Supporting Information

Near-infrared fluorescent and columnar liquid crystal: Synthesis, mesomorphic and photophysical properties of triphenylene-Bodipy-triphenylene triad

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1. General

All chemical reagents were obtained from commercial suppliers and used without further purification. The other organic solvents and inorganic reagents were purified according to standard anhydrous methods before use. TLC analysis was performed using pre-coated glass plates. Column chromatography was performed using silica gel (200-300 mesh). NMR spectra were recorded in CDCl₃ on a Bruker-ARX 600 instrument at 25°C. Chemical shifts are reported in ppm, using tetramethylsilane (TMS) as internal standard. MS spectra were obtained from Bruck mass spectrometer. UV-Vis and fluorescence spectra were recorded on Varian UV-Vis spectrometer. Fluorescence spectra were measured in a conventional quartz cell (10×10×45 nm) at 25 °C on a Hitachi F-4500 spectrometer equipped with a constanttemperature water bath, with excitation and emission slits 10nm wide. The fluorescence quantum yields were determined by the equation: $\Phi_{F(sample)} = (F_{sample}/F_{ref})(A_{ref}/A_{sample})(n_{sample^2}/n_{ref^2})\Phi_{F(ref)}$, where F, A, and *n* are the measured fluorescence (area under the emission peak), the absorbance at the excitation position, and the refractive index of the solvent, respectively. OH-Bodipy was used as the reference (Φ_F=0.94) (Liu, J. Y.; Yeung, H. S.; Xu, W.; Li, X. Ng, D. K. P. Org. Lett. 2008, 10, 5421-5424). The fluorescence absolute Φ_F values of solid films were measured using an Edinburgh Instruments FLS920 Fluorescence Spectrometer with a 6-inch integrating sphere. 4-(prop-2-ynyloxy)benzaldehyde 4 and 3, 4bis-(prop-2-ynyloxy)benzaldehyde 8 were prepare by the published procedure (Hongyu Guo, Xiaoting Fang, Fafu Yang, Yingmei Zhang, J. Incl Phenom. Macrocycl. Chem, 2016, 84(1), 79-86). Compounds 1, 2 and 3 were synthesized according to the reference (Bigiong Hong, Fafu Yang, Hongyu Guo, and Ziyu Jiao, Tetrahedron Letters, 2014, 55, 252-255).

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Scheme S1. The synthetic route for triphenylene-Bodipy dyad derivative 7

2.1. Synthesis of BODIPY derivative 5

Under N₂ atmosphere, four drops of trifluoroacetic acid were added in the CH₂Cl₂ solution(120 mL) of compound **4** (0.1 g, 0.6 mmol) and 2,4-dimethyl-1*H*-pyrrole (0.14g, 1.5 mmol). The solution was stirred at room temperature for 12 h. Subsequently, 2,3-dichloro-5,6-dicyano-p-benzoquinone (0.16 g, 0.7 mmol) was added in reaction system. The mixtural solution was reacted continuously for another 4 h. Furthermore, triethylamine (6 mL, 0.043 mol) and BF₃ • Et₂O (6 mL, 0.05 mol) was added into reaction system. The obtained mixture was further stirred at room temperature overnight. After reaction, 90 mL of water was added in the mixture and the orgainc layer was separated. The organic portion was further filteration, the solution was concentrated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: CH₂Cl₂/ petroleum ether =1:1). Compound **5** was obtained as an red solid in yield of 22%. Compound **5**: ¹H NMR (400 MHz, CDCl₃) δ ppm: 1.40 (s, 6H, CH₃), 2.54 (s, 1H, CH), 2.57 (s, 6H, CH₃), 4.74 (s, 2H, OCH₂), 5.97 (s, 2H, ArH), 7.02 (d, 2H, *J* = 8.0Hz, ArH), 7.18 (d, 2H, *J* = 8.0Hz, ArH). ¹³C NMR (100 MHz, CDCl₃) δ ppm: 14.55, 18.51, 29.71, 53.20, 55.76, 75.73, 77.80, 115.63, 120.99, 127.99, 129.24, 131.77, 141.61, 143.15, 155.37, 158.04. HR-MS(ESI) (C₂₂H₂₂BF₂N₂O) [M]⁺: Calcd.: 379.1791. found: 379.1789.



Figure S1. ¹H NMR spectrum of **5**



2.2. Synthesis of compound 6

CuSO₄·5H₂O (62 mg, 0.25 mmol) and sodium ascorbate (100 mg, 0.5 mmol) were added into the DMF solution (20 mL) of compound **5** (75 mg, 0.2 mmol) and compound **3** (72 mg, 0.1 mmol). The mixture was stirred at 100°C for 15 h. After cooling, the solution was partitioned by CHCl₃ (50 mL) and water (50 mL). The organic phase was separated, washed by water (3×30mL) and then dried over MgSO₄. After filtration, the solvent removed under reduced pressure. The residue was purified by silica gel column chromatography with the eluent of CH₂Cl₂/ petroleum ether (1:1). Compound **6** was collected as an red solid in yield of 66%.

Compound **6**: ¹H NMR (400 MHz, CDCl₃) δ ppm: 0.97 (t, 15H, CH₂<u>CH₃</u>), 1.35~2.03 (m, 44H, CH₂ and CH₃), 2.54 (s, 6H, CH₃), 4.22(bs, 12H, OCH₂), 4.43 (t, 2H, *J*=8.0Hz, NH₂), 5.19 (bs, 2H, OCH₂-trazole), 5.96 (s, 2H, ArH), 7.04 (d, 2H, *J* = 8.0Hz, ArH), 7.11-7.15 (m, 3H, ArH and H-trazole), 7.83 (bs, 6H, TpH); ¹³C NMR (100 MHz, CDCl₃) δ ppm: 13.97, 14.41, 22.42, 25.38, 26.10, 28.39, 29.15, 29.54, 29.85, 50.44, 61.67, 69.01, 69.48, 107.01, 115.21, 115.69, 120.60, 122.55, 123.68, 127.74, 129.02, 131.77, 141.35, 142.88, 143.65, 146.73, 148.93, 155.16, 158.80; MALDI-TOF-MS Calcd.for *m/z* = 1177.7, found: *m/z* = 1179.3(M+H)⁺. HR-MS(ESI) (C₇₁H₉₄BF₂N₅O₇) [M]⁺: Calcd.: 1177.7220. found:1077.7260. Elemental Analysis Calcd.: C 72.37, H 8.04, N 5.94%. Found: C 72.33, H 8.09, N 5.86%.









2.3. Synthesis of compound 7

Under N₂ atmosphere, compound 6 (200 mg, 0.17 mmol) and *p*-dimethylaminobenzaldehyde (45 mg, 0.30 mmol) were add in a round-bottomed flask with 30 mL of dry toluene. 2 mL of piperidine, 2 mL of glacial acetic acid and a small amount of 3 Å molecular sieves were added in the flask. The mixture was stirred and refluxed for 5 h. After cooling, the solvent was evaporated under reduced pressure. The residue was partitioned by CHCl₃ and the solution of NH₄Cl (10%). The organic layer was dried with NaCl (satd.) and anhydrous $MgSO_4$. After concentration, the product was purified by silica gel column chromatography using CH_2Cl_2 / petroleum ether (2:1) as eluent. A purple solid was obtained in the yield of 20%. Compound 7: ¹H NMR (400 MHz, CDCl₃) δppm: 0.99 (t, 15H, J = 8.0Hz, CH₃), 1.40~2.10 (m, 44H, CH₂ and CH₃), 2.59 (s, 3H, CH₃), 3.04 (s, 6H, NCH₃), 4.24 (bs, 12H, OCH₂), 4.45 (t, 2H, J = 8.0Hz, NCH₂), 5.19 (s, 2H, OCH₂), 5.96 (s, 1H, ArH), 6.59 (s, 1H, ArH), 6.68 (d, 2H, J = 8.0Hz, ArH), 7.00 (s, 1H, ArH), 7.02 (d, 2H, J = 8.0Hz, ArH), 7.05 (s, 1H, ArH), 7.17 (d, 2H, J = 8.0Hz, ArH), 7.51 (d, 2H, J = 8.0Hz, ArH), 7.66 (s, 1H, ArH), 7.85 (bs, 6H, ArH); ¹³C NMR (100 MHz, CDCl₃) δppm: 14.15, 22.72, 28.38, 29.39, 29.72, 29.94, 30.32, 31.86, 33.69, 40.26, 50.34, 61.98, 69.13, 69.52, 69.71, 69.94, 106.94, 107.25, 107.41, 112.01, 113.97, 115.14, 117.64, 120.37, 122.68, 123.59, 124.42, 124.74, 124.93, 127.91, 129.11, 129.55, 129.93, 130.85, 131.75, 133.48, 136.08, 137.58, 139.30, 148.60, 151.03, 152.78, 154.63, 158.57; MALDI-TOF-MS Calcd.for m/z = 1308.8, found: m/z = 1309.7 (M+H⁺). HR-MS(ESI) (C₈₀H₁₀₃BF₂N₆O₇) [M+H]⁺: Calcd.: 1309.8035. found:1309.8004. Elemental Analysis Calcd.: C 73.37, H 7.93, N 6.42%. Found: C 73.31, H 7.98, N 6.34%.



Figure S9. ¹³C NMR spectrum of 7



Figure S11. HR-MS (ESI) spectrum of 7



Scheme S2. The synthetic route for triphenylene-Bodipy-triphenylene triad derivative 11

2.4. Synthesis of compound 9

Under N₂ atmosphere, four drops of trifluoroacetic acid were added in the CH₂Cl₂ solution(120 mL) of compound 8 (0.13 g, 0.6 mmol) and 2,4-dimethyl-1*H*-pyrrole (0.14g, 1.5 mmol). The solution was stirred at room temperature for 12 h. Subsequently, 2,3-dichloro-5,6-dicyano-p-benzoquinone (0.16 g, 0.7 mmol) was added in reaction system. The mixtural solution was reacted continuously for another 4 h. Furthermore, triethylamine (6 mL, 0.043 mol) and BF₃ • Et₂O (6 mL, 0.05 mol) was added into reaction system. The obtained mixture was further stirred at room temperature overnight. After reaction, 90 mL of water was added in the mixture and the orgainc layer was separated. The organic portion was further washed with 100 mL of 5% aqueous NaHCO₃ solution and then dried over anhydrous MgSO₄. After filteration, the solution was concentrated under reduced pressure and the residue was purified by silica gel column chromatography (eluent: CH_2Cl_2 / petroleum ether =1:1). Compound 9 was obtained as an red solid in yield of 24%. Compound 9: ¹H NMR (400 MHz, CDCl₃) δppm: 1.50 (s, 6H, CH₃), 2.47 (s, 1H, CH), 2.56 (s, 1H, CH), 2.57 (s, 6H, CH₃), 4.79 (s, 2H, OCH₂), 4.85 (s, 2H, OCH₂), 6.00 (s, 2H, ArH), 6.91 (d, 1H, J = 8.0Hz, ArH), 7.00 (s, 1H, ArH), 7.18 (d, 1H, J = 8.0Hz, ArH). ¹³C NMR (100 MHz, CDCl₃) δppm:14.44, 14.60, 29.71, 53.48, 56.71, 57.06, 76.28, 76.33, 77.88, 78.11, 114.58, 115.49, 121.20, 121.58, 128.46, 131.63, 141.07, 143.07, 148.05, 155.39. HR-MS(ESI) (C₂₅H₂₄BF₂N₂O₂) [M]⁺: Calcd.: 433.1897. found:433.1869.







Figure S13. ¹³C NMR spectrum of **9**



2.5. Synthesis of compound 10

 $CuSO_4 \cdot 5H_2O$ (62 mg, 0.25 mmol) and sodium ascorbate (100 mg, 0.5 mmol) were added into the DMF solution (20 mL) of compound **9** (90 mg, 0.21 mmol) and compound **3** (72 mg, 0.1 mmol). The mixture was stirred at 100°C for 15 h. After cooling, the solution was partitioned by $CHCl_3$ (50 mL) and water (50 mL). The organic phase was separated, washed by water (3×30mL) and then dried over MgSO₄. After filtration, the solvent removed under reduced pressure. The residue was was purified by silica gel column chromatography with the eluent of $CH_2Cl_2/$ petroleum ether (1:1). Compound **10** was collected as an red solid in yield of 64%.

Compound **10**: ¹H NMR (400 MHz, CDCl₃) δppm: 0.99 (t, 30H, CH₂CH₃), 1.33~2.02 (m, 82H, CH₂ and CH₃), 2.54 (s, 6H, CH₃), 4.23(m, 24H, OCH₂), 4.38 (m, 4H, NH₂), 5.22 (bs, 4H, OCH₂-trazole), 5.96 (s, 2H, ArH), 6.79 (d, 1H, J = 8.0Hz, ArH), 6.94 (s, 1H, ArH), 7.05 (d, 1H, J = 8.0Hz, ArH), 7.70--7.73 (m, 2H, H-trazole), 7.84 (bs, 12H, TpH); ¹³C NMR (100 MHz, CDCl₃) δppm: 14.11, 14.35, 20.46, 22.57, 25.50, 25.63, 26.09, 26.38, 28.38, 29.70, 30.13, 31.32, 33.16, 50.27, 65.57, 69.17, 69.52, 69.69, 80.17, 107.02, 107.34, 107.41, 107.51, 115.59, 115.69, 121.16, 123.45, 123.68, 128.33, 128.81, 130.89, 131.58, 140.95, 142.99, 148.76, 148.89, 149.02, 155.38; MALDI-TOF-MS Calcd.for m/z = 2033.5, found: m/z = 2035.2 (M+H⁺). HR-MS(ESI) (C₁₂₃H₁₆₉BF₂N₈O₁₄) [M+Na]⁺: Calcd.: 2055.2754. found:2055.2663. Elemental Analysis Calcd.: C 72.68, H 8.38, N 5.51%. Found: C 72.63, H 8.44, N 5.42%.



Figure S16. ¹³C NMR spectrum of **10**







2.6. Synthesis of compound 11

Under N₂ atmosphere, compound **10** (200 mg, 0.98 mmol) and *p*-dimethylaminobenzaldehyde (36 mg, 2.5 mmol) were add in a round-bottomed flask with 30 mL of dry toluene. 2 mL of piperidine, 2 mL of glacial acetic acid and a small amount of 3 Å molecular sieves were added in the flask. The mixture was stirred and refluxed for 5 h. After cooling, the solvent was evaporated under reduced pressure. The residue was partitioned by CHCl₃ and the solution of NH₄Cl (10%). The organic layer was dried with NaCl (satd.) and anhydrous MgSO₄. After concentration, the product was purified by silica gel column chromatography using CH₂Cl₂/ petroleum ether (2:1) as eluent. A green solid was obtained in the yield of 21%.

Compound **11**: ¹H NMR (400 MHz, CDCl₃) δ ppm: 0.98 (t, 30H, CH₂<u>CH₃</u>), 1.35~2.03 (m, 82H, CH₂ and CH₃), 3.00 (s, 12H, NCH₃), 4.23 (m, 24H, OCH₂), 4.39 (m, 4H, NH₂), 5.24 (s, 2H, OCH₂-trazole), 5.27 (s, 2H, OCH₂-trazole), 6.59 (s, 2H, ArH), 6.66 (d, 4H, *J* = 8.0Hz, ArH), 6.84 (d, 1H, *J* = 8.0Hz, ArH), 6.96 (s, 1H, ArH), 7.05 (d, 1H, *J* = 8.0Hz, ArH), 7.16 (d, 2H, *J* = 12.0Hz, CH=), 7.49 (d, 4H, *J* = 8.0Hz, ArH), 7.53 (d, 2H, *J* = 12.0Hz, CH=), 7.69 (s, 1H, H-trazole), 7.76 (s, 1H, H-trazole), 7.83 (bs, 12H, TpH); ¹³C NMR (100 MHz, CDCl₃) δ ppm: 14.13, 15.77, 22.71, 23.75, 25.54, 26.75, 27.30, 28.49, 288.88, 29.53, 30.05, 31.94, 32.63, 33.07, 33.84, 38.78, 50.13, 63.39, 68.17, 69.18, 69.33, 69.85, 107.07, 107.31, 114.08, 116.45, 116.99, 119.92, 122.93, 123.54, 124.51, 126.40, 128.22, 128.85, 129.10, 129.46, 130.79, 131.60, 132.24, 132.42, 133.30, 133.94, 135.45, 138.11, 139.16, 140.41, 145.24, 148.91, 149.70, 151.53, 157.15, 157.95; MALDI-TOF-MS Calcd.for *m*/*z* = 2294.5, found: *m*/*z* = 2296.6 (M+H⁺). HR-MS(ESI) (C₁₄₁H₁₈₇BF₂N₁₀O₁₄) [M]⁺: Calcd.: 2294.4327. found:2294.4277. Elemental Analysis Calcd.: C 73.80, H 8.21, N 6.10%. Found: C 73.73, H 8.28, N 6.01%.



Figure S19. ¹H NMR spectrum of **11**





Figure S21. MALDI-TOF-MS spectrum of 11



Figure S22. HR-MS (ESI) spectrum of 11



Figure S23. Fluorescence emission spectra of compounds 6 and 7 in toluene at different temperature.



Figure S24. Fluorescence emission spectra of compounds 10 and 11 in toluene at different temperature.