1	<b>Oxidation modification of Ru-based catalyst for</b>					
2	acetylene hydrochlorination					
3	Baochang Man, <sup>a, b</sup> Haiyang Zhang, <sup>*a, b</sup> Jinli Zhang, <sup>*a, c</sup> Xing Li, <sup>a, b</sup> Na Xu, <sup>a,</sup>					
4	<sup>b</sup> Hui Dai, <sup>c</sup> Mingyuan Zhu <sup>a, b</sup> and Bin Dai <sup>a, b</sup>					
5						
6	Table of contents:					
7	Table S1 The catalytic performance of some potential non-mercury catalysts recently					
8	reported in the literatures.					
9	Table S2 The relative content of the elements in the catalysts, determined by XPS					
10	spectra.					
11	Figure S1. The schematic diagram of four different routes to synthesize Ru-based					
12	catalysts.					
13	Figure S2. FT-IR spectra of the fresh and used Ru/AC catalysts.					
14	Figure S3. Nitrogen adsorption-desorption isotherms of the catalysts.					
15	Figure S4. TG curves of the fresh and used catalysts recorded in air atmosphere.					
16	Figure S5. High-resolution XPS spectra of Ru 3p for the fresh and used Ru-based					
17	catalysts.					
18	Figure S6. XPS spectra of C1s for the fresh catalysts.					

\* Corresponding author Tel: 86-993-2057-277; Fax: +86 993 2057210; E-mail address: <u>zhy198722@163.com</u> (H.Y. Zhang), zhangjinli@tju.edu.cn (J.L. Zhang)

<sup>&</sup>lt;sup>a</sup> School of Chemistry and Chemical Engineering of Shihezi University, Shihezi, Xinjiang 832000, PR China.

<sup>&</sup>lt;sup>b</sup> Key Laboratory for Green Processing of Chemical Engineering of Xinjiang Bingtuan, Shihezi, Xinjiang 832000, PR China.

<sup>&</sup>lt;sup>c</sup> School of Chemical Engineering & Technology, Tianjin University, Tianjin 300072, P.R. China.

- 19 Figure S7. TEM images of the fresh and used catalysts: (a) Fresh Ru/AC, (b) Used
- 20 Ru/AC, (c) Fresh Ru/AC-O, (d) Used Ru/AC-O, (e) Fresh (Ru/AC)-O, (f) Used
- 21 (Ru/AC)-O, (g) Fresh Ru-O/AC-O, (h) Used Ru-O/AC-O.

Catalyst	Composition of catalysts	Reaction conditions	Initial catalytic activity <sup>a</sup>	Deactivation rate
				(% h <sup>-1</sup> ) <sup>b</sup>
Mo <sub>2</sub> N/AC [1]	1.5 wt% MoN	$T = 180 \text{ °C. GHSV}(C_2H_2) = 30 \text{ h}^{-1},$	$X_A = 80.0\%, Svc_M > 90.0\%$	4.28
		$V_{HCl}/V_{C2H2} = 1.15$		
Au-Bi/C <sup>[2]</sup>	0.30  wt% Au, n(Bi)/n(Au) = 3	T = 180 °C. GHSV( $C_2H_2$ ) = 600 h <sup>-1</sup> ,	$X_A = 87.0\%, Svc_M > 99.0\%$	0.500
		$\mathbf{V}_{\mathrm{HCl}}/\mathbf{V}_{\mathrm{C2H2}} = 1.1$		
PdCl <sub>2</sub> -KCl-LaCl <sub>3</sub> /C <sup>[3]</sup>	0.9 wt% Pd, 0.2 wt% KCl, 0.2	T = 120-180 °C. GHSV( $C_2H_2$ ) = 120 h <sup>-1</sup> ,	$X_A = 98.0\%, Svc_M > 99.5\%$	10.500
	wt% LaCl <sub>3</sub>	$V_{\rm HCl}/V_{\rm C2H2} = 1.15$		
Z4M1 <sup>[4]</sup>	1.5 at% N	$T = 180 \text{ °C. } \text{GHSV}(\text{C}_2\text{H}_2) = 50 \text{ h}^{-1},$	$X_A = 60.0\%$ , $Svc_M > 90.0\%$	0.300
		$V_{\rm HCl}/V_{\rm C2H2} = 1.15$		
Ru-in-CNT <sup>[5]</sup>	1 wt% Ru	$T = 170 \text{ °C. } \text{GHSV}(\text{C}_2\text{H}_2) = 90 \text{ h}^{-1},$	$X_A = 99.8\%, Svc_M > 99.0\%$	0.480
		$V_{HCl}/V_{C2H2} = 1.1$		
Ru-O/AC-O	1 wt% Ru	T = 180 °C. GHSV( $C_2H_2$ ) = 180 h <sup>-1</sup> ,	$X_A = 99.6\%, Svc_M > 99.9\%$	0.275
		$V_{\rm HCl}/V_{\rm C2H2} = 1.15$		

## 22 Table S1 The catalytic performance of some potential non-mercury catalysts recently reported in the literatures.

23 <sup>a</sup> X<sub>A</sub> represents the initial conversion of acetylene and S<sub>VCM</sub> represents the selectivity to VCM of the catalyst.

24 <sup>b</sup> Deactivation rate was defined as (the initial maximum C<sub>2</sub>H<sub>2</sub> conversion – the final C<sub>2</sub>H<sub>2</sub> conversion)/(deactivation period, h)

Catalysts	Ru (wt%)	O (wt%)	C (wt%)	Others (wt%)
Fresh AC		13.42	85.75	0.83
Used AC		5.68	93.53	0.79
Fresh AC-O		22.33	77.05	0.62
Used AC-O		15.05	84.37	0.58
Fresh Ru/AC	0.98	11.57	86.98	0.47
Used Ru/AC	0.92	6.46	92.23	0.39
Fresh Ru/AC-O	0.96	18.43	80.02	0.59
Used Ru/AC-O	0.91	13.51	85.06	0.52
Fresh (Ru/AC)-O	0.95	19.32	79.01	0.72
Used (Ru/AC)-O	0.93	14.85	83.56	0.66
Fresh Ru-O/AC-O	0.97	20.29	78.28	0.46
Used Ru-O/AC-O	0.95	17.58	81.09	0.38

**Table S2** The relative content of the elements in the catalysts, determined by XPS spectra.



29 Figure S1. The schematic diagram of four different routes to synthesize Ru-based catalysts.





Figure S3. Nitrogen adsorption-desorption isotherms of the fresh (a) catalysts and used (b) catalysts.



Figure S4. TG curves of fresh and used catalysts recorded in air atmosphere.





**Figure S5.** High-resolution XPS spectra of Ru 3p for the fresh and used Ru-based catalysts.





Figure S6. XPS spectra of C 1s for the fresh catalysts.



- Figure S7. TEM images of the fresh and used catalysts: (a) Fresh Ru/AC, (b) Used Ru/AC, (c)
- Fresh Ru/AC-O, (d) Used Ru/AC-O, (e) Fresh (Ru/AC)-O, (f) Used (Ru/AC)-O, (g) Fresh Ru-O/AC-O, (h) Used Ru-O/AC-O.

## 55 References

- 56 [1] H. Dai, M. Zhu, H. Zhang, F. Yu, C. Wang and B. Dai, J. Ind. & Eng. Chem.,
- 57 2017.
- 58 [2] K. Zhou, J. Jia, C. Li, H. Xu, J. Zhou, G. Luo, F. Wei, Green Chem., 2015, 17,
- *356-364*.
- 60 [3] Q. Song, S. Wang, B. Shen, J. Zhao, Pet. Sci. Technol., 2010, 28, 1825-1833.
- 61 [4] X. Li, J. Zhang and W. Li, J. Ind. & Eng. Chem., 2016, 44, 146-154.
- 62 [5] G. Li, W. Li, H. Zhang, Y. Pu, M. Sun and J. Zhang, Rsc Adv., 2014, 5, 9002-9008.