

Electronic Supplementary Information

**Synthesis and magnetic properties of nanostructured metallic Co, Mn and Ni oxide materials obtained from solid-state metal-macromolecular complexes precursors**

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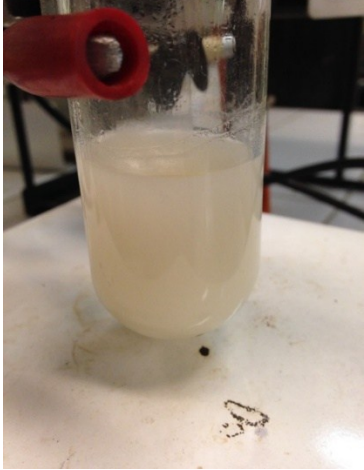
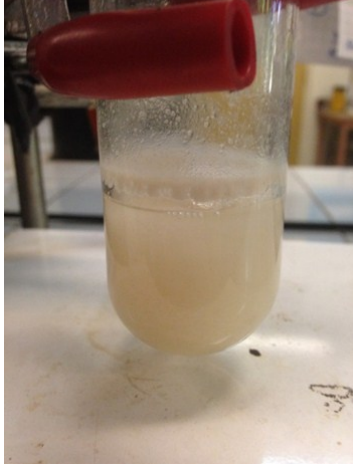
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S<sub>1</sub> Table 1. Experimental Details of the reaction formation of the precursors 1-3

| Number | Precursor                      | g. of Chitosan | g. of metallic salt | ml of solvent | Reaction time | Color of the product |
|--------|--------------------------------|----------------|---------------------|---------------|---------------|----------------------|
| 1.-    | Chitosan•(MnCl <sub>2</sub> )  | 0.52           | 0.6                 | 30            | 2 week        | white                |
| 2.-    | Chitosan•(CoCl <sub>2</sub> )  | 0.79           | 0.6                 | 30            | 2 week        | blue                 |
| 3.-    | Chitosan •(NiCl <sub>2</sub> ) | 0.43           | 0.6                 | 50            | 1 week        | green                |

S<sub>2</sub> Illustrations showing the color of the precursors

|                                                                                     |                                                                                      |
|-------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------|
| 1.- (Chitosan)(CoCl <sub>2</sub> ) <sub>x</sub> : a) at t=0                         | b) at t= after 7 days                                                                |
|    |    |
| Chitosan)(MnCl <sub>2</sub> ) <sub>x</sub> : a) at t=0                              | b) at t= after 7 days                                                                |
|   |   |
| (Chitosan)(NiCl <sub>2</sub> ) <sub>x</sub> (a) at t=0                              | b) at t= after 7 days                                                                |
|  |  |

**S<sub>3</sub> Table 2 Precursors and their estimated metal contents by TG/DSC and elemental Analysis**

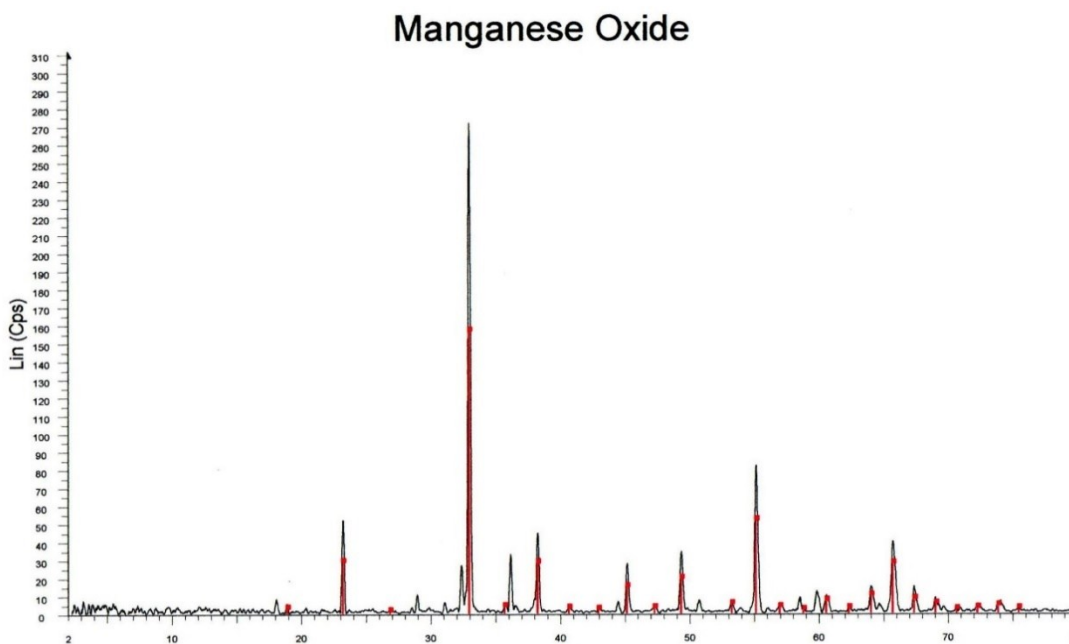
| Number | Precursor                     | % TG/DSC | % elemental analysis |
|--------|-------------------------------|----------|----------------------|
| 1.-    | Chitosan•(MnCl <sub>2</sub> ) | 45       | 46                   |
| 2.-    | Chitosan•(CoCl <sub>2</sub> ) | 86       | 89                   |
| 3.-    | Chitosan•(NiCl <sub>2</sub> ) | 89       | 92                   |

**S<sub>4</sub> Table 3  $\nu(\text{OH})$  data of the Chitosan•(ML<sub>n</sub>) precursor**

| Number | Precursor                     | $\nu(\text{OH}) \text{ cm}^{-1}$ |
|--------|-------------------------------|----------------------------------|
|        | Chitosan                      | 3448                             |
| 1.-    | Chitosan•(MnCl <sub>2</sub> ) | 3393                             |
| 2.-    | Chitosan•(CoCl <sub>2</sub> ) | 3382                             |
| 3.-    | Chitosan•(NiCl <sub>2</sub> ) | 3345                             |

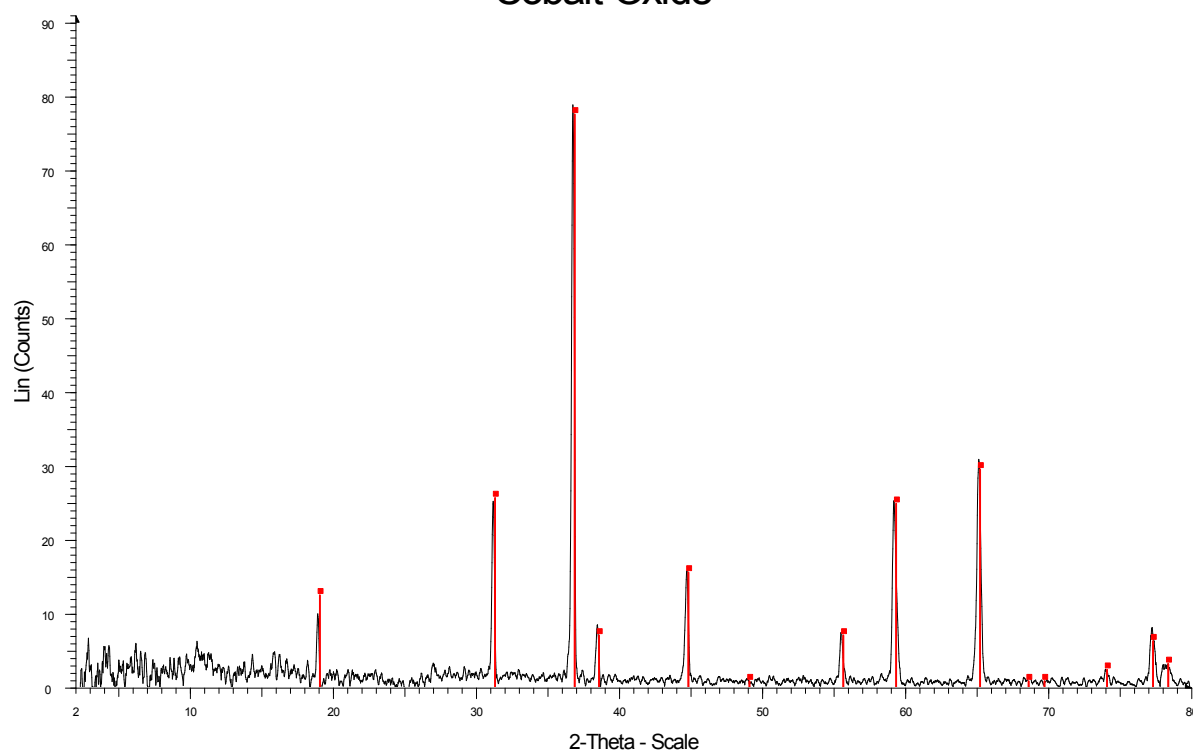
**S<sub>5</sub> XRD Patterns of Mn<sub>2</sub>O<sub>3</sub>, Co<sub>3</sub>O<sub>4</sub> and NiO.**

A) Mn<sub>2</sub>O<sub>3</sub>

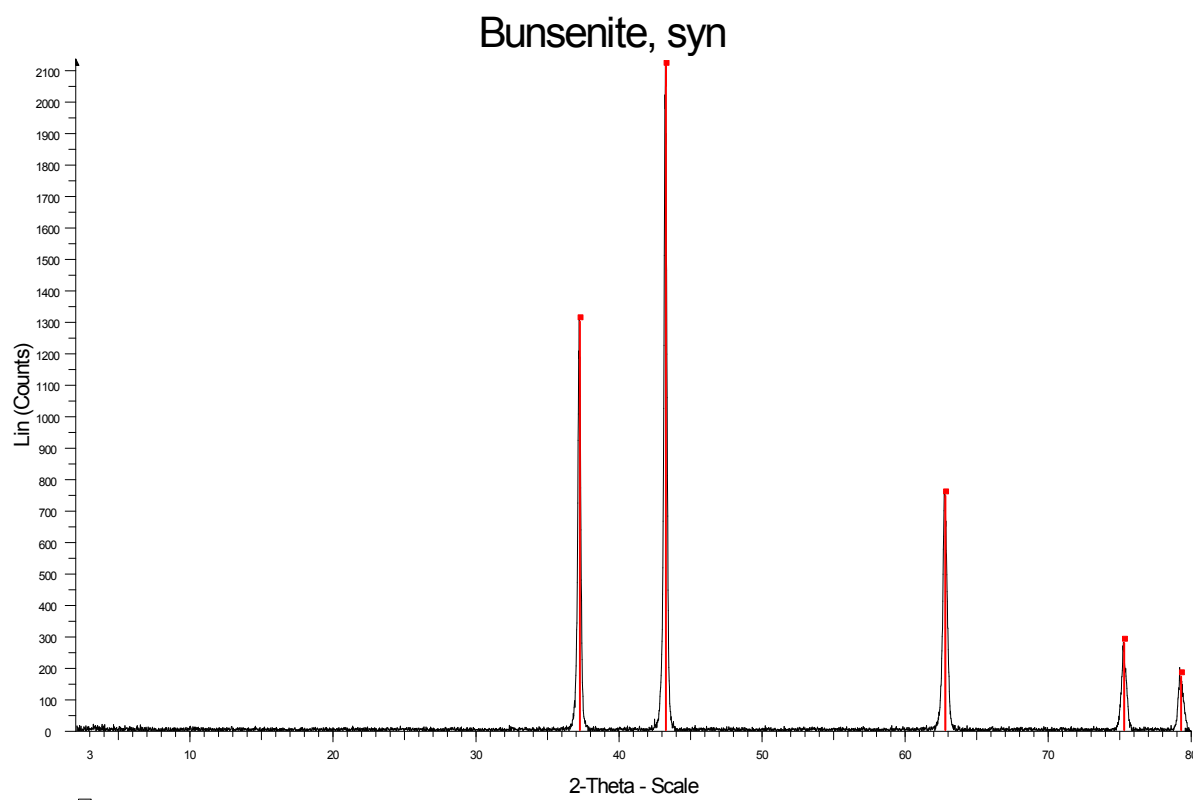


B)  $\text{Co}_3\text{O}_4$

# Cobalt Oxide



C) NiO



The red bars are the respective patterns from the library data.

**S<sub>6</sub> Table 4 ID/IG ratio measured for precursors 1,2 and 3 in several zones with a 15 mW laser power. Samples pirolized at 400 °C.**

| Precursor | Zone | ID/IG ratio |
|-----------|------|-------------|
| 1.-       | 1    | 2.29        |
|           | 2    | 2.04        |
| 2.-       | 1    | 1.43        |
|           | 2    | 1.49        |
| 3.-       | 1    | 2.12        |
|           | 2    | 1.82        |
|           | 3    | 2.38        |
|           | 4    | 1.99        |