

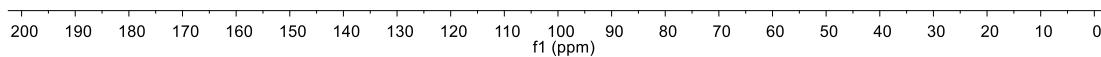
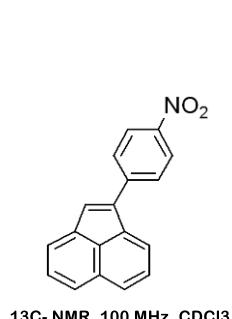
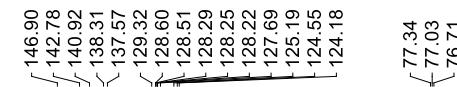
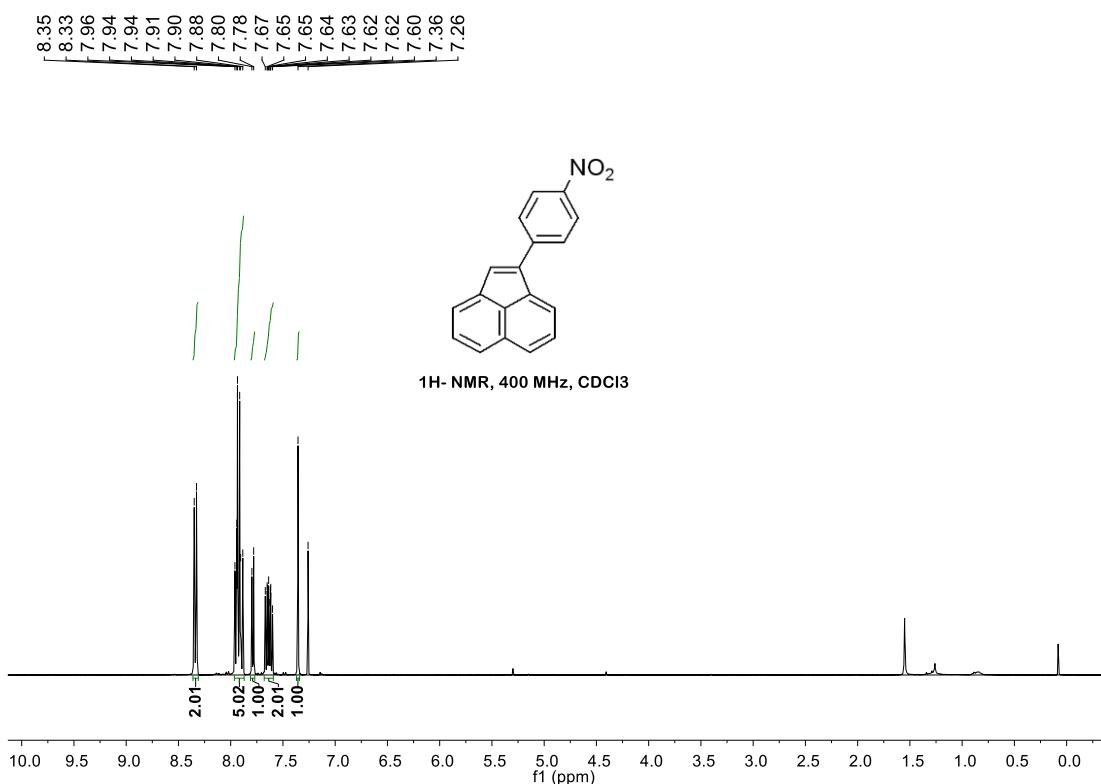
# Synthesis of mono- and di-arylated acenaphthylenes and programmed access to dibenzo[*j,l*]fluoranthenes *via* palladium-catalysed C-H bond functionalisation

Xinzhe Shi,<sup>a</sup> Thierry Roisnel,<sup>a</sup> Jean-François Soulé,<sup>\*a</sup> Henri Doucet<sup>\*a</sup>

<sup>a</sup> Institut des Sciences Chimiques de Rennes, UMR 6226 CNRS-Université de Rennes 1 "Organométalliques: Matériaux et Catalyse", 35042 Rennes Cedex, France. Tel: 33 (0)2 23 23 63 84; E-mail: jean-francois.soule@univ-rennes1.fr; henri.doucet@univ-rennes1.fr.

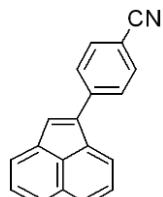
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2. X-ray analysis of <b>23</b> (CDC 1553883) and <b>33</b> (CDC 1553881)	S 35-44

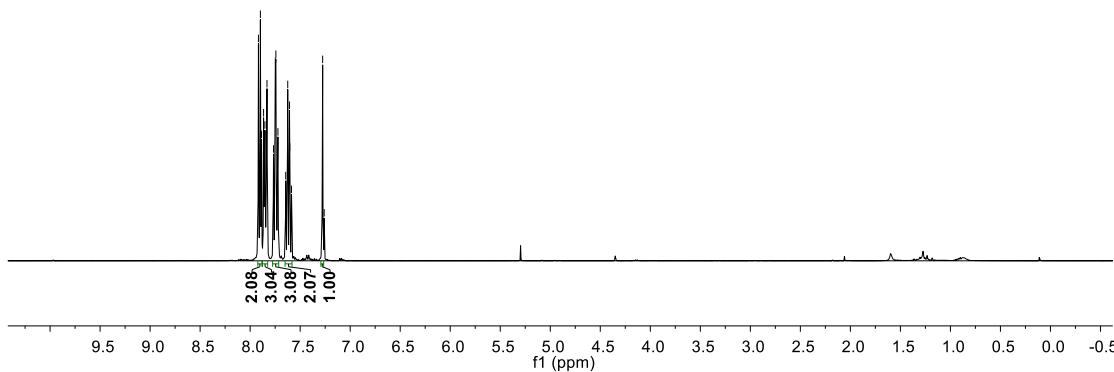
**1a**

**2**

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7.90  
7.89  
7.87  
7.85  
7.83  
7.77  
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7.74  
7.72  
7.64  
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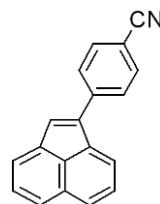


1H-NMR, 400 MHz, CDCl<sub>3</sub>

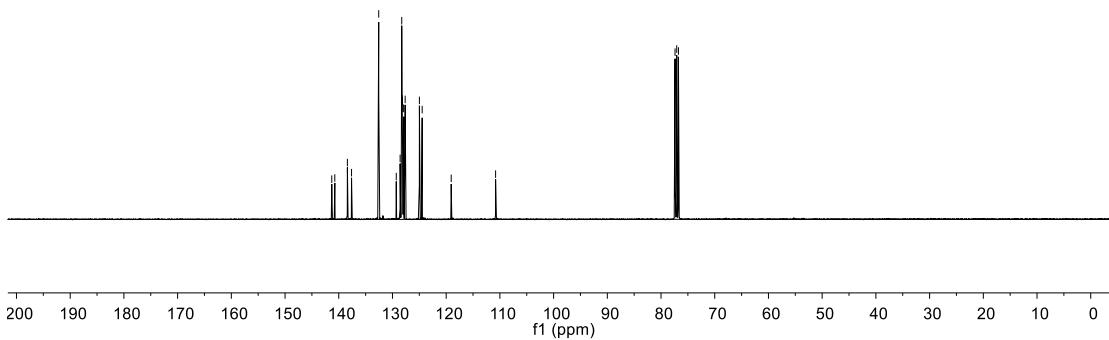


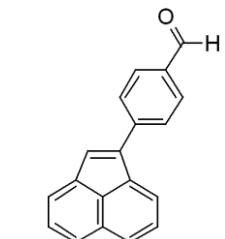
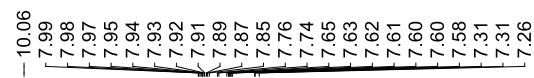
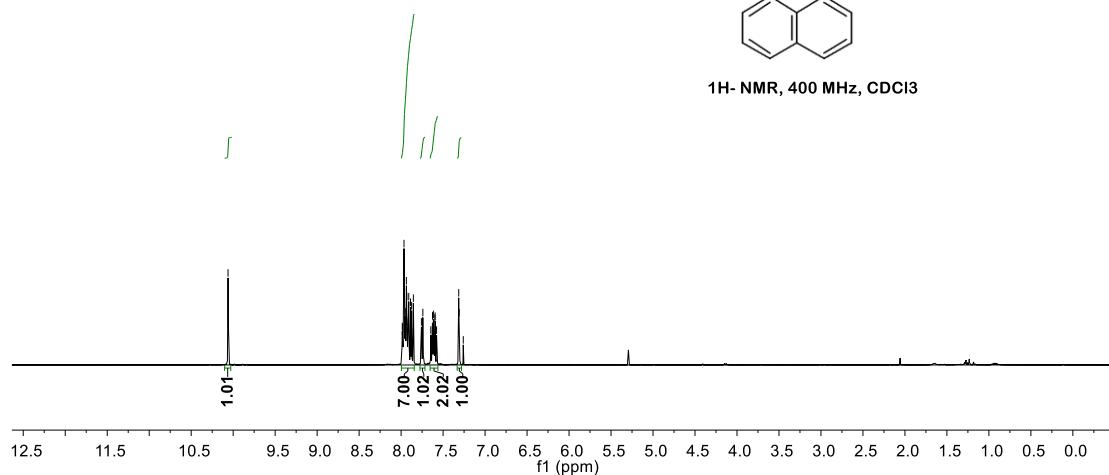
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127.64  
124.98  
124.47  
119.06  
110.80

77.39  
77.07  
76.76

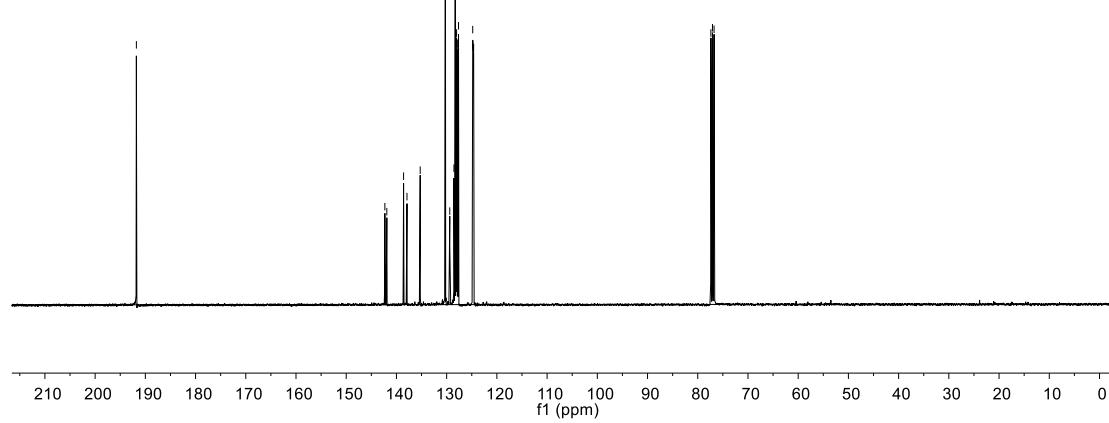


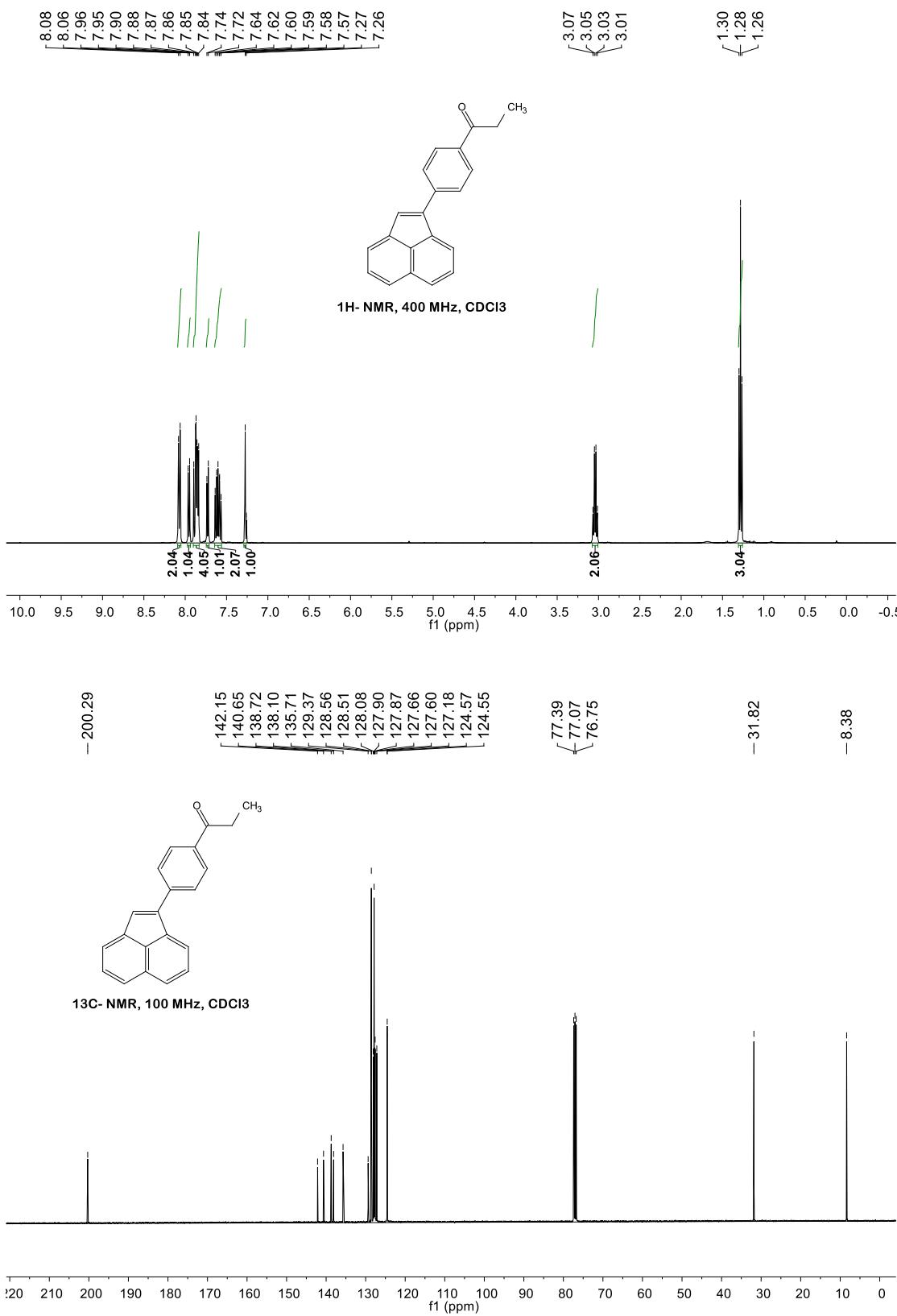
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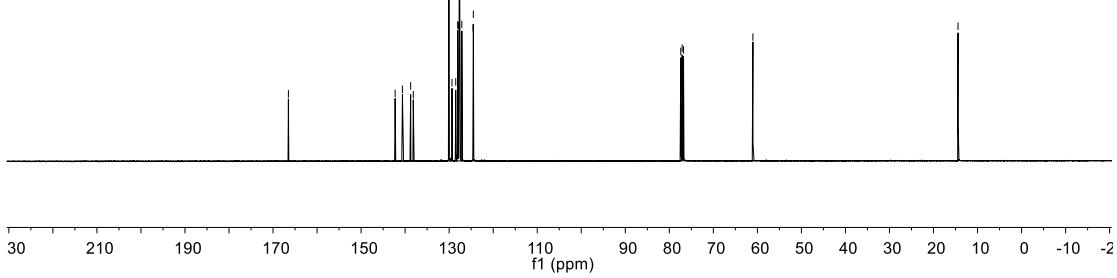
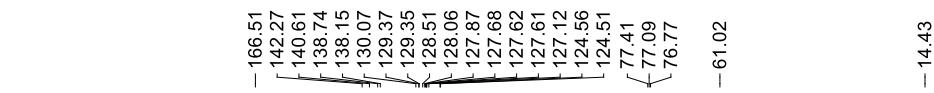
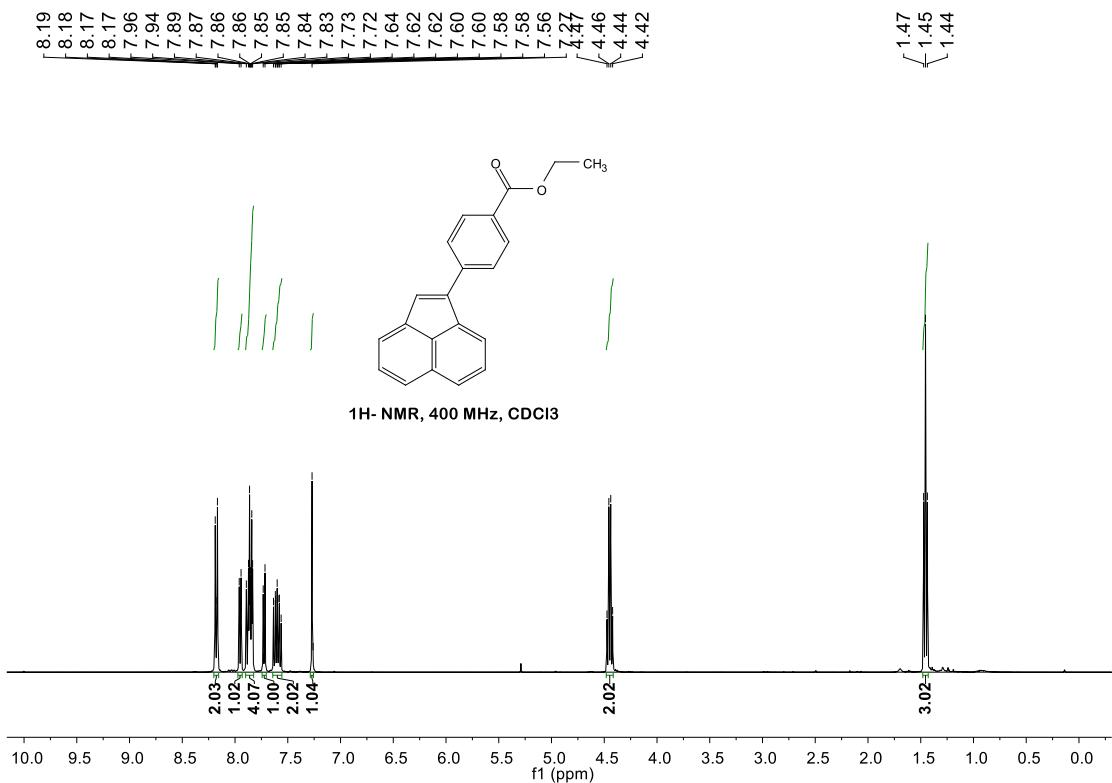


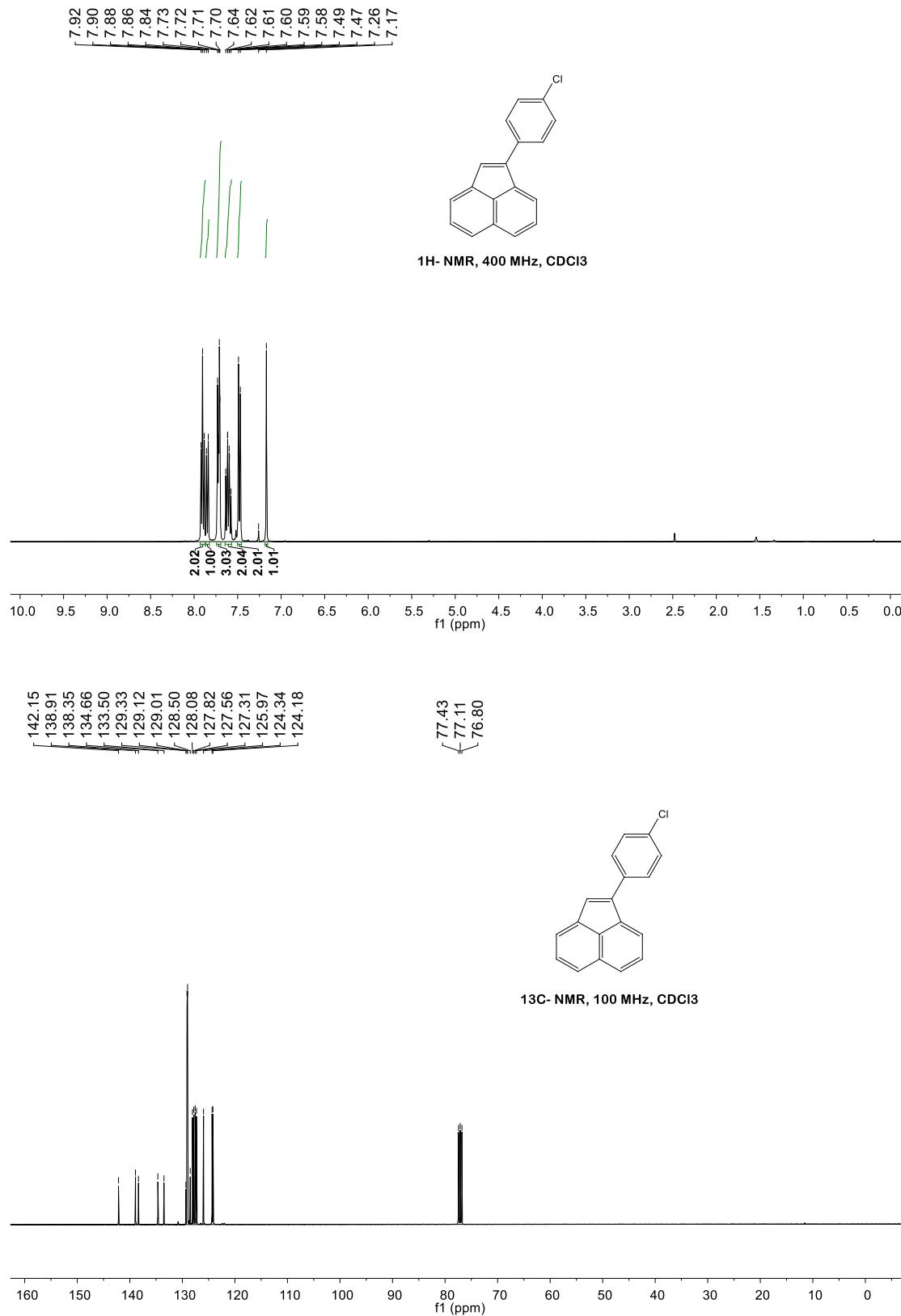
**3**1H- NMR, 400 MHz, CDCl<sub>3</sub>

— 191.79

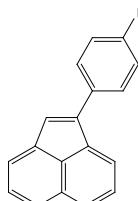
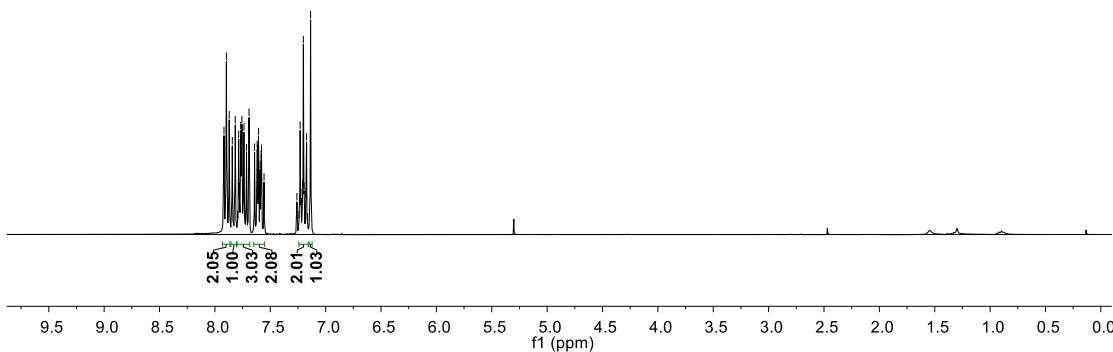
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135.27  
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128.02  
127.91  
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76.7513C- NMR, 100 MHz, CDCl<sub>3</sub>

**4**

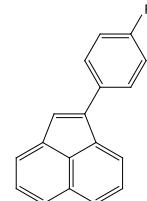
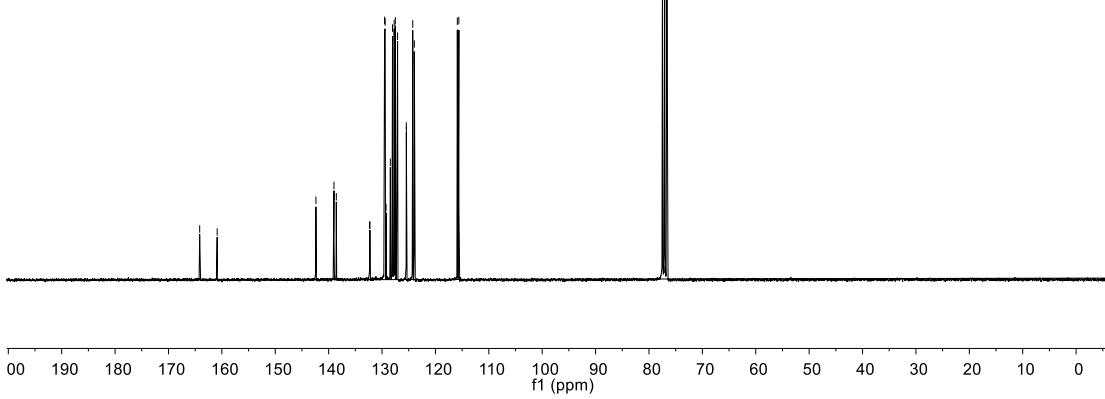
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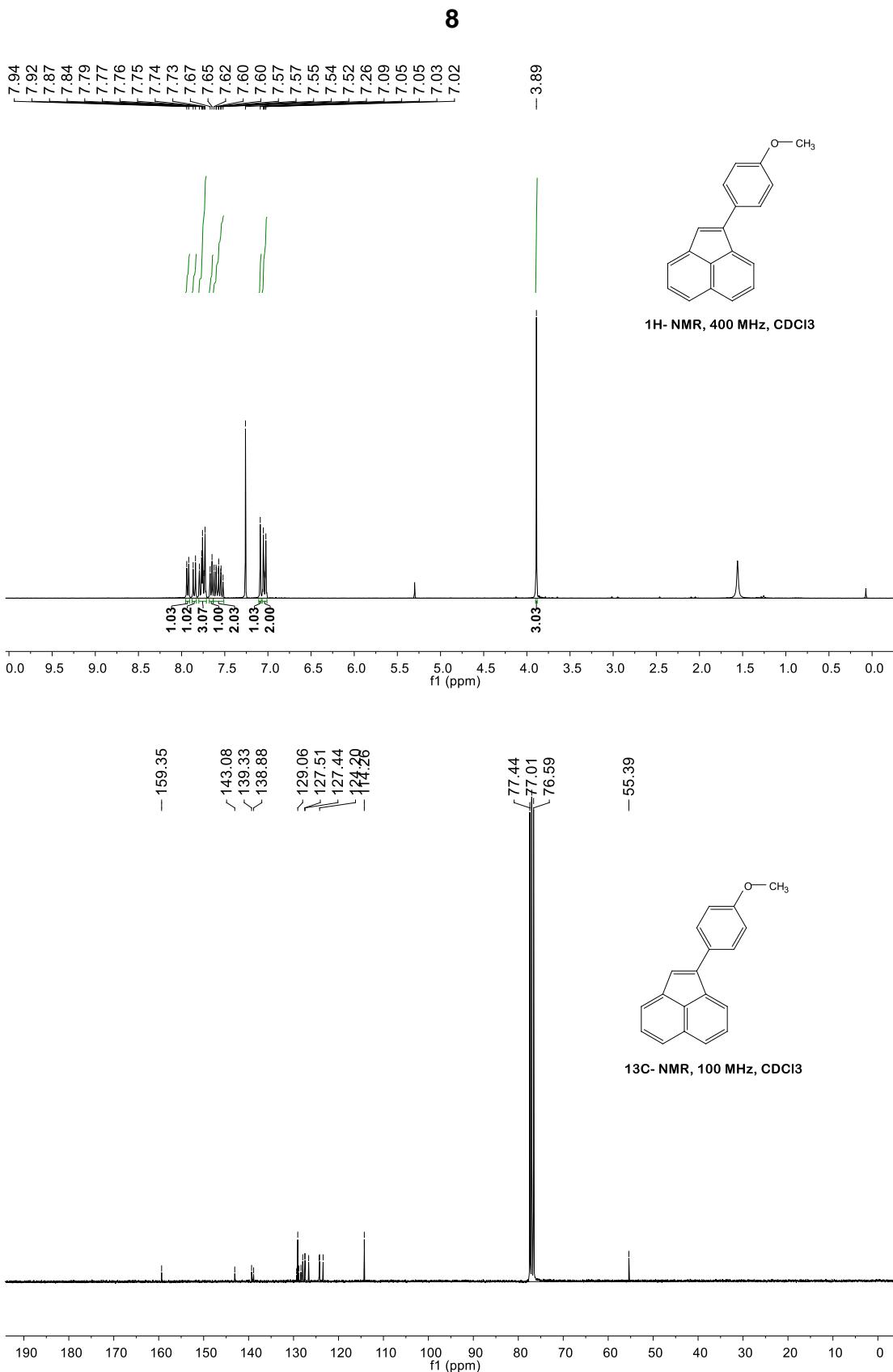
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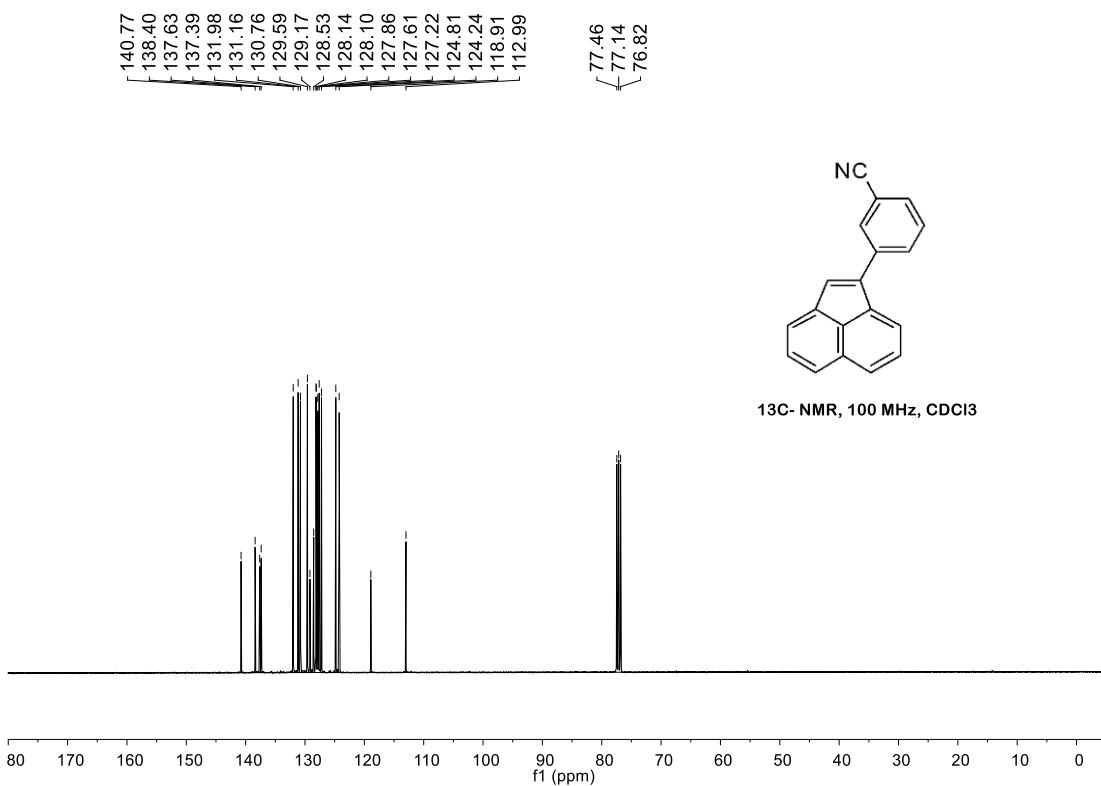
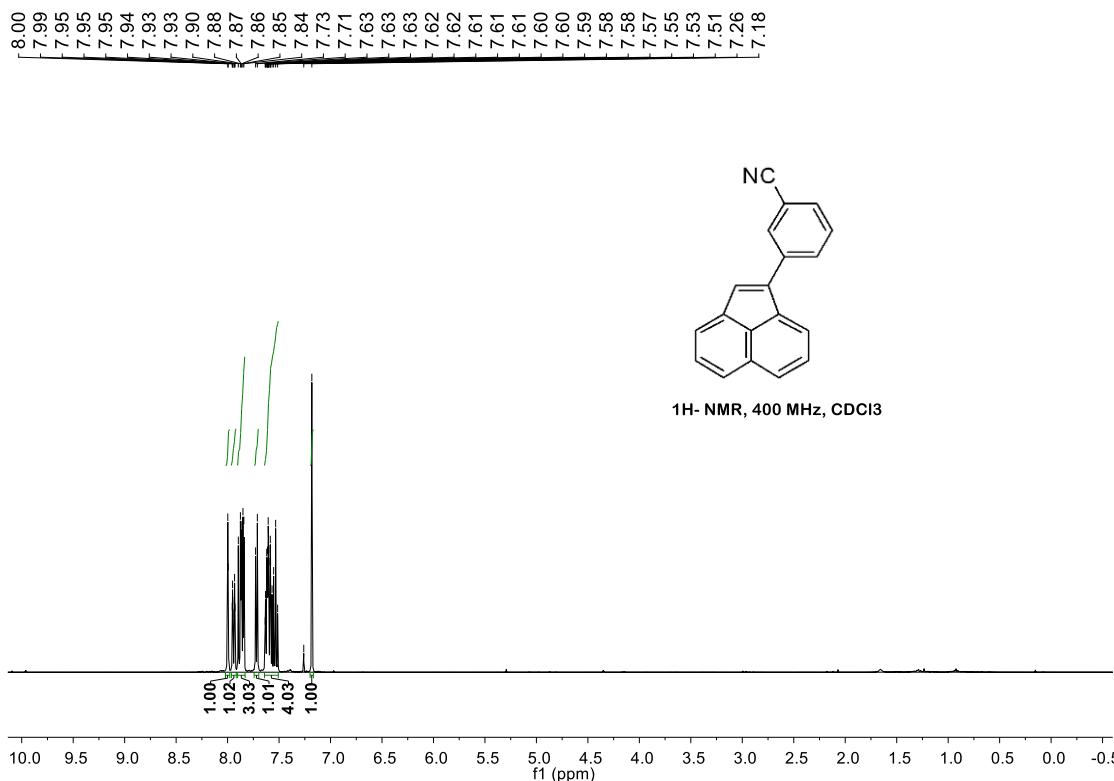
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7.90  
7.87  
7.84  
7.82  
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7.79  
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7.77  
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7.75  
7.74  
7.73  
7.71

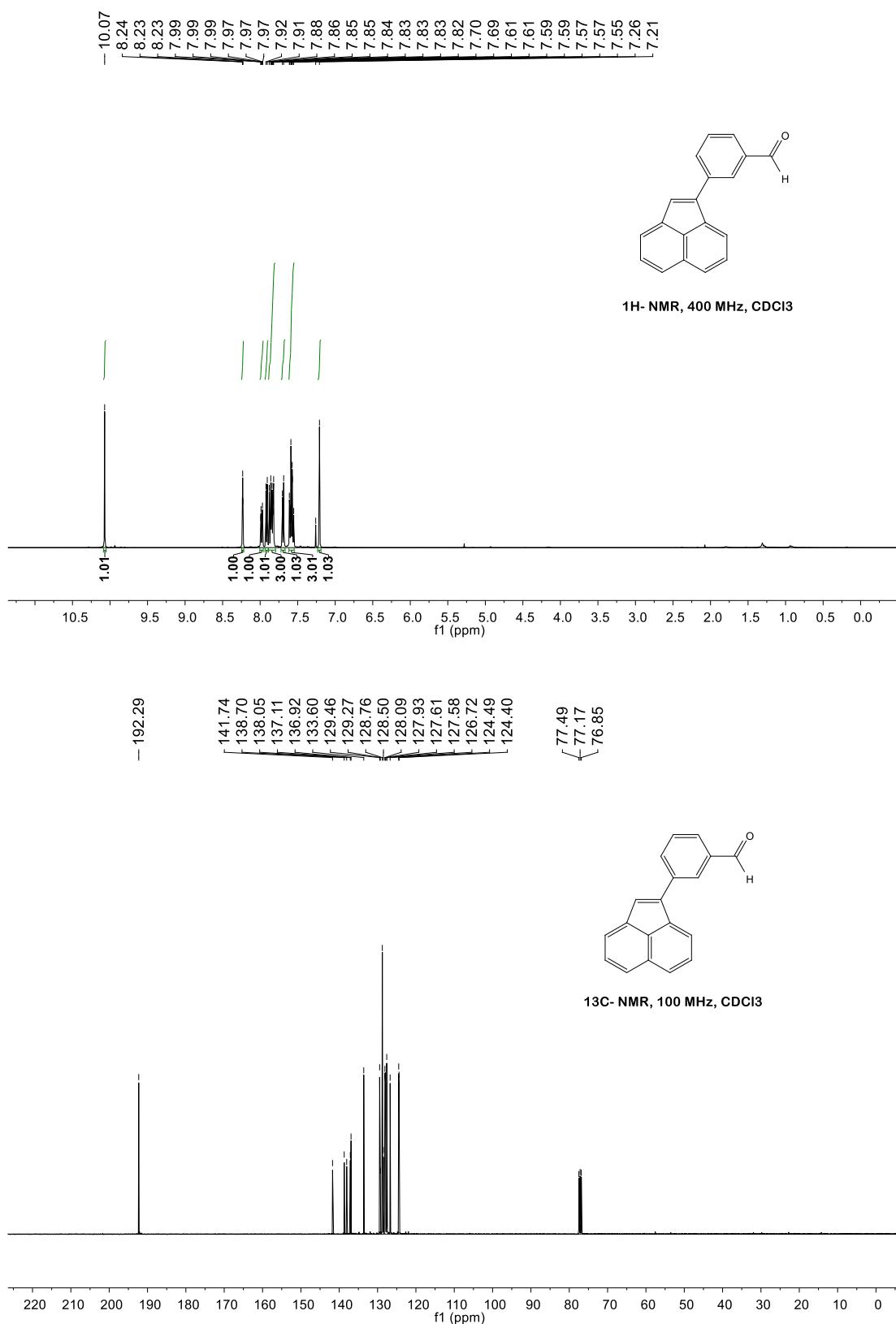
1H- NMR, 400 MHz, CDCl<sub>3</sub>

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-160.88  
142.38  
139.00  
138.56  
139.56  
132.27  
129.52  
129.41  
129.25  
128.44  
128.03  
127.73  
127.51  
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77.05  
76.63

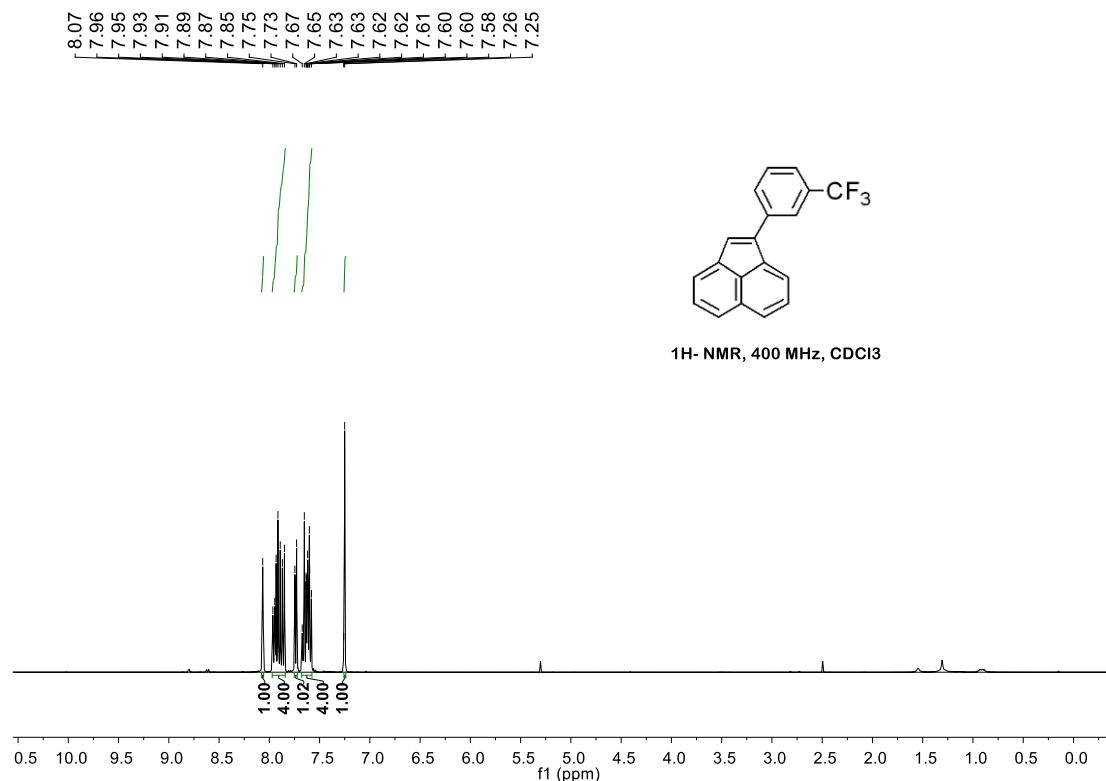
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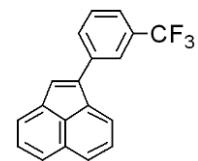
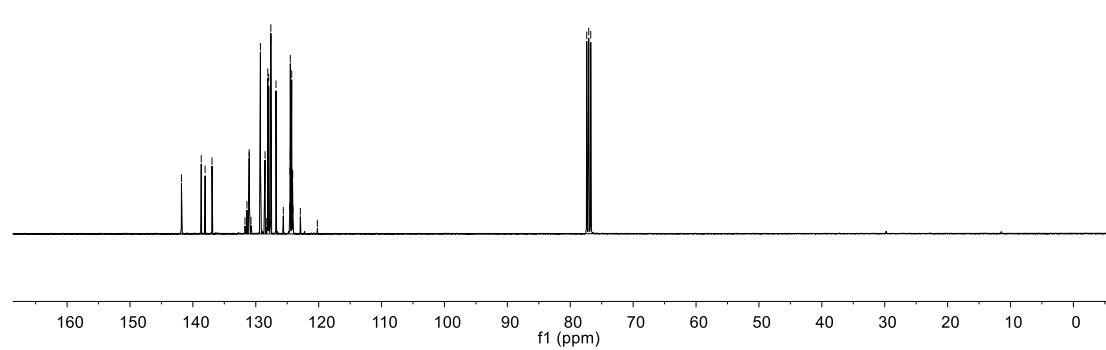
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**10**

11

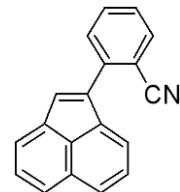
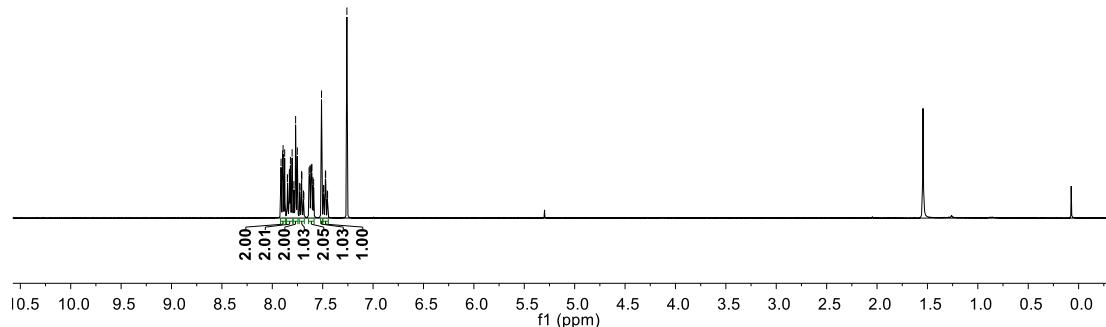


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138.67
138.06
136.96
131.40
131.05
131.04
129.26
128.53
128.09
127.95
127.60
127.59
126.78
125.62
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77.06
76.74

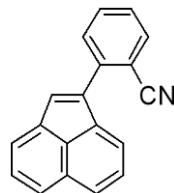
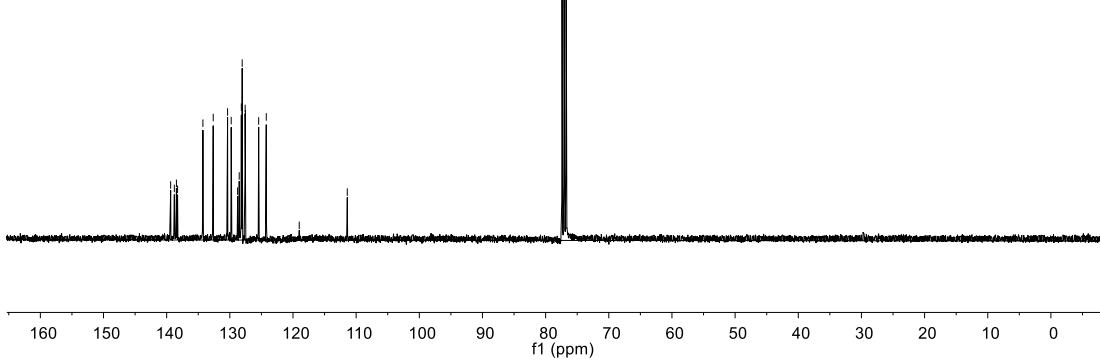
**13C-NMR, 100 MHz, CDCl<sub>3</sub>**

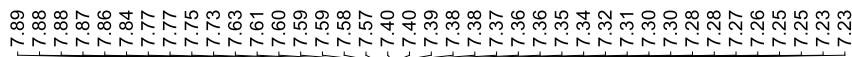
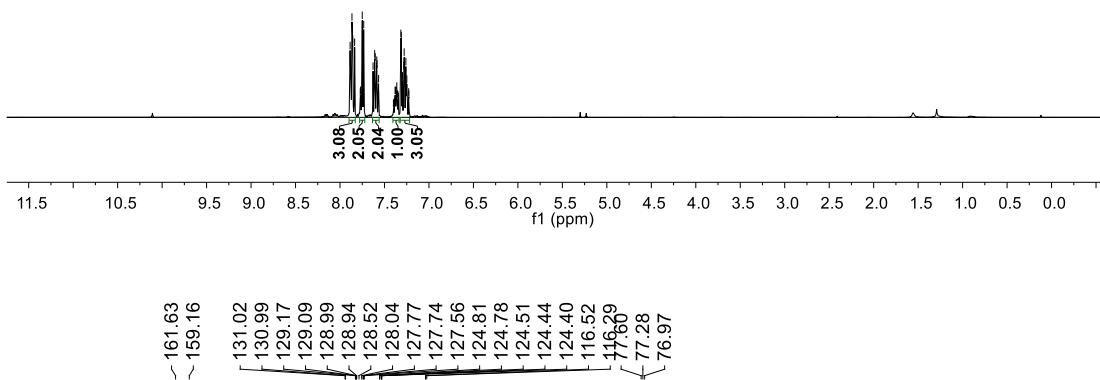
**12**

7.91
7.89
7.88
7.85
7.85
7.83
7.83
7.82
7.81
7.80
7.79
7.77
7.75
7.73
7.72
7.71
7.70
7.69
7.68
7.67
7.66
7.65
7.64
7.63
7.62
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7.45
7.26

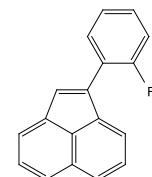
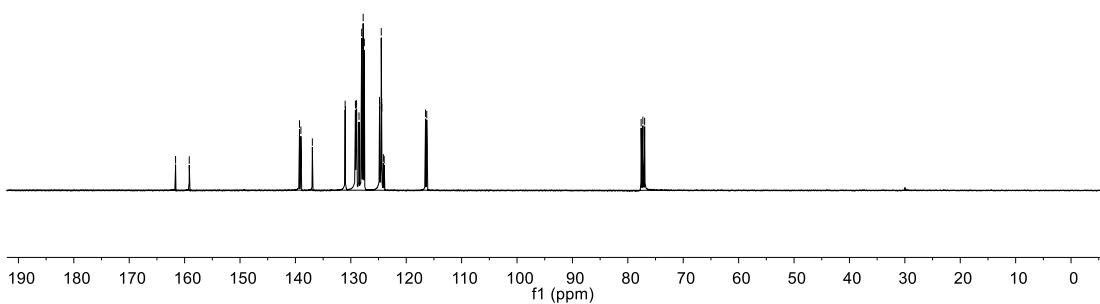
1H- NMR, 400 MHz, CDCl<sub>3</sub>

139.37
138.79
138.46
138.27
134.26
132.63
130.36
129.78
128.51
128.18
128.04
127.58
127.53
125.43
124.40

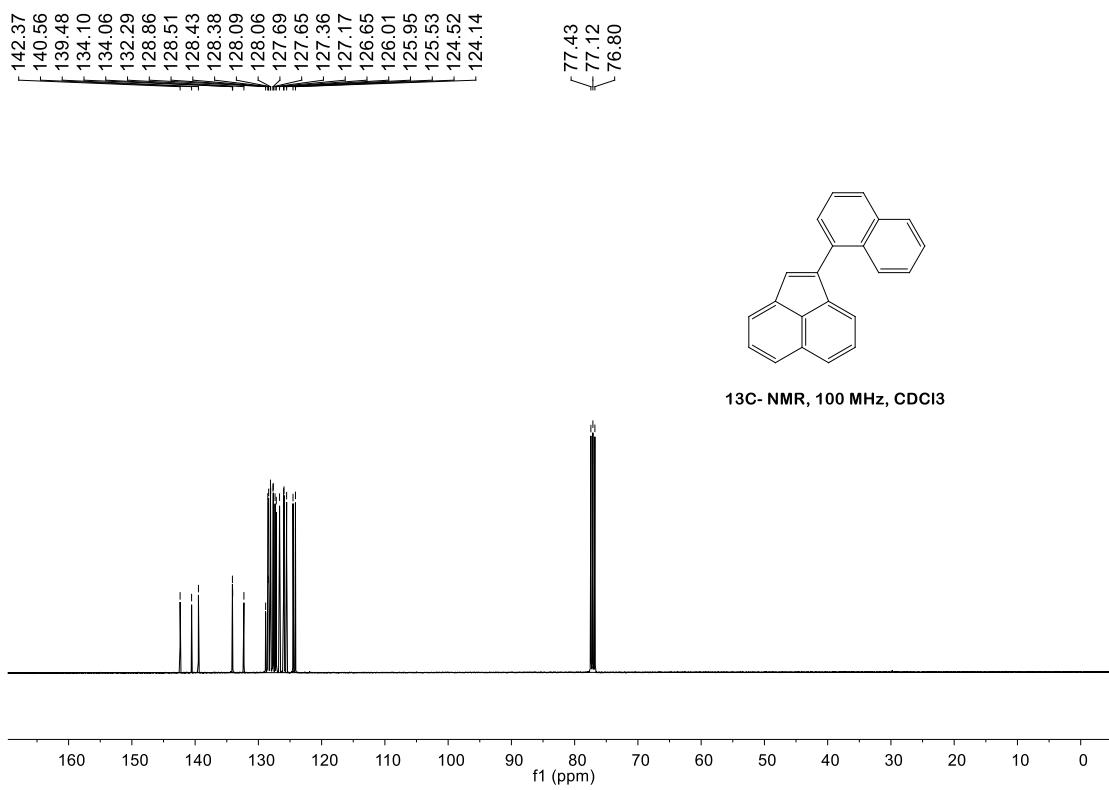
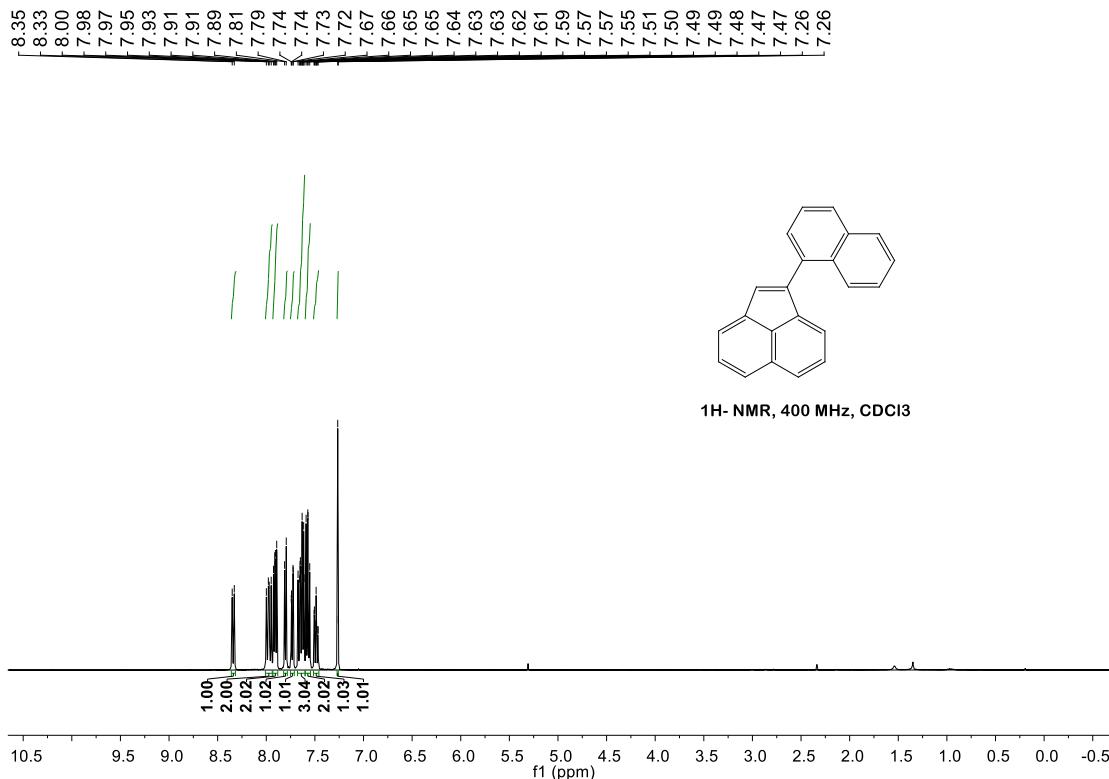
13C- NMR, 100 MHz, CDCl<sub>3</sub>

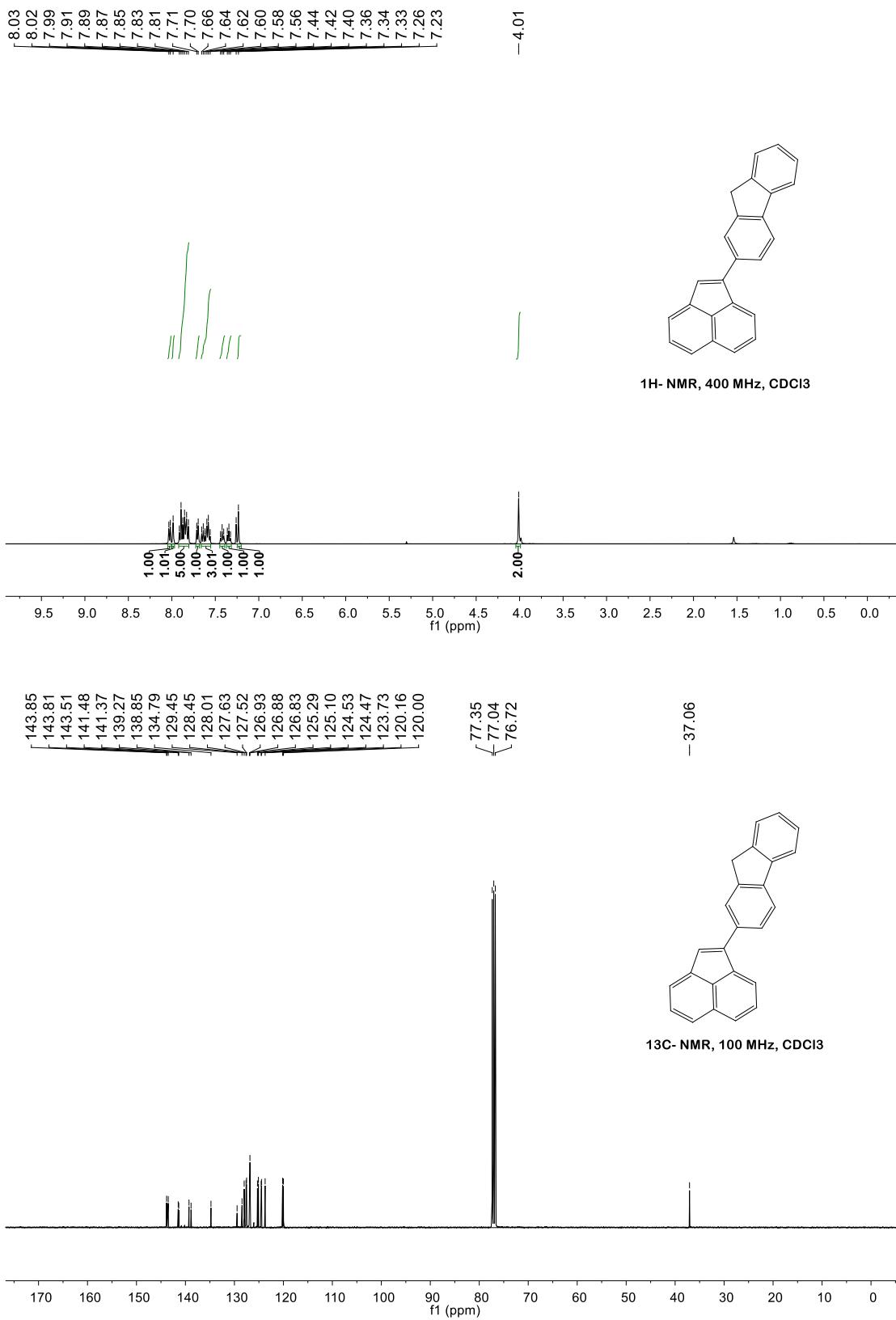
**13**<sup>1</sup>H- NMR, 400 MHz, CDCl<sub>3</sub>

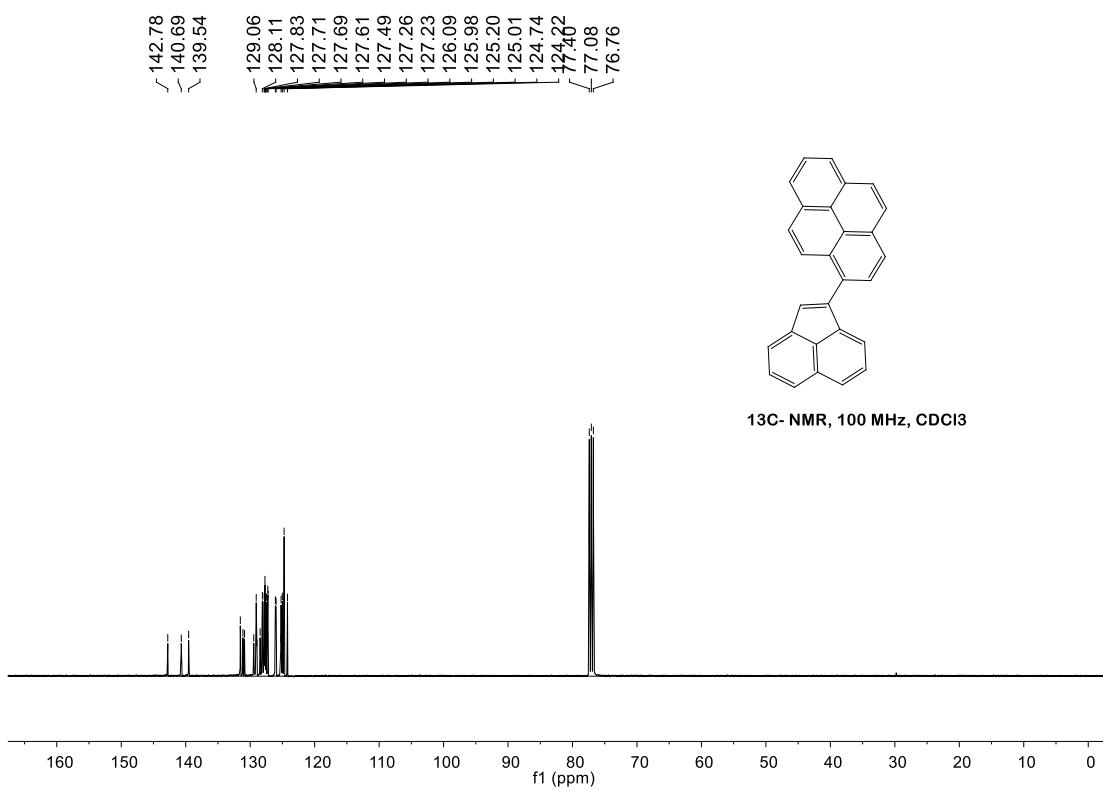
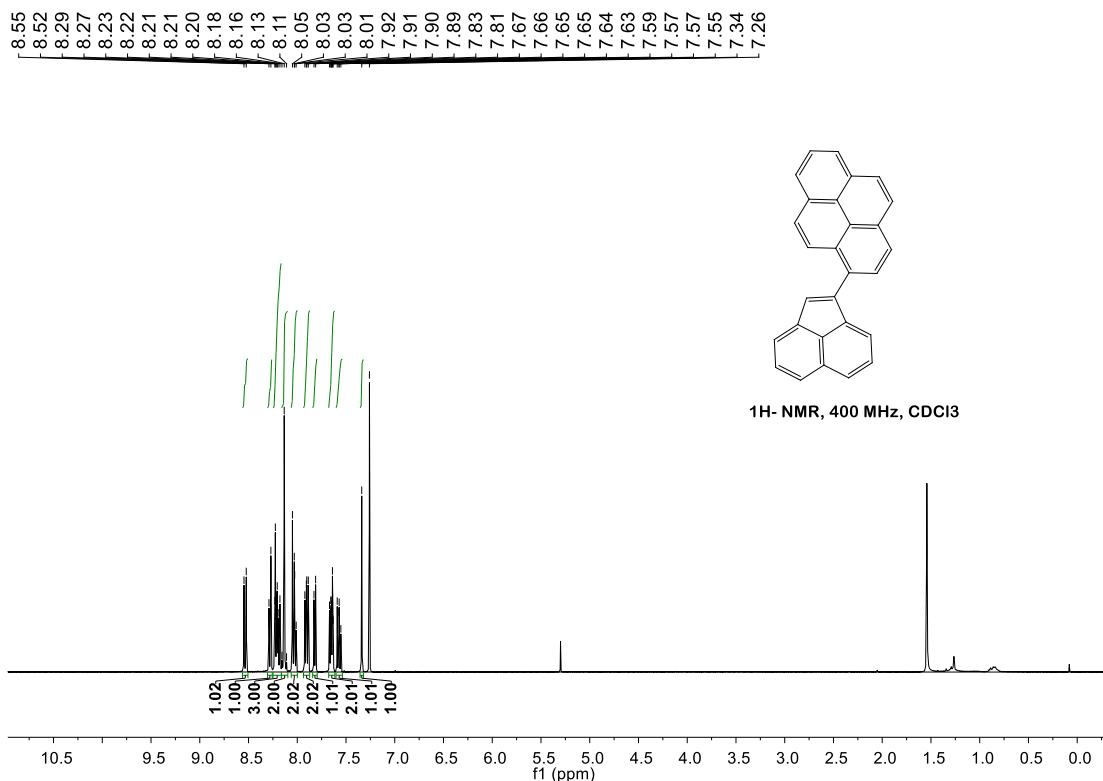
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~159.16  
131.02  
130.99  
129.17  
129.09  
128.99  
128.94  
128.52  
128.04  
127.77  
127.74  
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124.78  
124.51  
124.44  
124.40  
116.52  
116.29  
77.28  
76.97

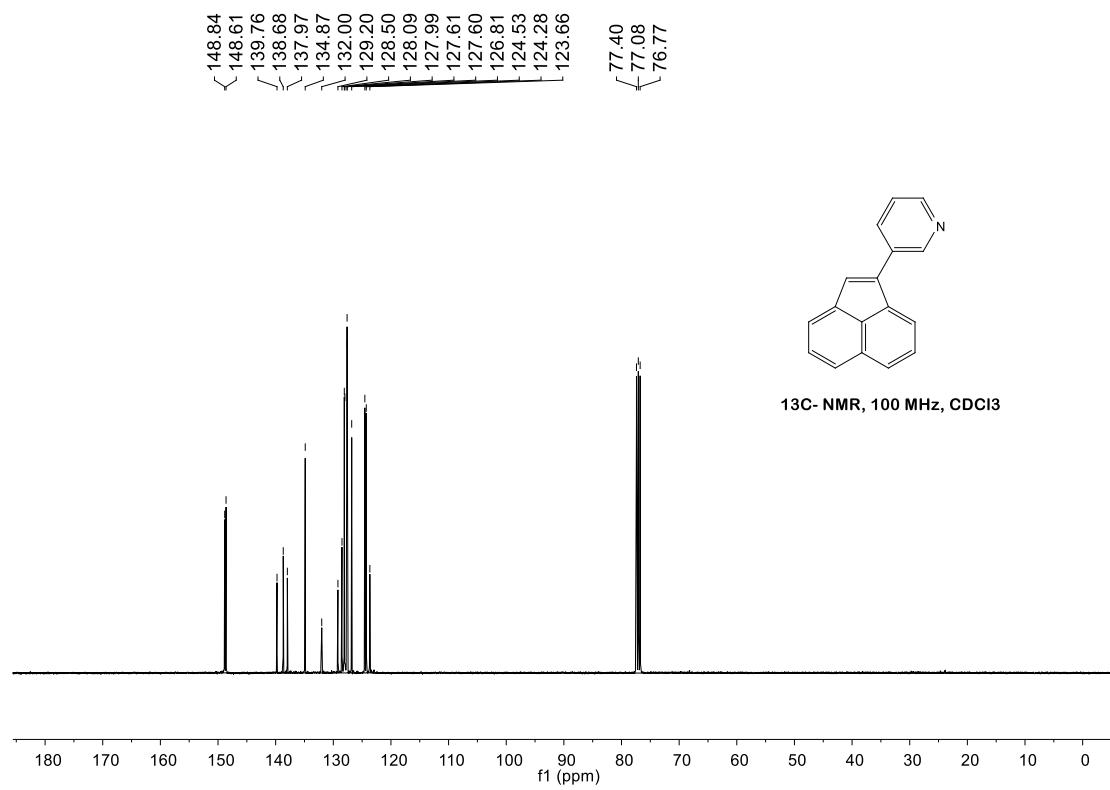
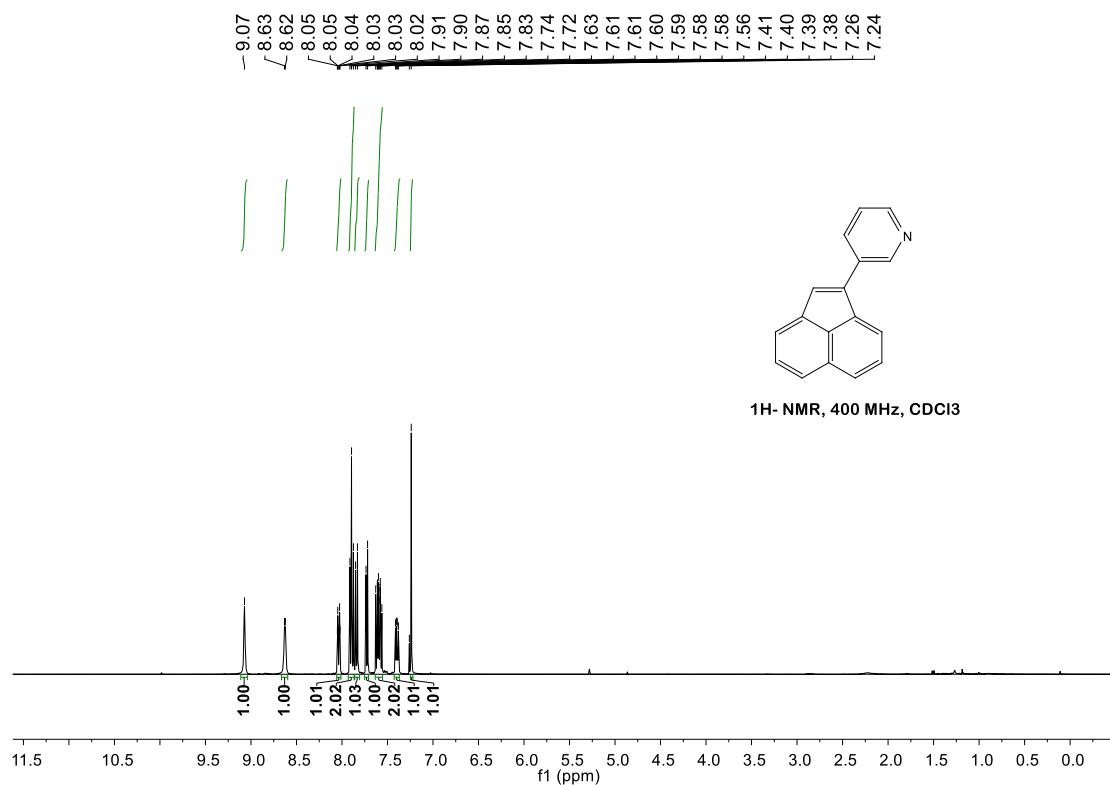
<sup>13</sup>C- NMR, 100 MHz, CDCl<sub>3</sub>

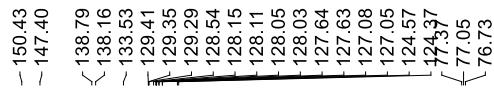
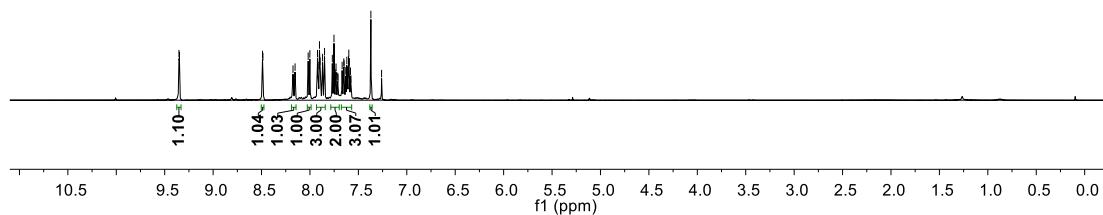
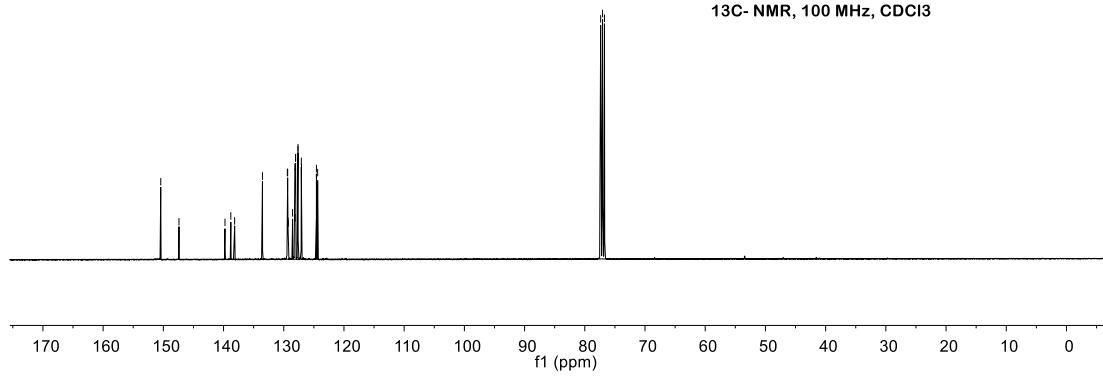
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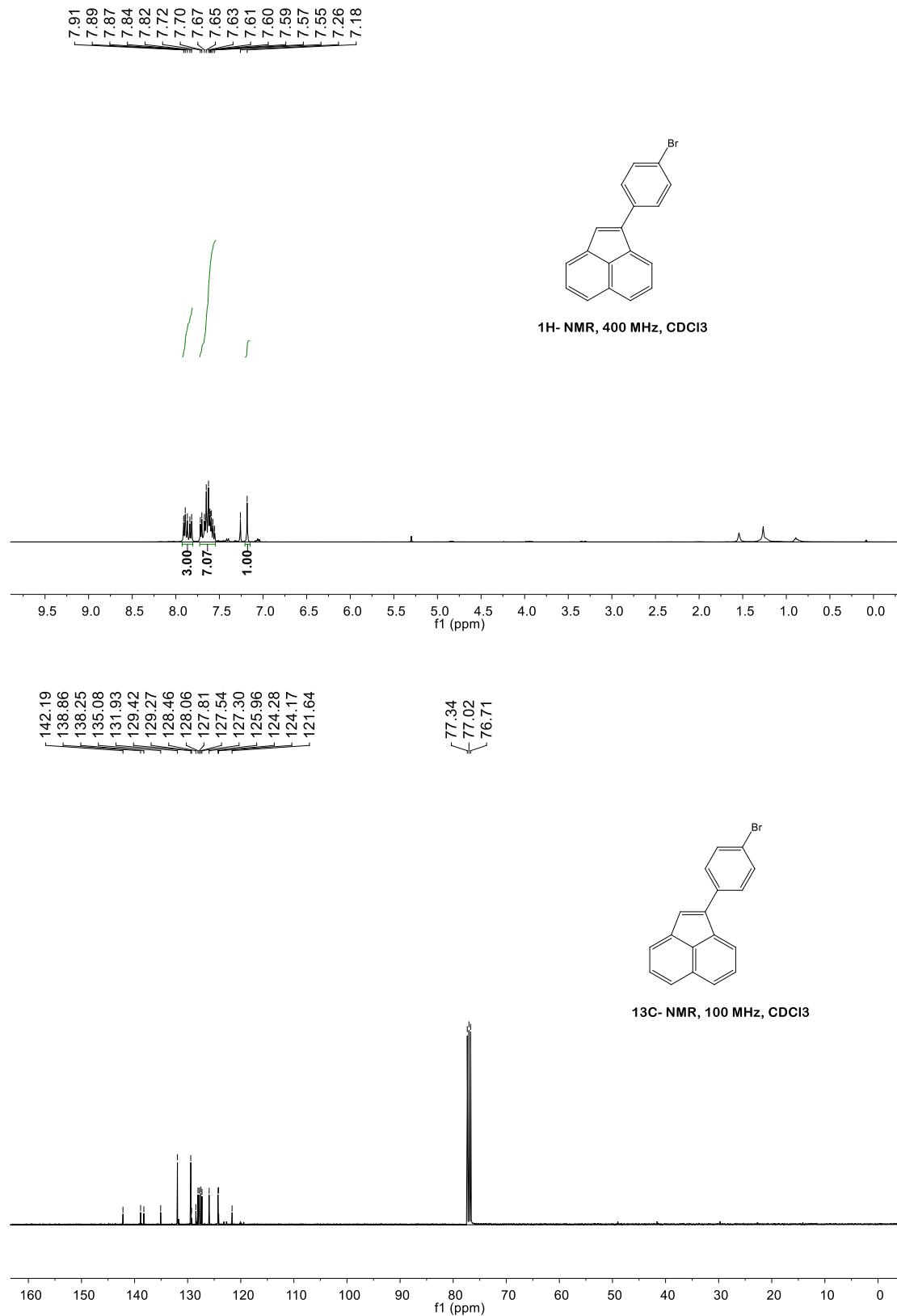
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**15**

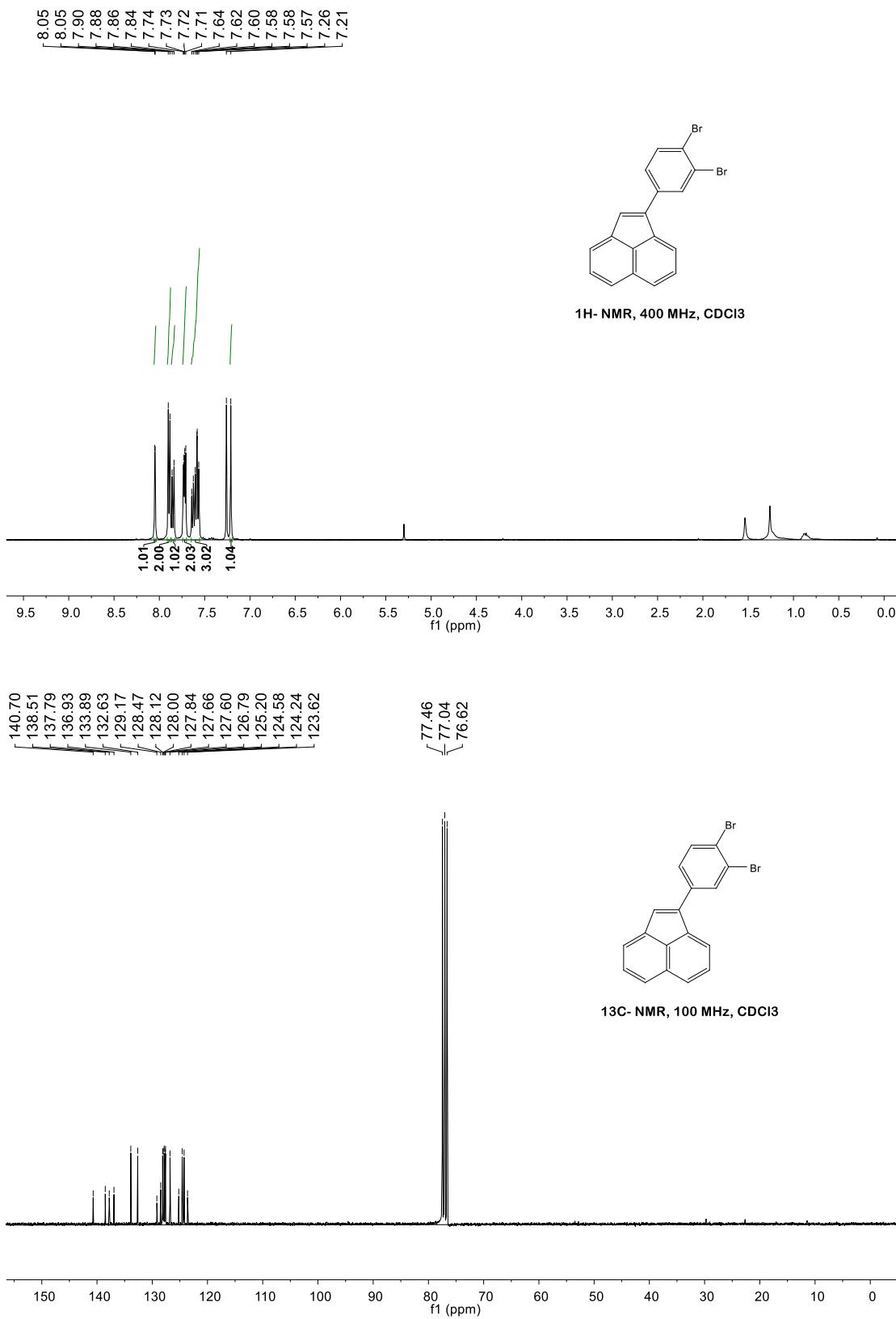
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**17**

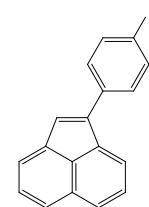
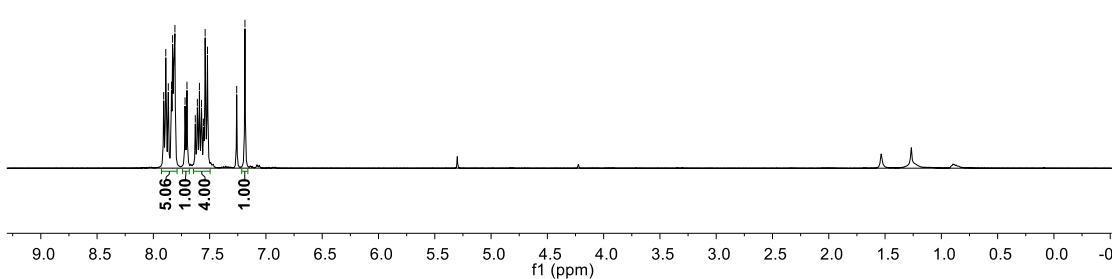
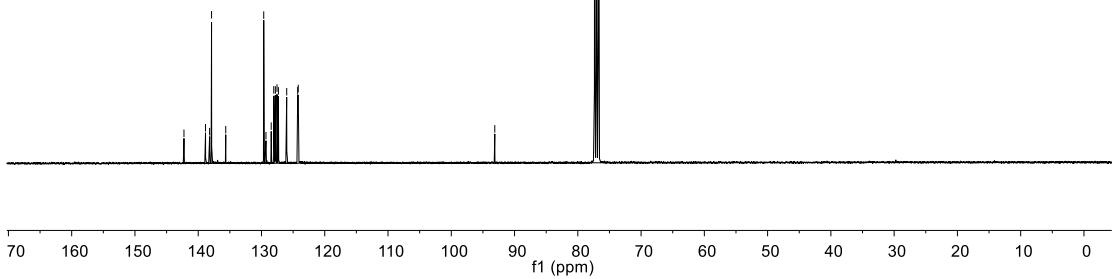
**18**<sup>1</sup>H- NMR, 400 MHz, CDCl<sub>3</sub><sup>13</sup>C- NMR, 100 MHz, CDCl<sub>3</sub>

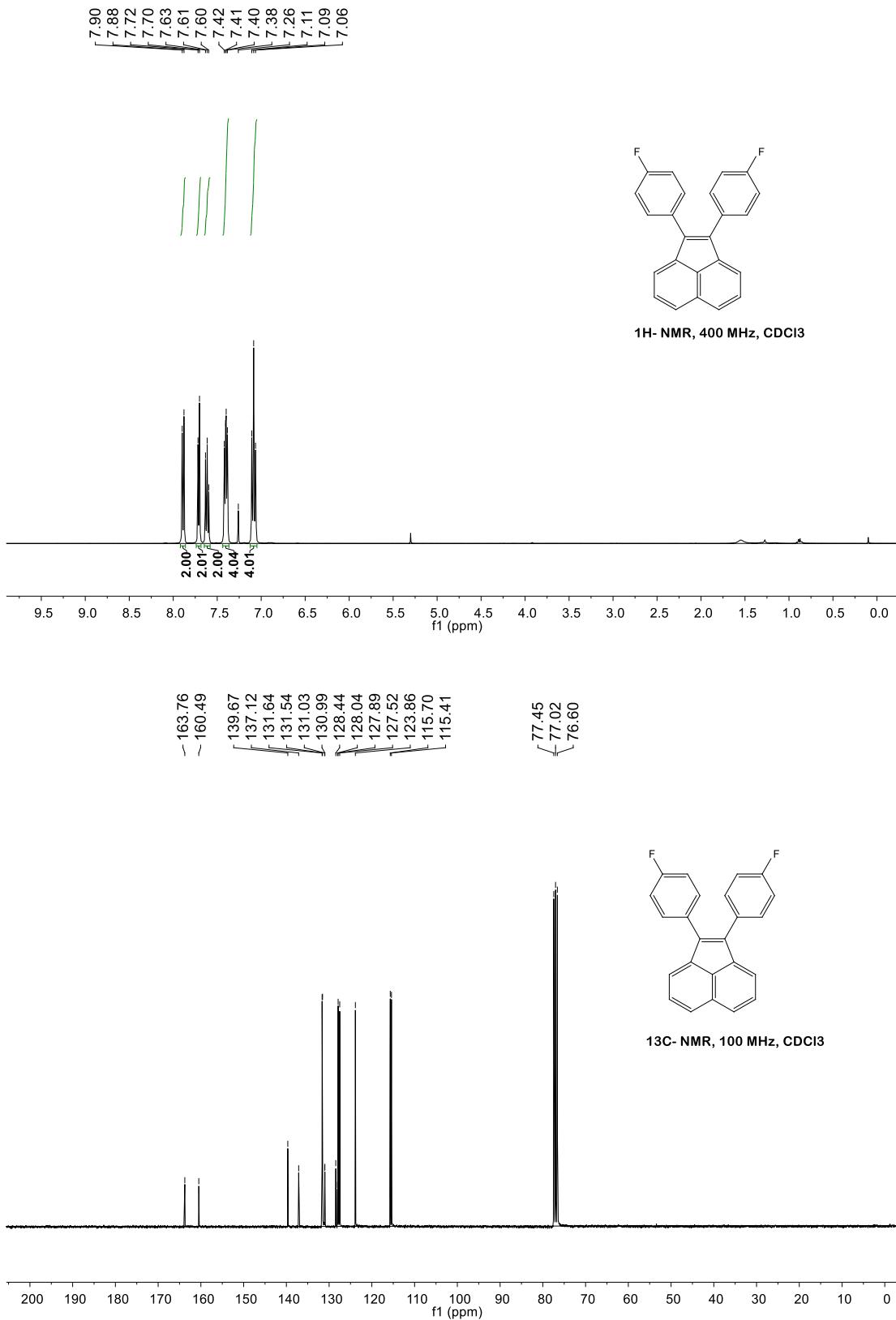
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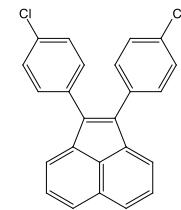
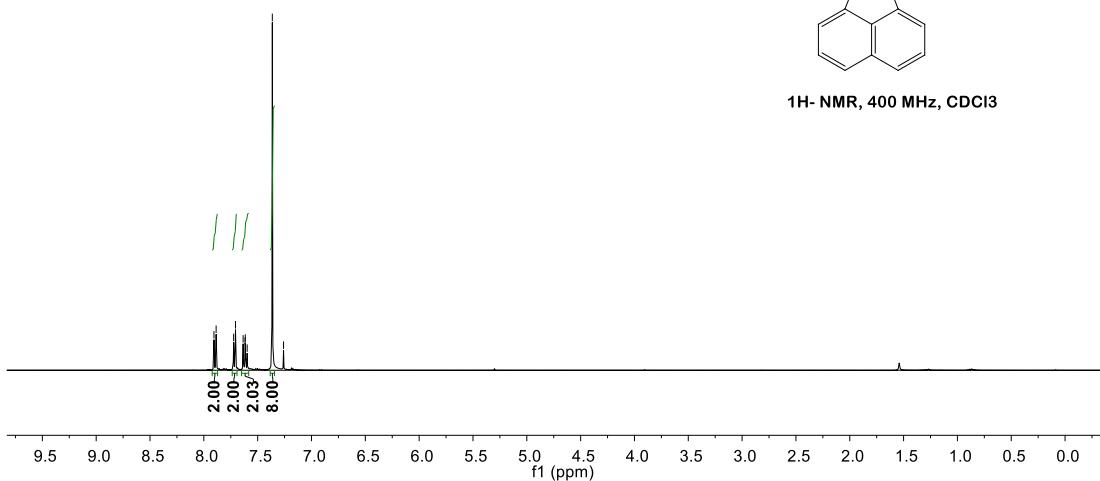


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1H- NMR, 400 MHz, CDCl<sub>3</sub>13C- NMR, 100 MHz, CDCl<sub>3</sub>

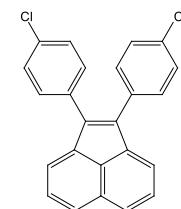
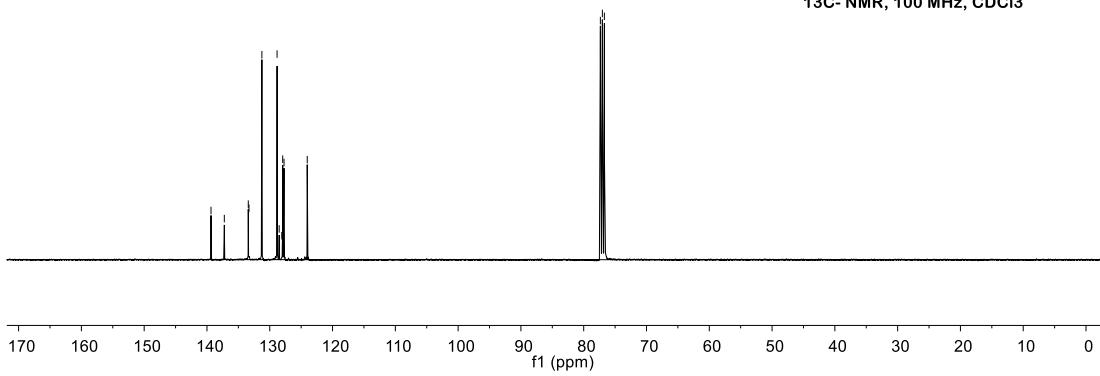


7.91  
7.89  
7.72  
7.71  
7.64  
7.62  
7.62  
7.60  
7.36  
7.26

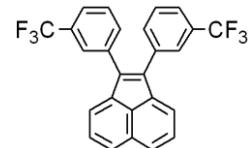
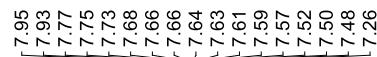
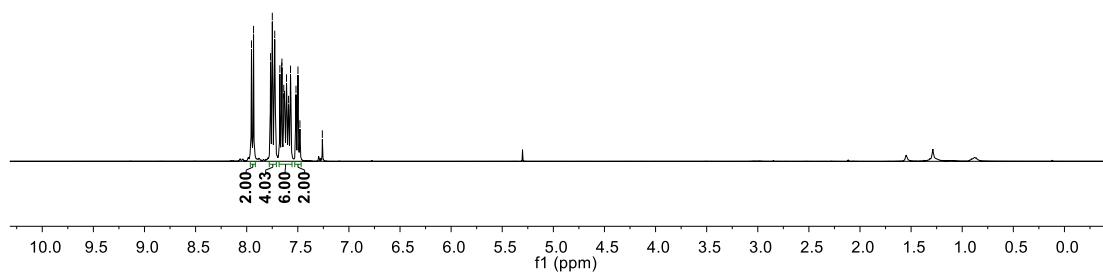
1H-NMR, 400 MHz, CDCl<sub>3</sub>

139.35  
137.24  
133.42  
133.31  
131.25  
128.83  
128.50  
128.08  
127.94  
127.73  
124.03

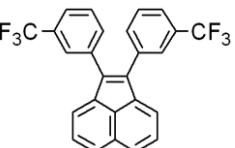
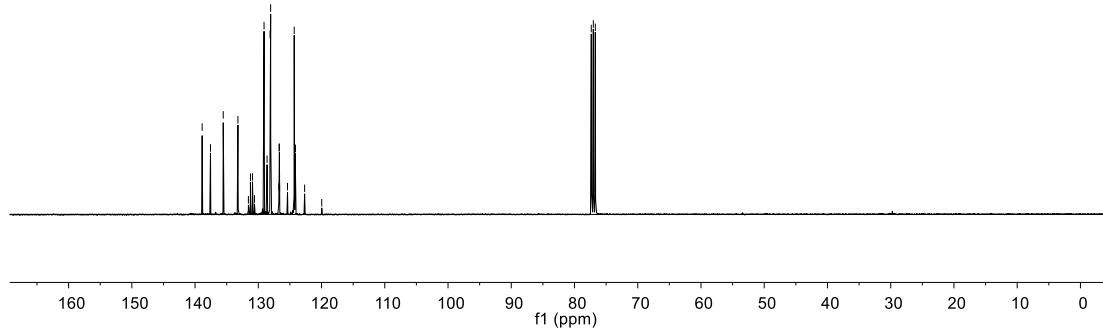
77.34  
77.03  
76.71

13C-NMR, 100 MHz, CDCl<sub>3</sub>

24

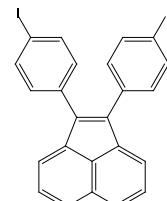
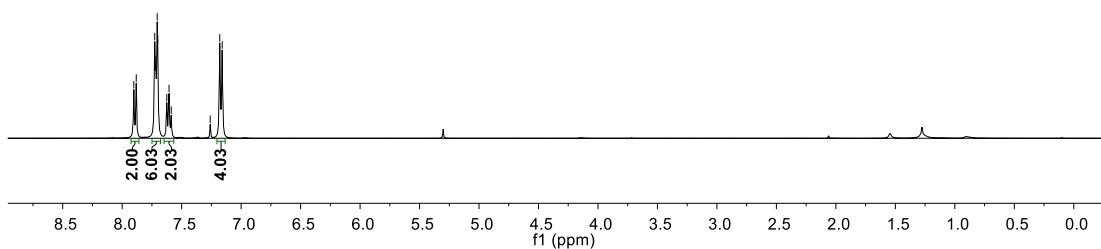
 $^1\text{H}$ -NMR, 400 MHz,  $\text{CDCl}_3$ 

138.88
137.56
135.55
133.22
131.57
131.24
130.92
130.60
129.09
128.61
128.14
128.05
126.75
126.71
126.67
126.63
125.38
124.32
124.17
124.13
122.68
119.97

 $^{13}\text{C}$ -NMR, 100 MHz,  $\text{CDCl}_3$ 

25

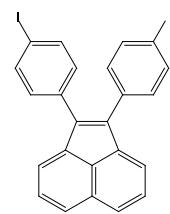
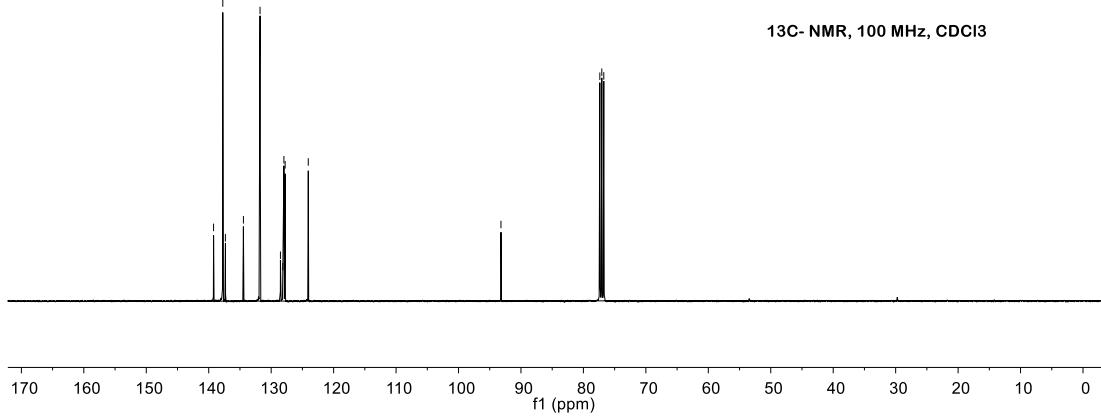
7.90  
7.88  
7.73  
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7.71  
7.70  
7.62  
7.61  
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7.26  
7.18  
7.16

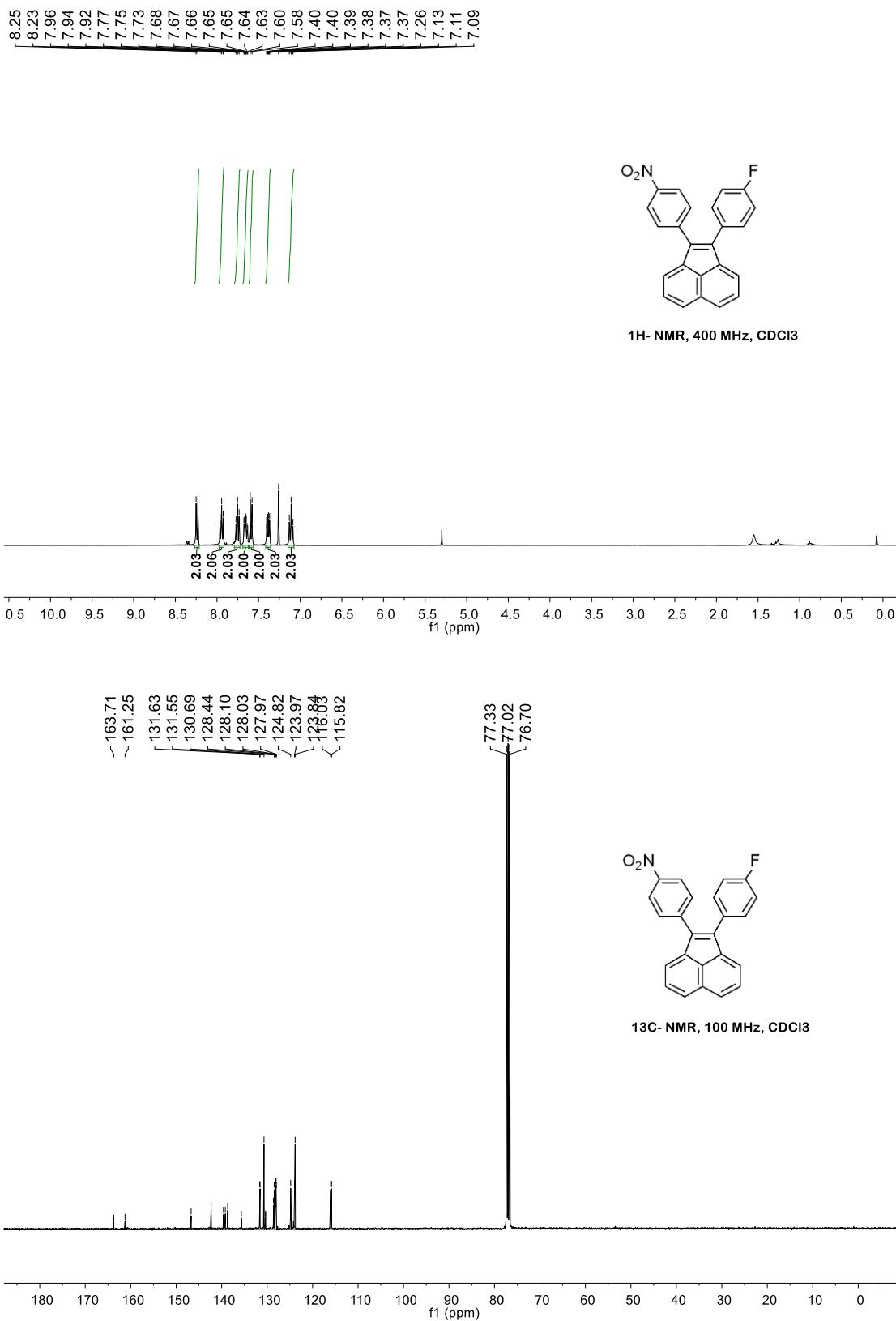
1H- NMR, 400 MHz, CDCl<sub>3</sub>

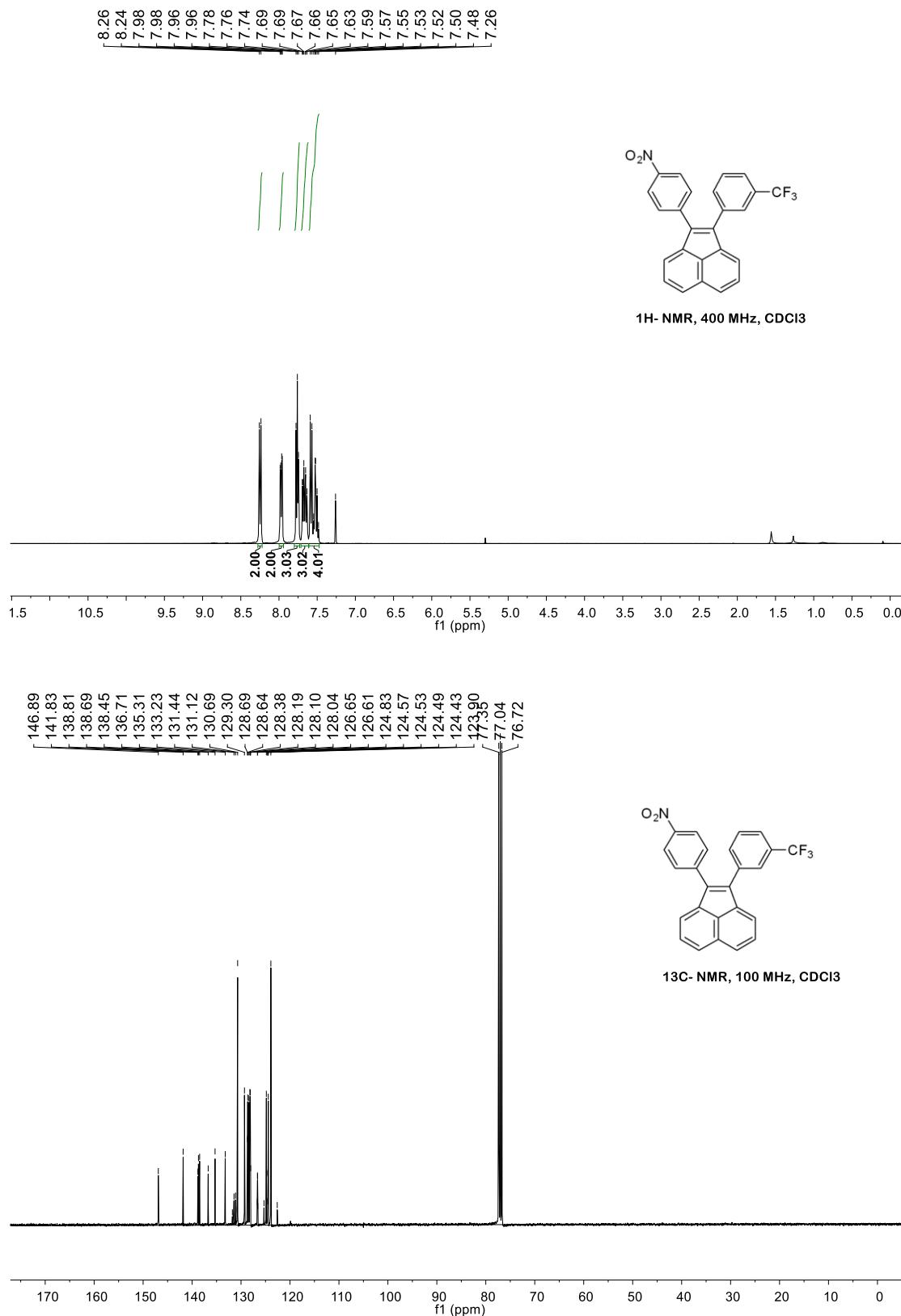
139.22  
137.75  
137.33  
134.44  
131.78  
128.52  
128.13  
127.96  
127.76  
124.06

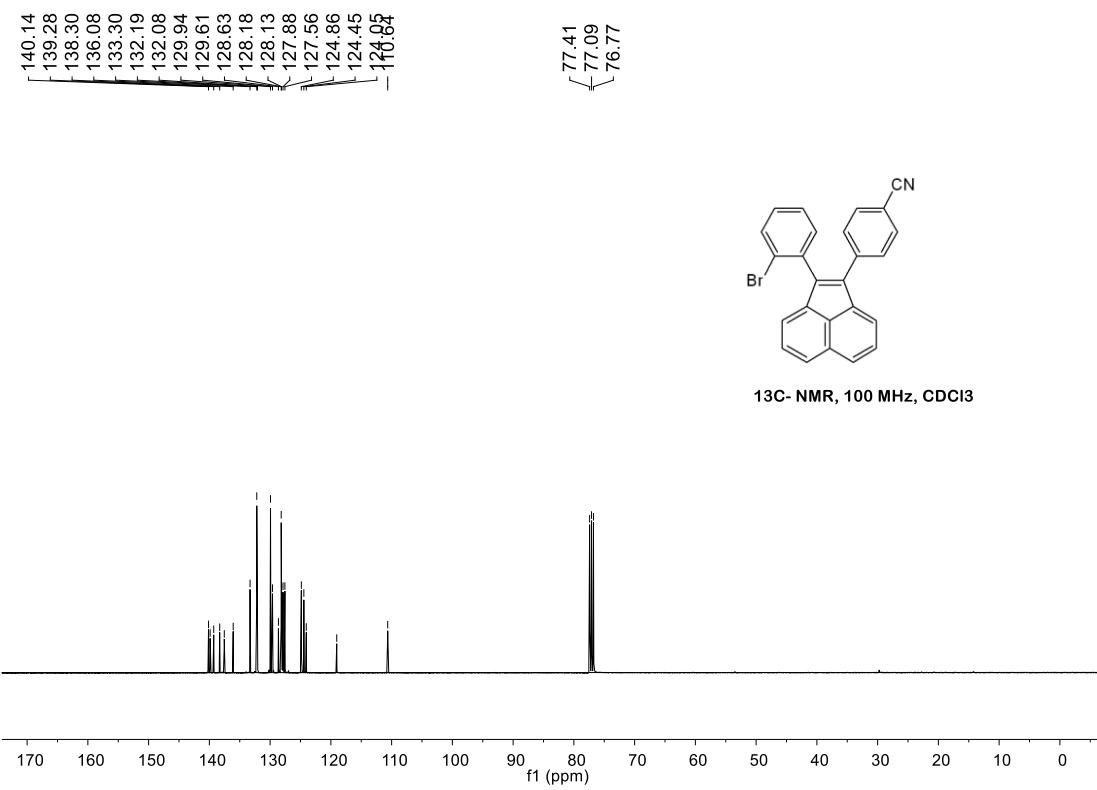
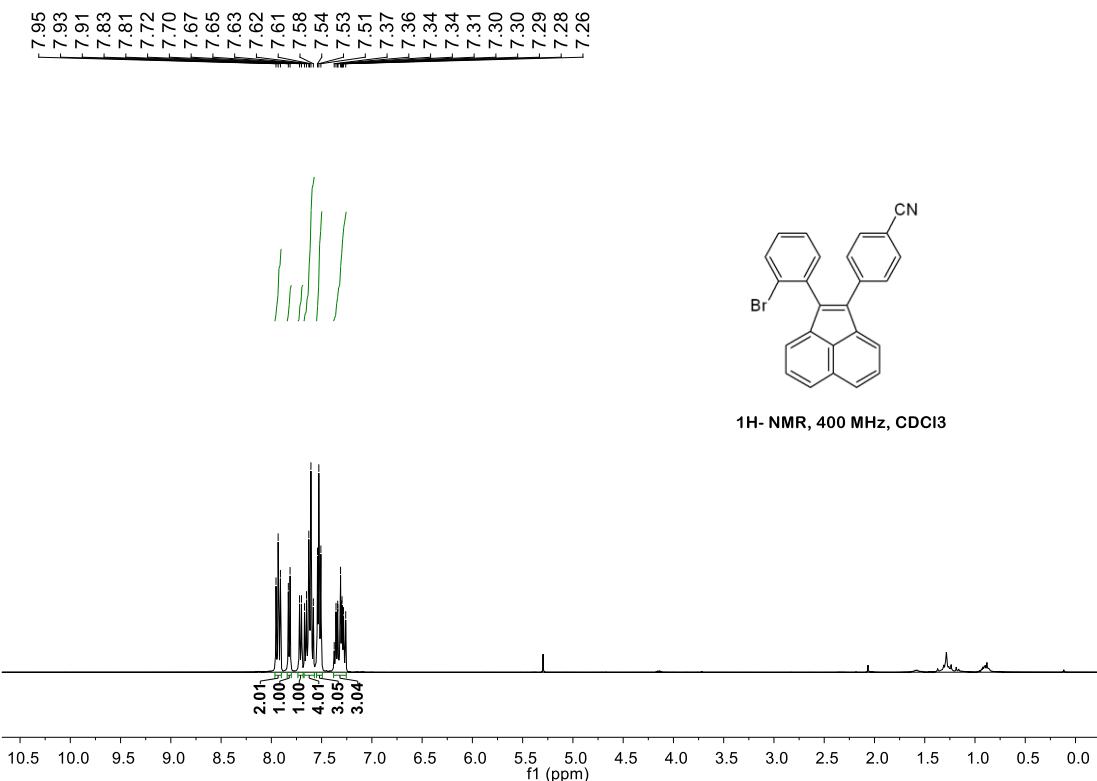
-93.20

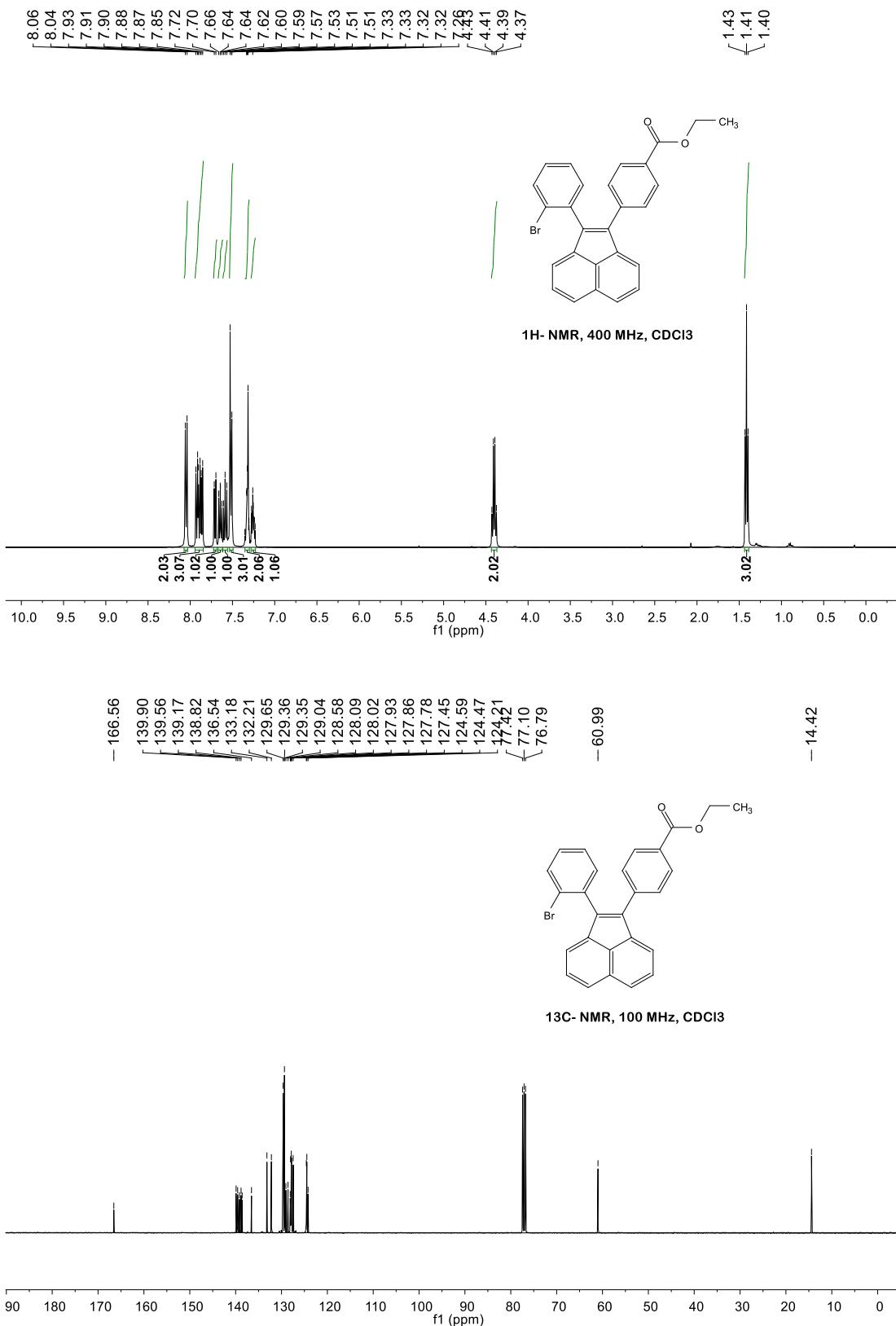
77.37  
77.06  
76.74

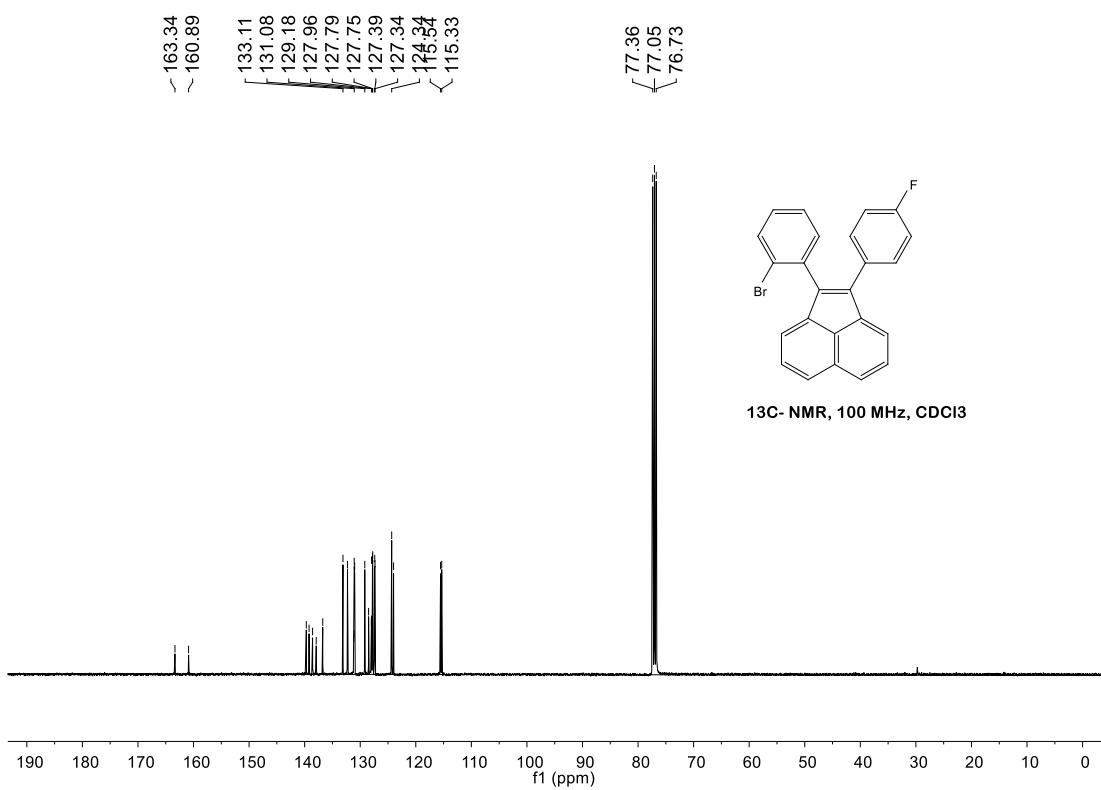
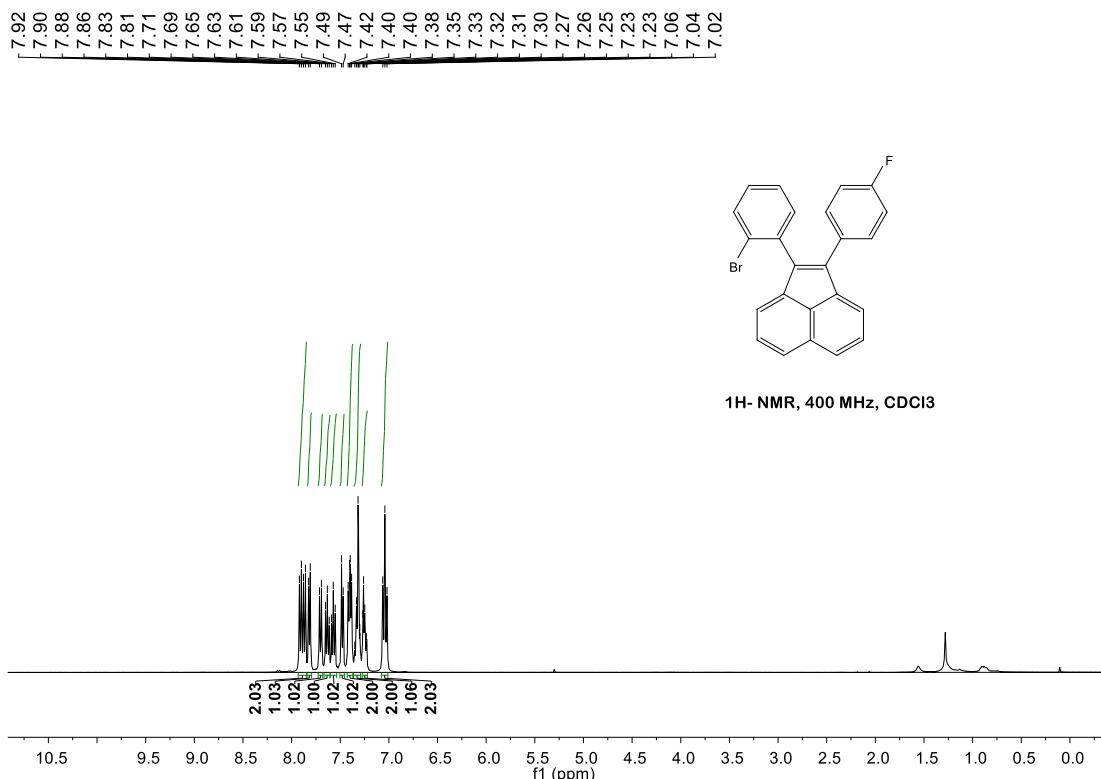
13C- NMR, 100 MHz, CDCl<sub>3</sub>

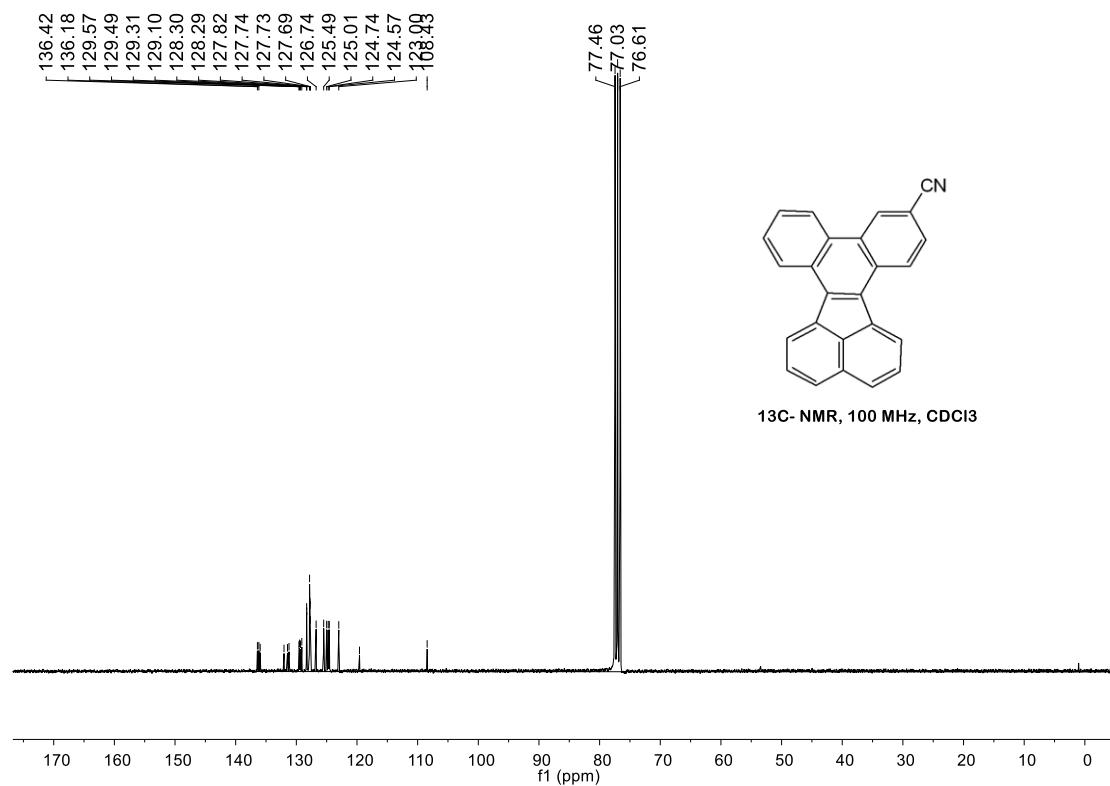
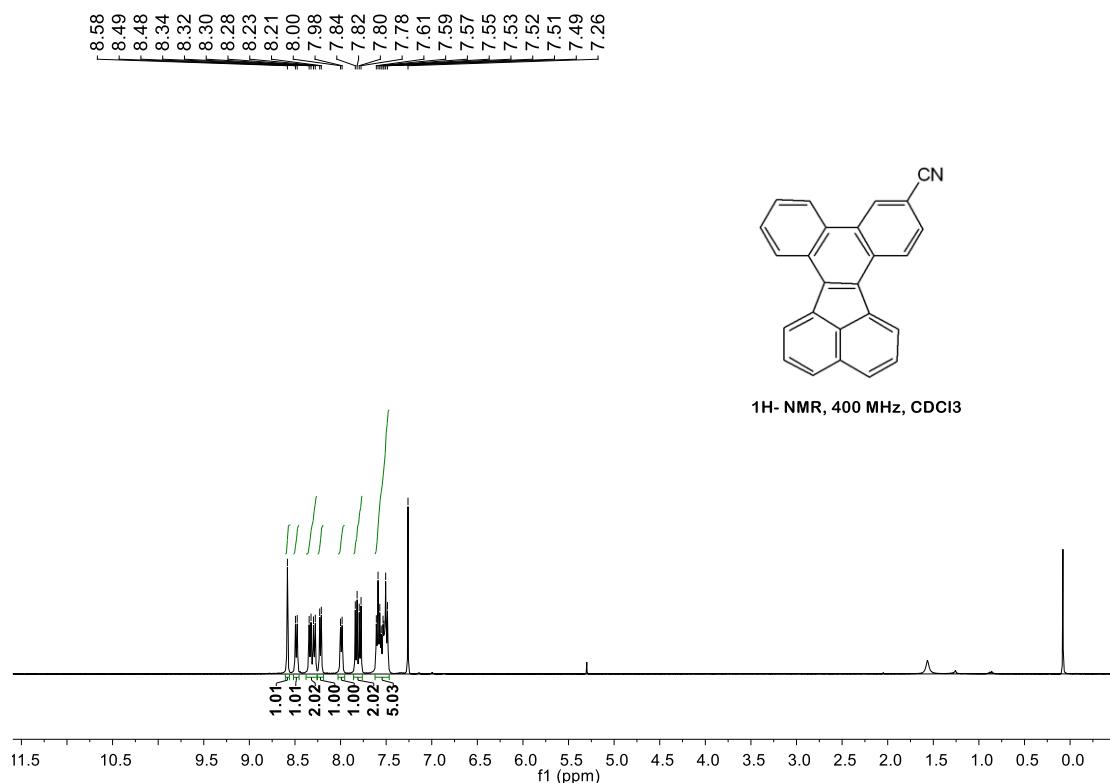




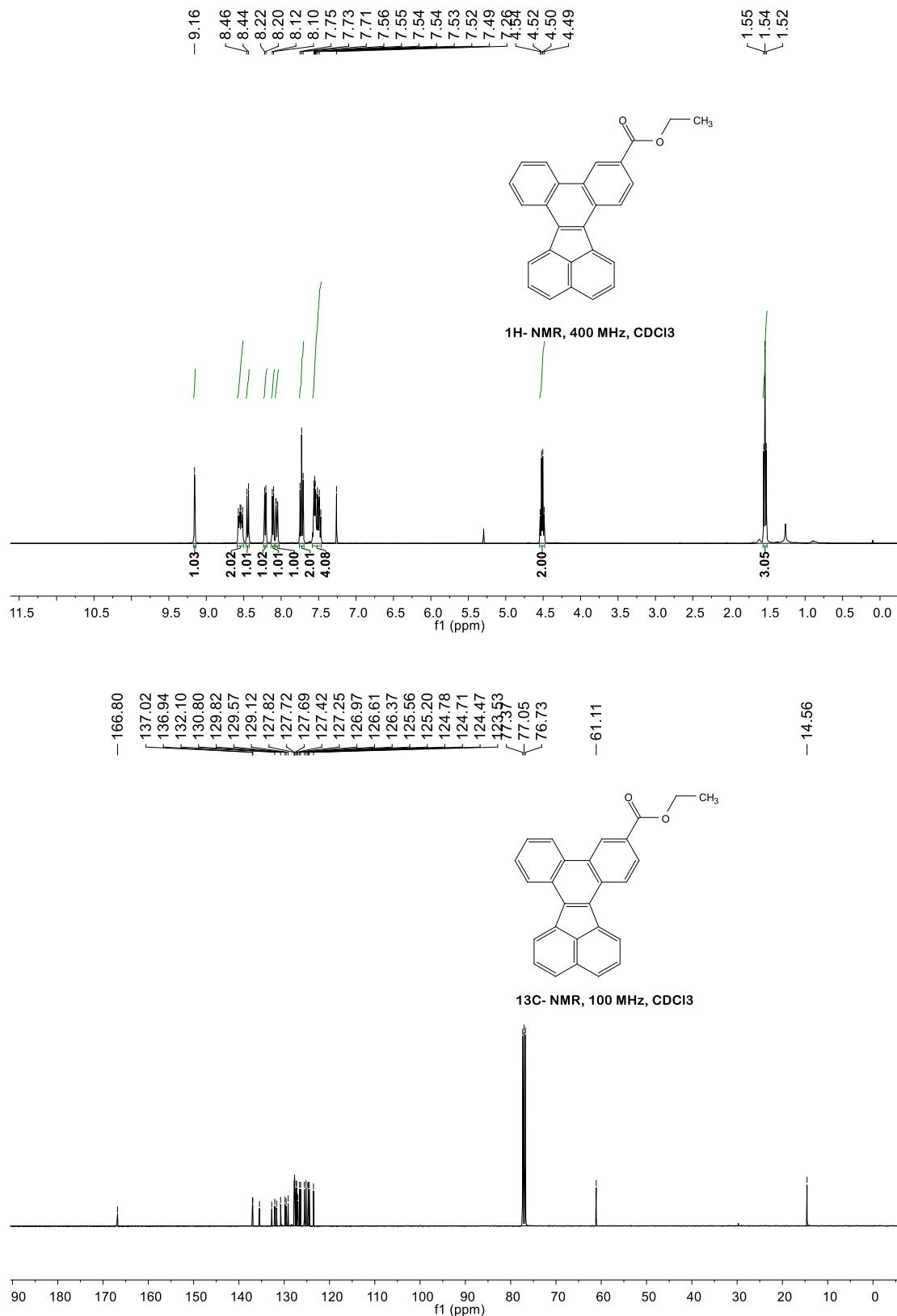
**28**

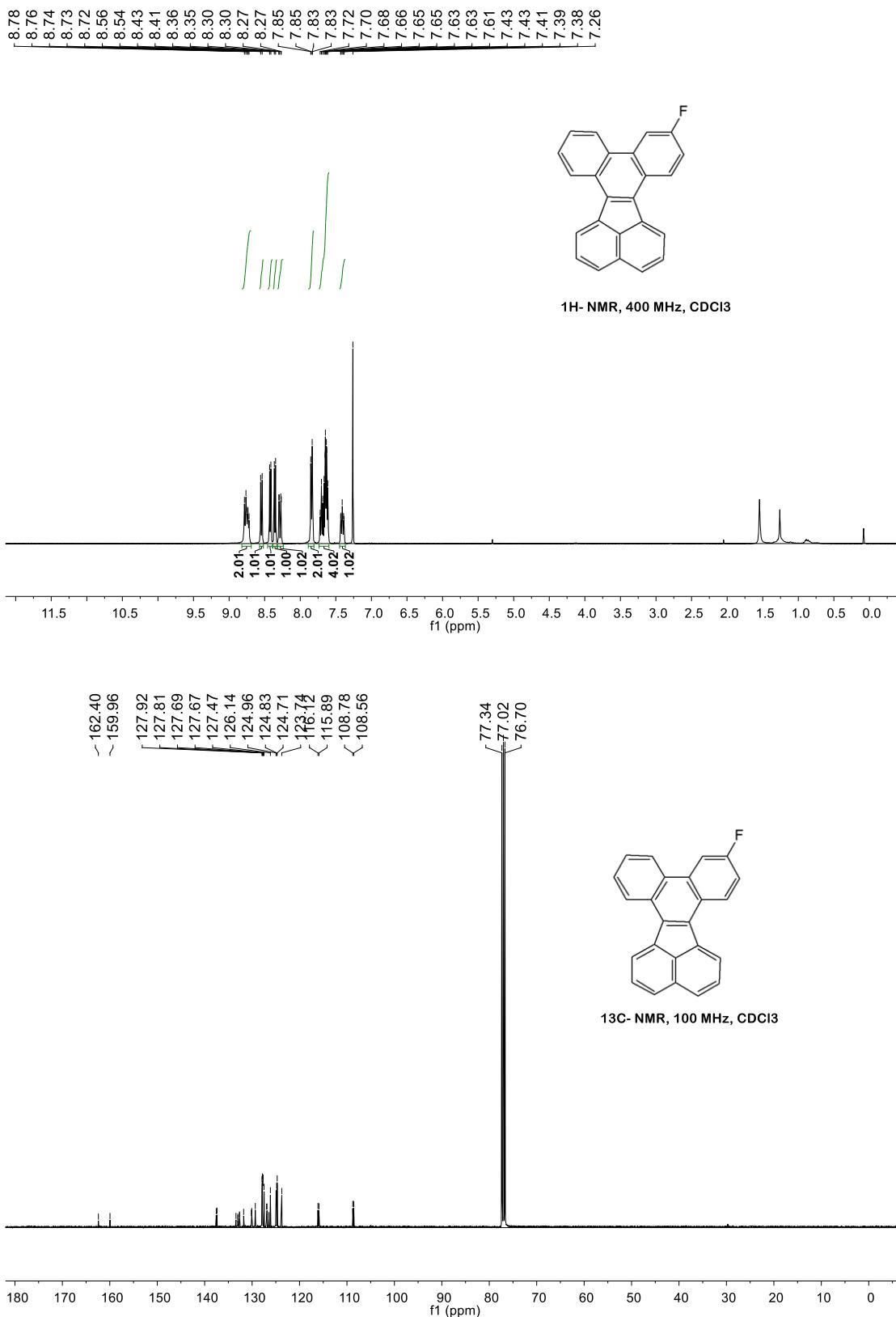
**29**

**30**

**31**

32



**33**

X-ray analysis of **23** (CDC 1553883) and **33** (CDC 1553881)

### Crystal structure report of compound 23:

#### CDC 1553883

##### X-ray crystallographic study



(C<sub>24</sub>H<sub>14</sub>Cl<sub>2</sub>);  $M = 373.25$ . APEXII, Bruker-AXS diffractometer, Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ),  $T = 150 \text{ K}$ ; monoclinic C 2/c (I.T.#15),  $a = 16.8237(12)$ ,  $b = 11.3330(7)$ ,  $c = 10.2994(8) \text{ \AA}$ ,  $\beta = 114.382(3)^\circ$ ,  $V = 1788.6(2) \text{ \AA}^3$ ,  $Z = 4$ ,  $d = 1.386 \text{ g.cm}^{-3}$ ,  $\mu = 0.367 \text{ mm}^{-1}$ . The structure was solved by dual-space algorithm using the *SHELXT* program [1], and then refined with full-matrix least-square methods based on  $F^2$  (*SHELXL*) [2]. All non-hydrogen atoms were refined with anisotropic atomic displacement parameters. H atoms were finally included in their calculated positions. A final refinement on  $F^2$  with 2048 unique intensities and 119 parameters converged at  $\omega R(F^2) = 0.0828$  ( $R(F) = 0.0331$ ) for 1756 observed reflections with  $I > 2\sigma(I)$ .

[1] G. M. Sheldrick, Acta Cryst. A71 (2015) 3-8

[2] Sheldrick G.M., Acta Cryst. C71 (2015) 3-8

##### Structural data

Empirical formula	C <sub>24</sub> H <sub>14</sub> Cl <sub>2</sub>
Formula weight	373.25
Temperature	150 K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, C 2/c
Unit cell dimensions	$a = 16.8237(12) \text{ \AA}, \alpha = 90^\circ$ $b = 11.3330(7) \text{ \AA}, \beta = 114.382(3)^\circ$ $c = 10.2994(8) \text{ \AA}, \gamma = 90^\circ$
Volume	1788.6(2) Å <sup>3</sup>
Z, Calculated density	4, 1.386 (g.cm <sup>-3</sup> )
Absorption coefficient	0.367 mm <sup>-1</sup>
$F(000)$	768
Crystal size	0.500 x 0.370 x 0.220 mm
Crystal color	orange
Theta range for data collection	3.479 to 27.476 °
$h_{\min}, h_{\max}$	-21, 21
$k_{\min}, k_{\max}$	-14, 14
$l_{\min}, l_{\max}$	-9, 13
Reflections collected / unique	4736 / 2048 [ $R(\text{int})^a = 0.0210$ ]
Reflections [ $I > 2\sigma$ ]	1756
Completeness to theta_max	0.995
Absorption correction type	multi-scan
Max. and min. transmission	0.916, 0.838
Refinement method	Full-matrix least-squares on $F^2$
Data / restraints / parameters	2048 / 0 / 119
<sup>b</sup> S (Goodness-of-fit)	1.039
Final $R$ indices [ $I > 2\sigma$ ]	$R1^c = 0.0331$ , $wR2^d = 0.0828$
$R$ indices (all data)	$R1^c = 0.0404$ , $wR2^d = 0.0889$

Largest diff. peak and hole      0.253 and -0.244 e·Å<sup>-3</sup>

$$^aR_{int} = \sum |F_o|^2 - <|F_o|^2> / \sum |F_o|^2$$

$$^bS = \{\sum [w(F_o^2 - F_c^2)^2] / (n - p)\}^{1/2}$$

$$^cR1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$^dWR2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$$

$$w = 1 / [\sigma(F_o^2) + aP^2 + bP] \text{ where } P = [2F_c^2 + \text{MAX}(F_o^2, 0)] / 3$$

**Atomic coordinates, site occupancy (%) and equivalent isotropic displacement parameters (Å<sup>2</sup>). U(eq) is defined as one third of the trace of the orthogonalized U<sub>ij</sub> tensor.**

Atom	x	y	z	occ.	U(eq)
C1	0.32828(8)	0.16558(11)	0.44358(15)	1	0.0203(3)
C2	0.34445(9)	0.26123(12)	0.37459(15)	1	0.0215(3)
H2	0.326399	0.260500	0.274177	1	0.026
C3	0.38762(9)	0.35836(12)	0.45508(14)	1	0.0201(3)
H3	0.398624	0.424790	0.408671	1	0.024
C4	0.41517(8)	0.36029(11)	0.60289(14)	1	0.0181(3)
C5	0.39573(8)	0.26281(12)	0.66821(14)	1	0.0201(3)
H5	0.412517	0.263489	0.768315	1	0.024
C6	0.35247(9)	0.16547(12)	0.58942(15)	1	0.0211(3)
H6	0.339618	0.099715	0.634671	1	0.025
C7	0.46354(9)	0.46293(11)	0.68616(14)	1	0.0182(3)
C8	0.43886(9)	0.58702(12)	0.64277(14)	1	0.0194(3)
C9	0.37326(9)	0.64179(13)	0.53047(15)	1	0.0240(3)
H9	0.329807	0.596977	0.457607	1	0.029
C10	0.500000	0.65790(17)	0.750000	1	0.0195(4)
C11	0.500000	0.78079(17)	0.750000	1	0.0229(4)
C12	0.56703(10)	0.83550(13)	0.86939(17)	1	0.0274(3)
H12	0.569725	0.919084	0.876953	1	0.033
C13	0.62801(10)	0.76743(13)	0.97401(16)	1	0.0281(3)
H13	0.672676	0.805752	1.052227	1	0.034
Cl1	0.27588(2)	0.04204(3)	0.34336(4)	1	0.02943(13)

#### Anisotropic displacement parameters (Å<sup>2</sup>)

The anisotropic displacement factor exponent takes the form: -2π<sup>2</sup> [ h<sup>2</sup> a<sup>\*2</sup> U<sub>11</sub> + ... + 2 h k a<sup>\*</sup> b<sup>\*</sup> U<sub>12</sub> ].

Atom	U11	U22	U33	U23	U13	U12
C1	0.0182(6)	0.0167(6)	0.0244(7)	-0.0038(5)	0.0070(5)	0.0003(5)
C2	0.0218(7)	0.0232(7)	0.0189(6)	-0.0003(5)	0.0078(5)	0.0014(5)
C3	0.0217(7)	0.0200(6)	0.0198(6)	0.0021(5)	0.0097(5)	-0.0002(5)
C4	0.0175(6)	0.0170(6)	0.0203(6)	-0.0003(5)	0.0083(5)	0.0013(5)
C5	0.0212(7)	0.0209(6)	0.0190(6)	0.0010(5)	0.0090(5)	0.0010(5)
C6	0.0210(7)	0.0172(6)	0.0263(7)	0.0030(5)	0.0108(6)	0.0013(5)
C7	0.0212(7)	0.0176(6)	0.0183(6)	0.0009(5)	0.0105(5)	0.0003(5)
C8	0.0220(7)	0.0190(6)	0.0196(6)	-0.0001(5)	0.0111(5)	-0.0003(5)
C9	0.0261(7)	0.0231(7)	0.0217(7)	0.0014(6)	0.0086(6)	0.0011(6)
C10	0.0206(9)	0.0196(9)	0.0204(9)	0.000	0.0105(7)	0.000
C11	0.0279(10)	0.0186(9)	0.0266(10)	0.000	0.0155(8)	0.000
C12	0.0349(8)	0.0177(6)	0.0333(8)	-0.0040(6)	0.0177(7)	-0.0038(6)
C13	0.0322(8)	0.0248(7)	0.0260(7)	-0.0077(6)	0.0109(6)	-0.0080(6)
Cl1	0.0323(2)	0.01927(19)	0.0297(2)	-0.00531(14)	0.00581(16)	-0.00290(13)

#### Bond lengths [Å]

C1 - C2	= 1.383(2)
C1 - C6	= 1.385(2)
C1 - Cl1	= 1.7468(14)
C2 - C3	= 1.3890(19)
C2 - H2	= 0.9500
C3 - C4	= 1.3966(19)
C3 - H3	= 0.9500
C4 - C5	= 1.4005(19)
C4 - C7	= 1.4747(18)
C5 - C6	= 1.3848(19)
C5 - H5	= 0.9500
C6 - H6	= 0.9500
C7 - C7_#1	= 1.379(3)
C7 - C8	= 1.4822(18)
C8 - C9	= 1.3737(19)
C8 - C10	= 1.4089(17)
C9 - C13_#1	= 1.425(2)
C9 - H9	= 0.9500

C10 - C11 = 1.393(3)  
 C11 - C12 = 1.4224(17)  
 C11 - C12\_#1 = 1.4224(17)  
 C12 - C13 = 1.377(2)  
 C12 - H12 = 0.9500  
 C13 - H13 = 0.9500

Symmetry transformations used to generate equivalent atoms:

#1 -x, y, -z+1/2      T = [1, 0, 1]

**Angles [°]**

C2 - C1 - C6 = 121.73(12)  
 C2 - C1 - Cl1 = 118.87(11)  
 C6 - C1 - Cl1 = 119.40(10)  
 C1 - C2 - C3 = 118.66(13)  
 C1 - C2 - H2 = 120.70  
 C3 - C2 - H2 = 120.70  
 C2 - C3 - C4 = 121.29(13)  
 C2 - C3 - H3 = 119.40  
 C4 - C3 - H3 = 119.40  
 C3 - C4 - C5 = 118.23(12)  
 C3 - C4 - C7 = 120.09(12)  
 C5 - C4 - C7 = 121.67(12)  
 C6 - C5 - C4 = 121.16(13)  
 C6 - C5 - H5 = 119.40  
 C4 - C5 - H5 = 119.40  
 C1 - C6 - C5 = 118.88(12)  
 C1 - C6 - H6 = 120.60  
 C5 - C6 - H6 = 120.60  
 C7\_#1 - C7 - C4 = 127.92(7)  
 C7\_#1 - C7 - C8 = 108.40(7)  
 C4 - C7 - C8 = 123.67(11)  
 C9 - C8 - C10 = 118.37(13)  
 C9 - C8 - C7 = 135.25(13)  
 C10 - C8 - C7 = 106.36(12)  
 C8 - C9 - C13\_#1 = 118.36(14)  
 C8 - C9 - H9 = 120.80  
 C13\_#1 - C9 - H9 = 120.80  
 C11 - C10 - C8\_#1 = 124.76(8)  
 C11 - C10 - C8 = 124.76(8)  
 C8\_#1 - C10 - C8 = 110.48(17)  
 C10 - C11 - C12 = 115.84(9)  
 C10 - C11 - C12\_#1 = 115.84(9)  
 C12 - C11 - C12\_#1 = 128.32(18)  
 C13 - C12 - C11 = 120.06(14)  
 C13 - C12 - H12 = 120.00  
 C11 - C12 - H12 = 120.00  
 C12 - C13 - C9\_#1 = 122.57(14)  
 C12 - C13 - H13 = 118.70  
 C9\_#1 - C13 - H13 = 118.70

Symmetry transformations used to generate equivalent atoms:

#1 -x, y, -z+1/2      T = [1, 0, 1]

**Torsion angles [°]**

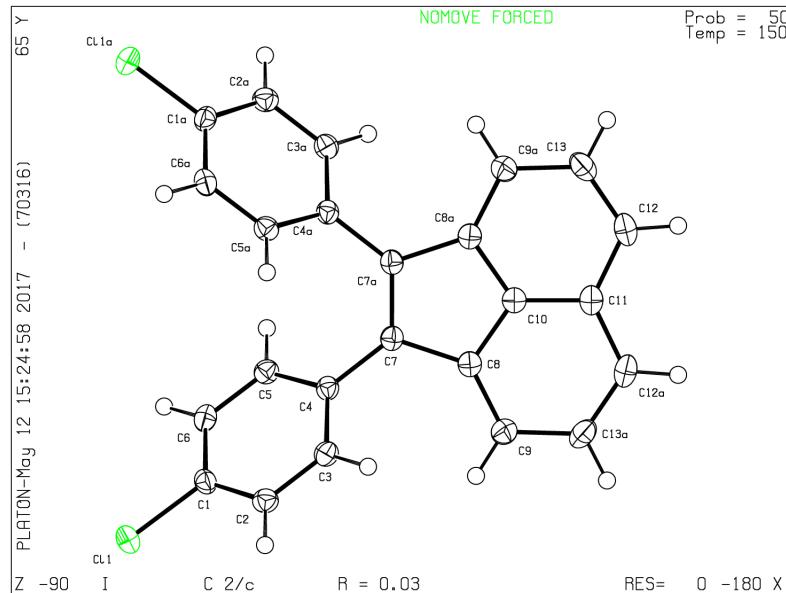
C6 - C1 - C2 - C3 = 1.20(2)  
 Cl1 - C1 - C2 - C3 = -178.78(10)  
 C1 - C2 - C3 - C4 = 0.50(2)  
 C2 - C3 - C4 - C5 = -2.00(2)  
 C2 - C3 - C4 - C7 = 177.88(12)  
 C3 - C4 - C5 - C6 = 1.70(2)  
 C7 - C4 - C5 - C6 = -178.11(12)  
 C2 - C1 - C6 - C5 = -1.50(2)  
 Cl1 - C1 - C6 - C5 = 178.55(10)  
 C4 - C5 - C6 - C1 = -0.10(2)  
 C3 - C4 - C7 - C7\_#1 = -136.55(18)  
 C5 - C4 - C7 - C7\_#1 = 43.30(2)  
 C3 - C4 - C7 - C8 = 44.05(19)  
 C5 - C4 - C7 - C8 = -136.10(14)  
 C7\_#1 - C7 - C8 - C9 = -178.68(16)  
 C4 - C7 - C8 - C9 = 0.80(2)  
 C7\_#1 - C7 - C8 - C10 = -0.10(17)  
 C4 - C7 - C8 - C10 = 179.40(11)  
 C10 - C8 - C9 - C13\_#1 = 1.91(19)  
 C7 - C8 - C9 - C13\_#1 = -179.65(15)

C9 - C8 - C10 - C11 = -1.10(14)  
 C7 - C8 - C10 - C11 = -179.96(6)  
 C9 - C8 - C10 - C8\_#1 = 178.90(14)  
 C7 - C8 - C10 - C8\_#1 = 0.04(6)  
 C8\_#1 - C10 - C11 - C12 = -0.68(10)  
 C8 - C10 - C11 - C12 = 179.32(10)  
 C8\_#1 - C10 - C11 - C12\_#1 = 179.32(10)  
 C8 - C10 - C11 - C12\_#1 = -0.68(10)  
 C10 - C11 - C12 - C13 = 1.59(16)  
 C12\_#1 - C11 - C12 - C13 = -178.41(16)  
 C11 - C12 - C13 - C9\_#1 = -0.80(2)

Symmetry transformations used to generate equivalent atoms:

#1 -x, y, -z+1/2      T=[ 1, 0, 1]

#### Structure visualisation



#### Crystal structure report of compound 33:

CDC 1553881

X-ray crystallographic study



(C<sub>24</sub>H<sub>13</sub>F);  $M = 320.34$ . D8 VENTURE Bruker AXS diffractometer [\*], Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ),  $T = 150 \text{ K}$ ; monoclinic  $P 2_1/n$  (I.T.#14),  $a = 10.4684(7)$ ,  $b = 7.4280(4)$ ,  $c = 18.7773(11) \text{ \AA}$ ,  $\beta = 90.086(3)^\circ$ ,  $V = 1460.11(15) \text{ \AA}^3$ .  $Z = 4$ ,  $d = 1.457 \text{ g.cm}^{-3}$ ,  $\mu = 0.092 \text{ mm}^{-1}$ . The structure was solved by dual-space algorithm using the *SHELXT* program [1], and then refined with full-matrix least-square methods based on  $F^2$  (*SHELXL*) [2]. All non-hydrogen atoms were refined with anisotropic atomic displacement parameters. H atoms were finally included in their calculated positions. A final refinement on  $F^2$  with 3338 unique intensities and 230 parameters converged at  $\omega R(F^2) = 0.1592$  ( $R(F) = 0.0647$ ) for 2348 observed reflections with  $I > 2\sigma(I)$ .

[1] G. M. Sheldrick, Acta Cryst. A71 (2015) 3-8

[2] Sheldrick G.M., Acta Cryst. C71 (2015) 3-8

[\*] Thanks to FEDER funds

#### Structural data

Empirical formula	C <sub>24</sub> H <sub>13</sub> F
Formula weight	320.34
Temperature	150 K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, $P 2_1/n$
Unit cell dimensions	$a = 10.4684(7) \text{ \AA}, \alpha = 90^\circ$ $b = 7.4280(4) \text{ \AA}, \beta = 90.086(3)^\circ$ $c = 18.7773(11) \text{ \AA}, \gamma = 90^\circ$
Volume	1460.11(15) Å <sup>3</sup>
Z, Calculated density	4, 1.457 (g.cm <sup>-3</sup> )
Absorption coefficient	0.092 mm <sup>-1</sup>
$F(000)$	664
Crystal size	0.390 x 0.340 x 0.045 mm
Crystal color	yellow
Theta range for data collection	3.363 to 27.484 °
$h_{\min}, h_{\max}$	-13, 13
$k_{\min}, k_{\max}$	-9, 8
$l_{\min}, l_{\max}$	-24, 23
Reflections collected / unique	12886 / 3338 [ $R(\text{int})^a = 0.0673$ ]
Reflections [ $I > 2\sigma$ ]	2348
Completeness to theta_max	0.993
Absorption correction type	multi-scan
Max. and min. transmission	0.996, 0.723
Refinement method	Full-matrix least-squares on $F^2$
Data / restraints / parameters	3338 / 0 / 230
<sup>b</sup> S (Goodness-of-fit)	1.047
Final $R$ indices [ $I > 2\sigma$ ]	$R1^c = 0.0647$ , $wR2^d = 0.1592$
$R$ indices (all data)	$R1^c = 0.0962$ , $wR2^d = 0.1789$
Largest diff. peak and hole	0.623 and -0.563 e.Å <sup>-3</sup>

$$^a R_{\text{int}} = \sum |F_o^2 - \langle F_o^2 \rangle| / \sum [F_o^2]$$

$$^b S = \{\sum [w(F_o^2 - F_c^2)^2] / (n - p)\}^{1/2}$$

$$^c R1 = \sum ||F_o - |F_c|| / \sum |F_o|$$

$$^d wR2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$$

$$w = 1 / [a(F_o^2) + aP^2 + bP] \text{ where } P = [2F_c^2 + \text{MAX}(F_o^2, 0)] / 3$$

Atomic coordinates, site occupancy (%) and equivalent isotropic displacement parameters (Å<sup>2</sup>). U(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

Atom	x	y	z	occ.	U(eq)
C1	0.3973(2)	0.1902(3)	0.46823(12)	1	0.0247(5)
C2	0.2835(2)	0.1289(3)	0.43615(13)	1	0.0332(6)
H2	0.284957	0.087315	0.388312	1	0.040
C3A	0.1710(3)	0.1286(4)	0.47323(14)	0.595(3)	0.0407(7)
H3A	0.095246	0.087019	0.450655	0.595(3)	0.049
C3B	0.1710(3)	0.1286(4)	0.47323(14)	0.405(3)	0.0407(7)
F4	0.0652(4)	0.0799(7)	0.4415(2)	0.405(3)	0.0558(7)
C5	0.1656(3)	0.1882(4)	0.54322(14)	1	0.0412(7)
H5	0.087139	0.185925	0.568525	1	0.049
C6	0.2748(2)	0.2505(3)	0.57564(13)	1	0.0315(6)
H6	0.270664	0.292199	0.623411	1	0.038
C7	0.3928(2)	0.2539(3)	0.53953(11)	1	0.0235(5)
C8	0.5084(2)	0.3177(3)	0.57184(11)	1	0.0217(5)
C9	0.5305(2)	0.3956(3)	0.64362(11)	1	0.0230(5)
C10	0.4587(2)	0.4345(3)	0.70307(12)	1	0.0297(5)

H10	0.370590	0.404410	0.704526	1	0.036
C11	0.5178(3)	0.5199(4)	0.76206(12)	1	0.0338(6)
H11	0.467190	0.546592	0.802667	1	0.041
C12	0.6443(3)	0.5656(3)	0.76312(12)	1	0.0313(6)
H12	0.679862	0.624784	0.803392	1	0.038
C13	0.7219(2)	0.5239(3)	0.70356(12)	1	0.0270(5)
C14	0.8546(3)	0.5588(3)	0.69658(13)	1	0.0346(6)
H14	0.899922	0.617417	0.733839	1	0.041
C15	0.9168(3)	0.5076(4)	0.63598(14)	1	0.0408(7)
H15	1.006252	0.527921	0.632738	1	0.049
C16	0.8531(2)	0.4252(4)	0.57756(14)	1	0.0359(6)
H16	0.899438	0.391923	0.536135	1	0.043
C17	0.7238(2)	0.3941(3)	0.58151(12)	1	0.0248(5)
C18	0.6621(2)	0.4406(3)	0.64588(11)	1	0.0238(5)
C19	0.6227(2)	0.3190(3)	0.53485(11)	1	0.0227(5)
C20	0.6295(2)	0.2593(3)	0.46229(12)	1	0.0244(5)
C21	0.7431(2)	0.2681(4)	0.42199(13)	1	0.0338(6)
H21	0.818165	0.315117	0.443526	1	0.041
C22	0.7480(3)	0.2110(4)	0.35283(14)	1	0.0415(7)
H22	0.824785	0.220393	0.326197	1	0.050
C23A	0.6387(3)	0.1390(4)	0.32237(13)	0.595(3)	0.0385(6)
F24	0.6447(3)	0.0784(5)	0.25537(14)	0.595(3)	0.0558(7)
C23B	0.6387(3)	0.1390(4)	0.32237(13)	0.405(3)	0.0385(6)
H23B	0.642451	0.094836	0.274945	0.405(3)	0.046
C25	0.5254(2)	0.1296(3)	0.35852(12)	1	0.0314(6)
H25	0.452022	0.080782	0.335804	1	0.038
C26	0.5171(2)	0.1920(3)	0.42931(11)	1	0.0239(5)

**Anisotropic displacement parameters ( $\text{\AA}^2$ )**

The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [ h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12} ]$ .

Atom	U11	U22	U33	U23	U13	U12
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C1	0.0323(12)	0.0204(10)	0.0214(11)	0.0061(8)	-0.0065(9)	-0.0041(9)
C2	0.0414(15)	0.0342(13)	0.0239(12)	0.0050(10)	-0.0122(11)	-0.0119(11)
C3A	0.0353(15)	0.0532(17)	0.0336(14)	0.0123(12)	-0.0124(11)	-0.0201(12)
C3B	0.0353(15)	0.0532(17)	0.0336(14)	0.0123(12)	-0.0124(11)	-0.0201(12)
F4	0.0446(14)	0.0897(19)	0.0330(12)	-0.0144(12)	-0.0059(10)	-0.0068(13)
C5	0.0325(14)	0.0579(18)	0.0331(14)	0.0133(13)	-0.0041(11)	-0.0166(13)
C6	0.0295(13)	0.0402(14)	0.0248(12)	0.0062(10)	-0.0018(10)	-0.0084(10)
C7	0.0278(12)	0.0226(11)	0.0201(11)	0.0060(8)	-0.0061(9)	-0.0037(9)
C8	0.0260(11)	0.0204(10)	0.0186(10)	0.0042(8)	-0.0039(8)	-0.0004(8)
C9	0.0289(12)	0.0232(11)	0.0168(10)	0.0055(8)	-0.0052(9)	-0.0032(9)
C10	0.0325(13)	0.0363(13)	0.0202(11)	0.0043(9)	-0.0014(9)	-0.0033(10)
C11	0.0418(15)	0.0410(14)	0.0185(11)	-0.0002(10)	0.0000(10)	0.0004(11)
C12	0.0469(15)	0.0279(12)	0.0189(11)	0.0019(9)	-0.0102(10)	-0.0034(10)
C13	0.0357(13)	0.0252(11)	0.0202(11)	0.0057(9)	-0.0096(9)	-0.0035(9)
C14	0.0382(14)	0.0385(14)	0.0270(12)	0.0065(10)	-0.0160(11)	-0.0106(11)
C15	0.0286(13)	0.0618(18)	0.0320(14)	0.0084(12)	-0.0102(11)	-0.0132(12)
C16	0.0269(13)	0.0550(17)	0.0259(12)	0.0021(11)	-0.0053(10)	-0.0027(11)
C17	0.0260(12)	0.0285(12)	0.0199(11)	0.0034(9)	-0.0066(9)	-0.0001(9)
C18	0.0286(12)	0.0236(11)	0.0193(10)	0.0050(8)	-0.0059(9)	-0.0013(9)
C19	0.0243(11)	0.0233(11)	0.0206(11)	0.0045(8)	-0.0056(8)	0.0007(9)
C20	0.0256(11)	0.0264(11)	0.0212(11)	0.0016(9)	-0.0045(9)	0.0060(9)
C21	0.0248(12)	0.0485(15)	0.0280(13)	-0.0035(11)	-0.0015(10)	0.0061(11)
C22	0.0353(14)	0.0613(18)	0.0279(13)	-0.0069(12)	0.0017(11)	0.0150(13)
C23A	0.0463(16)	0.0480(16)	0.0213(12)	-0.0088(11)	-0.0040(11)	0.0165(12)
F24	0.0446(14)	0.0897(19)	0.0330(12)	-0.0144(12)	-0.0059(10)	-0.0068(13)
C23B	0.0463(16)	0.0480(16)	0.0213(12)	-0.0088(11)	-0.0040(11)	0.0165(12)
C25	0.0382(14)	0.0325(13)	0.0236(12)	-0.0030(10)	-0.0111(10)	0.0062(10)
C26	0.0320(12)	0.0205(10)	0.0193(10)	0.0020(8)	-0.0073(9)	0.0034(9)

**Bond lengths [ $\text{\AA}$ ]**

C1 - C2	= 1.409(3)
C1 - C7	= 1.421(3)
C1 - C26	= 1.452(3)
C2 - C3B	= 1.370(4)
C2 - C3A	= 1.370(4)
C2 - H2	= 0.9500
C3A - C5	= 1.388(4)
C3A - H3A	= 0.9500
C3B - F4	= 1.308(5)
C3B - C5	= 1.388(4)
C5 - C6	= 1.374(3)

C5 - H5 = 0.9500  
 C6 - C7 = 1.411(3)  
 C6 - H6 = 0.9500  
 C7 - C8 = 1.433(3)  
 C8 - C19 = 1.384(3)  
 C8 - C9 = 1.485(3)  
 C9 - C10 = 1.377(3)  
 C9 - C18 = 1.418(3)  
 C10 - C11 = 1.417(3)  
 C10 - H10 = 0.9500  
 C11 - C12 = 1.367(4)  
 C11 - H11 = 0.9500  
 C12 - C13 = 1.418(4)  
 C12 - H12 = 0.9500  
 C13 - C18 = 1.395(3)  
 C13 - C14 = 1.420(4)  
 C14 - C15 = 1.366(4)  
 C14 - H14 = 0.9500  
 C15 - C16 = 1.422(3)  
 C15 - H15 = 0.9500  
 C16 - C17 = 1.375(3)  
 C16 - H16 = 0.9500  
 C17 - C18 = 1.414(3)  
 C17 - C19 = 1.482(3)  
 C19 - C20 = 1.435(3)  
 C20 - C21 = 1.411(3)  
 C20 - C26 = 1.420(3)  
 C21 - C22 = 1.367(3)  
 C21 - H21 = 0.9500  
 C22 - C23B = 1.386(4)  
 C22 - C23A = 1.386(4)  
 C22 - H22 = 0.9500  
 C23A - F24 = 1.338(4)  
 C23A - C25 = 1.369(4)  
 C23B - C25 = 1.369(4)  
 C23B - H23B = 0.9500  
 C25 - C26 = 1.410(3)  
 C25 - H25 = 0.9500

**Angles [°]**

C2 - C1 - C7 = 118.80(2)  
 C2 - C1 - C26 = 121.20(2)  
 C7 - C1 - C26 = 120.10(2)  
 C3B - C2 - C1 = 120.70(2)  
 C3A - C2 - C1 = 120.70(2)  
 C3A - C2 - H2 = 119.70  
 C1 - C2 - H2 = 119.70  
 C2 - C3A - C5 = 121.10(2)  
 C2 - C3A - H3A = 119.40  
 C5 - C3A - H3A = 119.40  
 F4 - C3B - C2 = 119.80(3)  
 F4 - C3B - C5 = 119.00(3)  
 C2 - C3B - C5 = 121.10(2)  
 C6 - C5 - C3A = 119.50(3)  
 C6 - C5 - C3B = 119.50(3)  
 C6 - C5 - H5 = 120.30  
 C3A - C5 - H5 = 120.30  
 C5 - C6 - C7 = 121.40(2)  
 C5 - C6 - H6 = 119.30  
 C7 - C6 - H6 = 119.30  
 C6 - C7 - C1 = 118.50(2)  
 C6 - C7 - C8 = 122.80(2)  
 C1 - C7 - C8 = 118.70(2)  
 C19 - C8 - C7 = 121.30(2)  
 C19 - C8 - C9 = 108.60(19)  
 C7 - C8 - C9 = 130.10(2)  
 C10 - C9 - C18 = 117.20(2)  
 C10 - C9 - C8 = 137.10(2)  
 C18 - C9 - C8 = 105.61(19)  
 C9 - C10 - C11 = 119.30(2)  
 C9 - C10 - H10 = 120.30  
 C11 - C10 - H10 = 120.30  
 C12 - C11 - C10 = 122.90(2)

C12	- C11	- H11	=	118.50
C10	- C11	- H11	=	118.50
C11	- C12	- C13	=	119.40(2)
C11	- C12	- H12	=	120.30
C13	- C12	- H12	=	120.30
C18	- C13	- C12	=	116.90(2)
C18	- C13	- C14	=	116.60(2)
C12	- C13	- C14	=	126.50(2)
C15	- C14	- C13	=	119.60(2)
C15	- C14	- H14	=	120.20
C13	- C14	- H14	=	120.20
C14	- C15	- C16	=	122.60(2)
C14	- C15	- H15	=	118.70
C16	- C15	- H15	=	118.70
C17	- C16	- C15	=	119.40(2)
C17	- C16	- H16	=	120.30
C15	- C16	- H16	=	120.30
C16	- C17	- C18	=	117.10(2)
C16	- C17	- C19	=	137.20(2)
C18	- C17	- C19	=	105.71(19)
C13	- C18	- C17	=	124.60(2)
C13	- C18	- C9	=	124.20(2)
C17	- C18	- C9	=	111.21(19)
C8	- C19	- C20	=	121.20(2)
C8	- C19	- C17	=	108.84(19)
C20	- C19	- C17	=	129.90(2)
C21	- C20	- C26	=	118.70(2)
C21	- C20	- C19	=	122.50(2)
C26	- C20	- C19	=	118.70(2)
C22	- C21	- C20	=	121.90(2)
C22	- C21	- H21	=	119.10
C20	- C21	- H21	=	119.10
C21	- C22	- C23B	=	118.60(3)
C21	- C22	- C23A	=	118.60(3)
C21	- C22	- H22	=	120.70
C23A	- C22	- H22	=	120.70
F24	- C23A	- C25	=	119.40(3)
F24	- C23A	- C22	=	118.50(3)
C25	- C23A	- C22	=	122.00(2)
C25	- C23B	- C22	=	122.00(2)
C25	- C23B	- H23B	=	119.00
C22	- C23B	- H23B	=	119.00
C23B	- C25	- C26	=	120.40(2)
C23A	- C25	- C26	=	120.40(2)
C23A	- C25	- H25	=	119.80
C26	- C25	- H25	=	119.80
C25	- C26	- C20	=	118.30(2)
C25	- C26	- C1	=	121.70(2)
C20	- C26	- C1	=	120.00(2)

**Torsion angles [°]**

C7	- C1	- C2	- C3B	=	-0.80(4)
C26	- C1	- C2	- C3B	=	-179.70(2)
C7	- C1	- C2	- C3A	=	-0.80(4)
C26	- C1	- C2	- C3A	=	-179.70(2)
C1	- C2	- C3A	- C5	=	-0.10(4)
C1	- C2	- C3B	- F4	=	176.60(3)
C1	- C2	- C3B	- C5	=	-0.10(4)
C2	- C3A	- C5	- C6	=	0.80(4)
F4	- C3B	- C5	- C6	=	-175.90(4)
C2	- C3B	- C5	- C6	=	0.80(4)
C3A	- C5	- C6	- C7	=	-0.60(4)
C3B	- C5	- C6	- C7	=	-0.60(4)
C5	- C6	- C7	- C1	=	-0.30(4)
C5	- C6	- C7	- C8	=	-179.90(2)
C2	- C1	- C7	- C6	=	1.00(3)
C26	- C1	- C7	- C6	=	179.90(2)
C2	- C1	- C7	- C8	=	-179.40(2)
C26	- C1	- C7	- C8	=	-0.60(3)
C6	- C7	- C8	- C19	=	179.90(2)
C1	- C7	- C8	- C19	=	0.40(3)
C6	- C7	- C8	- C9	=	-2.40(4)
C1	- C7	- C8	- C9	=	178.00(2)

C19	- C8	- C9	- C10	= 178.30(3)
C7	- C8	- C9	- C10	= 0.40(4)
C19	- C8	- C9	- C18	= -0.20(2)
C7	- C8	- C9	- C18	= -178.00(2)
C18	- C9	- C10	- C11	= 1.40(3)
C8	- C9	- C10	- C11	= -176.90(2)
C9	- C10	- C11	- C12	= -0.30(4)
C10	- C11	- C12	- C13	= -1.20(4)
C11	- C12	- C13	- C18	= 1.40(3)
C11	- C12	- C13	- C14	= -178.80(2)
C18	- C13	- C14	- C15	= -1.20(3)
C12	- C13	- C14	- C15	= 178.90(2)
C13	- C14	- C15	- C16	= 2.20(4)
C14	- C15	- C16	- C17	= -0.30(4)
C15	- C16	- C17	- C18	= -2.50(4)
C15	- C16	- C17	- C19	= 178.30(3)
C12	- C13	- C18	- C17	= 178.10(2)
C14	- C13	- C18	- C17	= -1.70(3)
C12	- C13	- C18	- C9	= -0.20(3)
C14	- C13	- C18	- C9	= 180.00(2)
C16	- C17	- C18	- C13	= 3.60(3)
C19	- C17	- C18	- C13	= -176.90(2)
C16	- C17	- C18	- C9	= -177.90(2)
C19	- C17	- C18	- C9	= 1.60(2)
C10	- C9	- C18	- C13	= -1.20(3)
C8	- C9	- C18	- C13	= 177.60(2)
C10	- C9	- C18	- C17	= -179.70(2)
C8	- C9	- C18	- C17	= -0.90(2)
C7	- C8	- C19	- C20	= 1.00(3)
C9	- C8	- C19	- C20	= -177.03(19)
C7	- C8	- C19	- C17	= 179.21(19)
C9	- C8	- C19	- C17	= 1.10(2)
C16	- C17	- C19	- C8	= 177.60(3)
C18	- C17	- C19	- C8	= -1.70(2)
C16	- C17	- C19	- C20	= -4.40(4)
C18	- C17	- C19	- C20	= 176.30(2)
C8	- C19	- C20	- C21	= 176.50(2)
C17	- C19	- C20	- C21	= -1.20(4)
C8	- C19	- C20	- C26	= -2.30(3)
C17	- C19	- C20	- C26	= 180.00(2)
C26	- C20	- C21	- C22	= -1.10(4)
C19	- C20	- C21	- C22	= -179.90(2)
C20	- C21	- C22	- C23B	= -1.40(4)
C20	- C21	- C22	- C23A	= -1.40(4)
C21	- C22	- C23A	- F24	= -178.10(3)
C21	- C22	- C23A	- C25	= 2.40(4)
C21	- C22	- C23B	- C25	= 2.40(4)
C22	- C23B	- C25	- C26	= -0.80(4)
F24	- C23A	- C25	- C26	= 179.70(3)
C22	- C23A	- C25	- C26	= -0.80(4)
C23B	- C25	- C26	- C20	= -1.80(3)
C23A	- C25	- C26	- C20	= -1.80(3)
C23B	- C25	- C26	- C1	= 177.60(2)
C23A	- C25	- C26	- C1	= 177.60(2)
C21	- C20	- C26	- C25	= 2.70(3)
C19	- C20	- C26	- C25	= -178.50(2)
C21	- C20	- C26	- C1	= -176.70(2)
C19	- C20	- C26	- C1	= 2.10(3)
C2	- C1	- C26	- C25	= -1.20(3)
C7	- C1	- C26	- C25	= 179.90(2)
C2	- C1	- C26	- C20	= 178.20(2)
C7	- C1	- C26	- C20	= -0.70(3)

**Structure visualisation**

