

Electronic Supplementary Information (ESI)

**The mechanism of ratiometric fluoride sensor and ESIPT  
process for 2,6-dibenzothiazolyphenol and its derivative**

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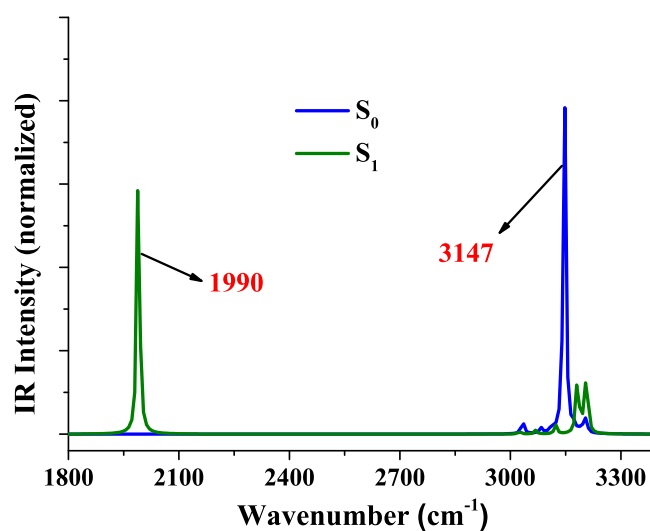


Figure S1: The theoretical IR spectra of 26DB in the spectra region of O1-H2 stretching vibrational mode in both S<sub>0</sub> and S<sub>1</sub> states.

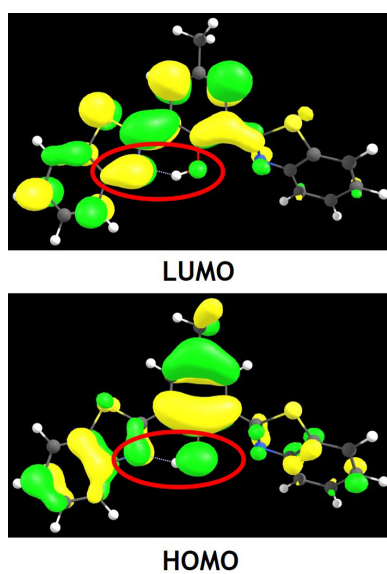


Figure S2: View of the frontier molecule orbitals (HOMO and LUMO) for 26DB molecule. Herein, the red cycle indicates the charge transfer moiety involved in hydrogen bond (O1-H2  $\cdots$  N3) of 26DB.

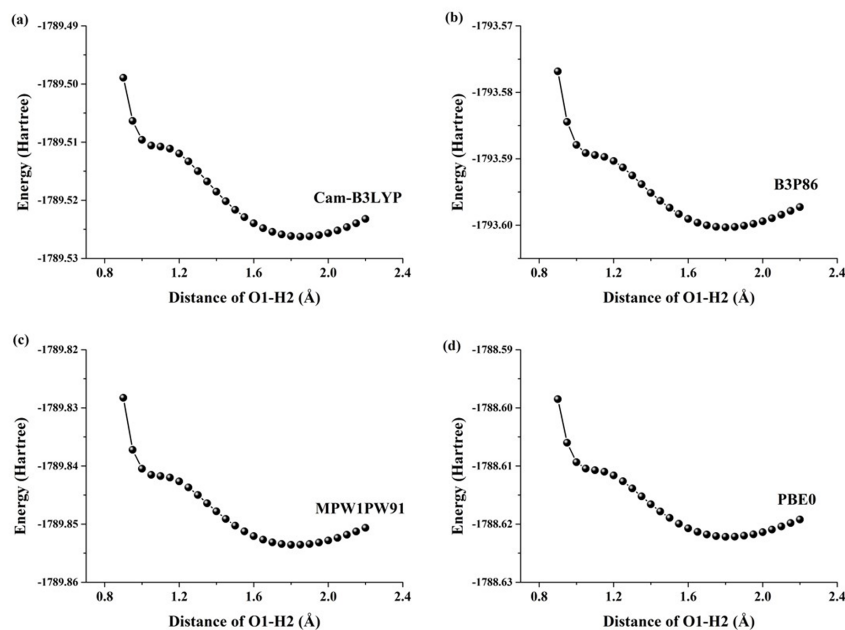


Figure S3: The theoretical  $S_1$ -state potential energy curves for 26DB system in step of 0.05 Å based on four kinds of functionals. (a) Cam-B3LYP; (b) B3P86; (c) MPW1PW91; (d) PBE0.

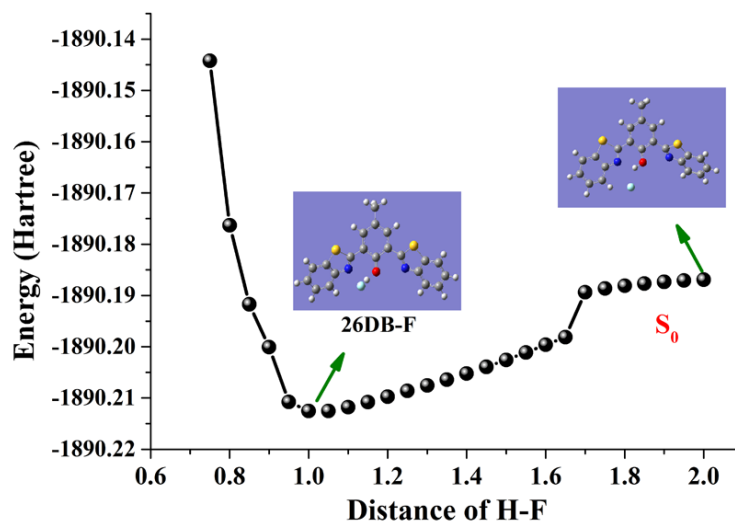


Figure S4: Potential energy curve of  $S_0$  state for 26DB with the addition of fluoride anion, which is the function of H-F distance.

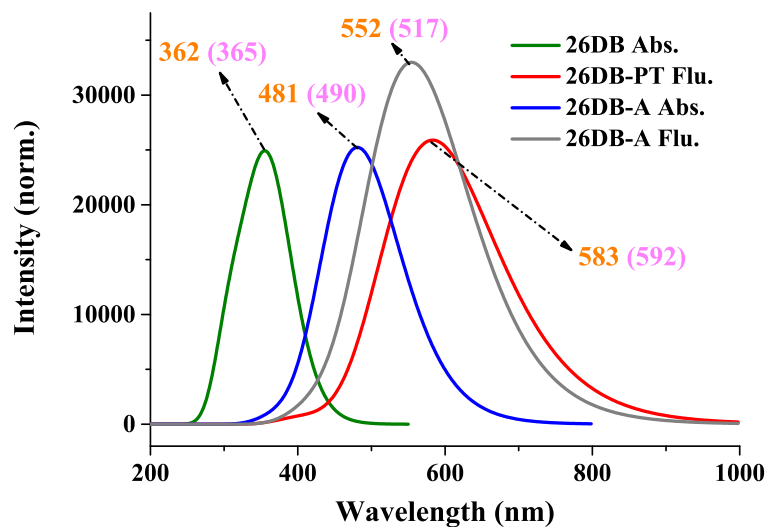


Figure S5: The theoretical electronic spectra of 26DB, 26DB-PT and 26DB-A structures in THF solvent. Herein, the previous experimental results are listed in the brackets.

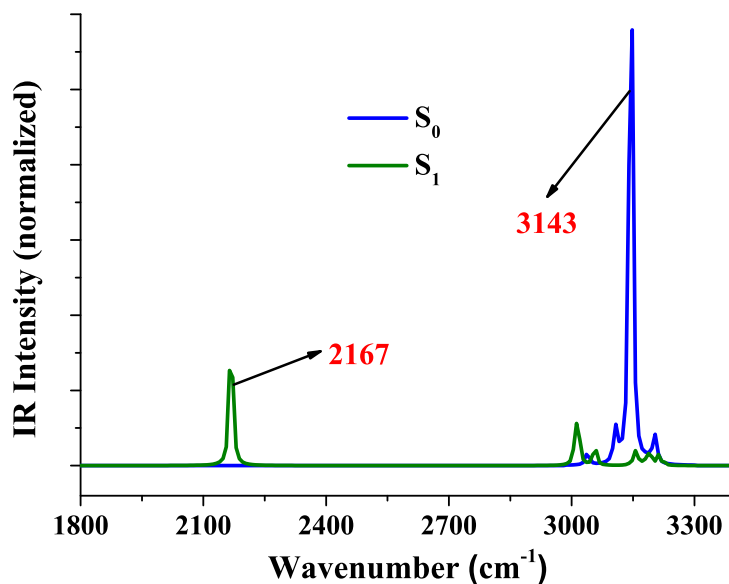


Figure S6: The theoretical IR spectra of Bis-26DB in the spectra region of synergetic O1-H2 and O4-H5 stretching vibrational mode in both  $S_0$  and  $S_1$  states.