

Supporting Information

Mn₃O₄ hollow microcubes and solid nanospheres derived from a metal formate framework for electrochemical capacitor applications

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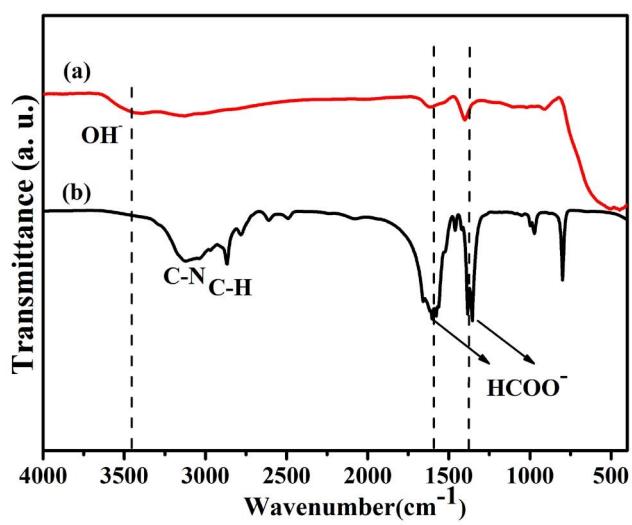


Fig. S1 FT-IR spectra of the precursor (a) and the obtained product by NaOH treatment of precursor (b).

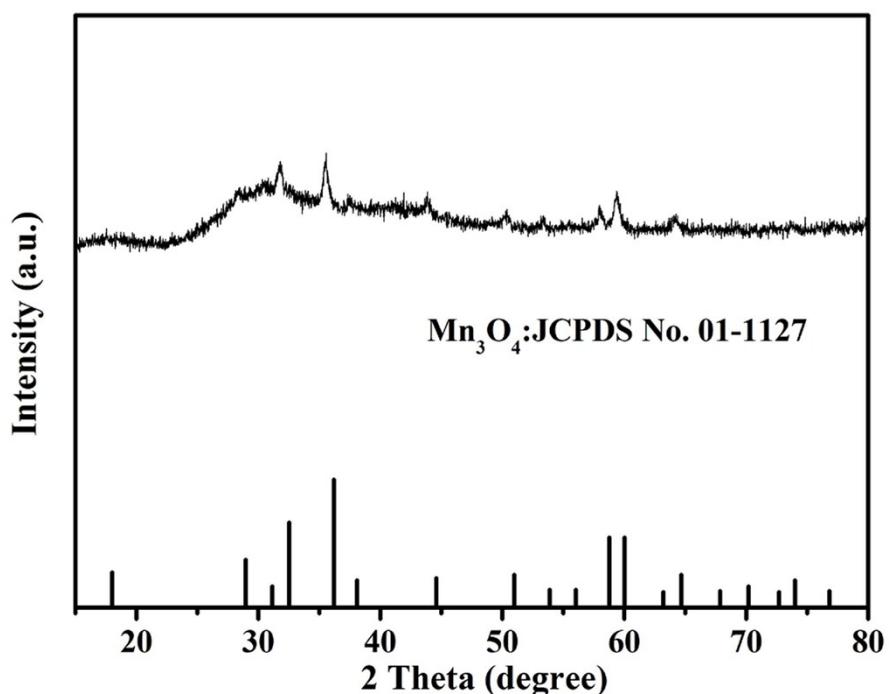


Fig. S2 XRD pattern of Mn_3O_4 hollow microcubes synthesized by reaction for 12 h.

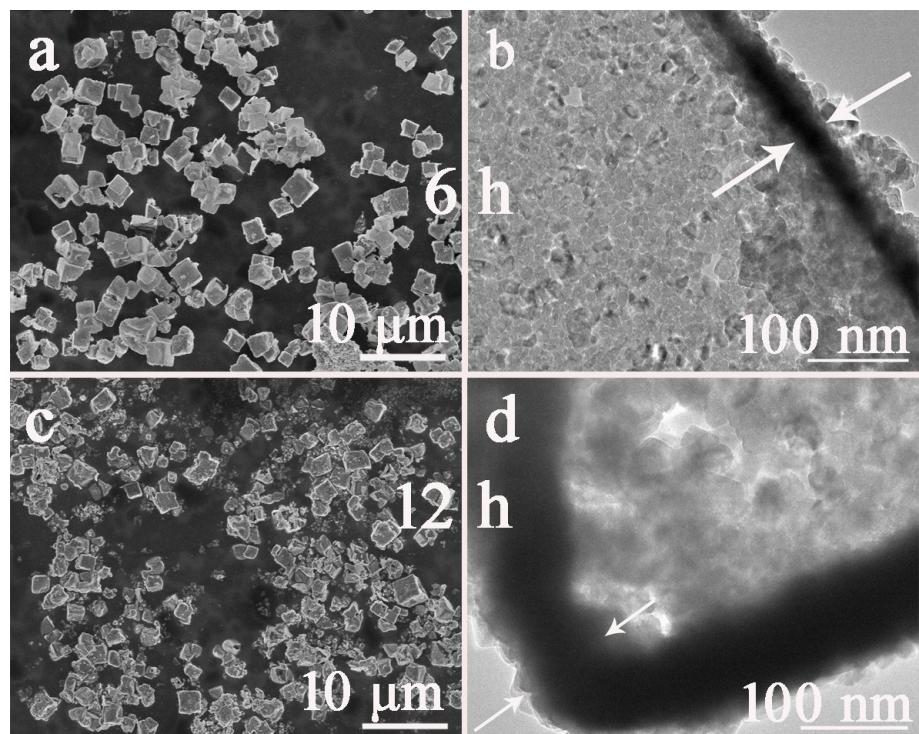


Fig. S3 SEM (a, c) and TEM images (b, d) of Mn_3O_4 hollow microcubes synthesized by reaction for 6 h and 12 h.

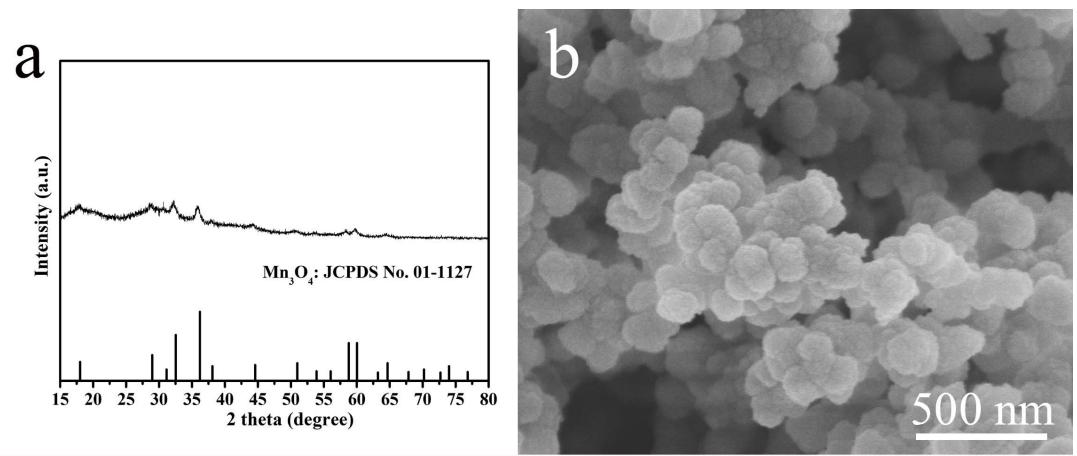


Fig. S4 XRD pattern (a) and SEM image (b) of Mn_3O_4 nanoparticles synthesized by reaction of precursor with alkaline solution in a molar ratio of 1:2 for 6 h.

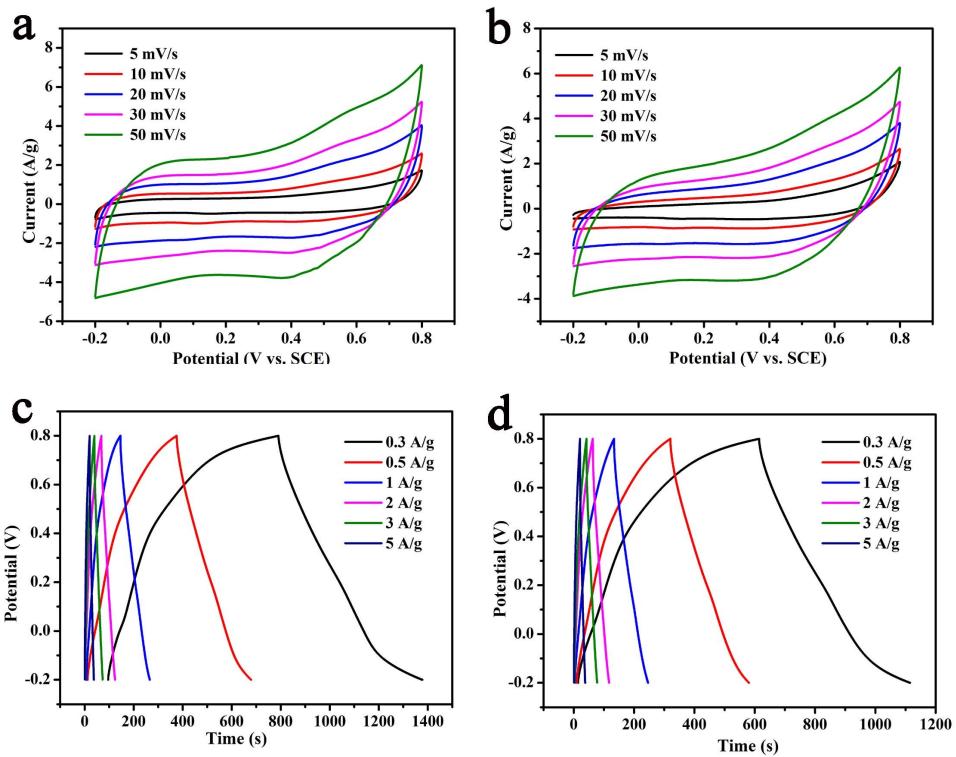


Fig. S5 CV (a, b) and GCD (c, d) curves of Mn_3O_4 hollow microcubes (a,c) and solid nanospheres (b,d).

Table S1. Comparison of the electrochemical performances of Mn_3O_4 electrode materials prepared in the present work with other reported Mn_3O_4 based electrode materials.

Materials	Electrolyte	Test condition	$C_s(\text{F/g})$	Specific capacitance retention after cycle	Ref
Mn_3O_4 hollow-tetrakaidecahedrons	1 M Na_2SO_4	5 mV/s	148	100% after 400 cycles	1
Mn_3O_4 hexagonal plate	1 M Na_2SO_4	0.5 A/g	82	100% after 1000 cycles	2
Mn_3O_4 nanorod/graphene	1 M Na_2SO_4	0.5 A/g	121	100% after 1000 cycles	3
nitrogen-doped carbon/ Mn_3O_4	1 M Na_2SO_4	0.5 A/g	73	94% after 1000 cycles	4
graphene/ Mn_3O_4	1 M Na_2SO_4	0.5 A/g	142	92% after 800 cycles	5
Mn_3O_4 solid nanospheres	1 M Na_2SO_4	0.5 A/g	131	86% after 8000 cycles	This work
Mn_3O_4 hollow microcubes	1 M Na_2SO_4	0.5 A/g	152	95% after 8000 cycles	This work

Supplementary references

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