

## **Supporting Information**

### **Reductive Dechlorination of Endosulfan Isomers and its Metabolites by Zero-Valent Metals: Reaction Mechanism and Degradation Products**

by

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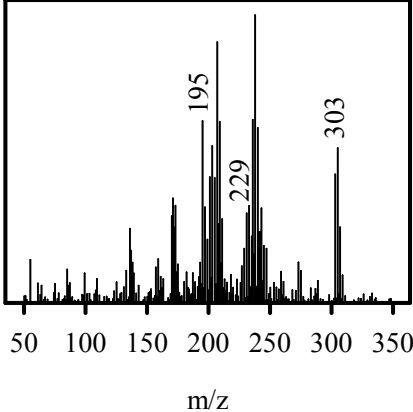
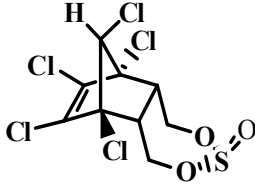
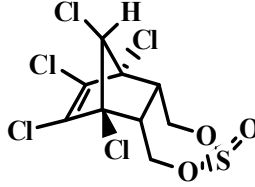
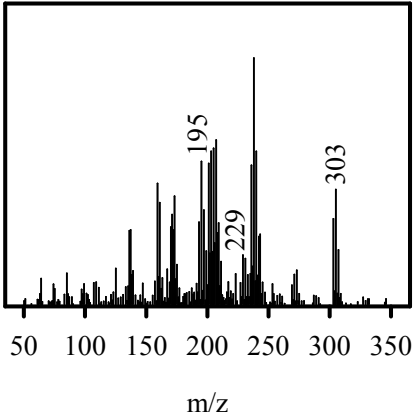
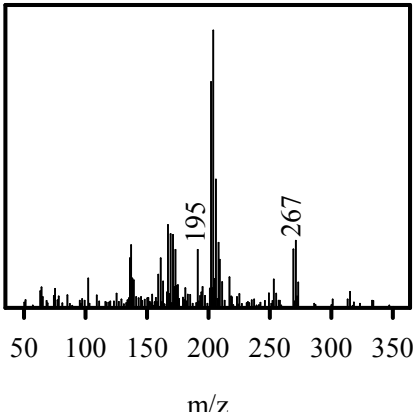
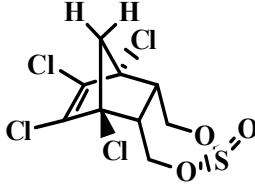
<sup>a</sup> Department of Civil Engineering, Indian Institute of Technology Kanpur-208016,  
India

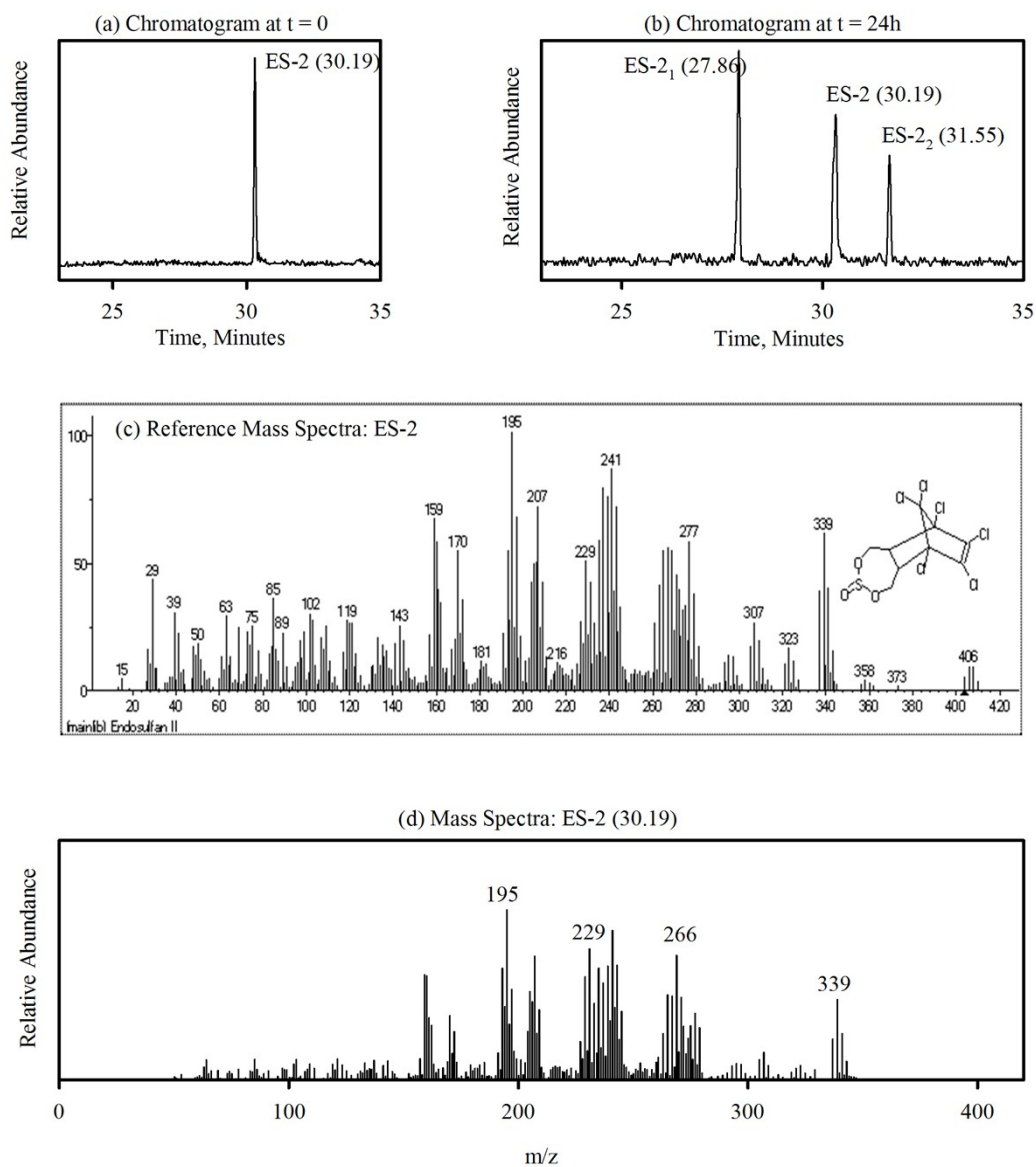
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Water Research

Ben-Gurion University of Negev, Sede-Boqer-89990, Israel

Supplementary data has 17 pages (S1-S17), 7 Figures. (S1-S8), and 7 tables (S1-S7).

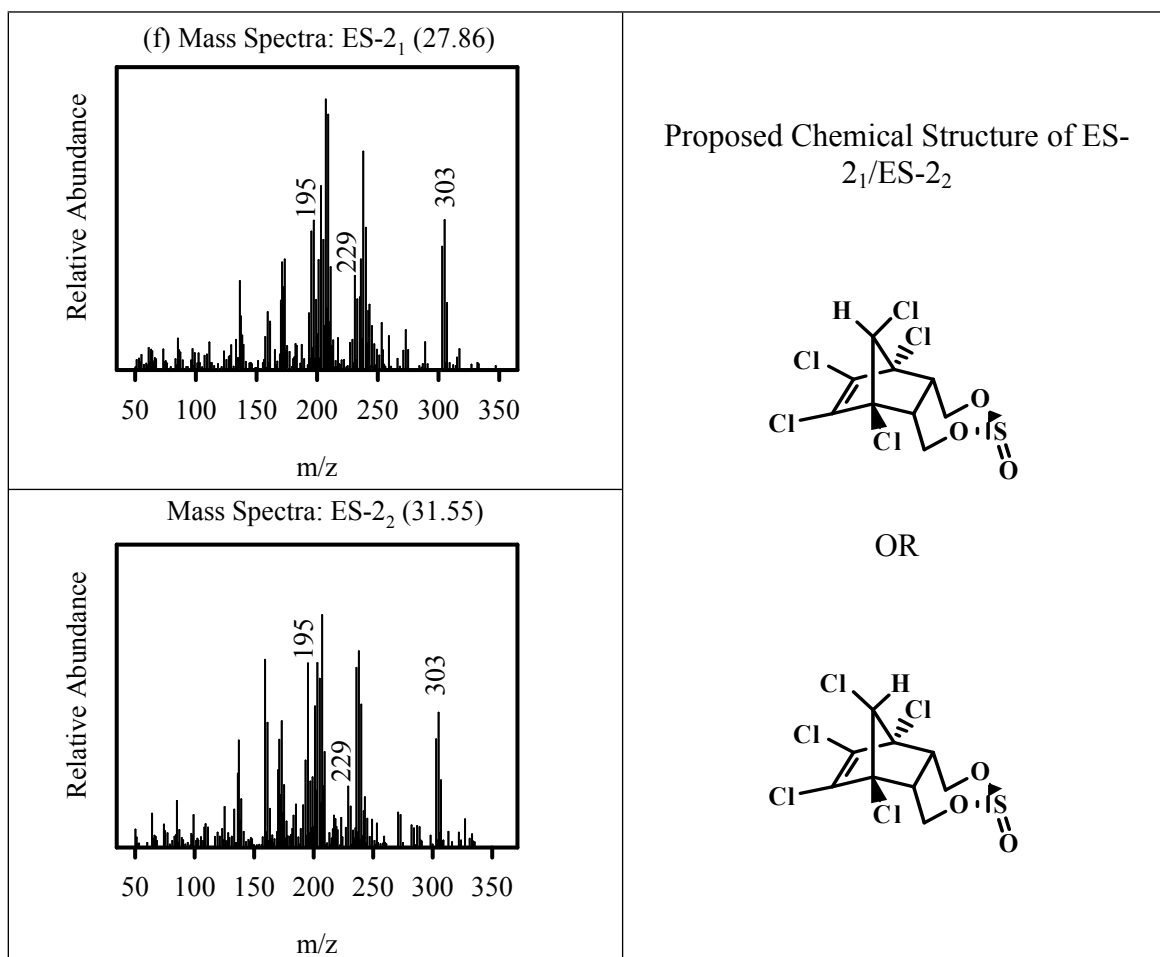
**Table SI.1** Reductive dechlorination of ES-1 by NZVI: degradation products.

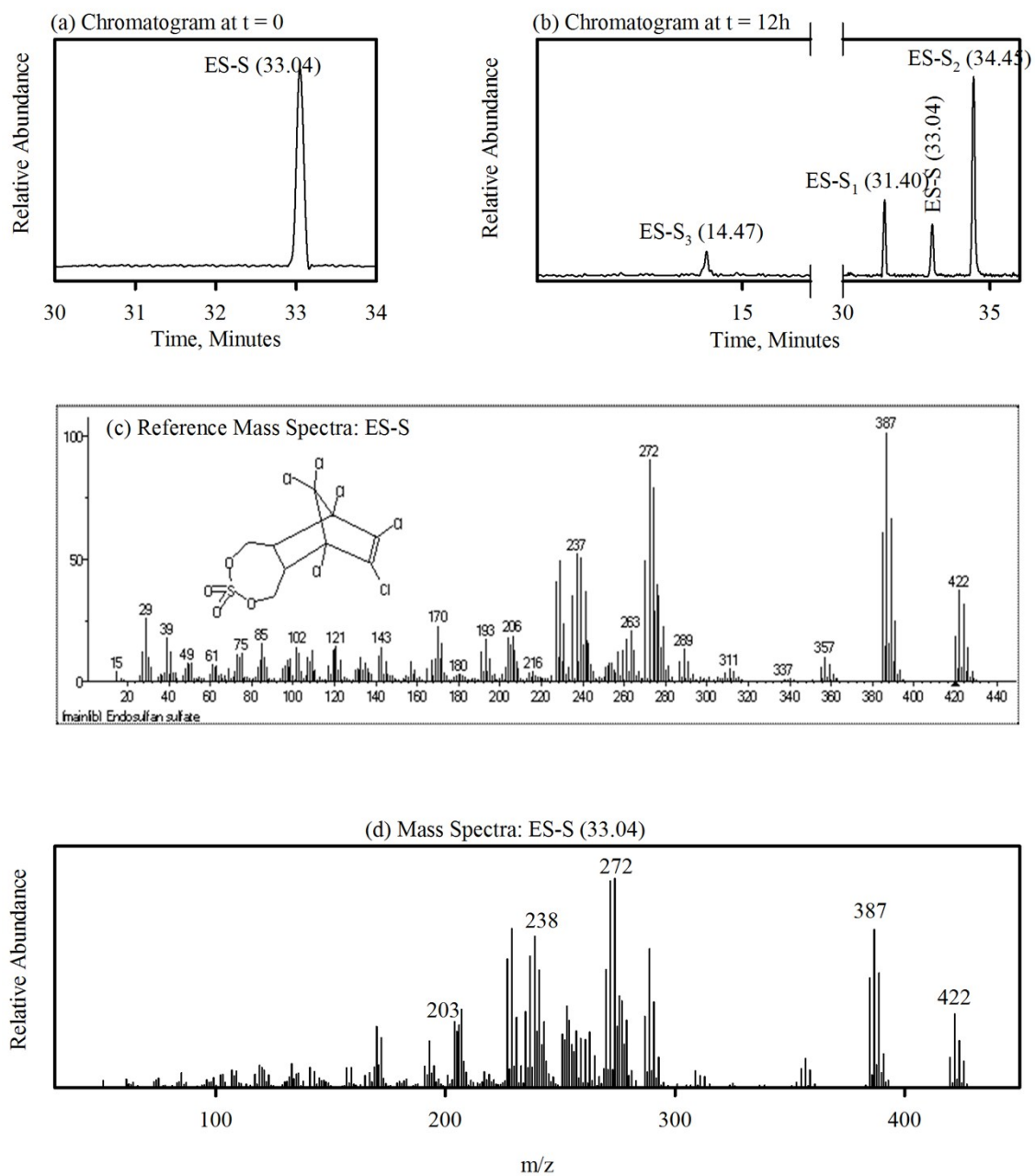
<p>Mass Spectra: ES-1<sub>1</sub> (25.40)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-1<sub>1</sub>/ES-1<sub>2</sub></p>  <p>OR</p> 
<p>Mass Spectra: ES-1<sub>2</sub> (26.65)</p>  <p>Relative Abundance</p> <p>m/z</p>	
<p>Mass Spectra: ES-1<sub>3</sub> (23.95)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-1<sub>3</sub></p> 



**Fig. S1.1.** Reductive dechlorination of ES-2 by NZVI.  
 a) Chromatogram at,  $t = 0$ ; b) Chromatogram at,  $t = 24\text{ h}$   
 c) Mass Spectra of ES-2 from NIST Library  
 d) Mass Spectra of ES-2 obtained in the Present Study

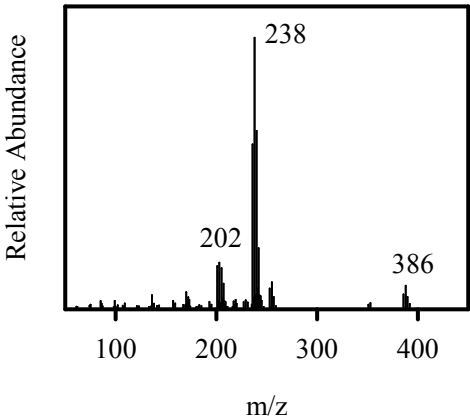
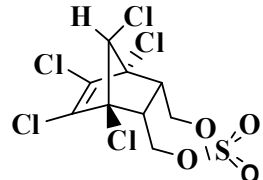
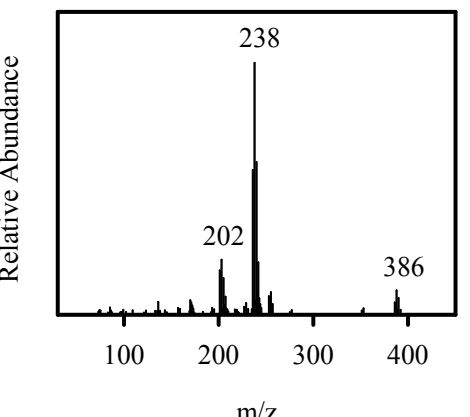
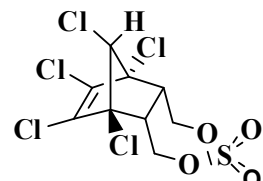
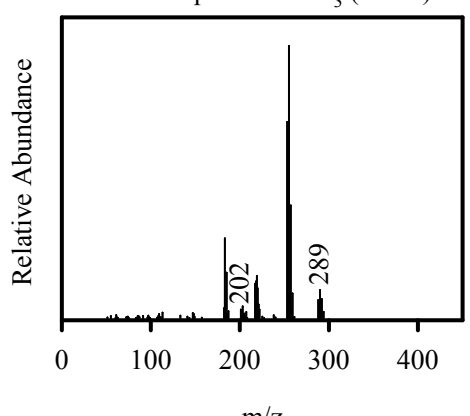
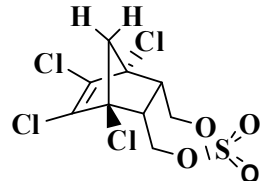
**Table SI.2** Reductive dechlorination of ES-2 by NZVI: degradation products.

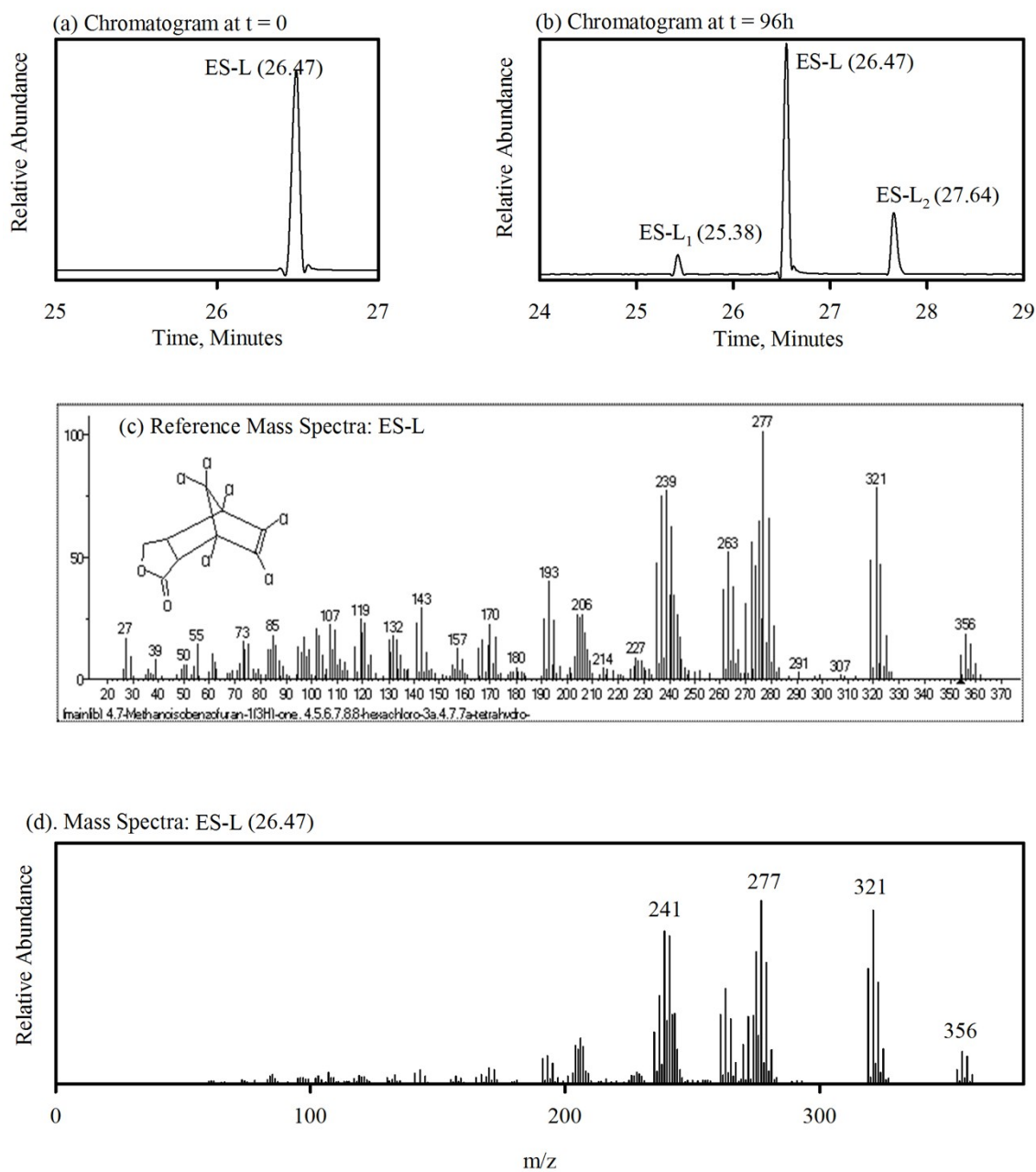




**Fig. SI.2.** Reductive dechlorination of ES-S by NZVI.  
 a) Chromatogram at,  $t = 0$ ; b) Chromatogram at,  $t = 12\text{ h}$   
 c) Mass Spectra of ES -S from NIST Library  
 d) Mass Spectra of ES -S obtained in the Present Study

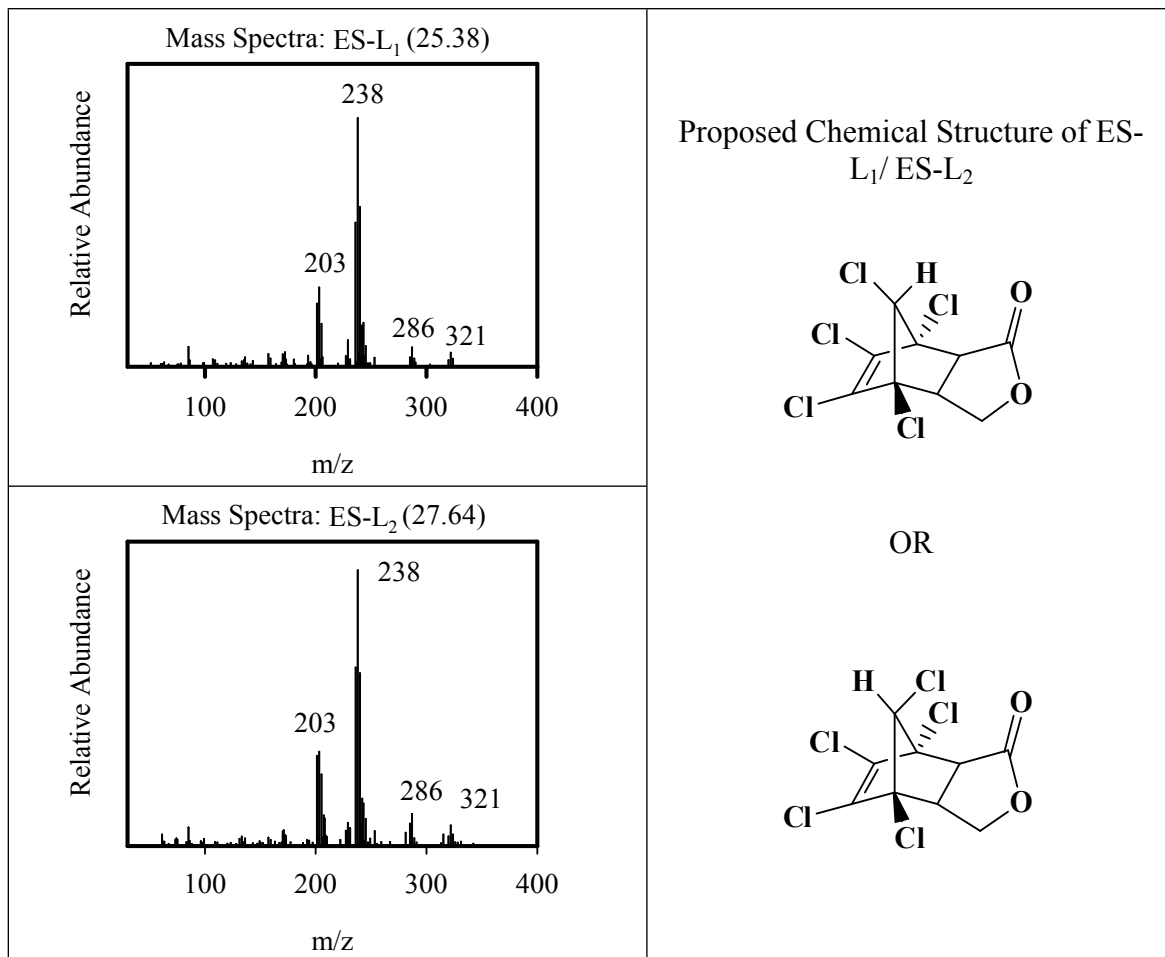
**Table SI.3** Reductive dechlorination of ES-S by NZVI: degradation products

<p>Mass Spectra: ES-S<sub>1</sub> (31.40)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-S<sub>1</sub>/ES-S<sub>2</sub></p> 
<p>Mass Spectra: ES-S<sub>2</sub> (34.45)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>OR</p> 
<p>Mass Spectra: ES-S<sub>3</sub> (14.47)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-S<sub>3</sub></p> 

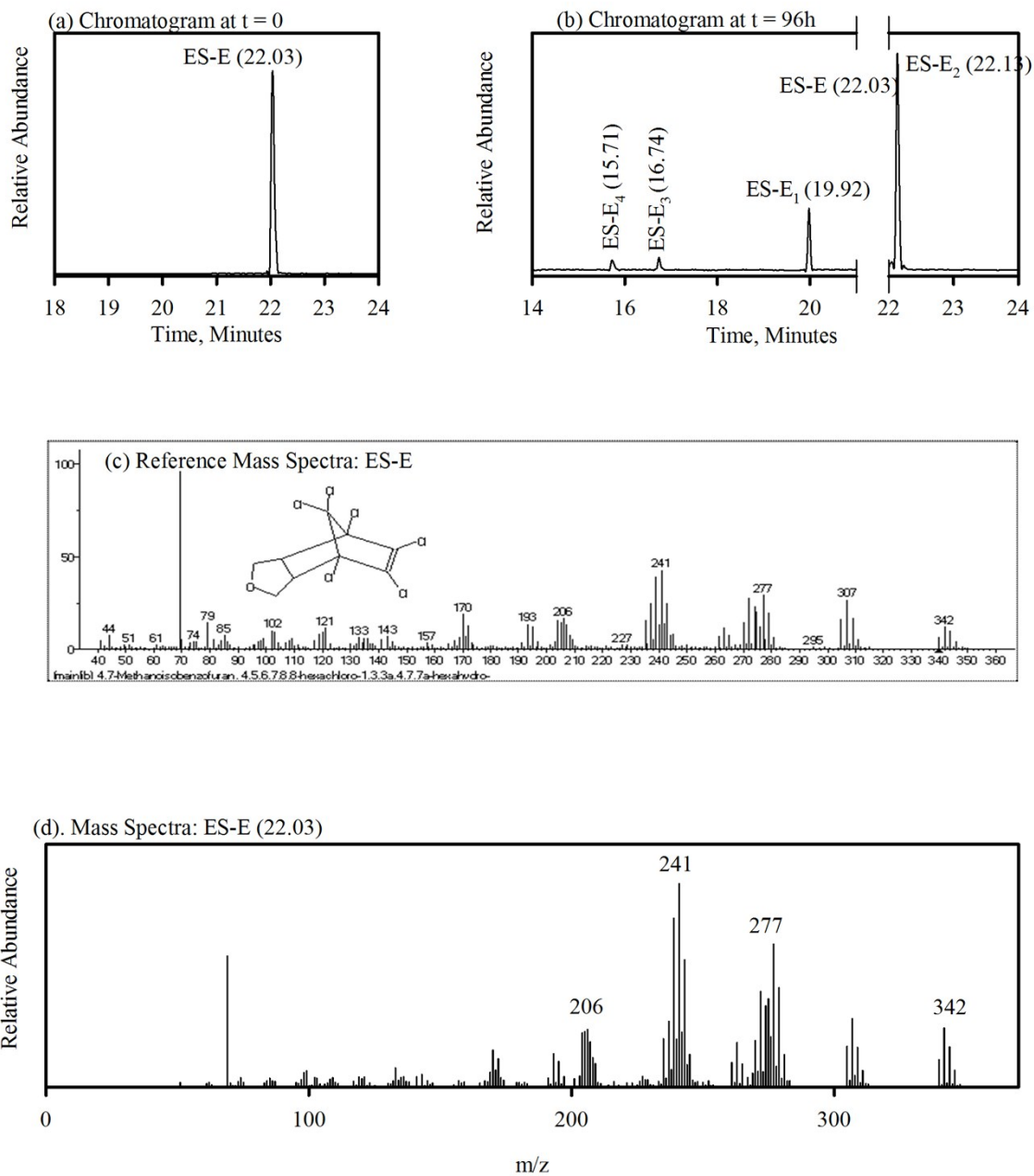


**Fig. SI.3.** Reductive dechlorination of ES-L by NZVI.  
 a) Chromatogram at,  $t = 0$ ; b) Chromatogram at,  $t = 96\text{ h}$   
 c) Mass Spectra of ES -L from NIST Library  
 d) Mass Spectra of ES -L obtained in the Present Study

**Table SI.4** Reductive dechlorination of ES -L by NZVI: degradation products.

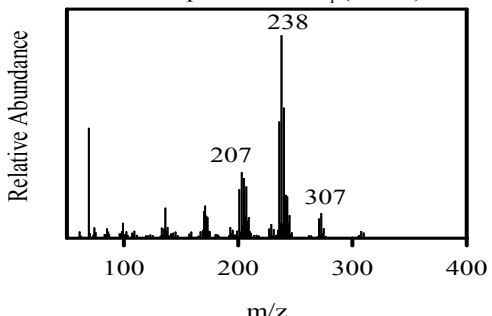
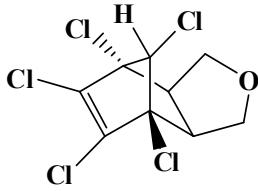
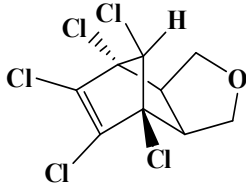
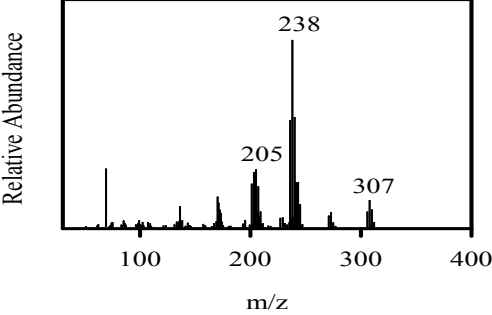
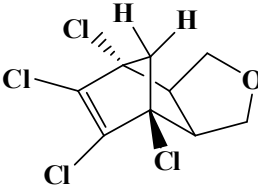
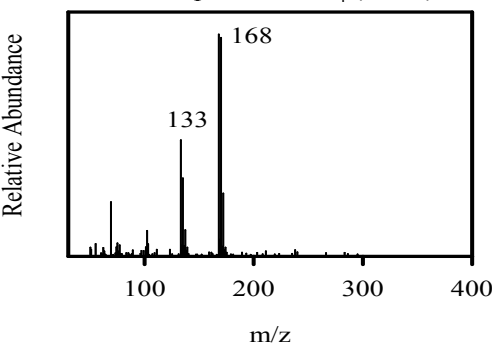
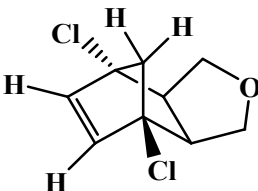




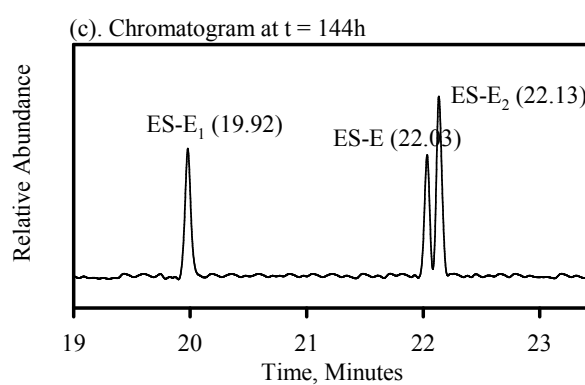
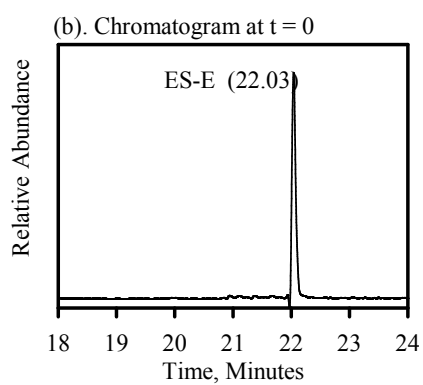
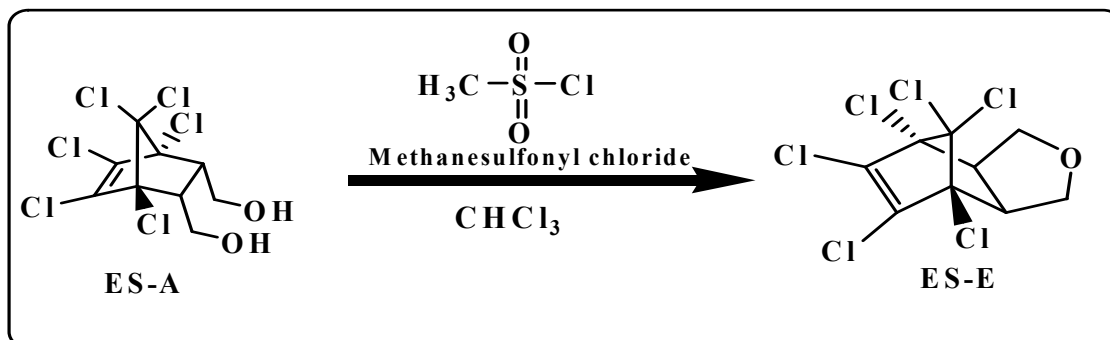


**Fig. SI.4.** Reductive dechlorination of ES-E by NZVI.  
 a) Chromatogram at,  $t = 0$ ; b) Chromatogram at,  $t = 96\text{ h}$   
 c) Mass Spectra of ES -E from NIST Library  
 d) Mass Spectra of ES -E obtained in the Present Study

**Table SI.5** Reductive dechlorination of ES -E by NZVI: degradation products.

<p>Mass Spectra: ES-E<sub>1</sub> (19.92)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>1</sub>/ES-E<sub>2</sub></p>  <p>OR</p> 
<p>Mass Spectra: ES-E<sub>2</sub> (22.13)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>3</sub></p> 
<p>Mass Spectra: ES-E<sub>4</sub> (15.71)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>4</sub></p> 

(a) Derivatization of ES-A

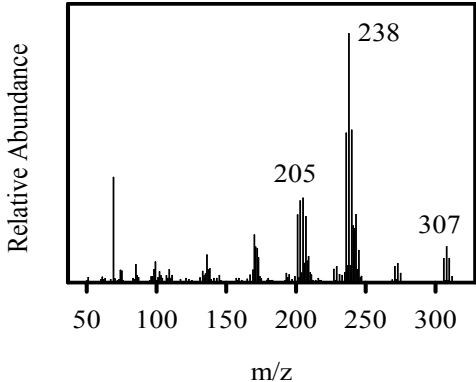
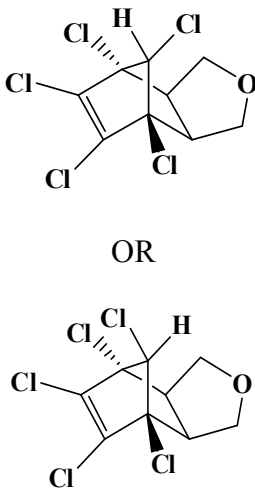
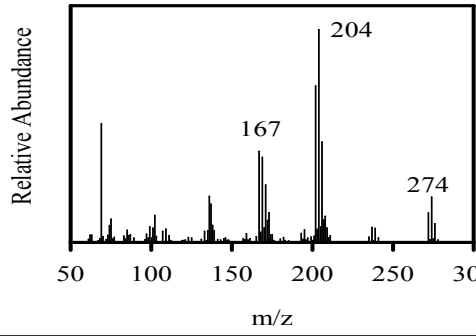
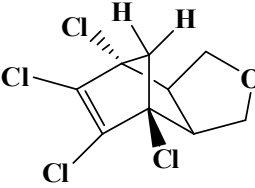
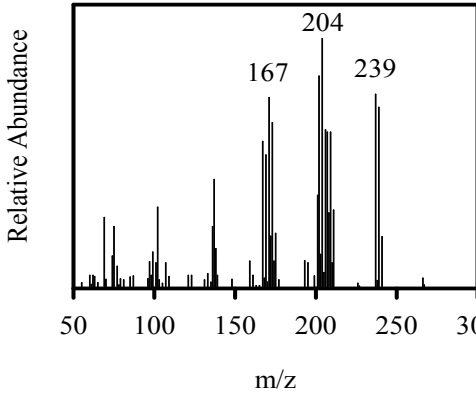
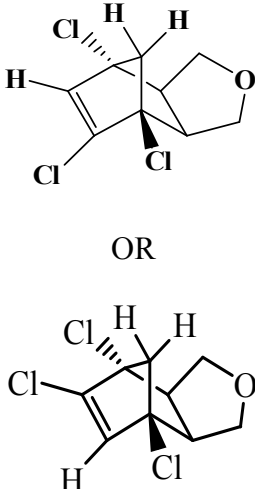


**Fig. SI.5.** Reductive dechlorination of ES-A by NZVI.

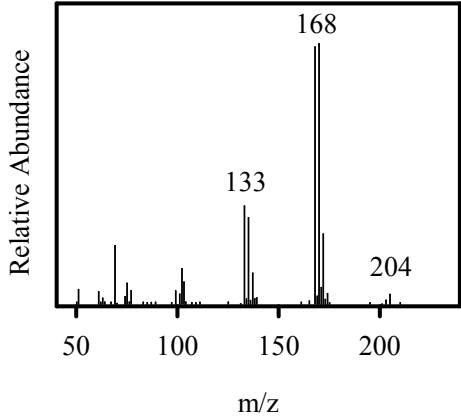
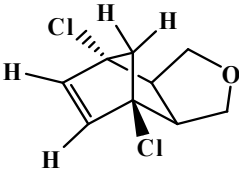
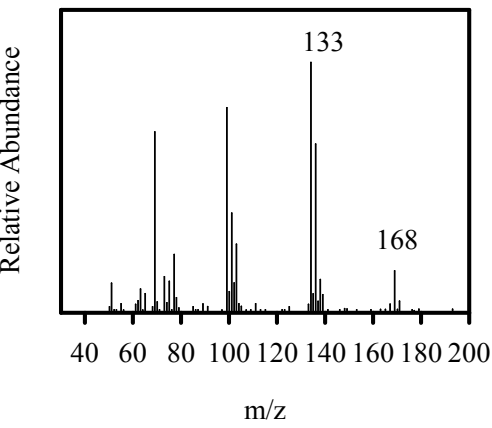
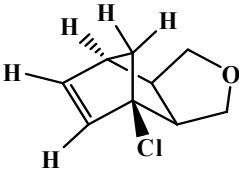
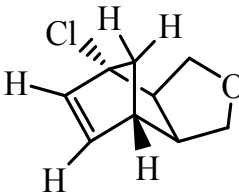
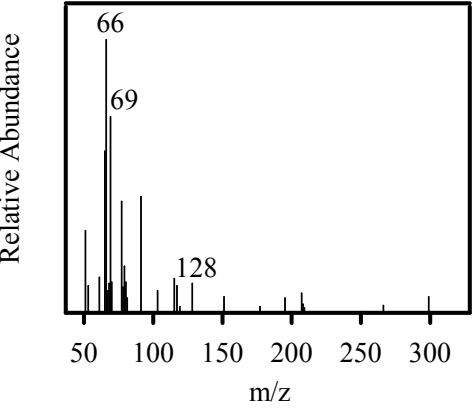
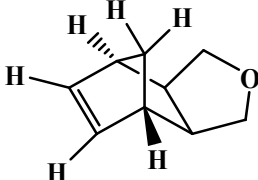
a) Derivatization procedure

b) Chromatogram at,  $t = 0$ ; c) Chromatogram at,  $t = 144\text{ h}$

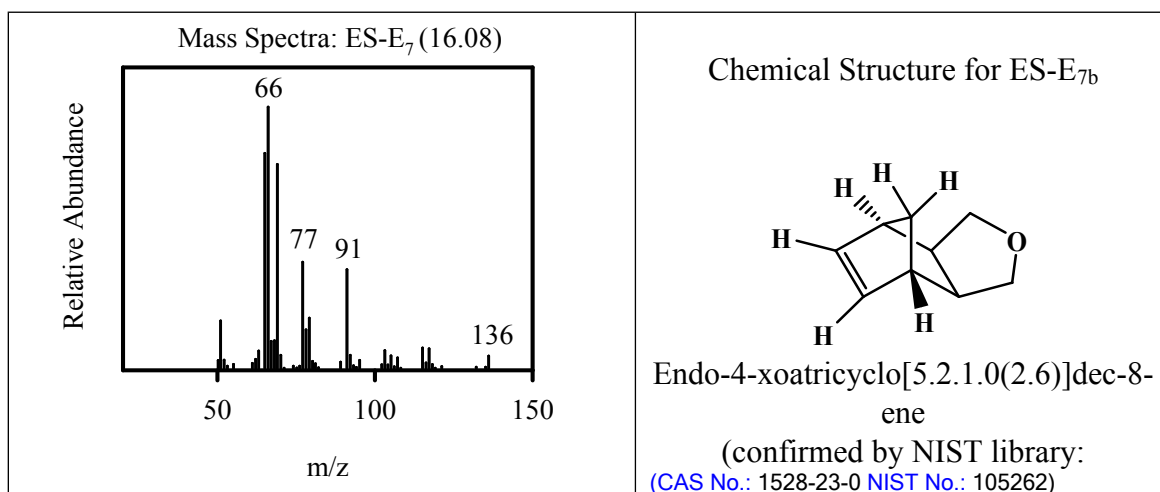
**Table SI.6** Reductive dechlorination of ES -E by Mg<sup>0</sup>: degradation products.

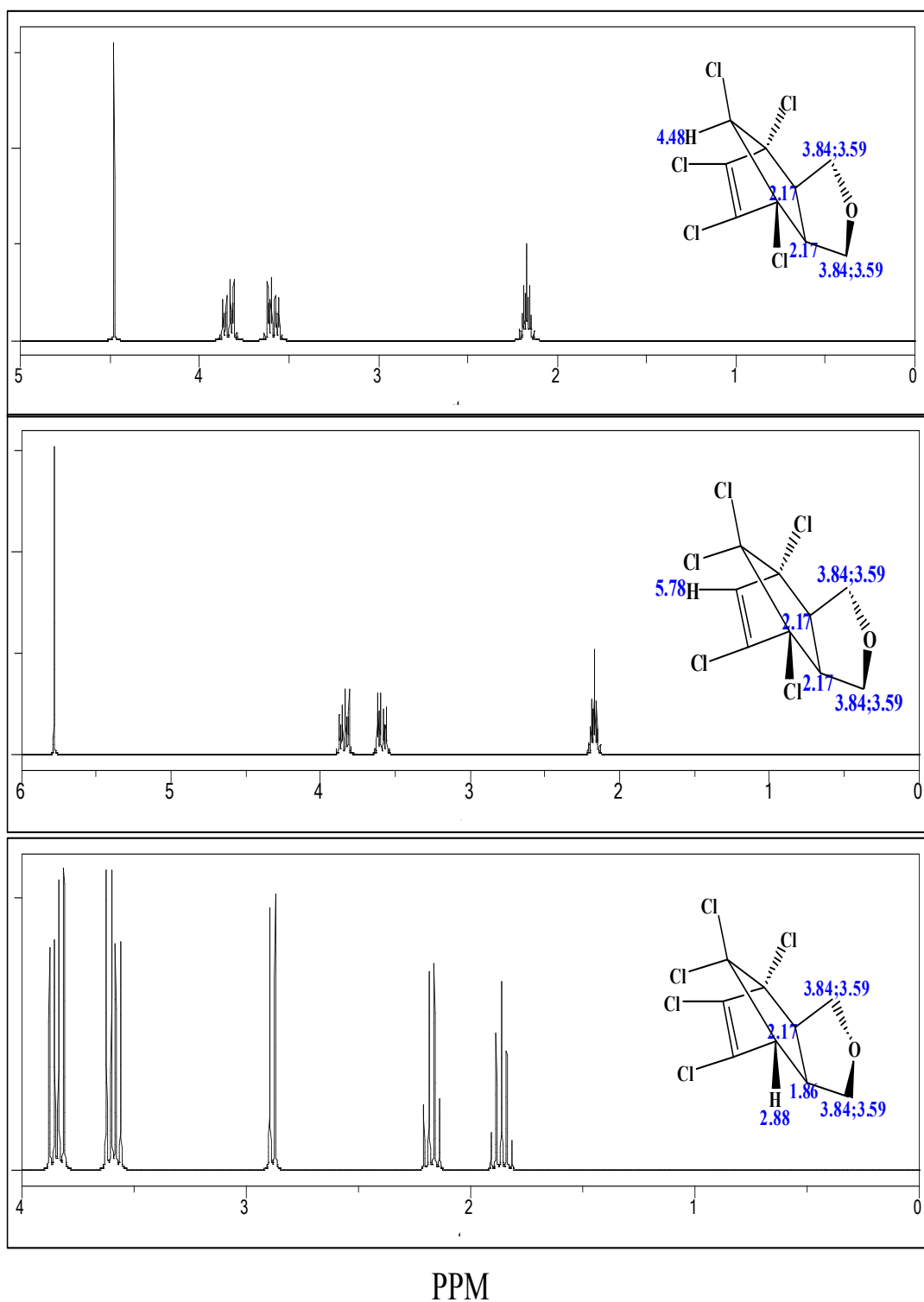
<p>Mass Spectra: ES-E<sub>1a</sub> (25.33)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>1a</sub>/ES-E<sub>2a</sub></p> 
<p>Mass Spectra: ES-E<sub>3a</sub> (23.91)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>3a</sub></p> 
<p>Mass Spectra: ES-E<sub>4a</sub> (22.82)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure of ES-E<sub>4a</sub></p> 

**Table SI.6 (Cont.).** Reductive dechlorination of ES-E by Mg<sup>0</sup>: degradation products.

<p>Mass Spectra: ES-E<sub>5a</sub> (21.11)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure for ES-E<sub>5a</sub></p> 
<p>Mass Spectra: ES-E<sub>6a</sub> (19.84)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure for ES-E<sub>6a</sub></p>  <p>OR</p> 
<p>Mass Spectra: ES-E<sub>7a</sub> (16.08)</p>  <p>Relative Abundance</p> <p>m/z</p>	<p>Proposed Chemical Structure for ES-E<sub>7a</sub> Endo-4-xoatricyclo[5.2.1.0(2.6)]dec-8-ene</p> <p>(confirmed by NIST library: <a href="#">CAS No.:</a> 1528-23-0 <a href="#">NIST No.:</a> 105262</p> 

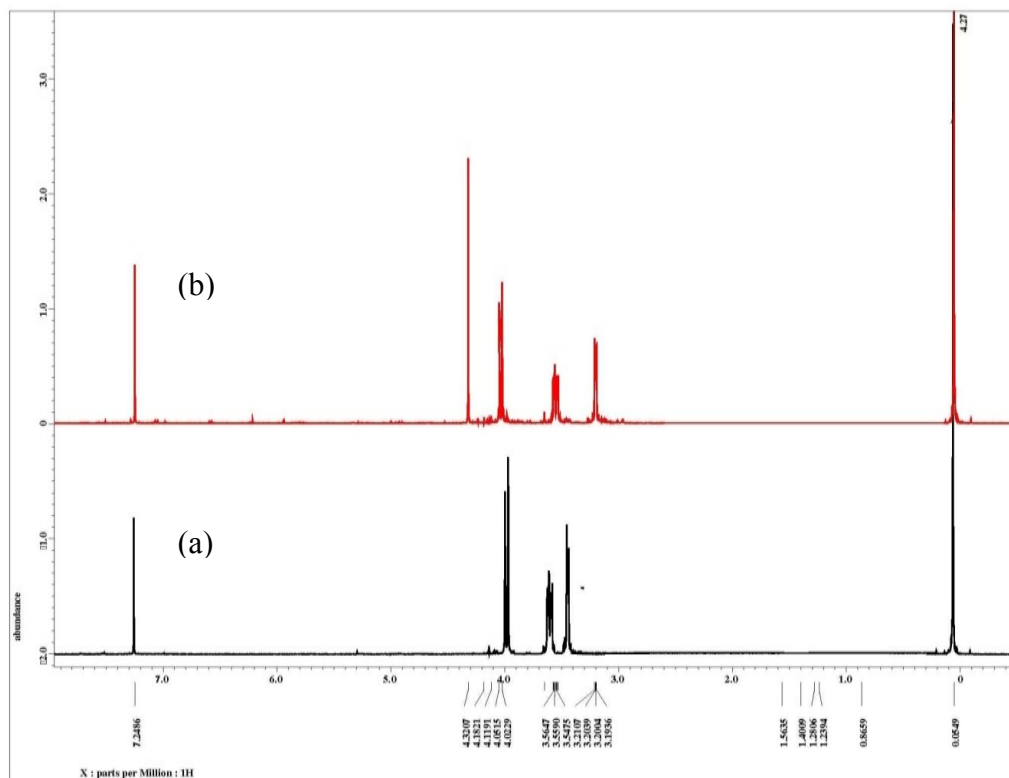
**Table SI.7.** Reductive Dechlorination of ES -E by Na<sup>0</sup>: Degradation Products





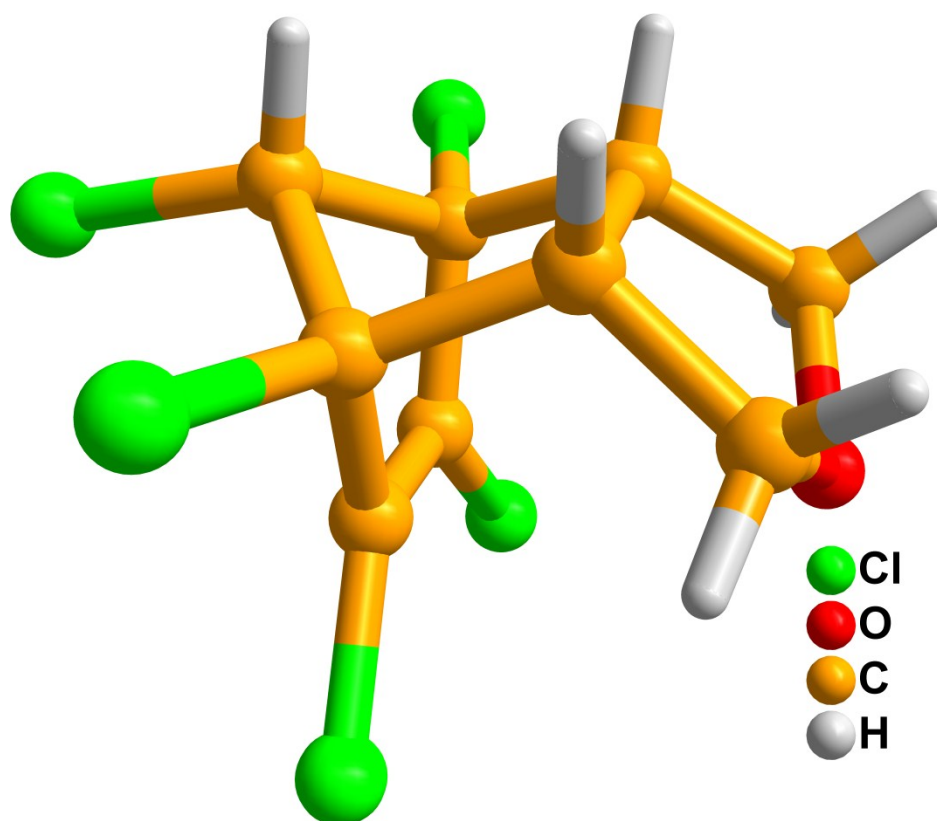
**Fig. SI.6.** Theoretical  $^1\text{H-NMR}$  of dechlorinated ES-E molecules.

- (a) Dechlorination at position-1
- (b) Dechlorination at position -2
- (c) Dechlorination at position -3

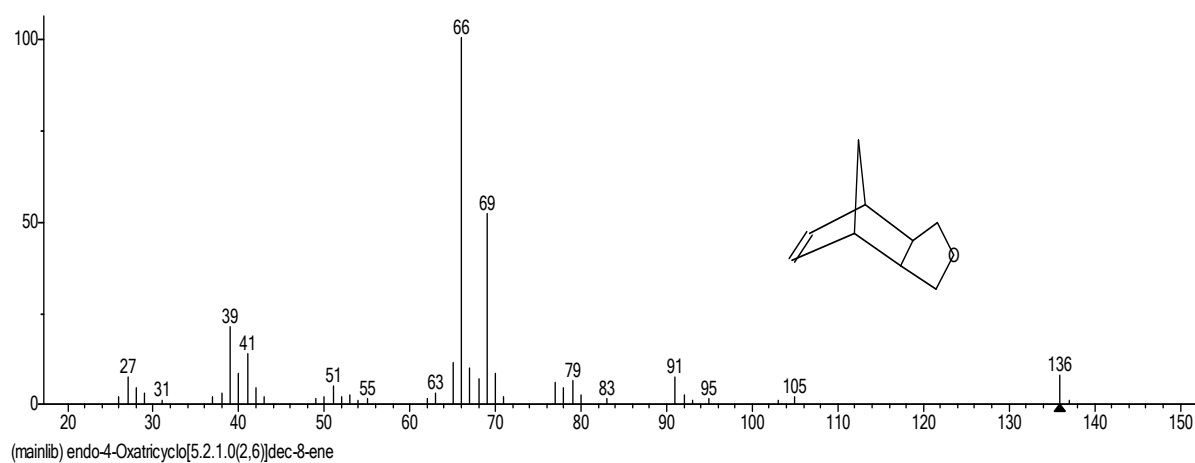


**Fig. SI.7.** Experimental <sup>1</sup>H-NMR for ES-E<sub>1a</sub>  
 (a) ES -E; (b) Dechlorinated ES -E





**Fig. SI.8.** X-ray crystallographic images (Diamond 3.2g) of dechlorinated ES-E<sub>1a</sub> molecules.



**Fig. SI.9** Mass Spectra of Endo-4-oxatricyclo[5.2.1.0(2.6)]dec-8-ene from NIST Library.