

Supporting Information

Tin Doped $\text{PrBaFe}_2\text{O}_{5+\delta}$ Anode Material for Solid Oxide Fuel Cells

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Table S1. Element molar ratios of as-prepared $\text{PrBaFe}_{(2-x)}\text{Sn}_x\text{O}_{5+\delta}$ ($x=0, 0.05, 0.1, 0.15, 0.2, 0.3$) powders obtained from ICP–OES experiment and analysis.

X value of Sn	Nominal compositions	ICP-OES tested results			
		Tested compositions	Pr/Ba	Fe/B a	Sn/Ba
0	$\text{PrBaFe}_2\text{O}_{5+\delta}$	$\text{Pr}_{0.97}\text{Ba}_{0.99}\text{Fe}_{2.00}\text{O}_{5+\delta}$	0.98	2.02	0
0.05	$\text{PrBaFe}_{1.95}\text{Sn}_{0.05}\text{O}_{5+\delta}$	$\text{Pr}_{0.99}\text{Ba}_{0.99}\text{Fe}_{1.93}\text{Sn}_{0.04}\text{O}_{5+\delta}$	1.00	1.95	0.04
0.1	$\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$	$\text{Pr}_{0.98}\text{Ba}_{0.99}\text{Fe}_{1.88}\text{Sn}_{0.09}\text{O}_{5+\delta}$	0.99	1.90	0.09
0.15	$\text{PrBaFe}_{1.85}\text{Sn}_{0.15}\text{O}_{5+\delta}$	$\text{Pr}_{0.99}\text{Ba}_{1.01}\text{Fe}_{1.86}\text{Sn}_{0.15}\text{O}_{5+\delta}$	0.98	1.84	0.15
0.2	$\text{PrBaFe}_{1.8}\text{Sn}_{0.2}\text{O}_{5+\delta}$	$\text{Pr}_{0.98}\text{Ba}_{0.98}\text{Fe}_{1.77}\text{Sn}_{0.20}\text{O}_{5+\delta}$	1.00	1.81	0.20
0.3	$\text{PrBaFe}_{1.7}\text{Sn}_{0.3}\text{O}_{5+\delta}$	$\text{Pr}_{0.99}\text{Ba}_{1.01}\text{Fe}_{1.74}\text{Sn}_{0.29}\text{O}_{5+\delta}$	0.98	1.72	0.29

Table S2. The oxygen content ($5+\delta$) in the as-prepared $\text{PrBaFe}_{(2-x)}\text{Sn}_x\text{O}_{5+\delta}$ ($x=0, 0.05, 0.1, 0.15, 0.2, 0.3$) and after reducing treatment in humidified gas mixture of 10 % H_2 + 90% N_2 at 800 °C for 24h.

x value of Sn	Oxygen content ($5+\delta$)	
	As-prepared	After reducing
0	5.7688	5.5853
0.05	5.8211	5.5571
0.1	5.8216	5.496
0.15	5.8386	5.4417
0.2	5.7605	5.4305
0.3	5.7458	5.3571

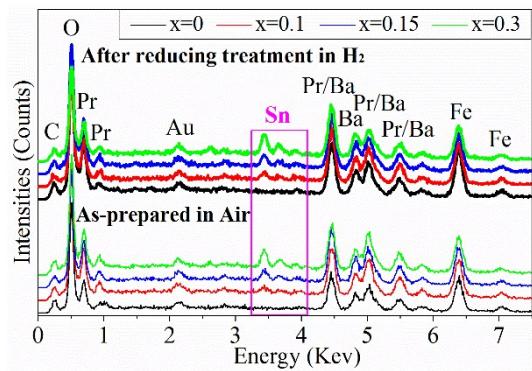


Figure S1. The EDS spectra of as-prepared $\text{PrBaFe}_{(2-x)}\text{Sn}_x\text{O}_{5+\delta}$ ($x=0, 0.1, 0.15, 0.3$) powders and after reducing treatment (humidified gas mixture of 10% H_2 + 90% N_2) at 800 °C for 24h.

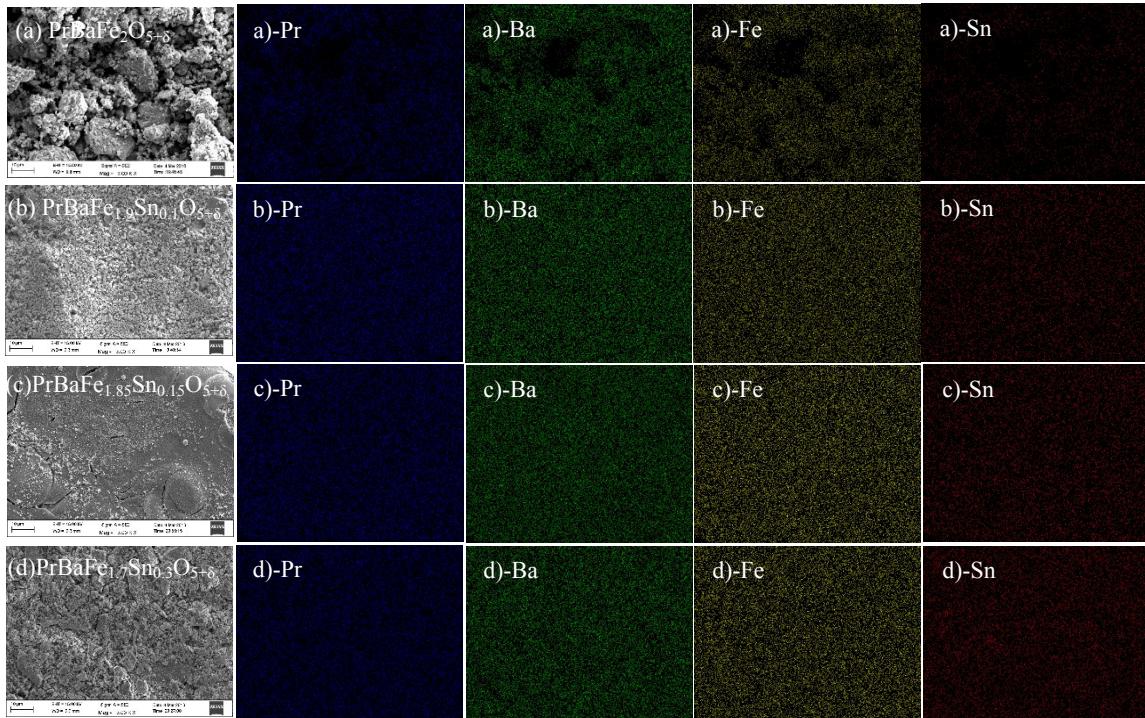


Figure S2. SEM images and EDS element mappings of as-prepared $\text{PrBaFe}_{(2-x)}\text{Sn}_x\text{O}_{5+\delta}$ ($x=0, 0.1, 0.15, 0.3$) powders. (a) $x=0$; (b) $x=0.1$; (c) $x=0.15$; (d) $x=0.3$. The scale length in all SEM images is 10 μm .

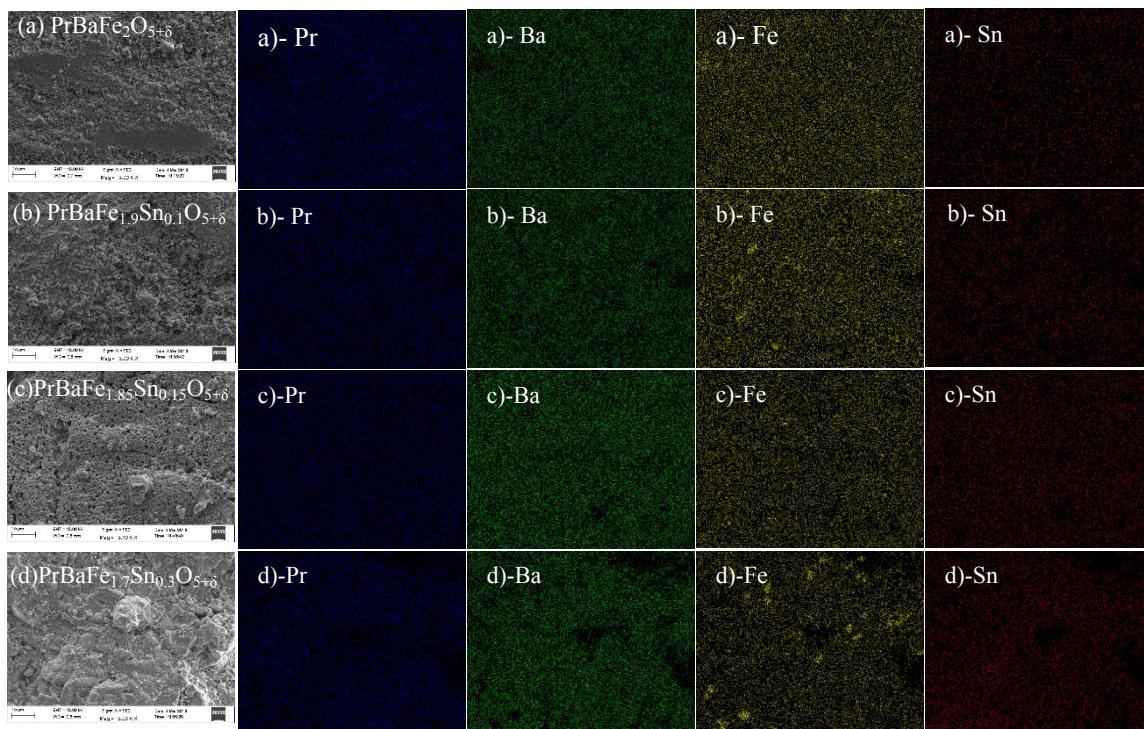


Figure S3. SEM images and EDS element mappings of $\text{PrBaFe}_{(2-x)}\text{Sn}_x\text{O}_{5+\delta}$ ($x=0, 0.1, 0.15, 0.3$) powders after reducing treatment at 800°C for 24 h in the humidified gas mixture of 10% H_2 + 90% N_2 . (a) $x=0$; (b) $x=0.1$; (c) $x=0.15$; (d) $x=0.3$. The scale length in all SEM images are 10 μm .

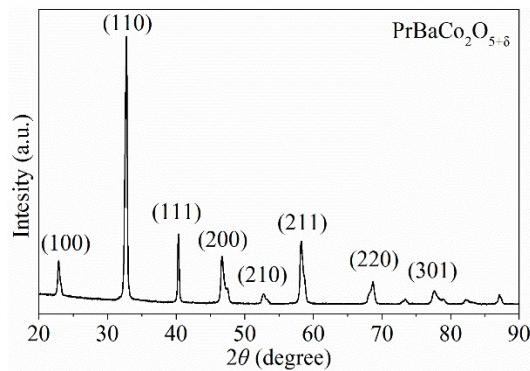


Figure S4. The XRD patterns of as-synthesized $\text{PrBaCo}_2\text{O}_{5+\delta}$ powder samples calcinated at 950°C in air for 5 h.

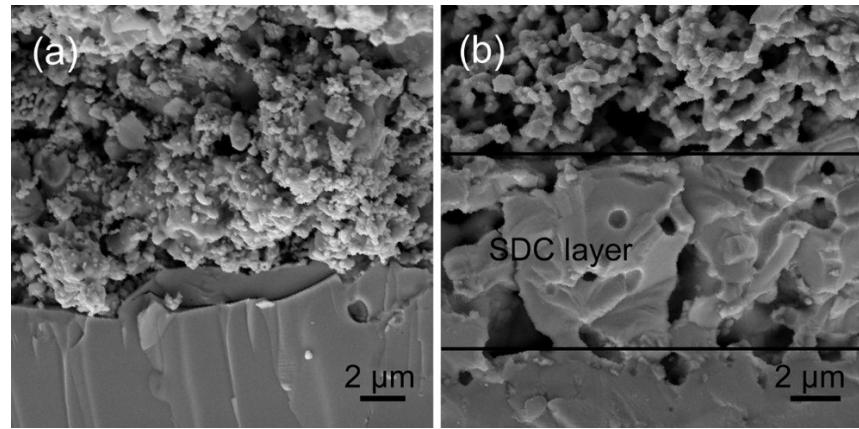


Figure S5. SEM micrographs of the single cell with configuration of $\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ | LSGM | SDC | PBCO after testing: Pr $\text{BaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ anode (a) and SDC buffer layer and PBCO cathode (b).

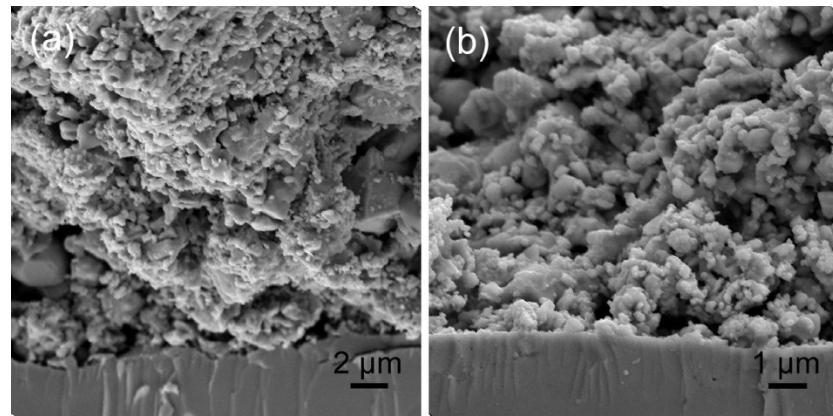


Figure S6. SEM micrographs of the symmetrical cell with configuration of $\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ | LSGM | $\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ after testing: $\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ anode (a) and $\text{PrBaFe}_{1.9}\text{Sn}_{0.1}\text{O}_{5+\delta}$ cathode (a).