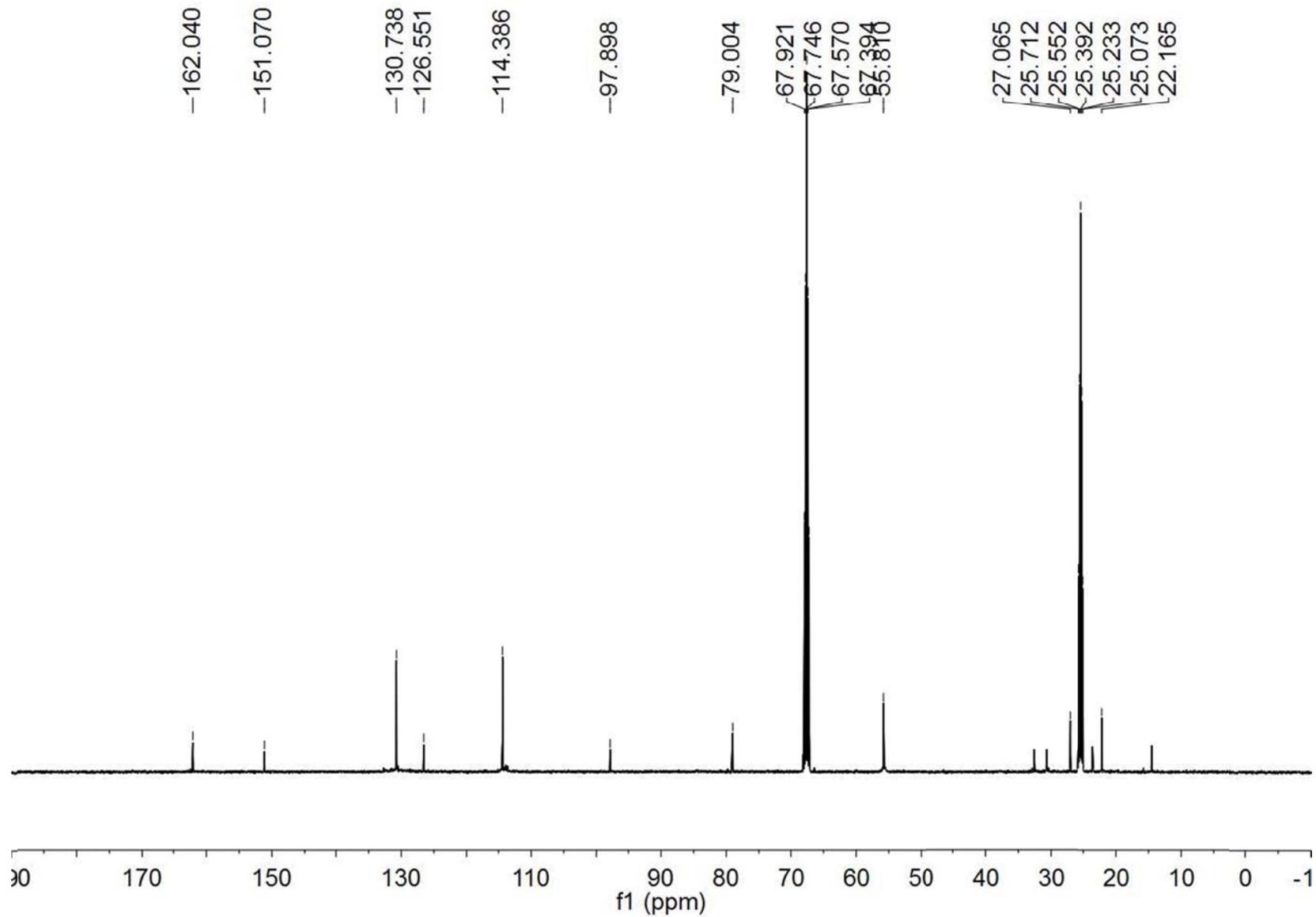
5a-<sup>13</sup>C NMR



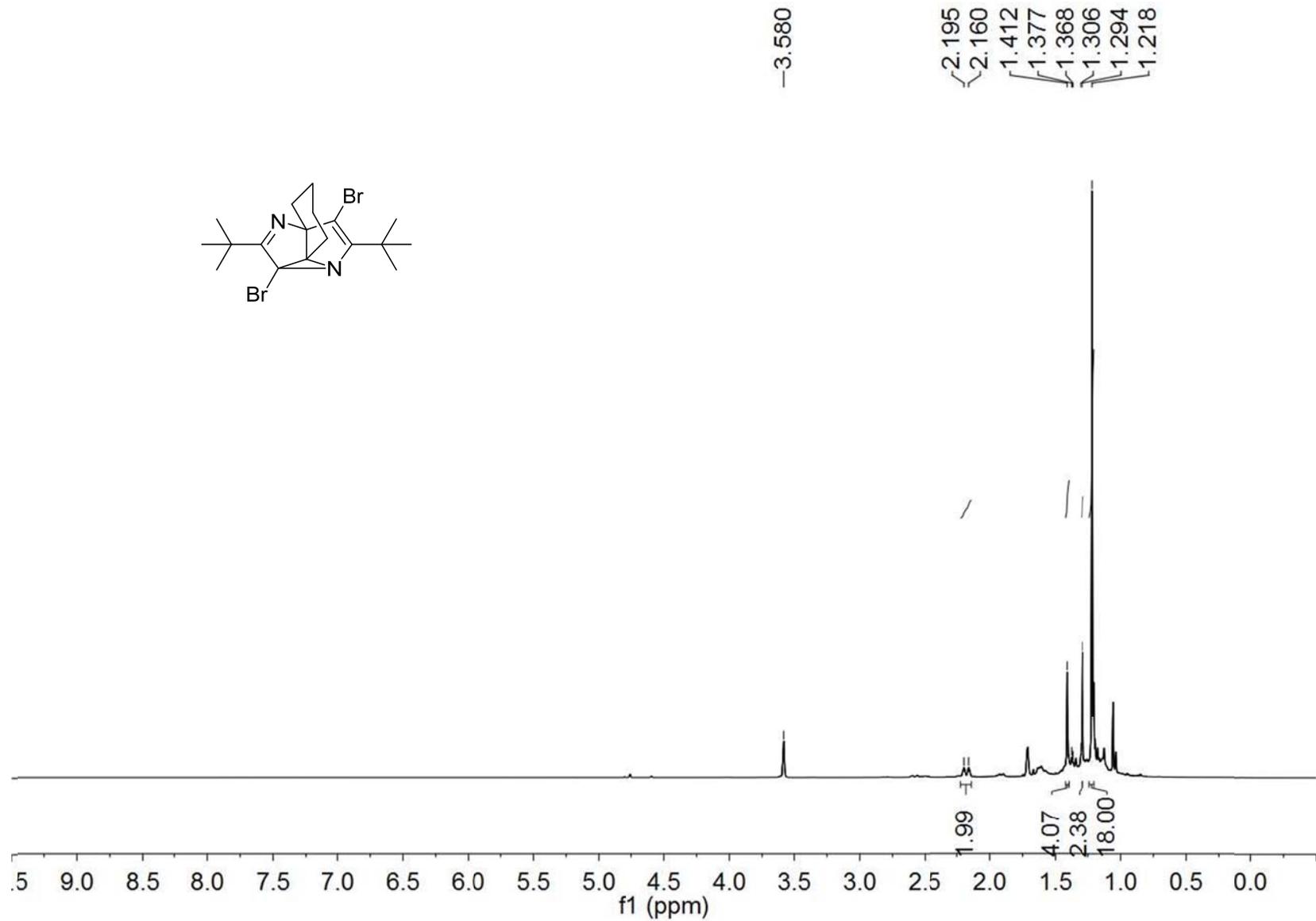
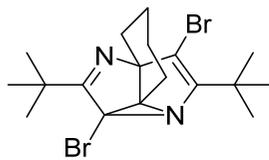
5b-<sup>1</sup>H NMR



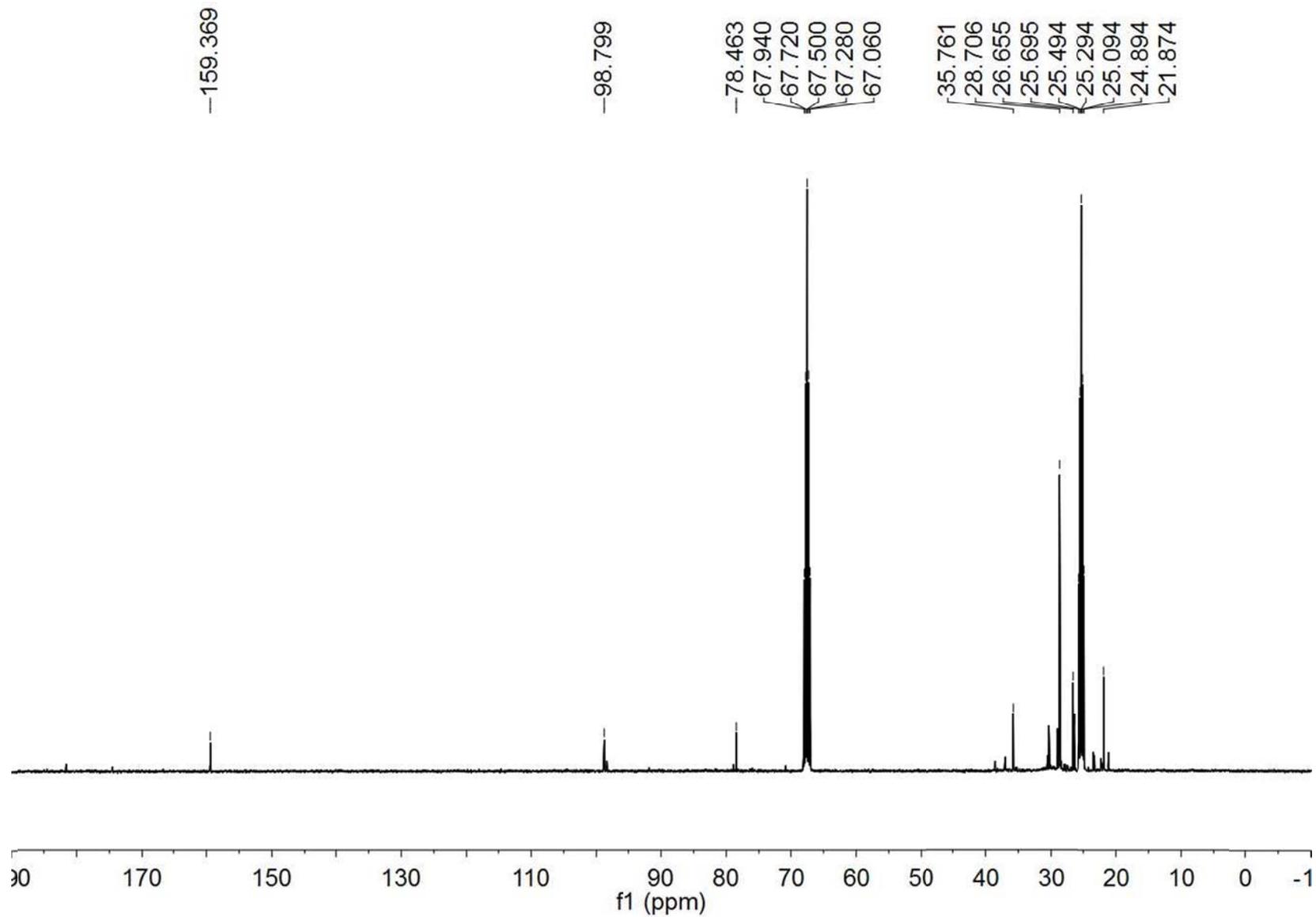
5b-<sup>13</sup>C NMR



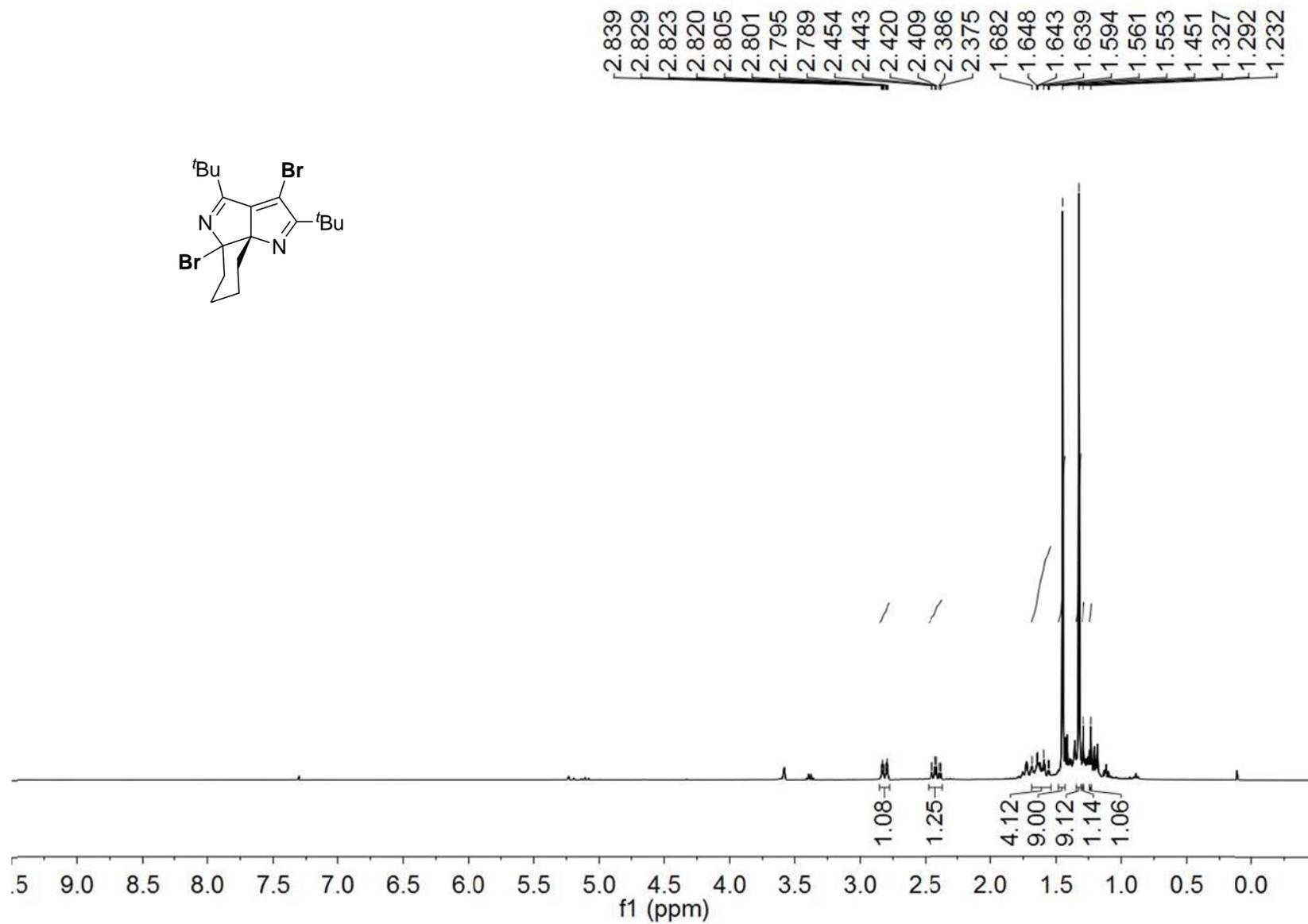
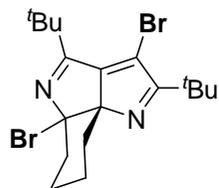
5c-<sup>1</sup>H NMR



5c-<sup>13</sup>C NMR



# 6-<sup>1</sup>H NMR



6-<sup>13</sup>C NMR

